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# Study on the Electrical, Structural, Chemical and Optical Properties of PVD Ta(N) Films Deposited with Different N<sub>2</sub> Flow Rates

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Abstract: By reactive DC magnetron sputtering from a pure Ta target onto silicon substrates, Ta(N) films were prepared with different N<sub>2</sub> flow rates of 0, 12, 17, 25, 38, and 58 sccm. The effects of N<sub>2</sub> flow rate on the electrical properties, crystal structure, elemental composition, and optical properties of Ta(N) were studied. These properties were characterized by the four-probe method, X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), and spectroscopic ellipsometry (SE). Results show that the deposition rate decreases with an increase of N<sub>2</sub> flows. Furthermore, as resistivity increases, the crystal structure transitions from  $\beta$ -Ta to TaN(111), and finally becomes the N-rich phase Ta<sub>3</sub>N<sub>5</sub>(130, 040). Studying the optical properties, it is found that there are differences in the refractive index (n) and extinction coefficient (k) of Ta(N) with different thicknesses and different N<sub>2</sub> flow rates, depending on the crystal size and crystal phase structure.

**Keywords:** tantalum nitride; electrical properties; structural properties; elemental composition; spectroscopic ellipsometry; optical properties

## 1. Introduction

Transition metal nitrides, especially tantalum nitride (TaN), are in high demand for a wide range of applications due to their high melting point, hardness, excellent wear and corrosion resistance, refractory character, mechanical and high-temperature stability, chemical inertness, and histocompatibility [1–6]. Some prominent examples of such applications are as a protective coating material against oxidation and corrosion [7], as a diffusion barrier for Al and Cu metallization in advanced microelectronics [8–11], in phosphide and nitride optoelectronics as ohmic contact [3,4], in artificial heart valves as histocompatibility materials [12], thin film resistors [13], as ceramic pressure sensors [14], and also different mechanical applications [5,6]. The large interest for TaN arises since it is considered recently as a high thermal conductive material in microelectronic chips for the  $\theta$ -TaN phase [15].

Furthermore, TaN belongs to the class of complex nitride phases of compounds which can vary from stoichiometry due to lattice vacancies. Diversifications in phase structure are common in TaN, from hexagonal-Ta<sub>2</sub>N, cubic-TaN, hexagonal-TaN, hexagonal-Ta<sub>5</sub>N<sub>6</sub>, tetragonal-Ta<sub>4</sub>N<sub>5</sub>, and orthorhombic-Ta<sub>3</sub>N<sub>5</sub> including TaN, which led to large differences in physical, chemical, and mechanical properties. The TaN alloy forms a variety of phases depending on the deposition technique and growth conditions [16,17]. Numerous reports have been published to characterize sputtered TaN films based on various sputtering parameters such as nitrogen (N<sub>2</sub>) partial pressure ratio [18,19], N<sub>2</sub>/Ar flow rate ratio [20–22], sputtering power [23], and substrate temperature [24] during



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**Copyright:** © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). deposition. By controlling these different parameters, their influence on the structural, chemical, electrical, and optical properties of the TaN film have been investigated.

Among them, there are quite a few studies on how the N<sub>2</sub> flow or N<sub>2</sub>/Ar flow rate ratio and the N<sub>2</sub>/(N<sub>2</sub> + Ar) partial pressure ratio affect the properties of TaN film. Chen et al. [25] used a magnetron sputtering low-power radio frequency deposition method with variable nitrogen flow rate to deposit TaN<sub>x</sub> barrier layers on silicon. They found that as the N<sub>2</sub> flow rate increases, the surface roughness of the deposited TaN<sub>x</sub> film is slightly increased, and the amorphous structure of TaN<sub>x</sub> is formed with good thermal stability. Zaman et al. [26] prepared a TaN film with a 3% to 25% N<sub>2</sub>/(N<sub>2</sub> + Ar) ratio on Si substrate by reactive magnetron sputtering and studied the effect of N<sub>2</sub> partial pressure on the crystal structure and hardness of the TaN film. It was found that the deposited films with 5% and 3% N<sub>2</sub> content show the highest hardness (33 Pa).

Although sputtered TaN films have been widely investigated for their different properties, their optical properties have not yet been analyzed as much. Recent studies show that spectroscopic ellipsometry (SE) can be used to characterize and measure thin film thickness because of its fast and non-destructive nature [2,27–29]. Aouadi et al. [2] have studied the effects of varying N<sub>2</sub> flow rates from 1 to 4 sccm on the structural and optical properties of TaN thin films. They report that the optical constants ( $\varepsilon_1$ ,  $\varepsilon_2$ ) may be used in conjunction with real-time SE to monitor and control the growth of tantalum nitride films. Cherfi et al. [30] deposited TaN films with an  $N_2$  flow rate of 0–12 sccm on Si (100) and glass substrates by DC magnetron sputtering to show the influence of nitrogen flow on the crystal structure and optical properties of TaN. It was shown that for low  $N_2$  flow (1 and 2 sccm), the TaN films show good conductor performance; a further increase of N<sub>2</sub> flow shows non-metallic behavior. At the same time, samples with similar structural properties have similar behaviors in terms of optical properties. Waechtler et al. show that SE can be used to examine the optical properties of Ta and TaN thin films from 75 to 380 nm thickness. They found a good agreement of optical properties with narrow-band data available for similar thin films. It was also shown that the optical properties of the films strongly depend on both substrate and film thickness [27]. Ma et al. studied the temperature-dependent dielectric function of TiN films by SE [28]. Recently, Xu et al. [31] used the method of comparing the measured the refractive index of low-k film under the Ta(N) diffusion barrier with the refractive index of the blank low-k film to study the integrity of the Ta(N) diffusion barrier using the approach developed by Shamiryan et al. [32]. The abbreviation Ta(N), used hereinafter, refers to the complex films containing both Ta and TaN with different nitrogen concentrations. The Ta/TaN stacks are used, for instance, as metal diffusion barriers in advanced microelectronics.

However, there has been limited study of the optical behavior of the different stoichiometries of thicker TaN films by SE with varying  $N_2$  flow rates in conjunction with electronic, structural, and chemical composition. The study of the optical properties of TaN can provide us with more information about TaN films and some potential possibilities for the development of new applications. Thus, the systematic study of TaN films for understanding the electrical, structural, chemical composition, and optical properties are required.

Therefore, we first focused our attention on the optical properties of TaN films by SE, a non-destructive testing method. By examining the optical properties of the sample, it can provide some guidance for the deposited sample. However, past studies did not focus on the influence of process conditions on either the refractive index (n) or extinction coefficient (k), which are important optical parameters for Ta and TaN films because n and k are regarded as "fingerprints of thin film materials". Then, the effects of deposition rate and N<sub>2</sub> flow on the deposition of Ta(N) films on the electrical, structural, and optical properties (n & k), as well as the elemental composition, of TaN films were studied by using the four-probe method, X-ray diffraction (XRD), and X-ray photoelectron spectroscopy. The observation of different phases and chemical composition evaluation observed by XRD, XPS were correlated with optical properties.

#### 2. Materials and Methods

Ta(N) films were synthesized using a standard magnetron sputtering (JS35-80G) system. The Ta(N) films were deposited on Si (100) wafers using magnetron sputtering of a Ta target (8.0 cm in diameter and 6.0 mm thick) of 99.95% purity. The substrate holder (located in the center of the chamber) was a 25 cm diameter plate, with rotation set to 10 rpm without heating the substrate for all of the depositions to improve the uniformity of the films. The target-to-substrate distance was 13.0 cm, and a negative bias was applied to the Ta target. After placing the Si (100) substrate in the deposition chamber, the chamber was evacuated to  $9.6 \times 10^{-4}$  Pa (by a turbo-molecular pump, Beijing Taiyueheng Vacuum Technology Research Institute, Beijing, China), the background vacuum was sufficient to ensure the vacuum required for Ta(N) film sputtering. Ar (99.999% purity) and  $N_2$ (99.999% purity) were introduced into the reaction chamber through a mass flow controller (Beijing Qixing Huachuang Electronics Co., Ltd., Beijing, China) and used as sputtering and reaction gases, respectively. The Ta target and the Si-substrate were sputter cleaned with Ar plasma before the Ta(N) films deposition for 5 min. Following cleaning, Ta(N) film was deposited at  $9.56 \times 10^{-4}$  Pa background pressure and 200 W DC power was applied in a mixture of Ar and N<sub>2</sub>.

To study the influence of different N<sub>2</sub> flow rates on the properties of the sputtered Ta(N) film, in all sputtering processes, the flow rate of Ar was fixed at 58 sccm with various N<sub>2</sub> flow rates from 0 to 58 sccm. The ratio of the reactive gas (N<sub>2</sub>, 0–58 sccm) to the sputter gas (argon, 58 sccm) were varied from 0 to 0.20, 0.29, 0.43, 0.65, and 1.00. The deposition time was also varied from 10 to 30 min to study whether the thickness of the film affects the optical properties, among other properties, of Ta(N) under each N<sub>2</sub> flow rate. However, when the N<sub>2</sub> flow was 12 and 17 sccm, the sputtering process commenced for only 10 min.

The thickness of the film was measured by a German Bruker Dektak step meter (Berlin, Germany). The resolution of the Dektak profilometer was 4 angstroms. The thickness was determined from the step height between the film and a masked substrate area. The resistivity of the sample was measured by a double-electric four-point resistance resistivity tester (FT-341) and was obtained from the current between two external probes and by measuring the voltage through the internal probes.

The crystallographic structure of the sputtered Ta(N) films was measured by an X-ray diffractometer (Rigaku Ultima IV, Tokyo, Japan) using a  $\theta$ -2 $\theta$  scan with a 1.54 Å wavelength Cu K $\alpha$  radiation, at room temperature, working at 40 kV and 30 mA, which recorded the diffraction intensity in the scattering angle range of 20°–60°.

X-ray photoelectron spectroscopy (XPS) was used to investigate the elemental composition and chemical states in Ta(N) films using PHI 5300 (PerkinElmer, Fremont, CA, USA) with an Mg K $\alpha$  (1253.6 eV) excitation source. This source was operated using a voltage of 12.5 kV and an emission current of 20 mA. The films were sputter cleaned in an Ar<sup>+</sup> environment with 89.45 eV pass energy for 5 min prior to measurement. Survey scans were conducted in the 0–1100 eV range.

The optical characterization of the films was carried out using a SENpro spectroscopic ellipsometer (SENTECH, Berlin, Germany) to measure phase ( $\Delta$ ) and amplitude ratio ( $\Psi$ ) changes in the reflected light. The incident angle was fixed at 70° in the spectral range from 400 to 1050 at 5 nm increments. In all cases, the ellipsometric data were processed using SpectraRay/3 software (SpectraRay/3 V2022-125) for the data analysis. By treating the  $\Psi - \Delta$  spectra, the refractive index (n) and extinction coefficient (k) of the corresponding Ta(N) films were extracted by using E. Kondoh ELLIPSHEET [33] for an infinitely thick film. The validity of this approach is proved below in the ellipsometry part of this paper. The specific details and film thicknesses are given in Table 1.

Film	Sputtering Time (min)	Argon Flow (sccm)	N <sub>2</sub> Flow (sccm)	Thickness (nm)	Sheet Resistance (Ω/sq)	Resistivity (μΩ·cm)
Та	10 20 30	58	0	223.18 449.63 664.35	8.67 4.39 2.92	193.5 197.4 193.9
TaN TaN	10 10	58 58	12 17	159.00 150.00	33.03 47.18	524.00 738.00
TaN	10 20 30	58	25	161.07 282.08 439.51	63.23 55.57 32.10	1018.4 1567.5 1410.8
TaN	10 20 30	58	38	129.84 246.84 399.00	2090.0 1020.0 174.83	27,136.6 25,177.7 6975.7
TaN	10 20 30	- 58 -	58 	97.64 195.93 305.19	160.70 7480 13,200	1568.9 146,555.6 402,850.8

**Table 1.** The thickness, sheet resistance, and resistivity of Ta(N) film deposited with different  $N_2$  flow rates and different sputtering times.

## 3. Results and Discussion

## 3.1. Deposition Rate and Resistivity

The deposition rate of Ta(N) films, sheet resistance, and resistivity depend on the flow of nitrogen. The film thickness decreases with an increasing flow of nitrogen, as shown in Figure 1a. The increase of sputtering time from 10 to 30 min makes the film thicker. However, the deposition rate does not depend on the sputtering time (Figure 1b). With the increase of nitrogen flow in the sputtering atmosphere of ionized Ar<sup>+</sup>, the intensity of ion bombardment of the Ta target decreases due to the reduction of the mean free path length. The number of sputtered Ta atoms also decreases, leading to a gradual reduction in deposition rate [34]. In addition, since there are a large number of active N atoms in the sputtering atmosphere (an increase of the N<sub>2</sub> flow rate), the number of active N atoms in the atmosphere gradually increases, which increases the chemical reaction between active N atoms and the surface of the Ta target [35,36] and thereby reducing the sputtering rate.



**Figure 1.** Change of Ta(N) film thickness measured by Dektak step meter (**a**) and the deposition rates (**b**) for different nitrogen flows and sputtering time.

The electrical resistivity of pure Ta films sputtered for 10, 20, and 30 min are nearly the same (193.5, 197.4, and 193.9  $\mu\Omega$ ·cm, respectively (Table 1)). It is interesting to notice that the measured resistivity for pure Ta films is similar to the values reported for tetragonal crystalline Ta ( $\beta$ -Ta) films (165  $\mu\Omega$ ·cm, Schauer et al. [37]; 210  $\mu\Omega$ ·cm, Cuong et al. [38])

and 242  $\mu\Omega$ ·cm, Arshi et al. [39]). Therefore, our pure Ta films are most likely  $\beta$ -Ta. The electrical resistivity of Ta(N) films deposited with different nitrogen flows is shown in Figure 2. It can be seen that the resistivity of TaN is higher than the resistivity of pure Ta observed at zero nitrogen flow. The introduction of nitrogen increases resistivity, changing linearly (embedded graph), with the thinner film offering lower resistivity Ta phase in the deposited Ta-N films and the increasing of the low resistivity N-rich TaN phase. A much more dramatic change of resistivity is observed in the films deposited with 58 sccm N<sub>2</sub> flow, especially after 30 min deposition (relatively thick films, see Table 1).



**Figure 2.** The resistivity TaN films at different nitrogen flow. The embedded graph is a zoom of data corresponding to the low nitrogen flow region.

Further increase of N<sub>2</sub> flow increases the resistivity of Ta(N) film and may depend on the formation of different surface topographies, grain sizes, changes of composition, amorphous structure formations, and defects/imperfections (scattering from grain boundary) [40]. The resistivity of TaN films in the current work is similar to the values reported in the literature [40,41]. When the  $N_2$  flow rate is 12 sccm, the resistivity of the sputtered TaN film is increased to 524  $\mu\Omega$ ·cm and close to FCC TaN [42–45], or cubic TaN(111) [39], or  $Ta_3N_5$  or  $Ta_5N_6$  [46–48]. After a further increase in the nitrogen flow to 25 sccm, the resistivity of the sputtered TaN film is close to the resistivity of Ta<sub>3</sub>N<sub>5</sub> (1126  $\mu\Omega$ ·cm, [39]). Remarkably, when the increase of  $N_2$  flow is increased to 38 and 58 sccm, the resistivity of the sputtered TaN films drastically increases, with the only exception being the 10 min deposition TaN films. Normally, this can be explained by increasing electron scattering from interstitial N atoms. [39]. However, this model does not explain so strong a difference between the films deposited with 38 and 58 sccm [49]. It is also well-known that an excess of  $N_2$  flow rate will decrease the mean free path of ionized Ta atoms, disturb the formation of TaN structures [40,50], and also increase electron scattering from interstitial N atoms. Therefore, the phase of the TaN film generated under our  $N_2$  flow rate will also be different. The existence of N-rich phases in the TaN films at higher nitrogen flows is consistent with both the XRD patterns and the XPS analysis.

#### 3.2. Structural Properties (XRD Analysis)

Figure 3 shows the X-ray diffraction (XRD) patterns of the Ta(N) films deposited for 10, 20, and 30 min with different N<sub>2</sub> flows in the gas mixture varying from 0 to 58 sccm. The XRD spectra of the Ta films (N<sub>2</sub> flow rate is 0) shows a mixed phase of  $\beta$ -Ta (221),  $\beta$ -Ta (002),  $\beta$ -Ta (330), and Ta (110). The diffraction intensity of (002) is the highest, and the peak area of the diffraction peak is the largest, which indicates that the Ta film that was sputtered is mainly  $\beta$ -Ta (002) preferred orientation (PDF#: 04-0788). Peaks at 35.40°, 37.04°, and 41.20° are indexed to be the TaN (111), TaN (111) [39], and TaN (200) (PDF#:



49-1283) structures respectively (Figure 3a). The peaks at  $31.86^\circ$ , and  $35.10^\circ$  correspond to Ta<sub>3</sub>N<sub>5</sub> (123) and Ta<sub>3</sub>N<sub>5</sub> (130) or (040) compounds respectively [51].

**Figure 3.** XRD patterns of Ta (20 min) and TaN films deposited for (**a**) 10 min, (**b**) 20 min, and (**c**) 30 min with different N<sub>2</sub> flows.

When the N<sub>2</sub> flow rate is 12 and 17 sccm, a mixed phase of TaN (111) and TaN (200) appears, and the diffraction peak of Ta<sub>3</sub>N<sub>5</sub> (023) also appears when the N<sub>2</sub> flow rate is 17 sccm. However, the diffraction peak of TaN (111) is the highest, and the diffraction peak area is also the largest, which indicates that the preferred orientation of the TaN film under these two flow rates is TaN (111). Similarly, when the N<sub>2</sub> flow rate is 38 and 58 sccm, the TaN film has a preferred orientation of Ta<sub>3</sub>N<sub>5</sub> (130,040). At the same time, when the N<sub>2</sub> flow rate increases, the diffraction peaks gradually widen, indicating that the grain size is gradually decreasing, leading to high resistivity (Table 1). Correspondingly, this could be attributed to the mixture of fcc TaN and its amorphous structure [46,52,53]. These phenomena observed by XRD are in line with our previous conjectures in the section on resistivity and are similar to those reported in an earlier study [43,44].

When the nitrogen flow rate is increased, the phase of the film evolves gradually from TaN (111) to Ta<sub>3</sub>N<sub>5</sub> (130), or Ta<sub>3</sub>N<sub>5</sub> (040) (35.40° to  $35.03^{\circ}$ ) [51]. Furthermore, the TaN (200) peaks are gradually decreasing. A broad peak corresponding to  $Ta_3N_5$  appears for the 58 sccm sample and significant broadening of the peaks may be due to the formation of a two-phase nanocomposite structure. This may be attributed to the high nitrogen fraction, which is known to inhibit the crystallization of nitrogen-rich  $TaN_x$  sputtered films. At the same time, the XRD patterns of TaN films with a nitrogen flow rate of 25–58 sccm and a sputtering time of 10–30 min were compared (Figure 3b,c), and it was found that the XRD patterns of TaN films with the same nitrogen flow rate but different sputtering times/different thicknesses did not change much, showing that the thickness of the film does not affect the formation of the crystal structure of the film. Generally, the films deposited with 58 sccm of nitrogen do not have pronounced patterns and this suggests that the films are losing their crystalline structure and becoming more amorphous (Table 2). The full width at half maximum (FWHM) was obtained by Gaussian fitting and corresponding crystallite size was calculated by a Debye–Scherrer equation (D =  $0.9\lambda/\beta\cos\theta$ ), where D is the crystal size,  $\lambda$  is the wavelength of X-ray,  $\theta$  is the Braggs angle in radians, and  $\beta$  is the FWHM of the peak in radians. For each peak of XRD spectra one sample was used. The crystallite sizes presented in Table 2 are obtained from Figure 3a samples.

Flow Rate of N <sub>2</sub> (sccm)	Peaks	Peak Position, 2θ (deg)	FWHM, θ (deg)	Average Crystalline size (±0.5 nm)
0	β-Ta(002)	33.72	0.266	31.22
12	TaN(111)	35.40	0.340	24.54
17	TaN(111)	35.28	0.394	21.15
25	Ta <sub>3</sub> N <sub>5</sub> (130,040)	35.04	0.543	15.34
38	Ta <sub>3</sub> N <sub>5</sub> (130,040)	35.12	0.890	9.37
58	$Ta_3N_5(130,040)$	34.24	3.394	2.45

Table 2. Calculated crystallite sizes of Ta and TaN films.

#### 3.3. Elemental Composition (XPS Studies)

XPS spectra were obtained to ascertain the elemental composition of the deposited TaN films. Figure 4 illustrates an evolution of the XPS survey spectrum of the deposited TaN films as a function of  $N_2$  flow rate in the binding energy range of 0–1150 eV and show the Ta, O, N, and C signals. It is confirmed that the survey does not contain additional components that arise in the Si2p spectrum because our films were thicker. The O and C signal in these spectra might come from the ambient atmosphere and/or the presence of background oxygen in the sputtering chamber during sputtering, and/or from organic residues during the storage, as already reported in the literature [35,39,53], which may vary depending on the different sputtering instruments used. There is a chemical shift for both O and C in the lower binding energy with an increase in the  $N_2$  flow rates due to the formation of Ta<sub>2</sub>O<sub>5</sub> (Figure 5a,b). It can be assumed that the most probable origin of these residues is associated with impurities present in the nitrogen and/or a decrease in the Ta sputtering rate, which increases the effect of residual gases.



**Figure 4.** Evolution of the XPS survey spectra in the binding energy range of 0–1100 eV as a function of N<sub>2</sub> flows (12–58 sccm, 10 min deposition) showing the Ta, O, N and C signals for TaN films.

The XPS core-level spectra of Ta4f, Ta4d, and N1s for different N<sub>2</sub> flow rates (12–58 sccm) are shown in Figure 6a–c, respectively. Figure 6a shows the XPS region of Ta4f, revealing that it is composed of three overlaying bonding environments: Ta4f<sub>5/2</sub> of Ta-O<sub>x</sub> (Ta bonded with O), Ta4f<sub>7/2,5/2</sub> of Ta-N, and Ta4f<sub>7/2</sub> of Ta-N located at 30.25, 28.15, and 25.97 eV, respectively. As shown in Figure 6a, increasing the flow of N<sub>2</sub> from 12 to 58 sccm is likely to chemically shift the Ta4f<sub>7/2</sub> peaks from 25.97 to 25.78 eV, which were attributed to TaN Ta4f<sub>5/2</sub> peaks. Furthermore, the chemical shift from 30.25 to 29.81 reveals that O-rich films composed of Ta<sub>2</sub>O<sub>5</sub> with an increase of N<sub>2</sub> flows were observed.



**Figure 5.** XPS core levels spectra of (**a**) O 1s, and (**b**) C 1s for the TaN films at different N<sub>2</sub> flows (12–58 sccm, 10 min deposition).



**Figure 6.** XPS core levels spectra of (**a**) Ta4f, (**b**) Ta4d and (**c**) N1s for the TaN films at different N<sub>2</sub> flows (12–58 sccm, 10 min deposition).

Figure 7 shows the deconvoluted spectra of Ta4f for the TaN films with an N<sub>2</sub> flow of 12 sccm. Compared with the referred Ta binding energy values of TaN and TaO<sub>x</sub>, the binding energy values in the Ta4f spectrum (Ta4f<sub>7/2</sub> = 25.7 eV and Ta4f<sub>5/2</sub> = 27.7 eV) accorded with the chemical state of Ta in Ta–O binding [54]. The Ta4f doublet at binding energy Ta4f<sub>7/2</sub> = 27.3 eV and Ta4f<sub>5/2</sub> = 29.0 eV matched Ta<sup>5+</sup> state in Ta<sub>2</sub>O<sub>5</sub> [55,56], while the corresponding Ta4f doublet peaks are located respectively at Ta4f<sub>7/2</sub> = ~25.1 eV and Ta4f<sub>5/2</sub> = ~27.3 eV that should be attributed to N-rich TaN phase [57,58]. The existence of the Ta–N bonding in the film is further confirmed by the N1s peak located around 397.0 eV [59]. As the nitrogen flow rate increases, more Ta–N bonds form and the N 1s peak also increases. This is consistent with the XRD spectra. It can be seen from Figure 6c that the N1s peak shifts from 397.55 to 398.85 eV for nitrogen flow from 12 to 58 sccm, while the Ta4p<sub>3/2</sub> peak also shifts from 406.4 to 404.95 eV. It is noteworthy that there are uncertainties on the exact value for Ta4p binding energy. Nyholm et al. reported a Ta4p

value of 400.9 eV [60] and Tian et al. [61] reported a relatively high value of 406.0 eV. This shift in the binding energy for the N1s peaks toward lower values are in agreement with a previous study [35,39], while the binding energies for the Ta4f doublets also shift toward lower energy values and do not agree with previous studies.



Figure 7. Deconvoluted spectra of Ta4f for the TaN films with  $N_2$  flow of 12 sccm for 10 min deposition.

## 3.4. Optical Properties

Spectroscopy ellipsometry (SE) is broadly used as an important tool for thin films thickness, refractive indices and optical properties analysis, and its basic principle is shown elsewhere [62]. In this study, the films were optically thick and the measured  $\Psi$  and  $\Delta$  of all Ta films and TaN films were used for direct calculation of optical characteristics by using the substrate model. The n and k values of the corresponding film were obtained by using E. Kondoh ELLIPSHEET [33]. For each curve (n and k) of SE, one sample was used for the calculation.

Figure 8a,b depict the complex refractive index (n) and extinction coefficient (k), respectively for Ta films and with a comparison with the literature report. After comparing with the data of Tompkins et al. [63]. and Waechtler et al. [27], it is found that the changing trends of n and k of Ta films with different thicknesses are roughly the same, and the thickness does have an effect on the optical properties of Ta films. This is different to Waechtler et al., who reported the same n, k of Ta films with different thicknesses.

In Figure 9, we compare the refractive index (n) and extinction coefficient (k) of our sputtered TaN film with an N<sub>2</sub> flow rate of 12–58 sccm with the data of Aouadi et al. [2] and Waechtler et al. [27]. Also, it is found that when the N<sub>2</sub> flow rate is 12 and 17 sccm, the n and k of our sputtered TaN film are close to the values in the reference. However, as the N<sub>2</sub> flow rate increases, the n and k values of our sputtered TaN film become further from the n and k values in the references (Figure 9a,b). Furthermore, different N<sub>2</sub> flow rates and different thicknesses of TaN films have significant differences in n and k, which shows that both N<sub>2</sub> flow rate and thickness affect the optical properties of TaN. In addition, we can see that there are differences in the n and k values and curve shapes of TaN films with different crystal structures. The influence of N<sub>2</sub> flow rate on the n and k of TaN film deposited with different N<sub>2</sub> flow rates. It is interesting that the extinction coefficient of TaN films deposited at high nitrogen flows (38 and 58 sccm) decreases starting from 700 nm and especially from 900 nm. The reduction of the extinction coefficient suggests that the films are becoming more dielectric-like and explains the drastic increase of their resistivity (Figure 2).



**Figure 8.** (a) Refractive index (n) and (b) extinction coefficient (k) of Ta films sputtered for 20 min (450 nm) and 30 min (664 nm) are compared with literature data.



**Figure 9.** (a) The refractive index (n) and (b) extinction coefficient (k) of TaN(12–58 sccm) films sputtered for 10 min are compared with those in the literature.

In this work, we used optically thick films to extract optical characteristics. The thicknesses of these films are measured by using a Bruker Dektak step meter that is used only for relatively thick film. However, many applications of Ta and TaN layers need very low thickness. For instance, when they are used as diffusion barriers in advanced microelectronics. Taking this into account, we examined the applicability of ellipsometry to measure the thickness of the thin Ta and TaN layers. For this purpose, we used the values of the optical characteristics of these layers found in literature and measured in our work. Then we calculated  $\Delta$ - $\Psi$  trajectories for the films with different thicknesses (Figure 10). The presented curves demonstrate that Ta(N) film thickness, but it can be very good for evaluation of Ta(N) films deposited as diffusion barriers for microelectronics application (d < 10 nm). It is also apparent that ellipsometry may be efficient for evaluation of d < 100 nm films, as it was in [31,32].



Figure 10. Optical characteristics (the calculated  $\Delta$ – $\Psi$  trajectories) of deposited (a) Ta and (b) TaN films with different thicknesses. The solid curves were calculated by using optical characteristics at 633 nm measured in our films deposited with 12 sccm N<sub>2</sub>. The dashed curves are based on optical characteristics reported in Ref. [27] (633 nm, Ar to N<sub>2</sub> ratio = 4:1, the flow rate is not reported).

## 4. Conclusions

The Ta(N) film with different  $N_2$  flow rates (0–58 sccm) and sputtering times of 10–30 min was deposited by the DC reactive magnetron sputtering method, and it was found that the Ta(N) film deposition rate, electrical, structural, chemical and optical properties depend on  $N_2$  flow rate. As the  $N_2$  flow rate increases from 0 to 58 sccm, the crystal structure of the sputtered film transitions from  $\beta$ -Ta to TaN (111) and finally becomes the N-rich phase  $Ta_3N_5$  (130) or  $Ta_3N_5$  (040). When the N<sub>2</sub> flow rate increases, the diffraction peaks gradually widen, which indicates that the grain size is decreasing, leading to higher resistivity (Table 1), and so correspondingly could be attributed to the mixture of fcc TaN and its amorphous structure The films deposited with 58 sccm of nitrogen lose specific crystallographic patterns and therefore become amorphous. In studying the optical properties, we can see that both the thickness and the N<sub>2</sub> flow rate affect the refractive index (n) and extinction coefficient (k) of TaN film, and have a greater impact on k. The curve shapes of n and k of similar crystal structures have a small difference. The influence of N<sub>2</sub> flow rate on the refractive index and extinction coefficient of TaN film may be caused by the different crystal structures and grain size of TaN film deposited with different N<sub>2</sub> flow rates. The extinction coefficient of the films deposited with 58 sccm of nitrogen decreases, which suggests the formation of a more dielectric-like nature of the deposited films. This fact explains the drastic increase in the resistivity of the films shown in Figure 2. The reason for the transformation to the dielectric state is the incorporation of Ta oxide that can be seen from XPS data. When the nitrogen flow is so large, it reduces the free path length of sputtered atoms, poisons the Ta target and therefore reduces the effective deposition rate. As a result, the role of the residual oxygen is drastically increased and the deposited film becomes tantalum oxynitride with much higher resistivity. This effect becomes more pronounced when the deposition time is longer, which is the reason for the strong difference in resistivity between the films deposited during 10 and 30 min. If deposition time is short, the target poisoning might be negligible.

Finally, the presented results suggest the resistivity of TaN films deposited by magnetron sputtering can be precisely controlled by changing nitrogen concentration during deposition. It is also important to note that too high a concentration of nitrogen may have the negative effect of increasing the impact of gas phase impurities. Author Contributions: Conceptualization, M.R.B.; methodology, Y.H. and M.R.; validation, J.Z., J.Y., and M.R.B.; formal analysis, Y.H., M.R., Y.W., J.Z., J.Y., and M.R.B.; resources, Y.W., J.Z., and J.Y.; data curation, Y.H. and M.R.; writing—original draft preparation, Y.H. and M.R.; writing—review and editing, Y.H., M.R., and M.R.B.; visualization, Y.H. and M.R.; supervision, Y.W., J.Z., J.Y., and M.R.B.; funding acquisition, J.Z., J.Y., and M.R.B. All authors have read and agreed to the published version of the manuscript.

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