

Supplementary Material

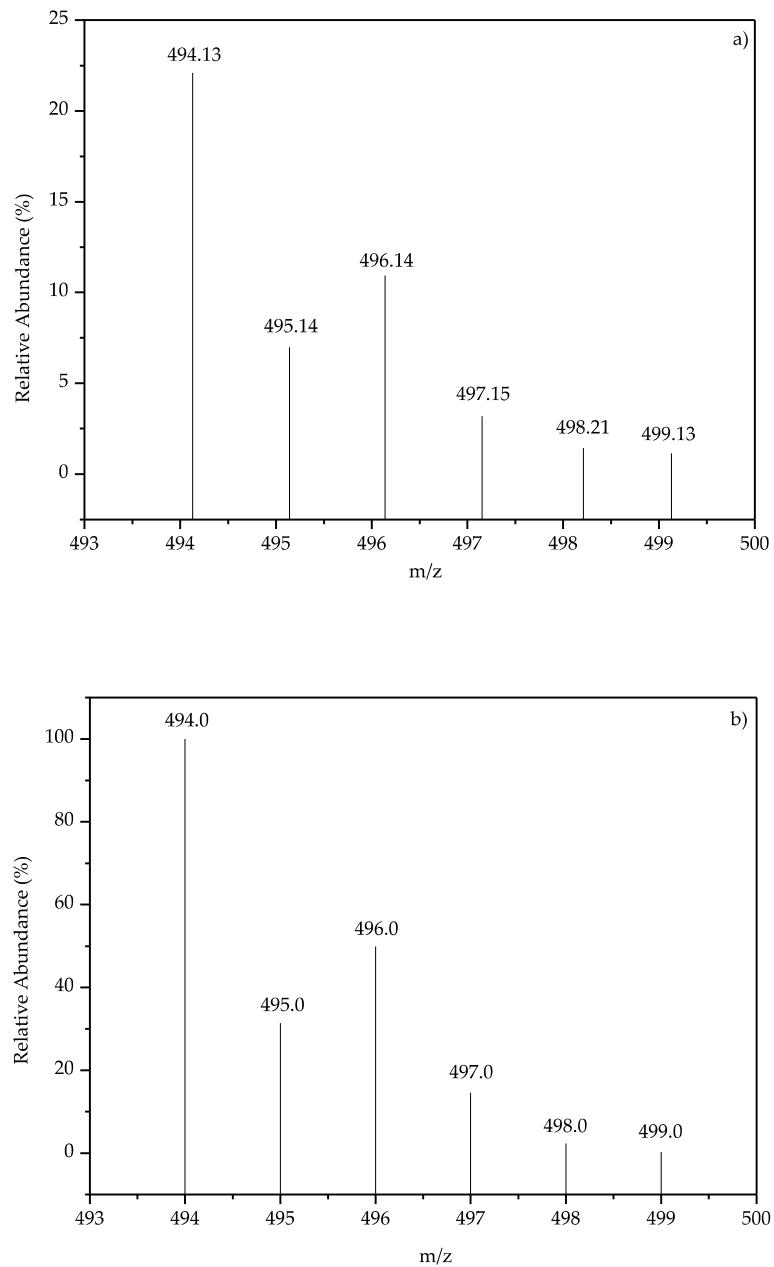


Figure S1. ESI spectrum of complex **1** in a CH_3OH . (a). Isotopic distribution calculated for the species $[\text{Cu}(\text{dmp})(\text{L})]^+ = [\text{Cu}(\text{C}_{14}\text{H}_{12}\text{N}_2)(\text{C}_{14}\text{H}_7\text{O}_3)]^+$ (b) (Qual Browser version 2.0.7copyright® Thermo Fischer Scientific Inc. 1998-2007).

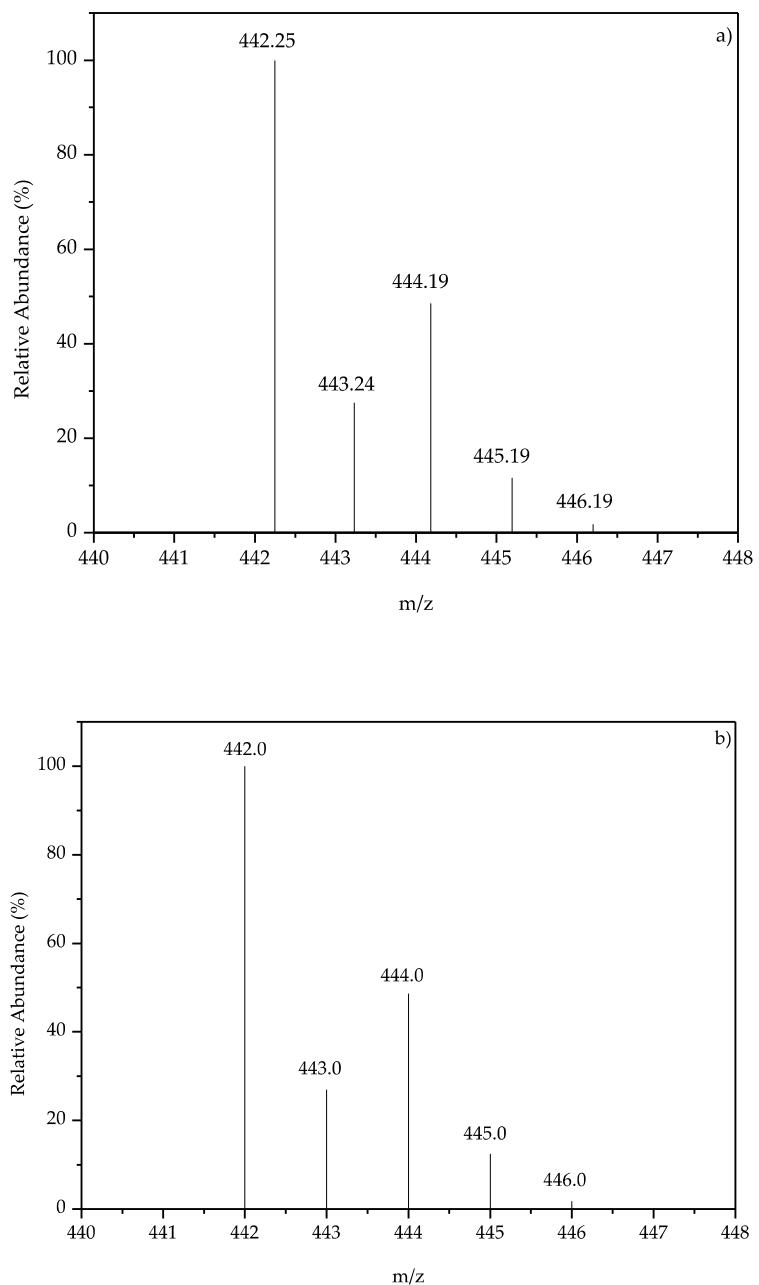
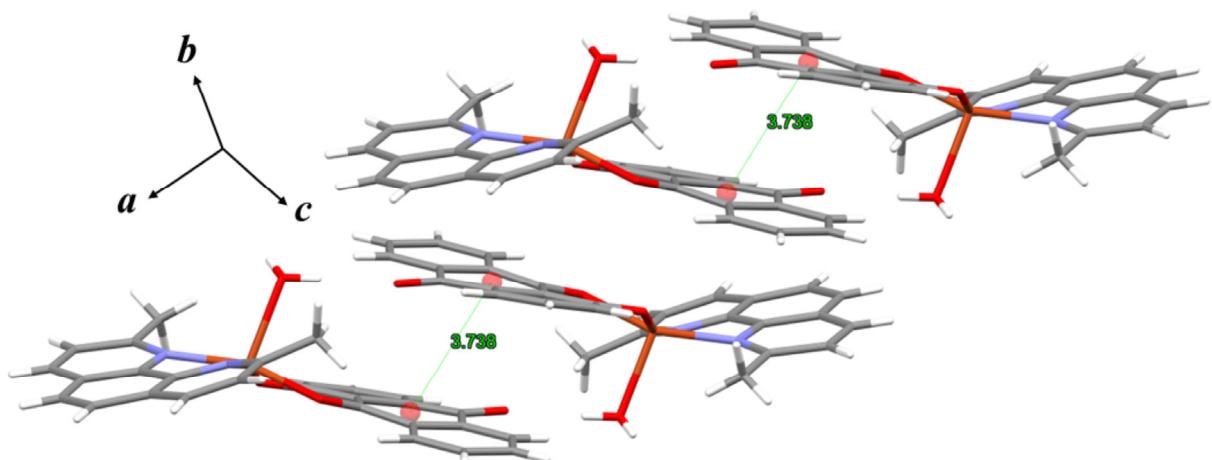
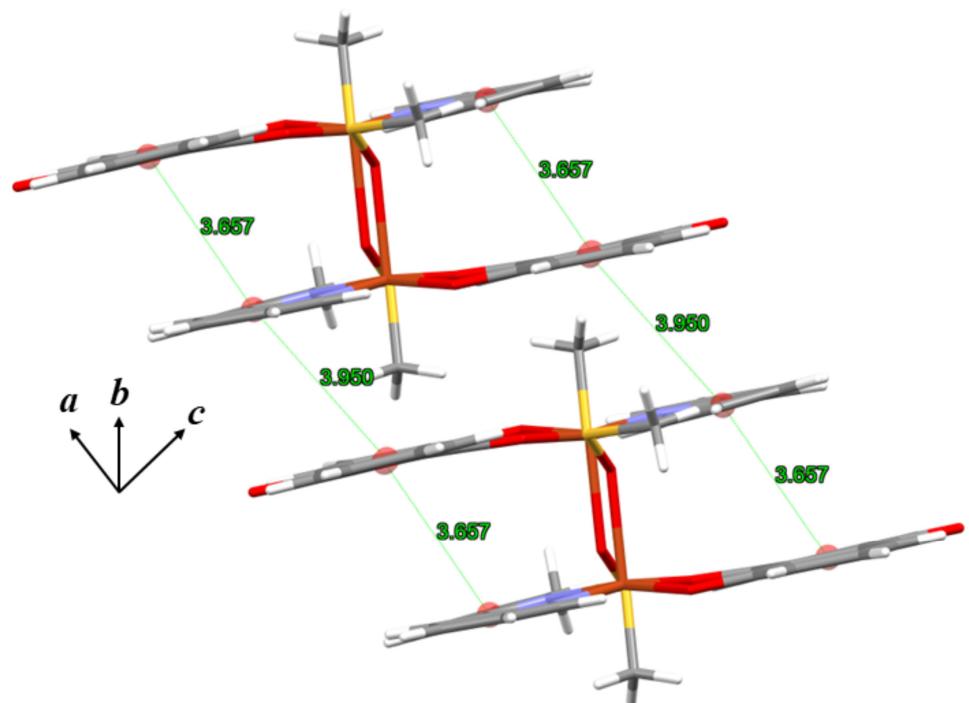


Figure S2. ESI spectrum of complex **2** in a CH_3OH . (a). Isotopic distribution calculated for the species $[\text{Cu}(\text{byp})(\text{L})]^+ = [\text{Cu}(\text{C}_{10}\text{H}_8\text{N}_2)(\text{C}_{14}\text{H}_7\text{O}_3)]^+$ (b) (Qual Browser version 2.0.7copyright® Thermo Fischer Scientific Inc. 1998-2007).



(a)



(b)

Figure S3. π -stacking interactions observed in a) complex 1 and b) complex 2.1.

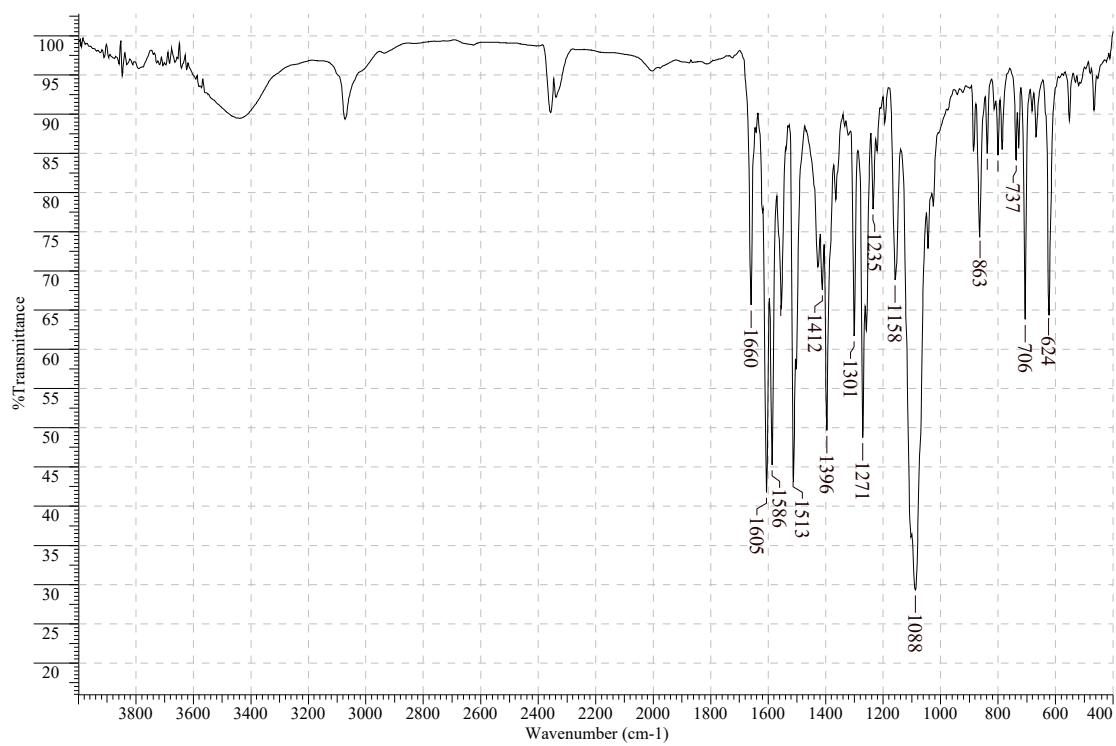


Figure S4. Infrared spectra ($4000\text{ cm}^{-1} - 400\text{ cm}^{-1}$), in KBr, of complex **1**, $[\text{Cu}(\text{dmp})(\text{L})(\text{H}_2\text{O})]\text{ClO}_4$.

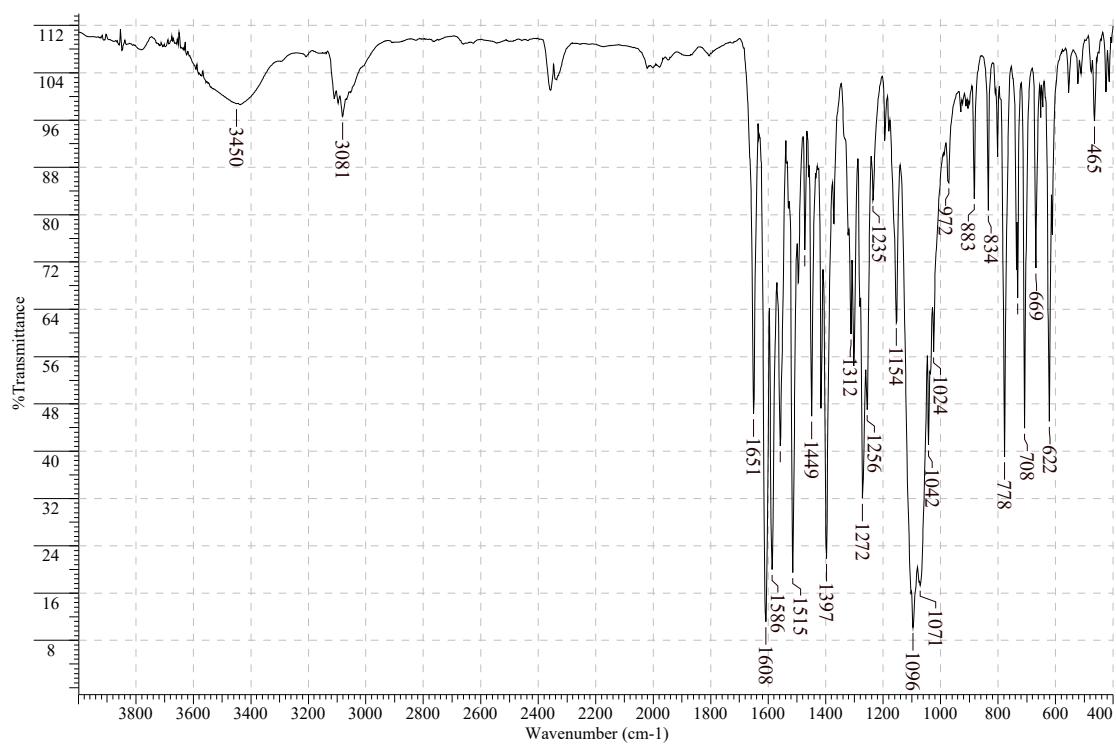


Figure S5. Infrared spectra ($4000 \text{ cm}^{-1} - 400 \text{ cm}^{-1}$), in KBr, of complex 2, $[\text{Cu}(\text{bpy})(\text{L})(\text{H}_2\text{O})(\text{ClO}_4)]$.

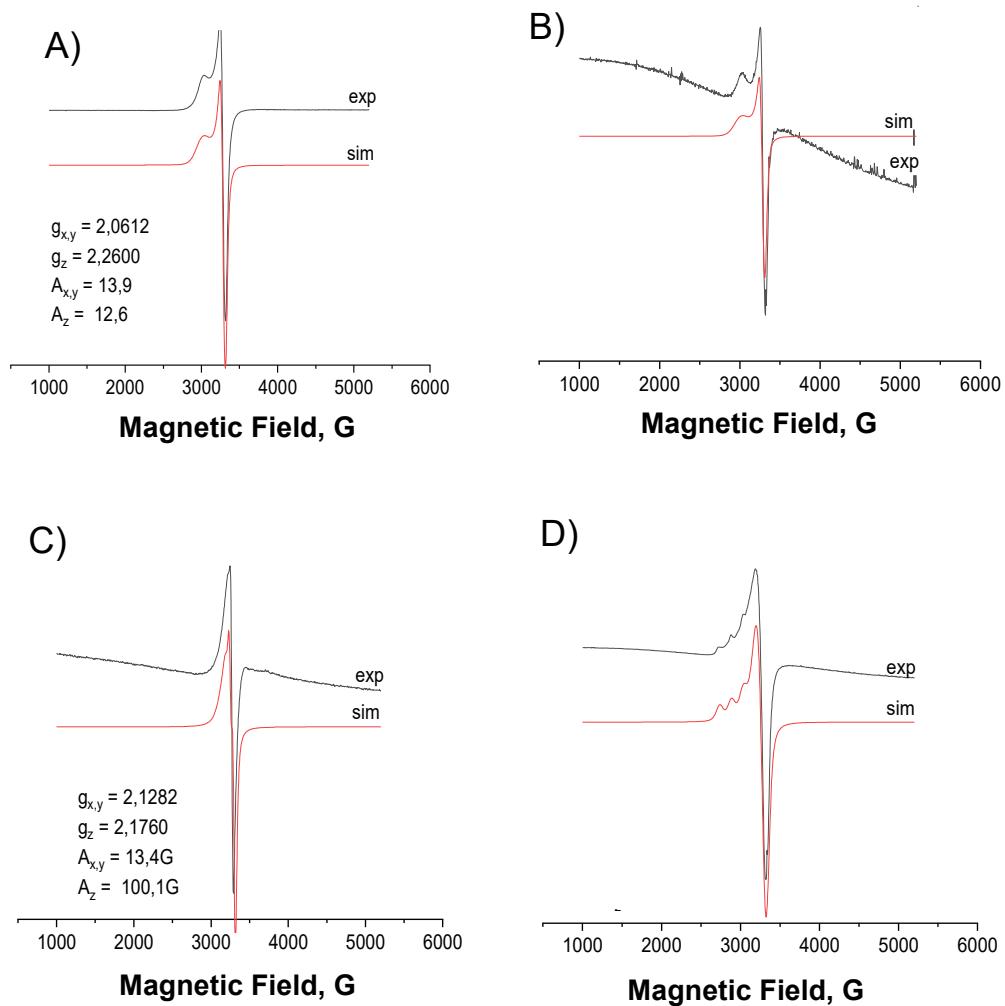


Figure S6. EPR spectra of complex **1**, $[\text{Cu}(\text{dmp})(\text{L})(\text{H}_2\text{O})](\text{ClO}_4)$, (—) and simulation (—) A) solid state, 298K; B) solid state, 77K; C) in dmso solution, 298K; D) in dmso solution, 77K

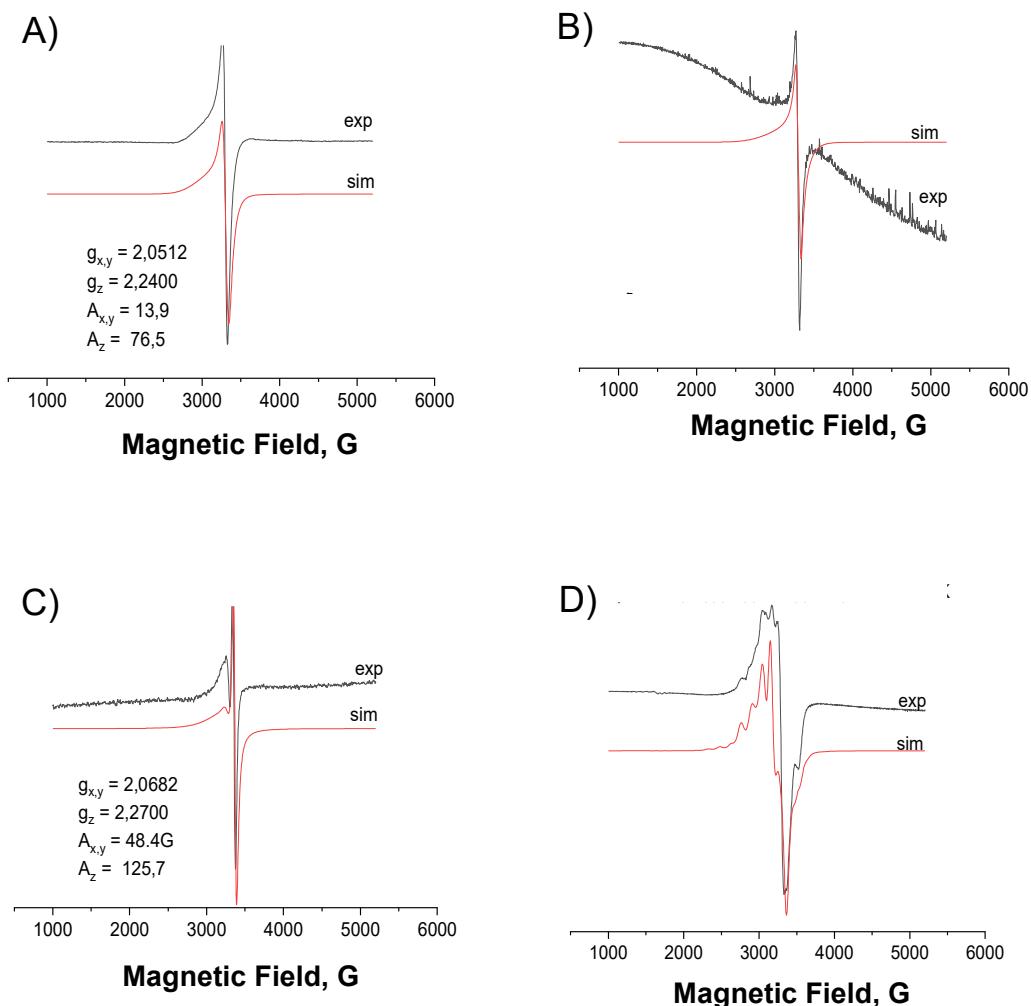


Figura S7. EPR spectra of complex **2**, $[\text{Cu}(\text{bpy})(\text{L})(\text{H}_2\text{O})](\text{ClO}_4)$, (—) and simulation (—) at A) solid state, 298K; B) solid state, 77K; C) in dmsol solution, 298K; D) in dmsol solution, 77K

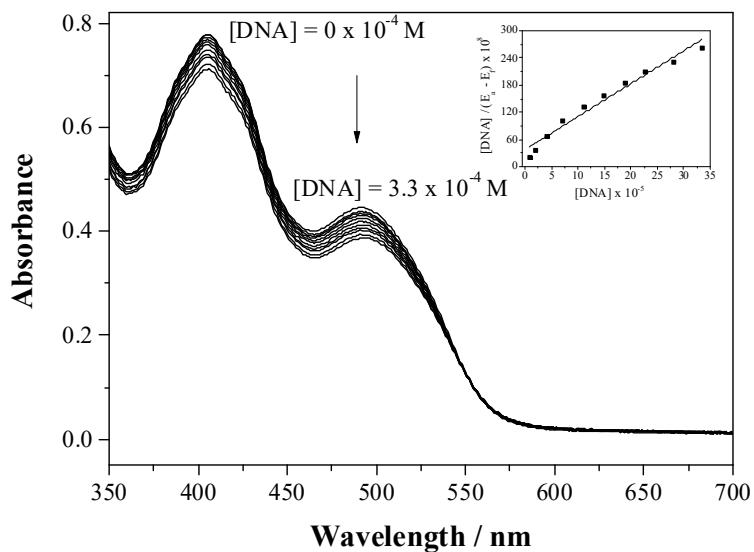


Figure S8. Spectra of solutions containing complex **2** (2.0×10^{-4} mol L⁻¹) and increasing concentrations of CT-DNA from 0 to 3.3×10^{-4} mol L⁻¹ dmso/ Tris-HCl, 1:2, pH = 7.2, I = 50 mmol L⁻¹. Inset: plot of [DNA] / ($\epsilon_a - \epsilon_f$) versus [DNA].