

Proceeding Paper

Efficient Detection of Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ Ions in Water by Zn-Tetrazolate-Based Two-Dimensional Metal-Organic Framework: A Comparative Study [†]

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Abstract: Excesses and shortages of some metal ions such as Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ pose a threat to human life. A Zn-based two-dimensional MOF has proven to be sensitive for Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ in water with Ksv values of 1.8×10^4 and 6.5×10^4 , respectively. The limit of detection (LOD) for these ions was found to be 2.5 μM and 0.15 μM for Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ ions, respectively. A comparative study was conducted using two 3D MOFs in terms of their detection ability, synthesis and structure.

Keywords: luminescent MOFs; fluorescence sensing; ion recognition; aminotetrazole



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1. Introduction

In order to detect trace amounts of Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ ions in water, relatively effective and fast methods of detection are required. Fluorescent sensors are suitable for detection as they are convenient and straightforward to use, with high sensitivity and accuracy [1]. Metal-organic frameworks (MOFs), or porous coordination polymers, are a class of highly ordered crystalline materials consisting of metal ions or metal clusters and organic linkers connected into three-dimensional lattices [2]. In recent years, metal-organic frameworks have become favorable sensing materials for cations/anions and small molecules [3] due to their several structural advantages over other materials [4]. The presence of metals such as Zn (II) and Cd(II) with π -conjugated organic linkers make MOFs good materials for fluorescence sensing [5]. Azolate-containing MOFs with uncoordinated nitrogen atoms can lead to strong interactions with heavy metals, which facilitates fluorescence quenching and consequently enhances the sensitivity of the MOF [6].

Given these considerations, a simple, small N,N-containing linker 5-aminotetrazole was selected as the secondary linker because it contains multiple binding sites, i.e., an amino group that can functionalize the MOF [7]. In this work, a 2D MOF (Zn (ATZ)(HCO₂)) BUT-25 (BUT = Beijing University of Technology) with a layer-pillared structure was synthesized [8] with a slightly modified method and applied for the detection of Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ ions in water. This MOF proved to be highly sensitive at detecting trace amounts of Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ in aqueous solution, with a limit of detection (LOD) of 0.25 μM and 0.15 μM for Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$, respectively. These values are less than the minimum values specified by US-EPA (15.7 μM and 2 μM (0.1 mg/L) [9,10]. In addition, a comparative study was performed with two different 3D MOFs, which also showed that BUT-25 is superior to the other two methods in terms of stability and recycling. Hence, we determined that BUT-25 can act as a selective and sensitive sensor for these ions.

2. Methods

The solid-state emission spectra of HATZ ligand and BUT-25 were investigated to check their fluorescence at room temperature. For the measurements, 20 mg of ground

BUT-25 was added into 20 mL of water and sonicated until a uniform suspension formed. Subsequently, 0.1 mL of 5mM cation solutions and 5 mM anion solutions were introduced into cuvettes preloaded with 1mL of BUT-25 suspension, and the mixture was shaken for consistency. The Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ detection limit was recorded with the decrease in fluorescence intensity of the BUT-25, which was observed after adding cation and anion aqueous solutions (Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$) with varying concentrations into 1 mL MOF suspensions in DMF, respectively.

3. Result and Discussion

3.1. Structural Description

The structures of all the synthesized MOFs have been reported previously [8,11]. BUT-25 consists of honeycomb-shaped bilayers, in which a Zn atom is tetrahedrally connected to three N atoms from three deprotonated ATZ- (5-aminotetrazole) and one oxygen atom from formate ions (HCOO^-). Adjacent Zn atoms are connected by deprotonated ATZ⁻, while bilayers are connected via formate ions, as shown in Figure 1a,b. For comparison, BUT-26 was synthesized through the sequential addition of thiophene-2,5-dicarboxylic acid (H_2TDC) to the crystals of BUT-25. The formate ions were replaced by H_2TDC without disturbing the original structure of the honeycomb-shaped layers. It appeared that these layers adjusted themselves to accommodate the sequentially inserted linker, as shown in Figure 1c. BUT-27 was synthesized through the sequential addition of isophthalic acid (H_2IPA) to the crystals of BUT-25. The linker did not replace any of the BUT-25 linkers; rather, it converted into a totally new structure similar to zeolite LTA topology, as shown in Figure 1d.

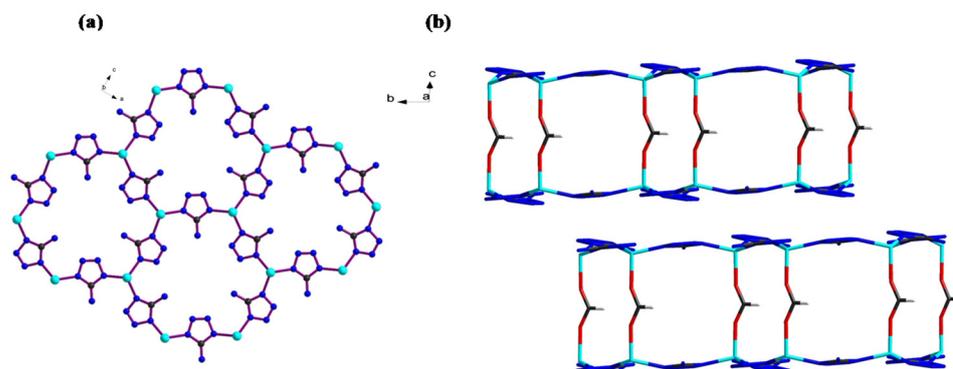


Figure 1. (a) honeycomb-shaped layer of BUT-25 (b) bilayers connected by formate ions in BUT-25.

3.2. Fluorescence Properties

Interestingly, Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ quenched the fluorescence of BUT-25 by nearly 80% and 90%, respectively; the rest of the ions featured lower quenching efficiencies, as shown in Figure S2. It is clear from Figure S3b,d that the S-V plots for Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ ions exhibited a good linear correlation ($R^2 = 0.994, 0.993$) in the concentration range of 0.1–0.1 μM for $\text{Cr}_2\text{O}_7^{2-}$ and 1–10 μM for Fe^{3+} ions. As the concentration increased, the S-V plots deviated from linearity and bent upwards, as shown in Figure S4a,b, indicating its high potential for practical application.

3.3. Sensing Mechanism

The PXRD patterns of BUT-25 before and after the fluorescence experiments were in accordance with each other, excluding the possibility of structural damage (Figure S1) [12]. The intense color change visible to the naked eye after the fluorescence tests indicates a strong interaction between the MOF and the target ions, as shown in Figures S5b,c, which might be responsible for detecting $\text{Cr}_2\text{O}_7^{2-}$ and Fe^{3+} ions [13]. The UV-Vis absorption spectra of selected anions clearly shows that the absorption spectrum of $\text{Cr}_2\text{O}_7^{2-}$ consists of two bands in the 200–600 nm range, which covered the excitation wavelength (290 nm)

and emission spectrum of BUT-25, while absorption spectrum of Fe^{3+} ion covered the excitation wavelength of BUT-25, as shown in Figures S6 and S7, pointing towards fluorescence quenching of BUT-25 due to the inner filter effect and resonance energy transfer mechanism [14]. This overlap would reduce the excitation of BUT-25, resulting in weaker fluorescence in the presence of $\text{Cr}_2\text{O}_7^{2-}$ and Fe^{3+} ions. However, no such overlap was observed for other anions and cations. Different metal cations may show different quenching behavior of luminescence because all cations feature different unsaturated electronic configurations. In comparison to BUT-26 and BUT-27, the performance of BUT-25 was quite satisfying in stability and recycling. The stability of both 3D structures was unsatisfactory as they could not recycle, even after washing with different organic solvents, such as DMF.

4. Conclusions

We successfully employed a 2D-MOF BUT-25 for the detection of Fe^{3+} and $\text{Cr}_2\text{O}_7^{2-}$ ions in water with Ksv values of 1.8×10^4 and 6.5×10^4 , respectively, which is quite a high value compared with many reported MOFs. The 2D-MOF BUT-25 also shows good selectivity, accompanied by color change visible to naked eye and recycling. The post-synthetic modification of BUT-25 through the sequential insertion of dicarboxylic linkers provides an example of how to post-synthetically modify a 2D MOF into a 3D MOF, which is helpful in widening the applications of MOF.

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