

Supplementary Materials

The Photocatalytic Activity of CaTiO₃ Derived from the Microwave-Melting Heating Process of Blast Furnace Slag

Jun Xie ¹, Qing Ye ^{1,2,*}, Jianghao Zhou ³, Yue Liao ¹ and Gongming Qian ^{1,2,*}

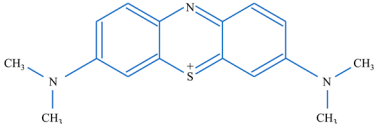
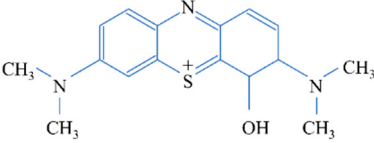
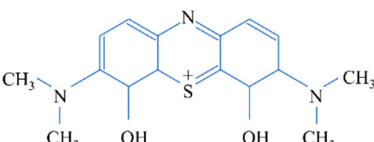
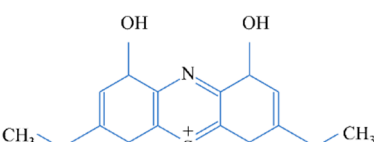
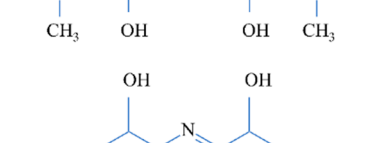
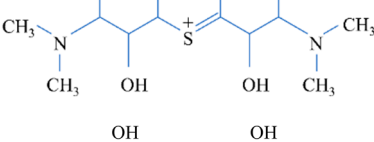
¹ School of Resource and Environmental Engineering, Wuhan University of Science and Technology, Wuhan 430081, China; junxie0mineral@wust.edu.cn (J.X.)

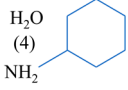
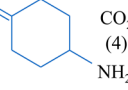
² Hubei Key Laboratory for Efficient Utilization and Agglomeration of Metallurgic Mineral Resources, Wuhan 430081, China

³ School of Materials and Metallurgy, Wuhan University of Science and Technology, Wuhan 430081, China

* Correspondence: yeqing@wust.edu.cn (Q.Y.); qiangongming@wust.edu.cn (G.Q.); Tel.: +86-13476834865 (G.Q.)

Table S1. The Gibbs free energy of MB molecular and intermediate products.

Compound	Gibbs free energy/ eV
	-1182.80
	-1258.60
	-1334.39
	-1486.06
	-1489.64
	-1710.00

SO_4^{2-} (1)	H_2O (4)		$\text{HN}=\text{C}$		CO_2 (4)	-2396.12
NO_3^- (0.5)	NH_4^+ (2.5)	SO_4^{2-} (1)	H_2O (16)	CO_2 (16)		-5223.41
