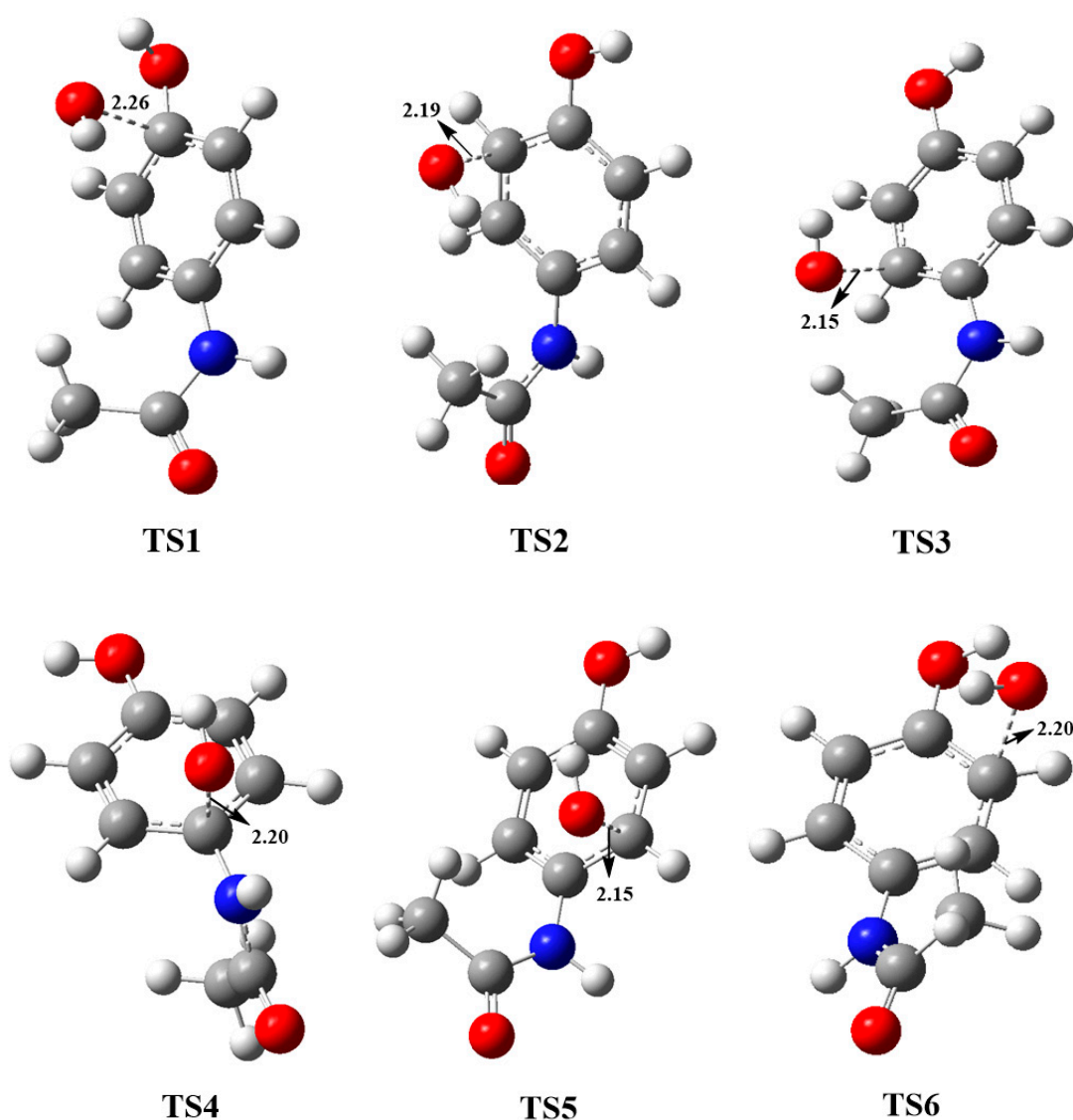
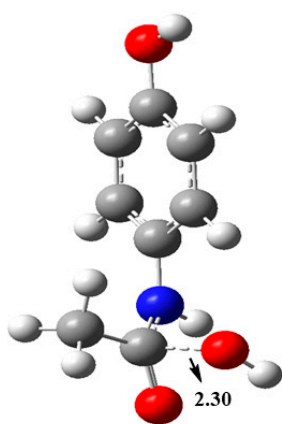


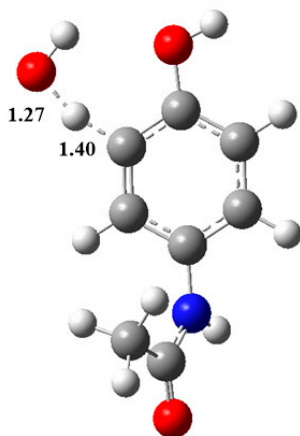
Supplementary Materials: Theoretical Calculation on the Reaction Mechanisms, Kinetics and Toxicity of Acetaminophen Degradation Initiated by Hydroxyl and Sulfate Radicals in the Aqueous Phase

Mengmeng Xu, Junfang Yao, Simei Sun, Suding Yan and Jingyu Sun

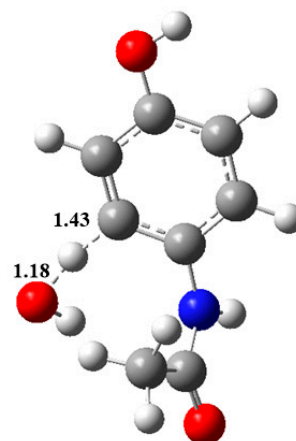




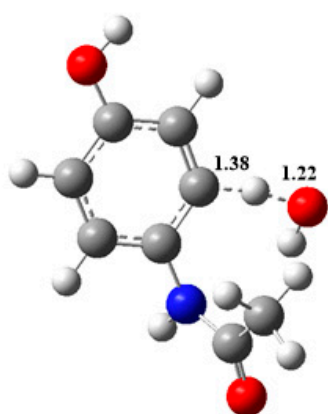
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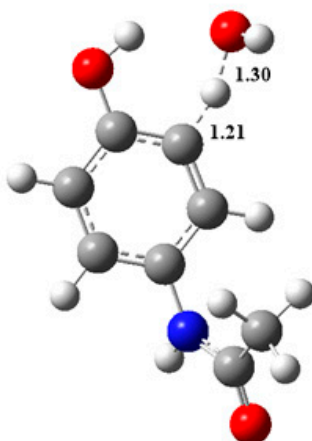
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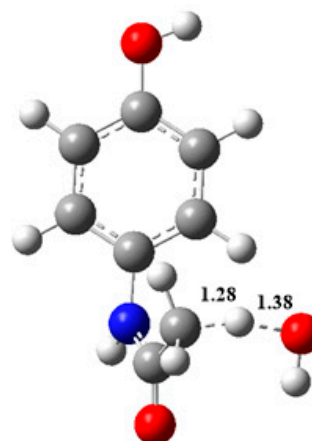
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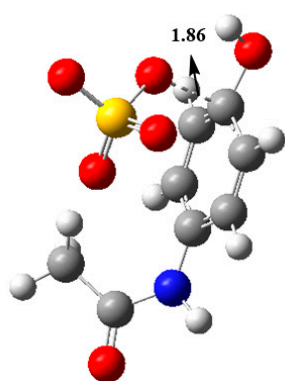
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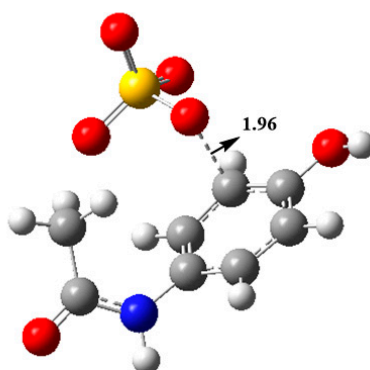
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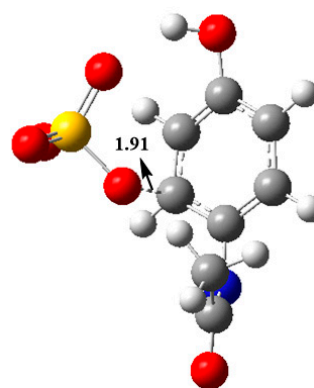
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TS8



TS9



TS10

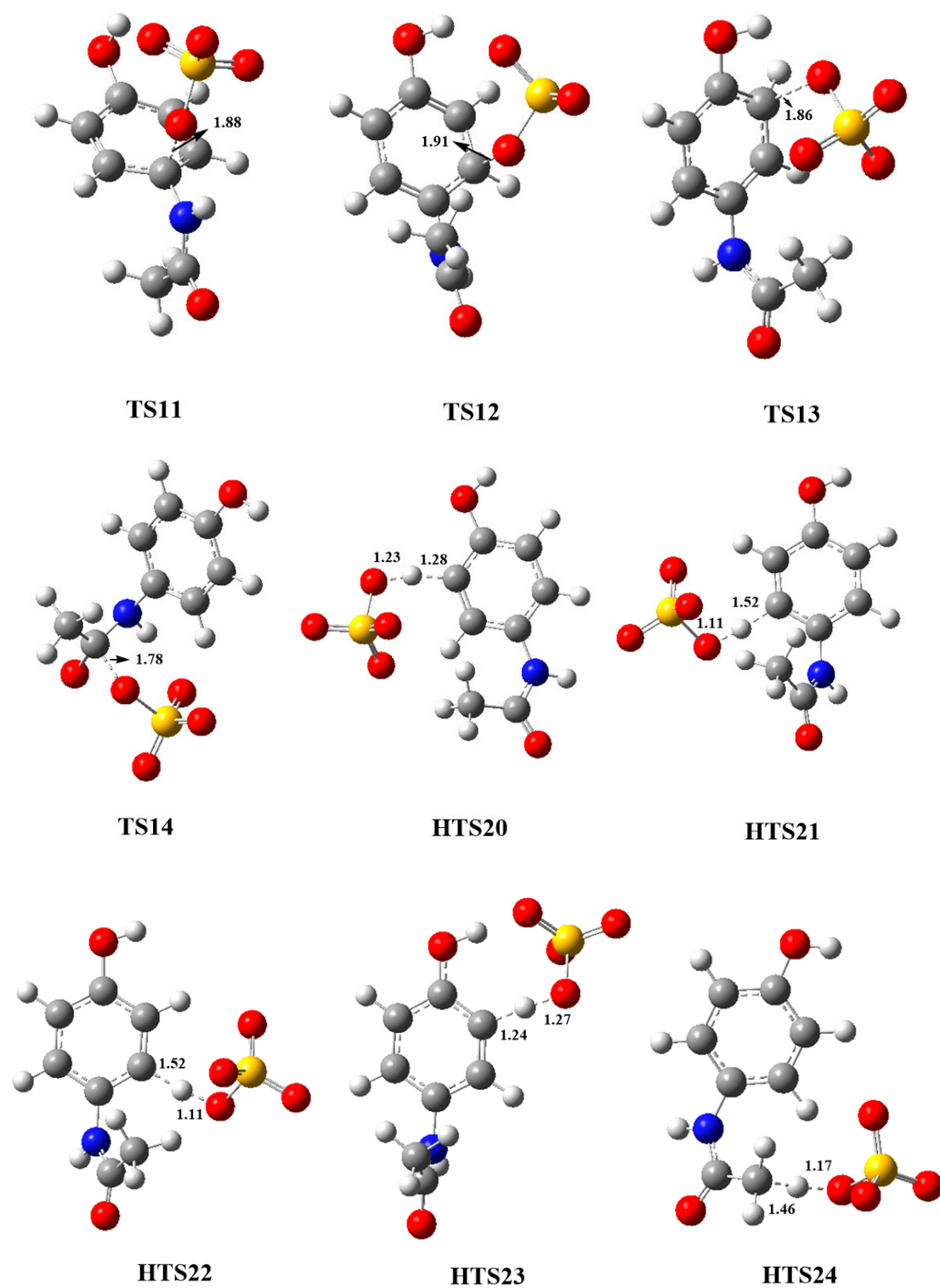


Figure S1. Optimized geometries involving the transition states of AAP with $\bullet\text{OH}$ and $\text{SO}_4^{\bullet-}$ at the M06-2X/6-31+G(d,p) level. Here, H (white), C (grey), O (red), S (yellow), N (blue).

Table S1. Calculated rate constants of AAP with •OH from 198 to 338 K and 1 atm.

Paths	198	218	238	258	278	298	318	338
Path1	8.04E+07	9.44E+07	8.82E+07	8.42E+07	8.17E+07	8.04E+07	7.99E+07	8.00E+07
Path2	1.87E+08	2.49E+08	2.24E+08	2.07E+08	1.95E+08	1.87E+08	1.82E+08	1.79E+08
Path3	1.51E+07	1.25E+07	1.30E+07	1.36E+07	1.43E+07	1.51E+07	1.59E+07	1.68E+07
Path4	3.22E+08	5.08E+08	4.32E+08	3.81E+08	3.46E+08	3.22E+08	3.04E+08	2.92E+08
Path5	3.33E+07	3.51E+07	3.40E+07	3.34E+07	3.32E+07	3.33E+07	3.36E+07	3.41E+07
Path6	9.02E+09	7.84E+09	6.01E+09	4.86E+09	4.10E+09	3.56E+09	3.18E+09	2.90E+09
Path7	6.75E+02	2.04E+01	6.00E+01	1.51E+02	3.36E+02	6.75E+02	1.25E+03	2.17E+03
Path15	7.36E-04	7.95E-03	5.80E-02	3.14E-01	1.34E+00	4.75E+00	1.44E+01	3.88E+01
Path16	4.40E-03	3.55E-02	2.03E-01	8.95E-01	3.21E+00	9.75E+00	2.60E+01	6.20E+01
Path17	4.51E-04	4.36E-03	2.89E-02	1.44E-01	5.74E-01	1.91E+00	5.51E+00	1.41E+01
Path18	1.28E+02	4.16E+02	1.11E+03	2.57E+03	5.30E+03	1.15E+04	1.75E+04	2.89E+04
Path19	9.61E+01	3.51E+02	1.03E+03	2.59E+03	5.73E+03	1.13E+04	2.11E+04	3.65E+04
<i>k</i> _{total-aq}	9.65E+09	8.74E+09	6.80E+09	5.58E+09	4.77E+09	4.20E+09	3.80E+09	3.51E+09

Table S2. Calculated rate constants of AAP with SO₄^{•−} from 198 to 338 K and 1 atm.

Paths	198	218	238	258	278	298	318	338
Path8	9.43E+16	6.50E+15	7.13E+14	1.12E+14	2.32E+13	6.00E+12	1.86E+12	6.67E+11
Path9	2.08E+14	2.85E+13	5.53E+12	1.40E+12	4.37E+11	1.61E+11	6.78E+10	3.19E+10
Path10	1.76E+14	2.87E+13	6.44E+12	1.85E+12	6.43E+11	2.60E+11	1.19E+11	6.04E+10
Path11	2.14E+13	3.50E+12	7.92E+11	2.29E+11	8.03E+10	3.28E+10	1.51E+10	7.74E+09
Path12	1.70E+14	2.78E+13	6.24E+12	1.79E+12	6.23E+11	2.52E+11	1.16E+11	5.85E+10
Path13	5.65E+18	2.66E+17	2.12E+16	2.53E+15	4.13E+14	8.65E+13	2.22E+13	6.76E+12
Path14	1.46E-11	3.47E-10	4.91E-09	4.67E-08	3.26E-07	1.77E-06	7.86E-06	2.95E-05
Path20	3.98E-02	1.93E-01	7.25E-01	2.25E+00	6.02E+00	1.43E+01	3.07E+01	6.10E+01
Path21	1.18E+00	4.02E+00	1.12E+01	2.69E+01	5.75E+01	1.11E+02	2.00E+02	3.37E+02
Path22	2.81E+00	8.72E+00	2.25E+01	5.06E+01	1.02E+02	1.88E+02	3.23E+02	5.24E+02
Path23	1.70E-02	8.03E-02	2.96E-01	9.03E-01	2.38E+00	5.55E+00	1.18E+01	2.31E+01
Path24	4.59E+09	1.71E+09	7.59E+08	3.84E+08	2.16E+08	1.33E+08	8.70E+07	6.05E+07
<i>k</i> _{total-aq}	5.74E+18	2.73E+17	2.19E+16	2.65E+15	4.38E+14	9.32E+13	2.44E+13	7.59E+12

Table S3. The acute and chronic toxicity class (mg L^{−1}).

Classification	Acute toxicity ¹	Chronic toxicity ²
Not harmful	LC ₅₀ >100 or EC ₅₀ >100	ChV>10
Harmful	10< LC ₅₀ <100 or 10< EC ₅₀ <100	1< ChV <10
Toxic	1< LC ₅₀ <10 or 1< EC ₅₀ <10	0.1< ChV <1
Very toxic	LC ₅₀ <1 or EC ₅₀ <1	ChV<0.1

¹Criteria set by the European Union (described in Annex VI of Directive 67/548/EEC);²Criteria set by the Chinese hazard evaluation guidelines for new chemical substances (HJ/T 154–2004).**Table S4.** Eco-toxicity values of AAP and its transformation intermediates to aquatic organisms (mg L^{−1}).

		AAP	IM1	IM13	IM8	IM6
Acute Toxicity	Fish (LC ₅₀)	323	7.08	186	120	21.4
	Daphnia (LC ₅₀)	63.1	21.4	933	562	77.6
	Green algae (EC ₅₀)	26.3	3.47	0.73	34.7	64.6
Chronic Toxicity (ChV)	Fish	26.3	5.50	6.2×10 ³	2.0×10 ³	64.6
	Daphnia	5.13	1.95	602	269	14.8
	Green algae	37.2	5.50	692	347	29.5