

Article

Photoacoustic Detection of Pollutants Emitted by Transportation System for Use in Automotive Industry

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Abstract: In photoacoustic spectroscopy, the signal is inversely proportional to the resonant cell volume. Photoacoustic spectroscopy (PAS) is an absorption spectroscopy technique that is suitable for detecting gases at low concentrations. This desirable feature has created a growing interest in miniaturizing PA cells in recent years. In this paper, a simulation of a miniaturized H-type photoacoustic cell consisting of two buffer holes and a resonator was performed in order to detect CO, NH₃, NO, and CH₄ pollutants. These gases are the main components of the air pollutants that are produced by the automotive industry. The linear forms of the continuity, Navier–Stokes equations, and the energy equation were solved using the finite element method in a gaseous medium. The generated pressure could be measured by a MEMS sensor. Photoacoustic spectroscopy has proven to be a sensitive method for detecting pollutant gases. The objectives of the measurements were: determining the proper position of the pressure gauge sensor; measuring the frequency response; measuring the frequency response changes at different temperatures; studying the local velocity at the resonant frequency; and calculating the quality factor. The acoustic quality coefficient, acoustic response (pressure), local velocity, frequency response, and the effect of different temperatures on the frequency response were investigated. A frequency response measurement represents the fact that different gases have different resonance frequencies, for which CO and NO gases had values of 23.131 kHz and 23.329 kHz, respectively. The difference between these gases was 200 Hz. NH₃ and CH₄ gases with values of 21.206 kHz and 21.106 kHz were separable with a difference of 100 Hz. In addition, CO and NO gases had a difference of 2000 Hz compared to NH₃ and CH₄, which indicates the characteristic fingerprint of the designed cell in the detection of different gases. Better access to high-frequency acoustic signals was the goal of the presented model in this paper.

Keywords: automotive industry; gas monitoring; mesoscale; photoacoustic cell; MEMS; pollutants; air quality; contamination; greenhouse gas emissions; photonics



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1. Introduction

Nowadays, automotive industry pollution is considered a major threat to the planet. Concentrations of gases caused by human activities—for instance, transportation—lead to consequences from the local scale to the global scale that affect the climate, the environment,

and human health [1–4]. As a result of increasing the number of automobiles on the roads, the transportation sector has become the second most considerable consumer of energy and has a major share in the production of air pollutants and greenhouse gas emissions [5,6].

Greenhouse gases have far-ranging environmental and health effects. They cause climate change by trapping heat. Climate change can disrupt the match between organisms and their local environment, reducing survival and reproduction, and causing subsequent impacts on populations and species distributions [7–10].

Emissions of materials such as carbon monoxide, nitrogen oxides, and hydrocarbons caused by the transportation sector have increased significantly in comparison to previous decades [11]. Carbon monoxide is a toxic gas that endangers human health and causes suffocation. The toxicity of carbon monoxide causes serious problems when it enters the car cabin, which can be avoided by installing a gas leak detector in the car [12]. Nitrogen oxides are important components of air pollution; in addition to helping to form acid rain and the greenhouse effect, at higher levels they have a great impact on human health and can lead to diseases of the respiratory system and, consequently, life-threatening diseases [13]. Ammonia and methane are also important pollutants in the atmosphere, and increasing amounts of them are emitted by road traffic [14]. NO_x and CO are the two main pollutants emitted by vehicles, which cause serious environmental pollution as well as health problems and threats (Wang 2018) [15]. Scientists have found that global warming is related to an increment in carbon dioxide emissions, which is another result of gasoline combustion (Karol Tucki 2019) [16]. Optimizing gasoline formulations in order to ensure the best performance and minimal emissions of pollutants has always been a challenge for ages (I. Schifter 2020) [17]. Air contamination caused by energy consumption poses a significant ultimatum for the environment. The study of emission sources is of great importance for the improvement of environmental problems (Miao Zhang 2021) [18]. Identifying and quantifying the effects of gases plays a significant role in the development and improvement of gas-metering technology. One of the main methods of optical spectroscopy is photoacoustic spectroscopy, which utilizes sound and light to investigate materials in solid, liquid, and gas phases [19]. This method is considered an effective and inexpensive method for studying the properties of a substance [20]. Considering that transportation is responsible for most of the pollutants emitted by human resources, photoacoustic spectroscopy is an appropriate method for examining ambient gases due to meeting many needs related to gas detection, including: high sensitivity, spectral selection, and multi-component diagnosis [21]. Photoacoustic spectroscopy has been used to study greenhouse gases in recent years [22,23]. Currently, atmospheric gas sensors based on the principle of spectroscopy have become important, owing to their sensitivity and selectivity. These devices have the ability to quickly monitor and investigate pollutants, the presence in the atmosphere of which may have harmful consequences, such as acid rain, global climate change, and endangering human health [24]. Photoacoustic spectroscopy has various considerable benefits that differentiate this particular detection mechanism from other methods of absorption spectroscopy [25]:

- Photoacoustic spectroscopy does not need an optical detector. Accordingly, it has no choice for detecting wavelengths; therefore, it is possible to apply it to all wavelengths, from ultraviolet to infrared and terahertz.
- The zero-background feature lets the detector and preamplifier operate on a zero signal in a wide dynamic range.

These considerable advantages of photoacoustic spectroscopy have led to a competitive benefit over other absorption spectroscopy technologies. As shown in Figure 1, the photoacoustic effect on gases consists of four stages [26]: (i) modulation of the laser radiation (both amplitude and frequency) at a wavelength for which the target species shows high absorption; (ii) excitation of the target molecules by absorption of the incident radiation; (iii) non-radiative transfer of the absorbed radiation energy via collisions with other molecules, hence resulting in periodic heating at the modulation frequency;

and (iv) detection of the acoustic waves produced by the periodic heating with sensitive microphones or pressure sensors.

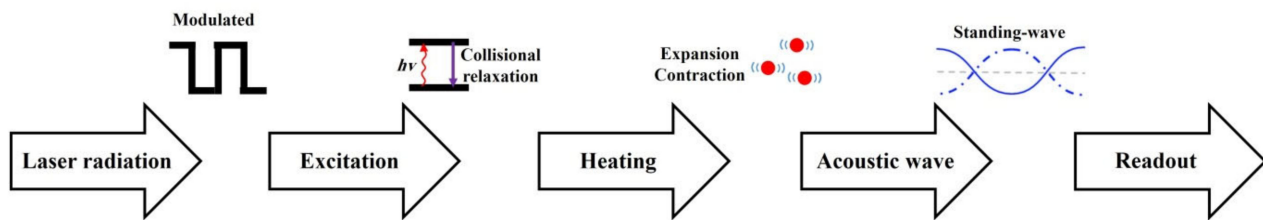


Figure 1. Schematic of the physical processes occurring in photoacoustic spectroscopy.

Advances in gas detection sensors for methane, propane, and other toxic gases in the automotive industry have been very promising. The use of detectors in cars in order to detect gas leakages has been considered due to an increase in mortality rates [27]. Optical gas sensors are powerful devices used to detect gases and monitor a variety of applications, such as environmental monitoring, industrial process control, medicine, and petrochemical industry applications [28]. Photoacoustic sensors use acoustic resonance to increase the signal level and improve the minimum concentration of detection [29]. In general, measures to control climate change or reduce air pollution have several advantages, comprising: improvements to energy efficiency, optimization of the environment, and improvements to social health and public benefits [30]. Here, the PS method was employed for the detection of pollutants in which a fingerprint feature was observed for gases. The investigation of toxic air pollutants emitted by vehicles was the main goal of this research. A miniaturized H-type photoacoustic cell was employed to the spectroscopy of CO, NH₃, NO, and CH₄ gases. The effects of important parameters such as the frequency response, the temperature, and the quality factor were numerically studied for the proposed system, and the results are presented in the Results and Discussion section.

2. Materials and Methods

2.1. Theory

This section explains the mathematical formulation of photoacoustic spectroscopy. The photoacoustic response was numerically investigated by solving the pressure field inside the cell, which is closely related to the acoustic waves produced by the photoacoustic effect [31,32]. Advances in numerical methods and the availability of user-friendly software allowing for the use of the finite element method (FEM) have paved the path for the direct solving of the Helmholtz equation and the optimization of the photoacoustic cell response [33]. Nowadays in general, it is accepted that the FEM is the most flexible numerical method [34]. The advantage of utilizing the FEM model is the potential to handle, in principle, any geometry of a photoacoustic cell. The essential equations for acoustics comprising thermal and viscous losses are applicable in the case where acoustic waves are produced by light absorption [35]:

$$\frac{\partial \rho}{\partial t} + \rho_0 \nabla \cdot \vec{u} = 0 \quad (1)$$

$$\rho_0 \frac{\partial \vec{u}}{\partial t} = -\nabla p + \left(\eta + \frac{4}{3} \right) \nabla \nabla \cdot \vec{u} - \mu \nabla \times \nabla \times \vec{u} \quad (2)$$

$$\rho_0 T_0 \frac{\partial s}{\partial t} = K \nabla^2 \tau + \rho_0 \epsilon \quad (3)$$

$$\epsilon = \epsilon_0 \exp\left(\frac{-r^2}{2\sigma^2}\right) \exp(i\omega t) \quad (4)$$

where the quantities $\rho, \vec{u}, p, \tau, s, \sigma, \eta, \mu$, and r represent the mass density, particle velocity, acoustic pressure, acoustic temperature, entropy, Gaussian spreading parameter, coefficient of shear viscosity, coefficient of bulk viscosity, and radial coordinate (the distance from the z -axis in the cylinders), respectively. Navier–Stokes linearized equations were used for modeling the sound wave propagation in the presence of viscous and thermal boundary layers. The harmonic changes were assumed to be negligible. In order to complete the partial differential equations, the continuity, motion, and energy conservation equations were used. As is obvious, the driving heat source was assumed to be of the Gaussian type. The heat source in a radial position respective to the center line of the resonator is expressed as Equation (4).

As discussed in [33,35], the produced heat in a PA cell by light absorption represents the source for a sound wave. Therefore, a corresponding source term has to be added to the Helmholtz equation:

$$\Delta^2 p(r, w) + k^2 p(r, w) = iw \frac{\gamma - 1}{c^2} H(r, w) \quad (5)$$

where p is the Fourier transform, $k = w/c$, and c is the sound velocity. By considering that the absorbing transition is not saturated, and that the modulation frequency is significantly smaller than the relaxation rate of the molecular transition, the relation $H(r, w) = aI(r, w)$ applies where $I(r, w)$ is the Fourier-transformed intensity of the electromagnetic field. It is known that the solution of the inhomogeneous wave equation can be indicated by Equation (6):

$$p(r, w) = \sum_j A_j(w) p_j(r) \quad (6)$$

The modes $p_j(r)$ and the corresponding eigenfrequencies $w_j = ck_j$ can be established by solving the homogeneous Helmholtz equation:

$$\Delta^2 p(r) + k^2 p(r) = 0 \quad (7)$$

2.2. Cell Design

The precision of a photoacoustic cell depends dramatically on its geometry. The H-type cell presented in the current study is simple in design and has two buffers that are connected to each other by a resonator. The buffer structure obviously affects the photoacoustic response. Selecting the dimensions of the buffer cylinders is a crucial issue in designing H-type photoacoustic cells [31,36]. Some advantages of investigating and using miniaturized photoacoustic cells are stated as follows:

- In photoacoustic spectroscopy, the signal is inversely related to the volume of the resonant cell, and this relation has led to the miniaturization of photoacoustic cells with great interest in recent years [37];
- Moreover, in photoacoustic spectroscopy, the detection limit is almost independent of the length of the laser–gas interaction, so this feature makes sensitivity and miniaturization possible in a photoacoustic spectroscopy system [38];
- Conventional photoacoustic gas analyzers mainly demand a large volume of the gas sample and are not suitable for detection applications in confined spaces [39].

Therefore, it is no wonder that a considerable portion of studies have focused on the design of miniaturized photo-acoustic sensors that can provide better portability and local detection without compromising the detection sensitivity. The cavity configuration was used because of the simple symmetry of the cylinder, which thoroughly matches the laser beam and propagates along the axis of the cylinder or one of its special modes, and also because of its easy processing [40]. The calculations were performed for a photoacoustic cell with a resonator length of $L_{res} = 5$ mm, a resonator radius of $R_{res} = 2.5$ mm, a buffer length of $L_{buf} = 2.5$ mm, and a buffer radius of $R_{buf} = 2.5 \times R_{res}$. The geometrical dimensions

of the parameters were chosen based on the MEMS-based microphone scales. Figure 2 represents a simulated photoacoustic cell.

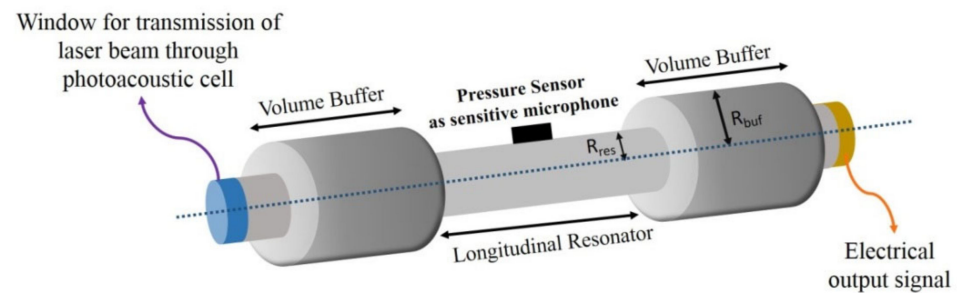


Figure 2. H-cell model as the photoacoustic cell.

2.3. Numerical Modeling

FEM studies are widely used to probe the behavior of forthcoming devices and their optimization [41]. The simulation of the photoacoustic response in a miniaturized H-type photoacoustic cell is discussed in this section. It should be noted that the FEM simulation of a PS cell as a gas detector was validated using the experimental results from [32]. Various environmental parameters have an impact on the signal of a gas sensor—for instance, pressure, mechanical vibration, background noise, electromagnetic fields, and temperature. As the laser beam passes through the windows of the photoacoustic cell, some amount of light may be absorbed by the windows, which can produce acoustic waves; these undesirable waves are called window noise. In this case, the length of the buffer cylinder should be $1/4$ acoustic wavelengths in order to minimize window noise [42]. The effects of mechanical vibrations or background noise are negligible at higher resonant frequencies because mechanical vibrations are usually at lower frequencies. Due to the optical readout, the effects of electromagnetic fields on the sensor signal are also negligible [43].

In general, two non-slip boundary conditions were applied to the velocity field and adiabatic conditions to temperature. These two conditions are known as standard boundary conditions for acoustic pressure. Standard boundary conditions of acoustic pressure were applied in order to prevent any thermal or adhesive boundary layers on the walls (layers are negligible) [44]. The other parameter that needed to be considered in the study of photoacoustic cells was the quality factor. The higher the loss rate, the lower the quality factor value [45]. The Q factor was determined as the ratio between the resonance frequency and the width of the resonance peak at half its resonance amplitude (FWHM). It was estimated using Equation (5) [46]:

$$Q = \frac{f_0}{f_h - f_l} \quad (8)$$

where f_0 represents the resonance frequency while f_l and f_h are the frequencies at which the value of the pressure amplitude has reduced to half of its value. $\Delta f (f_h - f_l)$ is the half width of the resonance peak. Therefore, Δf is also called the full width half maximum (FWHM) of the peak amplitude. The commercial FEM-based software COMSOL Multiphysics was utilized to model the miniaturized PA cell.

3. Results and Discussion

Due to the fact that vehicles are a major contributor to environmental pollution, in this study CO, NH₃, NO, and CH₄ gases were simulated. Then, the effect of various parameters was examined and analyzed when the miniaturized H-type photoacoustic cell was filled by these gases.

3.1. Acoustic Pressure

The acoustic pressure generated inside the cell was measured by a pressure sensor. Therefore, the first step in the study of gas was to detect the maximum amount of pressure in

the photoacoustic cell. This maximum pressure is where the resonant frequency occurs, or in other words, it is the most suitable place for the pressure sensor to be located. COMSOL Multiphysics software was used to evaluate the distribution of acoustic pressure inside the miniaturized photoacoustic cell filled by CO, NH₃, NO, and CH₄ gases. The results are presented schematically in Figure 3.

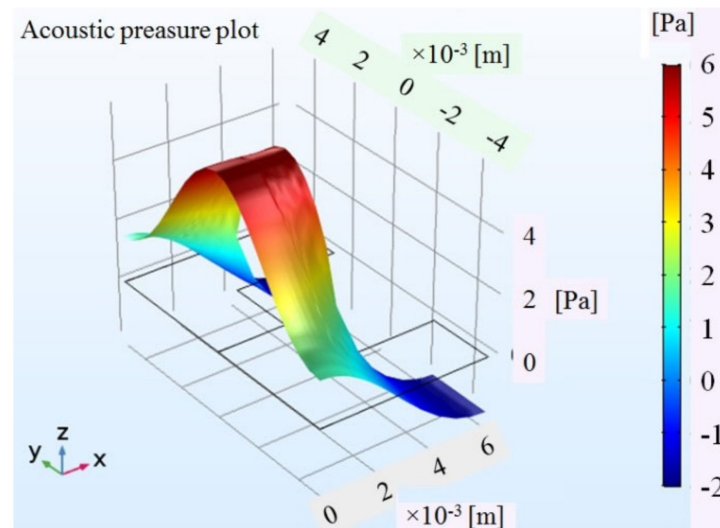


Figure 3. Acoustic pressure distribution corresponding to the first longitudinal state in a miniaturized acoustic cell.

Figure 3 represents the acoustic pressure distribution for the first excited longitudinal state, where the strongest acoustic pressure occurred in the middle of the resonant cavity. Hence, the first natural frequency was assumed to be the resonant frequency. The technical principle of PAS is to detect the energy of the acoustic wave generated during the molecular de-excitation process, which can be detected with a microphone, a quartz tuning fork (QTF), or a micro-cantilever. In 2021, a multicomponent gas sensor was proposed to detect CH₄, H₂O, CO₂, and C₂H₂ [47].

3.2. Frequency Response

Determining the frequency response of a photoacoustic cell is a common activity during cell design and examination [32]. The cell frequency response is examined in order to evaluate its suitability for specific measurements. A frequency analysis includes significant information for the design of a pressure sensor. The most important part of this information is determining the operation frequency of the sensor. The frequency of the acoustic wave, as stated in the previous section, is equal to the first natural frequency. Figure 4 shows the resonance frequencies for CO, NH₃, NO, and CH₄, which had values of 23.329, 21.206, 23.131, and 21.106 kHz, respectively.

3.3. Acoustic Velocity

Since the resonance frequency is directly related to the acoustic velocity, the local acoustic velocity at the resonant frequencies of the mentioned gases in Figure 4 were examined and the obtained results are shown in Figure 5. The r-coordinator in Figure 5 is the length of the resonator; the peaks are related to the frequency resonances of the resonators. The change in the peak of the frequency due to the change in acoustic velocity was different from the composition of the gas. It should be noted that the photoacoustic response of air components such as N₂, O₂, and CO₂ was also simulated: 18.101 kHz for CO₂; 23.684 kHz for N₂; and 22.112 kHz for O₂. Based on the results, these components have a frequency very different from the frequency of pollutants.

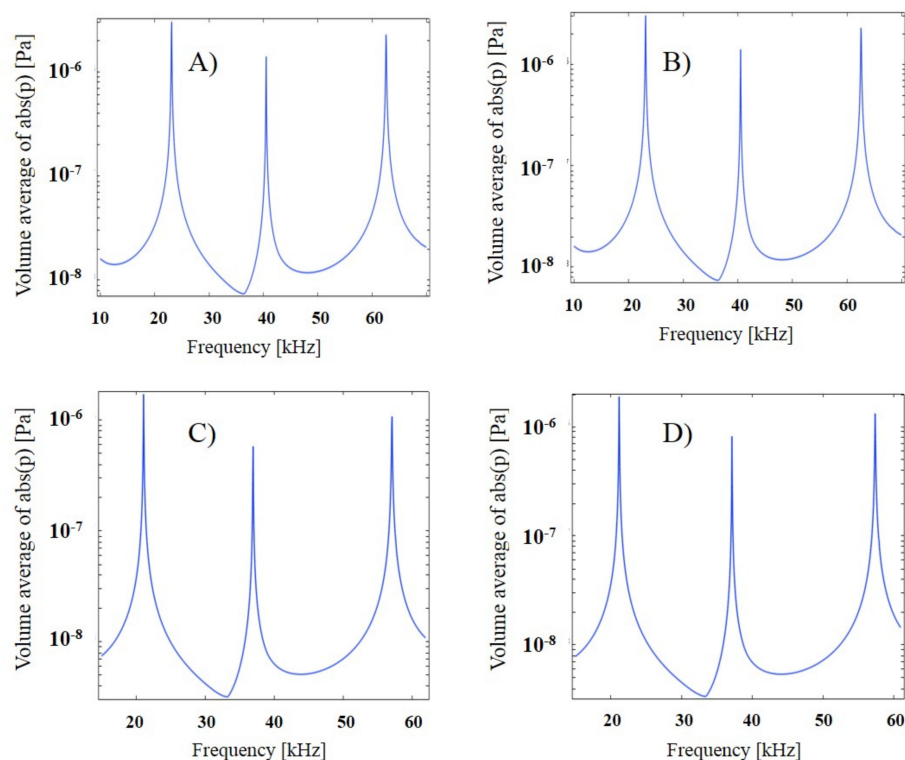


Figure 4. Frequency response; the cell filled by (A) NO, (B) CO, (C) CH₄, and (D) NH₃.

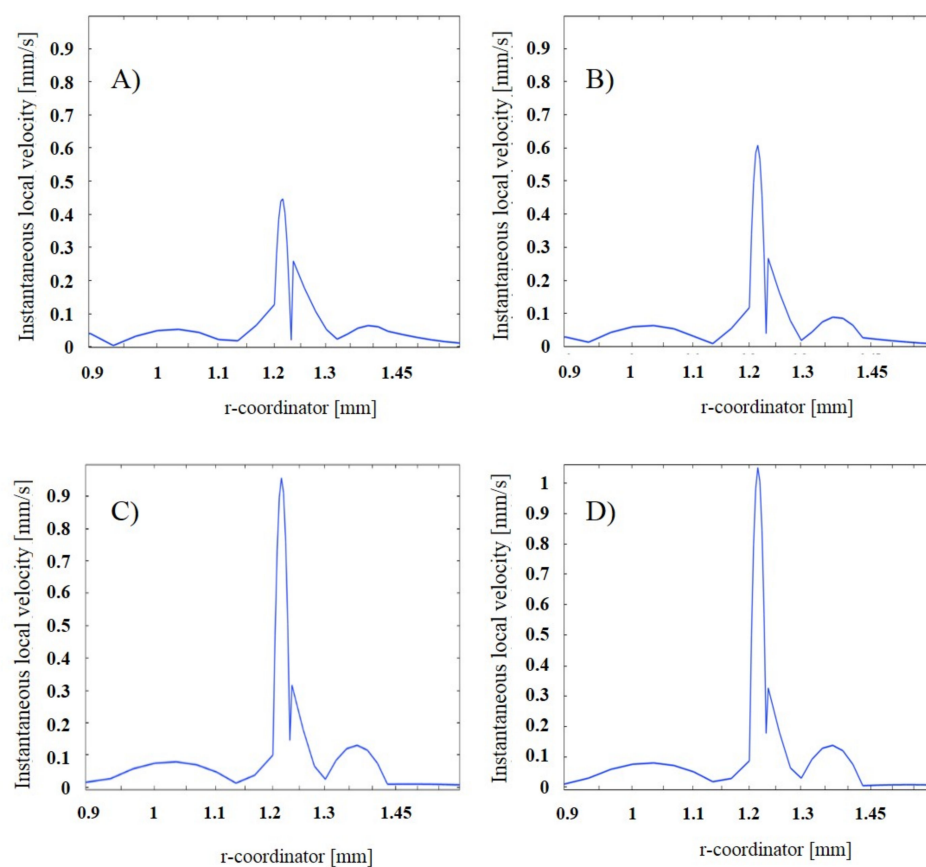


Figure 5. Instantaneous local velocity (mm/s), with the cell filled by (A) NO, (B) CO, (C) CH₄, or (D) NH₃.

The obtained results show that the local acoustic velocity at the resonance frequencies had a maximum value in the center of the cell. This indicates the function of the cell in resonance mode.

3.4. Frequency Response Changes with Temperature Changes

Temperature was one of the factors that affected the resonance frequency, as the acoustic velocity increased with temperature T [48,49]. Therefore, temperature changes could cause a change in the resonant frequency of PA cells. The effects of temperature on the sensor in the temperature range of 243 to 303 degrees Kelvin were studied specifically to investigate their effects on the frequency response. Figure 6 shows the rate of change in the frequency response for temperature change. Temperature monitoring was necessary to ensure that the PS cell was working in resonance mode.

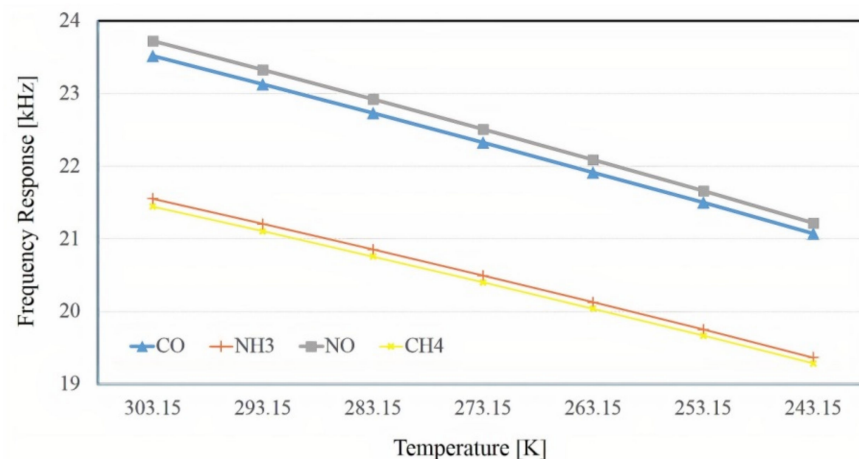


Figure 6. Frequency response verses temperature variations.

3.5. Quality Factor

The quality factor expresses the amount of energy stored in the resonator relative to the energy lost during a complete period [50,51]. Low losses increase the quality factor. A high quality factor guarantees the sensitivity of the detection. When the resonance peaks are small, it means that the quality factor is lower. Table 1 shows the results for the quality factors of CO, NH₃, NO, and CH₄ gases. The results indicate that miniaturized photoacoustic cells have a high sensitivity and performance. The mechanical noise was reduced by miniaturizing the resonator's diameter. Therefore, the QF increased significantly by decreasing the noise in the system. The achieved high QF indicates that the frequency response was highly sensitive for air pollutant gases.

Table 1. Comparison of QF.

Gas	First Natural Frequency (kHz)	Second Natural Frequency (kHz)	Third Natural Frequency (kHz)	Quality Factor
Nitric acid	23.329	40.891	63.158	33.32
Ammonia	21.206	37.157	57.4	30.19
Carbon monoxide	23.131	40.538	62.617	57.827
Methane	21.106	36.985	57.133	21.1

The laser photoacoustic spectroscopy method was an appropriate investigation technique which was capable of measuring gas concentrations at sub-ppb levels [52,53]. The smallest detectable concentration that could be detected with the system was calculated with the following equation:

$$C_{min} = \frac{V_n}{\alpha P_l C S_M} \quad (9)$$

where $V = V_n$ is the voltage of the photoacoustic signal for a signal-to-noise ratio equal to 1 (SNR = 1), α ($\text{cm}^{-1} \cdot \text{atm}^{-1}$) is the absorption coefficient for a given laser line, P_l (W) is the

unmodulated peak value of the power laser, C (Pa.cm.W⁻¹) is the cell constant, and S_M (mV/Pa) is the total responsivity of the microphones.

4. Conclusions

In this paper, a simulation of a miniaturized photoacoustic cell is presented in order to investigate pollutants produced by the automotive industry. The main advantage of this cell is its size, which facilitates the generation of a proper resonance frequency and quality factor. The PAS method is employed for the detection of pollutants in which a fingerprint feature is observed for gases. The proposed cell has an acoustic response in the center of the resonator, which is the maximum pressure for different gases at different resonant frequencies, and the values for NO, NH₃, CO, and CH₄ were 23.329, 21.206, 23.131, and 21.106 kHz, respectively. It should be noted that frequency measurements are a common and simple electronic way to measure different mechanical variations, and 100–200 Hz is significant and can be easily measured by MEMS-based oscillators. The dependence of the frequency response on the temperature, as well as the effects of local changes in velocity on the resonance frequency, were investigated to ensure the cell performance in the resonance state. It was shown that the eigenfrequency of photoacoustic cells can be defined by the finite element method. The obtained results in the study of CO, NH₃, NO, and CH₄ gases had excellent measured frequencies. The suggested cell shows a remarkable QF for the detection of air pollutants. The results show a high potential for miniaturizing photoacoustic cells to expand sensitive laser sensors. The results are useful and well-organized for energy science and fuel cell studies.

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