

Supplementary Materials: A Reliable Method for the Preparation of Multiporous Alumina Monoliths by Ice-Templating

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Acid Source and Acid/Al Ratio

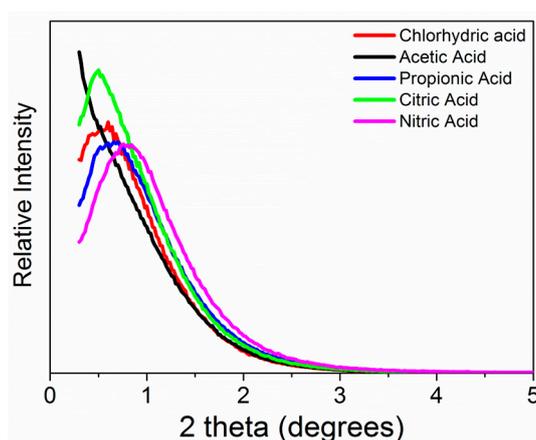


Figure S1. Small angle XRD for samples synthesized using different peptizing acids.

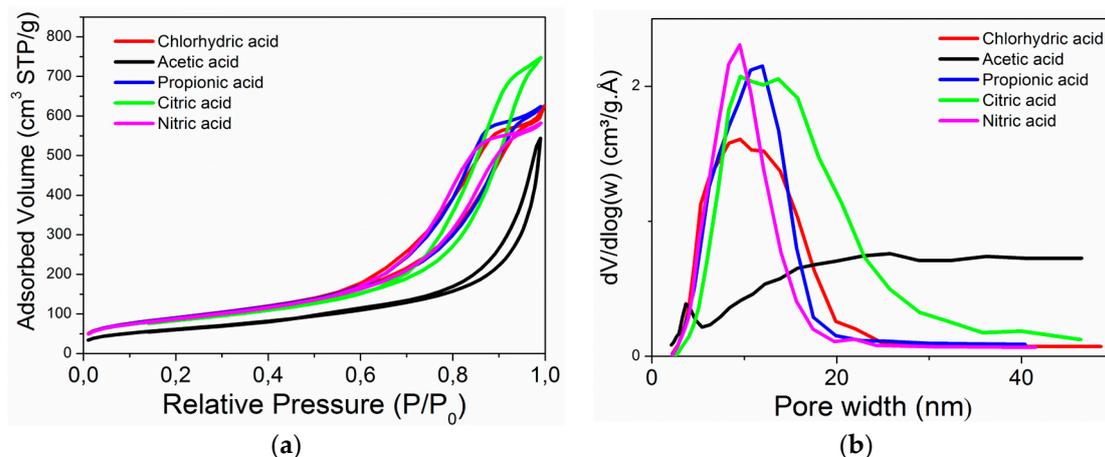


Figure S2. (a) Nitrogen adsorption—desorption isotherms and (b) pore size distributions (desorption branch) for samples synthesized using different peptizing acids.

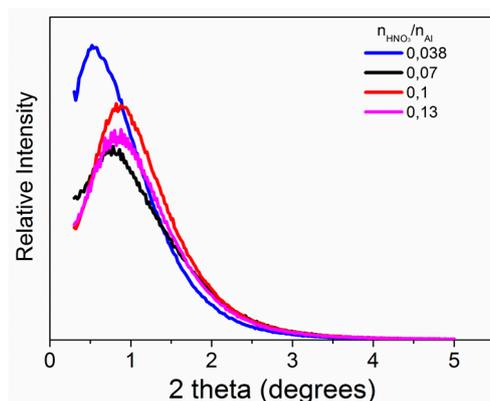


Figure S3. Small angle XRD for samples synthesized using different concentrations of nitric acid.

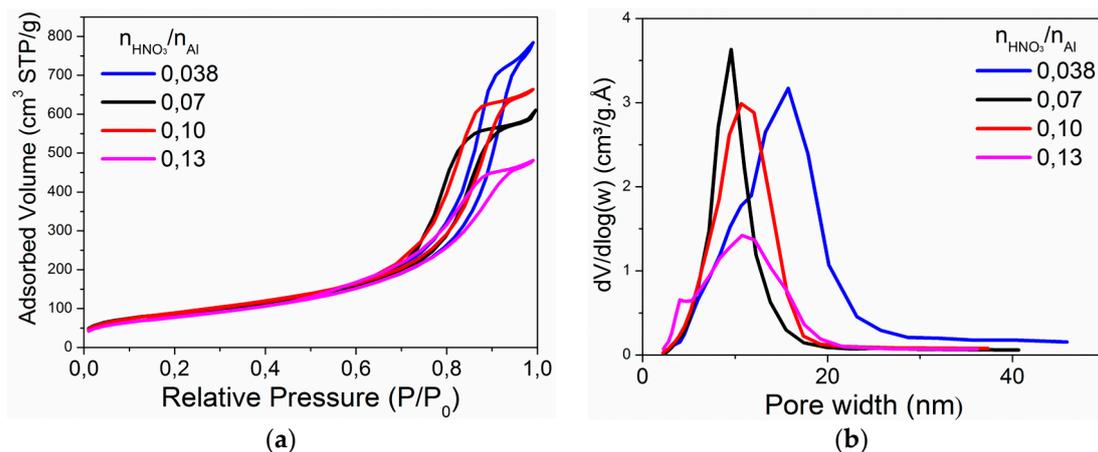


Figure S4. (a) Nitrogen adsorption—desorption isotherms and (b) pore size distributions for samples synthesized using different HNO₃/Al ratios.

Impact of Peptizing Duration and Temperature

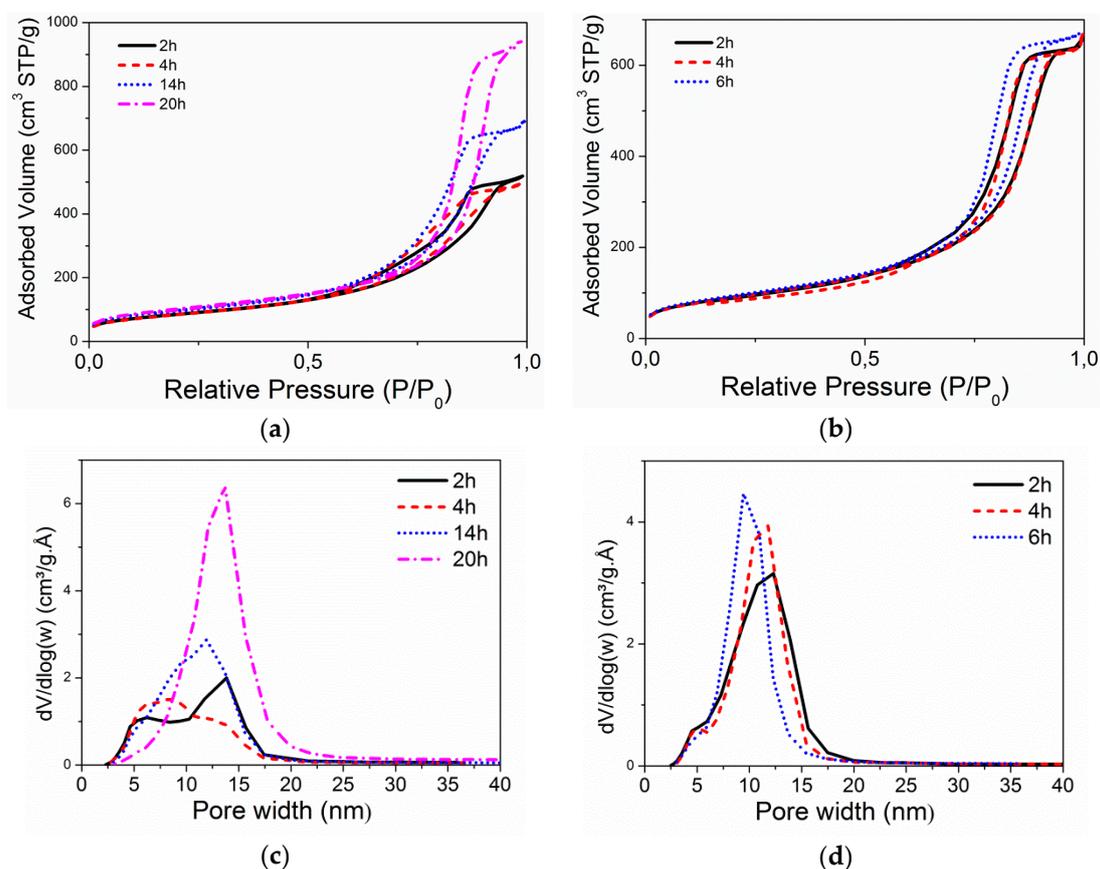


Figure S5. (a,b) Nitrogen adsorption—desorption isotherms and (c,d) pore size distributions for samples synthesized at (a,c) 85 °C and (b,d) 100 °C.

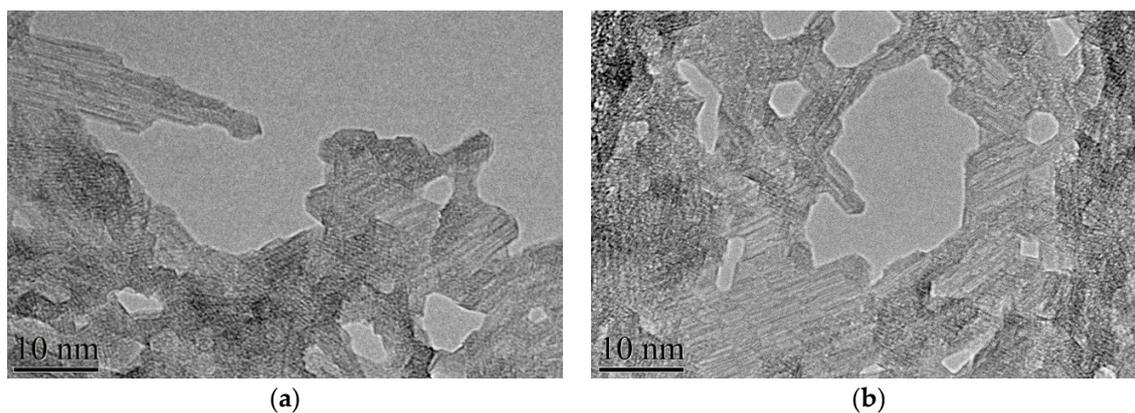


Figure S6. TEM images of the pseudoboehmite respectively after (a) 20 h at 85 °C and (b) 6 h at 100 °C.

Powder/Monolith Comparison

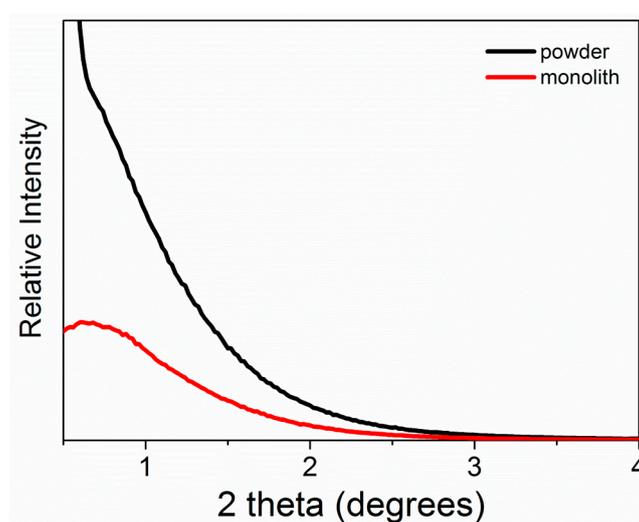


Figure S7. Comparative small angle XRD of powder and monolith synthesized under optimized conditions.

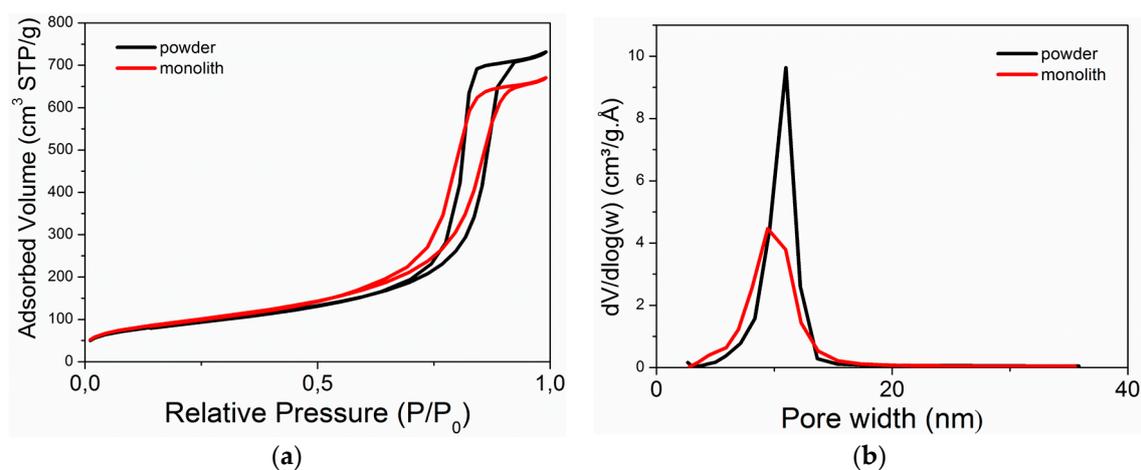


Figure S8. Comparative (a) Nitrogen adsorption–desorption isotherms and (b) corresponding pore size distributions of powder and monolith synthesized under optimized conditions.

Water Content

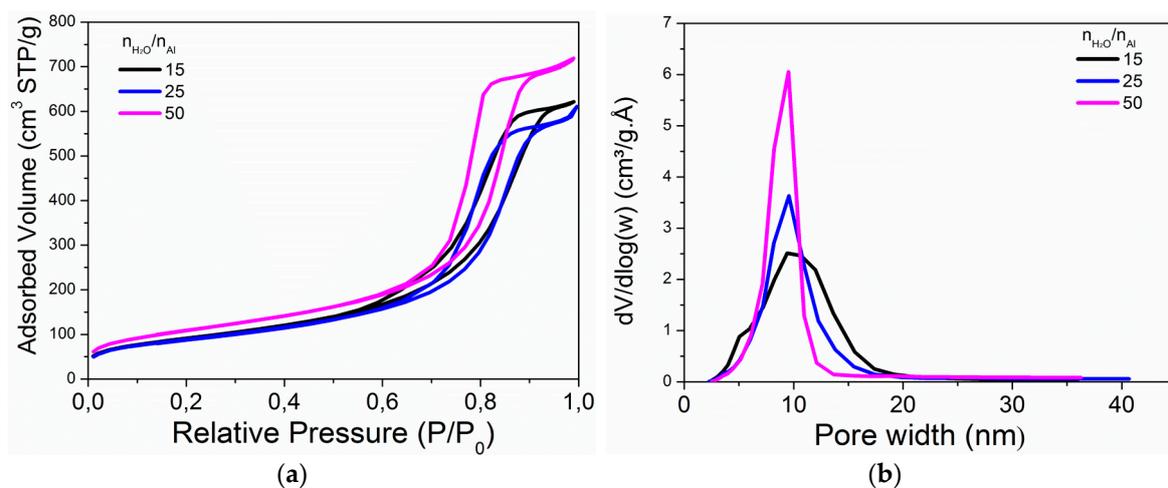


Figure S9. Impact of water content on (a) Nitrogen adsorption–desorption isotherms and (b) corresponding pore size distributions.

Thermal Treatment

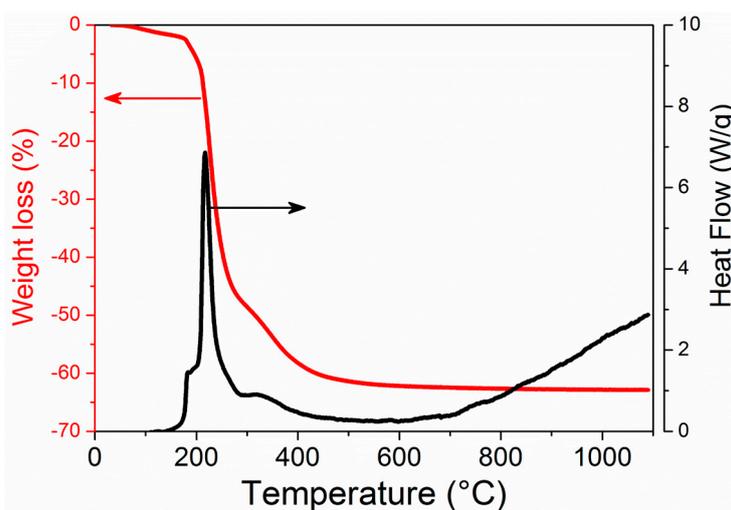


Figure S10. TGA-DSC analysis of ice-templated sample after sublimation of the ice crystals. An intense exothermic peak is typical of organics combustion.

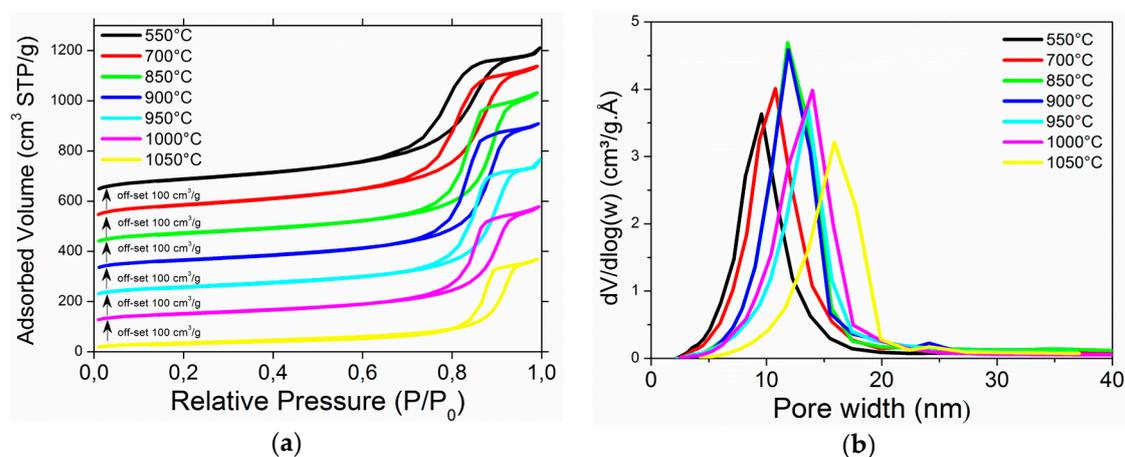


Figure S11. Impact of the thermal treatment (a) Nitrogen adsorption–desorption isotherms and (b) pore size distribution.

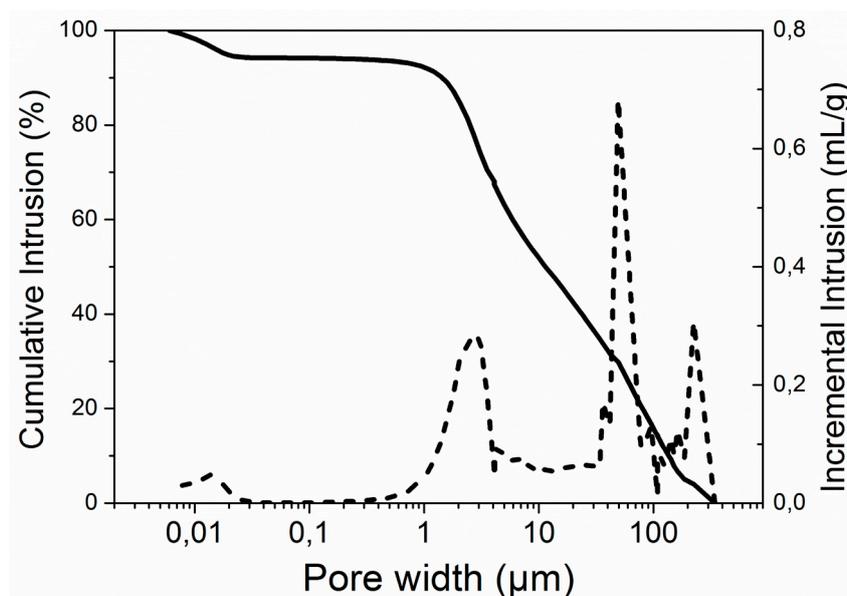


Figure S12. Mercury porosimetry of alumina monolith.