Supporting Information

Ta₂O₅/SiO₂ Multicomponent Dielectrics for Amorphous Oxide TFTs

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Compositional and structural analysis of the multicomponent dielectrics by Rutherford backscattering spectrometry (RBS), spectroscopic ellipsometry (SE), X-ray diffraction (XRD) and Atomic Force Microscopy (AFM).

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Ta ₂ O ₅	Density	SiO ₂	Ta ₂ O ₅	SiO ₂	Ar	Norm.	Norm.	Thickness
power	(1015	power	(mol.%)	(mol.%)	(at.%)	Ta ₂ O ₅	SiO ₂	(nm)
(W)	atm/cm ²)	(W)				(mol.%)	(mol.%)	
50	1668	150	37.5	59.1	3.4	39	61	243
75	1495	150	57.6	38.6	3.8	60	40	206
100	1796	150	65.7	30.0	4.3	69	31	245
125	1712	150	70.6	25.0	4.5	74	26	233
150	1923	150	71.7	23.9	4.4	75	25	257
100	1855	_	94.5	_	5.5	100	0	232

Table S1. Molar concentrations and film density obtained by RBS analysis of Ta_2O_5 , SiO₂ and Ar in the thin films deposited using different powers in the Ta_2O_5 and SiO₂ targets. Thickness of the films extracted by SE from [53].

While the Ar content is relatively similar across the film compositions presented in Table S1, a small tendency is still noticeable with the increase of the Ta₂O₅ content. The total atomic fraction of Ar incorporated in the material, x_{Ar} , can be described as the sum of the fractions incorporated in both the Ta₂O₅ material and the SiO₂ material, as shown in Equation (1),

$$x_{Ar} = x_{Ar@Ta_2O_5} x_{Ta_2O_5} + x_{Ar@SiO_2} x_{SiO_2}$$
(1)

where $x_{Ta_2O_5}$ and x_{SiO_2} are the non-normalized molar fractions of Ta₂O₅ and SiO₂, respectively, and $x_{Ar@Ta_2O_5}$ and $x_{Ar@Ta_2O_5}$ and $x_{Ar@Ta_2O_5}$ are the fraction of Ar incorporated in each of these materials, respectively. Knowing that the fractions of each material add to 1, as shown in Equation (2), Equation (1) can then be rewritten as Equation (3), describing the Ar content as a function of the Ta₂O₅ content, as presented in Figure S1. Trough linear fitting of this data, the fractions of Ar incorporation in each material,

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 $x_{Ar@SiO_2}$ and $x_{Ar@Ta_2O_5}$, are calculated as 1.9 % and 5.7 %, respectively, proving that Ta₂O₅ is much more prone for Ar incorporation than SiO₂.

$$1 = x_{Ar} + x_{Ta_2O_5} + x_{SiO_2} \tag{2}$$



Figure S1. Ar concentration as a function of the Ta₂O₅ concentration.

Capacitance-Voltage characteristics of MIS structures.

C-V curves measured from the MIS structures. The thicknesses (as described in Table S1) and MIS areas are also included in the figure.



Ta ₂ O ₅ (mol.%)	SiO ₂ (mol.%)	Thickness (nm)	Area (cm²)
39	61	243	1.60×10^{-3}
60	40	206	1.88×10^{-3}
69	31	245	1.88×10^{-3}
74	26	233	1.88×10^{-3}
75	25	257	1.88×10^{-3}
100	0	232	1.88×10^{-3}

Figure S2. C-V curves of the MIS employing the T_xS_{100} - x dielectric layers. The composition and geometrical properties of the MIS are presented in the table.





Figure S3. (a) V_{th} shift measured during gate biasing with V_{GS} = 10 V, V_{GS} = 0 V and V_{GS} = -10 V, sequentially, for the T₆₉S₃₁ composition. (b) V_{th} shift measured during negative gate biasing with V_{GS} = -10 V for an as-fabricated device with the T₃₉S₆₁ composition.

While the ΔV_{th} shift (observed only during the first minutes) during NGBS in as-fabricated devices is much lower than during PGBS, the channel is depleted during NGBS and thus only charges in the overlap areas between the gate and the source and drain electrodes are expected to respond to the biasing, making a direct comparison of these shift magnitudes unfeasible.

Power law time dependency of ΔV_{th} during positive gate bias stress



Figure S4. ΔV_{th} during positive gate bias stress (V_{GS} = 10 V) for the (a) T₆₀S₄₀, (b) T₆₉S₃₁ and (c) T₆₉S₃₁/SiO₂ compositions, showing power law time dependencies.