

*Supplementary material*

# Controlled growth of LDH films with enhanced photocatalytic activity in a mixed wastewater treatment

Zhongchuan Wang<sup>1</sup>, Pengfei Fang<sup>1</sup>, Parveen Kumar<sup>2</sup>, Weiwei Wang<sup>\*1</sup>, Bo Liu<sup>\*1,2</sup>, and Jiao Li<sup>1</sup>

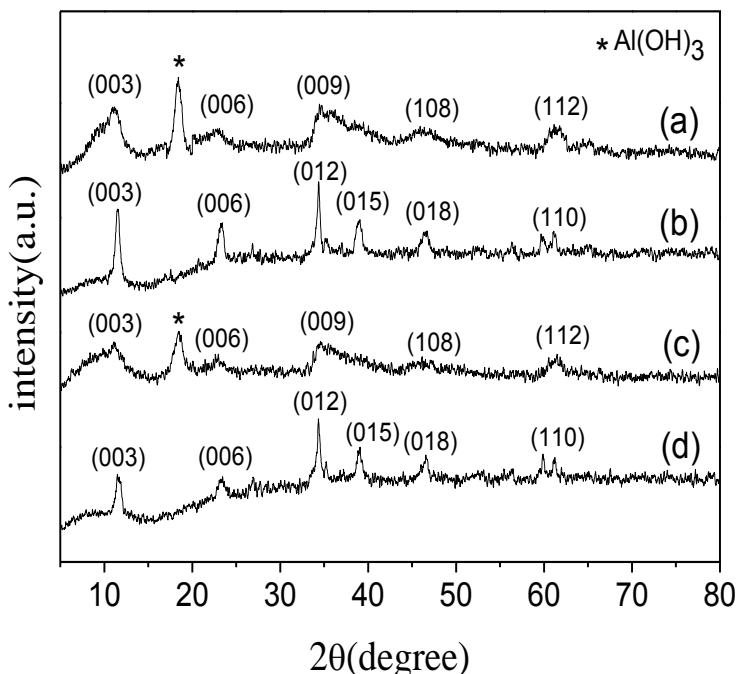
<sup>1</sup>School of Material Science and Engineering, Shandong University of Technology, Zibo 255000, China;

<sup>2</sup>Laboratory of Functional Molecular and Materials, School of Physics and Optoelectronic Engineering, Shandong University of Technology, Zibo 255000, China; wangzhongchuan1994@163.com (Z.W.); 17864373857@163.com (P.F.); kumar@sdut.edu.cn (P.K.); haiyan9943@163.com (J.L.)

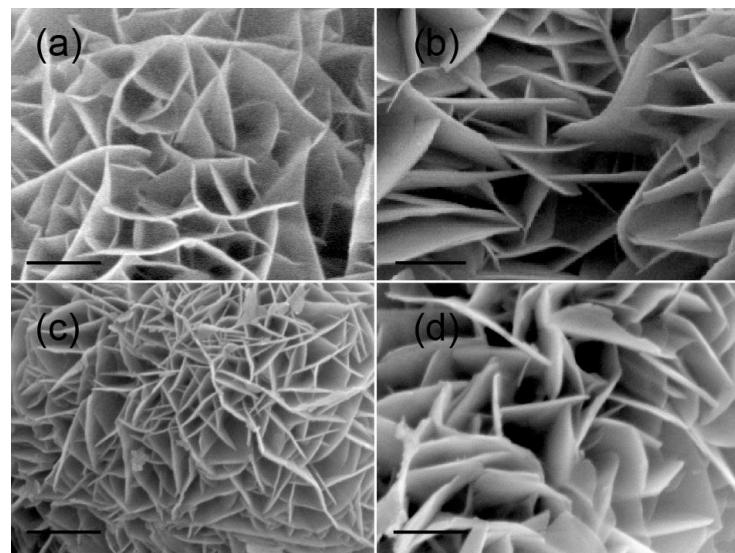
\*Correspondence: wangweiwei@sdut.edu.cn (W.W.); liub@sdut.edu.cn (B.L.); Tel.: +8615689078202(W.W.); +86-533-2783909 (B.L.)

**Table S1.** XPS Peak positions for Fe<sup>3+</sup> obtained from LDH films and Fe-doped LDH films.

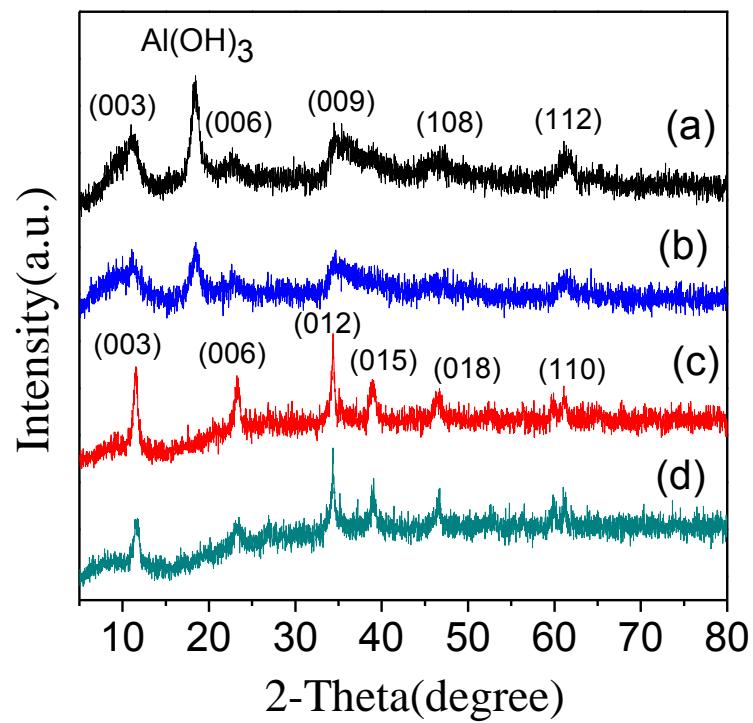
Film No.	Peak position (eV)	
	2p1/2	2p3/2
NiAl-LDH films	--	--
NiFe-LDH films	726.28	712.78
Fe-doped NiAl-LDH films	725.88	712.88
Fe-doped NiFe-LDH films	725.68	712.78



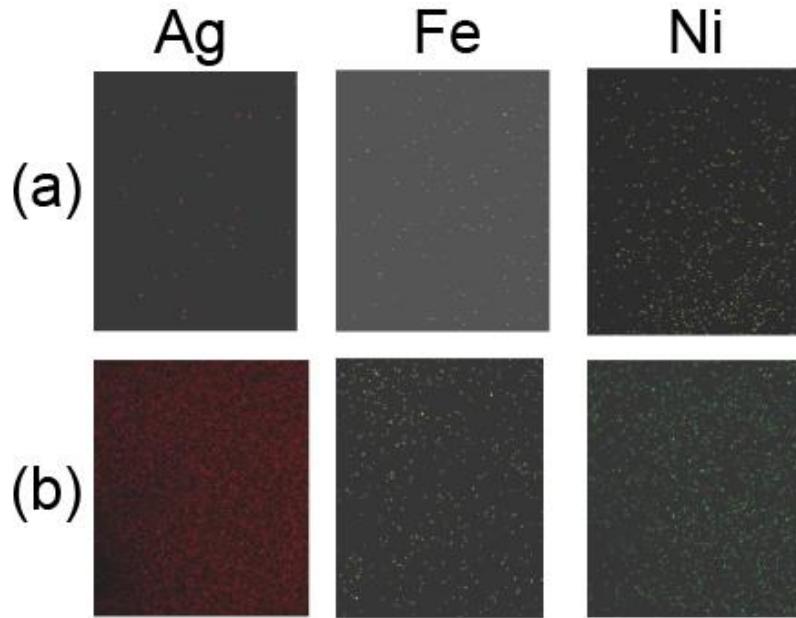
**Figure S1.** XRD patterns. (a) NiAl-LDH films, (b) NiFe-LDH films, (c) Fe-doped NiAl-LDH films, and (d) Fe-doped NiFe-LDH films.



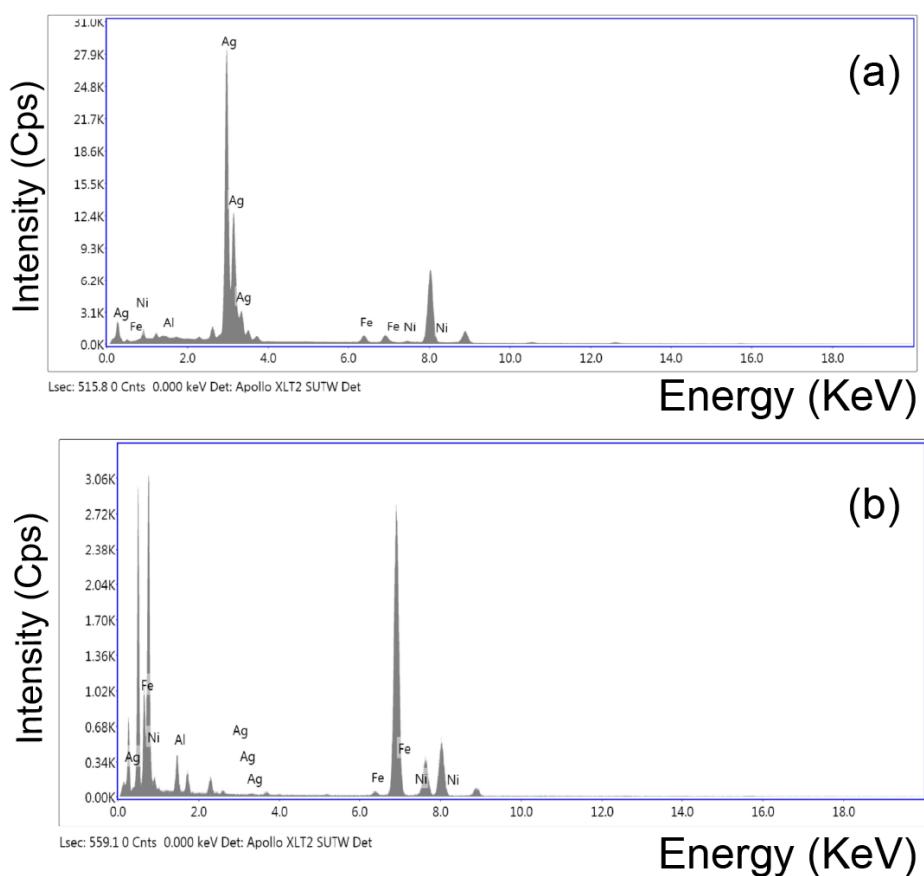
**Figure S2.** SEM images. (a) NiAl-LDH powders, (b) NiFe-LDH powders, (c) Fe-doped NiAl-LDH powders, and (d) Fe-doped NiFe-LDH powders.



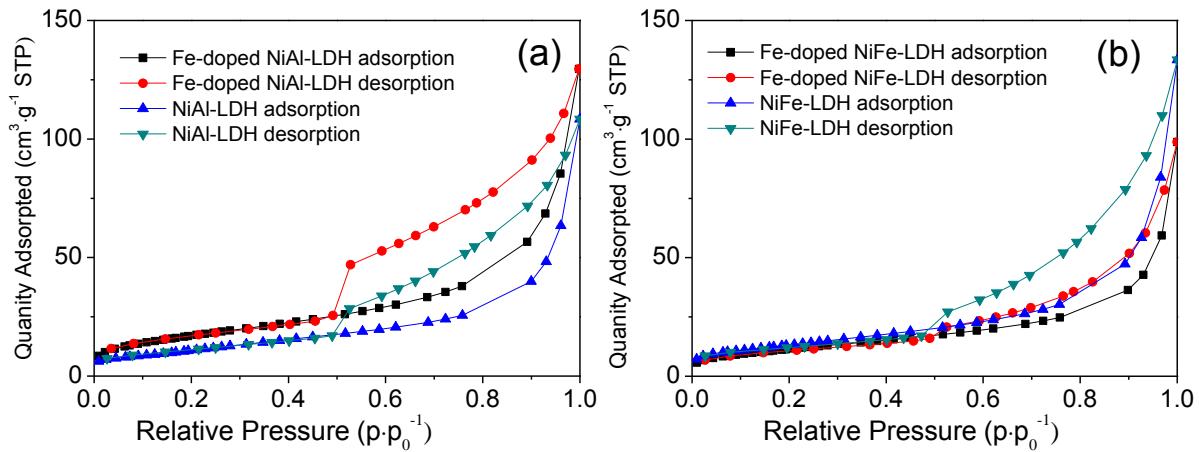
**Figure S3.** XRD patterns. (a) NiAl-LDH powders, (b) Fe-doped NiAl-LDH powders, (c) NiFe-LDH powders, and (d) Fe-doped NiFe-LDH powders.



**Figure S4.** EDS elements mapping for Fe-doped NiAl-LDH films after the photocatalytic degradation in the presence of methyl orange ( $20 \text{ mg}\cdot\text{L}^{-1}$ ) and Ag ions ( $5 \text{ mg}\cdot\text{L}^{-1}$ ). (a) Area without Ag particles, (b) area with Ag particles.



**Figure S5.** EDS spectra for Fe-doped NiAl-LDH films after photocatalytic reaction in the presence of  $5 \text{ mg}\cdot\text{L}^{-1}$  Ag ions. (a) Area with Ag nanoparticles, (b) area without Ag nanoparticles in Figure 7(c).



**Figure S6.** N<sub>2</sub> adsorption/desorption isotherms of (a) NiAl-LDH powders and Fe-doped NiAl-LDH powders, (b) NiFe-LDH powders and Fe-doped NiFe-LDH powders.

All LDH powders are a type IV isotherm with an H3 hysteresis loop, owing to the aggregation of LDHs nanosheets. The specific surface area for NiAl-LDH powders, NiFe-LDH powders, Fe-doped NiAl-LDH powders, and Fe-doped NiFe-LDH powders are 42.26 m<sup>2</sup>·g<sup>-1</sup>, 43.09 m<sup>2</sup>·g<sup>-1</sup>, 48.37 m<sup>2</sup>·g<sup>-1</sup>, and 42.82 m<sup>2</sup>·g<sup>-1</sup>, respectively. The pore size for NiAl-LDH powders, NiFe-LDH powders, Fe-doped NiAl-LDH powders, and Fe-doped NiFe-LDH powders are 14.35 nm, 12.92 nm, 10.38 nm, and 14.27 nm, respectively.