



Supplementary Materials

Fe₃O₄@C Nanoparticles Synthesized by In Situ Solid-Phase Method for Removal of Methylene Blue

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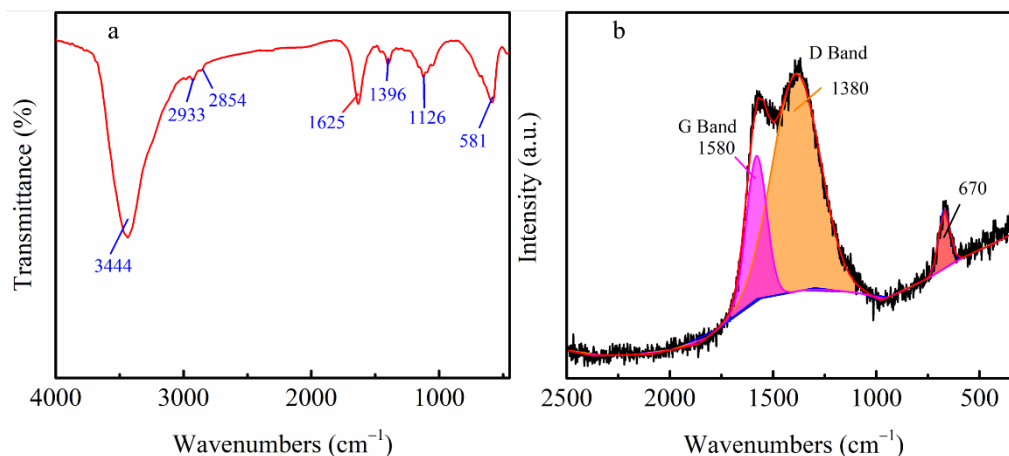


Figure S1. The (a) FT-IR spectroscopy and (b) Raman spectroscopy of the as-synthesized Fe₃O₄@C nanoparticles.

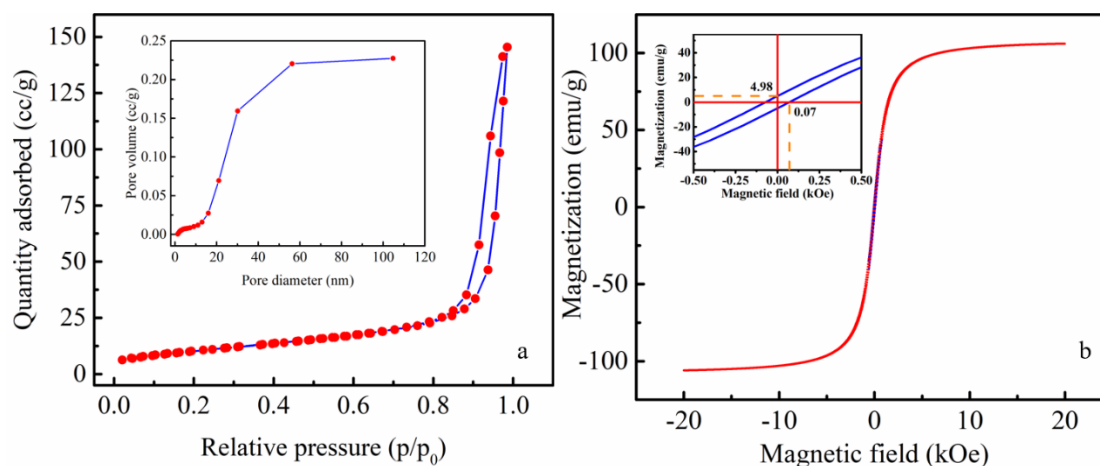


Figure S2. (a) The N₂ adsorption/desorption isotherm curves and (b) the magnetic property of the Fe₃O₄@C nanoparticles.

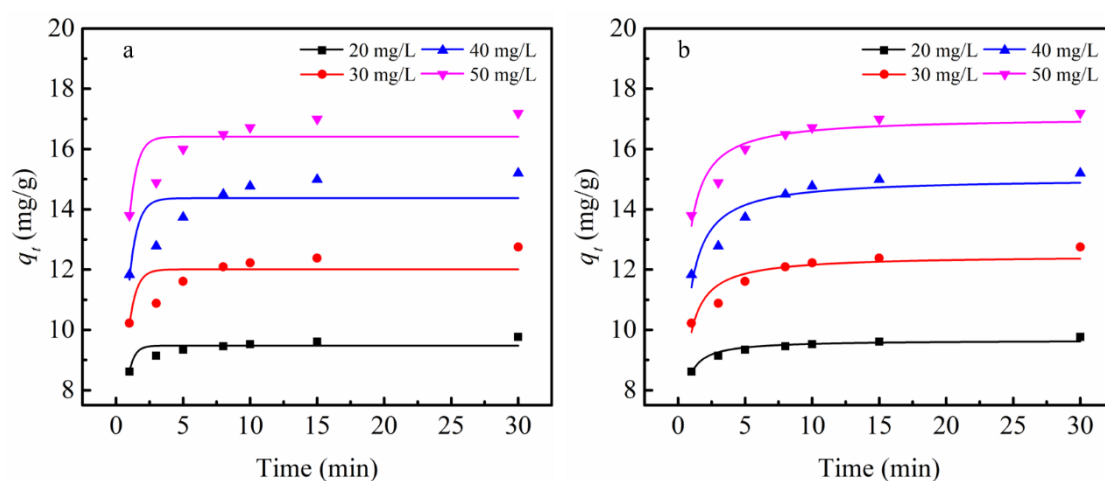


Figure S3. The non-linear forms of the kinetics model. (a) pseudo-first-order model, (b) pseudo-second-order model.

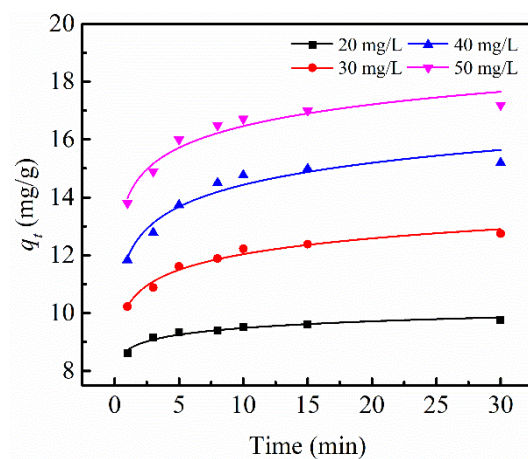


Figure S4. The Elovich kinetics model of the adsorption.

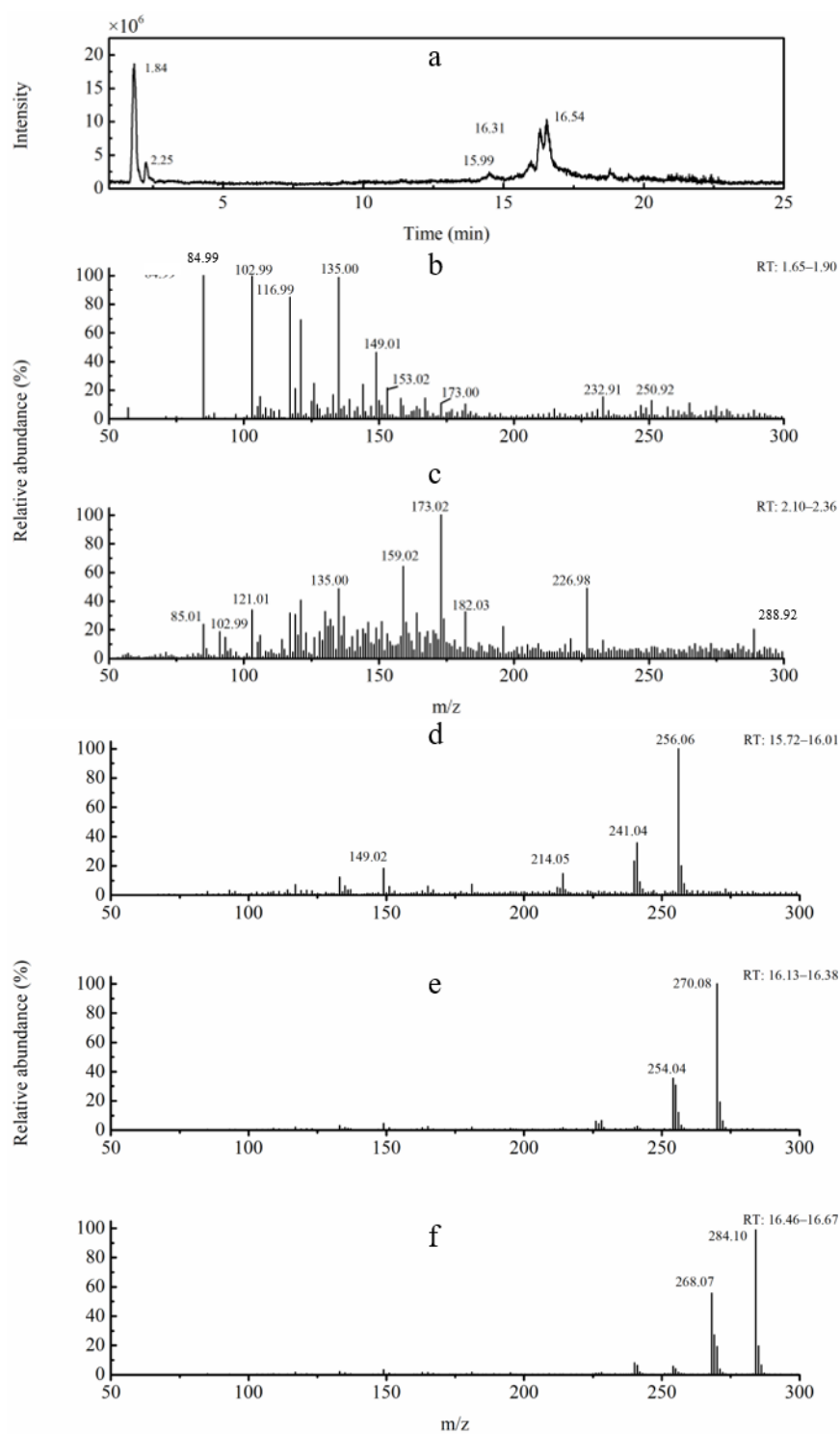


Figure S5. ESI mass spectra of different retention time at the reaction time of 1 h.

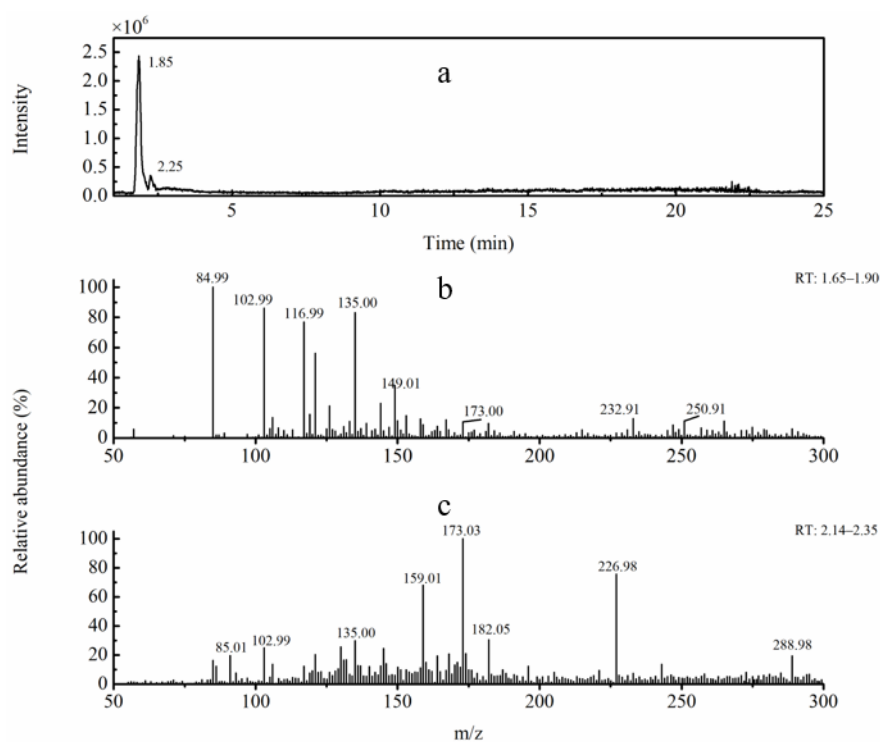


Figure S6. ESI mass spectra of different retention time at the reaction time of 3 h.

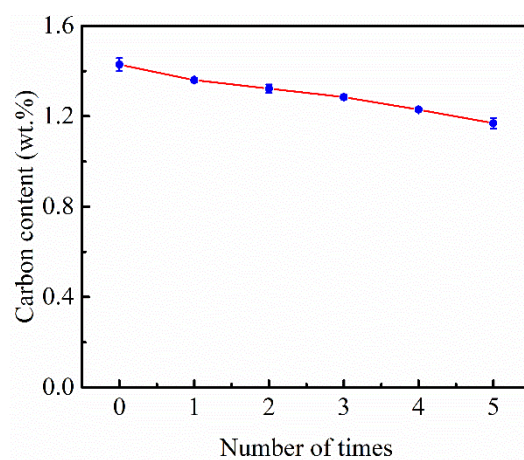


Figure S7. The carbon contents of the $\text{Fe}_3\text{O}_4@\text{C}$ nanoparticles after repeated use.

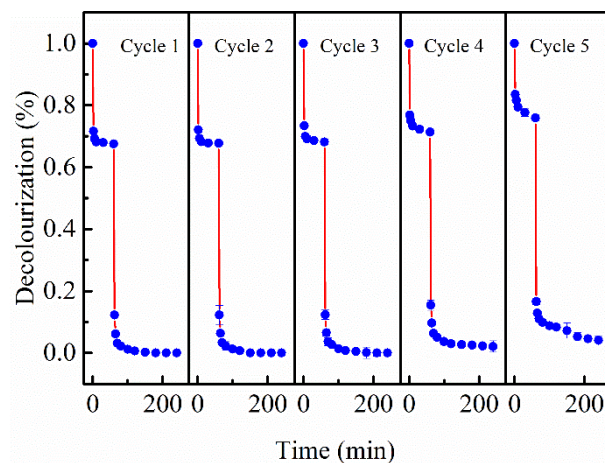


Figure S8. The recyclability test of the Fe₃O₄@C nanoparticles (Conditions: 100 mM MB, 30 mM H₂O₂, 2 g/L Fe₃O₄@C nanoparticles, the temperature of 40 °C, and initial pH value of 3.0).

Table S1. Kinetic parameters for adsorption of methyl blue on Fe₃O₄@C nanoparticles.

C_{ini} $\text{mg} \times \text{L}^{-1}$	$q_{e(expe)}$ mg/g	Pseudo-First-Order Kinetics			Pseudo-Second-Order Kinetics			
		k_1 min^{-1}	$q_{e(calc)}$ mg/g	R_1^2	k_2 $\text{g/mg} \times \text{min}$	$q_{e(calc)}$ mg/g	R_2^2	h $\text{mg/mg} \times \text{min}$
20	9.809	2.384	9.477	0.6822	0.5706	9.659	0.9326	53.23
30	12.84	1.863	12.01	0.4845	0.1651	12.95	0.8290	27.69
40	15.29	1.792	14.37	0.5084	0.1117	15.32	0.8470	26.22
50	17.26	1.672	16.41	0.5679	0.0918	17.38	0.8832	27.73

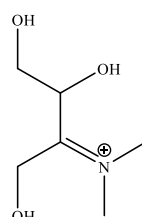
Table S2. The Elovich kinetic parameters for adsorption of methyl blue on Fe₃O₄@C nanoparticles.

C_{ini} $\text{mg} \times \text{L}^{-1}$	Elovich Model		
	α $\text{mg}/(\text{g} \times \text{min})$	β g/mg	R_1^2
20	6.928×10^{10}	2.984	0.9577
30	3.339×10^5	1.269	0.9706
40	5.569×10^4	0.924	0.9325
50	4.380×10^4	0.910	0.9296

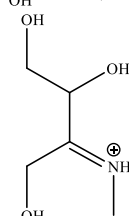
Table S3. The possible intermediate degradation products of MB.

No.	m/z	Structural Formula
1	289	
2	284	
3	270	
4	256	
5	227	
6	173	
7	159	

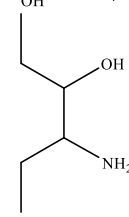
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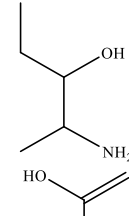
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