

Supplementary Materials

# Vanadium Oxide-Poly(3,4-ethylenedioxythiophene) Nanocomposite as High-Performance Cathode for Aqueous Zn-Ion Batteries: The Structural and Electrochemical Characterization

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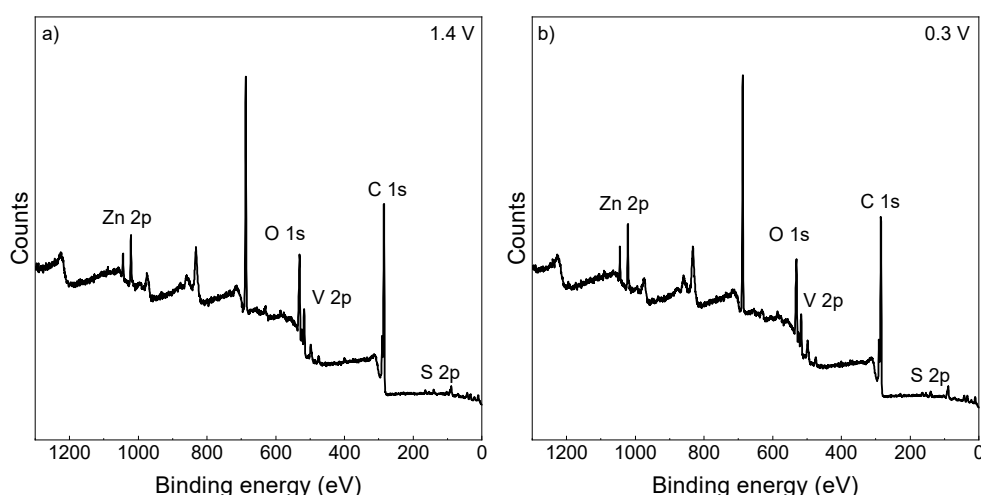
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**Figure S1.** Ex situ XPS survey spectra of VO@PEDOT electrode materials: (a) in the charged state; (b) in the discharged state.

**Table S1.** The comparison of the VO@PEDOT cathode with other V<sub>2</sub>O<sub>5</sub>-based cathodes for Zn-ion batteries.

Material	Electrolyte	Specific capacity, mA·h·g <sup>-1</sup> (current density)	Ref.
VO@PEDOT nanosheets	3 M ZnSO <sub>4</sub>	390 (0.3 A·g <sup>-1</sup> )	This work
		357 (1 A·g <sup>-1</sup> )	
		274 (5 A·g <sup>-1</sup> )	
		192 (10 A·g <sup>-1</sup> )	
Al-V <sub>10</sub> O <sub>24</sub> nanobelts	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	370 (0.5 A·g <sup>-1</sup> )	[1]
		327 (1 A·g <sup>-1</sup> )	
		232 (5 A·g <sup>-1</sup> )	
		290.5 (0.5 A·g <sup>-1</sup> )	
V <sub>10</sub> O <sub>24</sub> /C nanosheets	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	241 (1 A·g <sup>-1</sup> )	[2]
		148.6 (5 A·g <sup>-1</sup> )	
		116.7 (10 A·g <sup>-1</sup> )	
		134 (0.5 A·g <sup>-1</sup> )	
V <sub>10</sub> O <sub>24</sub> ·12H <sub>2</sub> O nanobelts	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	119 (1 A·g <sup>-1</sup> )	[3]
		90 (5 A·g <sup>-1</sup> )	
		80 (10 A·g <sup>-1</sup> )	
		243 (0.5 A·g <sup>-1</sup> )	
V <sub>10</sub> O <sub>24</sub> ·12H <sub>2</sub> O layered	2 M ZnSO <sub>4</sub>	200 (1 A·g <sup>-1</sup> )	[4]
		152 (2 A·g <sup>-1</sup> )	
		312 (0.5 A·g <sup>-1</sup> )	
		283 (1 A·g <sup>-1</sup> )	
V <sub>2</sub> O <sub>5</sub> /PANI nanocomposite	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	201 (6 A·g <sup>-1</sup> )	[5]
		278 (0.5 A·g <sup>-1</sup> )	
		264 (1 A·g <sup>-1</sup> )	
		202 (5 A·g <sup>-1</sup> )	
V <sub>2</sub> O <sub>5</sub> /PANI crisscrossed leaf-like	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	156 (10 A·g <sup>-1</sup> )	[6]
		333 (0.5 A·g <sup>-1</sup> )	
		307 (1 A·g <sup>-1</sup> )	
		121 (5 A·g <sup>-1</sup> )	
V <sub>2</sub> O <sub>5</sub> /THF nanobelts	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	360 (0.5 A·g <sup>-1</sup> )	[6]
		342 (1 A·g <sup>-1</sup> )	
		272 (5 A·g <sup>-1</sup> )	
		216 (10 A·g <sup>-1</sup> )	
PANI-V <sub>2</sub> O <sub>5</sub> nanosheets (100 °C)	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	297 (0.5 A·g <sup>-1</sup> )	[7]
		276 (1 A·g <sup>-1</sup> )	
		199 (5 A·g <sup>-1</sup> )	
		133 (10 A·g <sup>-1</sup> )	
PANI-V <sub>2</sub> O <sub>5</sub> nanosheets (120 °C)	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	295 (0.5 A·g <sup>-1</sup> )	[7]
		250 (1 A·g <sup>-1</sup> )	
		190 (5 A·g <sup>-1</sup> )	
		~340 (0.5 A·g <sup>-1</sup> )	
(PANI) <sub>x</sub> V <sub>2</sub> O <sub>5</sub> nanosheets	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	250 (1 A·g <sup>-1</sup> )	[8]
		190 (5 A·g <sup>-1</sup> )	
		~340 (0.5 A·g <sup>-1</sup> )	
		~340 (0.5 A·g <sup>-1</sup> )	
PANI/V <sub>2</sub> O <sub>5</sub>	2 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	~340 (0.5 A·g <sup>-1</sup> )	[9]

aggregated flakes		~330 (1 A·g <sup>-1</sup> ) ~290 (5 A·g <sup>-1</sup> ) 235 (20 A·g <sup>-1</sup> ) 380 (0.5 A·g <sup>-1</sup> )	
PANI/V <sub>2</sub> O <sub>5</sub> sponge-like particles	2 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	340 (1 A·g <sup>-1</sup> ) 247 (2 A·g <sup>-1</sup> ) 327 (0.5 A·g <sup>-1</sup> )	[10]
PPy/VOH nanosheets	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	303 (1 A·g <sup>-1</sup> ) 281 (2 A·g <sup>-1</sup> ) 186.4 (0.5 A·g <sup>-1</sup> )	[11]
V <sub>2</sub> O <sub>5</sub> @PPy	2 M ZnSO <sub>4</sub>	101.8 (1 A·g <sup>-1</sup> ) 65.3 (5 A·g <sup>-1</sup> ) 360 (1 A·g <sup>-1</sup> )	[12]
V <sub>2</sub> O <sub>5</sub> @PEDOT monolithic grains	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	280 (5 A·g <sup>-1</sup> ) 197 (10 A·g <sup>-1</sup> ) ~325 (0.5 A·g <sup>-1</sup> )	[13]
PEDOT-VO 3D flowers	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	~290 (5 A·g <sup>-1</sup> ) ~275 (10 A·g <sup>-1</sup> ) 282.2 (0.5 A·g <sup>-1</sup> )	[14]
V <sub>2</sub> O <sub>5</sub> @PEDOT/CC core/shell nanosheet arrays	2.5 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	274.5 (1 A·g <sup>-1</sup> ) 254.1 (5 A·g <sup>-1</sup> ) 240.2 (10 A·g <sup>-1</sup> ) 448.5 (0.5 A·g <sup>-1</sup> )	[15]
VO <sub>2</sub> /PEDOT nanobelts	3 M Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	430.4 (1 A·g <sup>-1</sup> ) 347.2 (5 A·g <sup>-1</sup> ) 231.2 (10 A·g <sup>-1</sup> ) 375 (0.5 A·g <sup>-1</sup> )	[16]
V <sub>2</sub> O <sub>5</sub> nanopaper	2 M ZnSO <sub>4</sub>	355 (1 A·g <sup>-1</sup> ) 290 (6 A·g <sup>-1</sup> ) 219 (10 A·g <sup>-1</sup> )	[17]

## References

- Li, Q.; Wei, T.; Ma, K.; Yang, G.; Wang, C. Boosting the Cyclic Stability of Aqueous Zinc-Ion Battery Based on Al-Doped V10O24·12H2O Cathode Materials. *ACS Appl. Mater. Interfaces* **2019**, *11*, 20888–20894, doi:10.1021/acsami.9b05362.
- Wu, W.; Wang, S.; Zhang, C.; Hou, S.; Zhang, L. Facile and Scalable Synthesis of 3D Structures of V10O24·12H2O Nanosheets Coated with Carbon toward Ultrafast and Ultrastable Zinc Storage. *ACS Appl. Mater. Interfaces* **2021**, *13*, 18704–18712, doi:10.1021/acsami.1c00749.
- Wei, T.; Li, Q.; Yang, G.; Wang, C. High-rate and durable aqueous zinc ion battery using dendritic V10O24·12H2O cathode material with large interlamellar spacing. *Electrochim. Acta* **2018**, *287*, 60–67, doi:10.1016/j.electacta.2018.08.040.
- Liu, W.; Dong, L.; Jiang, B.; Huang, Y.; Wang, X.; Xu, C.; Kang, Z.; Mou, J.; Kang, F. Layered vanadium oxides with proton and zinc ion insertion for zinc ion batteries. *Electrochim. Acta* **2019**, *320*, doi:10.1016/j.electacta.2019.134565.
- Du, Y.; Wang, X.; Man, J.; Sun, J. A novel organic-inorganic hybrid V2O5@polyaniline as high-performance cathode for aqueous zinc-ion batteries. *Mater. Lett.* **2020**, *272*, 127813, doi:10.1016/j.matlet.2020.127813.
- Yan, H.; Ru, Q.; Gao, P.; Shi, Z.; Gao, Y.; Chen, F.; Chi-Chun Ling, F.; Wei, L. Organic pillars pre-intercalated V4+-V2O5·3H2O

- nanocomposites with enlarged interlayer and mixed valence for aqueous Zn-ion storage. *Appl. Surf. Sci.* **2020**, *534*, 147608, doi:10.1016/j.apsusc.2020.147608.
7. Chen, S.; Li, K.; Hui, K.S.; Zhang, J. Regulation of Lamellar Structure of Vanadium Oxide via Polyaniline Intercalation for High-Performance Aqueous Zinc-Ion Battery. *Adv. Funct. Mater.* **2020**, *30*, 1–11, doi:10.1002/adfm.202003890.
  8. Li, R.; Xing, F.; Li, T.; Zhang, H.; Yan, J.; Zheng, Q.; Li, X. Intercalated polyaniline in V<sub>2</sub>O<sub>5</sub> as a unique vanadium oxide bronze cathode for highly stable aqueous zinc ion battery. *Energy Storage Mater.* **2021**, *38*, 590–598, doi:10.1016/j.ensm.2021.04.004.
  9. Yin, C.; Pan, C.; Liao, X.; Pan, Y.; Yuan, L. Regulating the Interlayer Spacing of Vanadium Oxide by In Situ Polyaniline Intercalation Enables an Improved Aqueous Zinc-Ion Storage Performance. *ACS Appl. Mater. Interfaces* **2021**, *13*, 39347–39354, doi:10.1021/acsami.1c09722.
  10. Wang, Z.; Tang, X.; Yuan, S.; Bai, M.; Wang, H.; Liu, S.; Zhang, M.; Ma, Y. Engineering Vanadium Pentoxide Cathode for the Zero-Strain Cation Storage via a Scalable Intercalation-Polymerization Approach. *Adv. Funct. Mater.* **2021**, *31*, 2100164, doi:10.1002/adfm.202100164.
  11. Feng, Z.; Sun, J.; Liu, Y.; Jiang, H.; Hu, T.; Cui, M.; Tian, F.; Meng, C.; Zhang, Y. Polypyrrole-intercalation tuning lamellar structure of V<sub>2</sub>O<sub>5</sub>·nH<sub>2</sub>O boosts fast zinc-ion kinetics for aqueous zinc-ion battery. *J. Power Sources* **2022**, *536*, 231489, doi:10.1016/j.jpowsour.2022.231489.
  12. Dong, R.; Zhang, T.; Liu, J.; Li, H.; Hu, D.; Liu, X.; Xu, Q. Mechanistic Insight into Polypyrrole Coating on V<sub>2</sub>O<sub>5</sub> Cathode for Aqueous Zinc-Ion Battery. *ChemElectroChem* **2022**, *9*, doi:10.1002/celc.202101441.
  13. Yao, Z.; Wu, Q.; Chen, K.; Liu, J.; Li, C. Shallow-layer pillaring of a conductive polymer in monolithic grains to drive superior zinc storage: Via a cascading effect. *Energy Environ. Sci.* **2020**, *13*, 3149–3163, doi:10.1039/d0ee01531h.
  14. Li, S.; Wei, X.; Wu, C.; Zhang, B.; Wu, S.; Lin, Z. Constructing Three-Dimensional Structured V<sub>2</sub>O<sub>5</sub>/Conductive Polymer Composite with Fast Ion/Electron Transfer Kinetics for Aqueous Zinc-Ion Battery. *ACS Appl. Energy Mater.* **2021**, *4*, 4208–4216, doi:10.1021/acsam.1c00573.
  15. Xu, D.; Wang, H.; Li, F.; Guan, Z.; Wang, R.; He, B.; Gong, Y.; Hu, X. Conformal Conducting Polymer Shells on V<sub>2</sub>O<sub>5</sub> Nanosheet Arrays as a High-Rate and Stable Zinc-Ion Battery Cathode. *Adv. Mater. Interfaces* **2019**, *6*, 1–8, doi:10.1002/admi.201801506.
  16. Liu, X.; Xu, G.; Zhang, Q.; Huang, S.; Li, L.; Wei, X.; Cao, J.; Yang, L.; Chu, P.K. Ultrathin hybrid nanobelts of single-crystalline VO<sub>2</sub> and Poly(3,4-ethylenedioxythiophene) as cathode materials for aqueous zinc ion batteries with large capacity and high-rate capability. *J. Power Sources* **2020**, *463*, doi:10.1016/j.jpowsour.2020.228223.
  17. Li, Y.; Huang, Z.; Kalambate, P.K.; Zhong, Y.; Huang, Z.; Xie, M.; Shen, Y.; Huang, Y. V<sub>2</sub>O<sub>5</sub> nanopaper as a cathode material with high capacity and long cycle life for rechargeable aqueous zinc-ion battery. *Nano Energy* **2019**, *60*, 752–759, doi:10.1016/j.nanoen.2019.04.009.