

Article

Strategies for Effective Management of Indoor Air Quality in a Kindergarten: CO₂ and Fine Particulate Matter Concentrations

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Abstract: The educational and play-related activities of children proceed mainly indoors in a kindergarten. High concentrations of indoor PM_{2.5} and CO₂ have been linked to various harmful effects on children, considerably impacting their educational outcomes in kindergarten. In this study, we explore different scenarios involving the operation of mechanical ventilation systems and air purifiers in kindergartens. Using numerical models to analyze indoor CO₂ and PM_{2.5} concentration, we aim to optimize strategies that effectively reduce these harmful pollutants. We found that the amount of ventilation required to maintain good air quality, per child, was approximately 20.4 m³/h. However, we also found that as the amount of ventilation increased, so did the concentration of indoor PM_{2.5}; we found that this issue can be resolved using a high-grade filter (i.e., a MERV 13 grade filter with a collection efficiency of 75%). This study provides a scientific basis for reducing PM_{2.5} concentrations in kindergartens, while keeping CO₂ levels low.

Keywords: kindergarten; CO₂; PM_{2.5}; mechanical ventilation; indoor air quality; numerical model



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1. Introduction

Indoor air quality management is essential, especially since people spend more than 80% of their lives indoors [1,2]. According to a report from the World Health Organization (WHO), indoor PM_{2.5} (fine particulate matter) was responsible for approximately 2.3 million deaths in 2020 [3]. In particular, children exposed to high PM_{2.5} levels have been shown to manifest a variety of negative health-related effects (e.g., pneumonia, high blood pressure, and heart disease) [4–6]. The respiration rate in children is relatively higher than that in adults when considering intake per unit body weight. Consequently, children are more vulnerable to the adverse effects of airborne toxic substances [6].

Indoor air pollutants can pose significant health risks to humans. Formaldehyde, emitted from adhesives in furniture and wallpapers, is a well-known indoor toxic gas that can cause symptoms associated with sick building syndrome [7]. Combustion gases released from gas stoves contain harmful substances such as carbon monoxide and nitrogen dioxide. Carbon monoxide can interfere with oxygen transport, leading to poisoning and even death [8]. Nitrogen dioxide can cause respiratory diseases [9]. Hydrogen sulfide is highly toxic and can cause irritation to the skin [10]. One of the prominent indoor toxic gases, radon, is prone to emanating from sources such as latex mattresses and furniture. Prolonged exposure to radon can lead to lung cancer and respiratory diseases [11]. Various studies have substantiated the harmful effects of these indoor toxic gases, and ventilation is primarily advocated as a solution [12,13]. However, managing such a diverse array of

hazardous gases individually poses a challenge. Hence, CO₂, which is relatively easy to measure, has been employed as an indicator of indoor toxic gases [14].

However, increasing ventilation rates indiscriminately to mitigate indoor toxic gases is not advisable. This is because higher ventilation rates can lead to an influx of external PM_{2.5} particles, causing an increase in indoor PM_{2.5} concentrations. Moreover, excessive ventilation can result in the reduced effectiveness of air purifiers, making it challenging to manage indoor PM_{2.5} concentrations during ventilation [15]. Therefore, it is necessary to develop strategies that simultaneously consider CO₂ and PM_{2.5} concentrations when managing indoor air quality. The Ministry of Environment in Korea has stipulated 1000 ppm of CO₂ as the standard to maintain good air quality in kindergartens [16], whereas the WHO recommends that average indoor PM_{2.5} concentrations should be less than 15 µg/m³; per day [3].

In four kindergartens surveyed in Korea, their indoor CO₂ concentrations exceeded 1000 ppm, respectively [17]. This sample of kindergartens highlights the necessity of determining the indoor ventilation rate that can maintain indoor CO₂ below 1000 ppm. In addition, previous studies have shown that indoor PM_{2.5} concentrations, on average, exceed 15 µg/m³; daily in various indoor spaces (e.g., classrooms and offices) [18,19].

Han et al. (2022) investigated indoor PM_{2.5} variations resulting from air purifier usage in an elementary school [19]. Noh and Yook (2016) used a numerical model to determine how to reduce indoor 0.3 and 3 µm particle concentrations using mechanical ventilation filters and air purifiers [20]. Nevertheless, these studies failed to provide a viable solution for reducing indoor CO₂ concentrations.

Peter et al. (2000) emphasized the importance of adequate ventilation in reducing indoor CO₂ concentration in middle schools [21]. Similarly, Franco et al. (2020) explored optimal ventilation methods while focusing on CO₂ concentration in university classrooms [22]. However, both studies did not account for indoor particulate matter. It is essential to note that increasing ventilation rates may inadvertently lead to higher concentrations of outdoor particulate matter infiltration.

Pacitto et al. (2020) studied the changes in indoor CO₂ and particulate matter in four different scenarios with or without window opening and air purifier operation [23]. In this study, only the measured data and the required ventilation rates were presented; there are no results regarding the expected concentration changes based on the provided ventilation rates.

Children are particularly vulnerable to particulate matter; thus, effective indoor air quality management in kindergartens is crucial. Nevertheless, there is a limited body of research addressing indoor air quality improvement specifically in kindergarten settings. Park et al. (2017) only measured the air quality (CO₂, PM₁₀) in kindergartens in Korea [17]. Nicolas et al. (2013) conducted a study measuring formaldehyde and benzene levels in a French kindergarten [24]. However, these studies did not suggest management methods for indoor air quality in kindergartens.

This study presents methods for simultaneously managing indoor CO₂ and PM_{2.5} concentrations in a kindergarten using numerical analysis modeling established through measurements. CO₂ and PM_{2.5} concentrations were measured throughout the day in the kindergarten and equations were developed to model the concentration changes over time within the test space. The accuracy of the equations was verified by comparing the measured concentrations with the concentrations obtained by numerical model. Overall, through the use of numerical modeling and experimental measurements, the appropriate mechanical ventilation rate to maintain CO₂ levels below 1000 ppm was suggested. In addition, to maintain low PM_{2.5} level under such a ventilation rate, the application of a high-grade filter in a mechanical ventilation system and a high-capacity air purifier were analyzed.

2. Materials and Methods

2.1. Experimental Setup

This study was conducted from January to March 2023, which is the winter season in Korea. The outdoor $\text{PM}_{2.5}$ concentrations in winter were higher than those in other seasons [25], potentially resulting in increased indoor $\text{PM}_{2.5}$ concentrations due to the inflow of outdoor $\text{PM}_{2.5}$ [26].

Figure 1 shows the floor plan of the experimental setup in a kindergarten in Daejeon, South Korea. The test was conducted in two classrooms. Both classrooms have the same area of 57.42 m^2 . In Classroom 1, which contained four windows, two air purifiers were arranged and the mechanical ventilation diffusers were equipped with two air supplies and two exhaust outlets. In Classroom 2, which featured three windows, a single air purifier was placed, as was a mechanical ventilation system similar to that in Classroom 1.

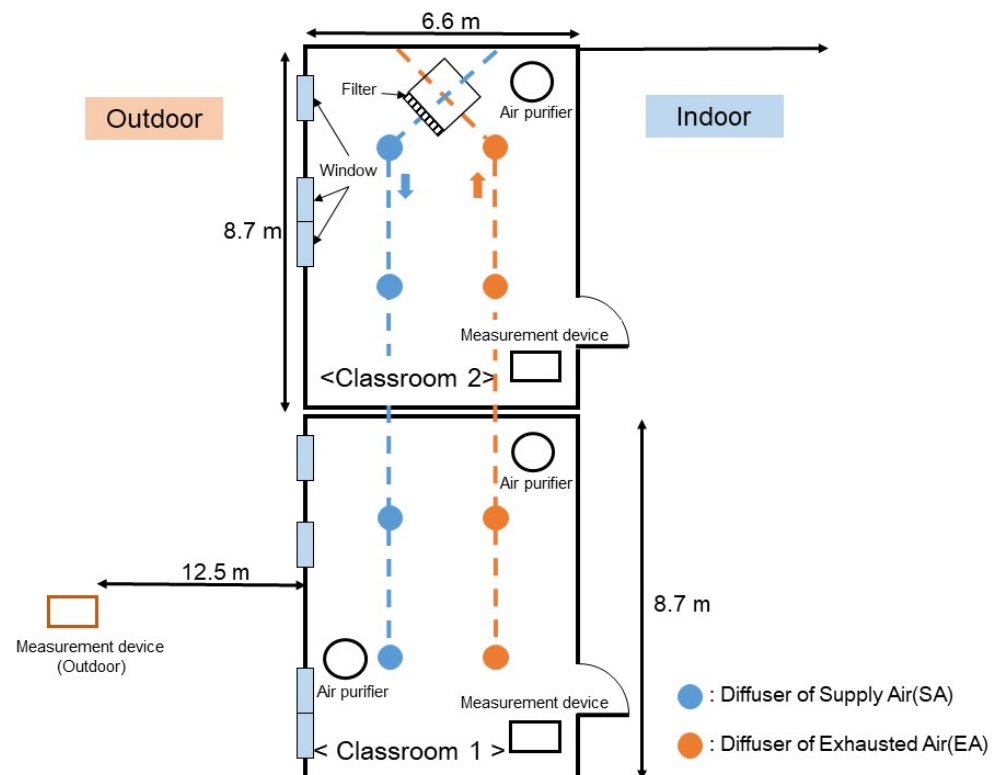


Figure 1. Schematic of the experimental setup: a floor plan view of the arrangement.

To prevent children from accidentally handling a device or breaking it, the measurement devices were installed on shelves at a height of 1.5 m. In practice, the doors remain closed during class; therefore, measurement errors caused by the doors are nearly negligible. The CO_2 concentrations were measured using an NDIR-type sensor (GTH53, EYC-tech, New Taipei City, Taiwan). The measuring range and linear accuracy of the CO_2 sensor is from 0 to 5000 ppm and ± 50 ppm, respectively. For measuring $\text{PM}_{2.5}$ concentrations, an optical particle counter (OPC, 1.109, Grimm Aerosol Technik Co., Ainring, Germany) was used. The OPC uses the light scattering method to measure the particle concentration. The measurable size range of OPC is from 0.25 to $32 \mu\text{m}$, and the sampling flow rate is 1.2 L/min. All the measurement devices used in this test were calibrated. Measurements were performed throughout the day in the kindergarten. Measurement devices were installed in Classrooms 1, 2, and outside, recording CO_2 and $\text{PM}_{2.5}$ concentration data at 1 min intervals. Additionally, the power consumption of ventilators and air purifiers were monitored to ensure their proper operation. The obtained parameters were utilized in the numerical analysis equation, and the measured data were then compared with the results from the numerical analysis.

2.2. Numerical Model

Figure 2 shows a schematic of the factors influencing indoor CO₂ concentrations and indoor PM_{2.5}. Based on this schematic, a mass balance equation was formulated to describe the variations in indoor CO₂ and PM_{2.5} concentrations.

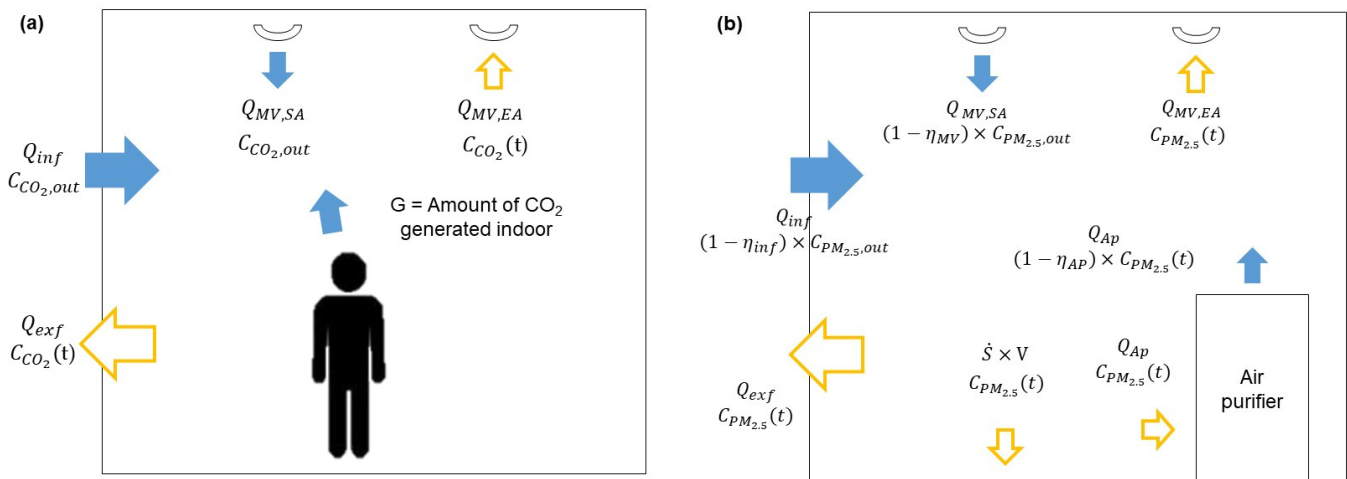


Figure 2. Diagrams of indoor (a) CO₂ model and (b) PM_{2.5} model.

Figure 2a presents a diagram of the factors that affect the indoor CO₂ concentration. These factors can be categorized into natural ventilation, mechanical ventilation, and human respiration. Here, natural ventilation is a method of supplying air to the classroom through the wall by opening the windows. Mechanical ventilation is a method of supplying the air through the mechanical ventilation system. In this study, balanced ventilation was used. Here, $C_{CO_2,out}$ represents the outdoor CO₂ concentration (ppm), and C_{CO_2} represents the indoor CO₂ concentration (ppm). Q_{inf} represents the air inflow through the walls and windows (m³/h), and Q_{exf} represents the air outflow through the walls and windows (m³/h). $Q_{MV,SA}$ represents the air inflow through the mechanical ventilation system (m³/h), and $Q_{MV,EA}$ represents the air outflow through the mechanical ventilation system (m³/h). V represents the volume of the indoor area (m³).

The parameter G represents the amount of CO₂ generated through respiration (ppm × m³/h) and was determined using Equation (1). In this equation, “Man”, “Woman”, and “Child” denote the number of adult men, adult women, and children indoors, respectively. Based on a study conducted by Cho et al. [27], the CO₂ generation rate through an adult man’s respiration is 305.3 ppm × m³/min, while it is 264.2 ppm × m³/min for adult women. Additionally, the study found that children generate approximately 73% of the CO₂ produced by adults [27]. This study assumes that each child produces the same amount of generated CO₂. Hence, in this study, the CO₂ generation rate in the children was determined as 268.9 × 0.73 ppm × m³/min. The indoor CO₂ concentration over time is given by Equation (2).

$$G = [\text{Man} \times 305.3 + \text{Woman} \times 264.2 + \text{Child} \times 268.9 \times 0.73] \quad (1)$$

$$C_{CO_2}(t) = \frac{C_{CO_2,out}(Q_{inf} + Q_{MV,SA}) + G}{Q_{exf} + Q_{MV,EA}} + \left[C_{CO_2}(0) - \frac{C_{CO_2,out}(Q_{inf} + Q_{MV,SA}) + G}{Q_{exf} + Q_{MV,EA}} \right] \times \exp^{-\frac{Q_{MV,SA} + Q_{inf}}{V} t} \quad (2)$$

Figure 2b shows the factors affecting indoor PM_{2.5} concentration. These factors can be categorized into natural ventilation, mechanical ventilation, air purifiers, and deposition. $C_{PM_{2.5},out}$ represents the outdoor PM_{2.5} concentration (μg/m³), while $C_{PM_{2.5}}$ signifies the indoor PM_{2.5} concentration (μg/m³). η_{inf} represents the collection efficiency through natural ventilation (–), η_{MV} represents the collection efficiency of the mechanical ventilation

system filter (–). η_{AP} represents the collection efficiency of the filter installed in an air purifier (–) and Q_{AP} represents the flowrate of an air purifier (m^3/h). So, $\eta_{AP}Q_{AP}$ represents the stated clean air delivery rate (CADR) of an air purifier (m^3/h). ϵ is the air mixing factor of an air purifier in a space [28]. $\dot{S}$ represents the deposition rate (h^{-1}). The indoor $\text{PM}_{2.5}$ concentration according to time is shown in Equation (3).

$$C_{PM_{2.5}}(t) = C_{PM_{2.5},out} \frac{(1 - \eta_{inf})Q_{inf} + (1 - \eta_{MV})Q_{MV,SA}}{Q_{exf} + Q_{MV,EA} + \dot{S}V + \epsilon\eta_{AP}Q_{AP}} + \left[C_{PM_{2.5}}(0) - C_{PM_{2.5},out} \frac{(1 - \eta_{inf})Q_{inf} + (1 - \eta_{MV})Q_{MV,SA}}{Q_{exf} + Q_{MV,EA} + \dot{S}V + \epsilon\eta_{AP}Q_{AP}} \right] \times \exp^{-\frac{Q_{inf} + Q_{MV,SA} + \dot{S}V + \epsilon\eta_{AP}Q_{AP}}{V} t} \quad (3)$$

2.3. Defining the Parameters

The parameters essential for calculating the numerical model from Equations (2) and (3) were applied as follows: The values of $C_{CO_2,out}$ and $C_{PM_{2.5},out}$ were derived from the measured data obtained from the measurement device placed outdoors. The value of V , which was the same in Classrooms 1 and 2, was determined to be 143.55 m^3 . G was determined by the number of occupants in the indoor space and substituting these values into Equation (1). $\dot{S}$ was set to 0.05 h^{-1} based on prior research [15,29].

As shown in Figure 3, the indoor CO_2 concentration was measured over time to determine the value of Q_{inf} . The measurement for Q_{inf} was conducted on 7 November 2022, with an outdoor wind speed of 1.7 m/s and a temperature difference of 21°C between the indoor and outdoor environments. As shown in Figure 3a, the initial indoor CO_2 concentration was set to 6000 ppm with all windows and doorways closed. Based on the results shown in Figure 3a, Q_{inf} was calculated to be $26 \text{ m}^3/\text{h}$ for Classrooms 1 and $16 \text{ m}^3/\text{h}$ for Classrooms 2. The higher Q_{inf} value for Classroom 1 compared to Classroom 2 can be attributed to the greater number of windows in Classroom 1, leading to potentially lower airtightness [30]. As shown in Figure 3b, the indoor CO_2 concentration was measured for all open windows. According to the data in Figure 3b, the calculated Q_{inf} values were $571 \text{ m}^3/\text{h}$ for Classrooms 1 and $207 \text{ m}^3/\text{h}$ for Classrooms 2. The study conducted by Cho et al. (2012) indicated that the natural ventilation rate tended to be proportional to the opening area [30]. As the area of the openings per window remained constant, Q_{inf} for Classroom 1 was determined to be $143 \text{ m}^3/\text{h}$ per open window, and for Classroom 2, it was determined to be $69 \text{ m}^3/\text{h}$ per opened window.

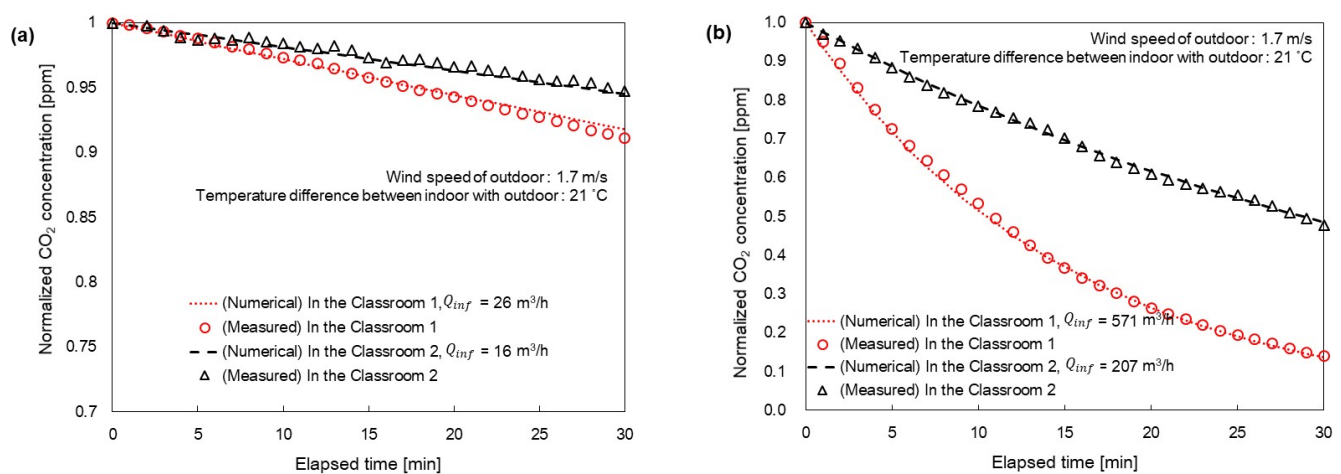


Figure 3. Measuring natural ventilation flow rate indoors when all windows are (a) closed and (b) opened. [Classroom 1, Classroom 2].

$Q_{MV,SA}$ were determined through airflow measurements in a diffuser using a flowmeter (Model 6750, KANOMAX, Osaka, Japan). The measurable flowrate is from 8 to 600 m³/h and the accuracy is $\pm 3\%$ from 8 to 350 m³/h, and $\pm 5\%$ from 350 to 600 m³/h. The mechanical ventilation system can be operated at two different flow rates, Levels 1 and 2. The airflow rates at each level were 142 m³/h and 230 m³/h in Classroom 1 and 157 m³/h and 254 m³/h in Classroom 2. The value of η_{MV} is 0.35, which has a filter grade of MERV 11. To satisfy the continuity equation, it was assumed that the total inflow into the classroom and total outflow were equivalent. This assumption is expressed through Equation (4).

$$Q_{inf} + Q_{MV,SA} = Q_{exf} + Q_{MV,EA} \quad (4)$$

Figure 4 shows the variation of indoor PM_{2.5} concentrations over time, with all windows and doorways closed. This approach allowed us to determine the in/out ratios. As time, t approaches infinity in Equation (3), the term $\exp\left(-\frac{Q_{inf} + Q_{MV,SA} + \dot{SV} + \varepsilon\eta_{Ap}Q_{Ap}}{V}t\right)$ becomes negligible, yielding the expression provided in Equation (5). This outcome enables us to derive the value of η_{inf} as mentioned in Equations (3) and (5). Through this analysis, the values of η_{inf} for Classroom 1 and Classroom 2 were determined as 0.404 and 0.562, respectively.

$$\frac{C_{PM_{2.5}}}{C_{PM_{2.5,out}}} = \frac{(1 - \eta_{inf})Q_{inf} + (1 - \eta_{MV})Q_{MV,SA}}{Q_{exf} + Q_{MV,EA} + \dot{SV} + \varepsilon\eta_{Ap}Q_{Ap}} \quad (5)$$

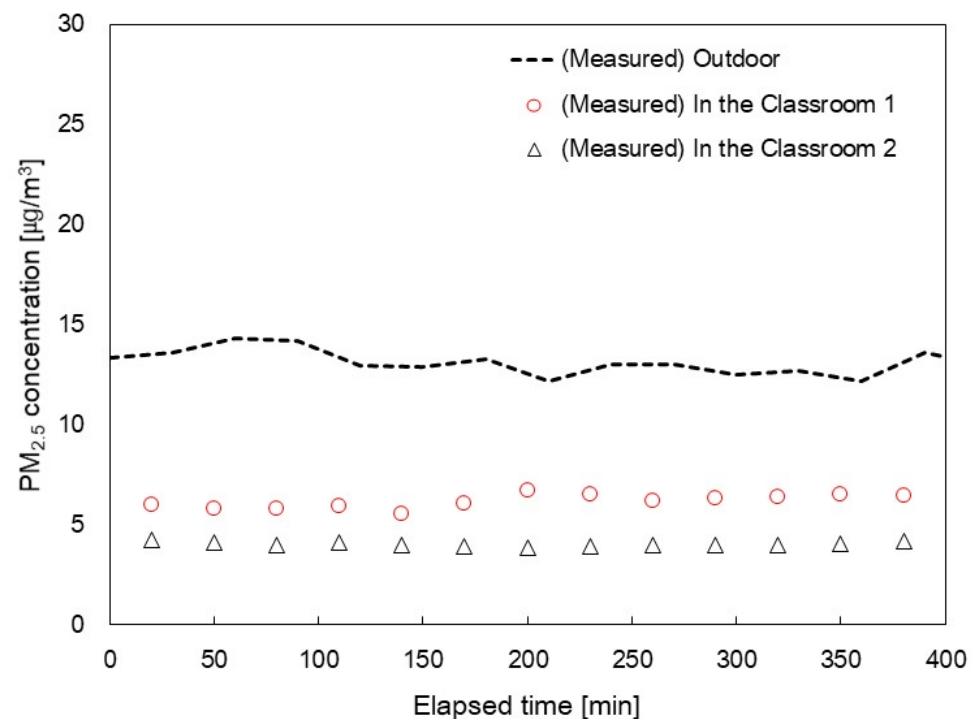


Figure 4. Saturated PM_{2.5} concentration indoors when an outdoor condition is stable according to time.

To obtain the value of $\eta_{Ap}Q_{Ap}$, we conducted measurements in a 30 m³ chamber using the standard test of the stated CADR (SPS-KACA002-0132 [31]). The air purifier within the kindergarten can be operated across flow rates of levels 1–4. The stated CADR of each level were 120, 210, 324, and 480 m³/h. However, the actual CADR measured from the classroom differ from the stated CADR measured from standard chamber. To determine the actual CADR, ε must be applied to account for the air circulation rate when the air purifier is in operation [20,28,29]. Figure 5 shows the ε values observed in both Classroom 1 and

Classroom 2. These measurements were conducted for 30 min and normalized to the initial concentration. During the measurements, one air purifier was operated in Classroom 1, whereas two air purifiers were operated in Classroom 2. All air purifiers were operated at level 2. The ε was calculated as 0.75 for Classroom 1 and 0.5 for Classroom 2. Given that the two air purifiers were situated at different locations within Classroom 1, it can be concluded that the air circulation rate in Classroom 1 was superior to that of Classroom 2.

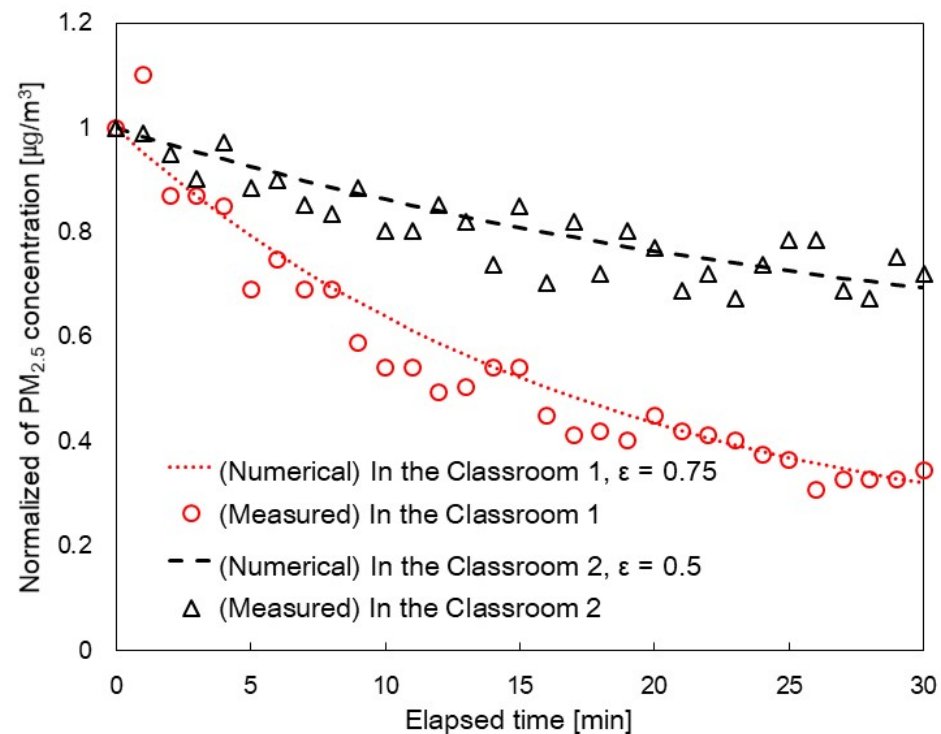


Figure 5. Measuring the EACR of air purifiers in Classrooms 1 and 2.

3. Results

3.1. Comparing the Numerical Model with Measured Data

In this study, we used two specific days within the January to March 2023 timeframe to compare the concentrations of indoor CO₂ and PM_{2.5} and those obtained via the numerical models. Malik (1978) demonstrated that natural ventilation is influenced by variables such as outdoor wind speed and the temperature differential between outdoor and indoor environments [32]. As such, 3 January 2023 and 11 January 2023, were selected because of their similarities to the outdoor wind speed and temperature difference observed on 7 November 2022, which served as a reference point in Figure 3.

Table 1 shows various parameters, including the number of occupants, Q_{inf} , Q_{MV} , and $\eta_{Ap}Q_{Ap}$ for both Classroom 1 and 2. The information presented in Table 1 was recorded during two distinct time periods: from 9:00 to 15:00, which encompassed the class time of the kindergarten, and 15:00–15:30, when the children left the premises. On 3 January 2023, the mechanical ventilation system operated at level 2, whereas on 11 January 2023, the system operated at level 1.

Table 1. Indoor condition of CO₂ and PM_{2.5} concentrations according to time for comparison in (a) Classroom 1 on 3 January 2023, (b) Classroom 2 on 3 January 2023, (c) Classroom 1 on 11 January 2023, (d) Classroom 2 on 11 January 2023.

Date	Classroom	Time	Number of People (Men/Women/Children)	Q_{inf} (m ³ /h)	$Q_{MV,SA}$ (m ³ /h)	$\eta_{Ap}Q_{Ap}$ (m ³ /h)
(a)						
3 January 2023	1	9:00~9:30	1/2/8	26	230	240
		9:30~10:30	1/2/11			
		10:30~12:20	1/2/0			
		12:20~14:50	1/2/11			
		14:50~15:30	1/2/0			
(b)						
3 January 2023	2	9:00~9:50	0/2/7	16	254	324
		9:50~11:00	0/2/11			
		11:00~12:40	1/2/0			
		12:40~15:10	0/2/11			
		15:10~15:30	0/1/4			
(c)						
11 January 2023	1	9:00~9:40	1/2/6	26	142	120
		9:40~11:00	1/1/0	143		
		11:00~12:00		26		
		12:00~13:00	1/2/10	143		
		13:00~15:00				
		15:00~15:30	0/2/0	571		
(d)						
11 January 2023	2	9:00~9:30	1/1/2	16	154	210
		9:30~11:00	1/2/7			
		11:00~12:30	1/2/0			
		12:30~15:00	2/2/8			
		15:00~15:30	1/2/0			

3.1.1. The Concentrations of CO₂

Figure 6 shows comparisons between the measured data of the indoor CO₂ concentrations and those obtained from the numerical model according to the time of day. Specifically, the graphs labeled as ‘In the Classroom 1’ and ‘In the Classroom 2’ represent the measured CO₂ concentration. On the other hand, the data labeled as ‘Numerical Model’ were obtained by substituting the recorded values from Table 1 into Equation (2) for CO₂ concentration.

Figure 6a presents the comparison data for Classroom 1 on 3 January 2023. Between 9:00 and 9:30, indoor CO₂ concentrations gradually increased to approximately 815 ppm as the children arrived at the kindergarten. During the history class from 9:30 to 10:30, characterized by a passive learning environment, the windows were closed, leading to a rise in CO₂ concentration to approximately 1105 ppm. Between 10:30 and 12:20, during which the children engaged in outdoor activities, indoor CO₂ concentrations considerably decreased to 585 ppm. From 12:20 to 14:50, there was lunchtime and playtime in the classroom, during which the children’s activity levels increased sharply, and the CO₂ concentration increased to 1180 ppm. Consequently, there were discrepancies between the

measured values and the numerical analysis results in Figure 6a. This can be attributed to the rapid increase in activity levels, which were not accurately reflected in the numerical analysis results. At 14:50, when the children went home, there was a decrease in CO₂ concentrations.

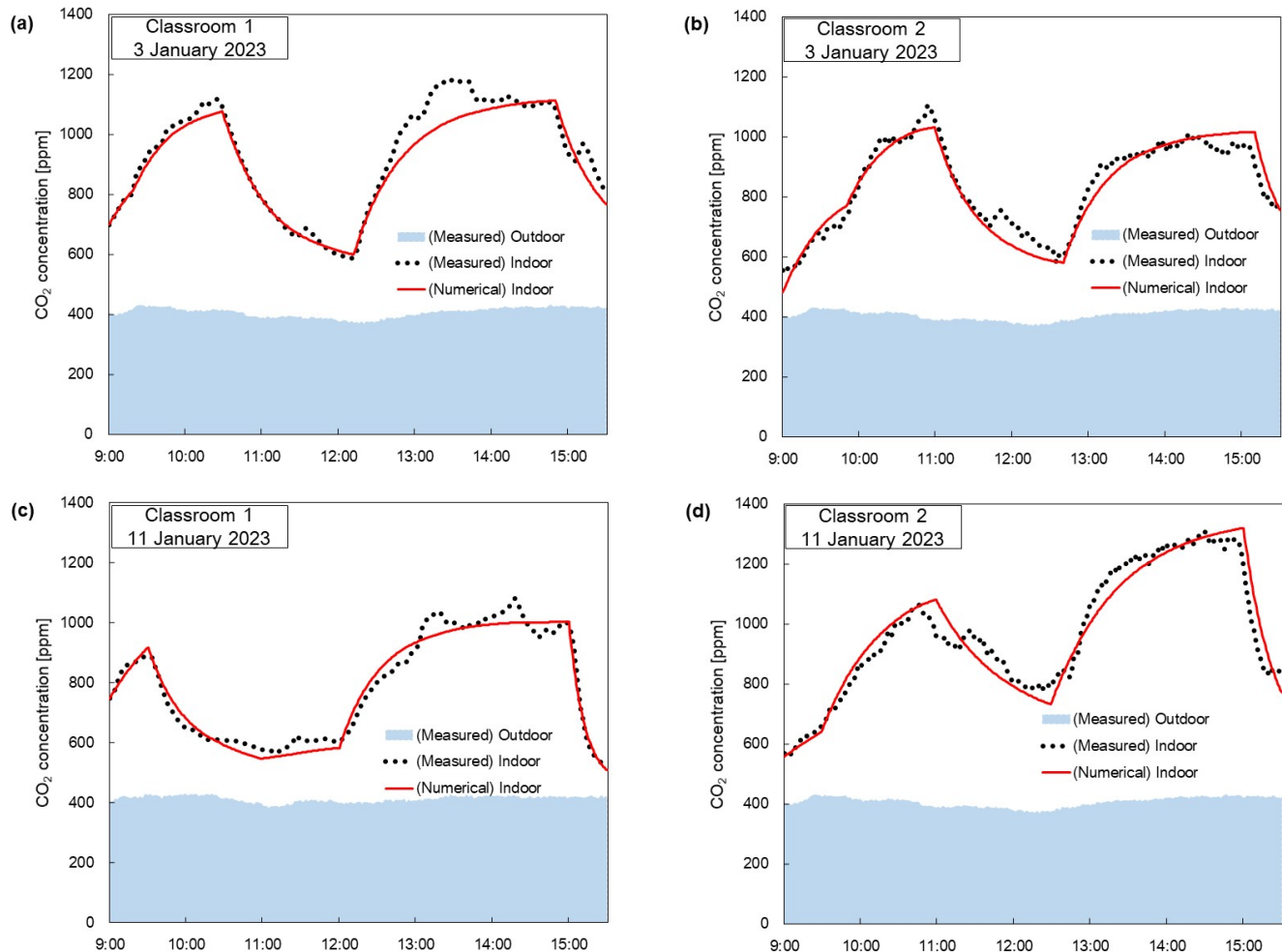


Figure 6. Comparison of CO₂ concentration obtained using the numerical model in (a) Classroom 1 on 3 January 2023, (b) Classroom 2 on 3 January 2023, (c) Classroom 1 on 11 January 2023, (d) Classroom 2 on 11 January 2023 with the measured CO₂ concentration data.

Figure 6b presents the comparison data for Classroom 2 on 3 January 2023. From 9:00 to 9:50, indoor CO₂ concentrations gradually increased to approximately 740 ppm as the children arrived at the kindergarten. During the history class from 9:50 to 11:00, the CO₂ concentration in Classroom 2 only increased to 1090 ppm. There were discrepancies between the measured values and numerical analysis results for 10:50–11:00; we assumed that there was high-level indoor activity as children were gearing up to go outside. Between 11:00 and 12:40, the children of Classroom 2 also engaged in outdoor activities, resulting in a considerable decrease in indoor CO₂ concentrations to 600 ppm. Lunchtime and naptime was from 12:40 to 15:10, unlike the Classroom 1. Therefore, there was no significant discrepancy between measured and numerical values, and the indoor CO₂ concentration only increased to 1005 ppm. From 15:10, the children went home, and the windows were opened, decreasing CO₂ concentrations.

Figure 6c shows the comparison data for Classroom 1 on 11 January 2023. From 9:00 to 9:40, indoor CO₂ concentrations gradually increased to approximately 870 ppm as the children arrived at the kindergarten. From 9:40 to 12:00, the children engaged in outdoor activities. From 9:40 to 11:00, one window was opened. Consequently, the indoor CO₂

concentrations decreased sharply, reaching approximately 565 ppm and converging around this concentration. From 11:00 to 12:00, the CO₂ concentrations slightly increased because of the closed window and the two teachers who remained in the classroom. From 12:00 to 15:00, there was a lunchtime and roleplay activity session with one window open in Classroom 1. The increased number of children initially led to a rise in CO₂ concentrations. However, there was an increase in CO₂ concentration initially to 1050 ppm after which the windows were opened and this concentration did not increase further. At 15:00, the children went home, and all the windows were opened, leading to decreased CO₂ concentrations.

Figure 6d shows the comparison data for Classroom 2 on 11 January 2023. From 9:00 to 9:30, indoor CO₂ concentrations gradually increased to approximately 640 ppm as the children arrived at the kindergarten. Most of the children had completed their arrival by 9:30, and they watched educational videos from 9:30 to 11:00 in Classroom 2 with all the windows closed. During this time, the CO₂ concentrations increased up to 1050 ppm. From 11:00 to 12:30, the children engaged in outdoor activities, and despite the presence of two teachers in the classroom, the indoor CO₂ concentration decreased to 790 ppm through mechanical ventilation. From 12:30 to 15:00, there was lunchtime and naptime. The indoor CO₂ concentration increased to 1230 ppm. Compared to Figure 6b, during the same period with a similar number of people and all the windows closed, the indoor CO₂ concentration was higher. This was due to a decrease in Q_{MV} from 254 to 157 m³/h. From 15:00, the children went home, and the windows were opened, leading to decreased CO₂ concentrations.

Figure 6 shows that, except for when the children's CO₂ generation increased due to high activity levels, the measured values and numerical analysis results were nearly the same, with an error rate of 6%. By analyzing Figure 6b,d, it was observed that increasing the ventilation system's airflow rates could also maintain lower indoor CO₂ concentrations.

3.1.2. The Concentrations of PM_{2.5}

Figure 7 shows a comparison between the measured data of indoor PM_{2.5} concentrations and those obtained from the numerical model according to the time of day. The data labeled 'Numerical Model' were obtained by substituting the recorded conditions from Table 1 into Equation (3) for PM_{2.5} concentrations.

Figure 7a shows the comparison data on PM_{2.5} concentrations for Classroom 1 on 3 January 2023. The windows were closed throughout the day, and the flow rates of mechanical ventilation and the air purifier were constant. Unlike Figure 6a, indoor PM_{2.5} concentrations vary with outdoor PM_{2.5} concentrations regardless of the number of people. The measured values and numerical analysis results are almost identical from 09:00 to 13:00. However, from 13:00 to 15:30, there are discrepancies between the measured and the numerical analysis results. During this time, the indoor PM_{2.5} concentration increased as the children's activity increased due to the children's playtime. Determining indoor activity levels solely based on the set parameters presented a challenge. Future research will be necessary to predict changes in indoor PM_{2.5} concentrations with consideration for children's activity levels.

Figure 7b presents the comparison data for Classroom 2 on 3 January 2023. Except when the children went home, the windows were closed, and the flow rates of mechanical ventilation and the air purifier were constant. Figure 7b shows that the measured values and numerical analysis results agreed. However, there is a slight discrepancy between the measured values and the numerical analysis results during the 09:30 to 09:40 and 12:30 to 13:00 timeframes. This was due to the PM_{2.5} generated by rearranging furniture for indoor classes. Unlike Figure 7a, the measured indoor PM_{2.5} concentration remains low between 13:00 and 15:00, consistent with the numerical analysis results shown in Figure 7b. This is because the activity level was low due to the children taking a nap. After 15:00, the indoor PM_{2.5} concentration increases sharply due to the inflow of PM_{2.5} from opening a window.

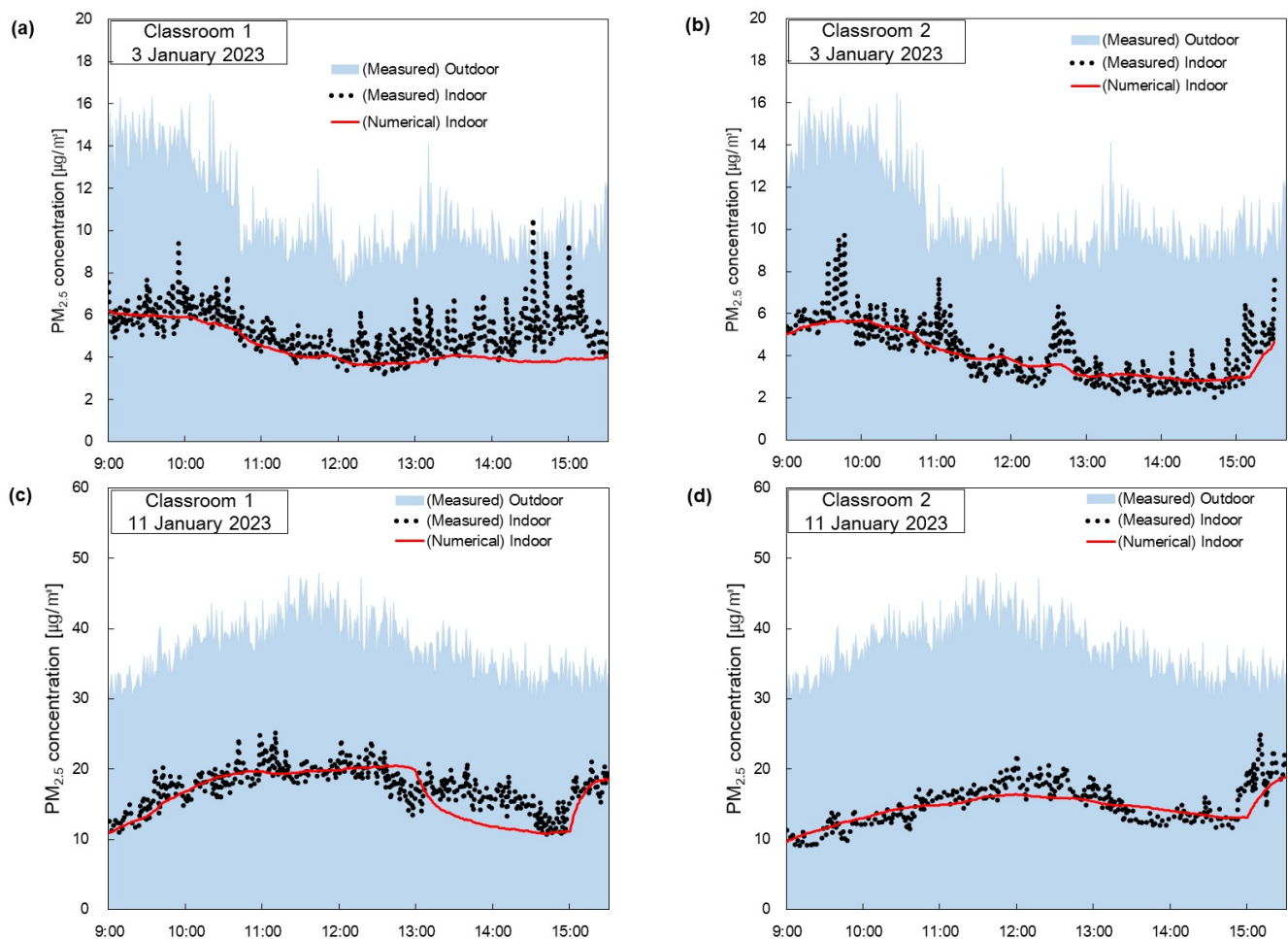


Figure 7. Comparison with measured data of PM_{2.5} concentration obtained by the numerical model in (a) Classroom 1 on 3 January 2023, (b) Classroom 2 on 3 January 2023, (c) Classroom 1 on 11 January 2023, (d) Classroom 2 on 11 January 2023 with the measured PM_{2.5} concentration data.

Figure 7c presents the comparison data for Classroom 2 on 3 January 2023. Only the flow rate of mechanical ventilation was constant, while the flow rates of the natural ventilation and air purifiers were varied. It can be seen that from 09:00 to 12:30, the measured values and numerical analysis results were generally consistent. However, from 12:30 to 15:00, there are discrepancies between the measured values and the numerical analysis results. Similar to Figure 7a, the increase in activity due to children's indoor activity results in higher PM_{2.5} concentrations than the numerical analysis results. After 15:00, the indoor PM_{2.5} concentration increases sharply due to the inflow of PM_{2.5} from opening the windows.

Figure 7d shows the comparison data for Classroom 2 on 11 January 2023. From 09:00 to 12:00, the indoor PM_{2.5} concentration gradually increased as the outdoor PM_{2.5} concentration increased, and in the afternoon, the indoor PM_{2.5} concentration decreased as the outdoor PM_{2.5} concentration decreased. From 15:00, the window was opened, and the indoor PM_{2.5} concentration increased sharply due to the inflow of PM_{2.5}. This shows that the indoor PM_{2.5} concentration is strongly affected by the outdoor PM_{2.5} concentration.

Figure 7 shows that indoor PM_{2.5} concentrations are generally affected by outdoor PM_{2.5} concentrations. In Figure 7a,b, we can see that on days with low outdoor PM_{2.5} concentrations, indoor PM_{2.5} concentrations stay below five µg/m³ on average. However, on days when the outdoor PM_{2.5} concentration is high, as shown in Figure 7c,d, the indoor PM_{2.5} concentration stays at an average of about 15 µg/m³ due to the inflow of PM_{2.5} despite the operation of the air purifier. Therefore, it is essential to reduce natural ventilation

and increase the dust collection efficiency of mechanical ventilation filters on days when the outdoor PM_{2.5} concentration is high to keep the indoor PM_{2.5} concentration low.

4. Discussion

4.1. Flowrate of Mechanical Ventilation

Having validated the accuracy of Equation (2) through the previous experiment, we aimed to utilize it to present the required $Q_{MV,SA}$ based on the number of children at which CO₂ concentrations below 1000 ppm can be maintained. Equation (2) can be simplified to Equation (6) as the term $\exp^{-\frac{Q_{MV,SA}+Q_{inf}}{V}t}$ becomes negligible because the volume of most classrooms in the kindergarten is small enough to make indoor CO₂ concentration almost be saturated within 2 h. And in Equation (6), the required $Q_{MV,SA}$ has been calculated to maintain the CO₂ concentration below 1000 ppm.

$$C_{CO_2}(t) = \frac{C_{CO_2,out}(Q_{inf} + Q_{MV,SA}) + G}{Q_{exf} + Q_{MV,EA}} \quad (6)$$

Figure 8 shows the airflow rates supplied by mechanical ventilation to maintain indoor CO₂ concentrations below 1000 ppm for varying numbers of children, as determined by Equation (6). The $C_{CO_2,out}$ was set to 423 ppm, acquired from the annual average atmospheric CO₂ concentration in 2021, as reported by the National Oceanic and Atmospheric Administration's Global Atmospheric Monitoring [33]. It was assumed that two women teachers were present in the classroom. Therefore, when no children were present indoors, a minimum ventilation of 40 m³/h was required. If classrooms in the kindergarten are filled to maximum capacity, which is two women teachers and 15 children, a ventilation rate of 345 m³/h is necessary for ideal ventilation. Calculating the slope of the graph further indicates that classrooms require 20.4 m³/h of ventilation per child.

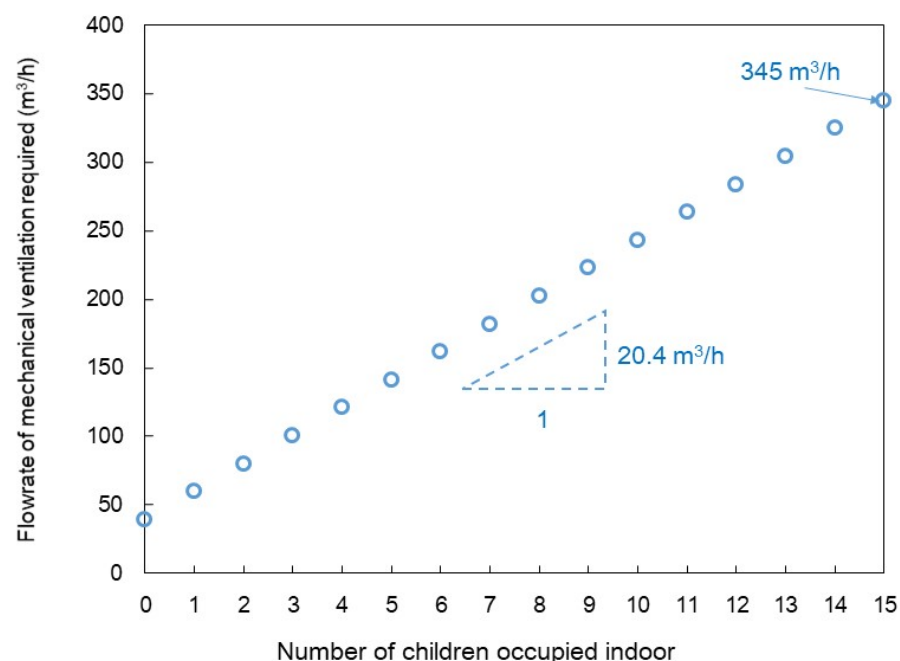


Figure 8. Flowrate of mechanical ventilation required, according to the number of children occupied indoor.

Figure 9 shows the days during the measurement period when the CO₂ concentrations were highest. Furthermore, the measurements on those specific days were compared with the $Q_{MV,SA}$ calculated using Equations (2) and (6) and applied to derive the values, which was denoted as “Scenario1”. As the conditions of Scenario 1, $Q_{MV,SA}$ is 345 m³/h that is

the highest flowrate in Figure 8, η_{MV} is 0.35 and $\eta_{AP}Q_{AP}$ is 210 m³/h. Table 2 shows the values applied to the numerical model over time for Scenario 1. Figure 9a shows the results of the measurements in Classroom 1. For the measured concentration in Classroom 1, the CO₂ concentration was almost over 1000 ppm. When Scenario 1 was applied, CO₂ concentrations were maintained below 1000 ppm during class time. Figure 9b shows the CO₂ concentrations measured on the same day for Classroom 2. Scenario 1, as calculated in Figure 9a, also allowed us to maintain CO₂ concentrations below 1000 ppm.

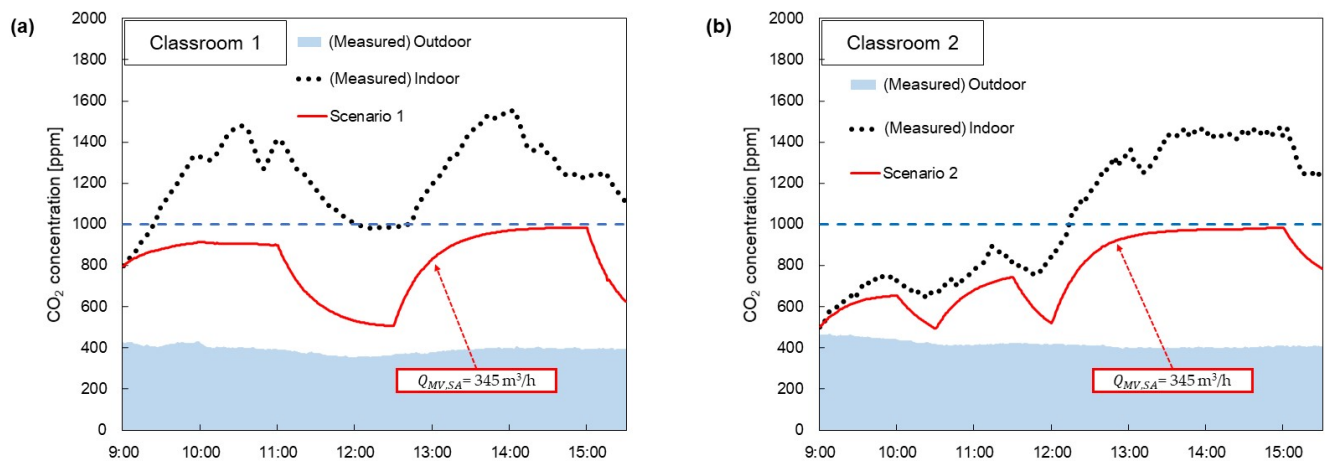


Figure 9. Graph showing that indoor CO₂ concentration was maintained below 1000 ppm in ‘Scenario 1’ and that the ideal amount of ventilation was applied on days when indoor CO₂ concentration was highest from January to March 2023 in (a) Classroom 1, (b) Classroom 2.

Table 2. Indoor condition of the Numerical Model for CO₂ in a day when the CO₂ concentration was highest from January to March 2023 in (a) Classroom 1, (b) Classroom 2.

(a)			
Classroom 1	Number of People (Men/Women/Children)	Q_{inf} (m ³ /h)	$Q_{MV,SA}$ (m ³ /h)
9:00~11:00	1/2/11	26	142 (Measured) Indoor 345 Scenario 1
11:00~12:30	1/2/0		
12:30~15:00	0/2/15		
15:00~16:00	0/2/0		
16:00~17:30	0/2/11		
17:30~18:00	1/2/0		
(b)			
Classroom 2	Number of People (Men/Women/Children)	Q_{inf} (m ³ /h)	$Q_{MV,SA}$ (m ³ /h)
9:00~10:00	0/2/4	16	154 (Measured) Indoor 345 Scenario 1
10:00~10:30	0/0/0		
10:30~11:30	0/2/8		
11:30~12:00	0/2/0		
12:00~15:00	0/2/15		
15:00~15:30	0/2/6		
15:30~16:30	0/2/13		
16:30~18:00	0/2/2		

4.2. Decreasing the Concentration of $PM_{2.5}$

Figure 10 shows the measured data for the indoor $PM_{2.5}$ concentration and the simulation under the conditions of scenarios 1, 2, and 3 on days with the highest indoor $PM_{2.5}$ concentrations. Under the conditions of scenario 2, $Q_{MV,SA}$ is $345 \text{ m}^3/\text{h}$, η_{MV} is 0.35 and $\eta_{AP}Q_{AP}$ is $480 \text{ m}^3/\text{h}$ that is a level 4 flowrate for the air purifier. Under the conditions of scenario 3, $Q_{MV,SA}$ is $345 \text{ m}^3/\text{h}$, η_{MV} is 0.75 which has a MERV 13 filter grade and $\eta_{AP}Q_{AP}$ of $210 \text{ m}^3/\text{h}$.

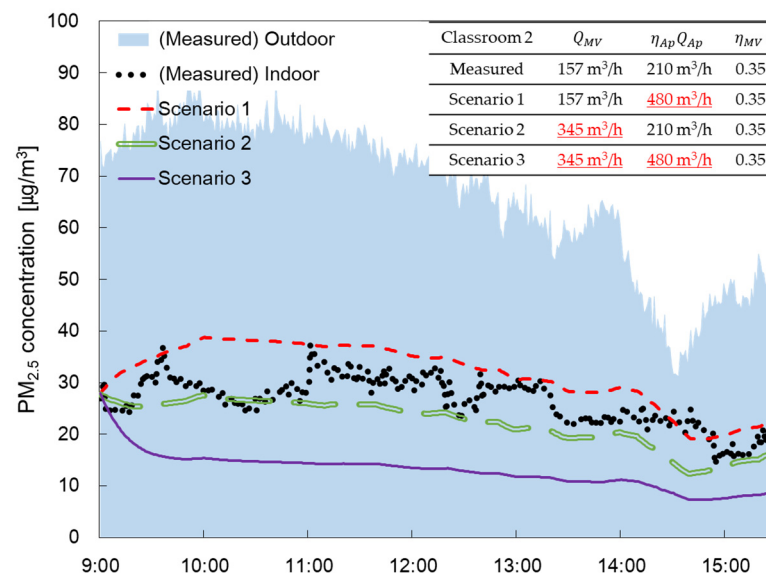


Figure 10. Graph comparing $PM_{2.5}$ concentration over time obtained from simulation under the scenario conditions with measured data to show the effect of decreasing $PM_{2.5}$ concentration.

Average $PM_{2.5}$ concentrations of a day under Scenario 1 ($Q_{MV,SA} = 345 \text{ m}^3/\text{h}$) were about 20% higher than the measured $PM_{2.5}$ concentration ($Q_{MV,SA} = 157 \text{ m}^3/\text{h}$). This indicated that increasing $Q_{MV,SA}$ leads to an influx of outdoor $PM_{2.5}$. To mitigate the increase in indoor $PM_{2.5}$ concentrations, it is necessary to either increase the CADR ($\eta_{AP}Q_{AP}$) of air purifiers or improve the filtration efficiency of the mechanical ventilation system's filters. Average $PM_{2.5}$ concentrations under Scenario 2 where $\eta_{AP}Q_{AP}$ of an air purifier is increased from $210 \text{ m}^3/\text{h}$ to $480 \text{ m}^3/\text{h}$ were only about 16% lower than the measured $PM_{2.5}$ concentration ($\eta_{AP}Q_{AP} = 210 \text{ m}^3/\text{h}$).

In Scenario 3, the filter in the ventilation system was replaced with a MERV13 filter and the flowrate of the mechanical ventilation system was increased to $345 \text{ m}^3/\text{h}$. The $PM_{2.5}$ concentrations ($\eta_{MV} = 0.75$, $Q_{MV,SA} = 345 \text{ m}^3/\text{h}$) obtained from Scenario 3 were 52% lower than the measured $PM_{2.5}$ concentration ($\eta_{MV} = 0.35$, $Q_{MV,SA} = 157 \text{ m}^3/\text{h}$). Even on the day that had significantly high indoor $PM_{2.5}$ concentrations (average $PM_{2.5}$ concentration = $26.5 \text{ µg}/\text{m}^3$), the average daily $PM_{2.5}$ concentration can be reduced to below $15 \text{ µg}/\text{m}^3$; under Scenario 3. The indoor $PM_{2.5}$ concentration was effectively reduced by enhancing the collection efficiency of the mechanical ventilation filter when the mechanical ventilation flow rate was high enough.

5. Conclusions

Our study was conducted in a kindergarten setting, recognizing the heightened vulnerability of children to indoor air quality issues compared to adults. Utilizing a numerical model, we have proposed an effective method for indoor air quality management employing air purifiers and ventilation devices. Furthermore, our study introduces strategies for reducing indoor $PM_{2.5}$ concentrations and lowering indoor CO_2 concentration levels.

The mechanical ventilation rate necessary to maintain a CO_2 concentration below 1000 ppm was calculated. It can be deduced that approximately $20.4 \text{ m}^3/\text{h}$ of ventilation

per child is required for kindergarten classrooms. These results can be useful in designing a mechanical ventilation system for rooms in a kindergarten.

If a high collection efficiency filter is equipped in mechanical ventilation system, then an increased flowrate of mechanical ventilation can lead to results that show a greater reduction in indoor PM_{2.5}. Consequently, a mechanical ventilation filter with a high collection efficiency for PM_{2.5} is advisable, especially in multi-use facilities such as kindergartens, where a significant amount of CO₂ is generated via respiration.

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Nomenclature

$C_{CO_2}(t)$	Indoor CO ₂ Concentration (ppm) = $C_{CO_2,in}$
$C_{PM_{2.5}}(t)$	Indoor PM _{2.5} Concentration (μg/m ³) = $C_{PM_{2.5},in}$
$C_{CO_2,out}$	Outdoor CO ₂ Concentration (ppm)
$C_{PM_{2.5},out}$	Outdoor PM _{2.5} Concentration (μg/m ³)
Q_{inf}	Indoor Infiltration (m ³ /h)
Q_{exf}	Indoor Filtration Rate (m ³ /h)
$Q_{MV,SA}$	Flow Rate of Supply via Mechanical Ventilation (m ³ /h)
$Q_{MV,EA}$	Exhaust Flow Rate by Mechanical Ventilation (m ³ /h)
G	Amount of CO ₂ Generated in Room (ppm × m ³ /min)
$\eta_{Ap}Q_{Ap}$	Clean Air Delivery Rate (m ³ /h = State CADR)
ϵ	Efficiency of Air Cleaning Rate (m ³ /h = EACR)
$\dot{S}$	Deposition Rate of PM _{2.5} (h ⁻¹)
V	Volume of Indoor (m ³)
η_{inf}	Collection Efficiency of PM _{2.5} Particles by Infiltration
η_{MV}	Collection Efficiency of PM _{2.5} Particles by Mechanical Ventilation

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