



X-ray Detectors Based on Halide Perovskite Materials

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Abstract: Halide perovskite has remarkable optoelectronic properties, such as high atomic number, large carrier mobility-lifetime product, high X-ray attenuation coefficient, and simple and low-cost synthesis process, and has gradually developed into the next-generation X-ray detection materials. Halide perovskite-based X-ray detectors can improve the sensitivity and reduce the detectable X-ray dose, which is applied in imaging, nondestructive industrial inspection, security screening, and scientific research. In this article, we introduce the fabrication methods of halide perovskite film and the classification and progress of halide perovskite-based X-ray detectors. Finally, the existing challenges are discussed, and the possible directions for future applications are explored. We hope this review can stimulate the further improvement of perovskite-based X-ray detectors.

Keywords: halide perovskite; X-ray detector; film fabrication methods; sensitivity

1. Introduction

X-ray detectors are commonly used in medical imaging, nondestructive industrial inspection, safety screening, and scientific research [1–3]. High sensitivity and high detection efficiency are believed to be critical figures of merit for X-ray detectors because weak X-ray signals can be detected, which greatly reduces the risk of medical examination [4–6]. The performance is closely related to the features of X-ray detection semiconductor materials including atomic number, charge carrier mobility, and carrier lifetime [7–9]. Various semiconductor materials have been utilized in X-ray detectors, such as silicon (Si), amorphous selenium (α -Se), germanium (Ge), and cadmium zinc telluride (CdZnTe). In addition, diamond is considered the elective material for X-ray detection and has a wide range of applications in the radiotherapy field mainly due to its "tissue equivalence" characteristic [10–13]. However, traditional materials such as α -Se, Ge, and CdZnTe suffer from many issues, including relatively small atomic numbers, the low attenuation coefficient of X-rays, and complex and costly fabrication processes [14–16]. Thus, it is necessary to explore new materials to replace the traditional materials for X-ray detectors.

Halide perovskites with a formula of ABX₃ (where A = CH₃NH₃⁺ (MA⁺), HC(NH₂)₂⁺ (FA⁺) and Cs⁺; B = Pb²⁺ or Sn²⁺; X = halides) are emerging candidates in X-ray detection. They have very high X-ray attenuation coefficients, a large carrier mobility-lifetime product ($\mu\tau$ product), a high atomic number, and a simple and low-cost synthesis process [17,18]. The summarization of the parameters of the materials for X-ray detectors is shown in Table 1. X-ray detectors employing organic–inorganic hybrid perovskite materials such as MAPbI₃ and MAPbBr₃ and all-inorganic perovskite materials such as CsPbBr₃ have been reported. According to the structural dimensionalities of halide perovskites, various types of materials with different characteristics of three-dimensional, low-dimensional, and three-dimensional /low-dimensional hybrid perovskite appear successively. Moreover, different techniques are used for the fabrication of high-quality halide perovskite films, which is the premise for achieving useful X-ray detectors. However, reviews that comprehensively



Citation: Tan, Y.; Mu, G.; Chen, M.; Tang, X. X-ray Detectors Based on Halide Perovskite Materials. *Coatings* 2023, 13, 211. https://doi.org/ 10.3390/coatings13010211

Academic Editors: Alicia de Andrés and Torsten Brezesinski

Received: 11 December 2022 Revised: 9 January 2023 Accepted: 12 January 2023 Published: 16 January 2023



Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). summarize the fabrication methods of halide perovskite films and the features of halide perovskite-based X-ray detectors with different structural dimensionalities and different organic and inorganic components are relatively rare.

Material	Atomic Number	Density (g cm ⁻³)	Band Gap (eV)	μτ Product (cm ² V ⁻¹)	Resistivity (Ω cm)	Ref.
Si	14	2.33	1.12	>1	10^{4}	[19]
α-Se	34	4.3	2.1-2.2	10^{-7}	$10^{4} - 10^{5}$	[20]
CdTe	48, 52	6.2	1.44	$1.5 imes 10^4$	$10^{8} - 10^{9}$	[21]
Cd _{0.9} Zn _{0.1} Te	48, 30, 52	5.78	1.57	10^{-2}	10^{11}	[21,22]
PbI ₂	82, 53	6.2	2.3-2.6	10^{-5}	10^{13}	[20]
HgI_2	80, 53	6.4	2.13	10^{-4}	10^{13}	[23]
Ğe	32	5.33	1.57	>1	50	[24]
CsPbBr ₃	56.4	4.55	2.2-2.33	$1.3 imes10^{-2}$	$8.5 imes 10^9$	[25]
MAPbI ₃	35.6	4.3	1.5-1.6	$10^{-4} - 10^{-7}$	-	[26]
MAPbBr ₃	35, 82	3.45	2.2-2.3	-	-	[5]
Diamond	12	3.52	5.25-5.5	-	-	[10-13,20]

Table 1. The summarization of the parameters of the materials for X-ray detectors.

This review introduces in detail the fabrication methods of halide perovskite films and then summarizes the progress of research and development of halide perovskite-based X-ray detectors. Figure 1 shows the main fabrication methods and types of detectors introduced in this review. Finally, according to the progress of halide perovskite-based X-ray detectors, the existing problems are discussed, and the possible directions for future applications are explored.



Figure 1. Progress in X-ray detectors based on halide perovskite materials.

2. Halide Perovskite Film Fabrication Methods

The quality of the film is the key factor to determine the performance of perovskitebased X-ray detectors. Fabrication methods are classified as a spin-coating process, dissolution and recrystallization method, spray deposition process, aerosol–liquid–solid method, solution growth method, scraper method, inkjet printing method, and laser direct writing and melt and crystallization. In the following, we describe these methods, introduce representative works, and analyze their features.

2.1. Spin-Coating Process

Spin coating is an early and widely used technology in perovskite film processing due to its simple and easy operation process. The process is mainly divided into two steps: dropping the solution on the cleaned substrate and accelerating the rotating plate to a required velocity. A uniform film is formed on the substrate because of the solution diffusion and solvent evaporation at high speed. The thickness of the film depends on the concentration, density, and spin velocity of the solution. In 2015, Yakunin et al. fabricated an X-ray detector based on methylammonium lead iodide perovskite (CH₃NH₃PbI₃) perovskite materials through the spin-coating process [27]. Uniform perovskite films deposited onto patterned electrode structures are accomplished through spin-coating process suffers from the difficulty of scaling up to large-area substrates. In addition, the rotational speed, solution concentration, etc., of different solutions need to be reexplored to obtain the desired thickness.



Figure 2. Spin-coating, dissolution and recrystallization, spray deposition, and aerosol–liquid–solid methods. (**a**) The cross-sectional optical microscopy of perovskite films and the scale bar is 100 μm. Reprinted with permission from Ref. [27]. 2015, Springer Nature. (**b**) CsPbBr₃ thick film preparation and repair process diagram. Reprinted with permission from Ref. [26]. 2019, Wiley. (**c**) SEM images of (i) the original CsPbBr₃ thick film, and the treated thick films with (ii) 5, (iii) 10, and (iv) 20 times dissolution and recrystallization. Reprinted with permission from Ref. [26]. 2019, Wiley. (**d**) Spraying device for depositing double-layer perovskite film. Reprinted with permission from Ref. [28]. 2020, Optica. (**e**) SEM cross-sections of different layers of perovskite solar cells. Reprinted with permission from Ref. [28]. 2020, Optica. (**f**) Comparison of traditional solution-based approach to film production (**left**) and the aerosol–liquid–solid method (**right**). This highlights the latter's unique ability to manufacture high-quality thick perovskite films. Reprinted with permission from Ref. [29]. 2021, Cell Press. (**g**) Cross-section (**upper**) and top-view SEM images (**lower**) of perovskite films fabricated by the aerosol–liquid–solid method after 4 (**left**) and 100 growth cycles (**right**). Reprinted with permission from Ref. [29]. 2021, Cell Press.

2.2. Dissolution and Recrystallization Method

The dissolution and recrystallization process is an efficient method for the preparation of high-quality perovskite films. During many dissolution and recrystallization processes, the sharp portion of the surface particles of the film is dissolved in the original perovskite solution. This dissolved material recrystallizes at the right temperature, filling the holes in the surface. After this repeated process, smooth and dense thick perovskite films can be obtained. In 2019, Zongyan Gou et al. ameliorated the film of the CsPbBr₃ microcrystalline after many dissolutions and recrystallizations, laying a good foundation for obtaining a high-performance X-ray detector (Figure 2b) [26]. The scanning electron microscope (SEM) images of the CsPbBr₃ films after repairing at different times are shown in Figure 2c. The surface of the original membrane is rough and full of holes. With the increase in dissolution and recrystallization times, the surface of the CsPbBr₃ film becomes microporous and smooth. This indicates that the rigid part of the CsPbBr₃ crystal on the surface of the film gradually dissolves into the original perovskite solution, leading to the filling of the holes.

2.3. Spray Deposition Process

The spray deposition process is one of the low-cost and large-scale production manufacturing methods for perovskite films (Figure 2d). The perovskite can be easily modified into ink to form a uniform perovskite film by a spray deposition process. In 2020, Koth Amratisha et al. developed the continuous spray deposition technique to realize the layerby-layer stacking structure of different perovskite materials (Figure 2e) [28]. The perovskite film stability can be well maintained in a humid environment owing to the spray deposition preparation method. Sequential spray deposition technology opens up a new way for perovskite film stacking design and mass production under economical ambient conditions.

2.4. Aerosol-Liquid-Solid Method

The aerosol–liquid–solid method is a streamlined circulation from aerosol to liquid to solid, which can be used for the preparation of perovskite film. A suite of technological parameters including temperature, aerosol supply rate, and composition can be precisely controlled for aerosol–liquid–solid method technology. Due to surface tension and viscosity limitations, it is tough to deposit very thick wet films on the substrate through traditional solution methods such as spin-coating methods; however, the aerosol–liquid–solid method could achieve dense, high crystallinity, low defect density thick perovskite film around 1 mm (Figure 2f). In 2021, Wei Qian et al. demonstrated aerosol–liquid–solid methods to enable the continued growth of homogeneous perovskite films [29]. The thickness and grain size of the film increase gradually with the increase in growth time, as shown in Figure 2g.

2.5. Solution Growth Method

The solution growth method has the advantages of a low-cost, large-scale, and fast growth rate, which is suitable for the growth of perovskite single crystal. In 2016, Haotong Wei et al. successfully fabricated solution-grown MAPbBr₃ single crystals with extraordinary optoelectronic properties [30]. In 2020, Xin Wang et al. achieved the solution-processed epitaxial growth of MAPbX₃ organic–inorganic hybrid perovskite single crystal with different halide components and no lattice mismatch [31]. Figure 3a shows the SEM images of the cross-section and surface characterization of the solution-processed epitaxial layer, revealing the growth of a single-crystal film from a small island into a flat film.

2.6. Scraper Method

The scraper method is one of the crafts for producing films. In 1952, a scraper method patent was obtained by scraping water-based and non-water-based slurry onto moving plasterboard. Recently, the scraper method has been applied in the fabrication of perovskite films. When a constant relative motion is established between the blade and the substrate, the perovskite slurry is spread over the substrate to form a thin perovskite film. The scraper

can run at a speed of several meters/per min. This operation mode is suitable for coating wet perovskite films with a thickness of tens of microns to hundreds of microns [32,33]. In 2017, Yong Churl Kim et al. used a scraper method to fabricate an 830 μ m-thick polycrystalline MAPbI₃ layer [34]. In 2022, Mengling Xia et al. applied a combination of the scraper method and a soft pressure-assisted cryogenic solution treatment to make high-quality films for thin-film transistor (TFT) integration (Figure 3b) [35]. The method realizes the compatibility between the perovskite material and the TFT substrate and helps to achieve compact perovskite films with a smooth surface and passivated grain boundaries.

2.7. Inkjet Printing Method

Inkjet printing technology represents a highly established method for thin-film manufacturing with applications in electronics, optics, bioengineering, and other fields [36]. This technology has the advantages of low manufacturing costs, a simple and flexible process, and no mask plate and lithography, which can be widely used in perovskite solution printing [37,38]. Figure 3c illustrates the basic components of an inkjet printer including an ink chamber, conducting nozzle, substrate, and translational stage [39]. When a voltage is applied between the substrate and the nozzle, a curved liquid surface is formed at the nozzle under the action of the induced electric field force [40]. As the voltage gradually increases, small droplets are formed and ejected onto the substrate (Figure 3d) [41]. The movement of the nozzle and the mode of the applied voltage can control the pattern printed on the substrate [39–41]. In 2019, Jingying Liu et al. homogeneously prepared perovskite films on different nature substrates by the inexpensive inkjet printing method (Figure 3e) [42]. The cheap and simple inkjet printing method enables the large-scale manufacturing of multi-channel arrays of perovskite-based X-ray detectors.

2.8. Laser Direct Writing

Laser direct writing is a common method for preparing nanostructured graphics. Laser direct writing technology is widely used in all-inorganic halide perovskite colloidal quantum dots. In 2017, Chen Jun et al. proposed a rapid preparation method of perovskite colloidal quantum-dot film based on laser direct writing. It is simple, fast, and does not require a mask [43]. It includes three steps: spin-coating perovskite colloidal quantum dots, laser writing, and solvent washing (Figure 3f). Moreover, a large-scale (100 mm \times 100 mm) perovskite colloidal quantum-dot patterning is demonstrated by the laser direct writing technology.

2.9. Melt and Crystallization

Perovskite has various forms such as nanocrystals, nanowires, and polycrystalline films, among which perovskite single crystal is the more stable form. Melt and crystallization are the preferred methods to obtain perovskite crystals. In this way, high-quality perovskite single crystals can be grown from the melt, which can meet the requirements of large diameters and cause less pollution during the synthesis process. In 2022, Andrii Kanak et al. studied the melting and crystallization process of all-organic perovskite CsPbBr₃, obtained massive single crystals from the melt, discussed the phase transition mechanism of the whole process, and proposed the two-stage melting mechanism of CsPbBr₃ perovskite. The effect of heating and cooling conditions on the crystallization process of large particle CsPbBr₃ was studied. This study is helpful to further understanding the crystal structure and crystallization mechanism of perovskite. It is important for understanding the growth and high-quality preparation of perovskite crystals and massive perovskite materials [44].



Figure 3. Solution growth, scraper, inkjet printing, and laser direct writing methods. (**a**) SEM photos of epitaxial layer cross-section and solution treatment at different growth times. Reprinted with permission from Ref. [31]. 2020, American Chemical Society. (**b**) Scraper method and a soft pressure-assisted cryogenic solution treatment. Reprinted with permission from Ref. [35]. 2022, Wiley. (**c**) Schematic of inkjet printer. Reprinted with permission from Ref. [39]. 2012, IOP Publishing. (**d**) Schematic illustration of inkjet printing system. Reprinted with permission from Ref. [41]. 2015, Royal Society of Chemistry. (**e**) Schematic diagram of a perovskite-based device manufactured by inkjet printing. Reprinted with permission from Ref. [42]. 2019, Wiley. (**f**) Schematic diagram of the modeling process and mechanism of perovskite colloidal quantum dots. Reprinted with permission from Ref. [43]. 2017, Wiley.

3. Halide Perovskite-Based X-ray Detector

Recently, there have been many high-impact review papers on introducing perovskitebased X-ray detectors. Zhizai Li et al. reviewed the current research progress on halide perovskite-based X-ray detectors [45]. Joydip Ghosh et al. summarized recent efforts on lead-free double perovskite-based X-ray detectors [46]. Haodi Wu et al. advised scalable fabrication of metal halide perovskite-based X-ray detectors [47].

Halide perovskite as an emerging X-ray detection material has a long carrier lifetime, large atomic numbers, and high X-ray attenuation coefficient [23,48]. An X-ray detector based on halide perovskite material possesses high detection sensitivity, which is crucial for detecting weak X-ray dose rates and greatly reducing the risk of medical examination [45]. Moreover, the synthesis process of halide perovskite is simple and low-cost as previously mentioned, which could form large-area flat panel detector arrays at low cost. Halide perovskite-based X-ray detector with high sensitivity and low cost has great application prospects in security, defense, medical imaging, industrial materials inspection, and nuclear power plants [49–51].

There are various structural dimensionalities of perovskites by controlling appropriate organic and inorganic components, such as 3D and low-dimensional (2D, 1D, 0D) perovskites. In recent years, various halide perovskite-based X-ray detectors have been developed by utilizing different dimensional perovskites. The three main types of halide perovskite-based X-ray detectors are described in detail below, including 3D, low, and 3D/low-dimensional mixed perovskite-based X-ray detectors. The summarization of the performance of X-ray detectors based on halide perovskite materials is shown in Table 2.

3.1. 3D Perovskites-Based X-ray Detectors

3D perovskites have a universal formula of ABX₃ (where A is a monovalent organic or inorganic cation, B is a divalent cation, and X is halide anion) and are composed of continuous corner-sharing metal halide $[BX_6]^{4-}$ octahedra [52]. According to the organic and inorganic components, 3D perovskites-based X-ray detectors are mainly divided into two representative types: organic–inorganic hybrid perovskites and all-inorganic perovskites-based X-ray detectors.

3.1.1. Organic–Inorganic Hybrid Perovskites-Based X-ray Detectors

Organic–inorganic hybrid perovskites have been shown to have good optoelectronic properties for X-ray detection, including a large appropriate band gap (1.6–3.0 eV) and a large $\mu\tau$ product on the order of ~10⁻² cm² V⁻¹ [53]. In 2017, Wei Wei et al. fabricated a MAPbBr₃ single-crystal X-ray detector and integrated it onto the Si substrates. This allows the electrical signal to be read directly from the Si (Figure 4a) [53]. The brominated (3-amino-propyl) triethoxy-silane (NH₃Br) end molecular layer is used to mechanically and electrically connect the MAPbBr₃ single crystals to the Si. No lattice matching with the Si substrate is required. The sensitivity of the MAPbBr₃ single-crystal X-ray detector is 2.1 × 10⁴ μ C Gy_{air}⁻¹ cm⁻² under 8 keV X-ray radiation, which is more than 1000 times that of the commercially available amorphous α -Se detector. As shown in Figure 4b, it is possible to achieve X-ray imaging at a low X-ray (8 keV) dose rate of <0.1 μ Gy_{air} s⁻¹.

Although halide perovskite X-ray detectors have had great success, the high concentration of Pb in perovskite poses a serious threat to human and biological systems due to its high water solubility. "Green" bi-element with the same electronic structure as the Pb attracts wide attention. In 2017, Weicheng Pan et al. demonstrated sensitive X-ray detectors using solution-processed double perovskite $Cs_2AgBiBr_6$ single crystals [54]. The $Cs_2AgBiBr_6$ single crystals use one Ag^+ and one Bi^{3+} to substitute two toxic Pb^{2+} in $CsPbBr_3$, which are very friendly to humans and biological systems (Figure 4c,d). In addition, after heat annealing and surface treatment, the Ag^+/Bi^{3+} disordering is eliminated and the crystal resistivity is enhanced. The $Cs_2AgBiBr_6$ single crystals exhibit a higher resistivity ($10^9-10^{11} \Omega$ cm) than MAPbX₃ (X = Cl, Br, I; $10^7-10^8 \Omega$ cm). As a result, the $Cs_2AgBiBr_6$ single-crystal X-ray detectors possess a low minimum dose rate of 59.7 nGy_{air} s⁻¹ under



5 V and a high sensitivity of 8 μ C Gy_{air}⁻¹ cm⁻² (1 V bias) to 30 keV X-ray photons and 105 μ C Gy_{air}⁻¹ cm⁻² at 50 V (Figure 4e).

Figure 4. 3D organic–inorganic hybrid perovskites-based X-ray detectors. (a) Preparation diagram of Si-integrated MAPbBr₃ single crystal. Reprinted with permission from Ref. [53]. 2017, Springer Nature. (b) This device was used for the X-ray imaging in stacked glass coverslips; a stainless-steel plate with etched-through lines, a wrapped metal spring, a section of the tail fin, and an "N" copper sign gave a dose rate of 247 nGy_{air} s⁻¹. Reprinted with permission from Ref. [53]. 2017, Springer Nature. (c) X-ray diffraction of Cs₂AgBiBr₆ single crystal. Illustration: Crystal structure. The small wine ball represents Br⁻, the large khaki ball represents Cs⁺, and the light green and purple octahedra represent the AgBr₆ and BiBr₆ octahedra, respectively. Reprinted with permission from Ref. [54]. 2017, Springer Nature. (d) Thermogravimetric analysis of Cs₂AgBiBr₆. Included is a photograph of a solution-processed Cs₂AgBiBr₆ single crystal. Reprinted with permission from Ref. [54]. 2017, Springer Nature. (e) The signal-to-noise ratio of the device is calculated from the standard deviation of the X-ray photocurrent. The red dashed line represents a signal-to-noise ratio of 3, and thus the detection limit is 59.7 nGy_{air} s⁻¹ at 5 V bias; it is represented by purple stars surrounded by red dotted circles. Reprinted with permission from Ref. [54]. 2017, Springer Nature.

3.1.2. All-Inorganic Perovskites-Based X-ray Detectors

Although the hybrid organic–inorganic perovskite-based X-ray detectors have realized outstanding performance, the stability of hybrid organic–inorganic perovskite is poor. This has an impact on their actual use, hindering their use in practical applications. The environmental conditions of their storage, fabrication, and device operation are very strict due to the extreme sensitivity to both oxygen and moisture. In addition, they are also unstable to light and heat due to the influence of organic groups. All-inorganic cesium halide lead perovskite (CsPbX₃) without organic components is considered to be the next generation of X-ray detector materials due to its superior stability to hybrid perovskite.

In 2018, Jin Hyuck Heo et al. prepared a CsPbX₃ perovskite nanocrystalline X-ray detector that is easy to commercialize and cost-effective (Figure 5a) [55]. The acquired X-ray detectors exhibit high stability over X-ray irradiation of 40 Gy_{air} s⁻¹. Moreover, the X-ray imaging possesses an excellent spatial resolution of 9.8 lp mm⁻¹ at modulation transfer function (MTF) = 0.2 and 12.5–8.9 lp mm⁻¹ for a linear line chart, which is higher than commercial terbium-doped gadolinium oxysulfide (GOS) detectors of spatial resolution = 6.2 lp mm⁻¹ at MTF = 0.2 and 6.3 lp mm⁻¹ for a linear line chart (Figure 5b,c).

In 2019, Zongyan Gou et al. fabricated X-ray detectors based on the CsPbX₃ microcrystal thick film. The multi-dissolution and recrystallization method is used to further improve the photoelectric performance of the detector (Figure 5d) [26]. The sensitivity of CsPbX₃-based X-ray detectors is 470 μ C Gy_{air}⁻¹ cm⁻² at zero bias under a remarkably low dose rate (0.053 μ Gy_{air} s⁻¹), which is over 20 times higher than that of α -Se X-ray detectors working at a much higher field of 10 V μ m⁻¹ (Figure 5e,f).

3.2. Low-Dimensional Perovskites-Based X-ray Detectors

Although 3D perovskites-based X-ray detectors achieve high sensitivity, the phase transformation and instability of 3D perovskites limit their development. To address these issues of 3D perovskites, low-dimensional perovskites have been developed and high-performance low-dimensional perovskites-based X-ray detectors have been realized.

3.2.1. Organic-Inorganic Hybrid Perovskites-Based X-ray Detectors

 $(NH_4)_3Bi_2I_9$ belongs to organic–inorganic hybrid perovskite-related materials $A_3M_2X_9$ (A = Cs, Rb, NH₄; M = Bi, Sb; X = Br, I). The $(NH_4)_3Bi_2I_9$ has a 2D layered structure where the BiI₆ octahedra corners share each other and are stacked in a close-packed fashion in the (001) plane (Figure 6a) [56]. In 2019, Renzhong Zhuang et al. verified the 2D $(NH_4)_3Bi_2I_9$ without toxic elements is a suitable material to obtain a high-performance X-ray detector [56]. Layered $(NH_4)_3Bi_2I_9$ grows easily in low-temperature solutions and can be cut along cleavage planes (Figure 6b,c). Two types of $(NH_4)_3Bi_2I_9$ -based X-ray detectors with parallel and perpendicular device structures were proposed, as shown in Figure 6d. The devices were exposed to a source with X-ray photon energy up to 50 keV and with a peak intensity of 22 keV. The sensitivity of the parallel direction device is high at 8.2 × $10^3 \ \mu C \ Gy_{air}^{-1} \ cm^{-2}$, which is higher than that of the perpendicular one (803 $\ \mu C \ Gy_{air}^{-1} \ cm^{-2}$). Moreover, the perpendicular direction device possesses a lower detection limit of 55 nGy_{air} s⁻¹ compared to the parallel one of 210 nGy_{air} s⁻¹ (Figure 6e).

3.2.2. All-Inorganic Perovskites-Based X-ray Detectors

0D perovskite quantum dots are ideal candidates for X-ray detectors and large-area flat or flexible panels with great application potential. CsPbBr₃ materials as a typical all inorganic perovskite possess superior stability and excellent optoelectronic properties. In 2019, Jingying Liu et al. synthesized 0D high-quality colloidal CsPbBr₃ perovskite quantum dots via the hot injection method at room temperature and demonstrated a flexible, printable CsPbBr₃-based X-ray detector (Figure 7a–c) [42]. Perovskite quantum-dot films are printed evenly on a variety of substrates using an inexpensive inkjet printing method to demonstrate large-scale manufacturing of X-ray detector arrays. The device was exposed to a synchrotron soft X-ray beamline with photon energies ranging from 0.1 to 2.5

keV. The detector has a sensitivity of 1450 μ Gy_{air}s⁻¹ cm⁻² at 0.0172 mGy_{air}s⁻¹ X-ray dose rate and a bias voltage of only 0.1 V, which is 70 times more sensitive than the α -Se device. Moreover, the CsPbBr₃ quantum-dot-based X-ray detectors possess outstanding flexibility and durability (Figure 7d–f).



Figure 5. 3D all-inorganic perovskites-based X-ray detectors. (a) Schematic diagram of the structure of a CsPbBr₃ perovskite nanocrystalline scintillation X-ray detector. Reprinted with permission from Ref. [55]. 2018, Wiley. (b) Photographic and X-ray images for nondestructive inspection. Images of a normal ballpoint pen (**left**) and defective ballpoint pen (**right**) with identical appearance, ballpoint pen X-ray images from conventional GOS scintillator X-ray detectors, ballpoint pen X-ray images from CsPbBr₃ scintillator X-ray detectors, ballpoint pen X-ray images from GOS and CsPbBr₃ scintillator X-ray detectors. Reprinted with permission from Ref. [55]. 2018, Wiley. (c) MTF of the original silicon-based detector, CsPbBr₃, and conventional GOS scintillation X-ray detector. Reprinted with permission from Ref. [55]. 2018, Wiley. (d) Thick film radiation detector structure. The charge-producing region lies inside the thick film and is used for X-ray excitation. Reprinted with permission from Ref. [26]. 2019, Wiley. (e) Zero bias light response of raw and treated film at different dose rates. Reprinted with permission from Ref. [26]. 2019, Wiley. (f) The sensitivity of a device under different bias voltages and fixed emissivity (0.053 μ Gy_{air} s⁻¹). Reprinted with permission from Ref. [26]. 2019, Wiley.



Figure 6. Low-dimensional organic–inorganic hybrid perovskites-based X-ray detectors. Reprinted with permission from Ref. [56]. 2019, Springer Nature. (**a**) Crystal structure of $(NH_4)_3Bi_2l_9$ along the c and b axes. (**b**) Photograph of a bulk $(NH_4)_3Bi_2l_9$ single crystal. (**c**) Procedure for crystal growth. (**d**) Illustration of parallel and perpendicular device structures. (**e**) The signal-to-noise ratio of the device in directions parallel to and perpendicular to the (001) surface. The dashed blue line indicates a signal-to-noise ratio of 3, so the detection limits are 210 nGy_{air} s⁻¹ for parallel devices and 55 nGy_{air} s⁻¹ for vertical devices.

3.3. Three-Dimensional/Low-Dimensional Hybrid Perovskites-Based X-ray Detectors

3D perovskites such as MAPbI₃, MAPbBr₃, and CsPbBr₃ are conducive to achieving high sensitivity due to their long carrier lifetime. However, baseline drifting problems arising from ion migration in the 3D perovskite material prevent accurate signal recording, hindering practical applications. Low-dimensional perovskites such as 2D BDAPbI₄ (BDA = $NH_3C_4H_8NH_3$) and (NH_4)₃Bi₂I₉ can effectively solve the baseline drifting prob-

lems. However, the sensitivity of low-dimensional perovskite is limited by its poor carrier transport performance. Thus, the combination of the 3D and low-dimensional perovskites could be an effective choice beyond the performance of X-ray detection with individual dimension perovskites.



Figure 7. Low-dimensional all-inorganic perovskites-based X-ray detectors. Reprinted with permission from Ref. [42]. 2019, Wiley. (a) Schematic diagram of a perovskite-based device manufactured by inkjet printing. (b) Photography of an X-ray detector array on a 4-inch wafer. (c) Optical imaging equipment with the printing of quantum-dot film. Scale: 5 μ m. Illustration: Low-power image. Scale: 50 μ m. (d) Chemical reactions of flexible perovskite-based X-ray detector arrays on polyethylene terephthalate (PET) substrates. (e) Photography of flexible devices under bending. (f) Current and voltage curves of flexible device arrays at different bending angles under X-ray irradiation with 7.33 mGv_{air}s⁻¹ and 0.1 V bias voltages. Inset: Real device showing 120° bending.

In 2021, Xiuwen Xu et al. developed a double-layer perovskite film with a properly aligned energy level, where 2D (PEA)₂MA₃Pb₄I₁₃ (PEA = 2-phenylethylammonium, MA = methylammonium) is cascaded with vertically crystallized 3D MAPbI₃ (Figure 8a–d) [57]. Combining the fast carrier transport of the 3D layer and mitigated ion migration of the 2D layer, X-ray detectors provide high sensitivity and very stable baselines. Moreover, the 2D layer increases the film resistivity and enlarges the energy barrier for the hole injection without compromising carrier extraction (Figure 8e–g). As a result, the obtained double-layer perovskite detector exhibits a high sensitivity ($1.95 \times 10^4 \,\mu\text{C Gy}_{air}^{-1} \,\text{cm}^{-2}$) and a low detection limit (480 nGy_{air} s⁻¹) (Figure 8h). The double-layer perovskite detector exhibits excellent stability and a highly repeatable X-ray response when the X-ray is periodically on/off (Figure 8i). X-ray images of the printed circuit boards obtained with a double-layer perovskite detector with an active area of 3 mm × 3 mm are shown in Figure 8j.



Figure 8. Three-dimensional/low-dimensional hybrid perovskites-based X-ray detectors. Reprinted with permission from Ref. [57]. 2021, Wiley. (a) Schematic diagram illustrating the structure of a double-layer perovskite thin film, where gradient 2D perovskite was prepared from a precursor solution with the nominal component (PEA)₂MA₃Pb₄I₁₃. (b–d) SEM image of double-layer perovskite film, shown in (b) is the grain size distribution. (e) Double-layer perovskite X-ray detector equipment configuration: Carbon electrodes were exposed to incident X-ray photons and negatively biased, whereas gradient 2D perovskite was prepared from a precursor solution with the designated component (PEA)₂MA₃Pb₄I₁₃. (f) Control the interfacial level alignment of a device made of a double-layer perovskite film. (g) Band edge position diagram of two-dimensional perovskite and three-dimensional MAPbI₃ with respect to fluorine-doped SnO₂ (FTO) and carbon. (h) The dose-rate-dependent signal-to-noise ratio of control and double-layer perovskite detectors. (i) Stability test of double-layer perovskite detector under different dose rate X-ray irradiation. (j) A digital photograph of the printed circuit board (size: $5.1 \text{ cm} \times 5.1 \text{ cm}$) and X-ray images of a printed circuit board obtained with a double-layer perovskite detector with an active area of 3 mm \times 3 mm.

Table 2. The summarization of the performance of halide perovskite-based X-ray detectors.

Туре	Material	Fabrication Method	Film Thickness (µm)	Sensor Area (mm ²)	Sensitivity (µC Gy _{air} ⁻¹ cm ⁻²)	Minimum Detectable Dose Rate (µGy _{air} s ⁻¹)	Ref.
3D perovskites-based X-ray detectors	MAPbBr ₃	Solution-processed molecular bonding method	150	0.044	21,000	<0.1 (-1 V)	[53]
		Solution growth method	2000	-	80	0.5 (-0.1 V)	[30]
	Cs ₂ AgBiBr ₆	Solution-processed method	1180	3.14	105 (50 V)	0.0597 (5 V)	[54]
	CsPbBr ₃	Laser direct writing	-	10,000	-	-	[43]
		Dissolution and recrystallization method	18	50	470	0.053 (2 V)	[26]
		Melt and crystallization	-	-	-	-	[44]

Туре	Material	Fabrication Method	Film Thickness (µm)	Sensor Area (mm ²)	Sensitivity (µC Gy _{air} ⁻¹ cm ⁻²)	Minimum Detectable Dose Rate (µGy _{air} s ⁻¹)	Ref.
	CH ₃ NH ₃ PbI ₃	Spin-coating process	10–100	-	-	-	[27]
	CsPbI ₂ Br	Aerosol–liquid– solid method	One micrometer- hundreds of micrometers	10,000	148,000	$0.28 (0.125 V \mu m^{-1})$	[29]
	MAPbI ₃	Scraper method	400	78,400	17,432 (500 V mm ⁻¹)	$0.067 (5 \text{ V mm}^{-1})$	[35]
Low-dimensional perovskites-based X-ray detectors	(NH ₄) ₃ Bi ₂ I ₉	Solution growth method	4900 (parallel); 1500 (perpendicular)	6.75 (parallel); 22 (perpendicular)	8200 (parallel); 803 (perpendicular)	0.21 (parallel 1 V); 0.055 (perpendicular 10 V)	[56]
	CsPbBr ₃	Inkjet printing method	0.6	0.06	1450	17.2 (0.1 V)	[42]
3D/low- dimensional hybrid perovskites-based X-ray detectors	2D (PEA) ₂ MA ₃ Pb ₄ I ₁₃ /3D MAPbI ₃	Spray deposition process	0.5	9	19,500	0.48 (<25 V mm ⁻¹)	[57]

Table 2. Cont.

4. Challenges and Perspectives

A variety of materials have been studied for X-ray detection because X-ray detectors have a wide range of applications including medical imaging, nondestructive industrial inspection, and safety screening. Diamond is considered the elective material for X-ray detection, mainly due to its "tissue equivalence" [10–13]. Recently, halide perovskites are emerging candidates in X-ray detection owing to the high X-ray attenuation coefficients, large $\mu\tau$ product, high atomic number, and the simple and low-cost synthesis process. Although halide perovskite-based X-ray detectors have made many remarkable signs of progress in the past few years, there are still some important problems that need further consideration to meet application requirements. The existing challenges are discussed, and the possible directions for future applications are explored in the following.

Firstly, compared with traditional materials such as Si, α -Se, and Ge, the stability of perovskite materials is still relatively poor, which seriously restricts their practical application. Therefore, more attention should be paid to improving the stability of perovskite materials in future research.

Secondly, the thickness of perovskite films with hundreds of micrometers is the prerequisite for complete X-ray attenuation. However, the fabrication of a thick perovskite film with good uniformity is still a challenge. Moreover, the cost of preparation of perovskitebased X-ray detectors should be further reduced. Hence, developing new preparation methods or optimizing the original methods is the future development trend.

Thirdly, the sensitivity and minimum detectable X-ray dose rate of perovskite-based X-ray detectors are not good enough for practical use. It is necessary to develop high-performance perovskite-based X-ray detectors with high sensitivity and low detection limits.

Finally, the development of large-area flexible next-generation X-ray detection and imaging technology based on perovskite is the future research direction.

Author Contributions: Conceptualization, X.T.; investigation, Y.T. and G.M.; writing—original draft preparation, Y.T. and G.M.; writing—review and editing, Y.T., G.M. and X.T.; supervision, X.T.; project administration, X.T.; funding acquisition, X.T., G.M. and M.C. All authors have read and agreed to the published version of the manuscript.

Funding: This work was funded by the National Key R&D Program of China (2021YFA0717600), the National Natural Science Foundation of China (NSFC No. 62035004 and NSFC No. 62105022) and the China Postdoctoral Science Foundation (2022M710396). X.T. is sponsored by the Young Elite Scientists Sponsorship Program by CAST (No. YESS20200163).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

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