



Article Improved Electrical Properties of EHD Jet-Patterned MoS₂ Thin-Film Transistors with Printed Ag Electrodes on a High-k Dielectric

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Abstract: Electrohydrodynamic (EHD) jet printing is known as a versatile method to print a wide viscosity range of materials that are impossible to print by conventional inkjet printing. Hence, with the understanding of the benefits of EHD jet printing, solution-based MoS₂ and a high-viscosity Ag paste were EHD jet-printed for electronic applications in this work. In particular, printed MoS₂ TFTs with a patterned Ag source and drain were successfully fabricated with low-k silica (SiO₂) and high-k alumina (Al₂O₃) gate dielectrics, respectively. Eventually, the devices based on Al₂O₃ exhibited much better electrical properties compared to the ones based on SiO₂. Interestingly, an improvement of around one order of magnitude in hysteresis was achieved for devices after changing the gate insulator from SiO₂ to Al₂O₃. In effect, the results of this work for the printed MoS₂ and the printed Ag source and drains for TFTs demonstrate a new approach for jet printing in the fabrication of electronic devices.

Keywords: MoS₂; Ag electrode; EHD jet printing; high-k dielectric; thin-film transistors



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1. Introduction

Graphene, the first material in the two-dimensional (2D) family of materials, has been used in a large number of scientific applications due to its superior and novel properties, e.g., mechanical, thermal, electrical, optical, etc. [1]. Likewise, as an emerging candidate in the crowd of 2D materials, transition metal dichalcogenides (TMDs) have drawn much attention due to their sizeable bandgap, which is definitely better than the unfavorable zero bandgap of graphene [2]. In particular, molybdenum disulfide (MoS₂) is one of the most studied TMDs, with diverse applications [3,4] because of its tremendously high intrinsic electron mobility and indirect-to-direct bandgap from 1.2 eV in bulk to 2.0 eV in the monolayer—especially in thin-film transistor device applications [5].

Researchers have devoted considerable efforts to synthesizing high-quality MoS_2 with a controllable number of layers, playing a significant role in the fundamental research and application explorations involving this material. Generally, a variety of methods have been proposed to produce 2D TMD materials, including mechanical/chemical exfoliation [6,7], chemical vapor deposition (CVD) [8], wet-chemical based methods [9], and so on. Even though the growth of atomically thin TMD films via the CVD method is one of the most popular ways of producing these films, controlling the respective concentrations of the precursors precisely during the growth process is still challenging. Next, exfoliation methods, despite their simple and low-cost features, face the issue of random shape/thickness of the resulting films from the mechanical strategy and the diminished semiconductor properties of the films. Therefore, they are unsuitable for the production of 2D TMD materials for large areas over wafer-scale and high-throughput applications. In this context, the wetchemical-based method seems to be the method of choice for synthesizing high-quality MoS_2 with a controllable number of layers in a relatively simple and easy way. In the solution method, reports of jet-printed MoS₂-based TFTs with printed source and drain (S/D) electrodes are still rare. So far, there have been several metal nanoparticle pastes that can be printed, such as nanoparticle pastes of Au, Al, Cu, Ag, etc. Additionally, gold is a highly conductive material, but its prohibitive price is disadvantageous for use in mass production of TFT₅. In contrast, aluminum and copper are preferred because of their low price and good conductivity, but they are easily oxidized in the atmosphere, resulting in the degradation of their electrical features. Meanwhile, silver shows many outstanding merits, such as having the highest electrical conductivity among the materials that can be printed, along with its chemical stability and affordable price, making Ag stand out from the other materials. The undeniable possibility of using patterned Ag for electrodes in TFT devices fabricated by EHD jet printing was confirmed by our previous research [10]. In this previous work, the sheet resistance of printed Ag was evaluated to be around 0.027 Ω^{-1} —comparable or even superior to that of Ag layers made by other methods, such as inkjet printing and screen printing [11,12]. Hence, we chose high-viscosity Ag paste for patterning the source and drain for the TFTs in this work.

The improvement of TFTs' characteristics by using a high-k dielectric has been studied previously [13]. However, the use of an Al₂O₃ dielectric layer in solution-based MoS₂ TFTs with printed Ag S/D contacts—especially in a back-gated configuration—has not been explored to date. Hence, this work presents the advantages of EHD jet-printing technology in patterning electrical elements from various viscous materials and investigates the properties of high-k dielectric-based solution-processed MoS₂ TFTs (Figure 1). This work demonstrates the suitability of direct EHD jet-printing technology for mass production of TFTs because of its low cost and high performance.



Figure 1. The whole process of fabricating TFT devices, including the EHD jet printing of the MoS_2 semiconductor, the PMMA-based transfer of MoS_2 onto Al_2O_3/Si , and the EHD jet printing of Ag as the S/D of the TFTs.

2. Experimental Details

2.1. Growth of MoS₂ Layers

 MoS_2 patterns were created by a combination of EHD jet printing and one-step annealing following the process described in our previous research [9]. First, the ammonium tetrathiomolybdate ((NH₄)₂MoS₄) precursor solution for printing was prepared with concentrations of 25, 50, 75, and 100 mM by stirring the (NH₄)₂MoS₄ in a group of solvents of ethanolamine and butylamine for 12 h. Subsequently, a S-rich solution was formed by preparing a 1 M sulfur solution with carbon disulfide (CS₂) and dissolving the sulfur solution in the (NH₄)₂MoS₄ precursor solution with N,N-dimethylformamide. All chemicals used in this formulation were purchased from Sigma-Aldrich (Milwaukee, WI, USA) and ThermoFisher (Fisher Scientific, Leicestershire, UK) and used without further purification.

For the printing of $(NH_4)_2MoS_4$ on a UV/O₃-irradiated 300 nm thick SiO₂/Si substrate, the as-prepared precursor solution was collected in a syringe pump connected to a vertically

movable plastic tip. The target substrate was then placed stably on a metal stage that could be moved on a horizontal plane, as described in our previous report [10]. Subsequently, for patterning the (NH₄)₂MoS₄ lines, the voltage was adjusted to stretch the meniscus of the solution at the tip's mouth into an upside-down cone shape, called the Taylor cone-jet mode. In particular, the printing parameters for patterning the (NH₄)₂MoS₄ lines were a tip height of 2 mm, an applied voltage of 1.8–1.9 kV, a substrate temperature of 50 °C, a solution flow rate of 0.0032 μ L s⁻¹, and a stage speed range of 2000–8000 μ m s⁻¹. After printing the line patterns from the S-rich (NH₄)₂MoS₄ solution, the patterns were pre-annealed at 150 °C for 20 min in ambient air using a hot plate. The pre-annealed patterns were then moved into a tube furnace for their annealing at a high temperature of 1000 °C for 1 h in a low vacuum (10⁻¹–10⁻² Torr), without sulfurization or further post-annealing, resulting in the final crystalline MoS₂ line patterns.

2.2. Transfer of MoS₂ onto Other Substrates

The printed MoS₂ on the SiO₂/Si substrate, after annealing, was covered with a PMMA (avg. Mol wt. ~350,000 g mol⁻¹ and ~996,000 g mol⁻¹) layer using spin-coating (at a spinning speed of 3000 rpm for 30 s). The spin-coated sample was then baked at 200 °C for 2 h on a hot plate in ambient air. Subsequently, the PMMA/MoS₂/SiO₂/Si wafer was placed on the surface of an etchant mixture (HF:BOE:DI water (1:1:1)) to remove the SiO₂. The resulting PMMA/MoS₂ membrane was then cleaned from the etchant solution using deionized (DI) water. After discarding the etchant contaminant, the PMMA/MoS₂ double layer was picked up on different target substrates for different purposes—for example, on Al₂O₃/Si for TFT fabrication, or on a Cu grid for transmission electron microscopy (TEM, Ultra-Corrected-Energy-Filtered -TEM Libra 200 HT Mc Cs) studies. Finally, the top PMMA layer was removed using acetone at 80 °C to obtain the patterned MoS₂.

2.3. Device Fabrication

Printed TFTs were fabricated by EHD jet printing with MoS₂ line patterns as the semiconductor layer and Ag line patterns as S/D electrodes in each TFT. For a bottom gate and top contact (BGTC) configuration of the TFT, first, 40 nm thick alumina was deposited by atomic layer deposition (ALD) on clean bare Si wafers. The as-grown/printed MoS₂ was then transferred carefully from the SiO₂/Si substrate onto a cleansed Al₂O₃/Si substrate using the above procedure involving PMMA. Notably, the smooth morphological surface of the MoS₂ transferred onto other substrates was shown using an atomic force microscope (AFM, Nano expert II EM4SYS) to have no wrinkles or damage [14]. Finally, linear Ag patterns/terminals were printed perpendicularly on the MoS₂ line patterns with the assistance of the pneumatic pressure due to the high viscosity of Ag. In particular, the silver line patterns were successfully EHD-printed with a tip–substrate gap of 1.5 mm, applied voltage of 1 kV, stage speed of 2000–2500 μ m s⁻¹, and pressure of 80 kPa, and they were sintered at 200 °C for 30 min in ambient air. The whole fabrication process, from the preparation of the MoS₂ line patterns to their transfer and the fabrication of the TFT devices, is sketched out in Figure 1.

3. Results and Discussion

3.1. Printed MoS₂ Line Patterns

Figure 2a shows the microscopic images of the MoS_2 line patterns prepared with different $(NH_4)_2MoS_4$ precursor concentrations using EHD jet printing. Visually, from left to right in this figure, the printed line patterns originating from increasing solution concentrations possess different thicknesses, which could be predicted from the color of the patterns. Moreover, all patterns showed smooth surfaces that were hole-free regardless of the concentrations (from 25 to 100 mM), demonstrating the printability and appropriately chosen concentration of the precursor solution. In addition, based on Figure 2b, the MoS_2 pattern was proven to be undamaged and without wrinkles after transferring it using the PMMA-assisted method. Therefore, it is guaranteed that the MoS_2 quality will be



unchanged when the MoS₂ is transferred to an arbitrary substrate to be used in further device fabrication.

Figure 2. (a) Optical images of printed $(NH_4)_2MoS_4$ linear patterns with four different concentrations. (b) Optical images of the same pattern of MoS_2 after thermal annealing of as-grown/printed $(NH_4)_2MoS_4$ and transfer onto SiO_2/Si substrates, respectively.

For the evaluation of the composition and thickness of the printed MoS_2 , three methods of analysis—Raman, photoluminescence (PL), and XPS spectroscopies—were carried out on the printed MoS₂. In particular, Raman spectroscopy was carried out under four different precursor solutions, from 25 to 100 mM. As shown in Figure 3a, two strong signals for the E_{2g}^1 (at around 380 cm⁻¹) and A_{1g} (at around 405 cm⁻¹) modes emerged in the Raman spectra regardless of the molar concentrations of the precursor solution. These two major modes assigned to the in-plane vibration of the Mo and S atoms and the out-of-plane vibration of the S atoms provided good evidence for the presence of the 2H phase of MoS₂. It was also found that the two modes exhibited a well-defined concentration dependence, with the modes shifting opposite to one another with increasing concentration. Indeed, as the concentration was increased from 25 to 100 mM, the frequency difference between the modes increased gradually from 23.07 to 25.82 cm⁻¹ (Figure 3b). Moreover, the Raman spectra suggested that the MoS₂ line patterns obtained from concentrations of 25 mM and higher each consisted of at least three or four layers. Figure 3c describes the PL spectra of a representative MoS₂ sample printed from the 50 mM precursor solution. Two specific peaks at about 686 and 632 nm in the spectra were attributed to the A_1 and B_1 excitons originating from the transition at the K-point of the Brillouin zone in the sample material, respectively.

The chemical composition of the printed MoS₂ was determined by XPS. Figure 3d,e show the Mo 3d and S 2p regions, respectively, for the MoS₂ that was printed from the 50 mM precursor solution and annealed. The Mo 3d XPS spectra were clearly observed to have two distinct peaks at 229.0 and 232.1 eV and a weak peak at 226.4 eV corresponding to the $3d_{5/2}$ and $3d_{3/2}$ of Mo⁴⁺ and S 2s, respectively. These peaks proved the presence of the 2H phase in MoS₂. In addition, the peaks observed at 162.1 and 163.4 eV belonged to the divalent sulfide ions (S²⁻) $2p_{3/2}$ and $2p_{1/2}$ in 2H-MoS₂, respectively.

Furthermore, typical nanoscale images of the MoS₂ line pattern printed from the 25 mM concentration precursor solution were captured by TEM to observe the plane-view and thickness of the pattern. Figure 4a,b show the morphological TEM images of the printed line pattern at low and high resolutions of the TEM, respectively, revealing the honeycomb MoS₂ surface. Moreover, this feature can be clearly seen in the top insert of Figure 4b. Furthermore, the fast Fourier transform (FFT) corresponding to the lower insert of Figure 4b depicts the polycrystallization of the printed MoS₂. The layer number of the

line pattern was found to be four monolayers from the cross-sectional view of the TEM image (Figure 4c). Meanwhile, the selected area of Figure 4c, magnified to the scale shown in Figure 4d, corresponded to the tetra-layer of the pattern.



Figure 3. (a) Raman spectra and (b) frequencies of the A_{1g} and E_{2g}^1 modes against the precursor concentration. (c) PL spectra and (d,e) XPS spectra of the printed MoS₂ line patterns printed from the 50 mM concentration precursor solution.



Figure 4. MoS₂ line patterns for TEM printed from the 25 mM precursor solution: (a) Plan-view TEM

image and (b) HR-TEM image of the selected MoS₂ surface in (a), and the upper and lower inserts of (b) are the magnification of a portion of (b) and the corresponding fast Fourier transform, respectively. (c) Cross-sectional view of the TEM image and (d) the magnified image of the selected area in (c) containing the printed 4-layer MoS₂ line patterns.

3.2. Printed Ag Line Patterns

The printing process of the Ag line patterns, illustrated in Figure 5a, clearly shows that the Taylor cone-jet mode was used to print the pattern. The width and clear shape of the Ag line patterns were found to be strongly dependent on the printing parameters—such as pressure, additive existence, and especially the stage speed—in our previous research [10]. Here, Figure 5b shows a typical microscope image of a Ag line pattern printed under the optimized conditions previously mentioned in the experimental section. The patterns were each observed to be 100–200 μ m in width, without serrated edges. Figure 5c shows the AFM height step image of the printed Ag line pattern on MoS₂, and the thickness of the pattern was measured to be 2 μ m. Finally, according to the SEM image of the printed Ag line pattern after sintering, as shown in Figure 5d, although a rough morphology of the pattern was seen in the image, the distribution of the Ag particles was found to cover the entire substrate surface.



Figure 5. (a) Photograph of the Taylor cone-jet mode used for printing the Ag line patterns. (b) Optical image of a printed Ag line pattern. (c) AFM image at the boundary area including the Ag and MoS₂ line patterns, showing the thickness of the printed Ag line pattern. (d) SEM images of the printed Ag line pattern after sintering.

3.3. Printed MoS₂ TFTs

The electrical properties of the printed MoS₂ were characterized in a thin-film transistor application in which the integration of an ALD 40 nm Al₂O₃ dielectric, a printed MoS₂ line pattern as a semiconductor, and printed Ag line patterns as top contacts was carried out. In this study, a counterpart TFT based on low-k SiO₂ was also fabricated for comparison with the aforementioned high-k Al₂O₃-based TFT. Figure 6a shows the schematic of the BGTC MoS₂ TFTs. Additionally, the top-view optical images of two typical TFTs, in which MoS₂ was grown on SiO₂ (k = 3.9) and transferred onto Al₂O₃ (k = 7.0), respectively, are shown in Figure 6b, c, respectively.

-5 Ó 5 10 15 20

Gate voltage (V)



Drain voltage (V) Figure 6. (a) Diagram of cross-sectional TFTs. (b,c) Top-view optical images of real printed MoS₂ TFTs based on SiO₂ and Al₂O₃, respectively, with Ag S/D electrodes. (d,e) Transfer curves and (f,g) output curves of MoS₂ TFTs based on SiO₂ (V_{GS}: -20 to 120 V) and Al₂O₃ (V_{GS}: 0 to 20 V), respectively, fabricated from the same 100 mM solution. (h,i) Transfer and output curves of Al₂O₃-based MoS₂ TFTs fabricated from 4 different concentrated precursor solutions.

6

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The field-effect mobility (μ) and threshold voltage (V_{TH}) were calculated in the linear regime at a drain voltage (V_{DS}) of 1 V for the TFT. The threshold voltage is defined as the intersection point of the V_{GS} axis and the extrapolation of the linear region of the transfer curve. The linear field-effect mobility from each device was then calculated from the gradient of the drain current versus the gate voltage according to the following equation:

$$\mu = \frac{L}{WCV_{DS}} \times \frac{dI_{DS}}{dV_{GS}} \bigg|_{V_{DS}=1 V}$$
(1)

where L and W are the patterned MoS_2 channel length and width, respectively, and C is the capacitance per unit area of the gate insulator. The length and width of the patterned MoS_2 channel were about 40–100 and 500–600 µm (for various devices), respectively, used for calculating the TFTs' performance in relation to the precursor concentrations and the dielectric layers, respectively.

For the primary evaluation of the TFTs, two TFT device groups with different dielectrics of Al₂O₃ (15.5 \times 10⁻⁸ F cm⁻²) and SiO₂ (10⁻⁸ F cm⁻²) were prepared from the same 100 mM MoS₂ precursor solution. The $I_{DS} - V_{GS}$ transfer curves of these MoS₂ devices are shown in Figure 6d,e. Due to the dissimilar materials and thicknesses of the dielectric layer, the applied voltage range should be different to prevent the breakdown of the devices. While sweeping V_{GS} from -20 to 120 V (and back) at various constant V_{DS} values of at least 1 V for the 300 nm SiO₂ TFT (Figure 6d), those for 40 nm Al₂O₃ were measured at about -10 V to 20 V (then back) and 0.1 V (Figure 6e), respectively. Notably, both TFT device types showed a clockwise hysteretic phenomenon associated with the electron trap at the MoS₂-dielectric and/or MoS₂-Ag interfaces. However, a dramatic reduction in hysteresis of one order of magnitude was achieved when using Al₂O₃ instead of SiO₂ in the TFT, revealing smaller trap charges at the interface between MoS_2 and Al_2O_3 . In addition, although the maximum gate leakage current was about 10^{-9} – 10^{-8} A for the low- and high-dielectric-based TFT devices, a current (I_{DS}) two orders of magnitude higher was obtained when using Al_2O_3 instead of SiO₂ as the gate insulator. Simultaneously, the on/off current ratio (I_{on}/I_{off}) of each device with Al₂O₃ (~10⁵) was obviously enhanced compared to that of each device with SiO_2 (~10²). Moreover, other electrical parameters of each MoS₂ TFTs showed a remarkable improvement after changing the dielectric from low-k SiO₂ to high-k Al₂O₃, such as a 30 times steeper subthreshold swing (SS), a negatively shifted threshold voltage, and an 80 times increased carrier mobility. Furthermore, all devices exhibited an n-channel transistor. This was consistent with the $I_{DS} - V_{DS}$ output characteristics shown in Figure 6f,g. The output curves were linear in the low-bias range $(V_{DS} < 20 \text{ and } 2 \text{ V for SiO}_2 \text{ and } Al_2O_3$, respectively) and saturated in the higher drain bias region.

Having confirmed the advantages of using Al_2O_3 as the dielectric, MoS_2 TFT devices were fabricated with different patterned MoS_2 channel thicknesses from different precursor concentrations. Figure S1 shows the hysteresis of the gate transfer and output characteristics of MoS_2/SiO_2 TFTs, among which the best performance belonged to the devices fabricated from the 50 mM precursor solution. The highlightable electrical properties of these 50 mM MoS_2 devices were an on/off current ratio of ~10⁴ and a mobility of 0.024 cm² V⁻¹ s⁻¹ (Table S1), which were two and one order of magnitude increases compared to those of the devices with a thicker MoS_2 layer, respectively. Moreover, as with the Al_2O_3 -based TFTs, the hysteresis became smaller with a thicker MoS_2 . This thickness-dependent behavior indicates that the surface of the MoS_2 plays a crucial role in hysteresis [15].

In general, each MoS₂ TFT with a SiO₂ dielectric possessed fluctuated and much lower features than its counterpart with an Al₂O₃ layer. Meanwhile, the MoS₂/Al₂O₃ devices' performance was commonly more stable under different precursor concentrations (Figure 6h). In addition, the fully saturated output curves obtained for each Al₂O₃-based TFTs were better than those obtained for the corresponding SiO₂-based TFTs that lacked a saturation region with an excessively thick MoS₂ layer. (Figures 6i and S1d–f).

The respective characteristics of all TFT devices with Al_2O_3 are summarized in Table 1. In particular, the Al_2O_3 -based TFTs exhibited optimal properties of $\mu \approx 0.9$ cm² V⁻¹s⁻¹, SS ≈ 1.0 V dec⁻¹, $\frac{I_{on}}{I_{off}} \approx 5 \times 10^5$, and $V_{TH} \approx 7.0$ V, comparable to the properties reported in some previous solution-based works [16–18]. On the other hand, it was also noted that the field-effect mobility of each of our devices was lower than that of devices using different synthesis methods of MoS_2 , such as the CVD methods [19] or other S/D electrodes [20]. This lower mobility was a result of the following factors stemming from using the printed Ag line pattern: (1) a rough interfacial contact between Ag and MoS_2 , (2) an imperfectly clean area between the S/D electrodes due to the Ag printing process, and (3) a different barrier of the channel layer and S/D electrodes. Eventually, the result of this work confirmed that Al_2O_3 and Ag could be used as dielectric and gate electrode materials for MoS_2 TFTs, respectively. In essence, using Al_2O_3 dielectrics instead of SiO₂ can improve the mobility, current ratio, S-S factor, and hysteresis of TFT devices based on them, due to the screening effect of the dielectric on carrier scattering.

Gate Dielectric	Concentration mM	I _{on} /I _{off}	S-S (V dec ⁻¹)	V _{TH} (V)	μ (cm ² V ⁻¹ s ⁻¹)	Hysteresis (V)
40 nm Al ₂ O ₃	25	$(2.1\pm1.7)\times10^5$	2.2 ± 0.9	13.1 ± 2.4	0.29 ± 0.23	4.0 ± 0.65
	50	$(2.7\pm1.7)\times10^5$	1.3 ± 0.4	11.0 ± 1.6	0.33 ± 0.06	4.1 ± 0.15
	75	$(7.5\pm2.1)\times10^4$	4.1 ± 0.9	10.1 ± 1.5	0.42 ± 0.14	2.5 ± 0.23
	100	$(1.7\pm1.1)\times10^5$	2.5 ± 0.5	9.5 ± 2.9	0.58 ± 0.3	3.2 ± 0.57

Table 1. Respective characteristics of the Al₂O₃-based printed MoS₂ TFTs.

4. Conclusions

We demonstrated that an EHD jet printer could be used for multi-printing of a MoS_2 semiconductor and Ag electrodes for TFT fabrication. The MoS_2 pattern was obtained from the printed precursor solution after simple annealing. The MoS_2 films proved to be undamaged without wrinkles after transferring them to another substrate. When employing a high-k gate dielectric, all electrical properties of the TFTs could be improved due to the screening effect of the dielectric on carrier scattering. A controllable hysteretic behavior achieved by varying the MoS_2 thickness and the dielectric materials showed the potential for electronic device applications. Concurrently, the application of a stable printing technique for 2D materials for the synthesis of semiconductors and commercial pastes for the fabrication of electronics on a large scale.

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/nano13010194/s1, Figure S1: (a–c) Transfer characteristic curves with hysteresis behavior and (d–f) output characteristic curves of SiO₂-based MoS₂ TFTs prepared from 50 mM, 75 mM and 100 mM solution concentrations.; Table S1: Characteristics of the SiO₃-based printed MoS₂ TFTs.

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