



# Systematic Review Effect of Different Irrigating Solutions on Root Canal Dentin Microhardness—A Systematic Review with Meta-Analysis

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Abstract: The aim of this study was to evaluate the effect of different irrigating solutions as well as their combination and activation modes on root canal dentin microhardness. The protocol was registered in PROSPERO and PRISMA guidelines were followed. The structured question was as follows: "Which type of irrigating solution used in endodontic treatment causes more change in dentin microhardness?" The literature was screened via PubMed, Google Scholar, Scopus, and Science Direct. The last search was carried out in February 2023 with English language restriction. Two reviewers independently performed screening and evaluation of articles. A total of 470 articles were retrieved from all the databases, whereas only 114 articles were selected for full-text analysis. After applying eligibility criteria, 44 studies were evaluated and included in this review. The results showed that with increased contact time with irrigants, dentin microhardness decreases. Increased contact time with sodium hypochlorite (NaOCl) was associated with more reduction in dentin microhardness compared with other irrigants. Other irrigants, with the exception of distilled water, including EDTA, citric acid, herbal irrigants, glycolic acid, phytic acid, etc., in this study significantly decreased dentin microhardness. The maximum reduction in dentin microhardness was seen with 2.5% NaOCl after 15 min of contact time. The use of irrigating solutions alters the chemical composition of dentin, thereby decreasing its microhardness, which affects the clinical performance of endodontically treated teeth.

Keywords: endodontic irrigants; root dentin microhardness; root canal irrigants; systematic review

# 1. Introduction

Endodontic therapy relies crucially on the thorough chemo-mechanical preparation of the root canal system, which combines precise instrumentation with the application of effective irrigating solutions [1]. Irrigation is fundamental not only during the mechanical shaping but also subsequently, as it aids in removing microorganisms, tissue fragments, and dentinal debris via a flushing action [2]. It also helps avoid the accumulation of debris in the apical zone and the spread of infection to the periapical tissues [3].

The complex anatomy of root canals, with their varied shapes, narrow fins, isthmuses, and lateral extensions, often hinders complete debridement with instruments alone [4].



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2 of 25

This underscores the importance of irrigation for ensuring the entire root canal is free from bacterial contamination and is an essential step for a successful endodontic outcome [5].

Commonly used endodontic irrigating agents include citric acid, hydrogen peroxide  $(H_2O_2)$ , ethylenediaminetetraacetic acid (EDTA), sodium hypochlorite (NaOCl), and chlorhexidine (CHX) [6,7]. These substances offer a spectrum of beneficial actions, from antimicrobial effects to removing the smear layer and dissolving organic tissues [6]. Significantly, NaOCl, CHX, and EDTA are preferred due to their ability to dissolve organic tissue, eliminate the smear layer, and exhibit potent antimicrobial effects [8,9].

However, these solutions can also alter the chemical structure of dentin, particularly the calcium content in its hydroxyapatite crystals, which can subsequently influence key tooth properties like microhardness [10]. By evaluating dentin microhardness, we can infer changes in the physical and chemical properties of dentin such as the mineral content and modulus of elasticity of dentin [11]. Reduced dentin microhardness leads to a reduction in the modulus of elasticity of dentin [12].

Dentin microhardness measurement assesses the alteration in the calcium–phosphorus ration of the dentin structure. This provides indirect evidence of mineral loss or gain in the dental hard tissue [11].

This review rigorously investigates not only the direct effects of these solutions, but also delves into the methodologies, potential synergistic effects of combined irrigation protocols, and the role of activation methods. This comprehensive review of multiple databases aims to bridge the gap in the existing literature, providing a robust foundation for future research.

### 2. Materials and Methods

The protocol for this systematic review is registered with PROSPERO under the registration number CRD42022354739. This review was carried out following the Preferred Reporting Items for Systematic Reviews and Meta-Analyses (PRISMA) statement guidelines [13].

#### 2.1. Eligibility Criteria

The eligibility criteria are presented in Table 1.

Table 1. Eligibility criteria.

	Inclusion Criteria	Exclusion Criteria							
•	In vitro studies investigating different irrigating solutions and their impact on dentin microhardness.	•	Review articles, letters to the editor, clinical studies, and case reports/case series.						
•	Studies published in the English language.	•	Articles investigating changes in other parameters (surface roughness, erosion, flexural strength, etc.) and not including the microhardness of dentin.						
•	Studies conducted on permanent healthy human tooth specimens.	•	Studies involving deciduous human teeth and bovine teeth.						

# Focused PICO Question

The research question was formulated as follows:

- Population (P): extracted healthy human permanent teeth;
- Intervention (I): the application of various irrigating solutions in endodontic therapy;
- Comparison (C): various irrigating solutions;
- Outcome (O): dentin microhardness.

#### 2.2. Literature Search

The search strategy followed PRISMA guidelines (Table 2). An electronic literature search was executed across four prominent databases: PubMed, Google Scholar, Science Direct, and Scopus, up to February 1, 2023. The search was restricted to articles published in the English language.

Database	Search Strategy
PubMed	(((((((root canal dentin) OR (radicular dentin)) OR (radicular dentinal surface)) AND (irrigating solution)) OR (irrigation)) OR (root canal irrigation)) OR (EDTA)) OR (CHX)) OR (sodium hypochlorite)) AND (microhardness)
Google Scholar	root canal dentin OR radicular dentin OR radicular dentinal surface AND irrigating solution OR irrigation OR root canal irrigants AND EDTA AND CHX AND NaOCl AND microhardness
Science Direct	root canal dentin AND irrigating solution AND EDTA AND CHX AND NaOCl AND microhardness
Scopus	root canal dentin AND irrigating solution AND microhardness

 Table 2. Search strategy.

The search strategy incorporated the following keywords: "root canal dentin," "radicular dentin", "radicular dentinal surface", "root dentin", "irrigating solution", "irrigation", "root canal irrigation", "root dentin irrigation", "EDTA", "CHX", "chlorhexidine", "NaOCl", "sodium hypochlorite", and "microhardness". After identifying the relevant articles, a thorough screening process was undertaken to determine which studies would be included in the review.

Search results were imported into a reference manager software (Ravman, version 5, Boston, MA, USA), where duplicates were removed by S.A. and L.M. Titles and abstracts were then reviewed against the inclusion criteria, and studies meeting the criteria proceeded to full-text screening for qualitative synthesis.

#### 2.3. Data Extraction

Data extraction involved three main categories: study characteristics, methodology, and outcomes/results. Study characteristics encompassed author names and publication years. Methodological variables included sample size, tooth specimens, tooth sectioning, irrigation protocol, microhardness test details (such as load and time), and percentage. Outcome variables comprised dentin microhardness levels at different points and changes in microhardness. Mean and standard deviation values were also documented from the included studies.

#### 2.4. Quality Assessment

The quality assessment tool for in vitro studies (QUIN tool) checklist for reporting in vitro studies was used to evaluate the internal methodological quality (risk of bias) of the included studies resulting from the selection process. Each of the 12 parameters considered in the quality assessment tool was assessed for individual studies and then the percentage of complied items was calculated as (score  $\times 100/2 \times$  number of criteria applied).

#### 3. Results

The initial electronic database search found 470 articles. After removing duplicates, the total was reduced to 230. Subsequent screening based on abstracts and titles resulted in a further assessment of 114 articles. Finally, 44 full-text articles met the eligibility criteria for this study (Figure 1).

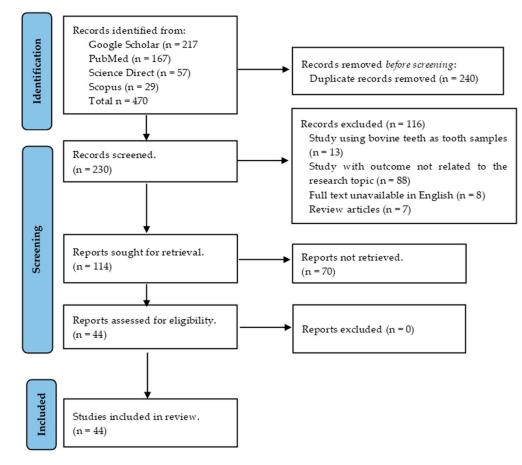


Figure 1. PRISMA flow diagram of literature search and selection process.

#### 3.1. Sample Size and Preparation

In total, 2267 healthy, extracted human teeth and 4534 sectioned tooth samples were utilized in the included studies. These predominantly employed single-rooted teeth like maxillary and mandibular incisors, canines, and mandibular premolars. The tooth sectioning process varied among studies using techniques such as diamond discs, low-speed and high-speed burs, cutting machines, and diamond saws. Most of the microhardness assessments were performed on longitudinally sectioned teeth with a few [11,14–17] opting for transverse cross-sections. Different storage media were used for tooth specimens including buffered saline [5,14,17–25] and 0.1% thymol [1,12,21,26–29] being the most common choices across different studies.

#### 3.2. Microhardness Testing Tools

Microhardness testing was performed using either Vickers or Knoop diamond indenters. The majority of studies utilized Vickers diamond indenters, with a few exceptions that employed Knoop indenters [22,30–32].

#### 3.3. Irrigating Solution Evaluation

Various irrigating solutions were evaluated for their impact on dentin microhardness (Table 3), with sodium hypochlorite and ethylenediaminetetraacetic acid being the most frequently studied solutions. They were tested at different concentrations and contact times.

		Sample Size, Type of			Load (g) Given		Ν	Aicrohardness V	alue (Mean $\pm$ SI	))		
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	vical	Mi	ddle	Apical		- MHN Test
		in Each Group			Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	-
			Saline	30		$46.6\pm 6.3$	$46.0\pm5.2$	$46.9\pm5.1$	$45.1\pm3.7$	$47.9\pm 6.8$	$43.7\pm7.3$	
			5% NaOCl + 18% HEBP	30	-	$43.7\pm5.0$	$36.2\pm5.4$	$45.5\pm5.5$	$35.7\pm4.1$	$46.1\pm3.7$	$40.0\pm5.7$	-
			2.5% NaOCl	30	-	$44.7\pm3.5$	$38.7\pm3.8$	$44.9\pm5.0$	$39.8 \pm 2.9$	$45.2\pm2.8$	$40.7\pm5.0$	-
			2.5% NaOCl + 17% EDTA	30 + 3	-	$47.5\pm6.4$	$30.7\pm3.5$	$47.3\pm3.7$	$34.5\pm5.4$	$47.2\pm3.6$	$35.3\pm4.0$	-
			2.5% NaOCl + 10% CA	30 + 3	-	$43.7\pm3.4$	$31.5\pm4.9$	$45.2\pm3.5$	$31.4\pm7.4$	$45.4\pm7.0$	$30.2\pm5.4$	_
1.	Tartari et al. 2013 [22]	45 SRT LS	2.5% NaOCl + 9% HEBP	30 + 5	25 and 15 s	$45.9\pm4.8$	$41.4\pm4.9$	$47.7\pm4.6$	$42.6\pm3.0$	$46.4\pm6.1$	$39.6\pm5.8$	- KHN
			2.5% NaOCL + 17% EDTA + 2.5% NaOCl	30 + 3 + 3		47.5 ± 6.4	$30.2\pm3.91$	$47.3\pm3.7$	$34.4\pm5.4$	47.2 ± 3.6	$35.7\pm5.2$	_
			2.5% NaOCl + 10% CA + 2.5% NaOCl	30 + 3 + 3	-	43.7 ± 1.8	$31.9\pm 6.8$	$45.6\pm2.9$	$29.8\pm 6.4$	$45.1\pm7.5$	$28.0\pm3.6$	-
			2.5% NaOCl + 9% HEBP + 2.5% NaOCl	30 + 5 + 3		45.9 ± 4.8	$39.1\pm4.76$	$47.7\pm4.6$	$41.8\pm4.2$	$46.4\pm6.1$	39.4 ± 4.9	-
							Pre Rx			After Rx		
			2.5% NaOCl + 5% EDTA	20 + 1			$66.01 \pm 5.66$			$56.69 \pm 1.21$		
			2.5% NaOCl + 15% EDTA	20 + 1	-		$66.15\pm5.58$			$59.76\pm3.42$		-
2	Pedersen et al.	24 Molars	2.5% NaOCl	2	-		$66.01 \pm 5.75$			$53.80\pm3.54$		-
2.	2020 [33]	LS	5% EDTA	1	- 300 and 20 s		$65.33\pm 6.88$			$65.18 \pm 5.52$		VHN
			15% EDTA	1	-		$65.59\pm 6.65$			$67.38 \pm 3.35$		-
			Saline	20	-		$65.72\pm8.17$			$65.33 \pm 8.46$		-
			1.3% NaOCl + 17% EDTA	20 + 1						$51.63 \pm 0.86$		_
3.	Dineshkumar	40 Mand PM	1.3% NaOCl + MTAD	20 + 5	- 300 and 20 s					$42.85 \pm 0.99$		
5.	et al. 2012 [34]	LS	1.3% NaOCl + HEBP	20 + 5	- 500 and 20 S					$53.74 \pm 1.18$		- VHN
			Distilled water	20	-					$66.65 \pm 1.04$		-
			1% PAA	15						$17.29\pm3.71$		
	Keine et al.	40 SRT	2.5% NaOCl	15	- -					$7.90 \pm 1.94$		_
4.	2019 [35]	LS	2.5% NaOCl + 17% EDTA + 2.5% NaOCl	15 + 3 + 1	25 and 10 s					$17.95\pm3.40$		- KHN
			0.9% saline (control group)	15	-					$0.37\pm0.24$		-

Table 3. Root dentin microhardness after contact with different irrigating solutions—data extraction from included studies.

		Sample Size, Type of			Load (g) Given	Microhardness Value (Mean $\pm$ SD)						
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Ce	ervical	Mi	ddle	Api	cal	
		in Each Group			Dwell Time -	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_
			3% NaOCl				$57.15 \pm 1.75$			$55.15 \pm 1.86$		
5.	Saha et al. 2017	80 PM	17% EDTA	15	- 300 and 20 s		$56.88 \pm 1.38$			$43.12\pm2.51$		– VHN
5.	[12]	LS	6% MCJ	- 15	300 and 20 s -		$57.92 \pm 1.78$			$56.91 \pm 2.11$		- VHN
			0.2% chitosan	_	-		$57.87 \pm 1.60$			$44.65\pm3.19$		_
			5.25% NaOCl							$51.74\pm6.03$		
			2.5% NaOCl	_						$50.86 \pm 5.08$		_
	Ari et al. 2004	90 Mand Ant	3% H <sub>2</sub> O <sub>2</sub>	_						$53.57 \pm 5.52$		_
6.	[20]	LS	17% EDTA	- 15	300 and 20 s					$53.66 \pm 3.87$		- VHN
		-	0.2% bCHX	_						$61.58 \pm 4.18$		
			distilled water (control group)	_						$61.86 \pm 11.70$		
	Elika et al.	40 SRT	Saline	15	- 200 and 20 s -		$55.98 \pm 3.94$			$55.07 \pm 4.15$		
7.			5% NaOCl + 17% EDTA				$54.03 \pm 5.88$			$48.00\pm5.32$		– VHN
7.	2021 [1]		Triphala	- 15			$47.40\pm5.53$			$43.60\pm5.95$		- 111
			Chloroquick	-	-		$43.46\pm4.43$			$38.80 \pm 4.90$		_
			distilled water							$45.27\pm7.25$		
8.	Asghari et al.	88 Mand PM	Triphala	- 15	200 and 15 s					$44.96\pm7.15$		– VHN
0.	2018 [16]	Transverse	2% CHX	- 15	200 and 15 S					$41.62\pm5.23$		- v111v
			5.25% NaOCl	_						$38.12\pm6.71$		_
			0.2% CHX				$51.59\pm8.98$			$53.15\pm8.20$		
9.	Prabhakar et al.	16 Mand PM LS	6% MCJ	- 15	300 and 10 s		$54.40\pm8.42$			$57.38 \pm 6.10$		– VHN
).	2013 [36]	10 Manu 1 M LS	6% MCJ + 0.2% CHX	15	500 and 10 5		$58.94 \pm 8.80$			$59.14 \pm 7.34$		- 111
			Saline				$52.70\pm8.15$			$55.68 \pm 6.86$		
	Earong at s1	00 CDT	Sapindus mukorossi							$60.07\pm0.49$		_
10.	Farooq et al. 2022 [37]	90 SRT LS	17% EDTA	15	300 and 10 s					$56.62\pm0.72$		VHN
		LS _	distilled water							$60.45\pm0.35$		

		Sample Size, Type of			Load (g) Given		Ν	licrohardness V	alue (Mean $\pm$ S	D)		
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	vical	Mie	ldle	Ар	ical	MHN Test
		in Each Group			Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_
			2.5% NaOCl							$36.90 \pm 2.46$		
	Patil and		3% H <sub>2</sub> O <sub>2</sub>	-						$57.20 \pm 4.65$		_
11.	Uppin 2012	120 Incisors LS	17% EDTA	15	300 and 20 s					$57.80 \pm 4.83$		VHN
	[21]		0.2% CHX	-						$65.05\pm4.29$		_
			Distilled water	-						$69.55\pm4.65$		_
		30 PM	Saline							$30.73 \pm 10.60$		
12.	Oliveira et al. 2007 [28]	LS	2% CHX	15	50 and 10 s					$20.89\pm10.24$		VHN
			1% NaOCl	-						$19.84 \pm 12.11$		_
			2.5% NaOCl solution			Cer	vical	Mie	ldle	Ар	ical	
13.	Garcia et al. 2013 [30]	24 Max Canines	2.5 % INdOCI Solution	15	25 and 10 s	0.58 =	= 11.32			0.58 ±	11.32	KHN
	2013 [30]	LS —	ChlorXTRA			0.67 =	22.57			0.67 ±	22.57	
			5.5% NaOCl gel	-	-	1.03 =	= 12.10			1.03 =	12.10	_
14.	Yaseen et al.	16 SRT	5.25% NaOCl + 13% GSE	15 + 15	- 300 and 20 s -			17.48	$\pm 2.53$			– VHN
14.	2020 [38]	LS	5.25% NaOCl + 17% EDTA	15 + 15	500 and 20 5			34.75	$\pm$ 1.61			- v111v
			2.5% NaOCl			0.11	± 0.02	0.10 :	± 0.01	0.13 :	± 0.02	
	Philip et al.	16 Max Canines	Miswak stick extract	-		0.28	$\pm 0.01$	0.27 :	± 0.01	0.14	± 0.02	_
15.	2021 [5]	LS	Cashew leaves extract	10	200 and 20 s	0.28	± 0.02	0.28 :	± 0.03	0.29 :	± 0.01	VHN
			Mango leaves extract	-		0.27	$\pm 0.01$	0.28 :	± 0.01	0.28 :	± 0.02	_
			Saline (control)	-		0.31	± 0.02	0.30 :	± 0.02	0.30 :	± 0.01	_
			2.5% NaOCl	5		10.0 =	= 21.15	8.92	± 1.08	8.36	± 1.16	
16.	Massoud et al.	40 Mand PM	17% EDTA + 2.5% NaOCl	10	25 110 -	32.98	$\pm 6.06$	30.37	$\pm$ 8.02	29.56	$\pm$ 8.01	– – VHN
10.	2017 [19]	LS	2.5% NaOCl + 2% CHX	10	25 and 10 s	19.15	± 3.09	17.68	$\pm$ 2.52	17.18	$\pm 2.35$	- V FIIN
			2.5% NaOCl + distilled water + 2% CHX	15	· · · · ·	15.16	± 1.25	13.82	$\pm 1.10$	13.23	$\pm 1.01$	_

		Sample Size, Type of			Load (g) Given		Ν	licrohardness V	alue (Mean $\pm$ S	D)		
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	rvical	Mi	ddle	Ар	ical	MHN Tes
		in Each Group		. ,	Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_
			2.5% NaOCl	10				52 :	± 2.0			
			6% MCJ + 17% EDTA	10 + 1				54 :	± 2.1			_
	Coole initiation	100 CDT	6% MCJ	10				53 :	± 2.2			_
17.	Saghiri et al. 2013 [39]	100 SRT LS	2.5% NaOCl + 17% EDTA	10 + 1	100 and 20 s			52 :	± 2.2			VHN
			1.3% NaOCl + MTAD	20 + 5				45 :	± 2.2			_
			2% CHX	5				4.1	± 1.1			_
			Saline (control group)	5				55.0	$\pm 1.1$			_
							Pre Rx			After Rx		
	Ibrahim et al.	54 SRT	2.5% NaOCl10				$83.56\pm2.97$			$59.15 \pm 1.76$		
18.	2021 [40]	LS	8% ethanolic extract of Olea europaea	10	300 and 20 s		$85.52 \pm 1.06$			$58.90 \pm 1.25$		VHN
			17% EDTA + 2.5% NaOCl	2 + 10			$50.32\pm2.3$			$47.76 \pm 4.05$		
19.	Kulkarni et al.	24 Ant	Saline	2	200 and 20 s					$54.39 \pm 3.59$		VHN
1).	2021 [31]	LS	2% NaF	2	200 and 20 3					$47.05\pm2.21$		_ ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
			2% CHG	2	-					$69.05\pm2.46$		_
			17% EDTA				$66.01 \pm 5.51$			$56.76 \pm 8.05$		
			REDTA	-	-		$59.76\pm3.28$			$50.44 \pm 4.23$		_
	Aslantas et al.	25 Mand 3rd Molars	6% NaOCl	-	-		$68.47 \pm 1.96$			$64.3\pm1.66$		_
20.	2014 [8]	LS	6% NaOCl with surface modifiers	- 5	300 and 20 s		$58.71 \pm 3.71$			$56.66 \pm 4.27$		- VHN
			2% CHX	-	-		$65.09\pm3.9$			$62.86 \pm 1.57$		_
			CHX-Plus	-	-		$60.26 \pm 1.91$			$60.04 \pm 4.80$		_
	De-Dues et al.		17% EDTA				$47.6\pm7.3$			$34.7\pm6.3$		
21.	2006 [14]	16 Max Canines Transverse	17% EDTAC	5	50 and 15 s		$49.9\pm9.0$			$36.6\pm3.8$		VHN
			10% Citric acid	-			$47.3\pm7.0$			$41.8\pm 6.2$		_
			17% EDTA	_			$55.5\pm8.4$			$23.88 \pm 4.59$		
22.	Kalluru et al.	40 Mand PM	17% EDTAC	- 5	- 50 and 15 s		$48.9\pm7.5$			$24.11\pm6.79$		– – VHN
	2014 [15]	Transverse	3% NaOCl	_ 0			$54.1\pm7.2$			$43.59\pm7.49$		
			MTAD	-	-		$51.3\pm7.0$			$45.78 \pm 6.39$		_

		Sample Size, Type of			Load (g) Given	Microhardness Value (Mean $\pm$ SD)						
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Ce	ervical	М	iddle	Api	cal	MHN Tes
		in Each Group			Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	Pre Rx         After Rx $01 \pm 2.56$	
			Saline (control group)				$56.95 \pm 3.40$			$53.91 \pm 2.56$		
	Duvvi et al.	75 Mand PM	2.5% NaOCl	-			$50.50\pm2.54$			$39.63 \pm 1.24$		
23.	2018 [41]	LS	5% NaOCl	5	300 and 20 s		$59.71 \pm 2.31$			$45.69\pm0.68$		VHN
			5% CaOCl <sub>2</sub>	_			$57.06 \pm 2.66$			$42.65 \pm 1.45$		
			10% CaOCl <sub>2</sub>	-			$56.96 \pm 1.84$			$39.03 \pm 2.17$		
			5% NaOCl + 17% EDTA + 2% CHX	5 + 5 + 5					64			
24.	Das et al. 2014	40 Incisors	6% MCJ + 17% EDTA	5 + 5	200 and 20 s			6	8.3			
	[27]	LS	5% NaOCl + QMix	5 + 5				6	9.9			
			Distilled water	5	-			5	'4.9			
			NaOCL-Extra					60 :	± 0.02			
	Dhawan et al.	120 PM	Pro-EDTA	-		55 ± 4.21						
25.	2019 [42]	120 PM LS	MTAD	5	200 and 20 s			59 :	± 0.01			VHN
			QMIx	-				63 :	± 0.01			
			CHX-Ultra	-				66 :	± 5.21			
			2.5% NaOCl	_				8.43	$\pm 2.58$			
			17% EDTA	-				21.59	$\pm 4.47$			
			17% EGTA	_				10.56	$\pm$ 3.34			_
			1% tetracycline hydrochloride					8.53	± 3.39			
26.	Sayin et al.	30 SRT	15% EDTAC	- 5	200 and 20 s			7.91	$\pm 1.34$			– VHN
20.	2007 [43]	LS	distilled water	- 5	200 and 20 5			3.42	$\pm 1.91$			— VIIIN
			17% EDTA + 2.5% NaOCl	-				27.54	± 5.05			_
			17% EGTA + 2.5% NaOCl					13.19	$\pm 5.08$			
			15% EDTAC + 2.5% NaOCl	-				11.81	$\pm 4.45$			
			1% tetracycline HCl + 2.5% NaOCl	-				11.06	± 3.76			_

		Sample Size, Type of			Load (g) Given		Ν	licrohardness V	alue (Mean $\pm$ SE	))		
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	vical	Mie	ddle	Aŗ	vical	MHN Tes
		in Each Group			Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_
						Cer	rvical	Mie	ddle	Ap	ical	
			Nano MgO	5		7.89	+ 0.74	8.88	+ 2.24	7.69	+ 2.28	
27.	Abdelrhman	16 Max Incisors	CHX loaded chitosan	5	- 200 and 20 s	13.74	+ 5.29	13.38	+ 2.39	13.28	+ 2.31	– – VHN
27.	et al. 2023 [44]	LS	5.2% NaOCl + 17% EDTA	3 + 2	200 and 20 5 -	19.47	' + 2.67	21.93 + 0.49		19.47 + 2.67		VIIIN
			Saline	5	-	0.56	+ 0.40	0.69 + 0.40		0.43 + 0.26		_
			2.25% NaOCl			70.92	± 0.83	66.84	± 1.22	76.86	$\pm 1.85$	
28.	Abdelgawad and Fayyad	40 Max Incisors	17% EDTA	Not mentioned	- 50 and 10 s	55.24	$\pm 0.45$	59.68	± 0.30	$65.24\pm0.577$		– – VHN
20.	2017 [45]	LS	Qmix	<ul> <li>clearly</li> </ul>	50 and 10 5	60.86	$\pm 0.15$	$63.02\pm0.49$		$69.72 \pm 1.188$		- 111
			0.2% Chitosan	_	-	63.80	$\pm 0.62$	65.00	$\pm 0.49$	73.88	± 0.79	_
			Saline			$63.73 \pm 2.85$		$73.10\pm12.74$		$60.57\pm3.16$		
	Khallaf et al.	100 DM	M. oleifera	<ul> <li>Not mentioned</li> </ul>	-	79.03	$\pm$ 9.92	71.30	± 3.02	$\begin{array}{c} 83.90 \pm 5.01 \\ \\ 95.60 \pm 7.61 \end{array}$		_
29.	2017 [25]	100 PM – LS –	M. oleifera and CHX	clearly	200 and 15 s	65.33	$\pm 5.10$	87.33	± 7.15			VHN
			CHX	_	-	89.23	± 6.22	82.87	± 12.97	$83.90\pm5.01$	_	
			NaOCl	_	-	72.30	$\pm 2.15$	76.77	± 3.24	61.37	$\pm$ 2.95	_
			distilled water					62.6	± 6.65			
		( <b>F</b> 0)))	EDTA	_	-			$54.92\pm 6.96$				_
30.	Alyahya et al. 2022 [46]	45 SRT LS	BioAKt	5	300 and 15 s			54.5 :	± 5.95			VHN
			40% citric acid	_	-			51.31 :	± 6.097			_
			10% citric acid	_	-			49.37	± 3.89			_
			5.25% NaOCl + 3% H <sub>2</sub> O <sub>2</sub>	5				5	50			
			5.25% NaOCl + SAEW	5 + 1	-			4	17			_
31.	Qing et al. 2006 [17]	43 SRT Transverse	5.25% NaOCl + distilled water	5 + 1	50 and 15 s			4	19			VHN
			5.25% NaOCl + SAEW	5 + 3	-			4	14			_
			5.25% NaOCl + 14.3% EDTA	5 + 1	-			44	4.5			_

		Sample Size, Type of			Load (g) Given		Ν	licrohardness V	alue (Mean $\pm$ SE	))		
No	Author	Teeth, and Section Used in Each Group	Irrigants	Contact Time (Minutes)	during Testing and Dwell Time	Cer	vical	Mi	ddle	Aj	pical	MHN Te
		in Each Group			Dwell lime -	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	
			distilled water					51.7	± 10.9			
			1% NaOCl	-	-			51.1	± 11.6			_
32.	Viapiana et al.	72 Canines	1% NaOCL + 17% EDTA	- 5	- 25 and 10 s			54.4	± 11.7			– – KHN
52.	2012 [32]	Transverse	without irradiation	- 5	25 and 10 5			45.0	± 9.7			
			Laser at 1.5 W/100 Hz	-	-			49.7	± 11.2			_
			Laser at 3 W/100 Hz	-	-			50.6	± 11.9			_
			5% NaOCl+ DW					77.39	± 2.16			
33.	Taneja et al.	10 PM	5% NaOCl + 17% EDTA	- 5+5	300 and 15 s			69.70	$\pm 4.14$			– vhn
55.	2014 [47]	LS	5% NaOCl + 2.25% PAA	- 3+5	500 and 15 s	$62.98 \pm 8.17$						
			5% NaOCl + Qmix	-	-			70.68	$\pm 4.97$			
			distilled water					39.33	$\pm$ 3.18			
			17% EDTA	— 3 300 and 20	_			39.28	$\pm 4.56$			
34.	Souza et al.	160 Incisors	Qmix		-			38.07	$\pm$ 4.01			– – VHN
54.	2021 [48]	LS	10% GA	- 5	500 and 20 5			35.62	$\pm$ 3.47			- 111
			17% GA	-	-			35.91	$\pm$ 3.24			_
			25% GA	-				35.98	$\pm$ 3.38			
			distilled water	3				0.00	± 2.77			
	Aranda-Garcia		17% EDTA	3	-			0.40 =	± 28.37			_
35.	et al. 2013 [49]	24 Max Canines LS	BioPure MTAD	5	25 and 10 s			1.94 =	± 25.72			KHN
			SmearClear	1				2.53 =	± 15.14			
			Qmix	2	-			1.10 =	± 41.13			
						Cer	vical	Mi	ddle	Aj	pical	
	Nikhil et al.	15 SRT	1% phytic acid			43.09	$\pm$ 7.40	43.59	$\pm$ 7.58	42.75	$\pm 6.87$	VHN
36.	2016	LS	17% EDTA	3	200 and 10 s	46.01	± 5.93	44.32	$\pm$ 4.12	44.2	± 3.69	_
	[50]		0.2% Chitosan		-	49.41	$\pm 5.56$	48.38	$\pm 5.16$	48.14	$\pm 4.63$	
			17% EDTA			55.64 50.17 41.15					1.15	
37.	Ballal el al. 2010 [51]	45 Max CI LS	7% maleic acid	1	200 and 20 s	52	2.85	48	3.75	52	2.85	VHN
			0.9% Saline	-		67	7.73	67	7.53	60	5.45	_

		Sample Size, Type of			Load (g) Given		Ν	Microhardness V	alue (Mean $\pm$ SE	))													
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	rvical	Mi	ddle	Aj	pical	MHN Tes											
		in Each Group			Dwell Time -	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_											
			5% EDTA			$7.30\pm8.35$																	
	41 10	25 Canines	5% EDTA + 0.25% cetrimide	-	-			8.78	$\pm 4.05$			_											
38.	Akcay and Sen 2012 [29]	LS	5% EDTA + 0.50% cetrimide	1	50 and 10 s			9.01	$\pm$ 4.14			VHN											
			0.25% cetrimide	-	-			4.59	$\pm 2.84$														
			0.50% cetrimide	-	-			7.77	± 3.83														
39.	Saleh and	18 Max Incisors LS	3% H <sub>2</sub> O <sub>2</sub> /5% NaOCl	. 1	100 and 15 s -			51.30	$\pm 0.02$														
57.	Ettman 1999 [18]	LS	17% EDTA	1	100 and 15 5			$47.30\pm0.02$															
			17% EDTA + 2.5% NaOCl					$55.80 \pm 3.65$															
	The action is here are	60 SRT Transverse	17% EGTA		-	$72.67 \pm 5.65$																	
40.	Unnikrishnan et al. 2019 [11]		MTAD	1	300 and 15 s	53.5 ± 2.78																	
			10% citric acid	_				48.30	$\pm$ 4.28			_											
			17% EDTA	-				72.00	$\pm 1.30$														
	Akbulut and	<b>70</b> (D)	2.5% NaOCl	_				662.76	$\pm$ 115.8														
41.	Terlemez 2019	72 SRT LS	17% EDTA	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s $541.41 \pm 150.96$	1 300 and 20 s $541.41 \pm 150.96$	1 300 and 20 s $541.41 \pm 150.96$	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s $541.41 \pm 150.96$	1 300 and 20 s $541.41 \pm 150.96$	300 and 20 s 541.41 ± 150.96	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s 541.41 ± 150.96	1 300 and 20 s $541.41 \pm 150.96$	1 300 and 20 s $541.41 \pm 150.96$	300 and 20 s		VHN			
	[24]		2% CHX	-				683.55	$\pm 152.13$			_											
			NI: 5% NaOCL + 17% EDTA + 5% NaOCl			1.68	$\pm 0.34$	1.8 ±	0.324	2.4 :	± 0.37												
	Arul et al. 2021	60 Max Incisors	PUI: 5% NaOCL + 17% EDTA + 5% NaOCl	-	100 and 10 s	$2.90 \pm 0.424$ $2.74 \pm 0.64$ $2.4 \pm 0.5$ $4.48 \pm 0.841$ $5.14 \pm 0.57$ $4.85 \pm 0.4$		± 0.50	– VHN														
42.	[26]	60 Max Incisors LS	EndoVac: 5% NaOCL + 17% EDTA + 5% NaOCl	-	100 and 10 5			5.14	± 0.57	4.85	± 0.43												
			Endovac + PUI: 5% NaOCL + 17% EDTA + 5% NaOCl	-	-	$5.06 \pm 0.680$ $5.15 \pm 0.54$ $4.82 \pm 0.54$				± 0.60	_												

		Sample Size, Type of			Load (g) Given		Ν	/licrohardness V	alue (Mean $\pm$ SE	))		
No	Author	Teeth, and Section Used	Irrigants	Contact Time (Minutes)	during Testing and	Cer	vical	Mie	ddle	Aj	pical	
		in Each Group			Dwell Time	Pre Rx	After Rx	Pre Rx	After Rx	Pre Rx	After Rx	_
			distilled water			$4.30\pm4.10$						
			17% EDTA + 5% NaOCl + DW	-				20.20	$\pm 3.36$			_
			17% EDTA + 60 s ultrasonic agitation + 5% NaOCl + DW	2	50 and 15 s 			23.60	$\pm$ 4.91			_
43.	Arslan et al. 2015 [23]	40 Max Ant LS	17% EDTA + 10 s agitation with laser + 5% NaOCl + DW			$18.62 \pm 7.66$						
	2010 [20]		17% EDTA + 20 s agitation with laser + 5% NaOCl + DW	-		$21.13 \pm 5.24$ $23.19 \pm 5.08$ $27.84 \pm 25$						_
			17% EDTA + 30 s agitation with laser + 5% NaOCl + DW									_
			17% EDTA + 40 s agitation with laser + 5% NaOCl + DW	-								_
			17% EDTA+ 5.25% NaOCl					53.11	$\pm$ 7.40			
44.	Eldeniz et al. 2005 [52]	45 Mand Ant LS	19% citric acid + 5.25% NaOCl	2.5 + 2.5	300 and 20 s	20 s 46.35 ± 5.77						VHN
			distilled water	-	-	69.73 ± 7.89						

SRT = single root tooth; Max = maxillary; Mand = mandibular; Ant = anterior; PM = premolar; CI = central incisor; LS = longitudinal section; NaOCl = sodium hypochlorite; EDTA = ethylenediaminetetraacetic acid; EGTA = ethylene glycol tetraacetic acid; EDTAC = EDTA + Cetavlon; DW = distilled water; MCJ = Morinda Citrifolia Juice; PAA = peracetic acid; CHX = chlorhexidine; CaOCl<sub>2</sub> = calcium hypochlorite; GA = glycolic acid; NaF = sodium fluoride; Chloroquick = 5% NaOCl + 18% etidronic acid; HEBP = (1-hydroxyethylidene-1,1-bisphosphonate); H<sub>2</sub>O<sub>2</sub> = hydrogen peroxide; Q-Mix = 2% CHX + 17% EDTA + detergent; NI = needle irrigation; PUI = passive ultrasonic irrigation; REDTA = cetrimide + EDTA; MTAD = 3% doxycycline, 4.25% citric acid, and detergent (Tween 80); NaOCl Extra = 6% NaOCl and surface modifiers; CHX-Extra = 2% CHX + surface modifiers; BioAkt = 4.8% citric acid, 0.003% silver electrolytes, detergents, water; Smear clear = 17% EDTA + cetrimide, surfactant; BioPure MTAD = 3% tetracycline isomer (doxycycline), 4.25% citric acid, 0.5% detergent; ChlorXTRA = sodium hypochlorite and surface modifiers (Triton X-detergent).

#### 3.4. Effect of Contact Time of Irrigating Solutions on Dentin Microhardness

The most significant reduction in microhardness was observed in the 2.5% NaOCl group with a 15 min contact time, with a Vickers Hardness Number value of  $36.90 \pm 2.46$ , compared to the control group that used distilled water, which had a microhardness value of  $69.55 \pm 4.65$  VHN [21]. The least reduction in microhardness was seen in the 0.2% CHX group with a 15 min contact time, with a value of  $61.58 \pm 4.18$  VHN, compared to the control group that used distilled water and had a microhardness value of  $61.86 \pm 11.70$  VHN [20].

# 3.5. Effect of Various Irrigating Solutions on Dentin Microhardness

# 3.5.1. Sodium Hypochlorite (NaOCl)

The reviewed studies used sodium hypochlorite concentrations ranging from 2.5% to 6%, with 2.5% NaOCl being the most tested [5,19–22,24,30,33,35,40,41,43]. At 15 min, 2.5% NaOCl significantly reduced dentin microhardness to  $36.90 \pm 2.46$  VHN versus the control group's  $69.55 \pm 4.65$  VHN [21]. A 5% NaOCl solution showed the greatest reduction over 5 min, lowering microhardness to  $45.69 \pm 0.68$  VHN from a pre-treatment level of  $59.71 \pm 2.31$  VHN [41]. Concentrations of 3% and 6% NaOCl also decreased dentin microhardness to  $43.59 \pm 7.49$  VHN and  $64.3 \pm 1.66$  VHN, respectively, after 5 min [8,15]. Conversely, 1% NaOCl achieved only a slight reduction after 5 min but a notable decrease to  $19.84 \pm 12.11$  VHN after 15 min, compared to the control saline's  $30.73 \pm 10.60$  VHN [28,32].

#### 3.5.2. Ethylenediaminetetraacetic Acid (EDTA)

Studies have assessed 17% EDTA as an irrigating solution, revealing it to be the second most examined. A notable decrease in dentin microhardness at  $57.80 \pm 4.83$  VHN was observed using 17% EDTA for 15 min when compared to the control's  $69.55 \pm 4.65$  VHN [21]. A reduction was also seen with a 5 min exposure, while a 3 min application did not result in a significant change, yielding  $39.28 \pm 4.56$  VHN versus the control's  $39.33 \pm 3.18$  VHN [48]. Lower concentrations of EDTA, specifically 5% and 15%, did not significantly alter microhardness after a 1 min contact time, with values recorded at  $65.18 \pm 5.52$  VHN and  $67.38 \pm 3.35$  VHN, respectively, against pre-treatment levels of  $65.33 \pm 6.88$  VHN and  $65.59 \pm 6.65$  VHN [33].

#### 3.5.3. Chlorhexidine (CHX)

The majority of studies focused on 2% chlorhexidine as an irrigant [8,16,28]. It reduced dentin microhardness the most to  $20.89 \pm 10.24$  VHN after a 15 min contact time compared with the control (saline) group's  $30.73 \pm 10.60$  VHN [28]. A minimal reduction to  $62.86 \pm 1.57$  VHN was noted after 5 min, versus the pre-treatment hardness of  $65.09 \pm 3.9$  VHN [8]. Conversely, 0.2% CHX did not yield a significant change in microhardness, even after 15 min, when compared to the control group [20,21,36].

#### 3.5.4. Herbal Irrigants

Among the evaluated herbal irrigants, extracts of miswak stick, cashew leaves, and mango leaves showed no significant reduction in dentin microhardness when compared to the control group's  $0.30 \pm 0.02$  VHN [5]. Similarly, other herbal solutions like Triphala and MCJ did not significantly affect dentin microhardness after a 15 min contact time [1,12,16,36,39]. Triphala's observed reduction was  $43.60 \pm 5.95$  VHN, not markedly different from the control's  $55.07 \pm 4.15$  VHN, and it had a lesser impact than 5% NaOCl and 17% EDTA [1,16]. Combining MCJ with chlorhexidine did not show a significant reduction from the pre-treatment hardness, but some studies noted a reduction when MCJ was paired with EDTA [27,36,39]. Herbal irrigants such as 8% ethanolic Olea europaea extract and 2% ethanolic Morus nigra extract did lower microhardness significantly compared to their pre-treatment levels [40]. However, Sapindus mukorossi had no impact when compared to the control [37]. Interestingly, M. oleifera alone and combined with CHX resulted in an increased dentin microhardness compared to the control group [25].

#### 3.5.5. Citric Acid

A few studies have examined 10% citric acid as an irrigant, observing a reduction in dentin microhardness [11,14,46]. The greatest decrease was to  $49.37 \pm 3.89$  VHN after 5 min, compared to the control group's  $62.6 \pm 6.65$  VHN [46]. It was found that there was no significant difference in the reduction of microhardness between 10% citric acid and 40% citric acid solutions [46]. In contrast, comparisons between 17% EDTA and 10% citric acid have yielded varied results. One study reported that 17% EDTA reduced microhardness significantly more to  $34.7 \pm 6.3$  VHN than 10% citric acid at  $41.8 \pm 6.2$  VHN after 5 min [14]. Another study found a minor reduction with 17% EDTA to  $72.0 \pm 1.3$  VHN compared to 10% citric acid's reduction to  $48.3 \pm 4.28$  VHN after just 1 min [11].

#### 3.5.6. Peracetic Acid (PAA)

Peracetic acid demonstrated a reduction in dentin microhardness to  $17.29 \pm 3.71$  KHN, which is comparable to the reduction observed with the NaOCI-EDTA-NaOCI sequence, at  $17.95 \pm 3.40$  KHN [35].

#### 3.5.7. Other Irrigants

Several different irrigating solutions have been studied for their effects on dentin microhardness, including 0.2% chitosan, glycolic acid, Qmix, hydrogen peroxide, MTAD, CaOCl<sub>2</sub>, Chlor XTRA, Smear Clear, and Chloroquick, among others. (Table 3) For instance, 15 min of contact with 0.2% chitosan resulted in a reduction in microhardness to  $44.65 \pm 3.19$  VHN from the initial 57.87  $\pm$  1.60 VHN [12]. Also, a comparison between 17% EDTA and 0.2% chitosan showed that EDTA had a significantly larger effect, decreasing microhardness to 59.68  $\pm$  0.30 VHN as opposed to 65.00  $\pm$  0.49 VHN for chitosan [45].

Hydrogen peroxide demonstrated a decrease in microhardness after a 15 min contact time, with one study highlighting a substantial reduction to  $57.20 \pm 4.65$  VHN compared to the distilled water control at  $69.55 \pm 4.65$  VHN [21]. Studies comparing hydrogen peroxide and EDTA revealed no significant difference in their ability to reduce microhardness, although in one study, EDTA showed a greater effect than a combination of 3% H<sub>2</sub>O<sub>2</sub>/5% NaOCl [18,20,21].

Other irrigants like Chlor XTRA and a 5.5% sodium hypochlorite gel caused reductions similar to a 2.5% sodium hypochlorite solution [30]. Meanwhile, MTAD also decreased microhardness, notably to  $45.78 \pm 6.39$  VHN after a 5 min contact time and was found to have a greater effect than the combination of NaOCl and EDTA [11,15,53]. However, 2% NaF did not present a significant difference compared with the control, while Smear Clear and QMix were similar in effectiveness to 17% EDTA [31,49]. Various concentrations of CaOCl<sub>2</sub> showed reductions in microhardness, with 10% CaOCl<sub>2</sub> marking the maximum decrease [41]. Glycolic acid, in its different concentrations, did not exhibit significant differences among the tested levels [48].

Moreover, the addition of surfactants to irrigating solutions was found not to alter root dentin microhardness [8,14,15,29]. EDTAC had a microhardness reduction value close to that of 17% EDTA, and the use of cetrimide with EDTA showed no significant difference from using EDTA alone [14,15,29].

Similarly, when using surface modifiers like Chlor-XTRA with NaOCl or REDTA (17% EDTA containing cetrimide), no significant differences were observed compared to the respective solutions without such additives [8]. Furthermore, nanoparticles such as CHX + CSNPs (chitosan-loaded nanoparticles) and MgO demonstrated a lower impact on reducing microhardness compared to a combination of 5% NaOCl with 17% EDTA [44].

#### 3.5.8. Activated Irrigating Solutions

Activation methods such as ultrasonic and laser agitation, including passive ultrasonic irrigation, were studied for their effects on dentin microhardness, yielding variable outcomes [23,26,32]. Irrigation solutions activated with ultrasonic agitation resulted in a decrease in microhardness to 23.6  $\pm$  4.91 VHN, which was not significantly different from the group without agitation at 20.2  $\pm$  3.36 VHN [23]. On the other hand, laser irradiation as an agitation method showed mixed results. One study indicated that laser activation led to less reduction in microhardness, at 50.6  $\pm$  11.9 VHN, compared to the group not subjected to laser irradiation, which had a microhardness of 45.0  $\pm$  9.7 VHN [32]. Yet, another study found that laser agitation used in conjunction with 17% EDTA and 5% NaOCl, followed by a rinse with distilled water, achieved the greatest reduction in microhardness at 18.62  $\pm$  7.66 VHN when compared to a combination without laser agitation, which resulted in a microhardness of 20.2  $\pm$  3.36 VHN [23].

#### 3.6. Effect of Combination of Irrigants on Dentin Microhardness

Nineteen studies [11,17,19,22,23,26,27,29,31,32,34–36,38,43,44,47,52,53] have examined the synergistic effects of various combinations of irrigating solutions, with sixteen [1,11,17,19,22,31–35,38,43,44,47,52,53] of these specifically comparing the effects of NaOCl and EDTA in different concentrations. The combination of 2.5% NaOCl with 17% EDTA was most frequently analyzed [11,19,22,31,35,43,53] followed by 5% NaOCl combined with 17% EDTA [44,47,52]. The findings indicate that the mix of 2.5% NaOCl with 17% EDTA, which showed a microhardness reduction value of  $30.7 \pm 3.5$  KHN, had a microhardness reduction of 2.5% NaOCl with 10% citric acid, which resulted in a microhardness reduction of  $31.5 \pm 4.9$  KHN after 30 min of contact time [22].

#### 3.7. Quality Assessment

The quality of the in vitro studies was assessed using the Quality Assessment Tool For In Vitro Studies (QUIN tool) checklist. The checklist includes 12 items which covers elements like the clarity and appropriateness of the study's objectives, the detailed characterization of the experimental model and conditions, the justification of sample sizes, the standardization of procedures, reproducibility of results, adequacy of statistical analysis, and transparency in reporting findings. The checklist aims to identify potential biases and determine the extent to which a study adheres to established scientific standards. Using such a tool in a systematic review ensures that conclusions are drawn from high-quality data, thereby contributing to the robustness of the evidence base in the field of study. The findings are summarized in Figure 2. The 44 studies assessed generally exhibited a consistent level of quality and a similar risk of bias. Most studies provided comprehensive rationales and clear objectives or hypotheses, and they typically detailed methodologies with defined study groups and outcome measures.

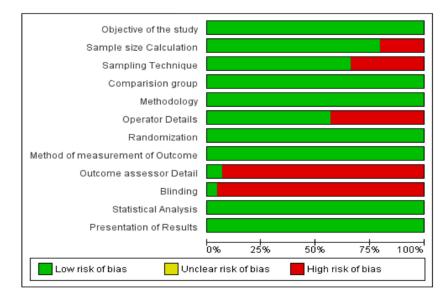


Figure 2. Quality assessment of included in vitro studies using the QUIN tool.

Method of measurement of outcome and the randomization process, presentation of results, and statistical analysis were uniformly reported. Only three studies discussed outcome assessor detail and two studies mentioned blinding. The percentage of checklist items met in the quality assessments ranged from 60% to 90% among the included articles.

#### 3.8. Risk of Bias in Included Studies

The risk of bias assessment using the QUIN tool with 12 items categorized twelve studies as low risk, and the remainder as medium risk (Figure 3).

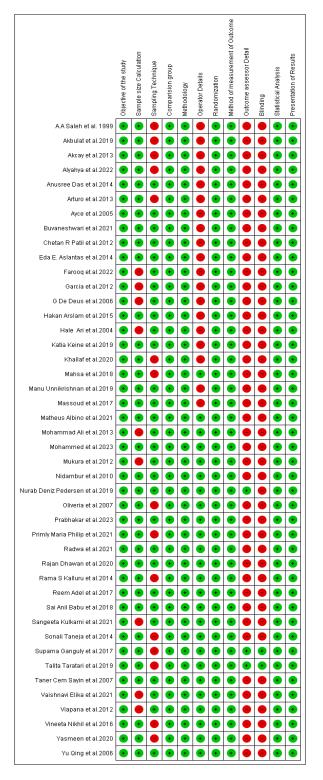
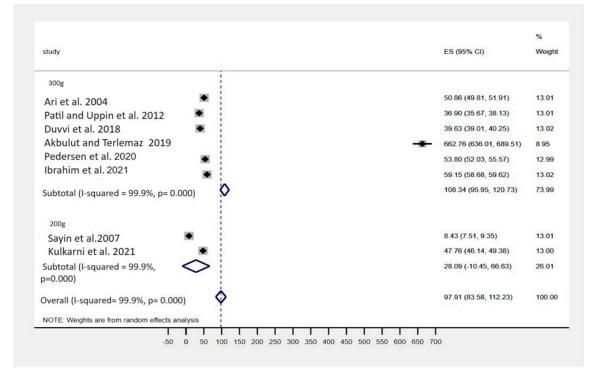


Figure 3. Risk of bias item for each included study using the QUIN tool.

#### 3.9. Meta-Analysis

The forest plot (Figure 4) shows the effect of 2.5% NaOCl on dentin microhardness under two different load conditions (300 g and 200 g). The studies indicate varying levels of reduction in dentin microhardness. Ari et al. [20] reported a mean reduction of approximately  $50.86 \pm 2.1$  under 300 g of load. Patil and Uppin [21] reported a mean reduction of approximately  $36.9 \pm 2.46$  under the same conditions. These results suggest that 2.5% NaOCl under a 300 g load has a significant impact on reducing dentin microhardness.



**Figure 4.** Forest plot showing the reduction in dentin microhardness using 2.5% NaOCl as an irrigating solution under 300 g [20,21,24,33,40,41] and 200 g of load [31,43].

#### 4. Discussion

In this systematic review and meta-analysis, we aimed to comprehensively assess the effect of various irrigating solutions on dentin microhardness, considering different contact times and concentrations. We synthesized data from 44 studies that met our eligibility criteria, examining a variety of irrigation solutions, including sodium hypochlorite (NaOCl), ethylenediaminetetraacetic acid (EDTA), chlorhexidine (CHX), herbal irrigants, citric acid, peracetic acid (PAA), and other novel irrigants. Our review also considered the activation methods used to enhance the effect of these solutions.

The inclusion of in vitro studies in this review helped in the detailed analysis of microhardness reduction by various irrigating solutions in a larger number of dentin samples than possible in human or animal trials.

This holds potential significance in the selection of endodontic irrigants as irrigants alter the chemical composition of dentin and can cause the formation and initiation of microcracks in dentin during endodontic procedures [53].

Success in the clinical performance of endodontically treated teeth is determined by the lesser impact of irrigating solutions on dentin microhardness as compared to the control group as a decrease in the microhardness of dentin weakens the tooth structure [34].

#### 4.1. Study Quality and Risk of Bias

The quality assessment of the studies included in this systematic review was performed using the QUIN tool, a checklist of items for reporting in vitro studies specifically for dental related studies. Twelve studies [12,17,22,33,36,40–45,51] were categorized as having a low risk, with the rest positioned within the medium risk category [1,8,11,15,16,18–21,23–29,31,33,35,38,43,46,48–50,52,53] as detailed in Figure 3.

This variability in the quality of the studies is an important factor to bear in mind as it influences the interpretation of the results. The assessment showed that while some studies adhered closely to the quality criteria set by the checklist, others deviated to varying degrees. This finding underscores the necessity of a critical approach to data analysis since the risk of bias can impact the overall conclusions drawn from this systematic review.

# 4.2. Effect of Sample Preparation and Testing Method, Load, and Dwell Time on Microhardness of Dentin

Most of the studies opted for longitudinal sectioning of the tooth, which divides the root into buccal and lingual segments, exposing the superficial dentin (Table 3). This mirrors clinical scenarios and ensures direct contact of the irrigating solution with the superficial layer in the root canal lumen. This choice significantly impacts microhardness testing by providing a more accurate representation of clinical conditions during treatment, enhancing the reliability of the results [54].

The hardness test measures the resistance of dentin to deformation caused by the penetration of an indenting stylus. The microhardness test is easy, quick, and requires only a tiny area of specimen surface for testing. The mineral content of dentin contributes to its hardness. Any irrigating solution which alters the Ca/P level of dentin alters the hardness value directly [55].

Nine studies [5,19,22,25,26,30,44,45,50] evaluated the microhardness value of the coronal, middle, and apical third separately. Four studies [19,44,45,50] concluded that there was a difference in the microhardness levels of the coronal, middle and apical thirds. Reductions in microhardness values were greater in the coronal third than the apical third. The possible reason for this could be that the microhardness of dentin depends on the tubular density which varies from one area to another on the root dentin surface. The tubular density affects microhardness, as the tubular density at the coronal section increases dentin microhardness decreases. The other studies which did not show differences in microhardness levels in the coronal, middle, and apical sections may have used a contact time of more than 10 min. This could have resulted in the overall deterioration of the internal structure of dentin to a significant extent [55].

The Knoop and Vickers testing methods differ in the shape of the indenter. The Vickers indenter penetrates approximately twice as far into the specimen as the shallower Knoop indenter [56] and is a widely accepted method as only one type of indentation is used for all types of surface treatment [12]. The Vickers Hardness Number is based on the mean of two diagonals, providing more reliable results, whereas the Knoop test relies only on one diagonal [12]. Therefore, most studies have used the Vickers Microhardness Test, except a few studies [22,30–32] which used the Knoop indenter.

The load applied during the microhardness testing of root canal dentin also plays a crucial role in the accuracy of results. Studies typically used loads ranging from 25 g to 300 g (Table 3). Due to dentin's elastic or viscoelastic nature, microhardness values at very low loads might be affected. Higher loads create larger impressions, aiding in indentation size measurement. This variation in microhardness with load is termed the Indentation Size Effect (ISE), which can be either normal, where microhardness decreases with increasing load, or reverse, where it increases. Comparing microhardness values obtained at different loads is not straightforward due to the various factors contributing to ISE, such as measurement accuracy, indenter geometry, and uncertainties in indentation area estimation, along with dentin's physical properties like elastic recovery or elastic–plastic deformation after indenter removal [57].

Another inconsistency in the methodologies of the included studies is that the load applied was often more than the root dentin can take. It has been reported that healthy caries-free coronal dentin microhardness ranges from 52 to 64 KHN or 46 to 53 VHN. The

root dentin has less mineral density compared to coronal dentin. Therefore, a higher load of more than 100 g may be impractical for a softer surface in the pre–post experiment because, after treatment, it produces a larger impression than the optical microscope can measure. The lowest loads, as small as 10 g for dentin, can create Vickers diagonals longer than 20  $\mu$ m [57].

Additionally, the variation of loading times (10, 15, and 20 s) might have contributed to heterogeneity in the microhardness values. A study performed to investigate the effect of indentation load and time on the Knoop and Vickers microhardness tests for enamel and dentin concluded that an indentation time of 10 s is sufficient for a permanent indentation on the tooth surface to take place.

It is evident from the results that there is no standard condition for dentin microhardness testing across the included studies. The heterogeneity in the selection of testing conditions depended on the researchers' decisions. The broad variation of hardness values can be produced by factors such as specimen preparation, diagonal length reading error, variation in chemical composition, age, and location in the tooth.

#### 4.3. Effect of Individual Irrigating Solutions on Microhardness of Dentin

In our systematic review, we meticulously examined the impact of various irrigating solutions on dentin microhardness, a critical aspect influencing the success of root canal treatments. Our comprehensive analysis revealed nuanced effects of each solution, shedding light on their potential implications in clinical practice.

Sodium hypochlorite (NaOCl) emerged as a potent agent for dentin microhardness reduction, particularly at a concentration of 2.5% [5,19–21,30,35,41,53]. The dissolution of intertubular dentin following NaOCl treatment led to tubule enlargement and increased vulnerability to structural compromise [27]. Moreover, our findings underscored the dose-dependent nature of NaOCl's effect, with higher concentrations and prolonged exposure exacerbating dentinal erosion and microhardness reduction [8,31].

Conversely, ethylenediaminetetraacetic Acid (EDTA) demonstrated significant dentinsoftening capabilities attributed to its chelating action on calcium ions [43]. However, the extent of softening varied with EDTA concentration, necessitating cautious consideration in treatment planning [12,20,22]. Notably, concerns regarding EDTA's potential to stimulate matrix metalloproteinase release raised questions regarding its long-term impact on dentin integrity [58].

Chlorhexidine (CHX) showcased dual-action properties, exhibiting both antimicrobial efficacy and dentin-softening capabilities [16]. While 2% CHX solutions altered dentin microhardness by disrupting the calcium–phosphate balance, lower concentrations released gradually over time facilitated canal shaping and sealing without compromising structural integrity [8,28].

Herbal irrigants, including Triphala and MCJ, offered intriguing alternatives to conventional solutions, albeit with milder dentin-softening effects [1,12,36]. Triphala's bacteriostatic properties, attributed to its citric acid content, and MCJ's organic acids demonstrated potential for application in specific clinical scenarios [1,16]. However, further research is warranted to validate their efficacy and safety profiles.

Citric acid, known for its chelating and smear layer removal properties, exhibited notable effects on dentin microhardness [11,14]. Its softening capabilities, dependent on pH rather than concentration, presented intriguing comparisons with EDTA, highlighting the need for nuanced evaluations in clinical settings [46].

Additionally, our review identified diverse effects of other irrigating solutions, such as MTAD [11,15,42,53], chitosan [12,45,50], CaOCl<sub>2</sub> [28,41], and QMix [42,45,48], on dentin microhardness. While some solutions showed promising results, further investigation is essential to elucidate their mechanisms of action and clinical implications comprehensively.

#### 4.4. Effect of Activation Methods of Irrigants

Studies [23,32] have looked into different activation methods for irrigating solutions, like laser irradiation and ultrasonic agitation. Ultrasonic agitation was found not to change dentin microhardness [23]. Lasers, however, with wavelengths between 810–980 nm, showed varying effects, largely depending on the irrigation solution used [23].

Some research has shown that laser agitation, especially when used with EDTA, can demineralize dentin, leading to a softer dentin structure. The laser works by vaporizing the dentin's organic matrix, creating pores and voids, which ultimately reduces its microhardness [59].

#### 4.5. Effect of Combinations of Irrigants

In endodontic treatments, irrigants are often used sequentially to enhance root canal cleaning [1]. The combination of sodium hypochlorite and EDTA is a common regimen [17]. This duo has been widely studied and is favored due to its synergistic effect on dentin microhardness [11,19]. The use of NaOCl followed by EDTA creates an alkaline environment which increases EDTA's efficiency in chelating calcium ions, thus leading to greater dentin demineralization [60].

The mechanism involves EDTA's chelation process, which targets the inorganic component, while NaOCl disrupts the organic matrix of dentin. Together, they reduce microhardness by softening the calcified tissues [60]. Moreover, the combination of NaOCl and EDTA was found to be comparable to the use of NaOCl with citric acid, since both EDTA and citric acid serve as chelating agents that demineralize dentin and facilitate the removal of calcium ions, altering the tooth's structural properties [22].

#### 4.6. Limitations of the Study

The ilimitations of the study encompass inconsistencies within the included studies and the necessity for additional studies to ascertain the practical significance of the observed effects.

Variability in factors such as the range of loads used for microhardness testing, differences in dentin properties across specimens, mode of delivery of irrigating solution, and variations in measurement techniques may introduce inconsistencies in the results.

Another possible limitation is the immersion treatment as the volume of the irrigant in a root canal clinically is small compared with the immersing root dentin in irrigating solutions. The experiments were also performed at room temperature and not body temperature.

This diversity in methodologies and experimental conditions of the included studies should be carefully considered when interpreting the collective findings of this review.

#### 4.7. Recommendations for Future Studies

For future studies aiming to assess the microhardness of root canal dentin, the following recommendations are proposed:

- 1. Standardize Load Range: the load while preforming microhardness tests should gradually increase from 10–50 g;
- 2. Control Indentation Time: Standardize the duration of indentation to 10 s to prevent variations in results due to differences in the duration of load application. Consistency in indentation time helps ensure reproducibility of results;
- Account for Dentin Properties: Take into account the inherent variability in dentin properties, such as its elastic or viscoelastic nature, which can influence microhardness measurements. Consider controlling for factors like dentin age, source (human or animal), and storage conditions to minimize variability;
- Use Consistent Measurement Techniques: employ standardized measurement techniques for assessing microhardness, such as Vickers or Knoop hardness testing, to ensure uniformity across studies;

- 5. Address Indentation Size Effect (ISE): Recognize the potential impact of ISE on microhardness measurements and consider its implications in the interpretation of results. Investigate the presence of normal or reverse ISE and its effect on dentin microhardness under different experimental conditions;
- 6. Report Methodological Details: Provide detailed descriptions of the experimental procedures, including the type of indenter used, the range of loads applied, indentation time, and any adjustments made to account for dentin properties or ISE. Transparent reporting facilitates reproducibility and enhances the reliability of study findings;
- 7. Consider Microstructural Analysis: complement microhardness measurements with microstructural analysis, such as scanning electron microscopy (SEM) or atomic force microscopy (AFM), to gain insights into the structural changes accompanying variations in microhardness.

#### 4.8. Clinical Implications

Irrigant solutions do alter the microhardness of root dentin which impacts the outcome of endodontic treatments. Despite their benefits like debris removal, disinfection, and smear layer removal, these solutions can also compromise dentin's physical properties, including microhardness. Reduced microhardness aids instrumentation but can weaken the root structure. Microhardness assessment offers insight into mineral substance changes in dental hard tissues.

## 5. Conclusions

The impact of various irrigants on dentin microhardness is complex, determined by factors such as their concentration, duration of contact, and inherent chemical characteristics. The broad variation of hardness values in the included studies is due to factors such as specimen preparation, diagonal length reading error, variation in chemical composition, age, and location in the tooth.

From the present systematic review, one can conclude that NaOCl and EDTA concentration and contact time with both the organic and inorganic components of dentin plays a significant role in the reduction of microhardness. Chlorhexidine also alters the calcium to phosphate ratio and influences dentin's structural integrity.

Interestingly, natural alternatives like Triphala present a gentler option with fewer adverse effects. The properties of other irrigants, such as glycolic acid, phytic acid, and chitosan, reflect their respective chemical compositions. Moreover, the choice of activation method can modify the outcomes of these irrigants, either enhancing or mitigating their effects on the microhardness of dentin.

However, more research is required to understand the complex interaction of irrigating solutions on the physical and mechanical properties of dentin using standardized methodologies.

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