



Review

# Metal Oxide Based Heterojunctions for Gas Sensors: A Review

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**Table S1.** The assembled strategies of n-n heterojunctions and their gas sensing performances.

Sensing composites	Type of junctions	Synthesis method	Working temperature (°C)	Sensing gas/conc. (ppm)	Sensor response ( $R_a/R_g$ ) or ( $R_g/R_a$ ) <sup>a</sup>	Response time (s)	Recovery time (s)	References
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> nanoparticles decorated ZnO nanoparticles	n-n	Sol-gel and spin-coating method	Room temperature	NH <sub>3</sub> /0.4	10000	20	20	[1]
SnO <sub>2</sub> nanoparticles modified ZnO nanorods	n-n	Wet chemical method	20	NO <sub>2</sub> /200	2.5	720	840	[2]
SnO <sub>2</sub> -ZnO composite Nanofibers	n-n	Electrospinning	300	H <sub>2</sub> /10	168.6	172	530	[3]
CeO <sub>2</sub> decorated ZnO nanosheets	n-n	Hydrothermal method combination with wet impregnation method	310	Ethanol/100	90	20	4	[4]
Sb-doped SnO <sub>2</sub> covered ZnO nano-heterojunction	n-n	Microwave hydrothermal method	NA	NO <sub>2</sub> /1	1.95	16	NA <sup>b</sup>	[5]
SnO <sub>2</sub> -coated ZnO nanowire	n-n	Solvothermal treatment followed with calcination	240	N-butylamine/200	9.2	< 30	< 55	[6]
TiO <sub>2</sub> nanoparticles decorated ZnO nanorods	n-n	Sol-gel technology followed by a spin coating method	120	NO <sub>2</sub> /50	112	14	6	[7]
ZnO tetrapod alloyed with Fe <sub>2</sub> O <sub>3</sub> nanoparticles	n-n	Flame transport synthesis	300	Ethanol/200	57.56	65	220	[8]
SnO <sub>2</sub> -TiO <sub>2</sub> composite oxide	n-n	Sol-gel method	280	Ethanol/200	62	~ 10	~ 8	[9]
CeO <sub>2</sub> -SnO <sub>2</sub> mixed oxide heterostructure	n-n	Hydrothermal method	300	H <sub>2</sub> /40	19.23	17	24	[10]
SnO <sub>2</sub> nanoparticles decorated SnS <sub>2</sub>	n-n	In-situ high-temperature oxidation	80	NO <sub>2</sub> /8	5.3	159	297	[11]
Ultrathin mesoporous ZnO-SnO <sub>2</sub> nanosheets	n-n	Urea decomposition method and subsequently calcination	240	Ethanol/50	80	7	42	[12]
TiO <sub>2</sub> nanoparticles decorated SnO <sub>2</sub> nanosheets	n-n	Hydrothermal process and PLD process	260	Triethylamine/100	52.3	16	22	[13]
Mg-doped TiO <sub>2</sub> /SnO <sub>2</sub> nanosheets	n-n	Hydrothermal method combined with and PLD	260	Triethylamine/50	30.4	9	95	[14]
SnO <sub>2</sub> -ZnO composites	n-n	Chemical route	90	H <sub>2</sub> /10000	~ 16	60	75	[15]
SnO <sub>2</sub> nanoparticles decorated MoS <sub>2</sub> nanoflowers	n-n	Hydrothermal method combined with PLD	300	Triethylamine/100	68.7	12	84	[16]
Fe <sub>2</sub> O <sub>3</sub> nanoparticles decorated SnO <sub>2</sub> nanowires	n-n	VSL combine with hydrothermal process	300	Ethanol/200	57.56	~ 65	~ 200	[17]
Mixed SnO <sub>2</sub> /TiO <sub>2</sub> included with carbon nanotubes	n-n	Sol-gel spin-coating method	250	Ethanol/1000	41	NA	NA	[18]
$\alpha$ -MoO <sub>3</sub> /TiO <sub>2</sub> core/shell nanorods	n-n	Hydrothermal method	180	Ethanol/10	4.8	< 40	< 40	[19]
Nano-coaxial Co <sub>3</sub> O <sub>4</sub> /TiO <sub>2</sub> heterojunction	n-n	Anodic oxidation combined with hydrothermal method	260	Ethanol/100	40	1.4	7.2	[20]
Brush-like ZnO-TiO <sub>2</sub> heterojunctions nanofibers	n-n	Electrospinning and hydrothermal process.	320	Ethanol/500	50	5	10	[21]

TiO <sub>2</sub> /V <sub>2</sub> O <sub>5</sub> branched nanoheterostructures	n-n	Electrospinning process combined with annealing	350	Ethanol/100	24.6	6	7	[22]
MoS <sub>2</sub> decorated TiO <sub>2</sub> nanotube	n-n	Anodization combined hydrothermal process	150	Ethanol/100	14.2	~ 15	~ 15	[23]
SnO <sub>2</sub> decorated TiO <sub>2</sub> nanotubes	n-n	Anodization combined with impregnation method.	250	H <sub>2</sub> /1000	1140	2	~ 400	[24]
TiO <sub>2</sub> nanotube arrays by ZnO modification	n-n	Anodization combined with impregnation method.	300	H <sub>2</sub> /100	340	22	2000	[25]
SnO <sub>2</sub> nanoparticles decorated TiO <sub>2</sub> nanofibers	n-n	Electrospinning technique	240	Ethanol/100	9.58	8	10	[26]
WO <sub>3</sub> nanorods decorated ZnO nanoplates	n-n	Hydrothermal treatment	250	NH <sub>3</sub> /300	24	60	50	[27]
WO <sub>3</sub> nanoclusters-SnO <sub>2</sub> film	n-n	Pulsed laser deposition technique	100	NO <sub>2</sub> /10	51000	67	17.05	[28]
SnO <sub>2</sub> nanoparticles modified WO <sub>3</sub> nanolamella	n-n	Acidification method combined with calcination	200	NO <sub>2</sub> /0.2	370	NA	NA	[29]
MoO <sub>3</sub> -ZnO core-shell nanorods	n-n	Hydrothermal method combined with ALD	350	Ethanol/200	7.62	44.88	119.87	[30]
MoS <sub>2</sub> -MoO <sub>3</sub> hybrid microflowers	n-n	Controlled vapor transport process	25	NO <sub>2</sub> /10	1.35	19	182	[31]
ZnO nanoparticles decorated	n-n	Hydrothermal method	250	Ethanol/100	18	2.5	5	[32]
MoO <sub>3</sub> /ZnO composite	n-n	Hydrothermal method	270	H <sub>2</sub> S/100	30	13	29	[33]
MoO <sub>3</sub> /Fe <sub>2</sub> O <sub>3</sub> composite	n-n	Hydrothermal strategy	206	Xylene/100	6.9	87	190	[34]
MoO <sub>3</sub> /V <sub>2</sub> O <sub>5</sub> composite	n-n	Spray pyrolysis deposition	200	NO <sub>2</sub> /100	1.8	118	1182	[35]
MoO <sub>3</sub> /SnO <sub>2</sub> composite	n-n	Wet-chemical method combined with calcination	115	H <sub>2</sub> S/10	43.5	22	10	[36]
Fe <sub>2</sub> O <sub>3</sub> /WO <sub>3</sub> nanocomposites	n-n	Two-step solution-based method	260	Acetone/100	105.8	7	20	[37]
In <sub>2</sub> O <sub>3</sub> / $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> heterostructure nanotubes	n-n	Single-capillary electrospinning method	225	Ethanol/100	24.41	3-5	6-10	[38]
TiO <sub>2</sub> nanoparticles decorated $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> nanorods	n-n	Thermal oxidation combined with solvothermal treatment	300	H <sub>2</sub> S/200	7.4	~ 150	~ 155	[39]
Fe <sub>2</sub> O <sub>3</sub> /ZnO core/shell nanorod	n-n	Solution phase controlled hydrolysis method	200	Ethanol/50	6.48	< 20	< 20	[40]
ZnO/Fe <sub>2</sub> O <sub>3</sub> heterostructure	n-n	Hydrothermal method combined with ALD	250	H <sub>2</sub> S/50	~ 100	19.1	156.5	[41]
ZnO nanoparticles decorated $\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	n-n	Hydrothermal method	200	Acetone/200	~ 43	19	20	[42]
TiO <sub>2</sub> -In <sub>2</sub> O <sub>3</sub> composite nanofibers	n-n	Electrospinning method	25	NO <sub>2</sub> /0.3	NA	0.1	10.3	[43]
Mixed Fe <sub>2</sub> O <sub>3</sub> -In <sub>2</sub> O <sub>3</sub> nanotubes	n-n	Electrospinning	250	Formaldehyde/100	33	5	25	[44]
Fe <sub>2</sub> O <sub>3</sub> nanoparticles modified In <sub>2</sub> O <sub>3</sub> nanowires	n-n	Thermal evaporation followed by solvothermal deposition	200	Acetone/500	~ 10	~ 55	~80	[45]
SnO <sub>2</sub> -In <sub>2</sub> O <sub>3</sub> composite	n-n	Solid-phase reaction method	160	Methanol/100	320.73	32	47	[46]
ZnO-modified In <sub>2</sub> O <sub>3</sub> heterojunction	n-n	Hydrothermal method combined	300	HCHO/100	46.8	6	7	[47]

with ultrasonic re-action								
In <sub>2</sub> O <sub>3</sub> -core/ZnO-shell nanorod	n-n	Thermal evaporation followed by ALD	300	H <sub>2</sub> S/100	1.28	530	500	[48]
Core-Shell In <sub>2</sub> O <sub>3</sub> /ZnO nanoarray	n-n	Hydrothermal process combined with a wet-chemical method	Room temperature	H <sub>2</sub> S/700	925	NA	NA	[49]
SnO <sub>2</sub> hollow spheres decorated with CeO <sub>2</sub> nanoparticles	n-n	Two-step hydrothermal strategy	225	Ethanol/100	37	2	70	[50]
Branch-like $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /TiO <sub>2</sub> nanofiber	n-n	Electrospinning technique and hydrothermal process	95	TMA/50	15	75	112	[51]
ZnO-SnO <sub>2</sub> composite	n-n	Hydrothermal method	26	O <sub>3</sub> /0.6	12	13	90	[52]
In <sub>2</sub> O <sub>3</sub> nanoparticles on ZnO hollow nanotubes	n-n	Hydrothermal method	260	Ethanol/100	68.19	8	10	[53]

<sup>a</sup>In this table, the R<sub>a</sub> and the R<sub>g</sub> were the resistances of the sensor in the air and the resistances of the sensor in the targeted gases, respectively. <sup>b</sup>The NA in the table means that the related information was not given in the references.

**Table S2.** The assembled strategies of n-p heterojunctions and their gas sensing performances.

Sensing composites	Type of junctions	Synthesis method	Working temperature (°C)	Sensing gas/conc. (ppm)	Sensor response ( $R_a/R_g$ ) or ( $R_g/R_a$ ) <sup>a</sup>	Response time (s)	Recovery time (s)	References
NiO nanoparticles modified ZnO nanorods	n-p	Two-step chemical bath deposition	330	Acetone/100	~ 13	~ 5	~ 10	[54]
NiO/ZnO nanofibers	n-p	Electrospinning	260	TMA/100	892	18	20	[55]
CuO nanoparticles decorated ZnO nanorods	n-p	Solution-based synthesis method	25	NO/10	3.65	100	> 600	[56]
CdO nanoparticles-decorated flower-like ZnO hollow microspheres	n-p	Two-step hydrothermal strategy	250	Ethanol/100	65.5	2	136	[57]
ZnO nanowire arrays/CuO nanospheres heterostructure	n-p	Hydrothermal method	122	Ethanol/80	NA <sup>b</sup>	41	99	[58]
ZnO/CuO composite	n-p	Solvothermal method	40	H <sub>2</sub> S/10	23.03	173	> 3000	[59]
PdO-decorated flower-like ZnO structures	n-p	Hydrothermal route combined with heat treatment	320	Ethanol/100	35.4	1	7	[60]
Cr <sub>2</sub> O <sub>3</sub> -functionalization ZnO nanorods	n-p	Carbothermal synthesis combined with solvothermal process	25	Ethanol/200	10.95	26	110	[61]
CuO nanoparticles decorated SnO <sub>2</sub> nanowires	n-p	Two-step CVD	200	H <sub>2</sub> S/1	700	NA	NA	[62]
Co <sub>3</sub> O <sub>4</sub> decorated flower-like SnO <sub>2</sub> nanorods	n-p	Hydrothermal method followed by chemical solution method	280	Xylene/100	47.8	98	107	[63]
NiO-SnO <sub>2</sub> composite nanofibers	n-p	Electrospinning	320	H <sub>2</sub> /100	~ 14	~ 3	~ 3	[64]
NiO/SnO <sub>2</sub> hollow sphere	n-p	Hydrothermal method followed by PLD process	220	Triethylamine/10	48.6	11	34	[65]
CuO nanoparticles decorate SnO <sub>2</sub> nanorods	n-p	Hydrothermal method	60	H <sub>2</sub> S/10	94000	~ 60	NA	[66]
NiO-SnO <sub>2</sub> microflowers	n-p	Hydrothermal route	200	Formaldehyde/100	~ 27.5	20	64	[67]
Bi <sub>2</sub> O <sub>3</sub> -branched SnO <sub>2</sub> nanowires	n-p	Vapor-liquid-solid method	250	NO <sub>2</sub> /2	56.92	~ 195	~ 20	[68]
Co <sub>3</sub> O <sub>4</sub> /SnO <sub>2</sub> flower-like structures	n-p	One-step hydrothermal technique	175	Trimethylamine/5	9.3	19	29	[69]
CeO <sub>2</sub> /TiO <sub>2</sub> core/shell nanorods	n-p	Hydrothermal method	320	Ethanol/1000	5.44	< 45	< 45	[70]
Co <sub>3</sub> O <sub>4</sub> /TiO <sub>2</sub> nanotube heterostructures	n-p	Anodization combined and cathodic deposition	200	H <sub>2</sub> /1000	1.06	~ 720	~ 600	[71]
CuO nanoparticles decorated WO <sub>3</sub> nanoplates	n-p	Hydrothermal method	100	H <sub>2</sub> S/5	223	73	450	[72]
NiO-decorated WO <sub>3</sub> nanocomposite	n-p	Hydrothermal process combined with solvothermal method	350	Ethanol/150	~ 80	~ 10	~ 5	[73]
CuO-modified WO <sub>3</sub> thin film	n-p	RF sputtering technique	300	H <sub>2</sub> S/10	534	2	~ 1440	[74]
CuO-functionalized WO <sub>3</sub> nanowires	n-p	Thermal evaporation combined with thermal annealing	300	H <sub>2</sub> S/100	6.73	~ 60	~ 70	[75]
Cr <sub>2</sub> O <sub>3</sub> nanoparticle modified WO <sub>3</sub> nanorods	n-p	Thermal evaporation followed by spin-coating	300	Ethanol/200	5.58	~ 50	~ 60	[76]
WO <sub>3</sub> /CuO composites	n-p	Hydrothermal method	80	H <sub>2</sub> S/5	105.14	42	3500	[77]
CuO nanoparticles decorated $\alpha$ -MoO <sub>3</sub> nanorods	n-p	Hydrothermal method combined with heating	180	H <sub>2</sub> S/50	160.8	~ 20	40	[78]
PdO nanoparticles decorated WO <sub>3</sub> nanoneedle	n-p	Aerosol assisted CVD	200	H <sub>2</sub> /500	1500	120	720	[79]
Fe <sub>2</sub> O <sub>3</sub> /Co <sub>3</sub> O <sub>4</sub> composite	n-p	Hydrothermal route	300	Ethanol/100	10.86	1.36	40.25	[80]
p-NiS/n-In <sub>2</sub> O <sub>3</sub> heterojunction nanocomposites	n-p	Solid-state grinding and calcination	300	Ethanol/100	10.8	8	20	[81]

Bi <sub>2</sub> O <sub>3</sub> nanoparticles decorated In <sub>2</sub> O <sub>3</sub> nanorods	n-p	One-step CVD process	200	Ethanol/200	17.74	23	200	[82]
NiO-In <sub>2</sub> O <sub>3</sub> composite nanofibers	n-p	Electrospinning and calcination method	300	Ethanol/100	78	~ 6	60	[83]
SnO <sub>2</sub> -Co <sub>3</sub> O <sub>4</sub> composite nanofibers	n-p	Electrospinning method	350	C <sub>6</sub> H <sub>6</sub> /1	18.7	13.58	13.65	[84]
ZnO/Co <sub>3</sub> O <sub>4</sub> composite nanoparticle	n-p	Solvothermal route	Room temperature	NO <sub>2</sub> /5	15	~ 17.5	~ 32.5	[85]
Porous flower-like CuO/ZnO nanostructures	n-p	Chemical solution method	220	Ethanol/100	25.5	6	42	[86]
CuO/ZnO heterostructured nanorods	n-p	A photochemical method	500	H <sub>2</sub> S/50	891	~ 820	~ 20	[87]
CuO-modified SnO <sub>2</sub> film	n-p	Screen printing technique combined with firing process	25	H <sub>2</sub> S/1	3672	< 15	~ 240	[88]
CuO nanoparticles modified SnO <sub>2</sub> nanowire	n-p	Chemical vapour deposition	200	H <sub>2</sub> S/2	3261	< 180	< 600	[89]
CuO nanoparticle decorated ZnO nanorod	n-p	Two-stage solution process	100	H <sub>2</sub> S/100	~ 38	~ 120	~ 120	[90]
CuO nanoparticles decorated ZnO nanoparticles	n-p	Hydrothermal method	200	Alcohol/200	3.3	62	83	[91]
CuO/ZnO heterojunction nanorods	n-p	Hydrothermal process combined with annealing	300	Ethanol/100	98.8	7	9	[92]

<sup>a</sup>In this table, the R<sub>a</sub> and the R<sub>g</sub> were the resistances of the sensor in the air and the resistances of the sensor in the targeted gases, respectively. <sup>b</sup>The NA in the table means that the related information was not given in the references.

**Table S3.** The assembled strategies of p-n heterojunctions and their gas sensing performances.

Sensing composites	Type of junctions	Synthesis method	Working temperature (°C)	Sensing gas/conc. (ppm)	Sensor response ( $R_a/R_g$ ) or ( $R_g/R_a$ ) <sup>a</sup>	Response time (s)	Recovery time (s)	References
NiO-SnO <sub>2</sub> foam	p-n	Dip-coating process	210	Toluene/10	19.2	9	8	[93]
NiO/ZnO heterojunction diode	p-n	Spin-coating followed by hydrothermal method	300	Toluene/95	20.1	100	60	[94]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /NiO heterojunction nanorods	p-n	Hydrothermal method	280	Acetone/100	290	28	40	[95]
Hierarchical $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /NiO composites	p-n	Hydrothermal method	200	Toluene/100	16.38	1	12	[96]
Cr <sub>2</sub> O <sub>3</sub> nanoparticles decorated In <sub>2</sub> O <sub>3</sub> nanorods	p-n	Thermal evaporation followed by solvothermal deposition	200	Ethanol/500	10.53	~ 17.5	~ 50	[97]
Fe <sub>2</sub> O <sub>3</sub> /Co <sub>3</sub> O <sub>4</sub> codecorated In <sub>2</sub> O <sub>3</sub> nanorod	p-n	Thermal evaporation followed by solvothermal deposition	200	Ethanol/200	35	10	120	[98]
Co <sub>3</sub> O <sub>4</sub> /ZnO nanocomposites	p-n	Chemical vapor deposition	400	CH <sub>3</sub> COCH <sub>3</sub> /100	1.2	NA <sup>b</sup>	NA	[99]
Co <sub>3</sub> O <sub>4</sub> / $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> heterostructure nanofibers	p-n	Coaxial electrospinning method	240	Acetone/50	11.7	2	20	[100]
Co <sub>3</sub> O <sub>4</sub> -SnO <sub>2</sub> hollow hetero-nanostructures	p-n	Galvanic replacement	275	Ethanol/5	13.5	NA	NA	[101]
Hierarchical Fe <sub>3</sub> O <sub>4</sub> @Co <sub>3</sub> O <sub>4</sub> core-shell microspheres	p-n	Hydrothermal method	160	Acetone/100	100	5	15	[102]
Rh <sub>2</sub> O <sub>3</sub> clusters decorated WO <sub>3</sub> crystallites	p-n	Wet-chemical method	300	CO/0.9	NA	NA	NA	[103]
Co <sub>3</sub> O <sub>4</sub> /SnO <sub>2</sub> core-shell nanospheres	p-n	Hydrothermal method	200	NH <sub>3</sub> /50	13.5	4	17	[104]
ZnO-Co <sub>3</sub> O <sub>4</sub> hollow polyhedrons heterostructure	p-n	Thermal decomposition	200	Ethanol/1000	106	7	236	[105]
ZnO/Co <sub>3</sub> O <sub>4</sub> microspheres	p-n	Solvothermal method combined annealing process	275	Ethanol/50	120	5.6	29	[106]
Co <sub>3</sub> O <sub>4</sub> -TiO <sub>2</sub> composite	p-n	Thermal conversion	115	Xylene/50	113	130	150	[107]
Ultra-thin CuO islands on sputtered SnO <sub>2</sub>	p-n	Sputtering followed by thermal evaporation	150	H <sub>2</sub> S/20	7400	15	118	[108]
CuO/ZnO thin film heterojunction	p-n	Sol-gel technique	300	H <sub>2</sub> /3000	266.5	NA	NA	[109]
CuO@ZnO microcubes	p-n	Two-step solution route	240	Ethanol/50	19.1	5-6	~ 18	[110]
ZnO branched p-Cu <sub>x</sub> O@n-ZnO nanowires	p-n	Hydrothermal method and ALD	250	Acetone/5	3.39	62	90	[111]
CuO-In <sub>2</sub> O <sub>3</sub> core-shell nanowire	p-n	Thermal oxidation combined with heating	300	CO/960	1.6	25	65	[112]
Fe <sub>2</sub> O <sub>3</sub> decorated CuO nanorods	p-n	Solvothermal route	240	Acetone/1000	10.9	64	80	[113]

<sup>a</sup>In this table, the  $R_a$  and the  $R_g$  were the resistances of the sensor in the air and the resistances of the sensor in the targeted gases, respectively. <sup>b</sup>The NA in the table means that the related information was not given in the references.

**Table S4.** The assembled strategies of p-p heterojunctions and their gas sensing performances.

Sensing composites	Type of junctions	Synthesis method	Working temperature (°C)	Sensing gas/conc. (ppm)	Sensor response ( $R_a/R_g$ ) or ( $R_g/R_a$ ) <sup>a</sup>	Response time (s)	Recovery time (s)	References
TeO <sub>2</sub> -core/CuO-shell composite	p-p	Thermal evaporation followed by sputter deposition	150	NO <sub>2</sub> /10	4.25	NA	NA	[114]
CuO-NiO core-shell microspheres	p-p	Two-step hydrothermal method	260	H <sub>2</sub> S/100	47.6	18	29	[115]
Cu <sub>2</sub> O/CuO sub-microspheres	p-p	Hydrothermal treatment	135	H <sub>2</sub> S/1	5.5	~ 60	~ 60	[116]
CuO-NiO nanotubes	p-p	one-pot synthesis	110	Glycol/100	10.35	15	45	[117]
NiFe <sub>2</sub> O <sub>4</sub> nanoparticles-decorated NiO nanosheets	p-p	Solvothermal condition	280	Acetone/50	23	-	-	[118]
NiO/NiCr <sub>2</sub> O <sub>4</sub> nanoparticles	p-p	Hydrothermal route	225	Xylene/100	66.2	1217	591	[119]
CuO nanoparticles decorated NiO nanosheets	p-p	A simple reflux and hydrothermal method	22	NO <sub>2</sub> /100	4.85	2	NA	[120]
Cu <sub>3</sub> Mo <sub>2</sub> O <sub>9</sub> @CuO nanorods	p-p	CVD	RT	NO <sub>2</sub> /5	1.6	49	241	[121]

<sup>a</sup>In this table, the  $R_a$  and the  $R_g$  were the resistances of the sensor in the air and the resistances of the sensor in the targeted gases, respectively. <sup>b</sup>The NA in the table means that the related information was not given in the references.

## References

- [1] Tang, H.; Yan M.; Zhang H.; Li S.; Ma X.; Wang M.; Yang D. A selective NH<sub>3</sub> gas sensor based on Fe<sub>2</sub>O<sub>3</sub>-ZnO nanocomposites at room temperature. *Sensors and Actuators B: Chemical* **2006**, 114, 910–915.
- [2] Lu, G.; Xu J.; Sun J.; Yu Y.; Zhang Y.; Liu F. UV-enhanced room temperature NO<sub>2</sub> sensor using ZnO nanorods modified with SnO<sub>2</sub> nanoparticles. *Sensors and Actuators B: Chemical* **2012**, 162, 82–88.
- [3] Katoch, A.; Kim J.-H.; Kwon Y.J.; Kim H.W.; Kim S.S. Bifunctional Sensing Mechanism of SnO<sub>2</sub>-ZnO Composite Nanofibers for Drastically Enhancing the Sensing Behavior in H<sub>2</sub> Gas. *ACS applied materials & interfaces* **2015**, 7, 11351–11358.
- [4] Hui, G.; Zhu M.; Yang X.; Liu J.; Pan G.; Wang Z. Highly sensitive ethanol gas sensor based on CeO<sub>2</sub>/ZnO binary heterojunction composite. *Materials Letters* **2020**, 278, 128453.
- [5] Wang, Z.; Zhi M.; Xu M.; Guo C.; Man Z.; Zhang Z.; Li Q.; Lv Y.; Zhao W.; Yan J.; Zhai C. Ultrasensitive NO<sub>2</sub> gas sensor based on Sb-doped SnO<sub>2</sub> covered ZnO nano-heterojunction. *Journal of Materials Science* **2021**, 56, 7348–7356.
- [6] Wang, L.; Li J.; Wang Y.; Yu K.; Tang X.; Zhang Y.; Wang S.; Wei C. Construction of 1D SnO<sub>2</sub>-coated ZnO nanowire heterojunction for their improved n-butylamine sensing performances. *Scientific reports* **2016**, 6, 35079.
- [7] Zou, C.W.; Wang J.; Xie W. Synthesis and enhanced NO<sub>2</sub> gas sensing properties of ZnO nanorods/TiO<sub>2</sub> nanoparticles heterojunction composites. *Journal of Colloid and Interface Science* **2016**, 478, 22–28.
- [8] Lupan, O.; Postica V.; Gröttrup J.; Mishra A.K.; De Leeuw N.H.; Adelung R. Enhanced UV and ethanol vapour sensing of a single 3-D ZnO tetrapod alloyed with Fe<sub>2</sub>O<sub>3</sub> nanoparticles. *Sensors and Actuators B: Chemical* **2017**, 245, 448–461.
- [9] Wen, Z.; Tian-Mo L. Gas-sensing properties of SnO<sub>2</sub>-TiO<sub>2</sub>-based sensor for volatile organic compound gas and its sensing mechanism. *Physica B: Condensed Matter* **2010**, 405, 1345–1348.
- [10] Motaung, D.E.; Mhlomo G.H.; Makgwane P.R.; Dhonge B.P.; Cummings F.R.; Swart H.C.; Ray S.S. Ultra-high sensitive and selective H<sub>2</sub> gas sensor manifested by interface of n-n heterostructure of CeO<sub>2</sub>-SnO<sub>2</sub> nanoparticles. *Sensors and Actuators B: Chemical* **2018**, 254, 984–995.
- [11] Gu, D.; Li X.; Zhao Y.; Wang J. Enhanced NO<sub>2</sub> sensing of SnO<sub>2</sub>/SnS<sub>2</sub> heterojunction based sensor. *Sensors and Actuators B: Chemical* **2017**, 244, 67–76.
- [12] Qin, S.; Tang P.; Feng Y.; Li D. Novel ultrathin mesoporous ZnO-SnO<sub>2</sub> n-n heterojunction nanosheets with high sensitivity to ethanol. *Sensors and Actuators B: Chemical* **2020**, 309, 127801.
- [13] Xu, H.; Ju J.; Li W.; Zhang J.; Wang J.; Cao B. Superior triethylamine-sensing properties based on TiO<sub>2</sub>/SnO<sub>2</sub> n-n heterojunction nanosheets directly grown on ceramic tubes. *Sensors and Actuators B: Chemical* **2016**, 228, 634–642.
- [14] Xu, H.; Ju D.; Chen Z.; Han R.; Zhai T.; Yu H.; Liu C.; Wu X.; Wang J.; Cao B. A novel hetero-structure sensor based on Au/Mg-doped TiO<sub>2</sub>/SnO<sub>2</sub> nanosheets directly grown on Al<sub>2</sub>O<sub>3</sub> ceramic tubes. *Sensors and Actuators B: Chemical* **2018**, 273, 328–335.
- [15] Mondal, B.; Basumatari B.; Das J.; Roychaudhury C.; Saha H.; Mukherjee N. ZnO-SnO<sub>2</sub> based composite type gas sensor for selective hydrogen sensing. *Sensors and Actuators B: Chemical* **2014**, 194, 389–396.



- [16] Li, W.; Xu H.; Zhai T.; Yu H.; Chen Z.; Qiu Z.; Song X.; Wang J.; Cao B. Enhanced triethylamine sensing properties by designing Au@SnO<sub>2</sub>/MoS<sub>2</sub> nanostructure directly on alumina tubes. *Sensors and Actuators B: Chemical* **2017**, 253, 97–107.
- [17] Choi, K.S.; Park S.; Chang S.-P. Enhanced ethanol sensing properties based on SnO<sub>2</sub> nanowires coated with Fe<sub>2</sub>O<sub>3</sub> nanoparticles. *Sensors and Actuators B: Chemical* **2017**, 238, 871–879.
- [18] Van Duy, N.; Van Hieu N.; Huy P.T.; Chien N.D.; Thamilselvan M.; Yi J. Mixed SnO<sub>2</sub>/TiO<sub>2</sub> included with carbon nanotubes for gas-sensing application. *Physica E: Low-dimensional Systems and Nanostructures* **2008**, 41, 258–263.
- [19] Chen, Y.-J.; Xiao G.; Wang T.-S.; Zhang F.; Ma Y.; Gao P.; Zhu C.-L.; Zhang E.; Xu Z.; Li Q.-H.  $\alpha$ -MoO<sub>3</sub>/TiO<sub>2</sub> core/shell nanorods: Controlled-synthesis and low-temperature gas sensing properties. *Sensors and Actuators B: Chemical* **2011**, 155, 270–277.
- [20] Liang, Y.Q.; Cui Z.D.; Zhu S.L.; Li Z.Y.; Yang X.J.; Chen Y.J.; Ma J.M. Design of a highly sensitive ethanol sensor using a nano-coaxial p-Co<sub>3</sub>O<sub>4</sub>/n-TiO<sub>2</sub> heterojunction synthesized at low temperature. *Nanoscale* **2013**, 5, 10916–10926.
- [21] Deng, J.; Yu B.; Lou Z.; Wang L.; Wang R.; Zhang T. Facile synthesis and enhanced ethanol sensing properties of the brush-like ZnO–TiO<sub>2</sub> heterojunctions nanofibers. *Sensors and Actuators B: Chemical* **2013**, 184, 21–26.
- [22] Wang, Y.; Zhou Y.; Meng C.; Gao Z.; Cao X.; Li X.; Xu L.; Zhu W.; Peng X.; Zhang B.; Lin Y.; Liu L. A high-response ethanol gas sensor based on one-dimensional TiO<sub>2</sub>/V<sub>2</sub>O<sub>5</sub> branched nanoheterostructures. *Nanotechnology* **2016**, 27, 425503.
- [23] Zhao, P.X.; Tang Y.; Mao J.; Chen Y.X.; Song H.; Wang J.W.; Song Y.; Liang Y.Q.; Zhang X.M. One-Dimensional MoS<sub>2</sub>-Decorated TiO<sub>2</sub> nanotube gas sensors for efficient alcohol sensing. *Journal of Alloys and Compounds* **2016**, 674, 252–258.
- [24] Xun, H.; Zhang Z.; Yu A.; Yi J. Remarkably enhanced hydrogen sensing of highly-ordered SnO<sub>2</sub>-decorated TiO<sub>2</sub> nanotubes. *Sensors and Actuators B: Chemical* **2018**, 273, 983–990.
- [25] Yu, A.; Xun H.; Yi J. Improving Hydrogen Sensing Performance of TiO<sub>2</sub> Nanotube Arrays by ZnO Modification. *Frontiers in Materials* **2019**, 6,
- [26] Chen, K.; Chen S.; Pi M.; Zhang D. SnO<sub>2</sub> nanoparticles/TiO<sub>2</sub> nanofibers heterostructures: In situ fabrication and enhanced gas sensing performance. *Solid State Electron* **2019**, 157, 42–47.
- [27] Nguyen, D.D.; Do D.T.; Vu X.H.; Dang D.V.; Nguyen D.C. ZnO nanoplates surfaced-decorated by WO<sub>3</sub> nanorods for NH<sub>3</sub> gas sensing application. *Advances in Natural Sciences: Nanoscience and Nanotechnology* **2016**, 7, 015004.
- [28] Sharma, A.; Tomar M.; Gupta V. WO<sub>3</sub> nanoclusters–SnO<sub>2</sub> film gas sensor heterostructure with enhanced response for NO<sub>2</sub>. *Sensors and Actuators B: Chemical* **2013**, 176, 675–684.
- [29] Kida, T.; Nishiyama A.; Hua Z.; Suematsu K.; Yuasa M.; Shimanoe K. WO<sub>3</sub> nanolamella gas sensor: porosity control using SnO<sub>2</sub> nanoparticles for enhanced NO<sub>2</sub> sensing. *Langmuir : the ACS journal of surfaces and colloids* **2014**, 30, 2571–2579.
- [30] Lee, W.I.; Bonyani M.; Lee J.K.; Lee C.; Choi S.-B. Volatile organic compound sensing properties of MoO<sub>3</sub>–ZnO core–shell nanorods. *Current Applied Physics* **2018**, 18, S60–S67.
- [31] Kumar, R.; Goel N.; Mishra M.; Gupta G.; Fanetti M.; Valant M.; Kumar M. Growth of MoS<sub>2</sub>–MoO<sub>3</sub> Hybrid Microflowers via Controlled Vapor Transport Process for Efficient Gas Sensing at Room Temperature. *Advanced Materials Interfaces* **2018**, 5, 1800071.
- [32] Li, J.; Liu H.; Fu H.; Xu L.; Jin H.; Zhang X.; Wang L.; Yu K. Synthesis of 1D  $\alpha$ -MoO<sub>3</sub>/0D ZnO heterostructure nanobelts with enhanced gas sensing properties. *Journal of Alloys and Compounds* **2019**, 788, 248–256.
- [33] Yu, H.-L.; Li L.; Gao X.-M.; Zhang Y.; Meng F.; Wang T.-S.; Xiao G.; Chen Y.-J.; Zhu C.-L. Synthesis and H<sub>2</sub>S gas sensing properties of cage-like  $\alpha$ -MoO<sub>3</sub>/ZnO composite. *Sensors and Actuators B: Chemical* **2012**, 171–172, 679–685.
- [34] Jiang, D.; Wei W.; Li F.; Li Y.; Liu C.; Sun D.; Feng C.; Ruan S. Xylene gas sensor based on  $\alpha$ -MoO<sub>3</sub>/ $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> heterostructure with high response and low operating temperature. *RSC Advances* **2015**, 5, 39442–39448.
- [35] Mane, A.A.; Nikam S.A.; Moholkar A.V. NO<sub>2</sub> gas sensing properties of sprayed composite porous MoO<sub>3</sub>-V<sub>2</sub>O<sub>5</sub> thin films. *Mater Chem Phys* **2018**, 216, 294–304.
- [36] Gao, X.; Ouyang Q.; Zhu C.; Zhang X.; Chen Y. Porous MoO<sub>3</sub>/SnO<sub>2</sub> Nanoflakes with n–n Junctions for Sensing H<sub>2</sub>S. *ACS Applied Nano Materials* **2019**, 2, 2418–2425.
- [37] Xue, D.; Zong F.; Zhang J.; Lin X.; Li Q. Synthesis of Fe<sub>2</sub>O<sub>3</sub>/WO<sub>3</sub> nanocomposites with enhanced sensing performance to acetone. *Chem Phys Lett* **2019**, 716, 61–68.
- [38] Zhao, C.; Zhang G.; Han W.; Fu J.; He Y.; Zhang Z.; Xie E. Electrospun In<sub>2</sub>O<sub>3</sub>/ $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> heterostructure nanotubes for highly sensitive gas sensor applications. *CrystEngComm* **2013**, 15, 6491.
- [39] Kheel, H.; Sun G.-J.; Lee J.K.; Lee S.; Dwivedi R.P.; Lee C. Enhanced H<sub>2</sub>S sensing performance of TiO<sub>2</sub>-decorated  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> nanorod sensors. *Ceram Int* **2016**, 42, 18597–18604.
- [40] Si, S.; Li C.; Wang X.; Peng Q.; Li Y. Fe<sub>2</sub>O<sub>3</sub>/ZnO core–shell nanorods for gas sensors. *Sensors and Actuators B: Chemical* **2006**, 119, 52–56.
- [41] Fan, K.; Guo J.; Cha L.; Chen Q.; Ma J. Atomic layer deposition of ZnO onto Fe<sub>2</sub>O<sub>3</sub> nanoplates for enhanced H<sub>2</sub>S sensing. *Journal of Alloys and Compounds* **2017**, 698, 336–340.
- [42] Kaneti, Y.V.; Moriceau J.; Liu M.; Yuan Y.; Zakaria Q.; Jiang X.; Yu A. Hydrothermal synthesis of ternary  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>–ZnO–Au nanocomposites with high gas-sensing performance. *Sensors and Actuators B: Chemical* **2015**, 209, 889–897.
- [43] Wang, L.; Gao J.; Wu B.; Kan K.; Xu S.; Xie Y.; Li L.; Shi K. Designed Synthesis of In<sub>2</sub>O<sub>3</sub> Beads@TiO<sub>2</sub>-In<sub>2</sub>O<sub>3</sub> Composite Nanofibers for High Performance NO<sub>2</sub> Sensor at Room Temperature. *ACS applied materials & interfaces* **2015**, 7, 27152–27159.
- [44] Chi, X.; Liu C.; Liu L.; Li S.; Li H.; Zhang X.; Bo X.; Shan H. Enhanced formaldehyde-sensing properties of mixed Fe<sub>2</sub>O<sub>3</sub>–In<sub>2</sub>O<sub>3</sub> nanotubes. *Mat Sci Semicon Proc* **2014**, 18, 160–164.
- [45] Kim, S.; Park S.; Sun G.-J.; Hyun S.K.; Kim K.-K.; Lee C. Enhanced acetone gas sensing performance of the multiple-networked Fe<sub>2</sub>O<sub>3</sub>-functionalized In<sub>2</sub>O<sub>3</sub> nanowire sensor. *Current Applied Physics* **2015**, 15, 947–952.
- [46] Li, Y.; Deng D.; Xing X.; Chen N.; Liu X.; Xiao X.; Wang Y. A high performance methanol gas sensor based on palladium-platinum-In<sub>2</sub>O<sub>3</sub> composited nanocrystalline SnO<sub>2</sub>. *Sensors and Actuators B: Chemical* **2016**, 237, 133–141.

- [47] Ma, L.; Fan H.; Tian H.; Fang J.; Qian X. The n-ZnO/n-In<sub>2</sub>O<sub>3</sub> heterojunction formed by a surface-modification and their potential barrier-control in methanal gas sensing. *Sensors and Actuators B: Chemical* **2016**, 222, 508–516.
- [48] Park, S.; Kim H.; Jin C.; Lee C. Enhanced gas sensing properties of multiple networked In<sub>2</sub>O<sub>3</sub>-core/ZnO-shell nanorod sensors. *Journal of nanoscience and nanotechnology* **2013**, 13, 3427–3432.
- [49] Zang, W.; Nie Y.; Zhu D.; Deng P.; Xing L.; Xue X. Core-Shell In<sub>2</sub>O<sub>3</sub>/ZnO Nanoarray Nanogenerator as a Self-Powered Active Gas Sensor with High H<sub>2</sub>S Sensitivity and Selectivity at Room Temperature. *The Journal of Physical Chemistry C* **2014**, 118, 9209–9216.
- [50] Liu, J.; Dai M.; Wang T.; Sun P.; Liang X.; Lu G.; Shimanoe K.; Yamazoe N. Enhanced Gas Sensing Properties of SnO<sub>2</sub> Hollow Spheres Decorated with CeO<sub>2</sub> Nanoparticles Heterostructure Composite Materials. *ACS applied materials & interfaces* **2016**, 8, 6669–6677.
- [51] Lou, Z.; Li F.; Deng J.; Wang L.; Zhang T. Branch-like hierarchical heterostructure (alpha-Fe<sub>2</sub>O<sub>3</sub>/TiO<sub>2</sub>): a novel sensing material for trimethylamine gas sensor. *ACS applied materials & interfaces* **2013**, 5, 12310–12316.
- [52] Da Silva, L.F.; M'peko J.C.; Catto A.C.; Bernardini S.; Mastelaro V.R.; Aguir K.; Ribeiro C.; Longo E. UV-enhanced ozone gas sensing response of ZnO-SnO<sub>2</sub> heterojunctions at room temperature. *Sensors and Actuators B: Chemical* **2017**, 240, 573–579.
- [53] Wei, N.; Cui H.; Wang X.; Xie X.; Wang M.; Zhang L.; Tian J. Hierarchical assembly of In<sub>2</sub>O<sub>3</sub> nanoparticles on ZnO hollow nanotubes using carbon fibers as templates: Enhanced photocatalytic and gas-sensing properties. *J Colloid Interface Sci* **2017**, 498, 263–270.
- [54] Liu, Y.; Li G.; Mi R.; Deng C.; Gao P. An environment-benign method for the synthesis of p-NiO/n-ZnO heterostructure with excellent performance for gas sensing and photocatalysis. *Sensors and Actuators B: Chemical* **2014**, 191, 537–544.
- [55] Li, C.; Feng C.; Qu F.; Liu J.; Zhu L.; Lin Y.; Wang Y.; Li F.; Zhou J.; Ruan S. Electrospun nanofibers of p-type NiO/n-type ZnO heterojunction with different NiO content and its influence on trimethylamine sensing properties. *Sensors and Actuators B: Chemical* **2015**, 207, 90–96.
- [56] Hwang, H.; Kim H.; Kim D. Fabrication and Characterization of CuO Nanoparticles/ZnO Nanorods Heterojunction Structure for Room Temperature NO Gas Sensor Application. *Journal of Nanoscience and Nanotechnology* **2016**, 16, 11608–11612.
- [57] Wang, T.; Kou X.; Zhao L.; Sun P.; Liu C.; Wang Y.; Shimanoe K.; Yamazoe N.; Lu G. Flower-like ZnO hollow microspheres loaded with CdO nanoparticles as high performance sensing material for gas sensors. *Sensors and Actuators B: Chemical* **2017**, 250, 692–702.
- [58] Cai, L.; Li H.; Zhang H.; Fan W.; Wang J.; Wang Y.; Wang X.; Tang Y.; Song Y. Enhanced performance of the tangerines-like CuO-based gas sensor using ZnO nanowire arrays. *Materials Science in Semiconductor Processing* **2020**, 118, 105196.
- [59] Wang, X.; Li S.; Xie L.; Li X.; Lin D.; Zhu Z. Low-temperature and highly sensitivity H<sub>2</sub>S gas sensor based on ZnO/CuO composite derived from bimetal metal-organic frameworks. *Ceramics International* **2020**, 46, 15858–15866.
- [60] Lou, Z.; Deng J.; Wang L.; Wang L.; Fei T.; Zhang T. Toluene and ethanol sensing performances of pristine and PdO-decorated flower-like ZnO structures. *Sensors and Actuators B: Chemical* **2013**, 176, 323–329.
- [61] Park, S.; Sun G.J.; Jin C.; Kim H.W.; Lee S.; Lee C. Synergistic Effects of a Combination of Cr<sub>2</sub>O<sub>3</sub>-Functionalization and UV-Irradiation Techniques on the Ethanol Gas Sensing Performance of ZnO Nanorod Gas Sensors. *ACS applied materials & interfaces* **2016**, 8, 2805–2811.
- [62] Giebelhaus, I.; Varechkina E.; Fischer T.; Rumyantseva M.; Ivanov V.; Gaskov A.; Morante J.R.; Arbiol J.; Tyrre W.; Mathur S. One-dimensional CuO-SnO<sub>2</sub> p-n heterojunctions for enhanced detection of H<sub>2</sub>S. *Journal of Materials Chemistry A* **2013**, 1, 11261.
- [63] Wang, H.; Chen M.; Rong Q.; Zhang Y.; Hu J.; Zhang D.; Zhou S.; Zhao X.; Zhang J.; Zhu Z.; Liu Q. Ultrasensitive xylene gas sensor based on flower-like SnO<sub>2</sub>/Co<sub>3</sub>O<sub>4</sub> nanorods composites prepared by facile two-step synthesis method. *Nanotechnology* **2020**, 31, 255501.
- [64] Wang, Z.; Li Z.; Sun J.; Zhang H.; Wang W.; Zheng W.; Wang C. Improved Hydrogen Monitoring Properties Based on p-NiO/n-SnO<sub>2</sub> Heterojunction Composite Nanofibers. *The Journal of Physical Chemistry C* **2010**, 114, 6100–6105.
- [65] Ju, D.; Xu H.; Xu Q.; Gong H.; Qiu Z.; Guo J.; Zhang J.; Cao B. High triethylamine-sensing properties of NiO/SnO<sub>2</sub> hollow sphere P-N heterojunction sensors. *Sensors and Actuators B: Chemical* **2015**, 215, 39–44.
- [66] Xue, X.; Xing L.; Chen Y.; Shi S.; Wang Y.; Wang T. Synthesis and H<sub>2</sub>S Sensing Properties of CuO-SnO<sub>2</sub> Core/Shell PN-Junction Nanorods. *The Journal of Physical Chemistry C* **2008**, 112, 12157–12160.
- [67] Meng, D.; Liu D.; Wang G.; Shen Y.; San X.; Li M.; Meng F. Low-temperature formaldehyde gas sensors based on NiO-SnO<sub>2</sub> heterojunction microflowers assembled by thin porous nanosheets. *Sensors and Actuators B: Chemical* **2018**, 273, 418–428.
- [68] Bang, J.H.; Choi M.S.; Mirzaei A.; Kwon Y.J.; Kim S.S.; Kim T.W.; Kim H.W. Selective NO<sub>2</sub> sensor based on Bi<sub>2</sub>O<sub>3</sub> branched SnO<sub>2</sub> nanowires. *Sensors and Actuators B: Chemical* **2018**, 274, 356–369.
- [69] Meng, D.; Si J.; Wang M.; Wang G.; Shen Y.; San X.; Meng F. One-step synthesis and the enhanced trimethylamine sensing properties of Co<sub>3</sub>O<sub>4</sub>/SnO<sub>2</sub> flower-like structures. *Vacuum* **2020**, 171, 108994.
- [70] Chen, Y.-J.; Xiao G.; Wang T.-S.; Zhang F.; Ma Y.; Gao P.; Zhu C.-L.; Zhang E.; Xu Z.; Li Q.-H. Synthesis and enhanced gas sensing properties of crystalline CeO<sub>2</sub>/TiO<sub>2</sub> core/shell nanorods. *Sensors and Actuators B: Chemical* **2011**, 156, 867–874.
- [71] Alev, O.; Kilic A.; Cakirlar C.; Buyukkose S.; Ozturk Z.Z. Gas Sensing Properties of p-Co<sub>3</sub>O<sub>4</sub>/n-TiO<sub>2</sub> Nanotube Heterostructures. *Sensors* **2018**, 18,
- [72] Yin, L.; Qu G.; Guo P.; Zhang R.; Sun J.; Chen D. Construction and enhanced low-temperature H<sub>2</sub>S-sensing performance of novel hierarchical CuO@WO<sub>3</sub> nanocomposites. *Journal of Alloys and Compounds* **2019**, 785, 367–373.
- [73] Long, H.; Zeng W.; Li T. Hierarchically solvothermal synthesis of WO<sub>3</sub>-based nanocomposite: Nature-inspired structure and enhanced gas-sensing property. *Physica E: Low-dimensional Systems and Nanostructures* **2017**, 88, 206–211.
- [74] Ramgir, N.S.; Goyal C.P.; Sharma P.K.; Goutam U.K.; Bhattacharya S.; Datta N.; Kaur M.; Debnath A.K.; Aswal D.K.; Gupta S.K. Selective H<sub>2</sub>S sensing characteristics of CuO modified WO<sub>3</sub> thin films. *Sensors and Actuators B: Chemical* **2013**, 188, 525–532.
- [75] Park, S.; Park S.; Jung J.; Hong T.; Lee S.; Kim H.W.; Lee C. H<sub>2</sub>S gas sensing properties of CuO-functionalized WO<sub>3</sub> nanowires. *Ceramics International* **2014**, 40, 11051–11056.
- [76] Choi, S.; Bonyani M.; Sun G.-J.; Lee J.K.; Hyun S.K.; Lee C. Cr<sub>2</sub>O<sub>3</sub> nanoparticle-functionalized WO<sub>3</sub> nanorods for ethanol gas sensors. *Applied Surface Science* **2018**, 432, 241–249.

- [77] He, M.; Xie L.; Zhao X.; Hu X.; Li S.; Zhu Z.-G. Highly sensitive and selective H<sub>2</sub>S gas sensors based on flower-like WO<sub>3</sub>/CuO composites operating at low/room temperature. *Journal of Alloys and Compounds* **2019**, 788, 36–43.
- [78] Wang, T.-S.; Wang Q.-S.; Zhu C.-L.; Ouyang Q.-Y.; Qi L.-H.; Li C.-Y.; Xiao G.; Gao P.; Chen Y.-J. Synthesis and enhanced H<sub>2</sub>S gas sensing properties of  $\alpha$ -MoO<sub>3</sub>/CuO p-n junction nanocomposite. *Sensors and Actuators B: Chemical* **2012**, 171–172, 256–262.
- [79] Annanouch, F.E.; Roso S.; Haddi Z.; Vallejos S.; Umek P.; Bittencourt C.; Blackman C.; Vilic T.; Llobet E. p-Type PdO nanoparticles supported on n-type WO<sub>3</sub> nanoneedles for hydrogen sensing. *Thin Solid Films* **2016**, 618, 238–245.
- [80] Mirzaei, A.; Park S.; Sun G.-J.; Kheel H.; Lee C.; Lee S. Fe<sub>2</sub>O<sub>3</sub>/Co<sub>3</sub>O<sub>4</sub> composite nanoparticle ethanol sensor. *J Korean Phys Soc* **2016**, 69, 373–380.
- [81] Huang, X.-Y.; Chi Z.-T.; Liu J.; Li D.-H.; Sun X.-J.; Yan C.; Wang Y.-C.; Li H.; Wang X.-D.; Xie W.-F. Enhanced gas sensing performance based on p-NiS/n-In<sub>2</sub>O<sub>3</sub> heterojunction nanocomposites. *Sensors and Actuators B: Chemical* **2020**, 304, 127305.
- [82] Park, S.; Kim S.; Sun G.-J.; Lee C. Synthesis, Structure, and Ethanol Gas Sensing Properties of In<sub>2</sub>O<sub>3</sub> Nanorods Decorated with Bi<sub>2</sub>O<sub>3</sub> Nanoparticles. *ACS applied materials & interfaces* **2015**, 7, 8138–8146.
- [83] Yan, C.; Lu H.; Gao J.; Zhu G.; Yin F.; Yang Z.; Liu Q.; Li G. Synthesis of porous NiO-In<sub>2</sub>O<sub>3</sub> composite nanofibers by electrospinning and their highly enhanced gas sensing properties. *Journal of Alloys and Compounds* **2017**, 699, 567–574.
- [84] Kim, J.-H.; Lee J.-H.; Mirzaei A.; Kim H.W.; Kim S.S. Optimization and gas sensing mechanism of n-SnO<sub>2</sub>-p-Co<sub>3</sub>O<sub>4</sub> composite nanofibers. *Sensors and Actuators B: Chemical* **2017**, 248, 500–511.
- [85] Park, S.; Kim S.; Kheel H.; Lee C. Oxidizing gas sensing properties of the n-ZnO/p-Co<sub>3</sub>O<sub>4</sub> composite nanoparticle network sensor. *Sensors and Actuators B: Chemical* **2016**, 222, 1193–1200.
- [86] Huang, J.; Dai Y.; Gu C.; Sun Y.; Liu J. Preparation of porous flower-like CuO/ZnO nanostructures and analysis of their gas-sensing property. *Journal of Alloys and Compounds* **2013**, 575, 115–122.
- [87] Kim, J.; Kim W.; Yong K. CuO/ZnO Heterostructured Nanorods: Photochemical Synthesis and the Mechanism of H<sub>2</sub>S Gas Sensing. *The Journal of Physical Chemistry C* **2012**, 116, 15682–15691.
- [88] Patil, L.A.; Patil D.R. Heterocontact type CuO-modified SnO<sub>2</sub> sensor for the detection of a ppm level H<sub>2</sub>S gas at room temperature. *Sensors and Actuators B: Chemical* **2006**, 120, 316–323.
- [89] Shao, F.; Hoffmann M.W.G.; Prades J.D.; Zamani R.; Arbiol J.; Morante J.R.; Varechkina E.; Rumyantseva M.; Gaskov A.; Giebelhaus I.; Fischer T.; Mathur S.; Hernández-Ramírez F. Heterostructured p-CuO (nanoparticle)/n-SnO<sub>2</sub> (nanowire) devices for selective H<sub>2</sub>S detection. *Sensors and Actuators B: Chemical* **2013**, 181, 130–135.
- [90] Wang, L.; Kang Y.; Wang Y.; Zhu B.; Zhang S.; Huang W.; Wang S. CuO nanoparticle decorated ZnO nanorod sensor for low-temperature H<sub>2</sub>S detection. *Materials Science and Engineering: C* **2012**, 32, 2079–2085.
- [91] Yu, M.-R.; Suyambrakasam G.; Wu R.-J.; Chavali M. Performance evaluation of ZnO–CuO hetero junction solid state room temperature ethanol sensor. *Mater Res Bull* **2012**, 47, 1713–1718.
- [92] Zhang, Y.-B.; Yin J.; Li L.; Zhang L.-X.; Bie L.-J. Enhanced ethanol gas-sensing properties of flower-like p-CuO/n-ZnO heterojunction nanorods. *Sensors and Actuators B: Chemical* **2014**, 202, 500–507.
- [93] Liu, T.; Yu Z.; Liu Y.; Gao J.; Wang X.; Suo H.; Yang X.; Zhao C.; Liu F. Gas sensor based on Ni foam: SnO<sub>2</sub>-decorated NiO for Toluene detection. *Sensors and Actuators B: Chemical* **2020**, 318, 128167.
- [94] Dey, S.; Nag S.; Santra S.; Ray S.K.; Guha P.K. Voltage-controlled NiO/ZnO p-n heterojunction diode: a new approach towards selective VOC sensing. *Microsystems & Nanoengineering* **2020**, 6, 35.
- [95] Wang, Z.; Zhang K.; Fei T.; Gu F.; Han D.  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/NiO heterojunction nanorods with enhanced gas sensing performance for acetone. *Sensors and Actuators B: Chemical* **2020**, 318, 128191.
- [96] Wang, C.; Cheng X.; Zhou X.; Sun P.; Hu X.; Shimanoe K.; Lu G.; Yamazoe N. Hierarchical  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/NiO composites with a hollow structure for a gas sensor. *ACS applied materials & interfaces* **2014**, 6, 12031–12037.
- [97] Park, S.; Kim S.; Sun G.-J.; Choi S.; Lee S.; Lee C. Ethanol sensing properties of networked In<sub>2</sub>O<sub>3</sub> nanorods decorated with Cr<sub>2</sub>O<sub>3</sub>-nanoparticles. *Ceram Int* **2015**, 41, 9823–9827.
- [98] Park, S.; Sun G.-J.; Kheel H.; Lee W.I.; Lee S.; Choi S.-B.; Lee C. Synergistic effects of codecoration of oxide nanoparticles on the gas sensing performance of In<sub>2</sub>O<sub>3</sub> nanorods. *Sensors and Actuators B: Chemical* **2016**, 227, 591–599.
- [99] Bekermann, D.; Gasparotto A.; Barreca D.; Maccato C.; Comini E.; Sada C.; Sberveglieri G.; Devi A.; Fischer R.A. Co<sub>3</sub>O<sub>4</sub>/ZnO nanocomposites: from plasma synthesis to gas sensing applications. *ACS applied materials & interfaces* **2012**, 4, 928–934.
- [100] Cao, J.; Wang Z.; Wang R.; Liu S.; Fei T.; Wang L.; Zhang T. Core-shell Co<sub>3</sub>O<sub>4</sub>/ $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> heterostructure nanofibers with enhanced gas sensing properties. *RSC Advances* **2015**, 5, 36340–36346.
- [101] Jeong, H.M.; Kim J.H.; Jeong S.Y.; Kwak C.H.; Lee J.H. Co<sub>3</sub>O<sub>4</sub>-SnO<sub>2</sub> Hollow Heteronanostructures: Facile Control of Gas Selectivity by Compositional Tuning of Sensing Materials via Galvanic Replacement. *ACS applied materials & interfaces* **2016**, 8, 7877–7883.
- [102] Qu, F.; Liu J.; Wang Y.; Wen S.; Chen Y.; Li X.; Ruan S. Hierarchical Fe<sub>3</sub>O<sub>4</sub>@Co<sub>3</sub>O<sub>4</sub> core-shell microspheres: Preparation and acetone sensing properties. *Sensors and Actuators B: Chemical* **2014**, 199, 346–353.
- [103] Staerz, A.; Kim T.-H.; Lee J.-H.; Weimar U.; Barsan N. Nanolevel Control of Gas Sensing Characteristics via p-n Heterojunction between Rh<sub>2</sub>O<sub>3</sub> Clusters and WO<sub>3</sub> Crystallites. *The Journal of Physical Chemistry C* **2017**, 121, 24701–24706.
- [104] Wang, L.; Lou Z.; Zhang R.; Zhou T.; Deng J.; Zhang T. Hybrid Co<sub>3</sub>O<sub>4</sub>/SnO<sub>2</sub> Core-Shell Nanospheres as Real-Time Rapid-Response Sensors for Ammonia Gas. *ACS applied materials & interfaces* **2016**, 8, 6539–6545.
- [105] Xiong, Y.; Xu W.; Zhu Z.; Xue Q.; Lu W.; Ding D.; Zhu L. ZIF-derived porous ZnO-Co<sub>3</sub>O<sub>4</sub> hollow polyhedrons heterostructure with highly enhanced ethanol detection performance. *Sensors and Actuators B: Chemical* **2017**, 253, 523–532.
- [106] Zhang, L.; Jing X.; Liu J.; Wang J.; Sun Y. Facile synthesis of mesoporous ZnO/Co<sub>3</sub>O<sub>4</sub> microspheres with enhanced gas-sensing for ethanol. *Sensors and Actuators B: Chemical* **2015**, 221, 1492–1498.
- [107] B., Z.; M. L.; Z. S.; H. K.; H. Y.; Q. L.; G. Z.; H. L. *Sens. Actuators, B* **2017**, 249, 558.

- [108] Chowdhuri, A.; Gupta V.; Sreenivas K. Fast response H<sub>2</sub>S gas sensing characteristics with ultra-thin CuO islands on sputtered SnO<sub>2</sub>. *Sensors and Actuators B: Chemical* **2003**, *93*, 572–579.
- [109] Mridha, S.; Basak D. Investigation of a p-CuO/n-ZnO thin film heterojunction for H<sub>2</sub>gas-sensor applications. *Semiconductor Science and Technology* **2006**, *21*, 928–932.
- [110] Yin, M.; Wang F.; Fan H.; Xu L.; Liu S. Heterojunction CuO@ZnO microcubes for superior p-type gas sensor application. *Journal of Alloys and Compounds* **2016**, *672*, 374–379.
- [111] Xue, X.-T.; Zhu L.-Y.; Yuan K.-P.; Zeng C.; Li X.-X.; Ma H.-P.; Lu H.-L.; Zhang D.W. ZnO branched p-Cu<sub>x</sub>O @n-ZnO heterojunction nanowires for improving acetone gas sensing performance. *Sensors and Actuators B: Chemical* **2020**, *324*, 128729.
- [112] Li, X.; Li X.; Chen N.; Li X.; Zhang J.; Yu J.; Wang J.; Tang Z. CuO-In<sub>2</sub>O<sub>3</sub>Core-Shell Nanowire Based Chemical Gas Sensors. *Journal of Nanomaterials* **2014**, *2014*, 1–7.
- [113] Park, S.; Kheel H.; Sun G.-J.; Ko T.; Lee W.I.; Lee C. Acetone Gas Sensing Properties of a Multiple-Networked Fe<sub>2</sub>O<sub>3</sub>-Functionalized CuO Nanorod Sensor. *Journal of Nanomaterials* **2015**, *2015*, 1–6.
- [114] Park, S.; Kim S.; Sun G.-J.; In Lee W.; Kim K.K.; Lee C. Fabrication and NO<sub>2</sub> gas sensing performance of TeO<sub>2</sub>-core/CuO-shell heterostructure nanorod sensors. *Nanoscale Res Lett* **2014**, *9*, 638.
- [115] Wang, Y.; Qu F.; Liu J.; Wang Y.; Zhou J.; Ruan S. Enhanced H<sub>2</sub>S sensing characteristics of CuO-NiO core-shell microspheres sensors. *Sensors and Actuators B: Chemical* **2015**, *209*, 515–523.
- [116] Meng, F.-N.; Di X.-P.; Dong H.-W.; Zhang Y.; Zhu C.-L.; Li C.; Chen Y.-J. Ppb H<sub>2</sub>S gas sensing characteristics of Cu<sub>2</sub>O/CuO sub-microspheres at low-temperature. *Sensors and Actuators B: Chemical* **2013**, *182*, 197–204.
- [117] Su, C.; Zhang L.; Han Y.; Ren C.; Zeng M.; Zhou Z.; Su Y.; Hu N.; Wei H.; Yang Z. Controllable synthesis of heterostructured CuO–NiO nanotubes and their synergistic effect for glycol gas sensing. *Sensors and Actuators B: Chemical* **2020**, *304*, 127347.
- [118] Xu, Y.; Fan Y.; Tian X.; Liang Q.; Liu X.; Sun Y. p-p heterojunction composite of NiFe<sub>2</sub>O<sub>4</sub> nanoparticles-decorated NiO nanosheets for acetone gas detection. *Materials Letters* **2020**, *270*, 127728.
- [119] Gao, H.; Guo J.; Li Y.; Xie C.; Li X.; Liu L.; Chen Y.; Sun P.; Liu F.; Yan X.; Liu F.; Lu G. Highly selective and sensitive xylene gas sensor fabricated from NiO/NiCr<sub>2</sub>O<sub>4</sub> p-p nanoparticles. *Sensors and Actuators B: Chemical* **2019**, *284*, 305–315.
- [120] Xu, H.; Zhang J.; Rehman A.U.; Gong L.; Kan K.; Li L.; Shi K. Synthesis of NiO@CuO nanocomposite as high-performance gas sensing material for NO<sub>2</sub> at room temperature. *Applied Surface Science* **2017**, *412*, 230–237.
- [121] Adamu, B.I.; Falak A.; Tian Y.; Tan X.; Meng X.; Chen P.; Wang H.; Chu W. p–p Heterojunction Sensors of p-Cu<sub>3</sub>Mo<sub>2</sub>O<sub>9</sub> Micro/Nanorods Vertically Grown on p-CuO Layers for Room-Temperature Ultrasensitive and Fast Recoverable Detection of NO<sub>2</sub>. *ACS Applied Materials & Interfaces* **2020**, *12*, 8411–8421.