



Facile Synthesis of Hierarchical $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ Nanoflowers Assembled by Nanosheets as a Novel Anode Material for High-Performance Lithium-ion Batteries

Xiao-Xu Ji ¹, Qing-Huai Zhao ¹, Hao Chen ², Xin-Wei Luo ², Yi Shang ² and Xiao-Di Liu ^{2,*}

¹ College of Physics and Electronic Engineering, Nanyang Normal University, Nanyang, 473061, China; xxji2010@163.com (X.-X.J.); zqh1262022@126.com (Q.-H.Z.);

² College of Chemistry and Pharmaceutical Engineering, Nanyang Normal University, Nanyang, 473061, China; xiaohao819@126.com (H.C.); 17527755065@163.com (X.-W.L.); s3030199101@163.com (Y.S.);

* Correspondence: 20122029@nynu.edu.cn

Abstract: As novel anodic materials for lithium-ion batteries (LIBs), transitional metal selenites can transform into metal oxide/selenide heterostructures in the first cycle, which can help to enhance the Li^+ storage performance, especially high discharge capacity. Herein, well-defined hierarchical $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ nanoflowers assembled by 10-nm-thick nanosheets are successfully synthesized *via* a facile one-step hydrothermal method. When used as anodic material for LIBs, the $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ nanoflowers exhibit a considerable high discharge capacity of 1064.1 mAh g^{-1} at a current density of 0.1 A g^{-1} . In addition, the obtained anode possesses good rate capability and cycling stability. Owing to the superior electrochemical properties, the $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ nanoflowers would be served as promising anodic materials for high-performance LIBs.

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Keywords: hydrothermal method; $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$; nanoflowers; nanosheets; lithium-ion batteries

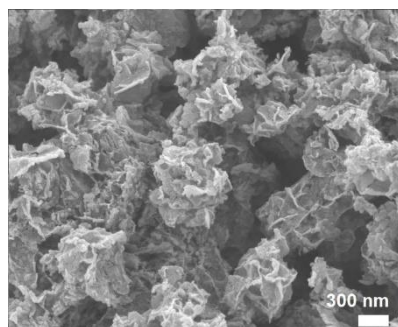


Figure S1. FESEM image of the $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ electrode after 180 cycles at 0.5 A g^{-1} .

Table S1. The comparison of the electrochemical performance the $\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ nanoflowers and other previously reported Co,Se-based anodes for LIBs.

Sample	Discharge Capacity	Cyclability	Ref.
$\text{CoSeO}_3 \cdot 2\text{H}_2\text{O}$ nanoflowers	$765.5 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$626 \text{ mAh g}^{-1} / 180\text{th} (0.5 \text{ C})$	this work
CoSe ₂ decorated NbSe ₂ nanosheets	$466 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$364.7 \text{ mAh g}^{-1} / 1500\text{th} (5 \text{ C})$	[1]
CoSe ₂ @MXene	$763 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$910 \text{ mAh g}^{-1} / 100\text{th} (0.2 \text{ C})$	[2]
ZnSe/CoSe ₂ @N-C composites	$399.6 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$755 \text{ mAh g}^{-1} / 100\text{th} (0.1 \text{ C})$	[3]
ZnSe/CoSe ₂ -C	$579 \text{ mAh g}^{-1} (0.8 \text{ A g}^{-1})$	$700 \text{ mAh g}^{-1} / 500\text{th} (1 \text{ C})$	[4]
Mesoporous hollow Co ₃ O ₄ hierarchical architecture	$629.4 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$555.7 \text{ mAh g}^{-1} / 1000\text{th} (1 \text{ C})$	[5]
Porous Co ₃ O ₄ spheres	$399.6 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$230 \text{ mAh g}^{-1} / 1000\text{th} (2 \text{ C})$	[6]
Urchin-like CoSe ₂	$390 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$410 \text{ mAh g}^{-1} / 1800\text{th} (1 \text{ C})$	[7]
NiCo ₂ O ₄ Hollow Spheres	$662 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$706 \text{ mAh g}^{-1} / 100\text{th} (0.2 \text{ C})$	[8]
MoSe ₂ /MoO ₂ Composite	$741 \text{ mAh g}^{-1} (1 \text{ A g}^{-1})$	$547 \text{ mAh g}^{-1} / 300\text{th} (2 \text{ C})$	[9]

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