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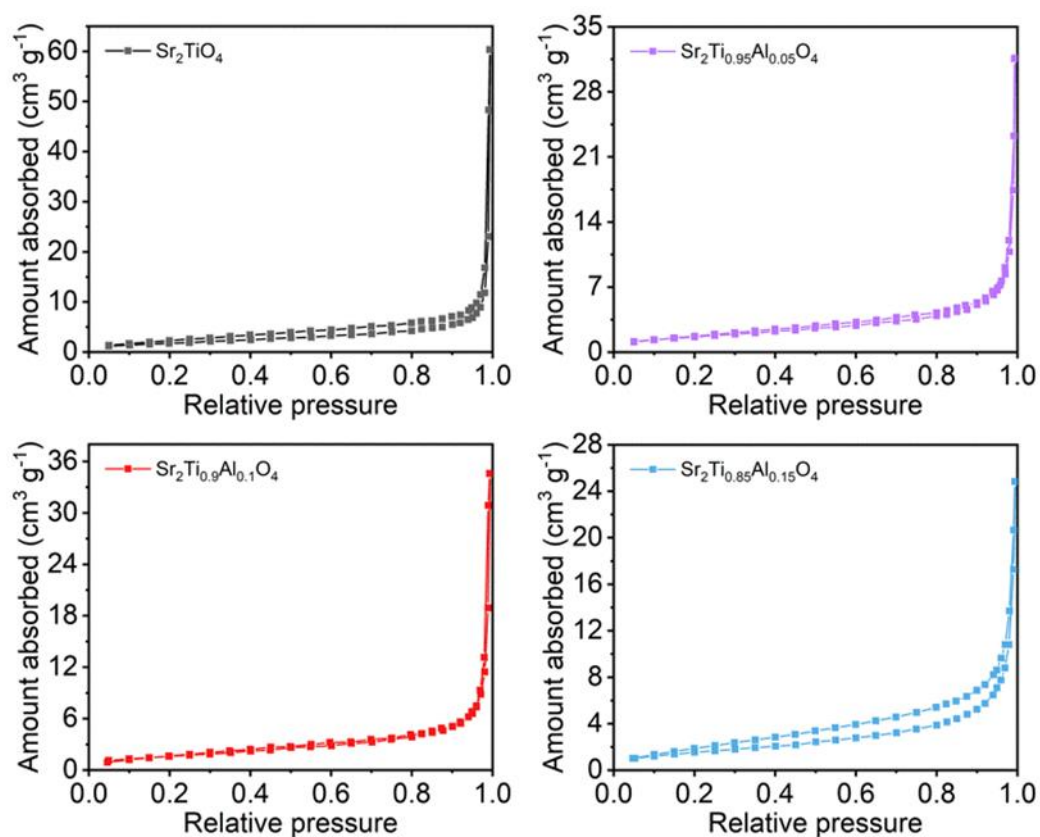
# Supplementary Materials: Aluminum Cation Doping in Rud-

## dlesden-Popper $\text{Sr}_2\text{TiO}_4$

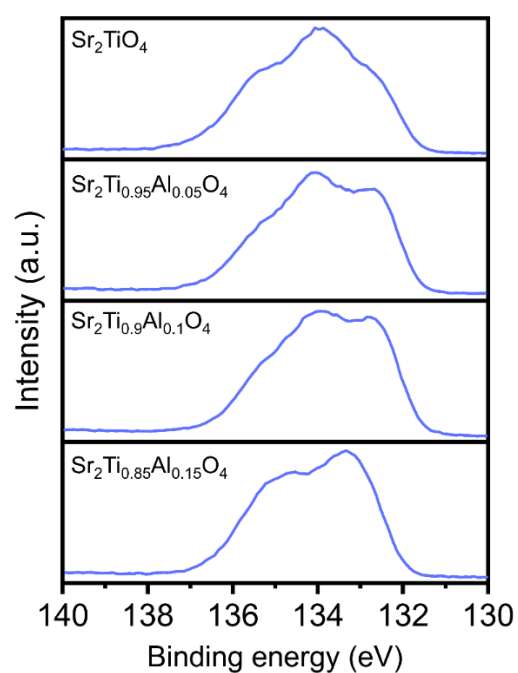
### Enables High-Performance Photocatalytic

### Hydrogen Evolution

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**Figure S1.**  $\text{N}_2$  adsorption and desorption curves of  $\text{Sr}_2\text{Ti}_{1-x}\text{Al}_x\text{O}_4$  ( $x = 0, 0.05, 0.1$  and  $0.15$ ).



**Figure S2.** Sr 3d XPS spectra of  $\text{Sr}_2\text{Ti}_{1-x}\text{Al}_x\text{O}_4$  ( $x = 0, 0.05, 0.1$  and  $0.15$ ).

**Table S1.** BET specific surface areas and band gap energies of various  $\text{Sr}_2\text{Ti}_{1-x}\text{Al}_x\text{O}_4$  samples ( $x = 0, 0.05, 0.1$  and  $0.15$ ).

Photocatalyst	BET specific surface area ( $\text{m}^2 \text{g}^{-1}$ )	Band gap energy (eV)
$\text{Sr}_2\text{TiO}_4$	6.729	3.32
$\text{Sr}_2\text{Ti}_{0.95}\text{Al}_{0.05}\text{O}_4$	6.093	3.20
$\text{Sr}_2\text{Ti}_{0.9}\text{Al}_{0.1}\text{O}_4$	5.933	3.06
$\text{Sr}_2\text{Ti}_{0.85}\text{Al}_{0.15}\text{O}_4$	5.797	3.16