

Computational Investigation of a NASICON-type Solid Electrolyte Material $\text{LiGe}_2(\text{PO}_4)_3$

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Table S1. Two body Buckingham potentials used for dopant oxides in $\text{LiGe}_2(\text{PO}_4)_3$.

Interaction	A (eV)	ρ (Å)	C (eV·Å ⁶)	Y (e)	K (eV·Å ⁻²)
Na ⁺ –O ²⁻	1497.830598	0.287483	0.00	1.000	99,999
K ⁺ –O ²⁻	1000.300	0.36198	10.56900	1.000	99,999
Rb ⁺ –O ²⁻	1010.80	0.3793	0.000	1.000	99,999
Al ³⁺ –O ²⁻	1114.9	0.3118	0.000	3.000	99,999
Ga ³⁺ –O ²⁻	2901.12	0.2742	0.000	1.000	99,999
Sc ³⁺ –O ²⁻	1575.85	0.3211	0.000	3.000	99,999
In ³⁺ –O ²⁻	1495.65	0.3327	4.33	3.000	99,999
Y ³⁺ –O ²⁻	1345.10	0.3491	0.000	3.000	99,999
Gd ³⁺ –O ²⁻	1885.75	0.3399	20.34	3.000	99,999
La ³⁺ –O ²⁻	1545.21	0.3590	0.000	−0.250	145.0
Si ⁴⁺ –O ²⁻	1283.910	0.32052	10.66	4.000	99,999
Sn ⁴⁺ –O ²⁻	1414.32	0.3479	13.66	4.000	99,999
Ti ⁴⁺ –O ²⁻	5111.7	0.2625	0.00	−0.10	314.0
Zr ⁴⁺ –O ²⁻	985.869	0.3760	0.00	1.35	169.617
Ce ⁴⁺ –O ²⁻	1986.83	0.3511	20.40	7.70	291.75