

# Electrodeposition of Cu-SWCNT Composites

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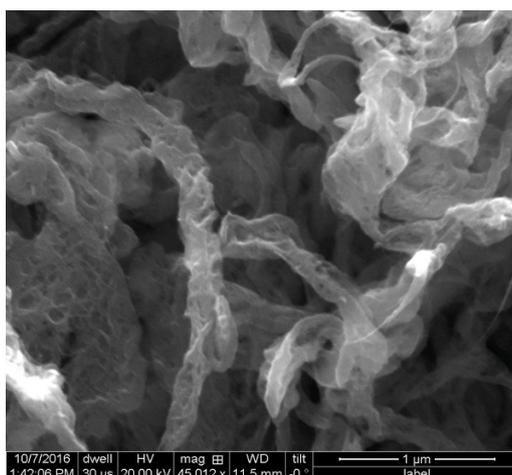
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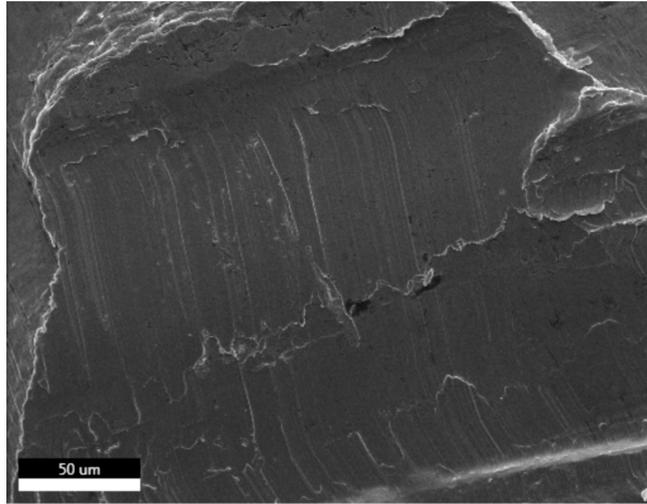
## Supplementary Materials



**Figure S1.** Image showing the association of the aggregates with the magnetic stir bar indicating the presence of the SWCNTs containing iron catalyst impurities.



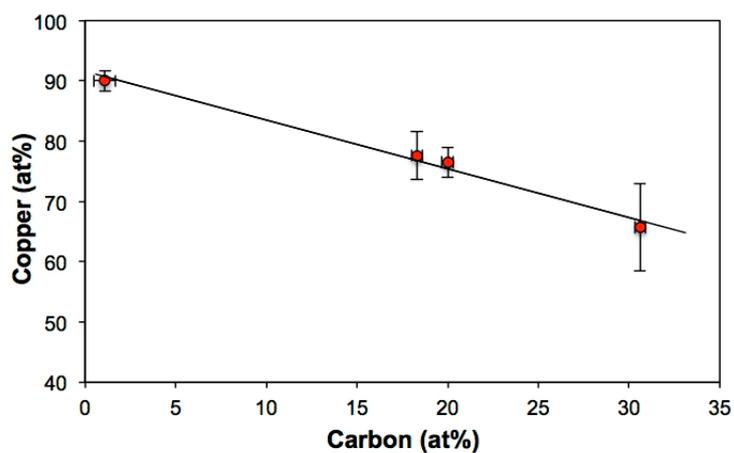
**Figure S2.** SEM image of HiPCO SWCNTs.



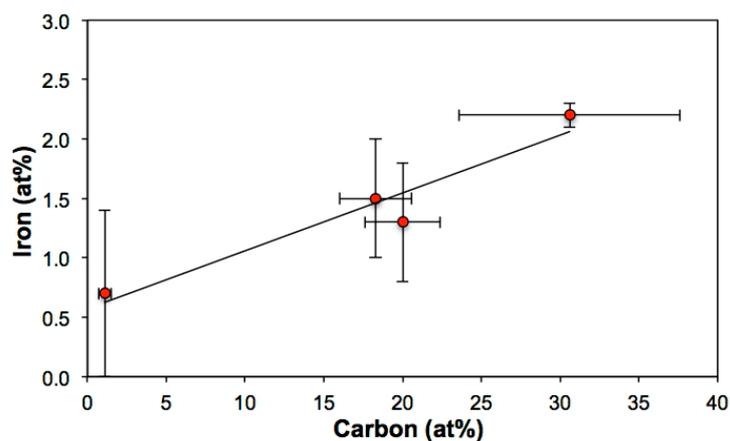
**Figure S3.** SEM of copper wire cathode showing scratches likely produced during manufacture and subsequent handling.



**Figure S4.** Photographic image of Cu-SWCNT composite deposited on the surface of the copper cathode after electrolysis using constant voltage (14 V) power supply:  $[\text{Cu}^{2+}] = \sim 0.095 \text{ M}$ ,  $[\text{SWCNT}] = 2.3 \text{ wt\%}$ .



**Figure S5.** Plot of copper (at%) versus carbon (at%) composition as determined by EDX analysis ( $R^2 = 0.98$ ). EDX shown as an average of 5 independent measurements per sample.



**Figure S6.** Plot of iron (at%) versus carbon (at%) composition as determined by EDX analysis ( $R^2 = 0.92$ ). EDX shown as an average of 5 independent measurements per sample.



**Figure S7.** Copper wire cathodes with hard coatings (Cu-SWCNT) before rinsing, and drying.