

*Supporting Information*

# Asymmetric Alternative Current Electrochemical Method Coupled with Amidoxime-Functionalized Carbon Felt Electrode for Fast and Efficient Removal of Hexavalent Chromium from Wastewater

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**Table S1.** Adsorption kinetics fitting parameters of Cr (VI) by Ami-CF

Models	Parameters	Values
Pseudo-first order model	$Q_{e, \text{exp}}/\text{mg}\cdot\text{g}^{-1}$	33.86
	$Q_{e, \text{cal}}/\text{mg}\cdot\text{g}^{-1}$	32.50
	$K_1/\text{h}^{-1}$	0.1486
	$R^2$	0.976
Pseudo-second order model	$Q_{e, \text{exp}}/\text{mg}\cdot\text{g}^{-1}$	33.86
	$Q_{e, \text{cal}}/\text{mg}\cdot\text{g}^{-1}$	36.13
	$K_2/\text{g}\cdot(\text{mg}\cdot\text{h})^{-1}$	0.0056
	$R^2$	0.993

**Table S2.** Fitting parameters of adsorption isotherm of Cr (VI) by Ami-CF

Models	Parameters	Values
Langmuir	$Q_m/\text{mg}\cdot\text{g}^{-1}$	105.79
	$K_L/\text{L}\cdot\text{mg}^{-1}$	0.01
	$R^2$	0.992
	$N$	0.39
Freundlich	$K_F/\text{mg}\cdot\text{g}^{-1}\cdot\text{L}^{1/n}\cdot\text{mg}^{-1/n}$	7.18
	$R^2$	0.968

**Table S3.** Adsorption capacity of adsorbents for Cr (VI)

Adsorbent	pH	Adsorption capacity	References
Mn-incorporated ferrihydrite	5.0	48.5 $\text{mg}\cdot\text{g}^{-1}$	[1]
TOCNF Grafted With PABS Copolymer	3.0	5.263 $\text{mg}\cdot\text{g}^{-1}$	[2]
Poultry manure-derived biochar	4.0	19.09 $\text{mg}\cdot\text{g}^{-1}$	[3]

Polyaniline nanowires-coated polypropylene filter	7	$66.7 \text{ mg}\cdot\text{g}^{-1}$	[4]
ZnFe <sub>2</sub> O <sub>4</sub>	2.5	$34.25 \text{ mg}\cdot\text{g}^{-1}$	[5]
Straw and sludge-based activated carbon	6.0	$15 \text{ mg}\cdot\text{g}^{-1}$	[6]
Fe-Al hydroxides	5.6	$20 \text{ mg}\cdot\text{g}^{-1}$	[7]
Chlorapatite	7.0	$10 \text{ mg}\cdot\text{g}^{-1}$	[8]
Carbon nano-onions	6.0	$60 \text{ mg}\cdot\text{g}^{-1}$	[9]
Tobacco petiole Pyrolytic biochar	6.0	$40 \text{ mg}\cdot\text{g}^{-1}$	[10]

**Table S4.** Influence of different concentrations of Cu (II), Zn (II) and Ca (II) on distribution percentage of the Cr (VI) species from the simulation results for 100 mg/L Cr (VI) solution at pH 2 by using Visual MINTEQ 3.1.

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