

# Supplementary Materials: Facile Fabrication of F-doped SnO<sub>2</sub> Nanomaterials for Improved Photocatalytic Activity

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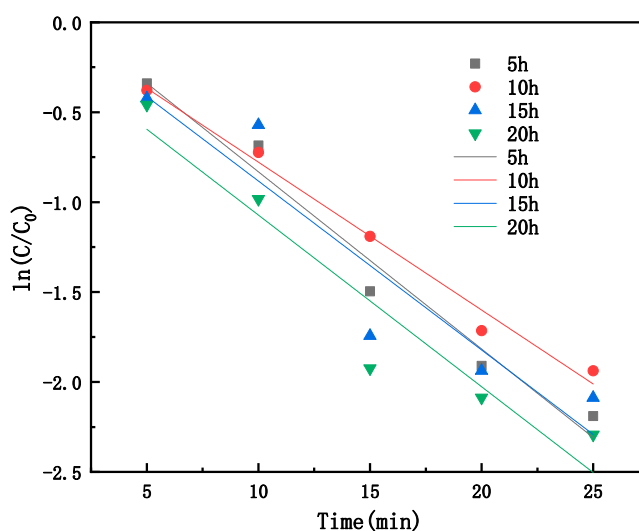
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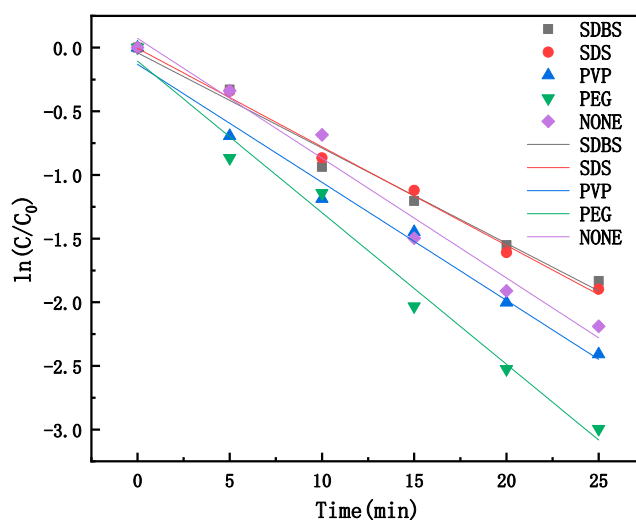
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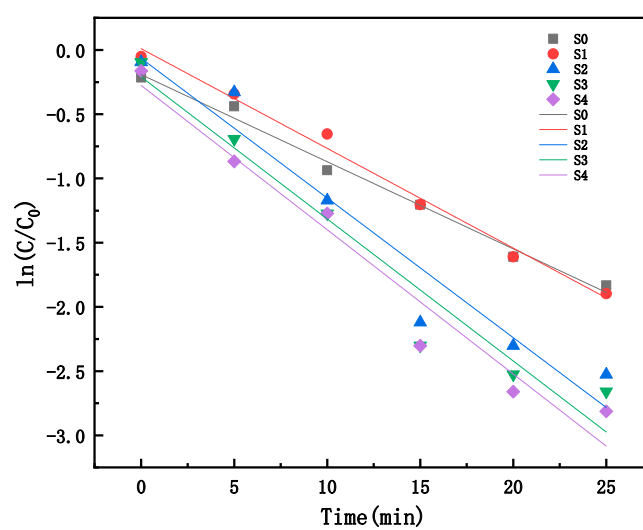
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(a)



(b)



(c)

**Figure S1.** pseudo-first-order kinetics of Rh B, (a) different solvothermal time (h); (b) different surfactants; (c) F/Sn.

**Table S1.** First-order kinetics fitting Equation and correlation coefficient ( $R^2$ ) for Different Solvothermal times.

Different Solvothermal Time (h)	Equation	The Correlation Coefficient ( $R^2$ )
5	$y = -0.09849x + 0.1532$	0.96041
10	$y = -0.08221x + 0.04404$	0.98316
15	$y = -0.09391x + 0.05623$	0.83797
20	$y = -0.09543x - 0.11787$	0.88489

**Table S2.** First-order kinetics fitting Equation and correlation coefficient ( $R^2$ ) for different surfactants.

Surfactants	Equation	The Correlation Coefficient ( $R^2$ )
SDBS	$y = -0.07482x - 0.04038$	0.98055
SDS	$y = -0.07738x - 0.00521$	0.99196
PVP	$y = -0.09273x - 0.13069$	0.98329
PEG	$y = -0.1191x - 0.1052$	0.98186
NONE	$y = -0.09411x + 0.07295$	0.97305

**Table S3.** First-order kinetics fitting Equation and correlation coefficient ( $R^2$ ) for different F/Sn ratio.

F/Sn	Equation	The Correlation Coefficient ( $R^2$ )
S0	$y = -0.06783x - 0.19131$	0.98756
S1	$y = -0.0776x + 0.01029$	0.98792
S2	$y = -0.10873x - 0.06459$	0.92503
S3	$y = -0.11058x - 0.20904$	0.92975
S4	$y = -0.11234x - 0.27547$	0.94825

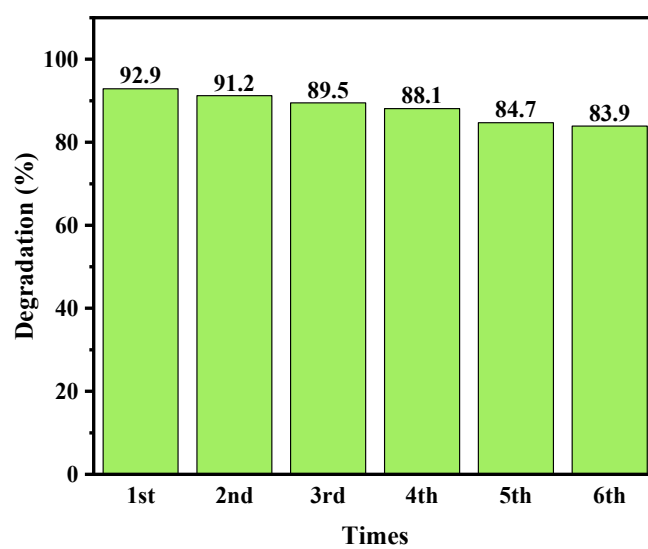


Figure S2. The reusability of the S4.