## **Supplementary Information**

## Carbon Quantum Dots Prepared with Chitosan for Synthesis of CQDs/AuNPs for Iodine Ions Detection

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## Test method for deacetylation degree of chitosan:

The degree of deacetylation (DD) of chitosan was determined by titration as follows:

0.20 g vacuum-dried sample was dispersed in 20 mL, 0.1 mol/L HCl and then titrated with 0.1 mol/L NaOH. The pH value and corresponding volume of NaOH solution was recorded and a titration curve was plotted. This plot had two inflection points, where the corresponding volumes were recorded as V1 and V2. The DD value was calculated according to the following equation

Degree of deacetylation (DD) = 
$$((V2-V1)\times c\times 0.016)/(0.094*W)$$
 (1)

Where c is the concentration of NaOH solution, and W is the sample weight (g).

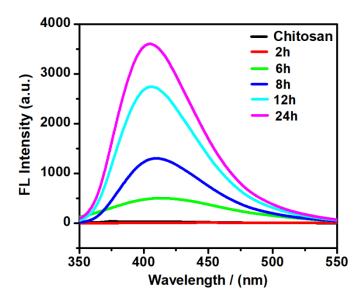


Figure S1 Fluorescence emission spectra of CQDs obtained by chitosan hydrothermal carbonization with different times (at a maximum excitation wavelength of 330 nm).

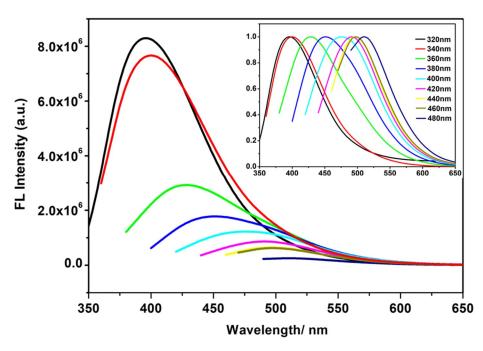


Figure S2 Fluorescence emission spectra of the CQDs obtained under excitations from 280 to 480 nm with 20 nm interval. Inset: Normalized Fluorescence emission spectra.

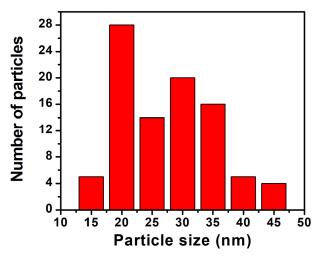


Figure S3 size distribution histogram of CQDs/AuNPs.



Figure S4 SAED of CQDS/AuNPs.

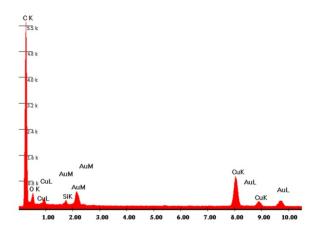


Figure S5 Energy dispersive X-ray spectrum (EDS) of the CQDS/AuNPs showing the peaks for C, O, Cu and Au.

Table S1 Results for the determination of iodide in salt samples (n=3).

Sample	Spiked (µM)	Measured (μM)	Recovery (%)
salt	0	-	-
	5	4.92	98.4
	10	10.74	107.4
	20	21.10	105.5

Table S2 Comparison of the proposed iodide detection method with other reported methods.

Methods	Probe	Linear range/(M)	LOD/(M)	Ref.
Colorimetric	ATTP-AuNPs	$5.0 \times 10^{-7}$ to $6.0 \times 10^{-6}$	$1.5 \times 10^{-8}$	[2]
Fluorescence	N-CQDs	$1 \times 10^{-4}$ to $2.0 \times 10^{-3}$	$1.0 \times 10^{-5}$	[17]
Colorimetric	Cu@Au Nanoparticles	_	$5.0 \times 10^{-6}$	[S1]
Colorimetric	FCD	_	4.3×10 <sup>-7</sup>	[S2]
Fluorescence				
Turn off and	$C_{20}H_{20}N_4O_5$	$6.25 \times 10^{-6}$ to $3.12 \times 10^{-5}$	$3 \times 10^{-8}$	[S3]
colorimetric				
Colorimetric	$C_{22}H_{22}N_4O_6S_2$	_	$5.31 \times 10^{-6}$	[S4]
Fluorescence	FITC-Au nanoparticles	$1.0 \times 10^{-7} - 1.0 \times 10^{-6}$	$5 \times 10^{-8}$	[S5]
Colorimetric	silver nanoparticles	$6.0 \times 10^{-6} - 8.0 \times 10^{-6}$	$2.5 \times 10^{-4}$	[S6]
Colorimetric	CQDs/AuNPs	$2.0 \times 10^{-5}$ - $4.0 \times 10^{-4}$	$2.3 \times 10^{-6}$	This work

## References

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