



Proceeding Paper Critical Variables Influencing the Ultrasound-Assisted Extraction of Bioactive Compounds—A Review ⁺

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Abstract: Ultrasound-assisted extraction (UAE) is a novel methodology, belonging to the so-called "Green Chemistry", which has gained interest in recent years due to the potential to recover bioactive compounds, especially those from plant matrices. It is widely recognized that the extraction of molecules by UAE gives rise to higher or similar yields than those obtained by traditional extraction methods. UAE has certain advantages inherent to Green Chemistry extraction methods, such as short extraction time and low solvent consumption. The aim of this review is to critically present the different variables and parameters that can be modified in UAE, such as ultrasound power, time, temperature, solvent, and solid to solvent ratio that influence yield and extraction performance.

Keywords: ultrasound-assisted extraction; critical variables; power; temperature; time; solvent

1. Introduction

Ultrasound-assisted extraction (UAE) is a technique that belongs to the group of novel extraction methods, together with microwave assisted extraction (MAE), enzyme assisted extraction (EAE) or high-pressure assisted extraction (HPAE) [1,2]. UAE promotes the extraction of compounds of interest, lowering the consumption of resources, such as solvent and energy, whereas achieving remarkably higher extraction yields [3,4]. In addition, UAE is a multipurpose method that lends itself to be combined with other extraction methods, both conventional and novel [5]. UAE has been applied to obtain extracts rich in bioactive compounds, such as phenolic compounds, pigments, polysaccharides, and amino acids, among others from plant matrices [1,3,6,7].

This methodology is based on the principle of cavitation, which leads to cell collapse of the matrix and allows the release of their inner substances. Several variables are relevant for the performance of UAE, including the solid–liquid ratio, the type of solvents used, the extraction time and the ultrasound power applied. Besides, ultrasound power and extraction time are closely linked to a fifth important factor, which is temperature. In practical terms, a correct optimization of these variables is essential to obtain a correct performance, resulting in a maximal extraction yield. In addition, temperature can affect the integrity of the bioactive compounds, since most of them are thermolabile. Considering that high ultrasound power linked to long extraction periods may lead to sample damage, temperature control is essential for a correct design of the cooling reactor and the optimization of UAE extraction protocols. Keeping all this in mind, this critical review is focused on



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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). the influence of all the variables that affect UAE, to analyze the critical factors involved in the optimization of this technique. In addition, response surface methodology (RSM) can be a representative tool to generate meta-models. RSM allows to analyze and optimize several variables at the same time and minimize the loss of matrices [8,9].

2. Variables Influencing Ultrasound-Assisted Extraction

To obtain good extraction yields, an optimization of the variables that influence the process is required. Among the variables that affect UAE, there are three types of parameters, as follows: physical, medium-dependent, and matrix-dependent parameters. Regarding the first, physical parameters are related to the ultrasonic waves applied during UAE and the equipment used. In this sense, those attributed to ultrasonic waves are ultrasound power, frequency, and ultrasound intensity (UI), whereas those related with ultrasound equipment are extraction time, and shape and size of the ultrasonic reactor. Medium-dependent parameters are related with the space in which ultrasound waves are transmitted from the emitting source to the matrix. Solvent properties, temperature and the presence of gases are examples of medium-dependent parameters. Finally, matrixdependent parameters are those that have a significant influence in the extraction of target compounds and considerably affect the effectiveness of the extraction. Type of matrix, structure, pre-treatment, particle size, or solid–liquid ratio are examples of those parameters [3,10]. Therefore, a correct design of the process, optimization of the variables, and appropriate equipment is needed to obtain extraction yields comparable to those obtained by the so-called traditional methods [11].

Regarding the ultrasound power, the use of high values usually improves extraction yields due to the generation of strong shear forces, so it is considered as one of the critical parameters to be optimized. Furthermore, higher ultrasound power reduces the time of extraction. For example, a study showed good results of the extraction of β -d glucans at a high extraction power (590 W), in only 58 min [12]. However, the use of high ultrasound power without control can overheat the reactor producing degradation of labile compounds and solvent evaporation [13]. In addition, the higher ultrasound power, the higher UI, so that when UI reaches the maximum value can produce liquid agitation and the consequent loss of ultrasound wave and the reduction of cavitation efficiency [14].

Regarding the extraction time, UAE allows to obtain good extraction yields with relatively short processing times (maximum 60 min) since longer times may cause undesirable changes in the extracted compounds. In this sense, optimized time commonly ranges between 20–60 min, minimizing the energy consumption and reducing the compounds' exposure to the process [15]. For example, one study shows that 7.25 min are enough to extract pigments from annatto seeds [16], while 37 min are needed to extract betacyanin and betaxanthin in bougainvillea flowers [17]. In the case of amino acid extraction, shorter extraction time was needed (6 min) [18,19], whereas other authors were able to extract polysaccharides from purple glutinous rice bran (*Oryza sativa*) with an extraction time of 20 min at 70 °C [20] (Table 1).

The type or polarity of the solvent used is closely linked to the nature of the compounds to be extracted. In addition, due to current concern for the environment, eco-friendly solvents are preferred. Predominantly, an aqueous medium is generally chosen for the extraction of polar compounds used in food matrices, while in the case of other organic compounds, ethanol, and methanol are usually employed. However, despite the use of methanol tends to obtain better extraction yields, ethanol is preferably chosen because of its lower toxicity [10]. For example, distilled water and ethanol are usually used for pigments extraction, while water is the most common solvent for the extraction of polysaccharides and amino acids [12,20,21] (Table 1). Furthermore, European Directive 2010/59/EU lists the solvents that can be used for the extraction of compounds from foodstuffs, as well as their uses and limitations [22]. In addition to the suitable solubility of the compounds of interest, it is also important to consider the vapor pressure, the surface tension, and the viscosity of the solvent, since those may affect cavitation and the extraction yield [3]. The

solid-to-solvent ratio used for each compound does not follow a certain pattern. It depends especially on the type of solvent and the matrix used. Table 1 shows that for the extraction of phenolic compounds, the most used solvent is ethanol and the solid-to-solvent ratio varies between 0.025 g/mL [23] and 0.1 g/mL [24], whereas ratios vary between from 0.058 g/mL for the extraction of betacyanin and betaxanthin in bougainvillea flowers [17] to 0.14 g/mL for the extraction of a natural pigment from annatto seeds [16].

Finally, high extraction temperatures not only affect the extraction yield but could also have negative effects due to the possible degradation of thermolabile compounds [14]. For this reason, the cooling system must allow the extraction of compounds avoiding the overheating of the medium by controlling the temperature of the system. For example, it is possible to extract phenolic compounds with temperatures up to 75 °C, β -d glucans at 81 °C or amino acids at 70 °C with optimal yields [12,18,25]. The increase in the temperature caused by the ultrasound probe itself is fundamentally produced when high ultrasound power is applied. The temperature increase produces a decrease in both viscosity and surface tension and induces an increase in the vapor pressure. Thus, too high temperatures can be harmful for the propagation of ultrasounds through the medium [13]. For these reasons, the optimization of the target components and improving the extractive properties of the solvent. Generally, the temperature does not exceed 80 °C, and it commonly works around 50 °C (Table 1).

Table 1. Optimized extraction conditions for phenolic acid	s, pigments, polysaccharides,	and amino acids from di	fferent
vegetal matrices by UAE.			

Compounds	Matrix	Solvent Type and Ratio	Optimized Parameters	Ref.
TFC	Blueberry pomace	EtOH/water 50%/50%; 0.05 g/mL	ET: 60 min; UPA: 64 W; T: 40 $^\circ \text{C}$	[24]
TFC	Zea mays waste	EtOH/water 70%/30% 0.09 g/mL	ET: 40 min; UPA: 50 W; T: ND	[23]
Stilbenes	Grape canes	EtOH/water 60%/40% 0.025 g/mL	ET: 10 min; UPA: 200 W; T: 75 $^\circ C$	[25]
Anthocyanins	Grape skins	EtOH 60% acidified pH = 3; 0.033 g/mL	ET: 28 min; UPA: 400 W; T: 50 $^\circ \text{C}$	[26]
Natural pigment	Annatto seeds	Dw, previously treated with chloroform/ratio solid solvent 0.14 g/mL	ET: 7.25 min; UPA: 200 W; T: 72.7 °C. (Duty cycle of 0.8 s)	[16]
Betacyanin and betaxanthin	Bougainvillea glabra flowers	Distilled water 100% 0.058 g/mL	ET: 37 min; UPA: 88 W; T: 55 °C	[17]
Natural yellow pigment	Physalis pubescens L.	EtOH 75% 0.083 g/mL	ET: 14 min; UPA: 180 W; T: ND; ultrasonic interval time of 10.55 s	[27]
B-d-glucan	Ganoderma lucidum	Distilled water 0.00004 g of fiber/mL	ET: 58 min; UPA: 590 W; T: 81 °C	[12]
Polysaccharides	Oryza sativa L.	Distilled water 0.05 g/mL	ET: 20 min; UPA: 150 W; T: 70 $^\circ C$	[20]
PSMP	Perilla seed meal	Distilled water 0.038 g/mL	ET: 52 min; UPA: 229 W; T: 43 $^{\circ}$ C	[21]
Amino acids	Grapes	Distilled water 0.1 g/mL to 0.05 g/mL not significant differences	ET: 6 min; UPA: 140 W; T: 70 $^\circ \mathrm{C}$	[19]
Amino acids	Apocynum venetum	Distilled water 0.00047 g/mL	ET: 32 min; UPA: 187 W; T: NI	[18]

Abbreviations: TFC: total flavonoid content. UPA: ultrasonic power amplitude. T: temperature; ET: extraction time; PSMP: perilla seed meal polysaccharides; NI: not included; Dw: dry weight.

3. Conclusions

UAE is a useful method for obtaining different compounds of interest from plant matrices, since remarkably higher extraction yields are obtained with short extraction times. This extraction method belongs to "Green Chemistry" because it allows to decrease the consumption of resources, such as solvent and energy. However, it is still necessary to optimize the more relevant variables that influence the effectiveness of UAE, such as ultrasound power, extraction time and temperature, type of solvent, and solid-to-solvent ratio. Different studies have shown extraction yields of different bioactive compounds, such as phenolic compounds, polysaccharides, pigments, and amino acids, using UAE with short extraction times (maximum 60 min), medium ultrasound power (between 200–500 W), temperatures around 50 °C (maximum 80 °C), and environmentally friendly solvents (distilled water and ethanol). Therefore, UAE can be appropriately applied to obtain bioactive compounds through an efficient and eco-friendly process, considering and optimizing the different critical variables that affect the process.

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