



Article Determination of Cordycepin Using a Stability-Indicating Greener HPTLC Method

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Abstract: A wide range of analytical techniques have been reported to determine cordycepin (CDN) in various sample matrices. Nevertheless, greener analytical approaches for CDN estimation are scarce in the literature. As a result, this study was designed to develop and validate a stabilityindicating greener "high-performance thin-layer chromatography (HPTLC)" technique for CDN determination in a laboratory-developed formulation. The greener eluent system for CDN detection was ethanol–water (75:25 v/v). At a wavelength of 262 nm, CDN was measured. The greenness scale of the proposed analytical technology was derived using the "Analytical GREENness (AGREE)" approach. The proposed stability-indicating HPTLC assay was linear for CDN analysis in the 50–1000 ng/band range with a determination coefficient of 0.9978. The proposed analytical technique for CDN analysis was simple, rapid, accurate, precise, robust, selective, stability-indicating, and greener. The AGREE score for the proposed stability-indicating HPTLC technique was calculated to be 0.79 using the AGREE calculator. The current protocol was able to detect CDN degradation products under various stress conditions, indicating its stability-indication characteristics and selectivity. The AGREE quantitative score indicated that the stability-indicating current protocol had outstanding greener characteristics. The amount of CDN in the laboratory-developed formulation was determined to be 98.84%, indicating the suitability of the current protocol in the assay of CDN in the formulations. These results suggested that CDN in a laboratory-developed formulation may be regularly determined using the stability-indicating greener HPTLC strategy.

Keywords: AGREE; cordycepin; greener HPTLC; validation

1. Introduction

Cordycepin (CDN) is a derivative of nucleoside adenosine, which is chemically known as 3'-deoxyadenosine (Figure 1) [1]. It was isolated from *Cordyceps militaris* for the very first time, but it is also isolated from some other species of *Cordyceps*, such as *Cordyceps sinensis* [2,3]. Adenosine and CDN have been reported as the main phytoconstituents of *C. militaris* [4]. In the literature, CDN has been reported to have several therapeutic activities, which include antioxidant [5], antimicrobial [6], antibacterial [7], antifungal [6], anti-inflammatory [8], neuroprotective [9], lung-protective [10], hepatoprotective [11], immunomodulatory [12], prosexual [13], and anticancer activities [14,15]. CDN is found in various pharmaceutical



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). formulations, food supplements, and mushroom products. As a consequence, both qualitative and quantitative methods must be used to identify CDN in a wide range of items, including food and pharmaceuticals.



Figure 1. Molecular structure of cordycepin (CDN).

A detailed literature survey revealed a wide range of analytical procedures for CDN determination in foods, pharmaceuticals, and dietary supplements. For the determination of CDN in various species of Cordyceps and dietary supplements, various "high-performance liquid chromatography (HPLC)" techniques have been developed and validated [16-23]. A few liquidchromatography mass-spectrometry (LC-MS) methods have also been developed to determine CDN in *C. militaris* and *C. sinensis*, either alone or in combination with other nucleosides [24,25]. An eco-friendly/greener LC-MS method has also been used to simultaneously determine CDN and iso-CDN in ten different kinds of Cordyceps [26]. An online solid-phase extraction (SPE)-HPLC technique was also developed for the simultaneous detection of CDN and 2'deoxyadenosine in C. militaris, C. sinensis, Hirsutella sinensis, and C. sobolifera [27]. Some column chromatography approaches have also been used to determine CDN in C. militaris and some other species of Cordyceps [28,29]. The separation and quantification of CDN and adenosine in *C. sinensis* have also been accomplished using a gradient ultra-high-performance liquid chromatography (UPLC) technique [30]. An UPLC approach coupled with mass spectrometry (UPLC-MS) was also used for the simultaneous determination of CDN and its metabolite 3'-deoxyinosine in a rat whole-blood sample and for their pharmacokinetic assessment [31]. Additionally, some "high-performance thin-layer chromatography (HPTLC)" techniques have been created to identify CDN in different types of Cordyceps, either alone or in combination with other adenosines [32–34]. Other techniques such as capillary electrophoresis [35], near-infrared spectroscopy [36], and nuclear magnetic resonance spectroscopy [37] have also been developed for the rapid determination of CDN in *C. militaris*.

The literature contains a wide variety of analytical techniques for CDN determination. However, neither a qualitative nor a quantitative assessment of the greenness profile of the reported analytical processes has been made. In addition, most of the assays of CDN analysis are based on liquid chromatography techniques, which are sensitive and accurate for CDN analysis, but these techniques are time-consuming, labor-intensive, complicated, expensive, and result in large amounts of solvent waste being added to the environment. So far, there are no reported greener HPTLC assays for CDN. Several qualitative and quantitative approaches are reported to forecast the greenness of analytical tests [38–42]. To forecast greenness characteristics, on the other hand, the "Analytical GREENness (AGREE)" technology takes into account all twelve green analytical chemistry (GAC) principles [40]. In order to determine the greenness profile of the current analytical protocol, the AGREE approach was used [40].

Based on all of these presumptions, this study's goal was to design and validate a stability-indicating (a method able to detect an analyte in the presence of degradation products), greener, reverse-phase HPTLC technique for CDN analysis using a formulation that the laboratory produced. The term greener HPTLC is a more suitable term for the current method due to the use of green solvents (ethanol and water) compared to methods in the literature. Therefore, we used the term greener HPTLC in this work. The current protocol for CDN detection was proven efficient using "The International Council for Harmonization (ICH)"-Q2-R1 guidelines [43].

2. Materials and Methods

2.1. Materials

The reference standard of CDN was provided by "Beijing Mesochem Technology Co., Ltd. (Beijing, China)". The chromatography grade ethanol and methanol were obtained from "E-Merck (Darmstadt, Germany)". The high pure water was collected from a "Milli- $Q^{\text{(B)}}$ (Milli-Q, Lyon, France)" unit. The nanoemulsion formulation of CDN was prepared via the aqueous-phase titration method using Triacetin (oil phase), Tween-80 (surfactant), Carbitol (cosurfactant), and water (aqueous phase). All additional reagents were of the analytical grade and procured from E-Merck (Darmstadt, Germany).

2.2. Instrumentation and Chromatographic Procedures

A "CAMAG HPTLC (CAMAG, Muttenz, Switzerland)" system was used for the determination of nanoemulsions of CDN. In order to apply the samples as 6 mm bands, a "CAMAG Automatic TLC Sampler 4 (ATS4) Sample Applicator (CAMAG, Geneva, Switzerland)" was used. The TLC plates used for the detection of CDN were "RP-60F254S plates (E-Merck, Darmstadt, Germany)". The "Hamilton microliter syringe (Hamilton, Bonaduz, Switzerland)" was loaded on to the sample applicator. Throughout the course of the experiment, the application rate for CDN detection remained constant at 150 nL/s. The TLC plates were set up at an 80 mm distance in a "CAMAG automated developing chamber 2 (ADC2) (CAMAG, Muttenz, Switzerland)". The greener eluent system was composed of the binary combination of ethanol and water (75:25 v/v). The greener eluent system's vapors were used to saturate the chamber for 30 min at 22 °C. At a wavelength of 262 nm, CDN was quantified. The slit size was adjusted to 4 × 0.45 mm², and the scan speed "(CAMAG TLC Scanner, Muttenz, Switzerland)" was set to 20 mm/s. For each procedure, three or six copies were applied. The "WinCAT's (version 1.4.3.6336, CAMAG, Muttenz, Switzerland)" program was used for the processing and interpretation of data.

2.3. Calibration Curve and Quality Control (QC) Sample for CDN

The required amount of CDN was dissolved into the greener eluent system's prescribed volume to obtain the CDN stock solution with the final concentration of 100 μ g/mL in triplicate (n = 3). From the stock solution, serial dilutions were made to obtain ten different CDN concentrations, which were 50, 100, 200, 300, 400, 500, 600, 700, 800, and 1000 ng/band (range = 50–1000 ng/band), in triplicate (n = 3). Approximately 200 μ L of each concentration was spotted onto reverse-phase TLC plates, and the respective chromatographic response was recorded. The CDN calibration plot was produced by graphing the CDN concentrations vs. the observed peak area (n = 6). Three distinct QC concentrations were freshly created for the evaluation of several parameters for validation studies.

2.4. Sample Processing for the Assay of CDN in Laboratory-Developed Nanoemulsion

Commercial formulations of CDN are not available in the Saudi Arabian market. As a result, a nanoemulsion formulation of CDN was developed and evaluated in the laboratory. The nanoemulsion formulation of CDN was prepared via the aqueous-phase titration method [44]. The components of the nanoemulsion were optimized by constructing a pseudo-ternary phase diagram [44,45]. The accurately weighed 10 mg of CDN was dissolved in 150 μ L of Triacetin (oil phase). Then, 250 μ L of Tween-80 (surfactant) and 250 μ L of Carbitol (cosurfactant) were added to the aqueous phase via vortexing. To this mixture, 350 μ L of water (aqueous phase) was added drop by drop until the clear and transparent nanoemulsion was obtained. One mL of a laboratory-developed nanoemulsion

with 10 mg/mL of CDN was appropriately diluted with methanol to produce 100 mL of stock solution in order to determine the CDN content. Following a suitable dilution with the greener eluent system and a 15-min sonication of this solution, the CDN content was determined using the current protocol.

2.5. Analytical Method Validation

The current protocol for CDN measurement was validated for distinct parameters according to ICH-Q2-R1 procedures [43]. By plotting the CDN concentrations vs. the measured peak area, CDN linearity was derived. The linearity of the current protocol for CDN was assessed in the 50–1000 ng/band range in six copies (n = 6).

For the current protocol of CDN analysis, the computation of the retardation factor (R_f), asymmetry factor (As), and theoretical plates number/meter (N/m) was employed to find the system appropriateness characteristics. The reported equations were used to obtain the " R_f , As, and N/m" values [42].

The intra-day and inter-day accuracy of the current protocol for CDN analysis was assessed using the % recovery approach. Six replicates (n = 6) of solutions were performed on the same day to examine the intra-day accuracy at three different QC concentrations, namely low QC (LQC = 100 ng/band), middle QC (MQC = 500 ng/band), and high QC (HQC = 1000 ng/band). On three distinct days, six replicates (n = 6) of CDN's LQC, MQC, and HQC solutions were used to determine inter-day accuracy. The percentage recovery for intra-day and inter-day accuracy was computed for the interpretation of the results.

The precision of the current protocol for CDN was determined as intra-day and interday variation. The intra-day variation for CDN was examined using the analysis of freshly produced CDN solutions at LQC, MQC, and HQC on the same day in six replicates (n = 6). The CDN inter-day precision was found by evaluating freshly generated CDN solutions at the same QC levels on three distinct days in six repetitions (n = 6). Both precisions were expressed in terms of the percentage of the relative standard deviation (%RSD).

By purposefully altering the composition of the greener eluent system, the CDN robustness was evaluated for the current protocol. The greener eluent system of ethanol–water (75:25, v/v) for CDN was altered to ethanol–water (77:23, v/v) and ethanol–water (73:27, v/v) for the current protocol, and the differences in the CDN chromatographic peaks and R_f were noted.

The sensitivity of the current protocol for CDN was calculated as the "limit of detection (LOD) and limit of quantification (LOQ)" by applying a standard deviation methodology. For the current protocol, the blank sample was analyzed in six replicates (n = 6), and the standard deviation was calculated. The LOD and LOQ for CDN were derived using the standard deviation method in six replicates (n = 6) using the following equations [46]:

$$LOD = 3.3 \times \frac{\sigma}{S}$$
(1)

$$LOQ = 10 \times \frac{\sigma}{S}$$
(2)

where σ is the standard deviation of the blank sample and S is the slope of the CDN calibration curve.

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In order to assess the specificity of the current protocol for CDN analysis, the R_f values and UV-absorption spectra of CDN in a laboratory-developed nanoemulsion formulation were compared to those of standard CDN.

2.6. Forced Degradation/Selectivity Studies

To investigate the selectivity and stability-indicating capabilities of the current protocol, forced degradation studies under a variety of stress conditions, including acidic (HCl), basic (NaOH), oxidative (H_2O_2), thermal, and photolytic stress conditions, were carried out [42,47]. The MQC concentration (500 ng/band) of CDN was freshly prepared into the greener eluent system for acid and base-induced degradation. By mixing 4 mL of

1M HCl and 4 mL of 1M NaOH into an aliquot (1 mL) of this solution, acid and base hydrolyses were applied. For the determination of CDN in the presence of its acid and base degradation products, these mixtures were refluxed for 48 h at 60 °C before being examined using the current protocol [42].

The MQC concentration (500 ng/band) of CDN was freshly prepared and introduced into the greener eluent system for oxidative degradation examination. This solution was oxidatively degraded by adding 2 mL of 30% H_2O_2 to an aliquot (1 mL) of it. For the detection of CDN in the presence of its oxidative degradation products, this mixture was refluxed for 48 h at 60 °C before being examined using the current protocol [42].

For the purpose of conducting thermal degradation tests, an aliquot of the MQC concentration (500 ng/band) was placed in a hot air oven for 48 h at 60 °C. The determination of CDN in the presence of its thermal degradation products was then performed utilizing the current protocol [42].

For photolytic degradation examinations, an aliquot of the MQC concentration (500 ng/band) was subjected to a UV chamber at 254 nm for 48 h. Then, CDN was determined using the current protocol while the photolytic degradation products were present [42].

2.7. Application of Current Protocol in the Determination of CDN in Laboratory-Developed Nanoemulsion

The processed samples of the laboratory-developed nanoemulsion formulation were spotted to reverse-phase TLC plates for the current protocol, and the peak areas of CDN in three copies (n = 3) were recorded. For the current protocol, the percent assay of CDN in the laboratory-developed nanoemulsion formulation was calculated using a CDN calibration plot.

2.8. Assessment of Eco-Friendliness of Protocol

The AGREE technique was used to derive the greenness/eco-friendliness of the current protocol for CDN determination [40]. Using "AGREE: The Analytical Greenness Calculator (version 0.5, Gdansk University of Technology, Gdansk, Poland, 2020)", the AGREE index (0.0–1.0) for the current protocol was obtained. The detailed AGREE protocol is included in the Supplementary Materials.

3. Results and Discussion

3.1. Analytical Method Development

The goal of the present study was to develop a stability-indicating greener HPTLC method to determine CDN. As a result, only greener solvents such as ethanol and water were examined as the eluent system. Toxic solvents such as methanol and acetonitrile were not examined due to their toxicity. Different concentrations of ethanol and water, including ethanol–water (35:65, v/v), ethanol–water (45:55), ethanol–water (55:45), ethanol–water (65:35), ethanol–water (75:25), ethanol–water (85:15), and ethanol–water (95:5, v/v), were examined as the greener eluent systems in order to establish a suitable band for CDN determination. All greener eluent systems were developed using the chamber saturation conditions. A representative TLC chromo-plate for the standard CDN, laboratory-developed nanoemulsion formulation, and degradation samples is shown in Figure 2.

It was discovered that the greener eluent systems, including ethanol–water (35:65, v/v), ethanol–water (45:55, v/v), ethanol–water (55:45, v/v), ethanol–water (65:35, v/v), ethanol–water (85:15, v/v), and ethanol–water (95:5, v/v), produced undesirable CDN chromatograms with undesirable As values (As > 1.15) when investigated. It was observed that the greener ethanol–water eluent system (75:5, v/v) produced a well-resolved and unbroken CDN chromatogram at $R_f = 0.77 \pm 0.02$ (Figure 3A) when investigated. Additionally, CDN was discovered to have an acceptable As value of 1.02 for CDN analysis. As a result, the final greener eluent system for the current protocol of CDN detection was chosen to be ethanol–water (75:25, v/v). The greatest chromatographic response for CDN was discovered at 262 nm when the spectral bands for CDN were calculated in densitometry



mode. Thus, 262 nm was used as the detection wavelength for all of the CDN investigations.

Figure 2. A representative thin-layer chromatographic (TLC) chromo-plate of standard CDN, formulation, and forced degradation study samples recorded using greener ethanol–water (75:25 v/v) eluent system for current protocol.



Figure 3. Representative chromatograms of (A) standard CDN and (B) CDN nanoemulsion for current protocol.

3.2. Analytical Method Validation

To obtain numerous validation parameters for CDN measurement, the ICH-Q2-R1 procedures were followed [43]. Table 1 summarizes the data from the linear regression analysis of the CDN calibration curve for the current protocol. The CDN calibration curve for the current protocol was linear in the 50–1000 ng/band range. In densitometry techniques, such as TLC and HPTLC, samples are spotted as bands. Hence, the most convenient unit for densitometry analysis is ng/band or μ g/band. Due to this reason, we used ng/band concentrations in this work. For the current protocol, CDN's determination coefficient (R²) and regression coefficient (R) were found to be 0.9978 and 0.9988, respectively. The results pointed to a strong correlation between the measured peak area and the CDN concentrations. The linearity of the current protocol for CDN analysis over a wider range of concentrations was demonstrated by these results.

The system appropriateness parameters for the current protocol are listed in Table 2. The current protocol's R_f , As, and N/m for CDN determination were found to be 0.77, 1.02, and 5321, respectively, which were reliable for CDN analysis.

Parameters	Value	
Linearity range (ng/band)	50–1000	
Regression equation	y = 11.718x + 294.98	
R^2	0.9978	
R	0.9988	
SE of slope	0.27	
SE of intercept	1.29	
95% CI of slope	10.53-12.91	
95% CI of intercept	289.41-300.54	
$LOD \pm SD$ (ng/band)	16.92 ± 0.14	
$LOQ \pm SD$ (ng/band)	50.76 ± 0.42	

Table 1. Resulting data for the linearity of cordycepin (CDN) for current protocol (mean \pm SD; n = 6).

R²: determination coefficient; R: regression coefficient; SE: standard error; CI: confidence interval; LOD: limit of detection; LOQ: limit of quantification.

Table 2. System appropriateness parameters of CDN for current protocol (mean \pm SD; n = 3).

Parameters	Values
R _f	0.77 ± 0.02
As	1.02 ± 0.01
N/m	5321 ± 4.92

R_f: retardation factor; As: asymmetry factor; N/m: theoretical plates number/meter.

The current protocol of the CDN measurement's intra-day and inter-day accuracy was calculated in terms of % recovery. The accuracy information regarding the current protocol is illustrated in Table 3. The intra-day % recoveries of CDN at three distinct QC levels were determined to be 98.24–101.14% using the current protocol. The inter-day % recoveries of CDN at three distinct QC levels were found to be 98.45–101.68% for the current protocol. These data suggested that the current protocol was accurate in the determination of CDN.

Table 3. Intra-day and inter-day accuracy data of CDN for current protocol (mean \pm SD; n = 6).

Conc. (ng/band)	Conc. Found (ng/band) \pm SD	Recovery (%)	RSD (%)
	Intra-day accuracy		
100	101.14 ± 1.17	101.14	1.15
500	491.23 ± 4.02	98.24	0.81
1000	988.71 ± 7.57	98.87	0.76
	Inter-day accuracy		
100		98.45	1.20
500	508.41 ± 4.32	101.68	0.84
1000	992.54 ± 7.87	99.25	0.79

The data for CDN analysis are expressed in terms of % RSD, and the intra-day and inter-day variation in the current protocol was investigated. The findings regarding the intra-day and inter-day variations for the current protocol of CDN analysis are shown in Table 4. The current protocol was shown to have a 0.77–1.01% RSD of CDN for intra-day precision. The current protocol was shown to have an RSD of CDN for the inter-day precision of 0.80–1.12%. The findings revealed the precision of the current protocol for CDN analysis.

The robustness of the current protocol for CDN analysis was evaluated by intentionally altering the composition of the greener eluent system. The findings regarding the data of robustness measurement for the current protocol are illustrated in Table 5. CDN % RSD for the current protocol was estimated to be 0.80–0.86%. CDN R_f values were found to be 0.76–0.78 for the current protocol. The findings revealed that the current protocol for CDN measurement was robust.

	Intra Day Provision			Inter Day Presiden		
Conc. (ng/band)	Conc. (ng/band) \pm SD	SE	RSD (%)	Conc. (ng/band) \pm SD	SE	(%) RSD
100	99.13 ± 1.01	0.41	1.01	98.02 ± 1.10	0.44	1.12
500	488.74 ± 3.89	1.58	0.79	486.32 ± 3.98	1.62	0.81
1000	1012.31 ± 7.86	3.20	0.77	987.41 ± 7.94	3.24	0.80

Table 4. Determination of CDN intra-day and inter-day fluctuation for current protocol (mean \pm SD; n = 6).

Table 5. Measurement of CDN robustness for current protocol (mean \pm SD; n = 6).

Conc. (ng/band)	Mobile Phase Composition (Ethanol–Water)			Results		
	Original	Used	Level	Conc. (ng/band) \pm SD	RSD (%)	R _f
		77:23	+2.0	489.31 ± 3.92	0.80	0.76
500	75:25 75 73	75:25	0.0	498.45 ± 4.28	0.85	0.77
		73:27	-2.0	508.94 ± 4.42	0.86	0.78

To evaluate the sensitivity of the current protocol for CDN analysis, the "LOD and LOQ" were computed. Table 1 summarizes the resulting values of "LOD and LOQ" of CDN for the current protocol. According to the results, the LOD and LOQ of CDN were computed to be 16.92 \pm 0.14 and 50.76 \pm 0.42 ng/band, respectively. The results demonstrated that the current protocol was sensitive for CDN measurement. The comparative evaluation of the Rf values and UV-absorption spectra of CDN in the laboratory-developed nanoemulsion with that of standard CDN allowed to assess the specificity of the current protocol for CDN assessment. The combined UV-absorption spectra of standard CDN and CDN in a laboratory-generated nanoemulsion are illustrated in Figure 4 for comparison. The peak response of standard CDN and the laboratory-developed nanoemulsion was measured at 262 nm. The initial peaks of the UV absorption spectra of the standard and nanoemulsion sample were not overlapped. However, the main peaks of CDN in both the samples were overlapped. The sample nanoemulsion contained some excipients. Due to the presence of some excipients, the initial peaks were not overlapped. The specificity of the current protocol for CDN assessment was demonstrated by finding the identical UV-absorption spectra, R_f values, and detection wavelengths of CDN in the standard and laboratorydeveloped nanoemulsion.



Figure 4. UV-absorption spectra of standard CDN and formulation, superimposed.

3.3. Selectivity and Forced Degradation Studies

The selectivity and degradation of the current protocol were investigated under various stress conditions. The outcomes of the current protocol are shown in Figure 5 and Table 6. The CDN peaks in the chromatographic peaks from degradation circumstances were clearly distinguished from other peaks of degradation products (Figure 5). Because there was zero percent of CDN left after acid degradation, the entire amount of CDN (100% of it) was degraded (Table 6 and Figure 5A). As a result, under acid degradation conditions, CDN was totally decomposed. Peaks 1 and 2 in Figure 5A, which represent the acid-induced degradation peaks, were separated by R_f values of 0.69 and 0.86, respectively. Under the base degradation scenario, CDN remained at 81.14 percent, while 18.86 percent was degraded (Table 6 and Figure 5B). As a result, CDN was stable enough to withstand base deterioration. Peaks 1, 2, and 4 in Figure 5B, which represent the base-induced degradation peaks, were separated by R_f values of 0.21, 0.69, and 0.82, respectively. Under base degradation conditions, CDN's R_f value was unchanged ($R_f = 0.77$). In total, 43.20 percent of CDN was decomposed, while 56.80 percent was still undergoing oxidative destruction (Table 6 and Figure 5C). Peaks 1, 2, 3, and 5 in Figure 5C, which were caused by H_2O_2 , were separated by R_f values of 0.48, 0.62, 0.69, and 0.82, respectively. Under oxidative degradation conditions, the R_f value of CDN was slightly shifted ($R_f = 0.76$). During thermal and photolytic degradation conditions, CDN was kept at 100.00 percent, and no deterioration was seen (Figure not shown). Because of this, CDN had a high level of resistance to thermal and photolytic deterioration. The resolution of CDN in Figure 5 is fine. However, the resolution of the degradation components is low. The current protocol was developed for CDN instead of its degradation products. Forced degradation studies were performed to evaluate the ability of current protocol to determine CDN in the presence of its degradation products. In this study, the CDN was successfully separated in the presence of its degradation products. Hence, the methodology was selective and stability-indicating. Using the current protocol, the maximum CDN degradation under acid degradation conditions was found. All of these results demonstrated that CDN can be detected by utilizing the current protocol in the presence of its degradation products. These findings and observations suggested that the current protocol had selectivity and stability-indicating properties.



Figure 5. Representative densitometry chromatograms of CDN derived under (**A**) acid-, (**B**) alkali-, and (**C**) oxidative-induced degradation of CDN.

Stress Condition	Number of Degradation Products (R _f)	CDN R _f	CDN Remaining (ng/band)	CDN Recovered (%)
1M HCl	2 (0.69, 0.86)	ND	0.00	0.00 ± 0.00
1M NaOH	3 (0.21, 0.69, 0.82)	0.77	405.70	81.14 ± 1.97
30% H ₂ O ₂	4 (0.48, 0.62, 0.69, 0.82)	0.76	284.00	56.80 ± 1.38
Photolytic	0	0.77	500.00	100 ± 0.00
Thermal	0	0.77	500.00	100 ± 0.00

Table 6. Results of forced degradation evaluation of CDN at various stress conditions for current protocols (mean \pm SD; n = 3).

ND: not detected.

3.4. Application of Current Protocol in the Determination of CDN in Laboratory-Developed Nanoemulsion

For the pharmaceutical assay of CDN in the laboratory-developed nanoemulsion, the current protocol was used as the alternative strategy to conventional analytical technologies. By contrasting the TLC band at $R_f = 0.77 \pm 0.02$ for CDN with standard CDN utilizing the current protocol, the chromatogram of CDN from the laboratory-developed nanoemulsion formulation was confirmed. The CDN peak in the laboratory-developed nanoemulsion had the same chromatographic profile as standard CDN when employing the current protocol. Furthermore, no additional peaks of nanoemulsion components were detected in the laboratory-developed formulation (Figure 3B), suggesting no interaction between CDN and formulation components. A CDN calibration plot was utilized to derive its content for the current protocol. Utilizing the current protocol, the % content of CDN in the laboratory-developed nanoemulsion was determined to be 98.84%. These results indicated the suitability of the current protocol for a CDN pharmaceutical assay.

3.5. Assessment of Eco-Friendliness of Protocol

Different qualitative and quantitative technologies are used to assess the greenness/ecofriendliness of analytical methods [38–42]. However, only AGREE uses all twelve GAC criteria to determine the eco-friendliness profile [40]. As a result, this approach was applied to gauge the analytical strategy's greenness profile. A typical diagram for the AGREE index of the current protocol is illustrated in Figure 6. The AGREE sheet for each component of GAC is presented in Figure S1. For the current protocol, the AGREE score was determined to be 0.79, indicating that the current protocol for CDN measurement had an outstanding greenness profile.



Figure 6. The representative pictogram for AGREE scale for the current protocol derived using AGREE calculator.

4. Conclusions

Greener analytical protocols for CDN measurement are scarce in the literature. Therefore, in order to design and validate a stability-indicating greener HPTLC protocol for CDN detection in a laboratory-developed nanoemulsion formulation, this study was carried out. The current protocol of CDN measurement is simple, accurate, precise, rapid, robust, selective, greener, and stability-indicating. The AGREE results suggested an outstanding greenness profile of the current protocol. The current protocol was able to detect CDN in the presence of its degradation products, indicating the selectivity and stability-indicating nature of the current protocol. These results indicated that the current protocol can be used to determine CDN in formulations.

Supplementary Materials: The following supporting information can be downloaded at: https:// www.mdpi.com/article/10.3390/separations10010038/s1. Figure S1: AGREE score sheet presenting AGREE score for twelve different components of green analytical chemistry for the current protocol of CDN analysis derived using AGREE calculator. The detailed AGREE protocol is also included in the Supplementary Materials.

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