



An Overview of the Valorization of Aquatic Plants in Effluent Depuration through Phytoremediation Processes

Nabila Khellaf¹, Hayet Djelal^{2,*} and Abdeltif Amrane³

- ¹ Department of Process Engineering, Faculty of Engineering, Badji Mokhtar University, P.O. Box 12, Annaba 23000, Algeria; khellafdaas@yahoo.fr
- ² UniLaSalle-Ecole des Métiers de l'Environnement, Cyclann, Campus de Ker Lann, 35170 Rennes, France
- ³ Ecole Nationale Supérieure de Chimie de Rennes, Université de Rennes, CNRS, ISCR, UMR6226,
 - 35000 Rennes, France; abdeltif.amrane@univ-rennes1.fr Correspondence: hayet.djelal@unilasalle.fr

Abstract: Environmental biotechnologies are a popular choice for using efficient, low-cost, low-waste, and environmentally friendly methods to clean up and restore polluted sites. In these technologies, plants (terrestrial and aquatic) and their associated micro-organisms are used to eliminate pollutants that threaten the health of humans and animals. They have emerged as alternative methods to conventional techniques that have become increasingly aggressive to the environment. Currently, all actors of the environment, whether governors, industrialists, or citizen associations are more interested in the application and development of these technologies. The present overview provides available information about recent developments in phytoremediation processes using specifically aquatic plants. The main goal is to highlight the key role of this technology in combating the drastic organic and inorganic pollution that threatens our planet daily. Furthermore, this study presents the valorization of aquatic plant after phytoremediation process in energy. In particular, this article tries to identify gaps that are necessary to propose future developments and prospects that could guarantee sustainable development aspired by all generations.

Keywords: aquatic plant; heavy metal; organic pollutant; phytoremediation; sustainable development

1. Introduction

During the last decades, biotechnological techniques have gained public recognition as a multidisciplinary research field. In addition to their use in the production of valuable products (pharmaceutical, food, and cosmetic products), they have been used in the protection of the environment with the development of depuration processes with micro-organisms, such as biofilters, activated sludge tanks, biological oxidation, etc. These traditional techniques, called environmental biotechnologies, have been very efficient in pollution treatment, due to the production of less waste and use of less energy than the physicochemical treatments. Nowadays, the knowledge of biological techniques explaining previously unknown phenomena has greatly increased, which has led to other biological purification processes based on the use of plants and their associated micro-organisms. Thus, phytotechnology, which is commonly referred to as phytoremediation, is an increasingly popular choice for using efficient, low-cost, low-waste, and environmentally friendly methods to clean up and restore polluted sites. Phytoremediation is a relatively recent technology that uses the natural ability of plants to extract, reduce, degrade, or immobilize toxic and/or undesirable organic and inorganic compounds present in contaminated media. It is used to reduce metals, pesticides, solvents, explosives, oil, radionuclides, nanomaterials, and other unwanted contaminants initially present in air, water, soils, and sediments [1-3]. The advantage of this green technology is to treat large areas or large volumes at a low cost without generating major disturbances of the environment. In addition, this process ensures a pleasant cover, which reduces visual pollution. The aesthetically pleasant aspect of



Citation: Khellaf, N.; Djelal, H.; Amrane, A. An Overview of the Valorization of Aquatic Plants in Effluent Depuration through Phytoremediation Processes. *Appl. Microbiol.* **2022**, *2*, 309–318. https:// doi.org/10.3390/ applmicrobiol2020023

Academic Editor: Katarzyna Turnau

Received: 21 March 2022 Accepted: 23 April 2022 Published: 26 April 2022

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). phytoremediation renders it more economical and ecological than conventional treatment methods [4–6].

During the 1990s, it was noted that soils contaminated with toxic metals and surface water charged with organic substances and nutrients were accompanied by an excessive multiplication of terrestrial or aquatic plants. Without intervention, urban and industrial wastelands were quickly colonized by vegetation [7]. Phytoremediation emerged as a scientific field from the study of vegetation at polluted sites and was the purpose of several studies. This was how the first works and studies on phytoremediation processes emerged in research and scientific circles.

The presence of hazardous pollutants in the environment and the severity of their effects on human health and the environment have forced governments to adopt standard values to protect human, fauna, and flora from potential exposure to these contaminants. This is highlighted by the large number of limit values existing for the atmosphere, water, aquatic environments, sewage sludge, sediments, and soils. These limit values are usually set by international environmental agencies (Environmental Protection Agency, World Health Organization, Climate Action Network, etc.).

2. Mechanisms of Phytoremediation and its Application in the Field

Phytoremediation technologies can be applied to organic and inorganic pollutants in soils, water, and atmosphere [8]. Globally, the purification processes and their selection are based on the type of pollutant, biological factors, and nature and environment of plant. These interrelated groups of factors (Table 1) play a role in the interaction between contaminants and biological barriers. Pollutant phytoremediation depends upon numerous biotic and abiotic factors.

Factors	Characteristic Properties		
Type of pollutant	Organic or inorganic, nature (gas, liquid, solid), concentration, chemical form, bioavailability.		
Biological factors	Exchange surface, accessibility of fixation sites, transport process, absorption capacity, growth, reproduction, nutrition, excretion.		
Nature of plant environment	Temperature, pH, light, concentration of suspended matter, salinity, ionic strength, organic and inorganic ligands.		

Table 1. Factors involved in the interactions between contaminants and biological barriers.

Temperature and pH are the most important environmental characteristics affecting chemical uptake and distribution within living plants [2,9]. Particularly, temperature governs the physiological processes of plants maintaining a relationship with growth rate and other functions; alternatively, pH affects the ionization state of some functional groups of the cell wall [10]. Several researchers reported that an acidic media had direct detrimental effects on the physiology of plants following a decrease in the electrochemical gradient across the plasma membrane [11,12].

During the physiological process of exchange with the surrounding environment, exogenous molecules penetrate through the biological barriers, namely from the external environment into the internal environment of the organism. When contamination occurs, these barriers show biological properties linked to their structure and to the physicochemical conditions of the environment. The biological factors of the organism, which are involved in the different interactions are: exchange surface and accessibility of binding sites, transport process and absorption capacity; in addition, the different physiological processes, such as growth, reproduction, nutrition, excretion, etc. can intervene on a larger scale [13].

Phytoremediation can be increased through using chelating agents, namely EDTA and specific acids; this specifically concerns metal pollutants that need to be absorbed by plants as complex compounds. In parallel, the addition of plant growth regulators (PGRs) can

improve plant phytoremediation efficiency by increasing plant biomass and decreasing the negative effects of the presence of contaminants in the plant [14].

All these factors (Table 1) are important to the design and implement effectively a sustainable phytoremediation technology.

2.1. Mechanisms of Phytoremediation

The interaction between chemicals and plants led to distinguishing different mechanisms, as summarized in Figure 1. Although not exhaustive, the figure gives some pollutants (in italic) that have been studied at the laboratory and field scale. The majority of scientific and popular articles researching this topic has summarized well the different techniques and explained the mechanisms involved. They have also identified key factors (nature of pollutant, concentration, and toxicity level) that are needed in order to design and effectively implement a sustainable phytoremediation technology [15].



Figure 1. Different techniques of phytoremediation.

It is important to understand how pollutants are accumulated, how they are being translocated to the plant cells, and how they are being detoxified [16]. In water, divalent metals are present in hydrated form, complexed by organic (fulvic and humic acids) or inorganic ligands, or are adsorbed on particles. The physicochemical characteristics of water (pH, light, temperature, ionic strength, content of organic, and inorganic ligands) affect the degree of dissociation between the complexed and ionic forms. In ionic form, metals can attach very quickly to the membrane surface of plants (their introduction into the cell is facilitated by diffusion). Depending on its nature, the metal element is integrated into the membrane or fixed on the sites reserved for nutrients (phosphates, for example). However, the exact mechanisms of absorption are still poorly understood, especially for some elements, such as metals considered as very toxic for plants (Cd, Pb, As, etc.). Nowadays, some metals are considered to be passively absorbed by certain proteins, while others are actively taken up by a selective, energy-requiring transport protein [17].

2.2. Field Application

Phytoremediation has been applied on a large scale in USA, Canada, and some European countries. In some African countries (Kenya, Zambia, South Africa, Nigeria, and Tanzania), some phytoremediation projects have been considered to remediate polluted sites with heavy metals [18]. Some field applications of extracting heavy metals using various terrestrial plants have been implemented in China [19], Portugal, and Russia [20].

Generally, the content of extractable metals decreased in soil after the field experiments. Specifically, Table 2 gives some in-situ decontamination operations of polluted sites.

 Table 2. Some in-situ phytoremediation of polluted sites.

Site Name	Treatment Process
Charmabul Ukraina	Decontamination of soil contaminated by
Chemobyl, Okraine	radioelements (uranium)
INCO Sudbury, Canada	Treatment of mine waste (nickel, copper)
-	Depollution of metal industry waste storage site
Jales, Portugal	Treatment of sites contaminated with non-ferrous metals
Bordeaux, France	Treatment of sites contaminated by sludge inputs
San Joaquin Valley, California	Decontamination of soil and water contaminated with selenium

3. Selection of Plant Species

In their review study, Akhtar et al. [21] asserted that more than 400 plant species have been tested and proved to have the ability to remediate soil and water pollution. For a good phytoremediation process, an appropriate selection of plant species is an important tool to ensure optimal performance of remediation. The selection is based on the type of pollutant element, the geographical location, the environmental conditions, and the knowledge of the element's ability to accumulate in plants. Generally, terrestrial plants possess greater biomass and a faster-growing root system than aquatic plants. Sunflowers and Indian mustard are considered as the first terrestrial plants tested at laboratory and field-scale [22]. Several research works reported the treatment of pollutants removal from soil using consortia of soil microbial flora and terrestrial plants. An in-situ phytoremediation of soil accumulated dyes from textile wastewaters was conducted successfully by Chandanshive et al. [23] using terrestrial plants on ridge beds of a constructed wetland. Their results revealed that the use of plants could be a wise strategy for soil clean-up affected by textile industry effluents. Additionally, during the process, plants absorbed contaminants from the soil and thus prevented horizontal contamination of groundwater [18].

Buscaroli [24] established in its review article a list of factors and indexes to evaluate the performance of terrestrial plants in phytoremediation processes. This interesting terminology avoided confusion in the interpretation of experimental data.

4. Aquatic Plants

Aquatic plants are plant species visible to the naked eye growing in wetlands, such as lakes, ponds, streams, etc. They are also found in wastewater streams considered as eutrophic media. Various species have been inventoried around the world in various aquatic environments. Their abundance and their distribution depend strongly on the hydraulic regime, nutrients and water quality [25].

Numerous studies have been published on the use of various aquatic plants for phytoremediation of both organic and inorganic contaminants. Macrophytes, with fixed or free culture (floating, submerged, and peri-aquatic) are excellent candidates for cleaning up contaminated sites with toxic or undesirable substances. Several species have been tested in laboratory and some have been used on a large scale.

4.1. Phytoremediation of Heavy Metals Using Aquatic Plants

Biogeochemical cycles of metals considered as normal constituents of the biosphere have been modified by anthropogenic activities, resulting in their presence in the different compartments of the environment. There are various sources of metals: natural erosion and corrosion processes, mining activity, metallurgical industry, electroplating, surface treatment, power generation, and the leaching of contaminated areas. These anthropogenic activities can be considered as the main cause of contamination of soils, streams, rivers, and even groundwaters. This contamination is a real concern since metals do not decompose, unlike most organic compounds, and as such they can accumulate in most systems by adsorption or complexation. They can also be bioaccumulated by both fauna and flora causing an increase in the concentration of metal in the living organisms. If the excretion phase is slow, it may result in a biological accumulation phenomenon. It has been shown that most metals undergo biological magnification as they progress through the food chain [26].

Heavy metals when entering aquatic and terrestrial ecosystems become potentially hazardous and tend to accumulate in sediments and to concentrate in aquatic food chains. From the point of view of impact assessment and risk assessment, it is necessary to study the inputs and loads, the distribution and the fate of contaminants into the receptacle systems. In particular, there is a need to study quantitative and qualitative characteristics, as well as the routes they take when they disperse into the environment, and their effects on the environment. However, to rationally manage and control this pollution, it is necessary to think of depuration processes of effluent before their release into the environment in order to limit as much of their polluting load as possible. Phytoremediation of water and soil polluted by heavy metals can be a technology of choice to clean up contaminated sites. It can be effective by an appropriate selection of plant species used for this purpose.

Inorganic pollutants that can be treated by aquatic plants are numerous. Table 3 gives some elements and compounds that have already been tested either in the laboratory or on a large scale during the last ten years.

Aquatic Plant	Species	Pollutant	Application	Reference
Water hyacinth	E. crassipes	Cr, Zn	Laboratory	[27]
		Fe, Mn, Zn, Cu, Pb, Cr, Cd	Field	[28]
Water lettuce	P. stratiotes	Cu, Fe, Hg	Laboratory	[29]
		Cu, Zn	Laboratory	[30]
		Fe, Mn, Zn, Cu, Pb, Cr, Cd	Field	[28]
Duckweed	L. minor	Co, Cd, Zn, Cr, Ni, Cu, Fe	Field	[31]
		Mn, N, P		[32]
		U, Th	Laboratory	[33]
		В	Laboratory	[34]
	L. gibba	Zn	Laboratory	[9]
		В	Laboratory	[34]
		Cd, Cu, Zn	Laboratory	[35]
	S. intermedia			
	S. polyrrhiza			
Water fern	W. velvet			
	S. natants			
	S. auriculata	Pb		[36]
	S. molestela	Fe, Mn, Zn, Cu, Pb, Cr, Cd	Field	[28]
	A. filiculoides	Pb, Hg		[3]
		Co, Cd, Zn, Cr, Ni, Cu, Fe	Field	[31]
		Mn		
Submerged macrophyte	Vallisneria natans	As		[37]

Table 3. Inorganic pollutants treated with aquatic plants as phytoremediators (2009–2020).

4.2. Phytoremediation of Organic Pollutants using Aquatic Plants

Our environment suffers enormous physical and chemical challenges caused by human activities and by the ecological imbalance observed in recent decades. The most important anthropogenic activities that lead to the contamination of soils and waters by organic pollutants concern the supply of sewage sludge and mineral fertilizers, various phytosanitary products, urban composts, and agri-industrial wastewaters and atmospheric releases near industrial sites. This danger has led governments to think about solutions by setting directive laws and stringent regulating measures to treat industrial effluents and encouraging scientists to look for the best methods for preserving the environment.

There are fewer organic pollutants treated by aquatic plants than inorganic pollutants, but large-scale application has been greater during the last ten years (Table 4).

Aquatic plant	Species	Pollutant	Application	Reference
Water lettuce	P. stratiotes	Chlorpyrifos	Laboratory	[38]
		Perchlorate	Laboratory	[39]
Duckweed	L. minor	Terbuthylazine	Laboratory	[40]
		Chlorpyrifos	Laboratory	[38]
		Chloroacetamide	Laboratory	[41]
		Wastewater	Laboratory	[42]
Water fern				
	Azolla filiculoides	Dyes	Laboratory	[43]
	Salvinia molesta	Dyes	Field scale	[44]
Aquatic ipomea	Ipomoea aquatica	Dyes	Field scale	[45]
Reed	Typha angustifolia Juncus fontanesii	Dyes	Field scale	[46]
Hydrilla	Hydrilla verticillata	Phenantrene, Pyrene	Laboratory	[47]
Potamot	Potamogeton crispus	Phenantrene, Pyrene	Laboratory	[48]
Aquatic plant	Scirpus grossus	Real sago mill effluent	Laboratory	[49]

Table 4. Organic pollutants treated with aquatic plants as phytoremediators (2010–2020).

5. Prospects and Future Developments

Phytoremediation provides effective methods for removing pollutants from water, soil, and air to provide a healthy and pleasant environment for humans. It had emerged around the 1980s in the United States and Canada and was developed in 1990s. The first patent (Phytotech Inc.) was filed in 1994 and was followed by research conducted at the laboratory-scale and tested at pilot and field-scale systems for remediation of uranium-contaminated water [22,50]. The large-scale application of this technology in the 2000s has allowed phytoremediation to move from the conceptual phase to the commercial phase. However, in Europe this technology is still slightly exploited despite certain projects supported by the European Commission (EU); these projects were mainly oriented towards basic research [51].

Due to its advantages (low cost, ease of operation, eco-friendly aspect), phytoremediation is supported by scientists, industrialists, ecological organizations, and citizens' communities. Other advantages of this technology include extraction of valuable metals (phytomining) and increased soil fertility/quality [18].

Firstly, future studies should focus on the potentials of aquatic plants for the removal of organic pollutants present in effluents, such as industrial wastewaters, acid mines, pulp and paper, and dairy effluents. According to the recent review of [8], only 11% of the pollutants treated by some aquatic species concern organic pollutants. *Lemna minor* is able to reduce chemical oxygen demand (COD) by 92% from wastewater blended from textile, distillery, and domestic sources [30].

Despite these strengths, phytoremediation possesses limitations and research studies must be conducted to expand their use. For this purpose, the exploitation of tolerance, uptake, and translocation of pollutants in plants should be considered. Additionally, an appropriate plant species can be considered as a phytoremediator if it displays some specific characteristics, namely (a) a high rate of accumulation of the pollutant, (b) an ability to accumulate large concentrations of contaminants, (c) an ability to accumulate various elements, (d) a rapid growth and high biomass production, and (e) a high resistance to diseases and harmful insects.

The selected plant species must display all these characteristics. Furthermore, the interaction plant–microbe for pollution bioremediation must be studied at all levels (laboratory, pilot, and large scale) to understand the symbiotic relationships of plants with rhizospheric organisms. In this context, the ramping up of the modern biotechnology in which manipulation of genes from different organisms led to modify genetically some plants to increase their capacity to tolerate and hyperaccumulate several pollutants can be considered as an opportunity. Indeed, the involvement of genetically modified micro-organisms can further increase the phytoremediation capacity leading to total purification of impacted ecosystems. However, genetic engineering has not yet gained public recognition as an ethical science.

Since traditional phytoremediation process poses some limitations regarding their applications at large scale, it is important to improve the efficiency removal by coupling with chemical or/and biological processes [6,52].

Another advantage of phytoremediation technology comes from the biomass value added, which presents economic opportunities in the form of biofuels and bioenergy [53]. Indeed, the production of energy and element recovery from biomass significantly increases the financial viability of this process and reduces the environmental impacts of disposal for contaminated biomass [54]. Aquatic weeds contain high amount of cellulose and hemicellulose and low lignin content, which give them a potential for biofuel and biomethane production [55]. It is economically possible to produce bioethanol using water hyacinth [56]. The anaerobic fermentation of the contaminated biomass is possible with biogas production containing 60% of methane [57]. Oil crops used for the phytoremediation of contaminated soils with heavy metals can be used for biodiesel production by the supercritical methanol method [58]. Mthethwa et al. investigated the valorization of an aquatic weed, Pistia stratiotes for biohydrogen production via a dark fermentation process [59]. The novel process, hydrothermal liquefaction (THL), is well adapted to the valorization of aquatic plant. Indeed, a study based on recycling of waste of the Zn accumulator Sedum plumbizincicola via the THL process showed the production of hydrochar, bio-oil, and carboxylic acids. Approximately 90% of Zn was released from biomass during THL at 200 °C. Consequently, hydrochar was poor in this metal and could be used as fertilizer [60].

Aquatic weeds are rich in nitrogen and phosphorus, so their potential to generate coproducts such as fertilizers and compost from their waste residues after biofuel production needs to be evaluated as it reduces the production cost and further improve the system efficiency [55]. In the same way, Shen et al. recognized the possibilities to valorize the contaminated biomass by pyrolysis, incineration, composting, and compaction, but alerted on the cost and security problems [61]. Moreover, it is important to improve the knowledge on the possible transfer of metals from contaminated aquatic plant to some by-products or residues during biofuel production [62].

Finally, after phytoremediation processes, various disposal and utilization methods, such as heat treatment (incineration, pyrolysis and gasification), extraction treatment (with liquid extractants), microbial treatment (compost and fermentation), compression landfill, and synthesis of nanomaterials (to use in electrochemistry, catalysis, medical industry, etc.) should be applied for the treatment of plant biomass containing organic and inorganic pollutants [63].

6. Conclusions

The use of natural and synthetic chemicals in human activities imposes the need to reduce their transfer and accumulation in aquatic and terrestrial ecosystems through the adoption of efficient and ecological processes of depuration. Furthermore, the cost-effective removal of pollutants from effluents remains a challenge in industrial and urban wastewater treatments. Phytoremediation emerged as an alternative method to conventional treatment processes and represents an interesting technology as it is considered a low-cost, low-waste, and environmentally friendly method for cleaning up polluted sites. The choice of plant species used as phytoremediator depends on several factors, including: its abundance, its growth rate, and its tolerance and capacity to accumulate and/or degrade the pollutant. Over a period of thirty years, a large number of research studies have been developed trying to find, with the contribution of genetic engineering, the most appropriate species for optimal use in phytoremediation. However, genetic engineering has not yet gained public recognition as an ethical science.

Author Contributions: N.K. and H.D. have writing the manuscript; A.A. provided comments to improve the manuscript. All authors have read and agreed to the published version of the manuscript.

Funding: This research received no external funding.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this article.

References

- Cota-Ruiz, K.; Delgado-Rios, M.; Martínez-Martínez, A.; Núñez-Gastelum, J.A.; Peralta-Videa, J.R.; Gardea-Torresdey, J.L. Current findings on terrestrial plants–Engineered nanomaterial interactions: Are plants capable of phytoremediating nanomaterials from soil? *Curr. Opin. Environ. Sci. Health* 2018, 6, 9–15. [CrossRef]
- Susarla, S.; Medina, V.F.; McCutcheon, S.C. Phytoremediation: An ecological solution to organic chemical contamination. *Ecolog. Eng.* 2002, 18, 647–658. Available online: http://cfpub.epa.gov/si/si_public_comments.cfm (accessed on 20 March 2022). [CrossRef]
- Hattink, J.; Wolterbeek, H. Accumulation of ⁹⁹T_C in duckweed *Lemna minor* L. as a function of growth rate and ⁹⁹T_C concentration. *J. Environ. Radioact.* 2001, 57, 117–138. [CrossRef]
- Ali, S.; Abbas, Z.; Rizwan, M.; Zaheer, I.E.; Yavas, I.; Unay, A.; Abdel-Daim, M.M.; Bin-Jumah, M.; Hasanuzzaman, M.; Kalderis, D. Application of floating aquatic plants in phytoremediation of heavy metals polluted water: A review. *Sustainability* 2020, 12, 1927. [CrossRef]
- 5. Ansari, A.A.; Naeem, M.; Gill, S.S.; AlZuailbr, F.M. Phytoremediation of contaminated waters: An eco-friendly technology based on aquatic macrophytes application. *Egypt. J. Aquat. Res.* **2020**, *46*, 371–376. [CrossRef]
- 6. Sarwar, N.; Imran, M.; Shaheen, M.R.; Ishaque, W.; Kamran, A.S.; Matloob, A.; Rehim, A.; Hussain, S. Phytoremediation strategies for soils contaminated with heavmetals: Modifications and future perspectives. *Chemosphere* **2017**, *171*, 710–721. [CrossRef]
- 7. Lemoine, G. Flores et pollinisateurs des villes et des friches urbaines ... Entre nature temporaire et biodiversité en mouvement. *Bull. Soc. Bot. N. Fr.* **2016**, *69*, 103–116.
- 8. Ekperusi, A.O.; Sikoki, F.D.; Nwachukwu, O.E. Application of common duckweed (*Lemna minor*) in phytoremediation of chemicals in the environment: State and future perspective. *Chemosphere* **2019**, *223*, 285–309. [CrossRef]
- 9. Khellaf, N.; Zerdaoui, M. Phytoaccumulation of zinc by the duckweed, *L. gibba* L.: Effect of temperature, pH and metal source. *Desalin. Water Treat.* **2013**, *51*, 5755–5760. [CrossRef]
- 10. Say, R.; Yilmaz, N.; Denizli, A. Removal of chromium (VI) ions from synthetic solutions by the fungus *Penicillium purpurogenum*. *Eng. Life Sci.* **2004**, *4*, 276–280. [CrossRef]
- Iqbal, J.; Baig, M.A. Effect of Nutrient Concentration and pH on Growth and Nutrient Removal Efficiency of Duckweed (*Lemna minor*) from Natural Solid Waste Leachate. *Inter. J. Health Medic.* 2016, 1, 1–7. Available online: https://researchplusjournals.com/index.php/IJHM/article/view/219 (accessed on 20 March 2022).
- 12. Mufarrege, M.M.; Di Luca, G.A.; Hadad, H.R.; Maine, M.A. Adaptability of *Typha domingensis* to high pH and salinity. *Ecotoxicology* **2011**, *20*, 457–465. [CrossRef] [PubMed]
- 13. Casas, S. Modélisation de la Bioaccumulation de Métaux Traces (Hg, Cd, Pb, Cu et Zn) Chez la Moule *Mytilus galloprovincialis*, en Milieu Méditerranéen. Ph.D. Thesis, de l'Université de Toulon Var, Toulon, France, 2005; 363p.
- 14. Rostami, S.; Azhdarpoor, A. The application of plant growth regulators to improve phytoremediation of contaminated soils: A review. *Chemosphere* **2019**, 220, 818–827. [CrossRef] [PubMed]
- 15. Oladoye, P.O.; Olowe, O.M.; Asemoloye, M.D. Phytoremediation technology and food security impacts of heavy metal contaminated soils: A review of literature. *Chemosphere* **2021**, *228 Pt 2*, 132555. [CrossRef] [PubMed]
- Mishra, A. Phytoremediation of heavy metal-contaminated soils: Recent advances, challenges, and future prospects. In *Bioreme*diation for Environmental Sustainability Toxicity, Mechanisms of Contaminants Degradation, Detoxification, and Challenges; Elsevier: Amsterdam, The Netherlands, 2021; Chapter 2; pp. 29–51.
- 17. Remon, E. Tolérance et Accumulation des Métaux Lourds par la Végétation Spontanée des Friches Métallurgiques: Vers de Nouvelles Méthodes de Bio-Dépollution. Ph.D. Thesis, de l'Université de Saint-Etienne, Saint-Etienne, France, 2006; 166p.
- Odoh, C.K.; Zabbey, N.; Sam, K.; Eze, C.N. Status, progress and challenges of phytoremediation-An African scenario. J. Environ. Manag. 2019, 237, 365–378. [CrossRef]
- Zhou, J. Present situation and prospects of technologies for remediation of heavy-metal-contaminated soil around Jiangxi Guixi Smelter. World Environ. 2016, 161, 48–53. [CrossRef]
- Kikuchi, R.; Gorbacheva, T.T.; Gerardo, R. Application of phytoremediation from experimental stage to practical stage: Comparative study in the Southern part and the Northern part of the European region. In *Handbook of Phytoremediation*; Nova Science Publishers, Inc.: Hauppauge, NY, USA, 2011; pp. 613–629.
- 21. Akhtar, A.B.T.; Yasar, A.; Ali, R.; Irfan, R. Phytoremediation using aquatic macrophytes. phytoremediation. In *Phytoremediation*; Ansari, A., Gill, S., Gill, R.R., Lanza, G., Newman, L., Eds.; Springer: Cham, Switzerland, 2017; pp. 259–276.

- 22. Dushenkov, V.; Kumar, P.N.; Motto, H.; Raskin, I. Rhizofiltration: The Use of Plants to Remove Heavy Metals from Aqueous Streams. *Environ. Sci. Technol.* **1995**, *29*, 1239–1245. [CrossRef]
- Chandanshive, V.; Kadam, S.K.; Khandare, R.V.; Kurade, M.B.; Jeon, B.-H.; Jadhav, J.P.; Govindwar, S.P. In situ phytoremediation of dyes from textile wastewater using garden ornamental plants, effect on soil quality and plant growth. *Chemosphere* 2018, 210, 968–976. [CrossRef]
- 24. Li, W. Environmental opportunities and contraints in the reproduction and dipersal of aquatic plants. *Aquatic Bot.* **2014**, *118*, 62–70. [CrossRef]
- 25. Mishra, V.K.; Tripathi, B.D. Concurrent removal and accumulation of heavy metals by the three aquatic macrophytes. *Bioresour. Technol.* **2008**, *99*, 7091–7097. [CrossRef]
- Mishra, V.K.; Tripathi, B.D. Accumulation of chromium and zinc from aqueous solutions using water hyacinth (*Eichhornia crassipes*). *J. Hazard. Mater.* 2009, *164*, 1059–1063. [CrossRef] [PubMed]
- 27. Lakra, K.C.; Lal, B.; Banerjee, T.K. Application of phytoremediation technology in decontamination of a fish culture pond fed with coal mine effluent using three aquatic macrophytes. *Inter. J. Phytoremed.* **2019**, *21*, 840–848. [CrossRef] [PubMed]
- Kumar, V.; Singh, J.; Saini, A.; Kumar, P. Phytoremediation of copper, iron and mercury from aqueous solution by water lettuce (*Pistia stratiotes* L.). *Environ. Sustain.* 2019, 2, 55–65. [CrossRef]
- Galal, T.M.; Eid, E.M.; Dakhil, M.A.; Hassan, L.M. Bioaccumulation and rhizofiltration potential of *Pistia stratiotes* L. for mitigating water pollution in the Egyptian wetlands. *Int. J. Phytoremed.* 2017, 20, 440–447. [CrossRef]
- 30. Amare, E.; Kebed, F.; Berihu, T.; Mulat, W. Field based investigation on phytoremediation potentials of *Lemna minor* and *Azolla filiculoides* in tropical, semi arid regions: Case of Ethiopia. *Int. J. Phytoremed.* **2017**, *20*, 965–972. [CrossRef]
- Sarkheil, M.; Safari, O. Phytoremediation of nutrients from water by aquatic floating duckweed (*Lemna minor*) in rearing of African cichlid (*Labidochromis lividus*) fingerlings. *Environ. Technol. Innov.* 2020, 18, 100747. [CrossRef]
- 32. Sasmaz, M.; Obek, E.; Sasmaz, A. Bioaccumulation of Uranium and Thorium by *Lemna minor* and *Lemna gibba* in Pb-Zn-Ag Tailing Water. *Bull. Environ. Contam. Toxicol.* **2016**, *97*, 832–837. [CrossRef]
- 33. Gür, N.; Türker, O.C.; Bocük, H. Toxicity assessment of boron (B) by *Lemna minor* L. and *Lemna gibba* L. and their possible use as model plants for ecological risk assessment of aquatic ecosystems with boron pollution. *Chemosphere* **2016**, *157*, 1–9. [CrossRef]
- 34. Megateli, S.; Semsari, S.; Couderchet, M. Toxicity and removal of heavy metals (cadmium, copper and zinc) by *Lemna gibba*. *Ecotoxicol. Environ. Saf.* **2009**, 72, 1774–1780. [CrossRef]
- 35. Espinoza-Quiñones, F.R.; Módenes, A.N.; Thomé, L.P.; Palặcio, S.M.; Trigueiros, D.E.G.; Oliveira, A.P.; Szymanski, N. Study of the bioaccumulation kinetic of lead by living aquatic macrophyte *Salvinia auriculata*. *Chem. Eng. J.* **2009**, *150*, 316–322. [CrossRef]
- 36. Arshadi, M.; Abdolmaleki, M.K.; Mousavinia, F.; Foroughifard, S.; Karimzadeh, A. Nano modification of NZVI with an aquatic plant *Azolla filiculoides* to remove Pb (II) and Hg (II) from water: Aging time and mechanism study. *J. Colloid Interface Sci.* 2017, 486, 296–308. [CrossRef] [PubMed]
- 37. Li, B.; Yang, Z.; Zhang, T. The role of submerged macrophyte in phytoremediation of arsenic from contaminated water: A case study on *Vallisneria natans* (Lour.) Hara. *Ecotox. Environ. Saf.* **2018**, *165*, 224–231. [CrossRef]
- Prasertsup, P.; Ariyakanon, N. Removal of chlorpyrifos by water lettuce (*Pistia stratiotes* L.) and duckweed (*Lemna minor* L.). *Int. J. Phytoremed.* 2011, 13, 383–395. [CrossRef] [PubMed]
- Bhaskaran, K.; Nadaraja, A.V.; Balakrishnan, S.; Shah, L.B.; Gangadharan, P.P.V. Phytoremediation of perchlorate by free floating macrophytes. J. Hazard Mater. 2013, 260, 901–906. [CrossRef] [PubMed]
- 40. Panfili, I.; Bartucca, M.L.; Del Buono, D. The treatment of duckweed with a plant biostimulant or a safener improves the plant capacity to clean water polluted by terbuthylazine. *Sci. Total Environ.* **2019**, *646*, 832–840. [CrossRef]
- 41. Obermeier, M.; Schröder, C.A.; Helmreich, B. The enzymatic and antioxidative stress response of *Lemna minor* to copper and a chloroacetamide herbicide. *Environ. Sci. Pollut. Res.* **2015**, *22*, 18495–18507. [CrossRef]
- 42. Amare, E.; Kebed, F.; Berihu, T.; Mulat, W. Wastewater treatment by *Lemna minor* and *Azolla filiculoides* in tropical semi-arid regions of Ethiopia. *Ecol. Eng.* **2018**, *120*, 464–473. [CrossRef]
- Vafaei, F.; Khataee, A.; Movafeghi, A.; Salehi, L.; Zarei, M. Bioremoval of an azo dye by *Azolla filiculoides*: Study of growth, photosynthetic pigments and antioxidant enzymes status. *Int. Biodeterior. Biodegrad.* 2012, 75, 194–200. [CrossRef]
- 44. Chandanshive, V.; Rane, N.; Gholave, A.; Patil, S.; Jeon, B.; Govindwar, S. Efficient decolorization and detoxification of textile industry effluent by *Salvinia molesta* in lagoon treatment. *Environ. Res.* **2016**, *150*, 88–96. [CrossRef]
- Rane, N.; Patil, S.; Chandanshive, V.; Kadam, S.; Khandare, R.; Jadhav, J.; Govindwar, S. *Ipomoea hederifolia* rooted soil bed and *Ipomoea aquatic* rhizofiltration coupled phytoreactors for efficient treatment of textile wastewater. *Water Res.* 2016, 96, 1–11. [CrossRef]
- Chandanshive, V.V.; Rane, N.R.; Tamboli, A.S.; Gholave, A.R.; Khandare, R.V.; Govindwar, S.P. Co-plantation of aquatic macrophytes *Typha angustifolia* and *Paspalum scrobiculatum* for effective treatment of textile industry effluent. *J. Hazard Mater.* 2017, 338, 47–56. [CrossRef] [PubMed]
- He, Y.; Chi, J. Phytoremediation of sediments polluted with phenanthrene and pyrene by four submerged aquatic plants. *J. Soil Sediment.* 2016, 16, 309–317. [CrossRef]
- Nash, D.A.H.; Abdullah, S.R.S.; Abu Hasan, H.; Idris, M.; Othman, A.R.; Al-Baldawi, I.A.; Ismail, N.I. Utilisation of an aquatic plant (*Scirpus grossus*) for phytoremediation of real sago mill effluent. *Environ. Technol. Innov.* 2020, 19, 101033. [CrossRef]

- Dushenkov, S.; Vasudev, D.; Kapulnik, Y.; Gleba, D.; Fleisher, D.; Ting, K.C.; Ensley, B. Removal of Uranium from Water Using Terrestrial Plants. *Environ. Sci. Technol.* 1997, 31, 3468–3474. [CrossRef]
- 50. Schwitzguébel, J.P.; van der Lelie, D.; Baker, A.; Glass, D.J.; Vangronsveld, J. Phytoremediation: European and American Trends. J. Soils Sediments 2002, 2, 91–99. [CrossRef]
- Yadav, K.K.; Gupta, N.; Kumar, A.; Reece, L.M.; Singh, N.; Rezania, S.; Khan, S.A. Mechanistic understanding and holistic approach of phytoremediation: A review on application and future prospects. *Ecol. Engin.* 2018, 120, 274–298. [CrossRef]
- 52. Lee, J.; Kaunda, R.B.; Sinkala, T.; Workman, C.F.; Bazilian, M.D.; Clough, G. Phytoremediation and phytoextraction in Sub-Saharan Africa: Addressing economic and social challenges. *Ecotox. Environ. Saf.* **2021**, 226, 112864. [CrossRef]
- 53. Jiang, Y.; Lei, M.; Duan, L.; Longhurst, P. Integrating phytoremediation with biomass valorization and critical element recovery: A UK contaminated land perspective. *Biomass Bioenergy* **2015**, *83*, 328–339. [CrossRef]
- Kaur, M.; Kumar, M.; Sachdeva, S.; Puri, S.K. Aquatic weeds as the next generation feedstock for sustainable bioenergy. *Bioresour. Technol.* 2018, 251, 390–402. [CrossRef]
- 55. Wang, Z.; Zheng, F.; Xue, S. The economic feasibility of the valorization of water Hyacinth for bioethanol production. *Sustainability* **2019**, *11*, 905. [CrossRef]
- 56. Hejna, M.; Onelli, E.; Moscatelli, A.; Bellotto, M.; Cristiani, C.; Stroppa, N.; Rossi, L. Heavy-metal phytoremediation from livestock wastewater and exploitation of exhausted biomass. *Int. J. Environ. Res. Public Health* **2021**, 18. [CrossRef]
- Ginneken, L.V.; Meers, E.; Guisson, R.; Ruttens, A.; Elst, K.; Tack, F.M.G.; Vangronsveld, J.; Diels, L.; Dejonghe, W. Phytoremediation for heavy metal-contaminated soils combined with bioenergy production. *J. Environ. Eng. Landsc. Manag.* 2007, 15, 227–236. [CrossRef]
- 58. Buscaroli, A. An overview of indexes to evaluate terrestrial plants for phytoremediation purposes. *Ecol. Indicators* **2017**, *82*, 367–380. [CrossRef]
- Mthethwa, N.P.; Nasr, M.; Bux, F. Utilization of Pistia stratiotes (aquatic weed) for fermentative biohydrogen: Electron-equivalent balance, stoichiometry, and cost estimation. *Int. J. Hydrog. Energy* 2018, 43, 8243–8255. [CrossRef]
- Lee, K.T.; Ofori-Boateng, C. Biofuels: Production technologies, global profile, and market potentials. In Sustainability of Biofuel Production from Oil Palm Biomass; Springer: Singapore, 2013; pp. 31–74. ISBN 978-981-4451-70-3.
- 61. Shen, X.; Dai, M.; Yang, J.; Sun, L.; Tan, X.; Peng, C.; Ali, I.; Naz, I. A critical review on the phytoremediation of heavy metals from environment: Performance and challenges. *Chemosphere* **2022**, *291*, 132979. [CrossRef]
- 62. Edgar, V.N.; Fabian, F.L.; Mario, P.C.J.; Ileana, V.R. Coupling plant biomass derived phytoremediation of potential toxic-metalpolluted soils to bioenergy production and high-value by-products—A review. *Appl. Sci.* 2021, *11*, 2982. [CrossRef]
- Liu, Z.; Khanh-Quang, T. A review on disposal and utilization of phytoremediation plants containing heavy metals. *Ecotox. Environ. Saf.* 2021, 226, 112821. [CrossRef] [PubMed]