

Article

# In-Situ Heat Treatment Study on the Nanocrystalline Cr<sub>2</sub>O<sub>3</sub> Film Using an Environmental Scanning Electron Microscope

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Academic Editor: Mark D. Soucek

Received: 9 November 2017; Accepted: 7 December 2017; Published: 8 December 2017

**Abstract:** In this work, the surface morphology changes of nanocrystalline Cr<sub>2</sub>O<sub>3</sub> film deposited on Si wafer during the heating process were observed in-situ by means of an environmental scanning electron microscope (ESEM). The Cr<sub>2</sub>O<sub>3</sub> film cracked at high temperature due to the cause of thermal stress; the corresponding crack area percentages on the film surface were real-time evaluated using image analysis software (SISC IAS V8.0) based on the principle of gray value analysis. In the meantime, the effects of the heating temperature on the crack area percentage, phase constituents, and grain size of the Cr<sub>2</sub>O<sub>3</sub> film were also studied in detail. The results showed that the percentage of crack area on film surface first increased with the heating temperature rise, and reached the maximum value at around 980 °C, and then gradually declined again. The above trend is closely related to the changes of thermal stress and grain growth in film. In addition, the heat treatment also had a strong influence on the grain size of the Cr<sub>2</sub>O<sub>3</sub> film.

**Keywords:** Cr<sub>2</sub>O<sub>3</sub> film; arc ion plating; in-situ observation; crack area percentage; ESEM; grain size

## 1. Introduction

In the past decades, the field of thin film materials has been developing rapidly, with more new types of films, and improved film quality. Thin film is not only an important asset for daily life services, but also for industrial progress [1]. Chromium oxide thin film is a very interesting research object in the field of thin film application. Because of its low activity in the structure, it is the most stable in all of the homogeneous crystals, so it has very good electrical performance and mechanical properties, which make it applied widely [2–5]. With the relatively low friction coefficient, high temperature resistance, stable chemical properties and so on, it has been used as a protection material for microelectronic components, and it also has become a barrier material. The Cr<sub>2</sub>O<sub>3</sub> films can also be used in bearings as well as in the combustion cylinder; in addition, such as hard disks and the appearance of digital magnetic recording media need to use this material [6–9]. As the Cr<sub>2</sub>O<sub>3</sub> films have stable chemical properties, good resistance to environmental corrosion, and have good optical performance, they are also used in fire prevention tools and optical instruments [10,11]. In addition, the Cr<sub>2</sub>O<sub>3</sub> films are also used to absorb solar energy for its higher solar absorbance and lower thermal emittance [12].

The Cr<sub>2</sub>O<sub>3</sub> thin films have been deposited by using various coatings preparing systems [13], such as plasma spraying, electrochemical deposition, arc ion plating, electron beam evaporation, metal–organic chemical vapor deposition, pulsed laser deposition, direct current sputtering, and radio frequency (RF) magnetron sputtering, etc. The AIP (Arc Ion Plating) is the most efficient technology among physical vapor deposition (PVD) techniques for its high ionization efficiency, high ion kinetic energy, and high preparation rate. Ji et al. [14] prepared the Cr<sub>2</sub>O<sub>3</sub> films employing AIP technique, and investigated the influence of oxygen flow rate on the film microstructure, morphology, and mechanical properties. It was found that the macro particles on Cr<sub>2</sub>O<sub>3</sub> film surface decreased gradually with the increase of O<sub>2</sub> flux. The most appropriate O<sub>2</sub> flux was about 130 sccm, in this experiment, the film possessed the maximum hardness of 36 GPa. Ho et al. [15] had studied Cr<sub>2</sub>O<sub>3</sub>/CrN biphasic coatings by cathode arc evaporation technique. The results indicated that the thin Cr<sub>2</sub>O<sub>3</sub> layer could serve as a thermal insulating layer at high temperatures, and to protect the inner coating. Accordingly, the maximum oxidation resistance temperature reached 900 °C in air. Wang et al. [16] had also investigated that Cr<sub>2</sub>O<sub>3</sub> is the only solid chromium oxide that keeps thermodynamically stable at temperatures in excess of 500 °C in the family of Cr<sub>x</sub>O<sub>y</sub>.

In previous works [17–20], the effect of substrate bias voltage on film growth mechanism, morphology, microstructure, mechanical properties, tribological properties and thermal stability of the nanocrystalline Cr<sub>2</sub>O<sub>3</sub> films were systematically investigated. Moreover, the double-layered and multilayered CrN/Cr<sub>2</sub>O<sub>3</sub> coatings were also developed. Accordingly the effects of the thickness ratio and the bilayer period on the microstructure, interface, and properties of the above coatings were studied systematically. However, few articles reported the characteristics of Cr<sub>2</sub>O<sub>3</sub> films at high temperatures, as well as the effect of annealing on the film microstructure, phase constituents, and grain size. Especially, how the microstructure and morphology change during the heating process is still unclear. Clarifying these issues is very helpful for the improvement of the Cr<sub>2</sub>O<sub>3</sub> film properties and their practical application.

Therefore, the changes of surface morphology of the AIP Cr<sub>2</sub>O<sub>3</sub> film during heat-treating process were in-situ observed, via an ESEM in a low vacuum atmosphere of 120 Pa, up to a temperature of 1100 °C. Because of the big difference of thermal expansion coefficient between the Si wafer and Cr<sub>2</sub>O<sub>3</sub> films, the big thermal stress was induced in the process of the heat treatment which led to film cracking. The effects of the heating temperature on the crack area percentage of the Cr<sub>2</sub>O<sub>3</sub> film surface were studied in detail. Moreover, the microstructure evolution, particle size, and phase structure of the Cr<sub>2</sub>O<sub>3</sub> film before and after heat treatment were also compared and discussed.

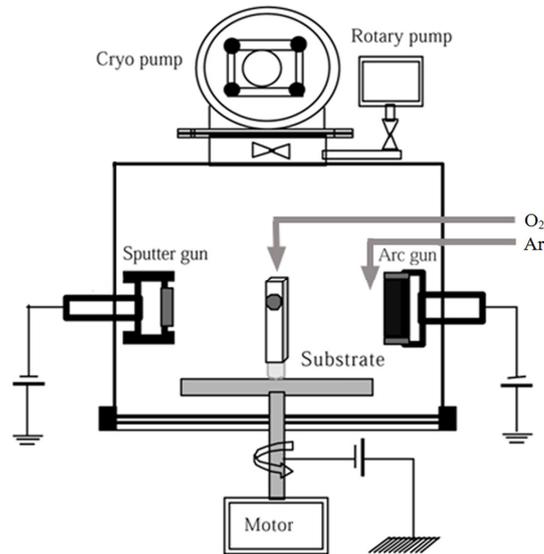
## 2. Experimental Details

### 2.1. Preparation of the Cr<sub>2</sub>O<sub>3</sub> Film

The Cr<sub>2</sub>O<sub>3</sub> film was deposited on single-crystalline Si wafer with (100) orientation by using a hybrid coating system. In this work, the only arc ion plating was used. The deposition chamber is a cylindrical shell structure (the diameter of 500 mm, height of 500 mm). The Cr target (99.99 wt % purity) was fixed at a distance of 350 mm from the rotating sample holder that with a diameter of 80 mm connected with arc cathode. The bias supply with a direct current (IDP-1010, KODIVAC Co., Ltd., Seoul, Korea) provided substrate bias voltage. A simulated diagram of the hybrid coating system is presented in Figure 1.

Before the deposition, the substrates were placed in acetone and ultrasonic solution to clean for 20 min, respectively. Next, they were placed on the rack with rotation. The substrate surface is parallel to the target surface and is held at the same level as the target. The samples were heated to 300 °C and then were cleaned by Ar ions bombardment at −800 V negative bias voltages for 5 min, which would remove contaminants and ensure strong adhesive adhesion between the film and substrate. In order to improve the adhesion and to reduce residual stresses, a Cr interlayer with the deposition time of 5 min was introduced. The inert gas (Ar) at a fixed flow ratio was input into the vicinity of Cr target to

avoid poisoning during the oxide film deposition process. The purity of Ar and reactive gas ( $O_2$ ) was 99.999%. The  $Cr_2O_3$  film thickness in this work was about  $3.5 \mu m$  by adjusting the preparation time. The detailed preparation parameters for nano-crystalline  $Cr_2O_3$  film fabricated by AIP technique are shown in Table 1.



**Figure 1.** Sketch map of the hybrid coating system.

**Table 1.** Detailed deposition parameters for nano-crystalline  $Cr_2O_3$  film fabricated by arc ion plating technique.

Parameters	Value
Base pressure (Pa)	$3.5 \times 10^{-3}$
Working pressure (Pa)	$3.0 \times 10^{-1}$
Deposition temperature ( $^{\circ}C$ )	300
Arc current (A)	55
DC bias voltage (V)	-100
Ar: $O_2$ gas flow ratio (sccm)	40:20
Substrate rotation speed (rpm)	25
Film thickness ( $\mu m$ )	$\sim 3.5$
Distance between the target and substrate (mm)	350

## 2.2. Characterizations of the $Cr_2O_3$ Film

The surface morphologies of the  $Cr_2O_3$  film during the heating process were in-situ tested by an environmental scanning electron microscope (ESEM, XL-30, Philips, Amsterdam, The Netherlands). A special hot stage with the circulation of cooling water was mounted in the sample chamber of ESEM for heating sample, which has been introduced in detail in [21]. The film sample was carefully placed in the platinum cup surrounding the thermocouple bead and the alumina and copper lids were placed in position. The heating rate was maintained at about  $5 \text{ }^{\circ}C/\text{min}$  in the first 20 min and gradually increased to over  $15 \text{ }^{\circ}C/\text{min}$  later on. The heating temperature ranged from room temperature to  $1100 \text{ }^{\circ}C$  which is less than the temperature limit for Si substrate ( $1300 \text{ }^{\circ}C$ ) [22]. And then the sample was cooled down to room temperature at a rate of  $20 \text{ }^{\circ}C/\text{min}$ . It should be noted that while saving images at low scanning speed, the film microstructure could change before the scan is completed, especially for high heating rates. During the heating process, due to thermal expansion, etc., image drift was very common and at high magnifications this could lead to losing sight of the film being studied. Thus, in order to get an optimized image, and let the visual field in the center of the screen, adjusting the instruments continuously were required. The increase in temperature of the sample

invariably led to a loss of contrast in the image, which could be compensated by choosing a higher spot size. All images of surface morphology with a magnification of  $1000\times$  were taken at different temperatures and time intervals during the heating process. Each image corresponds to an area of  $106 \times 80 \mu\text{m}^2$ . Based on the above topography images, the crack area percentage was evaluated using image analysis software (SISC IAS V8.0, SISC, Beijing, China) according to the principle of gray value analysis. The percentage values of crack area were calculated by the crack area on each image divided by the image area.

The transverse-section characteristics of as-deposited and heat-treated  $\text{Cr}_2\text{O}_3$  films were obtained using a field emission scanning electron microscope (FE-SEM, Hitachi S-4800, Hitachi, Tokyo, Japan). The phase constituent and microstructure analysis of the  $\text{Cr}_2\text{O}_3$  film before and after heat treatment, were conducted using an X-ray diffractometer (XRD, D8 ADVANCE, Bruker, Billerica, MA, USA) with monochromatic  $\text{Cu K}\alpha$  ( $\lambda = 0.154056 \text{ nm}$ ) radiation operated at 40 kV and 40 mA. A locked couple  $\theta$ - $2\theta$  mode with a  $0.02^\circ$  step size and a 0.2 s step time was employed. The analyzed range of diffraction angle  $2\theta$  was between  $30^\circ$  and  $80^\circ$ .

### 3. Results and Discussion

#### 3.1. Percentage of Crack Area for the Heat-Treated $\text{Cr}_2\text{O}_3$ Film

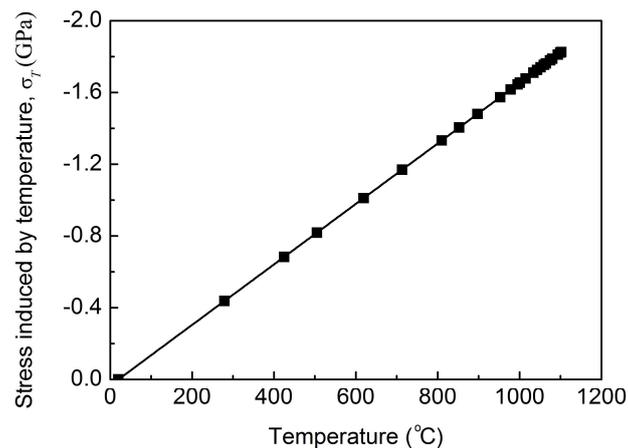
During the heating process, the large thermal stresses were introduced which could cause cracking and spalling of the film. Because the thermal expansion coefficient of the  $\text{Cr}_2\text{O}_3$  film ( $8.7 \times 10^{-6}$ – $9.6 \times 10^{-6} \text{ K}^{-1}$ ) and Si wafer ( $3.5 \times 10^{-6} \text{ K}^{-1}$ ) is very different [18], the mismatch of thermal expansion between them becomes more and more obvious which induced large thermal stress. For quantitative analysis, the thermal stress  $\sigma_T$  in  $\text{Cr}_2\text{O}_3$  film induced by temperature rise can be calculated by the following formula:

$$\sigma_T = -\frac{\Delta\alpha \times \Delta T \times E}{1 - \nu} \quad (1)$$

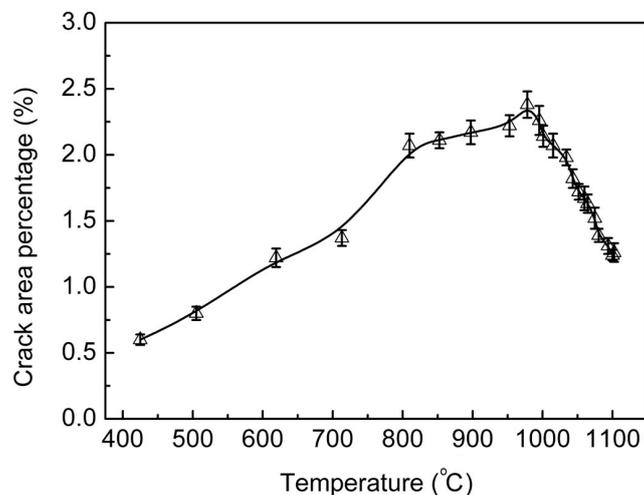
where  $\Delta\alpha$  is the difference of thermal expansion coefficient between the  $\text{Cr}_2\text{O}_3$  film and Si wafer,  $\Delta T$  is the difference between heating temperature and room temperature,  $E$  and  $\nu$  are elastic modulus and Poisson's ratio of the  $\text{Cr}_2\text{O}_3$  film, respectively. Here  $E = 230 \text{ GPa}$  and  $\nu = 0.25$  were chosen from reference [4], and the thermal expansion coefficient of the  $\text{Cr}_2\text{O}_3$  film chose  $9.0 \times 10^{-6} \text{ K}^{-1}$ . In this work, the thermal stress induced by temperature is negative which indicates the compressive stress generated, because the thermal expansion coefficient of  $\text{Cr}_2\text{O}_3$  film is larger than that of Si wafer.

Figure 2 shows the thermal stress variation in the  $\text{Cr}_2\text{O}_3$  film as the temperature rises, which was calculated by Equation (1). It can be seen that the compressive stress increased linearly with the increase of temperature. The maximum stress value was more than  $-1.8 \text{ GPa}$  as the temperature rose to  $1100^\circ\text{C}$ .

In addition, the crack area percentage on film surface was used to define the variation of  $\text{Cr}_2\text{O}_3$  film characteristics during the heating process, and the corresponding results are shown in Figure 3. It can be seen that the crack area percentage of the film first increased gradually with the increase of heating temperature, and reached the maximum value 2.38% at around  $980^\circ\text{C}$ . Since then, it sharply decreased again with the further increase of heating temperature. The thermal stress and grain growth mainly affected the percentage of crack area of the heat-treated film. It is known that usually there are some compressive residual stresses in AIP.



**Figure 2.** Variation of the thermal stress in the Cr<sub>2</sub>O<sub>3</sub> film with the increase of temperature.



**Figure 3.** Variation of the crack area percentage on the Cr<sub>2</sub>O<sub>3</sub> film surface with the increase of temperature.

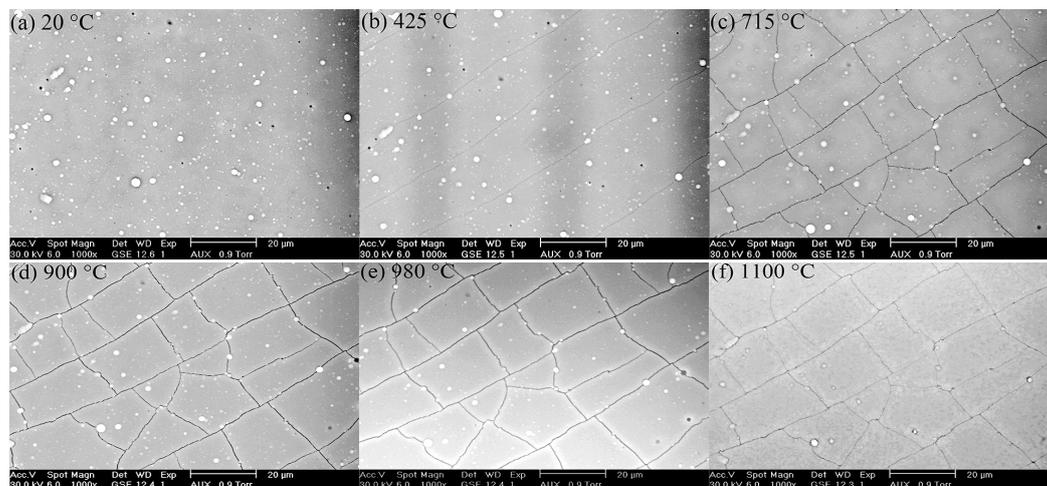
Cr<sub>2</sub>O<sub>3</sub> films [13], with the increase of heating temperature, the Si substrate expanded more than the Cr<sub>2</sub>O<sub>3</sub> film due to the former's larger thermal expansion coefficient. Thus, the large tensile thermal stresses were introduced into the film according to the principle of thermal expansion and contraction. At first, the compressive stresses in Cr<sub>2</sub>O<sub>3</sub> film can effectively prevent the occurrence and propagation of cracks which derived from thermal stress. The higher the temperature the larger the thermal stresses can be generated. When the thermal stresses reached a certain value the Cr<sub>2</sub>O<sub>3</sub> film started to crack. As the temperature rose further, the increasing thermal expansion difference between the film and substrate favored the propagation and expansion of cracks, which led to a gradual increase in the crack area percentage.

Another factor, the crystal growth can explain why the crack area percentage started to decline sharply as the heating temperature was above 980 °C. It was reported that the crystal growth was closely associated with the temperature and time of heat treatment. By differential scanning calorimetry and thermogravimetric analysis, He et al. [23] also found that the crystallization and vaporization temperatures of the AIP Cr<sub>2</sub>O<sub>3</sub> film are 360 °C and 940 °C, respectively. Their research results further revealed that the grain size of the Cr<sub>2</sub>O<sub>3</sub> film clearly increased after annealing above the crystallization temperature. In this work, the grain growth may be somewhat delayed due to the higher heating rate mentioned above, which brought the very transient grain growth time at different temperatures. The change of temperature would directly affect the grain size and grain boundary

sliding, which influenced the mechanical properties of films. In general, the higher temperature incurred the bigger transfer coefficient, grain size, and nucleation rate, but the too high temperature would not be conducive to crystal growth. Moreover, the crystal growth at different directions needed different temperatures to meet surface energy requirement. Usually, the grain boundaries have some certain inhibitory effects on crystal growth. However, there is no restriction to grain growth on both sides of the crack, because they have a free boundary. Thus, these grains would grow toward the crack, which results in a decrease in the crack area percentage of the  $\text{Cr}_2\text{O}_3$  film as the temperature exceeded  $980^\circ\text{C}$ . With the further increase of heating temperature, the  $\text{Cr}_2\text{O}_3$  grains were growing up and the crack area percentage declined rapidly.

### 3.2. Surface Morphologies

The in-situ surface morphologies of the  $\text{Cr}_2\text{O}_3$  film heated at several typical temperatures are shown in Figure 4a–f. From Figure 4a, it can be seen that some macro-particles range from a few hundred nanometers to a few micrometers, and a few hundred or dozens of nanometer pinholes were randomly distributed on the membranes surface at room temperature. The droplets emitted from the arc spots of the Cr target caused macro-particles. It had been reported that there are two kinds of different macroparticles on the AIP  $\text{Cr}_2\text{O}_3$  film surface. Some particles that have been deposited previously were embedded in the film, the latest macro-particles fell on the film surface randomly [13]. The growth of the deposited macro-particles was inhomogeneous (which flake from the substrate or shrink during deposition, etching, and heating/cooling in the film chamber), which could lead the pinholes in the films [24,25]. According to these morphology images, the size and amount of macroparticles did not change obviously with the temperature variation.



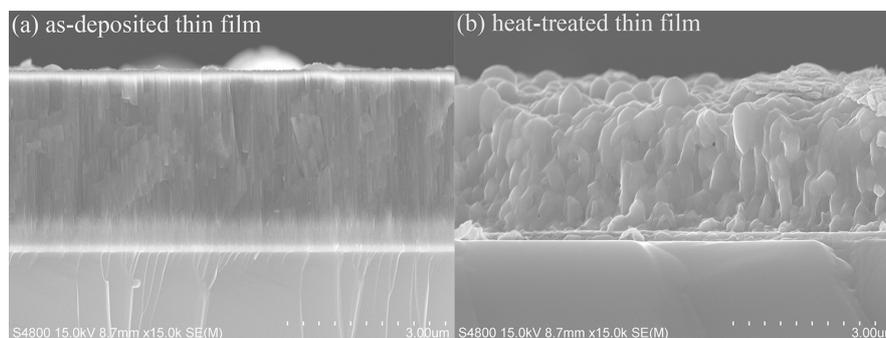
**Figure 4.** In-situ surface morphologies of the  $\text{Cr}_2\text{O}_3$  film heated at several typical temperatures: (a)  $20^\circ\text{C}$ ; (b)  $425^\circ\text{C}$ ; (c)  $715^\circ\text{C}$ ; (d)  $900^\circ\text{C}$ ; (e)  $980^\circ\text{C}$ ; and (f)  $1100^\circ\text{C}$ .

As the temperature was raised to about  $425^\circ\text{C}$  in Figure 4b, the  $\text{Cr}_2\text{O}_3$  film started to crack for the large thermal stresses generated during the heating process. These cracks were parallel to each other and propagated in the same direction. When the tip of crack encountered the macroparticles, it passed around the edge of macroparticles and did not separate them into two parts. That is to say, the initiation and propagation of cracks in the  $\text{Cr}_2\text{O}_3$  film may be associated with the macro-particle contamination to a certain extent. Moreover, it was also proved the preferred orientation growth of the  $\text{Cr}_2\text{O}_3$  film, which is accordance with the previous XRD study [13]. As the temperature increased to about  $715^\circ\text{C}$  in Figure 4c, some new cracks perpendicular to the original crack direction appeared. There were also a few cracks propagating and extending in other directions. The above cracks were connected together to form a network structure, which effectively released the large thermal stresses.

With the heating temperature increased further, the crisscrossing cracks in Figure 4d became wider than before. Namely the crack area percentage was increasing which was attributed to the thermal expansion coefficient of the Si substrate is bigger than the  $\text{Cr}_2\text{O}_3$  film, the latter expanded more already mentioned above. At about  $980^\circ\text{C}$  in Figure 4e, the maximum crack gaps could be found, accordingly the crack area percentage reached the maximum. In addition, the tiny drift of visual field and the loss of image contrast can also be observed, which were compensated by continuous adjustments and choosing a bigger spot size. In Figure 4f, the crack gaps reduced significantly. In this case, the heating temperature reached maximum at  $1100^\circ\text{C}$  and the  $\text{Cr}_2\text{O}_3$  grains also grew up after a long time heating. The grains on both sides of crack continuously grew toward the gaps, which led to the reduced crack gaps. Kacsich et al. [26] had also reported that the  $\text{Cr}_2\text{O}_3$  grain growth was depended on the outward lattice diffusion or grain boundary diffusion, and the inward diffusion of oxygen via grain boundaries.

### 3.3. Fractured Cross-Section

Figure 5a,b depicts the fractured cross-sectional SEM graphs of the as-deposited and heat-treated  $\text{Cr}_2\text{O}_3$  thin films. It can be seen clearly that the short and thin  $\text{Cr}_2\text{O}_3$  columnar units, Cr interlayer, and Si substrate from Figure 5a, and the film showed a dense microstructure. The AIP  $\text{Cr}_2\text{O}_3$  film possesses a nano-scale grain size that the average width of columnar grains is less than 50 nm. The Cr interlayer adhered well to the substrate. There are no pinholes and pores in the columnar structure. However, in Figure 5b, the columnar structure was coarsened significantly after heat treatment. The column width ranged from 200 to 500 nm, and some pores were observed in the film. These pores would seriously affect the film properties, even if its relatively dense microstructure and low defect density. According to previous work [18], the film generated many pores after heat treatment, mainly because the vaporization of  $\text{Cr}_2\text{O}_3$  phase at high temperature. Similarly, it had also been reported that the  $\text{Cr}_2\text{O}_3$  was the main stable product in the temperature range of  $600\text{--}950^\circ\text{C}$ . When the temperature increased further, the  $\text{Cr}_2\text{O}_3$  and  $\text{CrO}_3$  phases coexisted in the system, and the latter would gradually evaporate [27]. In addition, it is also observed that the grains in Cr interlayer have grown up due to high temperature oxidation reaction with residual oxygen in ESEM.

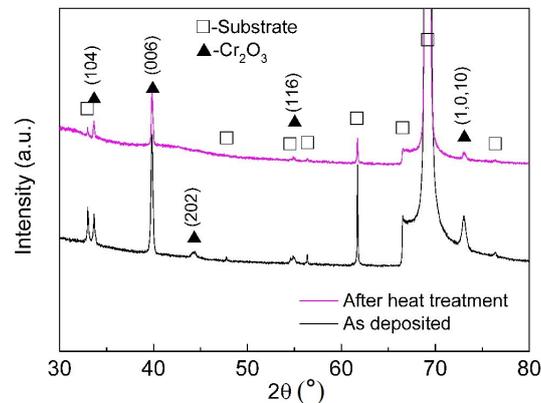


**Figure 5.** Fractured cross-sectional SEM images of the  $\text{Cr}_2\text{O}_3$  thin film: (a) as-deposited; (b) heat-treated.

### 3.4. XRD Analysis

Figure 6 illustrates the X-ray diffraction spectrum of the  $\text{Cr}_2\text{O}_3$  film before and after heat treatment. In contrast, the  $\text{Cr}_2\text{O}_3$  (202) diffraction peak at the position of  $2\theta = 44.26^\circ$  disappeared after heat treatment. The reasons are not clear at present. Additionally, these two spectral lines are very similar, and can be determined as the diffraction peaks of  $\text{Cr}_2\text{O}_3$  phase. The four diffraction peaks at the positions of about  $2\theta = 33.68^\circ$ ,  $39.81^\circ$ ,  $54.87^\circ$  and  $73.01^\circ$  corresponded to (104), (006), (116), and (1,0,10) crystal planes of rhombohedral corundum type  $\text{Cr}_2\text{O}_3$  were observed, respectively. Moreover, preferred orientation was still (006) crystal plane after heat treating, and the film was only composed of the  $\text{Cr}_2\text{O}_3$  phase. There were no other chromium oxide phases detected in the heat-treated film. As a

result, the phase constituents of the film after heat treatment did not change except for the reduction of peak intensity, which was attributed to the smaller sample size.

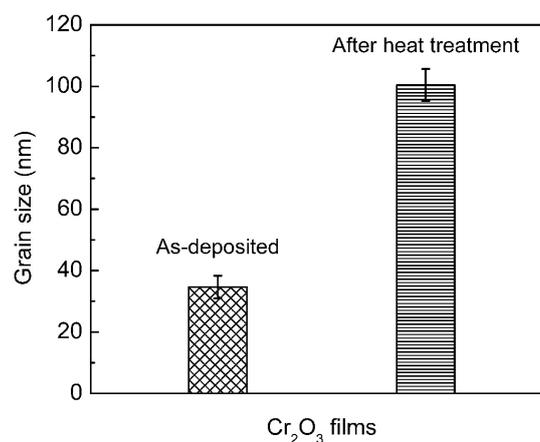


**Figure 6.** X-ray diffraction spectrum of the  $\text{Cr}_2\text{O}_3$  film before and after heat treatment.

The average grain size  $R_a$  of the as-deposited and heat-treated  $\text{Cr}_2\text{O}_3$  films were calculated using the full-width at half maximum (FWHM) value of the strongest (006) diffraction peak by the Scherrer's formula [28], given as:

$$R_a = \frac{0.9\lambda}{B \cos \theta} \quad (2)$$

where  $\lambda$  is the wavelength of Cu  $K\alpha$  radiation,  $B$  is the calibrated full-width at half maximum of a Bragg peak, and  $\theta$  is the Bragg angle. Here the diffraction peaks of Si single crystal in  $\theta$ - $2\theta$  mode spectra were used as the calibration reference. Figure 7 shows the average grain size of the as-deposited and heat-treated  $\text{Cr}_2\text{O}_3$  films. From the picture that the grain size increased nearly two times from 35 nm to about 100 nm after heat treatment. The fine  $\text{Cr}_2\text{O}_3$  grains accelerated the diffusion of grain boundary during the heating process, which resulted in the rapid growth of grains. Here, the interior diffusion of oxygen and external diffusion of chromium promoted the grain growth. Tabbal et al. [29] also obtained similar results, namely at the high temperature, it occurred that a significant improvement in the crystalline quality and a reduction of internal stress. Accordingly, the grain size varied from 50 to 200 nm in the temperature range of 20–950 °C.



**Figure 7.** Average grain size of the as-deposited and heat-treated  $\text{Cr}_2\text{O}_3$  films.

#### 4. Conclusions

The crack area percentage was evaluated by the ESEM topography images in-situ observed. With the increase of heating temperature, its value first increased gradually and reached the maximum at about 980 °C, and then sharply decreased again. The thermal stresses and Cr<sub>2</sub>O<sub>3</sub> grain growth explained above phenomenon well.

After heat treatment, the microstructure of the Cr<sub>2</sub>O<sub>3</sub> film was coarsened significantly. The column width ranged from 200 to 500 nm, and some obvious pores were left in the film which was attributed to the vaporization of Cr<sub>2</sub>O<sub>3</sub> phase at high temperature.

The grain size of Cr<sub>2</sub>O<sub>3</sub> film increased nearly two times from 35 nm to about 100 nm after the heat treatment. The high temperature promoted the rapid growth of Cr<sub>2</sub>O<sub>3</sub> grains by the interior diffusion of oxygen and external diffusion of chromium via the grain boundaries.

**Acknowledgments:** This work was supported by the Global Frontier Program through the Global Frontier Hybrid Interface Materials (GFHIM) of the National Research Foundation of Korea (NRF) funded by the Ministry of Science, ICT & Future Planning (2013M3A6B1078874). In addition, it was also funded by the National Nature Science Foundation of China (51301181), the Tianjin Key Research Program of Application Foundation and Advanced Technology (15JCZDJC39700).

**Author Contributions:** Tie-Gang Wang and Srinivasan Iyengar performed the experiments; Yanmei Liu and Zubing Yang analyzed the data; Kwang Ho Kim contributed reagents/materials/analysis tools; Tie-Gang Wang and Yu Dong wrote the paper.

**Conflicts of Interest:** The authors declare no conflict of interest.

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