

Review Green Synthesis of Gold Nanoparticles: An Eco-Friendly Approach

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Abstract: By virtue of their unique physicochemical properties, gold nanoparticles (AuNPs) have gained significant interest in a broad range of biomedical applications such as sensors, diagnosis, and therapy. AuNPs are generally synthesized via different conventional physical and chemical methods, which often use harmful chemicals that induce health hazards and pollute the environment. To overcome these issues, green synthesis techniques have evolved as alternative and eco-friendly approaches to the synthesis of environmentally safe and less-expensive nanoparticles using naturally available metabolites from plants and microorganisms such as bacteria, fungi, and algae. This review provides an overview of the advances in the synthesis of AuNPs using different biological resources with examples, and their profound applications in biomedicine. A special focus on the biosynthesis of AuNPs using different medicinal plants and their multifunctional applications in antibacterial, anti-inflammatory, and immune responses are featured. Additionally, the applications of AuNPs in cancer theranostics, including contrast imaging, drug delivery, hyperthermia, and cancer therapeutics, are comprehensively discussed. Moreover, this review will shed light on the importance of the green synthesis approach, and discuss the advantages, challenges, and prospects in this field.

Keywords: gold nanoparticles; green materials; biosynthesis; plant extracts; microorganisms; biomedical applications

1. Introduction

Nanotechnology is a promising field that integrates the various disciplines of science, engineering, and technology. The rapid scientific advances in this field have led to the development of different types of functional nanoparticles (NPs), with at least one dimension in the typical size range of 1 to 100 nm. Among the various metallic NPs, gold nanoparticles (AuNPs) have attracted huge attention due to their unique surface plasmon resonance properties, facile synthesis, tunable sizes, and multifunctional abilities with wellcharacterized properties [1,2]. They are versatile materials, relatively inert, biocompatible, and generally stable. Due to their well-defined surface chemistry, AuNPs can be easily conjugated with different molecules such as proteins, dyes, drugs, antibodies, enzymes, and nucleic acids [3-5]. AuNPs functionalized with different targeting moieties have enormous scope in various biomedical applications such as diagnosis, targeting, drug/nucleic acid delivery, imaging, and therapy (Figure 1). Furthermore, by employing the surface-enhanced Raman scattering technique, AuNPs are used as sensitive probes in Raman scattering and imaging applications [6]. The potential of AuNPs in biomedical fields has been tremendously increased by virtue of their applications in photothermal therapy, radiation therapy, computed tomography, biosensors, etc. [7]. Due to their intrinsic electrical and optical properties, as well as their ability to conjugate with different biomolecules, AuNPs-based biosensors with high sensitivity and selectivity are being developed [8,9]. In the last decade, AuNPs-based biosensors have attracted great attention in the diagnosis of various types of diseases. Recently, Antonio et al. [10] highlighted the various AuNPs-based biological assays for the detection and quantification of analytes in urinary samples, with a focus on protein analysis. Such assays using AuNPs are useful in the diagnosis of several illnesses such as kidney disorders, cancer, and heart diseases [4,11].



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Figure 1. Multifunctional applications of gold nanoparticles.

Gold nanomaterials can be synthesized via different techniques in various internal structure, sizes, and shapes structures, including nanospheres, nanorods, nanocubes, nanoshells, nanowires, nanocages, nanoflowers, etc. [12]. They exhibit exceptional properties such as fluorescence, attenuation of X-rays, etc., and act as excellent contrast agents for optical, fluorescence, X-ray, and photoacoustic imaging [13,14]. Their intrinsic features (optics, electronics, and physicochemical characteristics) can be altered by adjusting their size and shape. To improve their compatibility and stability in a biological environment, AuNPs are coated with different biomolecules such as phospholipids, proteins, and polymers such as polyethylene glycol [15,16]. By fine-tuning the aspect ratio (length/width of the particle), gold nanorods can be manipulated to absorb light very strongly in the near-infrared region, convert it into heat energy, and transmit it to the surrounding environment. This process, called photohyperthermia, is widely used to attenuate cancer cells, where gold nanorods are administered near the tumor region to destroy the cancer cells without causing much damage to the healthy neighboring cells [17,18]. These properties have made AuNPs a widely used nanomaterial for global academic research and in the production of various industrial products and medical devices.

2. General Methods for Synthesis of AuNPs

2.1. Physicochemical Methods

NPs are typically synthesized via two basic methods: "top-down" or "bottom-up" (Figure 2). In the top-down approach, the constituent bulk materials are initially broken down to powder form and subsequently reduced to fine nanoparticles using various techniques such as etching, grinding, sputtering, thermal/laser ablation, etc. On the other hand, the bottom-up method involves the self-assembly of atoms to form nuclei, which then transform into particles of nanoscale range. The bottom-up method is widely used to obtain NPs with uniform morphology and chemical composition. The Turkevich method is a conventional chemical synthesis method commonly used to produce spherical small AuNPs around 10 to 30 nm in diameter [19]. However, it was observed that for the synthesis of AuNPs above 30 nm size, the results were less reproducible and resulted in a broader size distribution of particles [20]. The major limitation of this method is the strict process

control that needs to be followed and the precise maintenance of temperature, pH, and salt concentration to obtain monodisperse NPs in a specific size range. Alternatively, the Brust method and seed-mediated growth method are widely used for the synthesis of AuNPs [3].



Figure 2. Top-down and bottom-up approaches for nanoparticles synthesis.

The different methods of AuNPs synthesis, such as physical (ultrasonication, irradiation, electrochemical, etc.), chemical (vapor deposition, sol-gel process, etc.), and biological methods (using plants and microbial sources), are shown in Figure 3. However, the various physical and chemical methods used for the synthesis of NPs are expensive and hazardous to human health due to the use of toxic components and the production of byproducts, either during the synthesis or during the capping/stabilization process of NPs. The use of chemicals raises serious environmental and toxicity concerns when administered to living organisms. For instance, the surfactant cetyltrimethylammonium bromide (CTAB), which acts as a stabilizer and template for the growth of gold nanorods in the seed-mediated method, is reported to be highly toxic to different types of cells [21]. Therefore, gold nanorods stabilized with CTAB are not suitable for biomedical applications and need to be properly surface-modified with biocompatible materials such as phospholipids or polyethy-



lene glycol [22]. Hence, there is a growing requirement to search for reliable, non-expensive, biocompatible, and environmentally friendly methods for the synthesis of NPs.

Figure 3. Different methods for synthesis of gold nanoparticles.

2.2. Biosynthetic Mechanism of AuNPs

The mechanism of AuNPs biosynthesis is a simple two-step process and does not require a dramatic increase in temperature and pressure. In the first step, the biological extract (e.g., plant, bacterial, or fungal extract) is mixed with the HAuCl₄ salt solution, which causes the reduction of gold (Au³⁺) ions to gold atoms (Au⁰). In the second step, growth and stabilization result in the AuNPs formation (Figure 4). Finally, the color change of the resulting solution indicates the formation of AuNPs [23,24]. The chemical reactions involved in the reduction of Au³⁺ to Au⁰ in the presence of H₂O molecules are expressed in the below reactions:

Dissociation: HAuCl₄ $\xrightarrow{H_2O}$ H⁺ + Au³⁺ + 4Cl⁻,

(Chloroauric acid)

Oxidation: $4Cl^- \rightarrow 2Cl_2 + 4e^-$,

Reduction: $Au^{3+} + 4e^- \longrightarrow Au^0 + e^-$.

A variety of biocompounds (enzymes, phenols, sugars, etc.) can participate both in the reduction and stabilization of different types of particles, including AuNPs [25]. Figure 5a,b show the biosynthesis mechanism of bacterial microorganisms, which can act as a "factory" for the production of AuNPs. The biosynthesis mechanism of the microorganisms can be either extracellular or intracellular based on the location of AuNPs production [26]. Extracellular biosynthesis occurs outside the bacterial cell by trapping and reducing metal ions in the presence of enzymes. On the contrary, in the intracellular method, metal ions are transported into the microbial cell to form NPs in the presence of enzymes [27].



Figure 4. Schematic representation of AuNPs biosynthesis mechanism.

(a) Extracellular AuNPs biosynthesis



Figure 5. Schematic mechanism of (a) extracellular and (b) intracellular AuNPs biosynthesis.

3. Green Synthesis of AuNPs

3.1. Importance of Green Synthesis of AuNPs

As the applications of AuNPs are increasing day by day, the demand for their synthesis also increases simultaneously. It has aroused interest among researchers worldwide to develop novel interdisciplinary routes for the synthesis of highly stable, monodisperse, and

safe AuNPs for various applications. To overcome the challenges in the conventional chemical synthesis method, an alternative approach to synthesize biocompatible NPs, termed "green synthesis", has evolved. It is an emerging branch of nanotechnology and has attracted huge attention among researchers and industries, as well as people concerned about environmental pollution and health hazards. Green synthesis techniques are important as they are an eco-friendly approach that involves the use of natural bioresources and avoids toxic chemicals to synthesize different types of NPs [28]. For instance, green synthesis of cobalt ferrite nanoparticles using extracts of grape peel, pulp, and honey-mediated synthesis of cobalt-zinc ferrite NPs were reported earlier [29,30].

A variety of plants, microorganisms, and biomolecules derived from them are used as a source for the synthesis of various types of NPs. Extracts from different parts of the plant such as leaves, roots, seeds, flowers, fruits, bark, etc., and microbes, including bacteria, fungi, and algae, are widely used to synthesize NPs with varying sizes and shapes using interdisciplinary routes [19,31]. The precursor gold salt solution is treated either with the microbial culture or plant extracts, which are then bioreduced to form AuNPs. Different metabolites and biomolecules such as sugars, fatty acids, proteins, enzymes, and phenols play a key role in the synthesis of the AuNPs [32,33]. Further, this biological approach involves the use of (i) an environmentally acceptable solvent medium for NPs synthesis, (ii) natural reducing agents, and (iii) nontoxic capping agents, mostly polyphenols and other secondary metabolites. Green synthesis of AuNPs using different bioagents and their applications is shown in Table 1.

Bioagent	Size (nm)	Shape	Application	Ref.
Areca catechu	13.7	Spherical	Anticancer, Antibacterial, Antioxidant, Catalyst	[34]
Mangifera indica Linn	6–18	Spherical	Drug delivery	[35]
Olive leaves	50-100	Spherical, Triangular, Hexagonal	Antioxidant	[36]
Citrus limon	15-80	Spherical, Triangular	Anticancer, Antimicrobial, Anti-inflammatory	[37]
Coreopsis lanceolata	20-30	Spherical	Detections of aflatoxins	[38]
Musa paradisiaca	<50	Spherical, Triangular	Anticancer	[39]
Zingiber officinale	5-20	Spherical, Triangular, Hexagonal	Antibacterial	[40]
Cocoa extract	150-200	Spherical, prismatic, Rod	Photothermal Therapy, contrast agents	[41]
Capsicum annuum var. grossum	6–37	Quasi-spherical, Triangular, Hexagonal	Catalyst	[42]
Čitrus maxima	25	Spherical	Catalyst	[43]
Trianthema decandra	17.9–79.9	Spherical, Hexagonal, Cubical	Antimicrobial	[44]
Mammea suriga	22-50	Spherical, Square	Antibacterial	[45]
Abelmoschus esculentus	45-75	Spherical	Antifungal	[46]
Shewanella oneidensis	2-50	Spherical	Antibacterial	[47]
Streptomyces sp.	90	Cubical	Antifungal	[48]
Gordonia amara	15-40	Spherical, Polycrystalline	Biosensor	[49]
Bacillus stearothermophilus	5-30	Spherical, Triangular and other	Biosensor	[50]
Pseudomonas fluorescens	5-50	Spherical	Bactericidal	[51]
Stenotrophomonas malophilia	40	Spherical	Bioremediation	[52]
Sporosarcina koreensis DC4	30-50	Spherical	Catalyst	[53]
Rhizopus oryzae	10	Nanocrystalline	Pesticides	[54]
Fusarium semitectum	18-50	Spherical	Optoelectronics	[55]
Candida albicans	20-80	Spherical, Non-spherical	Detection of liver cancer	[56]
Volvariella volvacea	20-150	Spherical, Triangular, Hexagonal	Therapeutic	[57]
Helminthosporum solani	2-70	Polydispersed	Anticancer drug	[58]
Penicillium brevicompactum	10-50	Spherical	Anticancer	[59]
Verticillium sp.	20	Spherical	Biomedical	[60]
Alternaria alternata	12 ± 5	Spherical, Triangular	Biomedical	[61]
Verticillium luteoalbum	<10	Spherical, Triangular, Hexagonal	Optics and Sensor	[62]
Fusarium oxysporum	46-70	Spherical, Triangular	Biomedical	[63]
Aureobasidium pullulans	29 ± 6	Spherical	Biomedical	[64]
Pichia jadinii	10-100	Spherical, Triangular, Hexagonal	Optics and Sensor	[65]
Yarrowia lipolytica	7.5–27	Spherical, Triangular, Hexagonal	Biomedical	[66]
Gracilaria corticata	5	Spherical	Antibacterial, antioxidant	[67]
Shewanella algae	9.6-200	Spherical, Nanoplates	Biomedical	[68]

Table 1. Green synthesis of AuNPs using different bioagents and their applications.

3.2. Advantages of Green Synthesis of AuNPs

The green synthesis method offers the following advantages over the chemical methods: (i) safety: this method avoids the exposure of chemicals or their toxic byproducts, either during the NPs synthesis step or during their stabilization process; (ii) cost-effective: no external stabilizing agent is required. On the contrary, chemical methods use expensive and hazardous chemicals and stabilizing agents; (iii) simplicity: biosynthesis of NPs from plant extract is a simple process; (iv) renewable feedstock: e.g., algal biomass; (v) easy availability of source materials; (vi) biocompatibility: since natural sources are used to synthesize and stabilize the particles, AuNPs synthesized via green synthesis methods are biocompatible to different cell types; (vii) the whole green synthesis process is dynamic, reproducible, and energy-efficient; (viii) suitable for large scale production of NPs for commercial applications; (ix) AuNPs synthesized via green synthesis method are also reported to exhibit antibacterial, antifungal, anticancer, and anti-inflammatory properties, and antioxidant and catalytic activity due to the presence of phytochemicals from the bioextract [69,70]. All these factors have rendered the green synthesis approach more rewarding than conventional methods.

4. Biosynthesis of AuNPs from Plant Sources

Biosynthesis of AuNPs from plant sources is facile and involves a single-step process in the one-pot method. To synthesize AuNPs from plant sources, different parts of the plant (leaves, fruits, bark, flower, peels, seed, rhizome root, etc.) are washed with distilled water, dried, ground into powder or chopped into small pieces, and boiled in distilled water to a specific temperature to obtain the extract. Then, filtration or centrifugation techniques are used to purify the extract, which is then simply mixed with various concentrations of gold salt solution (based on the plant parts and their species). The gold salt solution is reduced into AuNPs and the reaction completes in minutes to a few hours. The reaction mixture is further incubated to reduce the gold salt completely, and is visually monitored by color change. Finally, the synthesized AuNPs are purified by centrifugation and washed thoroughly in water for further use. The whole process is simple, eco-friendly, and can be scaled up easily.

Plants are rich in alkaloids, flavonoids, saponins, steroids, tannins, and other natural compounds [71]. The plant extract contains various secondary metabolites, which act as both reducing and stabilizing agents for the biogenesis of NPs [72]. Numerous reports have shown the successful synthesis of different types of NPs, such as silver, copper, gold, cobalt, palladium, magnetite, and zinc oxide [73,74]. Although various parts of plants have been reported in the biosynthesis of AuNPs, leaves are widely used. Variations in the level of metabolites content from different plant parts, and even variation among plants, play a crucial role in shaping the morphology of NPs.

Islam et al. [75] reported a reproducible green synthetic method to produce highly stable AuNPs, using the leaves extract of the plant Salix alba L. (syn: white willow), which belongs to the family Salicaceae. The leaves and bark of this plant are rich in phenolic contents such as salicin, which acts as a precursor in the development of aspirin. Hence, they are traditionally used for musculoskeletal pain relief and treatment of different ailments, owing to their antipyretic and anti-inflammatory properties. When the aqueous gold ions were treated with Salix alba L. leaves extract, they were reduced, leading to the synthesis of AuNPs. The UV–Vis absorption spectra data revealed that the synthesized AuNPs were found to be colloidally stable at different pH and salt concentrations. The AuNPs functionalized with the phytochemicals of leaf extracts exhibited good antifungal activity, pain-relieving, and muscle relaxant effect, which enhances their potential for various biomedical and pharmaceutical applications. Narayan et al. [76] reported the extracellular synthesis of AuNPs using coriander leaf extract as the reducing agent. Transmission electron microscopy (TEM) images have shown the formation of stable AuNPs in the size range of 6.75–57.91 nm (Figure 6) with varying shapes, such as spherical, triangle, truncated triangles, and decahedral morphologies.



Figure 6. (**A**,**B**) TEM images of AuNPs synthesized by leaf extract of coriander. Reprinted with permission from Ref. [76]. Copyright 2008 Elsevier.

Similarly, Boruah et al. [77] reported a simple, faster, low-cost, eco-friendly technique to biosynthesize AuNPs using fresh young leaves and leaf buds of Camellia Sinensis (widely used to prepare tea). When chloroauric acid was treated with tea extract prepared from young leaves and leaf buds at room temperature, they were reduced (Au³⁺ \rightarrow Au⁰) by polyphenols present in the extract, leading to the formation of AuNPs. Chen et al. [78] reported that AuNPs (~8–25 nm) synthesized from the aqueous leaf extract of Curcumae Kwangsiensis Folium exhibited excellent, dose-dependent anti-human ovarian cancer potential due to their antioxidant properties. Biogenic AuNPs synthesized using different plant extracts exhibit remarkable cytotoxic antibacterial properties. For instance, Chahardoli et al. [79] studied the reduction of gold ions into spherical AuNPs (3–37 nm) using Nigella arvensis leaf extract in the one-step green synthesis method. They showed the cytotoxicity effects against H1299 and MCF-7 cancer cell lines with an IC50 value of 10 and 25 µg/mL, respectively. Fourier transform infrared spectroscopy (FTIR) analysis confirmed the presence of phytocompounds involved in the reduction and stabilization of NPs.

Apart from leaves, other parts of the plants, such as barks, flowers, fruits, seeds, and even peels of fruits, are used in the synthesis of NPs. Bahram et al. [80] used the bark extract from the S. alba (willow tree) to synthesize AuNPs which exhibited high potential as colorimetric sensors for selective recognition and monitoring of cysteine, among other amino acids. Leon et al. [81] synthesized AuNPs through a one-pot synthesis method using Mimosa tenuiflora (Mt) bark extract, which is rich in different types of polyphenols. The synthesized AuNPs were characterized by a series of analytical techniques including FTIR and X-ray photoelectron spectrometry for functional group determination. The results indicated that AuMt (colloids formed by AuNPs and molecules of Mt) NPs interact mainly with carbonyl groups (ketones), in addition to hydroxyl groups of Mimosa tannins, saponins, and other molecules that participate in the reduction of Au³⁺ to Au⁰ and stabilization of nanomaterials. Due to the fluorescence property at low excitation power and a high cellular uptake, AuMtNPs synthesized with Mt bark extracts are good candidates for implementation as drug nanocarriers and fluorescent probes in cells. Elmiitwalli et al. [82] reported the green synthesis of AuNPs using cinnamon bark extract, which acts both as reducing agent and stabilizer.

Wang et al. [83] synthesized stable AuNPs in the size range 10–30 nm, utilizing lignin NPs at room temperature without the addition of chemicals. Lignin–AuNP composites exhibited enhanced stability in suspension for more than 7 days. Parida et al. [84] reported the synthesis of cost-effective and environment-friendly AuNPs using onion (Allium cepa) extract as reducing agent. Lee et al. [85] reported the synthesis of AuNPs (32.96 ± 5.25 nm) by the reduction of aqueous gold metal ions in contact with the aqueous peel extract of the plant *Garcinia mangostana* (*G. mangostana*). FTIR results revealed the presence of

phenols, flavonoids, benzophenones, and anthocyanins, which suggests that they may act as reducing agent.

Reports [86,87] have revealed that the temperature and pH of the reaction mixture play an important role in determining the final size and shape of the synthesized AuNPs. For example, Bogireddy et al. [86] described the formation of size-tunable crystalline AuNPs using sundried Coffea arabica seed (CAS) extract at room temperature. The influence of pH on the size of AuNPs was investigated by manipulating the pH of the reaction mixture (pH 5, 7, 9, and 11). The size, shape, and crystallinity of the NPs were analyzed using different techniques, including TEM and X-ray diffraction (XRD). The results showed the formation of larger NPs (\sim 69 nm) at lower pH value (\sim 5), which was probably due to the limited availability of capping agents (OH^{-} functional groups), whereas smaller, quasi-spherical NPs (~13 nm) were formed at higher pH values (>10). Thus, the obtained results stipulate the possibility to manipulate the size and shape anisotropy of NPs by controlling the pH of the reaction mixture. FTIR results revealed that the phenolic groups present in the CAS extract helped to reduce Au^{3+} to Au^{0} and stabilize the synthesized AuNPs. Similarly, Oueslati et al. [87] reported the synthesis of ultra-small and large AuNPs using polyphenol extracted from the Salvia officinalis plant. In both alkaline ($pH \sim 11$) and acidic media ($pH\sim5$), polyphenols induced rapid reduction of the Au (III) salt and led to the formation of highly monodisperse, ultra-small (~ 6 nm), and larger (~ 27 nm) spherical AuNPs, respectively. FTIR results revealed that different polyphenols were capped onto the surface of NPs favoring high colloidal stability.

Anbu et al. [88] synthesized spherical-shaped AuNPs with an average size of 15 nm using Platycodon grandiflorum (balloon flower plant) extracts and evaluated their antibacterial potential against Escherichia coli and Bacillus subtilis. The synthesized AuNPs significantly inhibited bacterial growth and demonstrated their antibacterial applications. Sett et al. [89] reported a novel method of AuNPs synthesis using aqueous fruit extract of *Dillenia indica*. The high phenolic content of the aqueous core extract of *D. indica* with a strong antioxidant property helped in the reduction of gold ions to AuNPs. The phytochemicals present in the fruit extract as an effective reducing and capping agent to synthesize AuNPs. TEM images of AuNPs revealed an average size range of 5–50 nm, which is very promising for most biological applications. The synthesized AuNPs did not show any form of cytotoxicity in the normal fibroblast cell line L929, thus proving their compatibility.

Elia et al. [90] compared the biocompatibility and stability of AuNPs synthesized using the extracts of the following four different plants: Salvia officinalis, Lippia citriodora, Pelargonium graveolens, and Punica granatum. When chloroauric ions were treated with the extract of different plants, the gold ions were reduced to gold atoms, which then aggregated to form AuNPs. TEM images have shown the formation of smaller spherical/triangular NPs starting from about 10 nm in size, whereas larger particles (~150 nm) were also formed with different geometrical shapes, such as triangles, pentagons, and hexagons. The cytotoxicity studies of all the synthesized AuNPs on L-cells (a murine fibroblast cell line) did not show deleterious effects and expressed biocompatibility as well as high stability for over 3 weeks. Therefore, the synthesized NPs have the potential to be applied in biomedical applications. Literature data have shown the formation of AuNPs of varying geometrical shapes when extracts from different plants were used [91,92].

Rao et al. [93] reported the green synthesis of AuNPs (20–30 nm) by reducing chloroauric acid with flower and leaf extracts of Ocimum tenuiflorum, leaves of Azadirachta indica and Mentha spicata, and peel of Citrus sinensis plants. The synthesized AuNPs were tested on pathogenic Staphylococcus aureus (Gram-positive) and Pseudomonas aeruginosa (Gram-negative) bacteria, which are dangerous to humans and other living organisms. The phytochemicals from the plant extract formed *in situ* capping on the NPs surface and exhibited antibacterial properties up to 99%. The toxicity study inferred that the phytochemicals-capped AuNPs ruptured the bacterial cell wall and affected the normal metabolic process of pathogenic bacteria. Interestingly, the phytochemicals-capped AuNPs produced via green synthesis exhibited higher antibacterial activities than the other metallic NPs produced by chemical methods. This proves the benefits of NPs synthesized through green routes compared to chemical methods and their superior features for biomedical applications.

5. Biosynthesis of AuNPs Using Microorganisms

Green synthesis of AuNPs using different types of microorganisms such as bacteria, fungi, algae, actinomycetes, etc., has triggered great interest in industrial microbiology owing to the numerous benefits they offer. Easy handling and processing, low-cost medium for their growth, and ability to adsorb and reduce various metal ions into NPs are some of the attractive reasons [19,94,95]. Large-scale cultivation of microbes in bulk fermenters will enable the surplus extraction of enzymes and various secondary metabolites in a less economical way. Various fungal strains can be cultivated on different substrates such as cellulosic wastes, coir-pith, and agricultural wastes, thereby enabling the usage of less-expensive raw materials for their growth, helping in waste recycling and reducing environmental pollution [96].

Microorganisms such as bacteria, filamentous fungi, yeast, algae, and actinomycetes have huge scope in the bioremediation process and have the potential to degrade contaminants, such as heavy metals, dyes, and toxic chemicals, that pose environmental and human risks [97,98]. In other words, microbes can be effectively utilized to biologically degrade harmful pollutants into nontoxic substances.

5.1. Fungi and Algae

Fungi are excellent candidates for large-scale production of NPs because of the simplicity, high scalability, downstream processing, easy handling, and cost-efficiency of fungal growth on both the laboratory and the industrial scale. The filamentous fungi are well known for their high metal tolerance and bioaccumulation properties. Fungal cell walls possess different functional groups such as amine, carboxyl, sulfhydryl, hydroxyl, and phosphate groups, that act as ligands and help to chelate metal ions [99]. Further, they secrete a wide array of proteins and enzymes such as ATPase, 3-glucanase, hemicellulose, glyceraldehyde-3-phosphate dehydrogenase, cell wall lytic enzyme β -1, etc., which play an important role in the feasible, large-scale synthesis of metallic NPs. Fungi such as Penicillium chrysogenum, Fusarium oxysporum, and Verticillium sp. are reported in the biosynthesis of metallic NPs such as platinum, silver, silicon, and titanium [100].

Numerous reports have elaborated the biogenesis of AuNPs using unicellular and multicellular fungi [56,101]. Extracellular or intracellular extracts of different fungi such as Candida albicans, Aspergillus niger, Aspergillus clavatus, and Penicillium sp. are widely used for the synthesis of AuNPs [102,103]. Priyadarshini et al. [104] reported an ecofriendly, ambient temperature protocol for size-controlled synthesis of AuNPs, using the fungus Aspergillus terreus IF0. AuNPs were formed immediately by adding chloroauric acid to the aqueous fungal culture extract. TEM results have revealed that the particles were found to be in the size range of 10–19 nm. FTIR analysis has indicated the presence of carboxyl, amino, and thiol functional groups from the fungal extracts, which were responsible for both bioreduction and stabilization of NPs. The synthesized AuNPs demonstrated excellent antibacterial activity against the Gram-negative bacteria, Escherichia coli, and have exciting scope in clinical applications.

Quite recently, Nguyen et al. [105] demonstrated the green-synthesis of silver and gold NPs using Ganoderma lucidum, mushroom extract, as reducing and capping agents. The synthesized NPs showed excellent catalytic, antibacterial activity, and colorimetric detection of Fe^{3+} ions in real water systems and exhibited their outstanding properties in environmental and biotechnological applications. Sastry et al. [106] observed the intracellular and extracellular production of AuNPs using two different genera of fungi, Verticillium sp. and *Fusarium oxysporum*. When the aqueous gold and silver ions were exposed to *Verticillium* sp., the metal ions were reduced intracellularly to form gold and silver NPs in the size range 2–20 nm. On the other hand, the same aqueous gold and silver

ions were reduced extracellularly in the case of *F. oxysporum*, leading to the formation of gold and silver NPs around 2–50 nm in size. The active biomolecules produced by the fungi, the concentration of precursor gold salt solution, and optimization of the experimental conditions play an important role in controlling the size distribution, shape, and biochemical composition of the synthesized NPs. For instance, Dhanasekar et al. [107] explained a simple and eco-friendly approach to synthesize AuNPs of different sizes (7–93 nm) and shapes by exposing the cell-free filtrate of filamentous fungus Alternaria sp. to three different concentrations (0.3, 0.5, and 1 mM) of chloroauric solution. In all cases, the Au³⁺ ions were reduced to Au⁰, leading to the formation of stable AuNPs. TEM analysis has revealed the presence of spherical, square, rod, pentagonal, and hexagonal morphologies for 1 mM chloroauric solution and quasi-spherical and spherical NPs for lower concentrations (0.3 and 0.5 mM) of chloroauric solution. FTIR analysis has revealed the presence of aromatic primary amines, amino acids such as tryptophan/tyrosine, or phenylalanine as the capping and stabilizing agents on the surface of AuNPs.

Different approaches have been used for the biosynthesis of AuNPs from fungal extracts. However, there is no clear knowledge about the limitations of all these methods. To gain better understanding, Molnar et al. [108] investigated 29 different thermophilic filamentous fungal strains to compare the AuNPs formed using either the extracellular fraction, the autolysate of fungi, or the intracellular fraction of fungi. The results have shown that AuNPs of varying sizes (6-40 nm) with high standard deviations ranging between 30% and 70% were formed based on the difference in the fungal strain and environmental conditions. Mishra et al. [59] reported the fungus-mediated synthesis of AuNPs using an industrially important fungus Penicillium rugulosum. TEM results revealed that the size of synthesized NPs was in the range of 20-80 nm. The AuNPs were then conjugated with isolated genomic DNA of bacteria Escherichia coli and Staphylococcus aureus. Stability analysis results have shown that DNA-conjugated AuNPs were highly stable and monodispersed, which infers that the presence of genomic DNA on the surface of NPs prevents them from aggregation due to their negatively charged phosphate backbone. Such surface modification of AuNPs will improve their shelf life during in vivo applications and enhance their scope in biomedicine.

A variety of algae, such as Turbinaria conoides, Spirulina platensis, Galaxaura elongate, and Shewanella algae, are used as bionanofactories for the synthesis of AuNPs [67,68]. Singh et al. [109] reported the synthesis of AuNPs using aqueous extract of Dunaliella salina, a unicellular, halotolerant microalga. The synthesis, characterization, and in vitro anticancer activity of the biosynthesized AuNPs is shown in Figure 7. The anticancer potential of AuNPs was tested against the breast cancer cell line (MCF7) and normal breast epithelial cell line (MCF 10A), and commercial anticancer drug cisplatin was used as a positive control. The cell viability results (Figure 8A–D) have indicated that AuNPs synthesized using D. salina selectively attenuated cancer cells and were not detrimental to the normal cell line, whereas cisplatin affected normal cells as well at 48 h exposure. Chellapandian et al. [110] demonstrated a facile one-pot synthesis of AuNPs using an aqueous solution of the marine red seaweed, Gracilaria verrucosa. The biocompatibility of the synthesized AuNPs was assessed using human embryonic kidney (HEK-293) cells. The fluorescence microscopy images using Trypan blue exclusion and AO/EB staining have shown (Figure 9) that the cells treated with biosynthesized AuNPs (100 μ g/mL) appear similar to control cells, indicating the cell viability.



Figure 7. Green synthesis of gold nanoparticles from Dunaliella salina, UV–Vis characterization spectra, and in vitro anticancer activity on breast cancer cell line. Reprinted with permission from Ref. [109]. Copyright 2019 Elsevier.



Figure 8. Cell viability using MTT assay after treatment with AuNPs and cisplatin on MCF 10A (**A**,**B**) and MCF 7 cell lines (**C**,**D**). Reprinted with permission from Ref. [109]. Copyright 2019 Elsevier.



Figure 9. Biocompatibility of HEK-239 cells with AuNPs under fluorescence microscope evidenced by Trypan blue exclusion and AO/EB staining. (a) Control cells appear green with no evidence of cell death. (b) Cells treated with AuNPs (100 μ g/mL) appear similar to control cells. Reprinted with permission from Ref. [110]. Copyright 2019 Elsevier.

5.2. Bacteria

Different types of bacterial species have been reported to play an important role in several biotechnological applications such as food processing, bioremediation, biofuels, genetic engineering, and biomining [111]. They are actively involved in the production of inorganic NPs such as silver, gold, and selenium [112]. Bacteria are good candidates in NPs synthesis due to their abundance in the environment, rapid growth, and ability to survive in extreme conditions. Prokaryotic bacteria and actinomycetes have been extensively used to synthesize AuNPs, either intracellularly or extracellularly. Various enzymes, fatty acids, and sugars present in the bacterial cell can reduce metal ions to their respective NPs. Reports have shown the biosynthesis of AuNPs using different bacteria such as Bacillus subtilis, Escherichia coli, Rhodopseudomonas capsulate, Lactobacillus, Pseudomonas aeruginosa, Bacillus megaterium, and Desulfovibrio desulfuricans [113–115].

Kumari et al. [116] demonstrated the formation of AuNPs of various sizes (2–500 nm) and shapes (spheres, triangles, pentagons, hexagons, and nanosheets) by modulating different physical parameters using Trichoderma viride filtrate. The synthesized NPs were characterized by different techniques, including dynamic light scattering, UV–visible spectroscopy, FTIR, TEM, and X-ray diffraction. Experimental studies have indicated that various parameters such as pH, temperature, time, and culture filtrate concentration play a major role in altering the morphology of NPs.

He et al. [117] reported the extracellular biosynthesis of stable AuNPs using the bacterial Rhodopseudomonas capsulate, which secretes cofactor NADH- and NADHdependent enzymes that induce bioreduction of Au³⁺ to Au⁰ and the subsequent formation of AuNPs. TEM results have demonstrated the formation of spherical AuNPs in the range of 10–20 nm at pH 7. On the contrary, few gold nanoplate structures were formed when the pH was reduced to 4. Interestingly, the results indicate that the pH plays an important role in determining the size and shape of NPs. At low pH, the functional groups (amino, carboxyl, sulfhydryl, etc.) possess more positive charge, and the reducing power of the biomass is weak. This leads to a very slow reaction rate and strong Au-biomass biosorbent, which would possibly result in the formation of nanoplate structures. On the other hand, an increase in the pH increases the reducing power and reaction rate and contributes to the formation of spherical-shaped NPs, which are thermodynamically favorable. Similarly, Sathiyanarayanan et al. [118] reported the extracellular synthesis of AuNPs using the Bacillus megaterium MSBN04. The exopolysaccharide (EPS) produced from this bacterium acts as both reducing and stabilizing agents. TEM and XRD analysis have confirmed the spherical crystalline nature of AuNPs (5–20 nm), which were capped with an EPS layer. Nadaf et al. [119] demonstrated a facile bacteriogenic route for the extracellular synthesis of AuNPs from Bacillus marisflavi YCIS MN 5, which showed their potential in catalytic dye degradation of congo red and methylene blue in the presence of sodium borohydride.

Few bacterial species can survive at toxic metal ion concentrations and extreme temperatures. They have unique defense mechanisms to tolerate such high stress and toxicity of metal ions. Ahmad et al. [120] reported the intracellular synthesis of AuNPs using a novel alkalotolerant actinomycete (Rhodococcus species). When the cells were exposed to chloroauric acid, gold ions were rapidly reduced. TEM analysis of thin sections of actinomycete cells has shown that highly monodispersed NPs in the size range of 5–15 nm were formed on the cytoplasmic membrane. The synthesized AuNPs were not toxic to the cells and the cells continued to grow even after the biogenesis of AuNPs. Llanten et al. [121] reported the biosynthesis of AuNPs using a thermophilic bacterium belonging to the genus Geobacillus, strain ID17, isolated from Antarctica. When exposed to Au^{3+} ions, the bacterial cells turned from colorless into a dark purple color. This bioconversion process is enzymatically mediated by the reductase enzyme and NADH as cofactors. TEM results have shown intracellular accumulation of quasi-hexagonal-shaped AuNPs with sizes ranging from 5–50 nm. Sharma et al. [94] exploited a novel strain of Marinobacter pelagius, which belongs to marine bacteria and can tolerate high salt concentration and can evade toxicity of different metal ions for the production of AuNPs. TEM images have shown the formation of stable and monodisperse AuNPs around 10 nm size upon exposure of the chloroauric acid solution to whole cells. The result indicated that this bacterial strain can synthesize stable, quick, and monodisperse AuNPs around 2–6 nm in size.

Several bacterial strains have the potential of adsorbing/binding metal ions and reducing them into NPs by enzymes produced during metabolic processes in cells, and this property helps to enhance their applications in bioremediation and bioleaching. Nangia et al. [52] identified a new bacterial strain, Stenotrophomonas malophilia (Au-Red02), and isolated it from gold-enriched soil samples, which can synthesize well-dispersed AuNPs of size about 40 nm. FTIR results showed that the synthesized AuNPs were capped with negatively charged phosphate groups, which improves their stability in the aqueous medium. Kunoh et al. [122] reported that the cells of Leptothrix (iron-oxidizing bacteria) released extracellular RNA which has the ability to reduce Au (III) to form spherical AuNPs when treated with an aqueous chloroauric acid solution under ambient conditions.

There are some drawbacks in the usage of bacteria for the synthesis of AuNPs. First, maintaining the bacterial culture is a tedious process. Second, safety measures have to be strictly followed in a clean environment to prevent them from mass contamination. Third, the reduction process is slow and takes time from hours to days. Hence, they are not a preferable choice for the commercial synthesis of AuNPs. Nevertheless, few recent reports have highlighted that the AuNPs synthesized from various bacterial strains exhibited superior properties when compared to the NPs prepared by chemical methods. For instance, Li et al. [123] reported that the AuNPs synthesized from Deinococcus radiodurans showed significant antibacterial activity against both Gram-positive and Gram-negative bacteria. This opens up additional scope for the NPs as an antibacterial agent. Similarly, Shabani et al. [124] reported the enzymatic synthesis of AuNPs (~10 nm) using *Escherichia coli*. The synthesized NPs exhibited strong antifungal properties against various human pathogenic fungi and nontoxicity for Vero and Hep-2 cell lines in vitro at concentrations ranging from 0.31 to 10%.

6. Green Synthesis of Different Types of Nanoparticles

Apart from AuNPs, the green synthesis technique is commonly used to synthesize different types of nanoparticles. For instance, Vinodhini et al. [125] demonstrated the green synthesis of silver nanoparticles (AgNPs) in the size range of 40–57 nm from the leaf extracts of medicinal herbs such as *Tabernaemontana divaricate, Basella alba,* and *Allium fistulosum*. The biosynthesised AgNPs were nontoxic and exhibited antibacterial, antifungal, antioxidant, and antidiabetic properties due to the presence of phenolics and other active components present in these medicinal plants. Similarly, Tyagi et al. [126] reported the biosynthesis of magnetic iron oxide nanoparticles (FeNPs) from spinacia oleracea (spinach) and musa acuminata (banana). Iqbal et al. [127] demonstrated the production of zinc oxide nanoparti-

cles (ZnONPs) using *Elaeagnus angustifolia* leaf extracts, and Gulbagca et al. [128] reported the green synthesis of palladium nanoparticles (PdNPs) using Urtica plant extracts. Among the various types of nanoparticles, AuNPs are highly preferred in biomedical applications as they exhibit unique properties such as surface plasmon resonance, nontoxicity, and high biocompatibility.

7. Biosynthesized AuNPs in Cancer Theranostics: Imaging, Drug Delivery, and Treatment

Cancer is a disease characterized by abnormal and unrestricted cell growth with potential to spread to other parts of the body. The World Health Organization (WHO) reported nearly 10 million deaths in 2020 and this is expected to increase to an estimated 12 million deaths by 2030 [129]. Among the different types of nanomaterials used to treat cancer, biosynthesized AuNPs functionalized with targeting ligands and anticancer drugs are considered as promising candidates in diagnosis and cancer therapy [24,130]. Kim et al. [131] explored this concept to prove the efficacy of AuNPs in treating glioblastoma multiforme (GBM), the most common primary grade 4 brain tumor. They demonstrated that the oral delivery of AuNPs conjugated with milk protein lactoferrin and polyethylene glycol (PEG), a biocompatible polymer, was able to cross the blood-brain barrier and bind with lactoferrin receptors that are highly expressed in the brain tumor cells in mice models. Further, when irradiated with laser light, the administered AuNPs increased the temperature in GBM due to photothermiaproperties and induced a significant reduction in the tumor volume. A schematic illustration for oral absorption of lactoferrin (Lf)–PEGconjugated AuNPs and the mechanism of action of Lf-PEG-AuNP targeting Glioblastoma multiforme (GBM) through lactoferrin receptor pathway of the small intestine, the bloodbrain barrier, and GBM cells is given in Ref. [131].

AuNPs, either individually or in combination with other treatment modalities such as radio/chemotherapy, have the ability to induce hyperthermia or deliver the drug in the targeted region or cell to produce a synergetic effect and thus help to facilitate cancer treatment. For instance, Rezaeian et al. [132] used a green chemistry approach to synthesize curcumin-coated AuNPs and performed in vitro study to the compare nanoparticlemediated photothermal therapy and radiofrequency electric field hyperthermia on mouse colorectal cancer (CT26) cell lines. The results have shown that the NPs induced apoptosis cell death considerably using both photothermal therapy and radiofrequency electric field hyperthermia treatments. Another study used the green chemistry method to synthesize stable AuNPs coupled with 5-Fluorouracil, a chemotherapeutic drug that is widely used for the treatment of liver cancer [133]. This study focused on in vivo toxicity induced by AuNPs in the zebrafish embryo model, in vitro drug release behavior, and efficacy of the NPs in human liver cancer (HepG2) cell lines. In vivo biodistribution analysis indicated that a higher amount of AuNPs accumulated in the liver induced significant cytotoxicity in HepG2 cell lines, which signifies that the AuNPs could be used as a tool for both imaging and targeted drug delivery with minimal side effects of liver cancer. Various reports specifying anticancer activity of AuNPs biosynthesized from different plant sources are mentioned in Table 2.

Plant	Extract Source	Anticancer Type	Size (nm)	Ref.
Jasminum auriculatum	leaf extract	Human cervical cancer cell line (HeLa)	8–37	[134]
Turbinaria decurrens	hydromethanolic extract	MCF-7, HEPG-2, and HCT-116 cell lines	10–19	[135]
Tecoma capensis	leaf extract	Human breast cancer cell line (MCF7)	10–35	[136]
Scutellaria barbata	aqueous extract of plant	Human pancreatic cancer cell lines (PANC-1)	400-1000	[137]
Couroupita guianensis	flower extract	Human leukemia cell line (HL-60)	7–48	[138]
Catharanthus roseus	leaf extract	MCF7 and HepG2 cell line	15–28	[139]
Tabebuia argentiea	flower extract	Liver cancer cell line (HEPG2)	56	[140]
Benincasa hispida	aqueous extract	Human cervical cancer cell line (HeLa)	22.18 ± 2	[141]
Coleous forskohlii	root extract	Liver cancer cell line (HEPG2)	10-30	[142]
Orchid	whole plant extract	Breast cancer cell line (AMJ 13)	14–50	[143]
Hevea brasiliensis	latex extract	Ovarian cell line (CHO-K1)	9	[144]
Antigonon leptopus	leaf extract	Human breast cancer (MCF-7) cells	13–28	[145]
Bauhinia purpurea	leaf extract	Lung carcinoma cell line (A549)	20-100	[146]
Petroselinum crispum	leaf extract	Human colorectal cell line (COLO- 201)	20-80	[147]
Indigofera tinctoria	leaf extract	Lung carcinoma cell line (A549)	6–29	[148]
Lonicera 4 japonica	flower extract	Human cervical cancer cell line (HeLa)	10-40	[149]
Podophyllum hexandrum	leaf extract	Human cervical cancer cell line (HeLa)	5–35	[150]
Gymnema sylvestre	leaf extract	Huma colorectal cancer cell line (HT29)	72.8	[151]
Brazilian Red Propolis	hydroethanolic extract	Bladder (T24), prostate (PC-3) cancer cell line	8–15	[152]
Mentha Longifolia	leaf extract	Breast cancer cell lines (MCF7, Hs 578Bst, Hs 319.T, UACC-3133)	36.4	[153]

Table 2. Green synthesis of AuNPs using different bioagents and their applications.

8. Bioapplications of Medicinal-Plants-Based AuNPs

Medicinal plants, which are widely used in traditional health practices, play a vital role in the treatment of various kinds of human diseases. Their extracts from different plant parts such as leaves, flowers, and roots are used in the development of novel therapeutic drugs [154,155]. It is estimated that above 60% of anticancer drugs that are currently used for cancer treatment are isolated from medicinal plants and almost 3000 medicinal plants worldwide have been reported to have anticancer properties [156,157]. Medicinal plants are rich in various phytoconstituents including polyphenols, flavonoids, glycosides, terpenoids, fatty acids, alkaloids, saponins, and tannins [136,158]. When these materials are used as a source for NPs synthesis, the surface of the NPs is either coated or conjugated with these beneficial compounds. Hence, NPs synthesized using medicinal plant extracts show interesting antibacterial, anti-inflammatory, antidiabetic, and cytotoxic properties [159].

Further, AuNPs synthesized via different green techniques using environmentally friendly reagents exhibit excellent immune response regulation and efficacy in therapy against immune system-associated diseases such as cancer, inflammatory, and autoimmune diseases [160]. However, scientific research on the anticancer properties of medicinal plants is limited. The exact mechanism and the key components of the plant extracts that induce biomedical effects are not well explored. Wang et al. [137] reported the green synthesis of AuNPs using a medicinal plant Scutellaria barbata, which is widely used in the Chinese system of medicine to treat various human ailments. Biosynthesized AuNPs showed effective anticancer activity against human pancreatic cancer cell lines (PANC-1).

Geetha et al. [138] revealed a simple, cost-effective, and one-step process to biosynthesize AuNPs using flower extract of a pharmacologically important tree Couroupita guianensis, commonly known as cannon ball tree, which has innumerable medicinal applications including antibiotic, antiseptic, and anti-inflammatory activity. The anticancer potential of biosynthesized AuNPs was evaluated using a variety of techniques such as MTT assay, DNA fragmentation, apoptosis by 4,6-diamidino-2-phenylindole (DAPI), a fluorescent staining, and comet assay for DNA damage, against human leukemia cell (HL-60) line. Leukemia is a type of blood cancer and a leading cause of cancer-related mortality worldwide [161]. Becerril et al. [162] reported the green synthesis of AuNPs using the aqueous extract from Turnera diffusa, a native desert plant used for traditional medicine in Mexico. This plant has great pharmacological significance including anti-obesity, antioxidant, antibacterial, anti-inflammatory, antidiabetic, antimycotic, and cytotoxic activities. The cytotoxicity and immunomodulatory effects of AuNPs synthesized using this plant extract were investigated on the leukocytes of longfin yellowtail Seriola rivoliana, a marine fish, and antibacterial activity against Vibrio parahaemolyticus and Aeromonas hydrophila, Gram-negative bacteria. The results indicated that the AuNPs increased the phagocytosis activity, attenuated the reactive oxygen species in leukocytes production, and increased the cellular antibacterial mechanism mediated by nitric oxide production.

Yasmin et al. [163] employed green techniques for rapid synthesis of spherical AuNPs in the size range of 16–30 nm using Hibiscus rosasinensis, a medicinal plant that has a lot of beneficial applications such as anti-infectious, anthelmintic, anti-inflammatory, diuretic, and antipyretic properties. The synthesized particles were found to be stable for up to a few months and have the potential to be used for medical and biosensor applications. A recent study reported the synthesis of AuNPs using the aqueous Mentha longifolia leaf extract. The major constituents of this plant include polyphenols, alkaloids, organic acids, terpenoids, etc., and this plant has been used as an antihypertensive and antitussive drug in traditional medicine. The biosynthesized nanoparticles were found to be effective against various breast cancer cell lines, such as breast adenocarcinoma (MCF7), breast carcinoma (Hs 578Bst), breast infiltrating ductal cell carcinoma (Hs 319.T), and breast infiltrating lobular carcinoma (UACC-3133) cell lines, without causing cytotoxicity against a normal cell line (HUVEC). Therefore, AuNPs synthesized using Mentha longifolia leaf aqueous extract can be tested as an anti-breast cancer drug in humans in the near future [153]. Another report described the biogenic synthesis of AuNPs using the Jasminum auriculatum leaf extract. The leaves of this plant have numerous medicinal applications such as antilithiatic, wound healing activity, and diuretic activity. They are used in the treatment of leprosy, skin diseases, ulcers, and wounds. The MTT assay performed using these NPs against human cervical cancer (HeLa) cells revealed significant anticancer activity without the requirement of doping additional molecules [134].

Hasan et al. [135] biosynthesized AuNPs from Turbinaria decurrens, which is an Egyptian marine brown macroalga, which has a diverse group of phytochemicals with unique bioactivities and is widely used as food and medicine. The authors compared the chemical composition and antioxidant and anticancer activities of both the hydromethanolic extract (HME) of this plant and the HME–AuNPs on three different cancer cell lines (MCF-7, HEPG-2, and HCT-116) using MTT assay. The results showed the strong anticancer activity of AuNPs against all the three studied cell lines. Their findings indicated that the biosynthesized AuNPs could be used as a source for the discovery of novel therapeutic agents in the biomedical field to treat oxidative stress-related diseases, particularly cancer. Hosny et al. [136] explained a phytofabrication technique to synthesize AuNPs that remained stable for up to three months using the aqueous leaf extract of Tecoma capensis, a flowering plant commonly found in tropical and subtropical areas of Africa. The anticancer efficacy of T. capensis–AuNPs was tested against human breast cancer cell line (MCF7) using MTT assay and the results revealed the excellent potency of AuNPs in preventing the development and proliferation of MCF7 cells. All these studies explain that the synthesis of AuNPs based on medicinal plants is beneficial to express their health and medical benefits along with their multifunctional potential in treating different diseases, including cancer.

9. Challenges and Future Prospects

In the recent decade, the green synthesis approach has been successfully used to synthesize a variety of NPs, including AuNPs, with varying morphology and properties. Numerous research articles have been published worldwide on the interdisciplinary routes in the synthesis of NPs from different plants and microorganisms. However, there are several limitations and drawbacks in the green synthesis method which limit their largescale production for commercial purposes and diminish their subsequent applications in biotechnology and nanomedicine. The current challenges are summarized below:

- Lack of sufficient data to control the size and shape of NPs: Many reports have demonstrated the formation of AuNPs with different morphologies during the synthesis process. The formation of NPs with uniform morphology and narrow size distribution is essential for pharmacological and biological applications. Proper knowledge to optimize the reaction mixture, experimental time, pH, temperature, rotational speed, concentrations of chloroauric acid concentrations, etc., have to be taken into care to solve this problem.
- Lack of standardized protocols to reproduce NPs with the same characteristics: NPs from different parts even of the same plant have variations in their structure and properties, as, e.g., AuNPs synthesized from different plant parts exhibit different levels of cytotoxicity due to the difference in the antioxidant/metabolite contents.
- Lack of clear knowledge on the mechanism of NPs synthesis is a major drawback, and identification of key components present in different metabolites from plants and microbial sources that play an active role in the synthesis of NPs are challenging.
- Separation and purification of NPs from the complex reaction mixture is another important aspect that still remains a hurdle.
- Scalability of synthesized NPs from a laboratory approach to meet the huge demands in the industrial and pharmaceutical scale is a major concern.
- A detailed toxicological study of the biosynthesized NPs is crucial to enhance their scope in diverse fields.
- Technical barriers and regulatory policies for the commercial synthesis of NPs limit the scale-up process, which needs to be overcome.
- Stability and functionalization: More research on green synthesis is required to surfacemodify the synthesized NPs to improve their stability in biological media and functionalize them with specific antibodies or peptides to improve their applications in drug delivery and cancer therapy.

Even though the green synthesis approach is very popular, overcoming the above challenges could improve the global acceptability and adaptability of the commercial synthesis of AuNPs. Future research and development in this sector should be directed towards overcoming the present hurdles and coming up with novel standardized protocols, designing smart and safe AuNPs functionalized with different biomolecules, and targeting moieties for multifunctional applications.

10. Conclusions

Due to their unique characteristics, AuNPs have enormous applications in various fields such as electronics, catalysis, optics, sensors, and biology. Though different physicochemical methods are used in the synthesis of NPs, the green synthesis method is a promising approach to produce different types of NPs, including gold nanomaterials, in a simple, eco-friendly, and cost-effective manner. This method has several advantages over the conventional physical and chemical methods used for NPs synthesis, such as safety, and does not involve the use of hazardous chemicals or the addition of external harmful substances during the synthesis/durability of NPs. Different parts of plants and a variety of microbes, including bacteria, fungi, algae, and yeast, are used as a natural source in the biosynthesis process. Further, NPs synthesized using biological extracts have several beneficial properties, such as high anticancer, antimicrobial, anti-inflammatory, antioxidant, and catalytic activity, etc., which find exciting applications in nanomedicine.

This review article focuses on the "state-of-the-art" research on the "green synthesis" of AuNPs, different sources of green materials, with special emphasis on biosynthesis of NPs from different parts of the plant, intracellular and extracellular synthesis from microbes, and their overall applications. Further, the advantage of the green synthesis method over the conventional chemical synthesis methods, current challenges in this field, and the prospects are discussed in detail. Green synthesis is an emerging field and current research on the biogenesis of AuNPs, characterization, and functionalization of the synthesized particles are still in the developing phase. A thorough understanding of the basic principles of green

chemistry and more research work is required to gain sufficient knowledge in this field. Addressing the current challenges in this field and overcoming them with standardized protocols and innovative techniques can revolutionize the synthesis of AuNPs on both laboratory and commercial scales.

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