

Table S1. Comparison with other catalysts

Semiconductor	Solvothermal method	Morphology	Sacrificial reagent	Light source (nm)	Optimal hydrogen evolution ($\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$)	Refs.
ZnIn ₂ S ₄ @NH ₂ -MIL-125(Ti)	Solvothermal method	Micropores and mesopores	Na ₂ S/Na ₂ SO ₃	300 W Xe lamp $\lambda > 420 \text{ nm}$	2204.2 40 wt% NH ₂ -MIL-125(Ti)	1
g-C ₃ N ₄ @ZnIn ₂ S ₄	Solvothermal method	2D/2D g-C ₃ N ₄ nanosheet@ZnIn ₂ S ₄ nanoleaf	Triethanolamine	300 W Xe lamp $\lambda > 420 \text{ nm}$	2780	2
Co(dmgh) ₂ pyC/ZnIn ₂ S ₄	Impregnating method	Microspheres	Triethanolamine	300 W Xe lamp $\lambda > 420 \text{ nm}$	3840 3.0 wt% Co(dmgh) ₂ pyCl/ZnIn ₂ S ₄	3
CQDs/ZnIn ₂ S ₄	Microwave hydrothermal method	Microspheres	N/A	350 W Xe lamp $\lambda > 400 \text{ nm}$	1032.2	4
MoS ₂ /ZnIn ₂ S ₄	Hydrothermal method	Nanoparticles	Na ₂ S/Na ₂ SO ₃	300 W Xe lamp $\lambda > 420 \text{ nm}$	2080 0.5 wt% MoS ₂ /ZnIn ₂ S ₄	5
In ₂ S ₃ /ZnIn ₂ S ₄	Ion-exchange method	Microflowers	Na ₂ S/K ₂ SO ₃	300 W Xe lamp $\lambda \geq 420 \text{ nm}$	678	6
CdS/ZnFe ₂ O ₄ /ZnIn ₂ S ₄	Solvothermal processes & Ionic layer adsorption-reaction method	Nanosheet stereoscopic films	Na ₂ S/Na ₂ SO ₃	300 W Xe lamp $\lambda > 420 \text{ nm}$	79.0 $\mu\text{mol h}^{-1}$ 1-CdS/ZnFe ₂ O ₄ /ZnIn ₂ S ₄	7

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Figure S1. TG results of the Co₃O₄(20)@ZIS

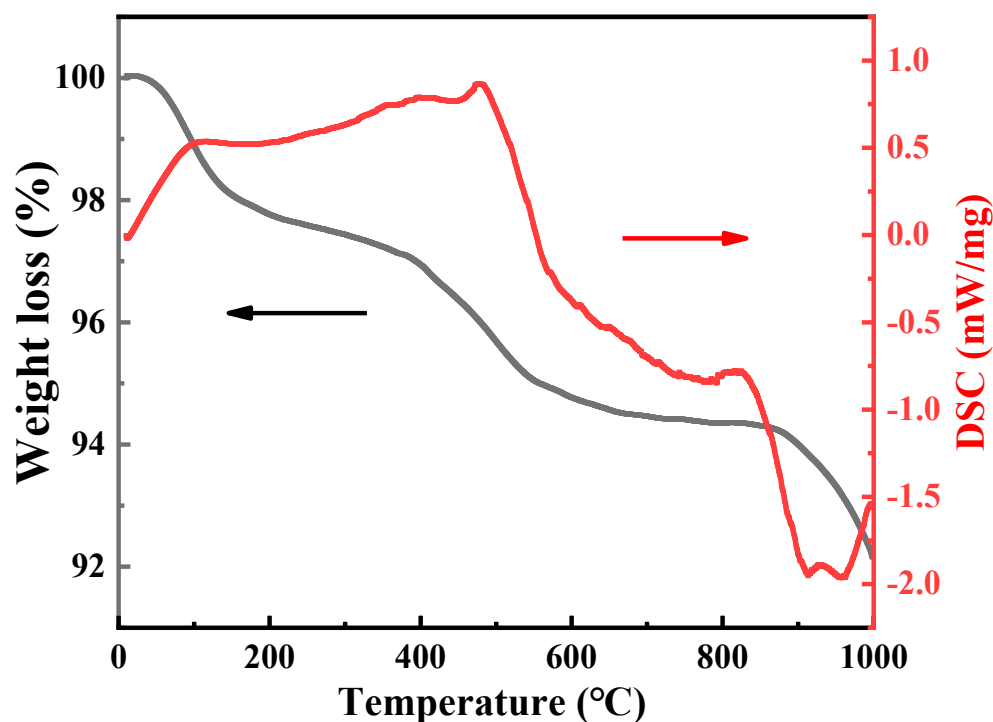


Figure S2. Three cycles of experiment of the $\text{Co}_3\text{O}_4(20)\text{@ZIS}$

