

*Supporting Information*

# Bimetallic PdAu Catalysts within Hierarchically Porous Architectures for Aerobic Oxidation of Benzyl Alcohol

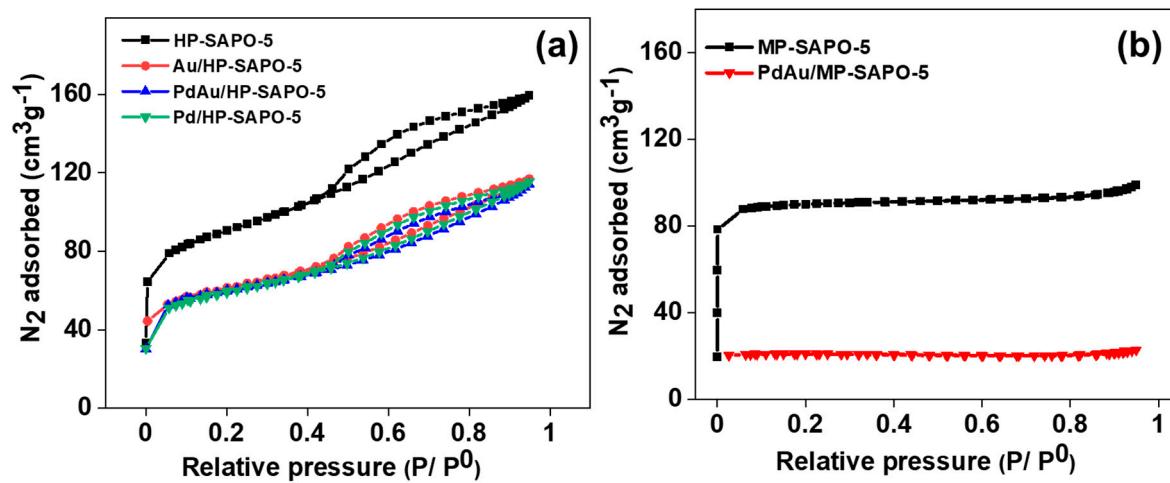
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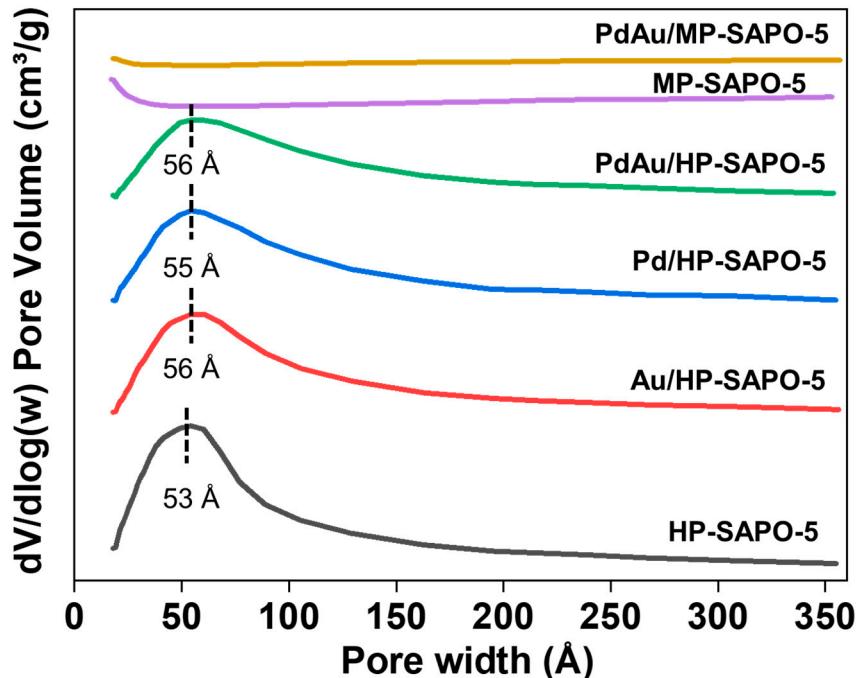
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**Table S1.** Benzyl alcohol oxidation to benzaldehyde using various catalysts.

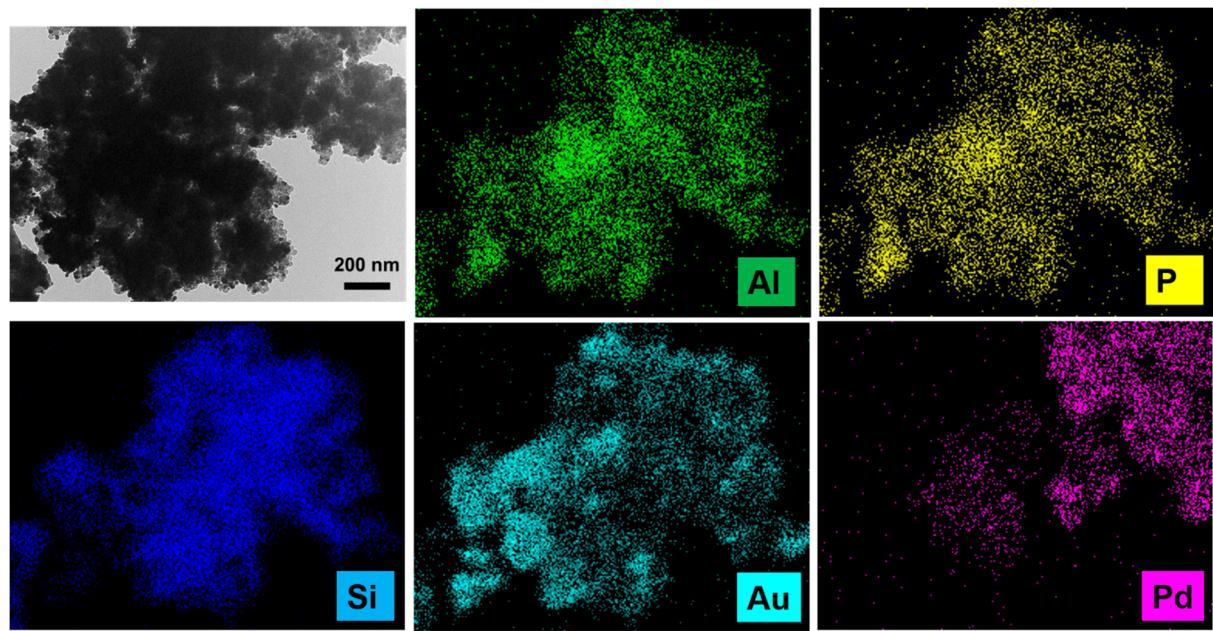
Catalyst	Type of metal	Type of support	NPs size	Reaction conditions	Catalytic activity (%)		Reference
					Conv	selectivity	
Au/Fe <sub>2</sub> O <sub>3</sub>	Au	Fe <sub>2</sub> O <sub>3</sub>	--	Toluene, 2 bar O <sub>2</sub> , 100 °C	7.1	87.6	[1]
Au/TiO <sub>2</sub>	Au	TiO <sub>2</sub>	--	Toluene, 2 bar O <sub>2</sub> , 100 °C	0.65	100	[1]
Au/TiO <sub>2</sub> -NR	Au	TiO <sub>2</sub> nanorod	6.2	K <sub>2</sub> CO <sub>3</sub> , Toluene, O <sub>2</sub> , 100 °C	30	900	[2]
Au/SiO <sub>2</sub>	Au	SiO <sub>2</sub>	--	Toluene, 2 bar O <sub>2</sub> , 100 °C	2.4	94.3	[1]
Au/C	Au	Carbon	--	Toluene, 2 bar O <sub>2</sub> , 100 °C	2.3	90.4	[1]
Au/CeO <sub>2</sub>	Au	CeO <sub>2</sub>	--	Toluene, 2 bar O <sub>2</sub> , 100 °C	3.4	100	[1]
Au/CeO <sub>2</sub> NR	Au	CeO <sub>2</sub> nanorod	3.6	Toluene, O <sub>2</sub> , 100 °C	89	94	[3]
Au/RGO	Au	Graphene oxide	5.4	H <sub>2</sub> O, NaHCO <sub>3</sub> , O <sub>2</sub> , 100 °C	65	93	[4]
Ru/C	Ru	Carbon	--	Toluene, O <sub>2</sub> , 50 °C	100	98	[5]
Pt/C	Pt	Carbon	3.5	Toluene, pO <sub>2</sub> = 150 psi, 120 °C	2.8	90.7	[6]
Pt/Carbon Hybrid	Pt	Carbon hybrid	2.83	KOH, Toluene, O <sub>2</sub> , 80 °C	99	99	[7]
PdPt/C	Pd, Pt	Carbon	2.2	Toluene, pO <sub>2</sub> = 150 psi, 120 °C	14.8	84.7	[6]
AuPd-PVP	Pd, Au	--	2.7	O <sub>2</sub> , 100 °C	14.8	90.5	[8]
PdAu/MSNs	Pd, Au	Silica	3.6	Toluene, O <sub>2</sub> 0.5 MPa, 90 °C	98	98	[9]



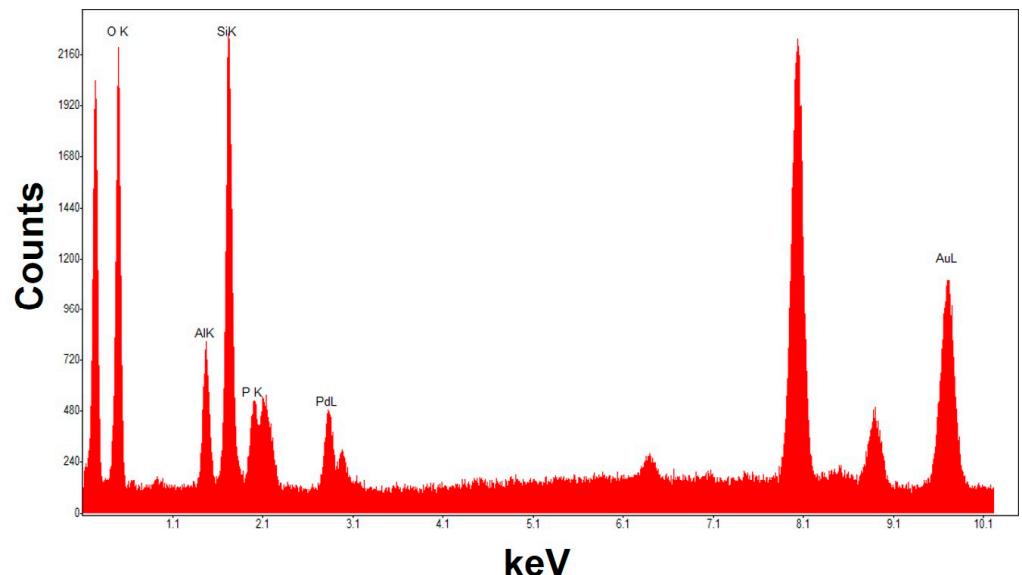
**Figure S1.** N<sub>2</sub> physisorption isotherms of prepared catalysts before and after NP deposition on (a) HP-SAPO-5 and (b) MP-SAPO-5.



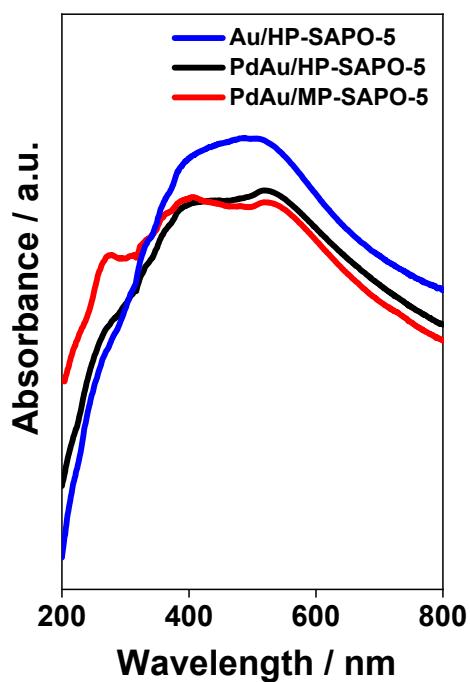
**Figure S2.** BJH pore size distribution for the hierarchically porous (HP) and microporous (MP) support systems before and after NP deposition. Plots are stacked for clarity.



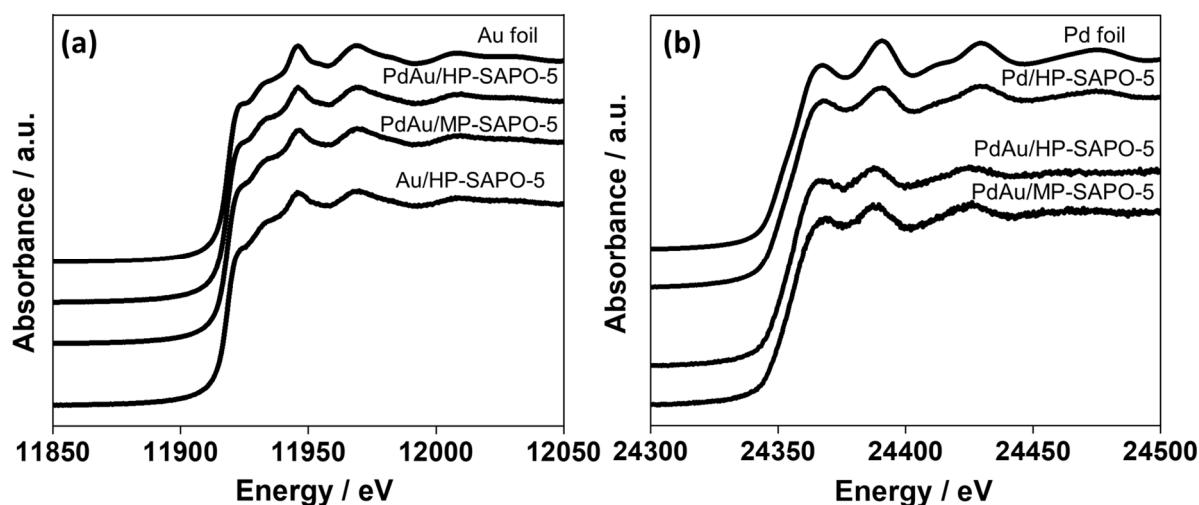
**Figure S3.** TEM micrograph of PdAu/HP-SAPO-5 along with the elemental mapping displaying the presence of Al, P, Si, Au and Pd.



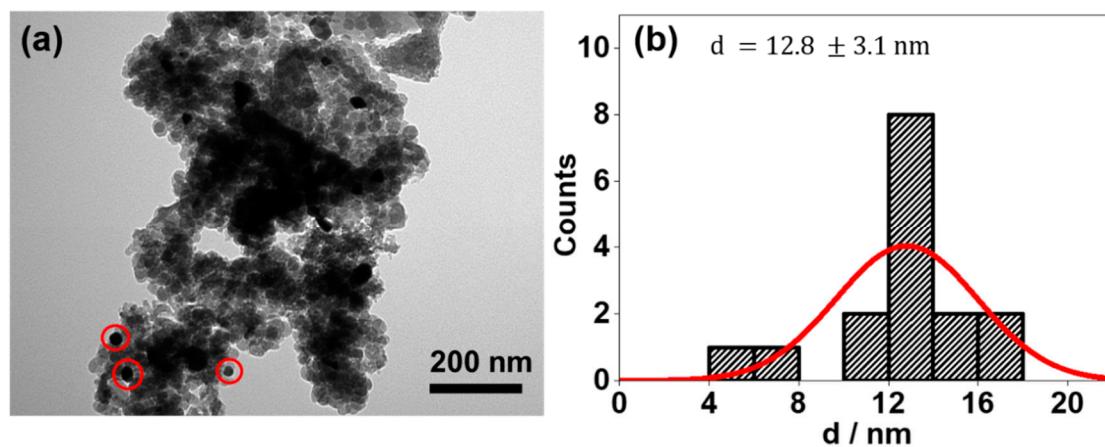
**Figure S4.** EDX spectrum of PdAu/HP-SAPO-5 displaying the presence of Al, P, Si, Au and Pd.



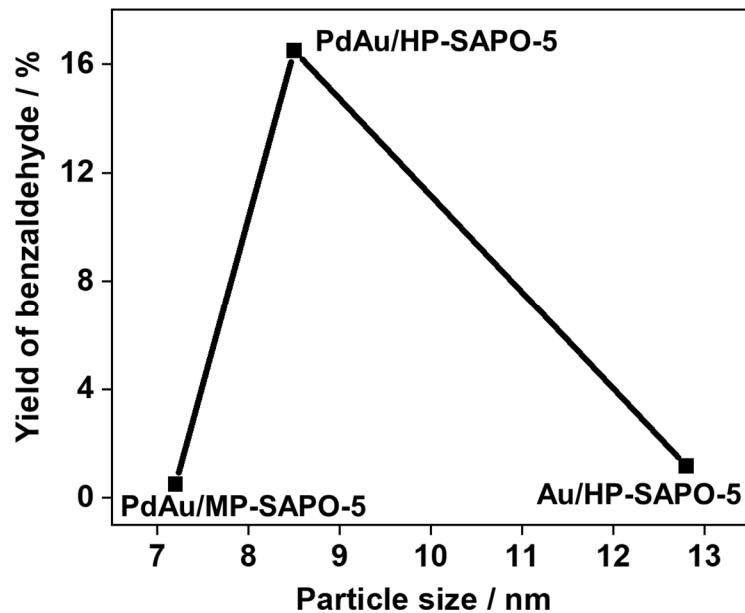
**Figure S5.** UV-vis spectra of PdAu bimetallic NPs on hierarchically porous and microporous SAPO-5.



**Figure S6.** The (a) Au L<sub>III</sub>-edge and (b) Pd K-edge X-ray absorption near edge structure (XANES) spectra of prepared catalysts.



**Figure S7.** (a) TEM image and (b) size distribution of Au/HP-SAPO-5.



**Figure S8.** Relationship between particle size and catalytic yield for monometallic and bimetallic catalysts.

**Table S2.** The obtained binding energy (B.E.) values in the Au 4f and Pd 3d XPS spectral analysis.

Sample	Au 4f B.E. (eV)		Pd 3d B.E. (eV)	
	4f <sub>7/2</sub>	4f <sub>5/2</sub>	3d <sub>5/2</sub>	3d <sub>3/2</sub>
Au/HP-SAPO-5	84.0	87.6	--	--
Pd/HP-SAPO-5	--	--	335.2	340.4
PdAu/HP-SAPO-5	83.6	87.3	335.4	340.6
PdAu/MP-SAPO-5	83.8	87.5	335.5	340.7

**References:**

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