

Supplementary Materials: Bi-Phase NiCo₂S₄-NiS₂/CFP Nano-composites as a Highly Active Catalyst for Oxygen Evolution Reaction

Jintang Li ^{*}, Yongji Xia, Xianrui Luo, Tianle Mao, Zhenjia Wang, Zheyu Hong and Guanghui Yue ^{*}

College of Materials, Xiamen University, Xiamen 361005, China
* Correspondence: leejt@xmu.edu.cn (J.L.), yuegh@xmu.edu.cn (G.Y.)

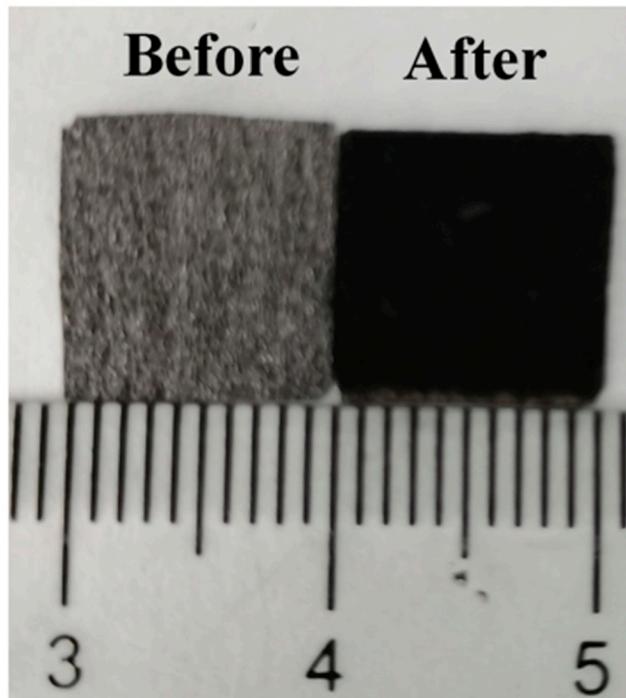


Figure S1. The surface change after as prepared NiCo₂S₄-NiS₂ anchored on the surface of CFP.

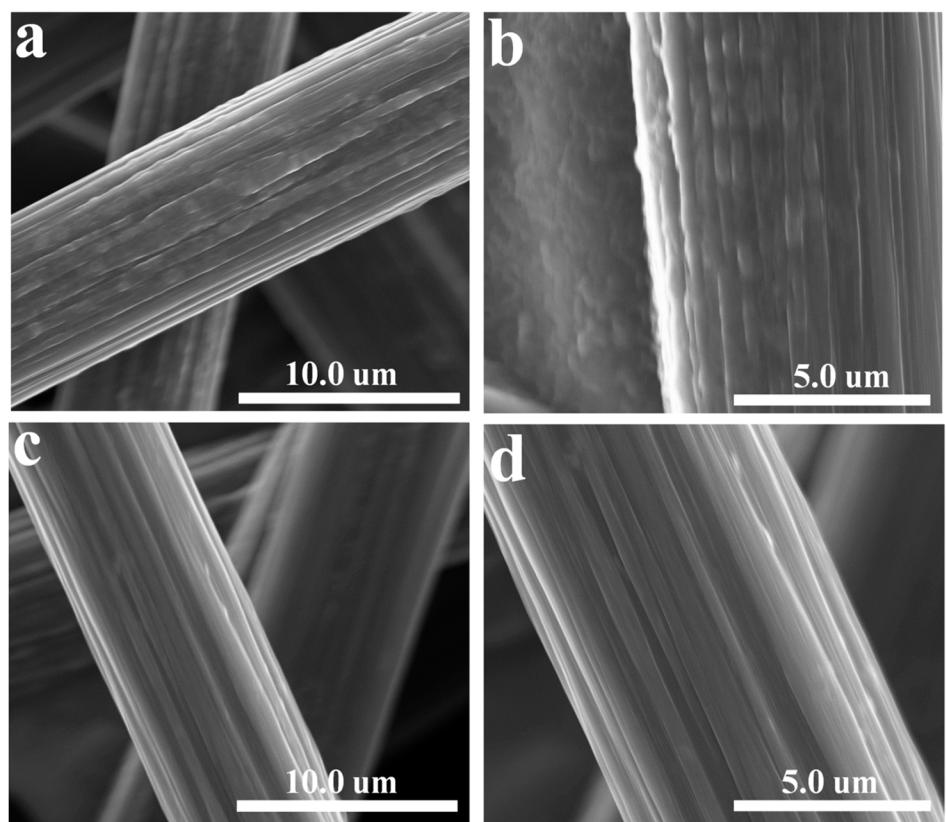


Figure S2. Low-magnification SEM images of (a) CFP and (c) OCFP. High-magnification SEM images of (b) CFP and (d) OCFP.

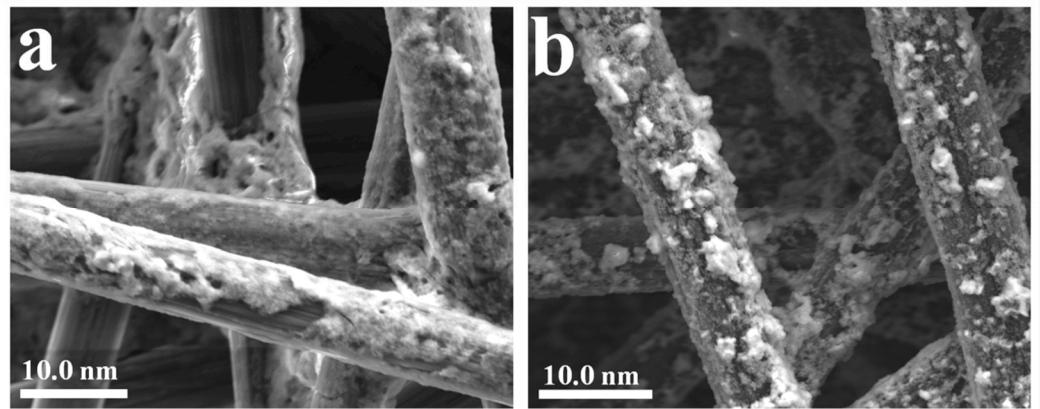


Figure S3. The morphologies of (a) Ni_3S_4 - NiS_2 /CFP and (b) CoS_2 /CFP.

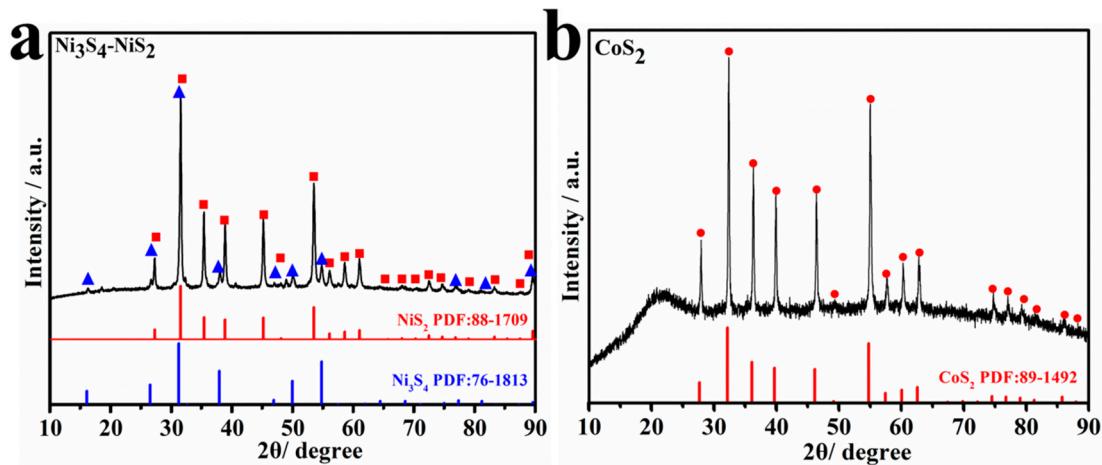


Figure S4. XRD patterns of (a) $\text{Ni}_3\text{S}_4\text{-NiS}_2$ and (b) CoS_2 .

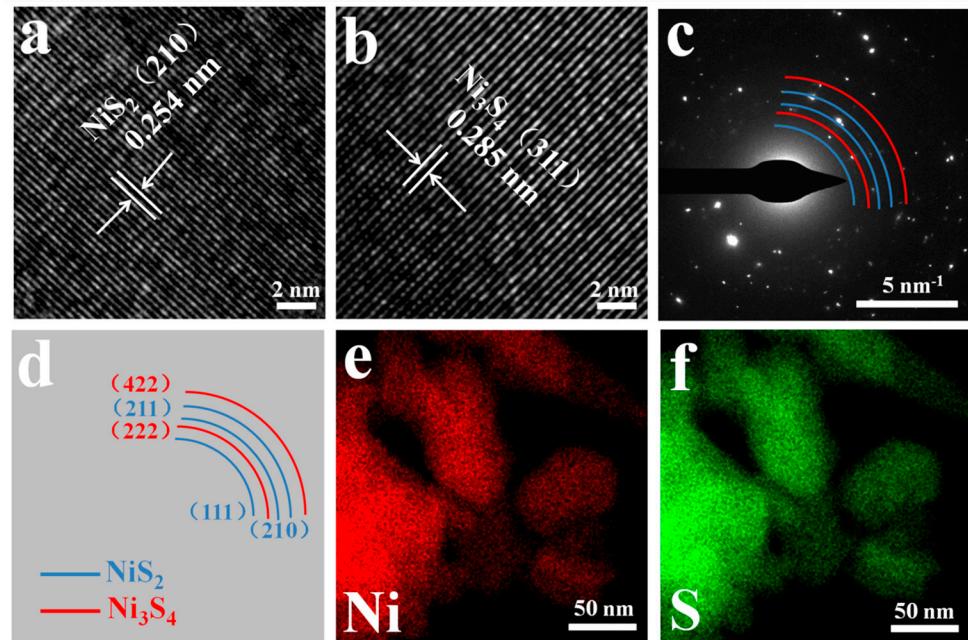


Figure S5. (a),(b) HRTEM image of $\text{Ni}_3\text{S}_4\text{-NiS}_2$. (c),(d) Selected area electron diffraction (SAED) diffraction ring of $\text{Ni}_3\text{S}_4\text{-NiS}_2$. (e),(f) The area chosen to do EDX elemental mapping and the corresponding EDX elemental mapping images for Ni and S respectively.

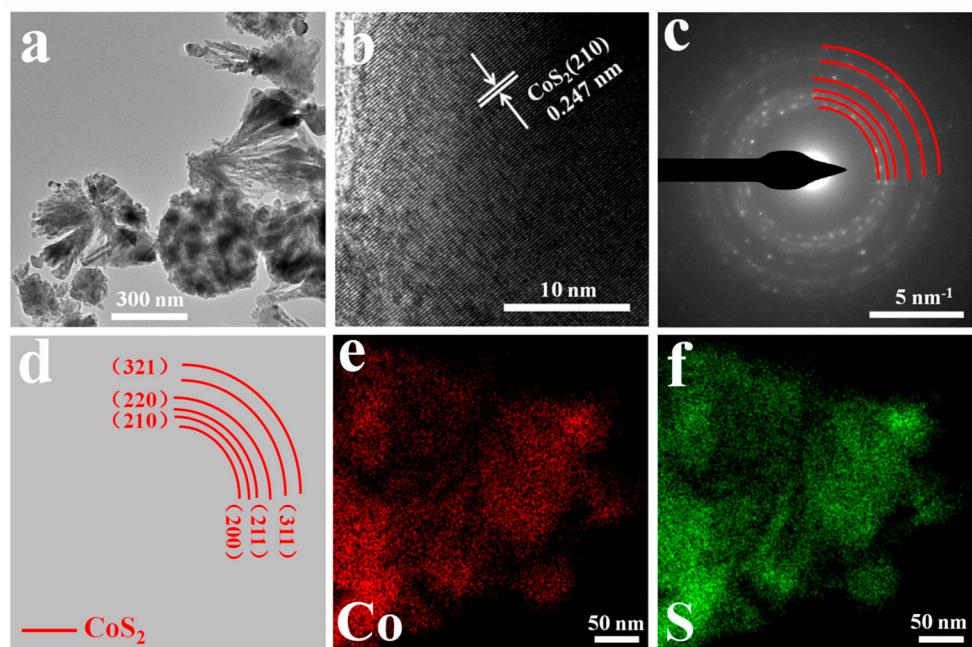


Figure S6. (a) The TEM image of CoS_2 . (b) HRTEM image of CoS_2 . (c),(d) Selected area electron diffraction (SAED) diffraction ring of CoS_2 . (e),(f) The area chosen to do EDX elemental mapping nd the corresponding EDX elemental mapping images for Co and S respectively.

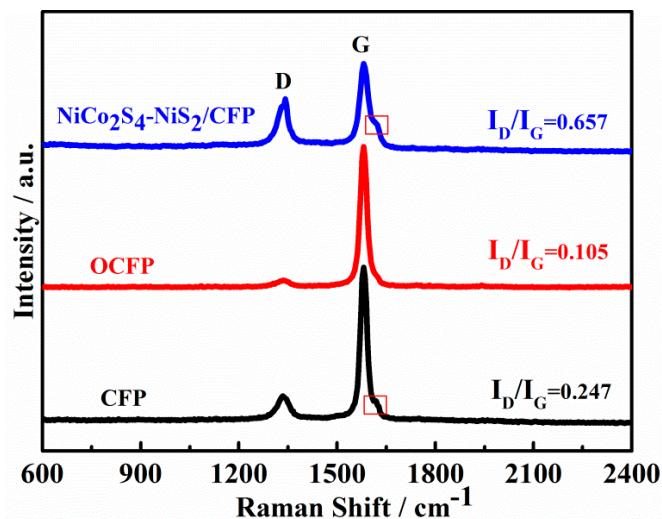


Figure S7. Raman spectra of CFP, OCFP and $\text{NiCo}_2\text{S}_4\text{-NiS}_2/\text{CFP}$.

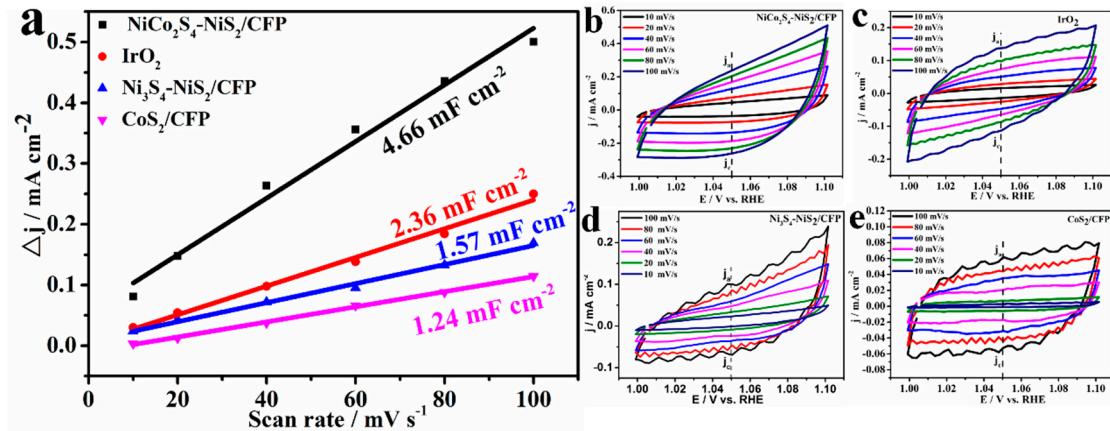


Figure S8. Double-layer capacitances (a) and CV scans at different scan rates (b–e).

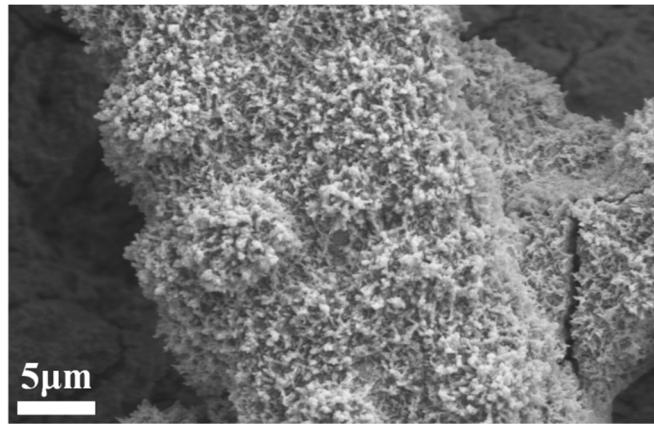


Figure S9. The morphology of $\text{NiCo}_2\text{S}_4\text{-NiS}_2/\text{CFP}$ after 12 h durability test.

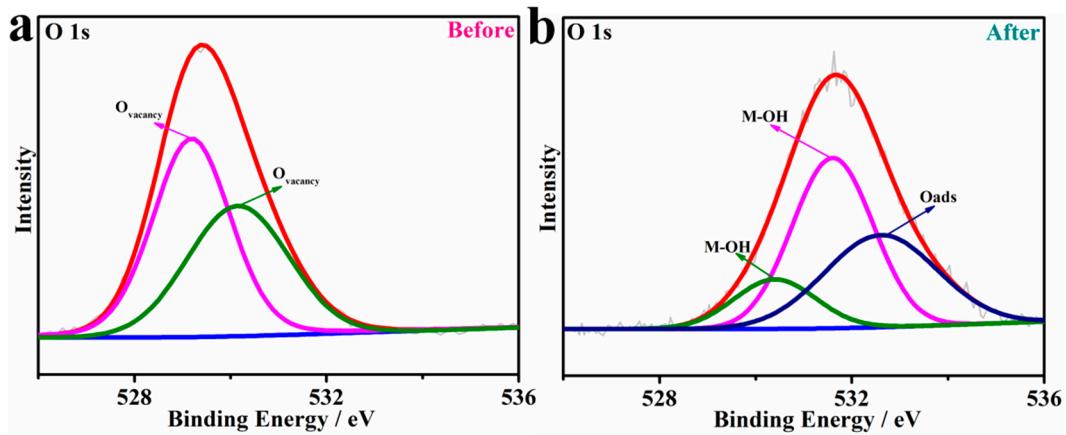


Figure S10. The O XPS spectra of as-prepared catalyst before (a) and after (b) 12 h durability test.

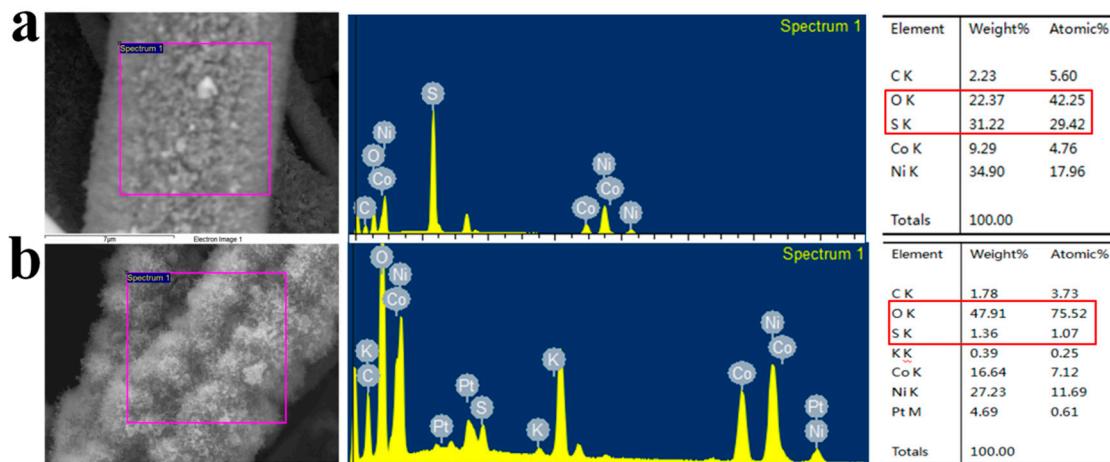


Figure S11. The EDS results of NiCo₂S₄-NiS₂/CFP (a) before 12 h durability test. (b) After 12 h durability test.

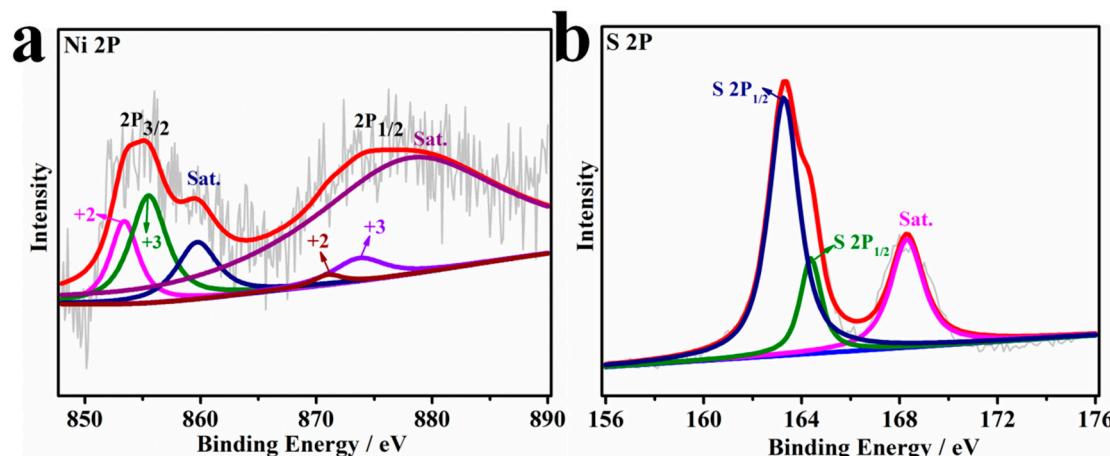


Figure S12. The XPS spectra of (a) Ni 2p, (b) S 2p in Ni₃S₄-NiS₂/CFP.

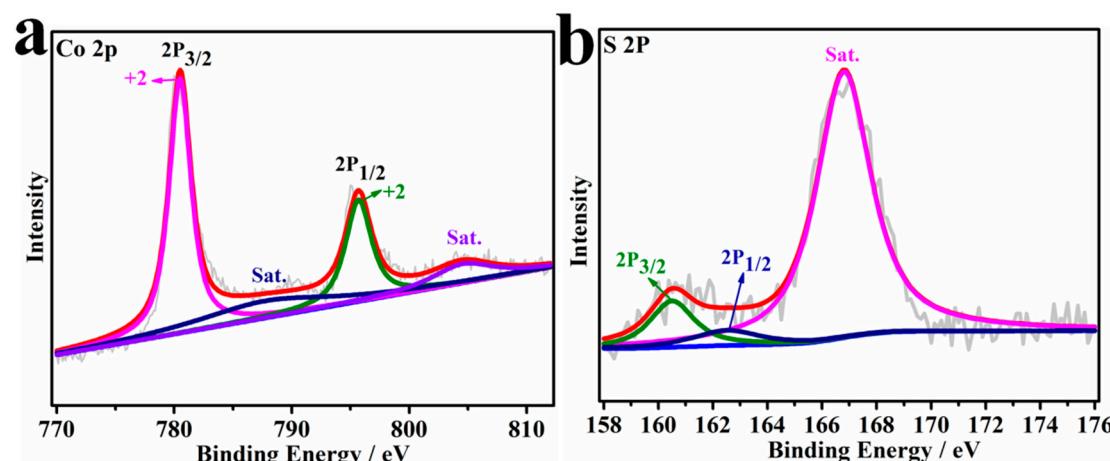


Figure S13. The XPS spectra of (a) Co 2p, (b) S 2p in CoS₂/CFP.

Table S1. Comparison of OER activity of NiCo₂S₄-NiS₂/CFP with that of most reported nickel-cobalt sulfide catalysts tested in alkaline solution.

Catalysts	E ₁₀ (V vs. RHE)	Tafel slope b (mV/dec)	Reference
NiCo ₂ S ₄ -NiS ₂ /CFP	1.395	81.54	This work

Ni ₃ S ₄ -NiS ₂ /CFP	1.485	179.2	This work
Ni-Fe-Co-S NSs/NNCs	1.437	63	[1]
C _x Ni _{1-x} S ₂	1.52	46	[2]
NCT-NiCo ₂ S ₄	1.56 (100 mA/cm ²)	86.8	[3]
NiCoDH/NiCoS	1.533 (20 mA/cm ²)	77.6	[4]
NiCoS/Ti ₃ C ₂ T _x	1.595	58.2	[5]
Ni _{4.3} Co _{4.7} S ₈	1.378	90	[6]
H-3DRG@NiCo-LDH	1.519	80.3	[7]
H-3DRG@NiCo ₂ S ₄	1.494	93.5	[7]
Co _{1.22} Ni _{0.78} P ₂ O ₇ -C/RGO	1.513	51	[8]
NiCoS-3	1.55	58.8	[9]
NiCo ₂ S ₄	1.567	64	[10]

E₁₀ is the potential to reach 10mA cm⁻² for OER.

References

1. Darband G. B.; Aliofkhazraei M.; Hyun S.; Rouhaghdam A. S.; Shanmugam S. Electrodeposition of Ni-Co-Fe mixed sulfide ultrathin nanosheets on Ni nanocones: a low-cost, durable and high performance catalyst for electrochemical water splitting. *Nanoscale* 2019, 11, 16621-16634.
2. Hong Y.R.; Mhin S.; Kim K.M.; Han W.S.; Choi H.; Ali G.; Chung K. Y.; Lee H. J.; Moon S.I.; Dutta S.; et al. Electrochemically activated cobalt nickel sulfide for an efficient oxygen evolution reaction: partial amorphization and phase control. *J. Mater. Chem. A* 2019, 7, 3592-3602.
3. Li F., Xu R., Li Y., Liang F., Zhang D., Fu W.F., Lv X.J. N-doped carbon coated NiCo₂S₄ hollow nanotube as bifunctional electrocatalyst for overall water splitting. *Carbon* 2019, 145, 521-528.
4. Zhao H.; Yang Y.; Dai X.; Qiao H.; Yong J.; Luan X.; Yu L.; Luan C.; Wang Y.; Zhang X. NiCo-DH nanodots anchored on amorphous NiCo-Sulfide sheets as efficient electrocatalysts for oxygen evolution reaction. *Electrochim. Acta* 2019, 295, 1085-1092.
5. Zou H.; He B.; Kuang P.; Yu J.; Fan K. Metal-Organic Framework-Derived Nickel-Cobalt Sulfide on Ultrathin Mxene Nanosheets for Electrocatalytic Oxygen Evolution. *ACS Appl. Mater. Interfaces* 2018, 10, 22311-22319.
6. Tang Y.; Yang H.; Sun J.; Xia M.; Guo W.; Yu L.; Yan J.; Zheng J.; Chang L.; Gao F. Phase-pure Pentlandite Ni_{4.3}Co_{4.7}S₈ Binary Sulfide as Efficient Bifunctional Electrocatalyst for Oxygen Evolution and Hydrogen Evolution. *Nanoscale* 2018, 10, 10459-10466.
7. Qin K.; Wang L.; Wen S.; Diao L.; Liu P.; Li J.; Ma L.; Shi C.; Zhong C.; Hu W.; et al. Designed synthesis of NiCo-LDH and derived sulfide on heteroatom-doped edge-enriched 3D rivet graphene films for high-performance asymmetric supercapacitor and efficient OER. *J. Mater. Chem. A* 2018, 6, 8109-8119.
8. Zhao C.; Li P.; Shao D.; Zhang R.; Wang S.; Zhu Z.; Zhao C. Phytic acid-derived Co_{2-x}Ni_xP₂O₇-C/RGO and its superior OER electrocatalytic performance. *Int. J. Hydrog. Energy* 2019, 44, 844-852.
9. Yu Z.; Bai Y.; Zhang S.; Liu Y.; Zhang N.; Sun K. MOF-directed templating synthesis of hollow nickel-cobalt sulfide with enhanced electrocatalytic activity for oxygen evolution. *Int. J. Hydrog. Energy* 2018, 43, 8815-8823.
10. Jiang J.; Yan C.; Zhao X.; Luo H.; Xue Z.; Mu T. A PEGylated deep eutectic solvent for controllable solvothermal synthesis of porous NiCo₂S₄ for efficient oxygen evolution reaction. *Green Chem.* 2017, 19, 3023-3031.