

Supplementary Materials: A Test of Various Partial Atomic Charge Models for Computations on Diheteroaryl Ketones and Thioketones

Piotr Matczak

Supplementary Material contains three sections: (S1) additional tables with the results of dipole moment calculations; (S2) selection of a quantum chemical method for calculating the reference full-density dipole moments of **1a–5a** and **1b–5b** from Figure 1; (S3) additional tables with the results of E_{elst} calculations.

S1. Additional Tables with the Results of Dipole Moment Calculations

Table S1. Dipole moments (μ in Debyes) calculated using the HF/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1a–5a** from Figure 1.

Conformer	Full Density	μ									
		Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	5.24	5.75	8.54	6.28	19.63	3.74	5.29	5.26	4.78	5.22	5.24
<i>ct-1a</i>	4.26	3.81	6.07	4.57	11.73	3.27	4.13	4.28	4.10	4.29	4.27
<i>tt-1a</i>	2.99	1.55	2.78	2.38	0.79	2.68	2.65	3.04	3.00	3.03	3.01
<i>cc-2a</i>	4.75	3.81	2.11	2.10	6.68	2.60	3.93	4.75	4.37	4.65	4.74
<i>ct-2a</i>	3.75	3.43	4.92	4.91	7.78	2.86	3.65	3.79	3.58	3.75	3.76
<i>tt-2a</i>	2.63	3.06	7.33	9.19	8.91	3.17	3.39	2.64	2.55	2.71	2.65
<i>cc-3a</i>	4.61	3.92	0.93	4.98	3.33	2.95	4.28	4.69	3.97	4.41	4.61
<i>ct-3a</i>	3.64	3.35	5.09	6.27	7.94	2.77	3.38	3.67	3.27	3.63	3.64
<i>tt-3a</i>	2.56	2.75	8.33	11.80	11.78	2.63	2.45	3.67	2.70	2.70	2.55
<i>cc-4a</i>	0.60	1.70	2.56	1.92	12.75	0.67	1.23	0.68	0.54	0.65	0.62
<i>ct-4a</i>	3.51	3.00	4.80	3.88	9.29	2.85	3.37	3.56	3.45	3.53	3.53
<i>tt-4a</i>	5.53	3.85	6.55	5.40	5.63	4.48	4.99	5.57	5.41	5.56	5.57
<i>cc-5a</i>	0.12	0.34	1.71	0.11	9.29	0.15	0.41	0.19	0.05	0.20	0.15
<i>ct-5a</i>	3.70	3.47	4.93	4.54	8.12	3.08	3.53	3.69	3.60	3.66	3.70
<i>tt-5a</i>	5.48	5.19	6.97	6.64	7.64	4.66	5.27	5.50	5.48	5.48	5.50

Table S2. Dipole moments (μ in Debyes) calculated using the HF/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1b–5b** from Figure 1.

Conformer	Full Density	μ									
		Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	5.21	5.38	5.82	1.70	9.55	3.68	4.86	5.30	4.69	5.26	5.25
<i>ct-1b</i>	4.58	3.96	3.92	1.31	5.58	3.44	3.98	4.67	4.64	4.75	4.63
<i>tt-1b</i>	3.63	2.41	0.79	1.17	7.44	3.08	2.79	3.77	3.83	3.83	3.71
<i>cc-2b</i>	4.92	3.76	0.21	6.02	2.80	2.72	3.73	5.05	4.34	4.95	4.97
<i>ct-2b</i>	4.11	3.64	2.90	4.04	1.87	3.05	3.54	4.26	4.00	4.22	4.17
<i>tt-2b</i>	3.18	3.51	4.70	4.83	0.61	3.38	3.36	3.29	3.03	3.38	3.24
<i>cc-3b</i>	4.80	3.89	1.32	8.70	5.86	2.89	3.85	4.97	4.23	4.80	4.86
<i>ct-3b</i>	3.98	3.54	3.39	6.15	3.68	2.85	3.19	4.07	3.90	4.09	4.03
<i>tt-3b</i>	3.09	3.21	5.63	7.24	2.01	2.86	2.42	3.12	3.65	3.34	3.12
<i>cc-4b</i>	1.94	3.04	1.36	0.92	4.57	1.68	1.90	2.04	1.62	2.06	1.98
<i>ct-4b</i>	4.33	3.77	3.19	1.16	2.27	3.47	3.68	4.45	4.27	4.45	4.39
<i>tt-4b</i>	6.01	4.19	4.27	1.23	3.18	4.81	4.98	6.15	5.65	6.15	6.09
<i>cc-5b</i>	1.61	1.52	0.79	2.49	1.20	1.02	1.34	1.72	1.61	1.75	1.66
<i>ct-5b</i>	4.71	4.43	3.98	3.52	0.33	3.80	4.10	4.77	4.48	4.73	4.73
<i>tt-5b</i>	5.84	5.12	4.45	2.66	1.39	4.85	5.04	5.94	5.81	5.94	5.87

Table S3. Dipole moments (μ in Debyes) calculated using the B3LYP/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1a–5a** from Figure 1.

Conformer	μ										
	Full Density ^a	Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	4.73	4.77	7.86	5.68	16.06	3.17	4.72	4.74	4.24	4.69	4.72
<i>ct-1a</i>	3.91	3.12	5.62	4.20	9.86	2.90	3.75	3.93	3.69	3.92	3.91
<i>tt-1a</i>	2.91	1.57	2.75	2.37	1.50	2.64	2.61	2.94	1.57	1.57	2.92
<i>cc-2a</i>	4.03	2.89	1.54	2.59	5.52	1.97	3.30	4.03	3.62	2.89	4.01
<i>ct-2a</i>	3.37	2.87	4.62	4.80	6.86	2.52	3.28	3.38	3.20	3.35	3.37
<i>tt-2a</i>	2.69	2.95	7.15	9.02	8.19	3.11	3.32	2.71	2.57	2.74	2.71
<i>cc-3a</i>	3.65	2.31	0.34	5.54	2.92	2.20	3.53	3.66	2.99	3.47	3.65
<i>ct-3a</i>	3.21	2.77	4.86	6.29	6.94	2.41	2.97	3.22	2.94	3.19	3.21
<i>tt-3a</i>	2.81	3.23	8.17	11.67	10.26	2.67	2.49	2.81	2.92	2.91	2.80
<i>cc-4a</i>	0.41	1.81	2.15	1.63	9.35	0.39	0.95	0.43	0.31	0.44	0.42
<i>ct-4a</i>	3.31	2.70	4.54	3.66	7.69	2.63	3.15	3.35	3.22	3.32	3.33
<i>tt-4a</i>	5.31	3.31	6.36	5.21	6.07	4.26	4.77	5.35	5.14	5.31	5.33
<i>cc-5a</i>	0.13	0.11	1.28	0.44	6.89	0.43	0.13	0.09	0.11	0.06	0.10
<i>ct-5a</i>	3.55	3.11	4.73	4.44	6.94	2.96	3.38	3.55	3.42	3.48	3.54
<i>tt-5a</i>	5.29	4.76	6.78	6.50	7.46	4.49	5.10	5.33	5.23	5.26	5.30

^a Values taken from [1].

Table S4. Dipole moments (μ in Debyes) calculated using the B3LYP/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1b–5b** from Figure 1.

Conformer	μ										
	Full Density ^a	Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	4.69	4.57	5.25	1.19	8.25	3.13	4.31	4.75	4.57	4.70	4.72
<i>ct-1b</i>	4.09	3.30	3.40	1.00	4.56	2.96	3.48	4.16	4.03	4.21	4.13
<i>tt-1b</i>	3.35	2.42	0.68	1.28	4.84	2.85	2.57	3.45	3.43	3.49	3.41
<i>cc-2b</i>	4.13	2.91	0.76	6.51	1.70	2.03	3.04	4.21	3.49	4.13	4.16
<i>ct-2b</i>	3.62	3.17	2.76	4.27	1.08	2.63	3.09	3.70	3.46	3.68	3.66
<i>tt-2b</i>	3.08	3.48	4.50	4.62	0.87	3.19	3.16	3.16	2.84	3.21	3.12
<i>cc-3b</i>	3.80	2.41	1.86	9.25	4.01	2.09	3.06	3.90	3.21	3.78	3.85
<i>ct-3b</i>	3.46	3.05	3.36	6.42	2.82	2.39	2.68	3.53	3.38	3.53	3.49
<i>tt-3b</i>	3.17	3.72	5.46	7.08	2.82	2.77	2.34	3.22	3.61	3.35	3.19
<i>cc-4b</i>	1.54	3.13	0.86	1.32	3.55	1.22	1.45	1.60	3.13	1.62	1.57
<i>ct-4b</i>	3.96	3.48	2.92	1.11	1.64	3.11	3.31	4.04	3.84	4.04	4.00
<i>tt-4b</i>	5.66	3.76	4.08	1.43	0.69	4.47	4.65	5.75	5.20	5.74	5.71
<i>cc-5b</i>	1.23	1.00	0.33	2.86	1.13	0.62	0.94	1.30	1.17	1.32	1.27
<i>ct-5b</i>	4.44	4.08	3.90	3.59	1.12	3.60	3.90	4.49	4.16	4.43	4.46
<i>tt-5b</i>	5.62	4.84	4.38	2.57	0.69	4.66	4.86	5.69	5.47	5.66	5.63

^a Values taken from [1].

Table S5. Dipole moments (μ in Debyes) calculated using the MP2/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1a–5a** from Figure 1.

Conformer	μ										
	Full Density	Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	4.39	4.03	7.25	5.75	16.53	2.80	4.35	4.40	3.92	4.36	4.39
<i>ct-1a</i>	3.60	2.81	5.12	6.19	10.02	2.58	3.43	3.61	3.39	3.62	3.60
<i>tt-1a</i>	2.65	1.60	2.43	11.43	1.14	2.40	2.37	2.68	2.62	2.67	2.66
<i>cc-2a</i>	3.71	2.49	0.87	2.95	5.80	1.66	2.99	3.72	3.31	3.63	3.69
<i>ct-2a</i>	3.10	2.63	4.29	4.71	6.91	2.26	3.01	3.12	2.95	3.09	3.10
<i>tt-2a</i>	2.50	2.75	6.98	8.88	8.02	2.88	3.10	2.49	2.38	2.53	2.51
<i>cc-3a</i>	3.40	2.55	0.21	5.37	3.56	2.07	3.40	3.49	2.77	3.26	3.41
<i>ct-3a</i>	2.96	2.55	4.57	3.90	6.91	2.15	2.73	2.98	2.72	2.95	2.96
<i>tt-3a</i>	2.57	2.50	7.92	2.11	9.79	2.28	2.10	2.51	2.67	2.65	2.56
<i>cc-4a</i>	0.07	0.70	1.62	1.31	9.75	0.05	0.61	0.14	0.02	0.11	0.08
<i>ct-4a</i>	3.11	2.43	4.17	3.44	7.81	2.42	2.92	3.15	3.03	3.11	3.13
<i>tt-4a</i>	5.08	3.51	6.02	5.01	5.91	4.03	4.55	5.10	4.91	5.08	5.10
<i>cc-5a</i>	0.57	0.71	0.72	0.86	6.98	0.85	0.29	0.51	0.74	0.50	0.54
<i>ct-5a</i>	3.45	3.26	4.47	4.38	6.98	2.87	3.27	3.42	3.31	3.38	3.44
<i>tt-5a</i>	5.11	4.72	6.47	6.34	7.47	4.30	4.90	5.13	5.05	5.08	5.12

Table S6. Dipole moments (μ in Debyes) calculated using the MP2/def2-QZVPP molecular wave functions/electron densities and the partial atomic charges derived from these wave functions/electron densities for the conformers of **1b–5b** from Figure 1.

Conformer	μ										
	Full Density	Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	4.17	3.71	4.36	9.73	8.04	2.61	3.78	4.23	3.61	4.19	4.19
<i>ct-1b</i>	3.54	2.78	2.62	6.57	4.55	2.44	2.96	3.61	3.49	3.66	3.57
<i>tt-1b</i>	2.83	2.11	0.03	6.56	5.88	2.40	2.11	2.94	2.91	2.97	2.88
<i>cc-2b</i>	3.62	2.34	1.73	7.10	2.07	1.56	2.57	3.74	3.02	3.64	3.66
<i>ct-2b</i>	3.14	2.68	2.53	4.53	1.35	2.19	2.63	3.25	3.01	3.21	3.17
<i>tt-2b</i>	2.65	2.99	3.99	4.21	0.02	2.75	2.73	2.72	2.41	2.76	2.68
<i>cc-3b</i>	3.34	2.43	2.75	0.67	4.06	1.76	2.73	3.50	2.82	3.35	3.40
<i>ct-3b</i>	2.98	2.57	3.20	1.10	2.66	1.93	2.24	3.04	2.94	3.06	3.02
<i>tt-3b</i>	2.68	2.74	4.88	1.81	1.67	2.18	1.74	2.69	3.11	2.84	2.69
<i>cc-4b</i>	1.09	1.90	0.13	1.79	3.36	0.80	1.03	1.18	0.75	1.18	1.12
<i>ct-4b</i>	3.61	3.07	2.46	1.28	1.33	2.79	2.98	3.69	3.51	3.68	3.65
<i>tt-4b</i>	5.22	3.71	3.45	1.00	1.51	4.08	4.26	5.33	4.77	5.30	5.27
<i>cc-5b</i>	0.67	0.40	0.49	3.46	0.64	0.08	0.40	0.75	0.61	0.75	0.71
<i>ct-5b</i>	4.23	4.15	3.70	2.65	0.88	3.43	3.71	4.26	3.96	4.20	4.24
<i>tt-5b</i>	5.24	4.64	3.76	2.19	0.05	4.30	4.49	5.32	5.09	5.28	5.25

S2. Selection of a Quantum Chemical Method for Calculating the Reference Full-Density Dipole Moments of **1a–5a** and **1b–5b**

Since the experimental dipole moments of the gas-phase conformers of **1a–5a** and **1b–5b** from Figure 1 are unknown (to be precise, for some of these compounds their dipole moments were measured experimentally for the mixtures of conformers in various solvents), the reference values of μ are obtained from the full molecular electron densities calculated at the HF/def2-QZVPP, B3LYP/def2-QZVPP and MP2/def2-QZVPP levels of theory. As a result, three sets of reference μ values are obtained and these sets are used for calculating the MSE and RMSE whose values are presented in Figure 3. If a single set of reference μ values is required, it is hard to ascertain whether HF/def2-QZVPP or B3LYP/def2-QZVPP or MP2/def2-QZVPP provides the most reliable full-density values of μ , because the lack of experimental dipole moments for the gas-phase conformers of **1a–5a** and **1b–5b** from Figure 1 makes it impossible. A reasonable solution to this problem is to consider a test set of some other molecules for which the dipole moments were measured experimentally in the gas phase. In our case such a test set is composed of seven simple molecules that are the building

blocks of **1a–5a** and **1b–5b**. The molecules of formaldehyde, thioformaldehyde, furan, thiophene, selenophene, pyrrole and 1-methylpyrrole are taken into account. Their μ values calculated from the HF/def2-QZVPP, B3LYP/def2-QZVPP and MP2/def2-QZVPP full densities, together with their μ values measured experimentally in the gas phase, are listed in Table S7. Additionally, the MSE and RMSE in the calculated μ values relative to the corresponding experimental data are shown in the last two rows in Table S7. These values of the MSE and RMSE clearly indicate that B3LYP/def2-QZVPP performs best in reproducing the experimental dipole moments of the seven molecules. It can be assumed that this level of theory also provides reliable values of μ for larger molecules possessing structural fragments taken from the seven molecules. Therefore, only the μ values obtained from the B3LYP/def2-QZVPP electron densities of the conformers of **1a–5a** and **1b–5b** from Figure 1 are used as the reference values to determine the MSE and RMSE in Figure 4.

Table S7. Calculated and experimental μ values for seven simple molecules. The last two rows give the MSE and RMSE of the calculated μ values from the experimental ones. All values in Debyes.

Molecule	μ			
	HF/def2-QZVPP	B3LYP/def2-QZVPP	MP2/def2-QZVPP	Experiment
Formaldehyde	2.80	2.36	2.32	2.33 [2]
Thioformaldehyde	2.12	1.72	1.63	1.65 [3]
Furan	0.78	0.66	0.64	0.66 [2]
Thiophene	0.74	0.50	0.48	0.55 [2]
Selenophene	0.76	0.37	0.40	0.39 [4]
Pyrrole	1.87	1.86	1.88	1.84 [2]
1-Methylpyrrole	2.16	2.14	2.24	2.12 [5]
MSE	0.24	0.01	0.01	0.00
RMSE	0.30	0.04	0.06	0.00

S3. Additional Tables with the Results of Eelst Calculations

Table S8. Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1a–5a** from Figure 1 in their three conformations. All results obtained at the HF/def2-QZVPP level of theory.

Conformer	ΔE	μ	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	2.01	5.24	13.94	12.91	12.52	51.43	3.86	5.75	0.00	0.00	0.00	0.00
<i>ct-1a</i>	0.00	4.26	4.32	0.39	0.86	1.36	2.03	2.18	8.15	15.64	14.23	47.75
<i>tt-1a</i>	0.88	2.99	0.00	0.00	0.00	0.00	0.00	0.00	28.80	14.44	31.94	60.08
<i>cc-2a</i>	0.00	4.75	3.97	0.00	0.00	0.00	0.00	1.06	0.00	111.04	0.00	0.00
<i>ct-2a</i>	0.91	3.75	2.88	3.87	10.47	7.29	0.22	0.75	66.47	0.00	34.60	134.34
<i>tt-2a</i>	2.48	2.63	0.00	7.14	26.76	13.43	0.39	0.00	96.69	161.25	59.15	146.93
<i>cc-3a</i>	0.00	4.61	3.29	0.00	0.00	0.00	1.23	4.56	0.00	0.00	0.00	0.00
<i>ct-3a</i>	1.39	3.64	3.22	6.53	12.84	13.23	0.75	1.81	113.05	68.39	32.25	166.92
<i>tt-3a</i>	3.36	2.56	0.00	13.50	38.58	29.05	0.00	0.00	116.82	136.80	74.43	199.66
<i>cc-4a</i>	0.00	0.60	0.00	0.00	0.00	14.46	0.00	0.00	7.93	40.04	0.13	6.46
<i>ct-4a</i>	3.66	3.51	15.67	10.28	4.61	0.00	2.19	2.53	0.00	0.00	0.00	0.00
<i>tt-4a</i>	8.43	5.53	15.60	17.29	6.57	13.11	4.07	2.34	30.42	5.69	26.22	32.50
<i>cc-5a</i>	0.00	0.12	0.00	0.00	0.00	7.82	0.00	0.99	18.97	43.43	40.16	0.00
<i>ct-5a</i>	5.59	3.70	13.56	3.82	5.20	3.41	0.58	0.27	0.00	101.71	0.00	33.67
<i>tt-5a</i>	9.75	5.48	26.62	9.36	9.96	0.00	1.64	0.00	54.84	0.00	22.04	118.23

Table S9. Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1b–5b** from Figure 1 in their three conformations. All results obtained at the HF/def2-QZVPP level of theory.

Conformer	ΔE	μ	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	1.32	5.21	6.21	0.00	0.00	0.00	2.25	2.11	60.55	189.18	9.66	47.41
<i>ct-1b</i>	0.00	4.58	0.00	1.67	3.72	9.04	1.23	0.39	0.00	164.13	11.22	31.04
<i>tt-1b</i>	0.62	3.63	2.15	13.12	19.26	61.54	0.00	0.00	8.82	0.00	0.00	0.00
<i>cc-2b</i>	0.00	4.92	0.00	0.00	2.16	0.00	0.00	0.00	0.00	18.63	0.00	0.00
<i>ct-2b</i>	0.91	4.11	2.08	2.80	0.00	4.75	0.30	0.33	52.86	0.00	33.33	111.86
<i>tt-2b</i>	2.42	3.18	1.72	6.14	3.01	9.49	0.61	0.40	53.38	16.95	34.58	119.60
<i>cc-3b</i>	0.00	4.80	0.00	0.00	4.08	0.00	0.90	2.71	0.00	42.90	0.00	0.00

<i>ct-3b</i>	1.23	3.98	2.59	4.19	0.00	2.04	0.62	0.95	78.22	39.68	41.37	146.38
<i>tt-3b</i>	3.14	3.09	2.54	9.88	7.38	6.95	0.00	0.00	97.52	0.00	59.73	169.79
<i>cc-4b</i>	0.00	1.94	1.63	0.00	3.35	0.00	0.00	3.14	0.68	47.60	0.00	33.45
<i>ct-4b</i>	3.85	4.33	7.74	4.30	0.00	23.12	1.53	3.13	11.94	0.00	9.93	40.02
<i>tt-4b</i>	8.70	6.01	0.00	5.95	11.56	78.04	2.50	0.00	0.00	35.08	27.30	0.00
<i>cc-5b</i>	0.00	1.61	0.00	0.00	3.77	0.00	0.00	0.29	136.14	32.22	45.83	210.62
<i>ct-5b</i>	2.98	4.71	10.19	4.11	2.71	0.19	1.01	0.00	49.45	41.03	12.17	123.41
<i>tt-5b</i>	4.56	5.84	16.69	4.87	0.00	17.38	1.74	0.20	0.00	0.00	0.00	0.00

Table S10. Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1a–5a** from Figure 1 in their three conformations. All results obtained at the B3LYP/def2-QZVPP level of theory.

Conformer	ΔE^{a}	μ^{b}	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	2.14	4.73	13.78	9.13	8.91	35.42	1.81	3.50	0.07	0.00	0.00	0.00
<i>ct-1a</i>	0.00	3.91	4.13	0.00	0.25	0.95	1.10	1.38	0.47	6.88	2.83	19.65
<i>tt-1a</i>	0.34	2.91	0.00	0.08	0.00	0.00	0.00	0.00	0.00	35.45	25.88	25.15
<i>cc-2a</i>	0.00	4.03	0.93	0.00	0.00	0.00	0.00	0.00	0.00	94.79	93.35	0.00
<i>ct-2a</i>	0.77	3.37	0.30	5.13	10.54	5.29	0.68	0.32	44.47	0.00	0.00	64.19
<i>tt-2a</i>	1.85	2.69	0.00	10.54	27.76	10.58	1.47	0.33	50.28	146.08	11.30	69.59
<i>cc-3a</i>	0.00	3.65	0.00	0.00	0.00	0.00	0.00	3.07	0.00	0.00	0.00	0.00
<i>ct-3a</i>	1.41	3.21	1.01	6.72	12.43	10.11	0.15	1.02	53.46	20.77	13.24	68.89
<i>tt-3a</i>	2.93	2.81	1.42	15.42	39.39	22.96	0.05	0.00	69.25	109.74	35.93	87.14
<i>cc-4a</i>	0.00	0.41	0.00	0.00	0.00	5.38	0.00	0.00	0.00	36.85	0.85	0.00
<i>ct-4a</i>	3.82	3.31	10.99	7.20	3.95	0.00	1.75	2.05	6.79	0.00	0.00	1.51
<i>tt-4a</i>	8.99	5.31	9.77	10.97	5.57	8.28	3.34	1.37	26.92	8.62	19.90	25.70
<i>cc-5a</i>	0.00	0.13	0.00	0.00	0.00	3.35	0.00	0.93	0.00	90.53	29.48	0.00
<i>ct-5a</i>	5.37	3.55	8.00	1.44	4.43	3.30	0.50	0.41	36.19	80.44	0.00	44.15
<i>tt-5a</i>	9.63	5.29	28.21	5.27	8.35	0.00	1.36	0.00	99.21	0.00	16.44	106.26

^a Values taken from [6];^b Values taken from [1].

Table S11. Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1b–5b** from Figure 1 in their three conformations. All results obtained at the B3LYP/def2-QZVPP level of theory.

Conformer	ΔE^a	μ^b	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	1.46	4.69	7.79	0.00	0.00	0.00	0.80	0.95	51.02	142.69	14.85	42.63
<i>ct-1b</i>	0.00	4.09	0.00	0.60	3.30	2.22	0.62	0.13	10.35	134.71	9.03	11.94
<i>tt-1b</i>	0.31	3.35	0.75	9.21	16.12	36.07	0.00	0.00	0.00	0.00	0.00	0.00
<i>cc-2b</i>	0.00	4.13	0.00	0.00	3.06	0.00	0.00	0.00	0.00	21.19	0.00	0.00
<i>ct-2b</i>	0.99	3.62	1.02	2.81	0.00	2.09	0.48	0.53	37.25	0.00	11.95	49.47
<i>tt-2b</i>	2.22	3.08	1.23	5.97	3.04	4.48	1.15	0.95	35.82	36.44	8.01	51.96
<i>cc-3b</i>	0.00	3.80	0.00	0.00	5.46	0.00	0.25	2.18	0.00	6.28	0.00	0.00
<i>ct-3b</i>	1.44	3.46	2.09	3.19	0.00	1.51	0.27	0.63	56.32	25.36	15.35	57.83
<i>tt-3b</i>	3.00	3.17	3.05	8.09	7.57	5.66	0.00	0.00	65.31	0.00	20.18	67.72
<i>cc-4b</i>	0.00	1.54	0.64	0.61	9.19	0.00	0.00	3.74	27.94	0.00	0.00	31.98
<i>ct-4b</i>	3.92	3.96	5.86	1.76	5.52	11.12	1.22	3.52	37.82	3.33	6.97	41.82
<i>tt-4b</i>	9.41	5.66	0.00	0.00	0.00	38.50	1.95	0.00	0.00	27.77	15.71	0.00
<i>cc-5b</i>	0.00	1.23	0.00	0.00	3.70	0.00	0.00	0.18	180.49	25.54	36.69	180.68
<i>ct-5b</i>	3.05	4.44	10.21	2.39	2.54	0.01	0.91	0.00	117.01	34.11	13.65	113.67
<i>tt-5b</i>	5.04	5.62	17.64	2.19	0.00	7.42	1.60	0.14	0.00	0.00	0.00	0.00

^a Values taken from [6]; ^b Values taken from [1].**Table S12.** Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1a–5a** from Figure 1 in their three conformations. All results obtained at the MP2/def2-QZVPP level of theory.

Conformer	ΔE	μ	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1a</i>	1.29	4.39	7.20	7.39	7.95	35.06	1.22	2.74	7.80	3.80	0.00	0.00
<i>ct-1a</i>	0.00	3.60	2.08	0.00	0.00	0.00	0.79	1.02	0.00	0.00	2.25	16.81
<i>tt-1a</i>	0.41	2.65	0.00	1.60	0.29	1.30	0.00	0.00	10.67	12.67	8.79	20.76
<i>cc-2a</i>	0.00	3.71	0.60	0.00	0.00	0.00	0.00	0.00	0.00	76.66	0.00	0.00
<i>ct-2a</i>	0.81	3.10	1.11	5.57	10.24	4.28	0.66	0.34	18.05	0.00	12.75	50.74
<i>tt-2a</i>	1.51	2.50	0.00	11.46	27.61	8.14	1.48	0.40	30.62	136.71	22.00	57.09

<i>cc-3a</i>	0.00	3.40	0.57	0.00	0.00	0.00	0.25	3.28	0.00	0.00	0.00	0.00
<i>ct-3a</i>	1.20	2.96	1.46	7.03	11.78	7.65	0.24	1.13	25.15	26.77	10.72	50.31
<i>tt-3a</i>	2.12	2.57	0.00	15.69	37.93	16.94	0.00	0.00	17.49	107.68	28.46	67.53
<i>cc-4a</i>	0.00	0.07	0.00	0.00	0.00	7.21	0.00	0.00	0.00	32.18	0.00	0.00
<i>ct-4a</i>	3.48	3.11	10.46	6.70	3.88	0.00	1.69	2.11	2.84	0.00	0.68	6.79
<i>tt-4a</i>	8.60	5.08	7.23	9.81	5.38	9.26	3.23	1.40	23.45	6.04	20.18	26.32
<i>cc-5a</i>	0.00	0.57	0.00	0.00	0.00	3.59	0.00	0.83	24.23	37.43	31.81	0.00
<i>ct-5a</i>	4.54	3.45	10.02	1.23	4.23	2.65	0.46	0.39	0.00	90.75	0.00	44.26
<i>tt-5a</i>	7.59	5.11	18.93	3.98	7.92	0.00	1.24	0.00	52.42	0.00	15.32	101.37

Table S13. Relative electron energies (ΔE in kcal/mol), dipole moments from the full density (μ in Debyes) and relative electrostatic energies (ΔE_{elst} in kcal/mol) for **1b–5b** from Figure 1 in their three conformations. All results obtained at the MP2/def2-QZVPP level of theory.

Conformer	ΔE	μ	ΔE_{elst}									
			Mulliken	NPA	GDMA	AIM	Hirshfeld	CM5	MKS	CHELP	CHELPG	HLY
<i>cc-1b</i>	0.40	4.17	2.84	0.00	0.00	0.00	0.40	0.63	69.81	161.55	20.11	47.08
<i>ct-1b</i>	0.00	3.54	0.00	1.05	3.14	5.30	0.39	0.00	0.00	148.13	11.77	12.40
<i>tt-1b</i>	0.62	2.83	2.40	10.93	16.02	45.35	0.00	0.11	15.88	0.00	0.00	0.00
<i>cc-2b</i>	0.00	3.62	0.00	0.00	4.16	0.00	0.00	0.00	62.83	16.08	0.34	0.00
<i>ct-2b</i>	1.15	3.14	3.34	1.97	0.00	1.52	0.41	0.48	3.80	0.00	6.20	36.07
<i>tt-2b</i>	2.25	2.65	3.88	4.56	2.34	3.23	1.08	0.90	0.00	34.68	0.00	36.91
<i>cc-3b</i>	0.00	3.34	0.00	0.00	6.60	0.00	0.30	2.10	0.00	0.09	0.00	0.00
<i>ct-3b</i>	1.41	2.98	3.40	2.20	0.00	0.35	0.28	0.59	8.93	30.38	9.32	42.15
<i>tt-3b</i>	2.69	2.68	4.09	5.94	6.19	2.63	0.00	0.00	9.40	0.00	10.02	48.09
<i>cc-4b</i>	0.00	1.09	3.15	3.60	10.64	0.00	0.00	3.56	2.68	32.81	0.00	33.73
<i>ct-4b</i>	3.65	3.61	7.68	3.59	6.27	10.93	1.23	3.48	25.56	0.00	9.20	49.17
<i>tt-4b</i>	9.07	5.22	0.00	0.00	0.00	40.99	1.96	0.00	0.00	20.82	18.04	0.00
<i>cc-5b</i>	0.00	0.67	0.00	0.00	4.64	0.00	0.00	0.03	119.65	31.20	44.73	195.95
<i>ct-5b</i>	2.48	4.23	9.31	1.17	6.96	0.46	0.93	0.00	70.21	31.51	18.08	128.64
<i>tt-5b</i>	3.11	5.24	13.88	1.60	0.00	9.66	1.59	0.19	0.00	0.00	0.00	0.00

References

1. Matczak, P.; Domagała, M. Charge distribution in conformers of diheteroaryl ketones and thioketones. Submitted.
2. Nelson, R.D., Jr.; Lide, D.R., Jr.; Maryott, A.A. *Selected Values of Electric Dipole Moments for Molecules in the Gas Phase*; U.S. National Bureau of Standards: Washington, DC, USA, 1967.
3. Hellwege, K.H.; Hellwege, A.M. *Molecular Constants from Microwave, Molecular Beam, and Electron Spin Resonance Spectroscopy*; Springer-Verlag: Berlin, Germany, 1974.
4. Fringuelli, F.; Marino, G.; Taticchi, A. Tellurophene and Related Compounds. *Adv. Heterocycl. Chem.* **1977**, *21*, 119–173.
5. Arnold, W.; Dreizler, H.; Rudolph, H.D. Mikrowellenspektrum, hinderungspotential der internen rotation, dipolmoment und quadrupolkopplungskonstanten des n-methylpyrrols. *Z. Naturforsch. A* **1968**, *23*, 301–306.
6. Matczak, P.; Domagała, M.; Domagała, S. Conformers of diheteroaryl ketones and thioketones: A quantum chemical study of their properties and fundamental intramolecular energetic effects. *Struct. Chem.* **2015**, doi:10.1007/s11224-015-0643-3.