

Article

Effects of Some Hill Reaction-Inhibiting Herbicides on Nitrous Oxide Emission from Nitrogen-Input Farming Soil

Yuta Takatsu, Sharon Y. L. Lau [†] , Li Li and Yasuyuki Hashidoko *

Research Faculty of Agriculture, Hokkaido University, Kita 9 Nishi 9, Kita-ku, Sapporo 060-8589, Japan; record@eis.hokudai.ac.jp (Y.T.); lauyuling@gmail.com (S.Y.L.L.); blueocean0301@gmail.com (L.L.)

* Correspondence: yasu-h@abs.agr.hokudai.ac.jp

† Present address: Sarawak Tropical Peat Research Institute, 94300 Kota Samarahan, Sarawak, Malaysia.

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Abstract: Nitrous oxide (N₂O) emission-suppressing activity of some electron-transport inhibitors of the Hill reaction system was investigated. The Hill reaction inhibitors—paraquat, isouron, bromacil, diquat, and simazine—all of which have been or are currently being used as herbicides in farming activity are expected to inhibit the electron-transporting pathways of nitrate respiration in denitrifying bacteria. Using N₂O-emitting soil bed (5.0 g of fresh weight) from a continuously manured Andisol corn farmland in Hokkaido, Japan, which was autoclaved and further supplemented with an active N₂O-emitter, *Pseudomonas* sp. 5CFM15-6D, and 1 mL of 100 mM NH₄NO₃ or (NH₄)₂SO₄ solution as the sole nitrogen source (final concentration, 0.2 mM) in a 30 mL gas-chromatography vial, the effects of the five herbicides on N₂O emission were examined. Paraquat and isouron (each at 50 μM) showed a statistically significant suppression of N₂O emission in both the nitrification and the denitrification processes after a 7-day-incubation, whereas diquat at the same concentration accelerated N₂O emission in the presence of NO₃[−]. These results suggest that paraquat and isouron inhibited both the nitrification and the denitrification processes for N₂O generation, or its upstream stages, whereas diquat specifically inhibited N₂O reductase, an enzyme that catalyzes the reduction of N₂O to N₂ gas. Incomplete denitrifiers are the key players in the potent emission of N₂O from Andisol corn farmland soil because of the missing *nosZ* gene. The electron relay system-inhibiting herbicides—paraquat and isouron—possibly contribute to the prevention of denitrification-induced nitrogen loss from the farming soil.

Keywords: electron-transport inhibiting herbicide; Hill reaction inhibitors; denitrification inhibition; Andisol farmland soil; N₂O emission suppression

1. Introduction

Nitrogen, an essential element required for plants and other living creatures, is mainly provided in the form of amino acids (as degraded protein in soil organic matters), ammonium (NH₄⁺, stable in soil), or nitrate (NO₃[−], easily lost from soil through leaching and denitrification) [1,2]. Plants or fungi inhabiting nitrogen-deficient soil acquire available nitrogen as follows: (1) Establishing symbiosis with free-living nitrogen fixing bacteria, (2) decomposing organic substances actively, or (3) suppressing nitrification of bacteria or archaea in soil, known as biological nitrification inhibition (BNI) [3,4]. *Brachiaria humidicola* and *Sorghum bicolor* of the family Poaceae (also known as Gramineae), release diterpene (brachialactone) and paraquinone derivatives (soligoquinones), respectively, from the plant roots as a BNI mechanism [5,6]. This mechanism involves the suppression of nitrogen loss from soil by the selective inhibition of ammonia oxidase in nitrifiers [4,6]. Whereas, denitrification is the most

important process associated with nitrogen loss in natural or farming soil [7,8]. To the best of our knowledge, natural products or synthesized chemical agents that can suppress this process, towards the reduction of nitrate respiration, are rarely known [9,10].

N₂O emission in farmland is attributable to human activities and it has a huge effect on global warming, because N₂O is a potent greenhouse gas which accounts for 6–8% of global warming [11]. N₂O is also a major factor affecting ozone depletion [12]. In acidic and fertilized soil utilized for farming activity, the final denitrification process, which reduces N₂O to N₂ through the catalytic reduction by N₂O reductase (NosZ), is often inhibited at the level of gene (*nosZ*) transcription [13,14]. In addition, lowered enzymatic activity in the acidic region can also inhibit the final denitrification process [15]. Hence, acidic soils such as acidic peat soils, acid-sulphate soil, coniferous forest bed soils, excess ammonium sulphate-containing-soil, tropical red soil, or volcanic ash soil, often become strong N₂O emission spots [16–19].

Nevertheless, some agricultural farmlands reclaimed from peat swampy forests, including oil palm plantation soil in Sarawak, Malaysia, showed relatively low N₂O emission despite the large addition of nitrogen and mineral fertilizers to the soil [20]. Similar woody peatland in Sumatra, Indonesia, which was converted to acacia plantation, particularly those in the mature plantation soil, suppressed N₂O emission despite the N-fertilizer input [21]. Using our culture-based N₂O emission assay, we came to the conclusion that practical N₂O flux from the soil was suppressed due to the introduction of an acceptable amount of paraquat to the plantation [20]. If any herbicide selectively inhibited nitrate respiration process treating NO_x as an electron acceptor instead of O₂, the herbicide may block the reduction process of denitrification to suppress N₂O emission, same as paraquat.

Thus, some Hill reaction-inhibiting herbicides showed potentials to suppress N₂O emission of nitrifiers and denitrifiers in the soil. According to the results of our bioassay, we show some evidence for Tollefson's short comments on the hidden effect of chemical pesticides [22,23].

2. Materials and Methods

2.1. Chemicals

Five electron transport inhibiting herbicidal compounds, methyl viologen dichloride (paraquat **1**; reagent grade), 3-(5-(tert-butyl)isoxazol-3-yl)-1,1-dimethylurea (isouron **2**; reagent grade), 5-bromo-3-*sec*-butyl-6-methyluracil (bromacil **3**, reagent grade), 1,1-ethylene-2,2'-bipyridinium (diquat **4**; reagent grade) [24], and 2-chloro-4,6-bis(ethylamino)-*S*-triazine (simazine **5**; reagent grade) were purchased from Wako (Osaka, Japan). Tropolone (**6**) was also a product from Wako (Figure 1). All the herbicides used are Hill reaction inhibitors preventing electron transport in the photosynthetic electron relay system [25]. In the soil incubation assay, N₂O-emitting soil of Andisol corn farmland, a recognized N₂O emission hotspot [21,26,27], was exposed to 50 μM of each test compound.

2.2. N₂O Emitting Soil for N₂O Emission Assay

Andisol at 0–15 cm of depth was collected from the fertilized dent corn farmland at Hokkaido University Shizunai Experimental Livestock Farm in Hokkaido, Japan (42°26' N, 142°28' E) in late April 2016, before tillage and at the first fertilization. *Pseudomonas* species, which are incomplete denitrifiers and N₂O emitters, have been isolated from the soil [21]. In addition, indirect evidence from another report shows that some unculturable soil microorganisms highly contribute to the active N₂O emission from the soil in early spring time [26].

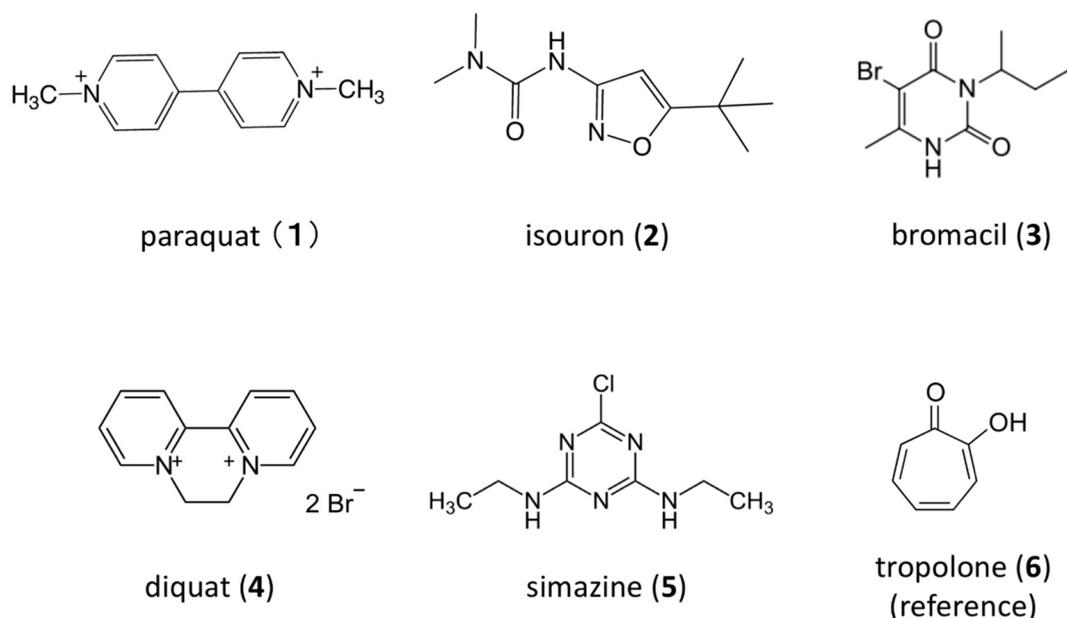


Figure 1. Test compounds.

2.3. Culture-Based N_2O -Emission Inhibition Assay Using Hill Reaction Inhibitors towards an Incomplete Denitrifier Isolated from the Andisol

Winogradsky's mineral solution-based 0.3% gellan gum soft gel culture (10 mL) was supplemented with 0.5 mg/mL sucrose and 1.44 mg/mL (14 mM) KNO_3 as carbon and excessive nitrogen sources, respectively. The culture was adjusted to pH 6.0 and sealed in gas-chromatography vials (Nichiden-Rika Glass Co., Kobe, Japan). After being autoclaved (121 °C for 15 min), 890 μ L of *Pseudomonas* sp. 05CFM15-6D cell suspension ($OD_{660} = 0.1$) and 110 μ L of 1.0 mM chemical solution (as 10% aq. DMSO) were added to the medium (finally 1.0% aq. DMSO). Hence, the headspace volume was 21.5 mL (27 mm of inner diameter, 490 mm² of opening area). Because the denitrification events are effective respiration processors under anaerobic conditions where NO_3^- is an electron acceptor, some electron transport-inhibiting herbicides isouron (2), bromacil (3), diquat (4), and simazine (5) were tested at 10 μ M, along with the same concentration of tropolone (6). Incubation was done at 25 °C in the dark for 2 weeks.

2.4. Soil Bed-Based N_2O Emission with Sucrose Supplementation

For the N_2O production assay in a soil bed, 5.0 g of the raw soil was put into a 30 mL gas chromatography vial, and 100 μ L of 100 mM NH_4NO_3 solution was added to the soil at a final concentration of 2 mM nitrogen sources (NH_4^+ and NO_3^-). To induce a stable N_2O emission by incomplete denitrifiers, sucrose was added to the soil as an additional carbon source [21,27,28]—(a) 100 μ L of 100 mg/mL sucrose solution was added to make a final concentration of 0.19% (*w/w*). Water-filled weight-base measurement of headspace volume was performed as described in the literature [29], the headspace was 27.3 mL ($n = 10$). (b) Alternative sucrose content (0.01–1.0%) was tested [28]. Incubation was done at 25 °C in the dark for 7 days.

2.5. Soil Bed-Based N_2O Emission Inhibition Assay

In the N_2O emission suppression assay, a *Pseudomonas* sp. isolated from Shinhidaka Andisol corn farm soil as an incomplete denitrifier was used [21]. N_2O emission from 5 g of the soil bed in a gas-chromatography vial and its suppression by Hill reaction-inhibiting herbicides were tested. We used the autoclaved soil inoculated with *Pseudomonas* sp. 05CFM15-6D and minimized 0.05% sucrose relatively stable for the N_2O emission, according to the results of Section 2.4. To the autoclaved soils containing inorganic nitrogen salt (0.2 mM NH_4NO_3 or 0.2 mM $(NH_4)_2SO_4$) as possible substrates of

N₂O, and 0.05% sucrose as the carbon source, 50 µM of each Hill reaction-inhibiting herbicide and 890 µL of the bacterial cell suspension (OD₆₆₀, 0.1) were added, and the mixture was vortexed before incubation. N₂O accumulated in the headspace (27.3 mL) was analyzed quantitatively by using a gas chromatography instrument as described next.

2.6. Quantitative Measurement of N₂O in Headspace

The level of N₂O in the headspace gas of the cultured vials with inoculates was measured using electron capture detector (ECD)-gas chromatography (GC-14B equipped with ECD-2014, Shimadzu, Kyoto, Japan). The gas chromatograph was equipped with an ECD maintained at 340 °C using a 1 m Porapak N column (Waters, Milford, MS, USA) maintained at 60 °C, with argon supplemented with 5% CH₄ as the carrier gas. After 2 weeks incubation, a portion of the headspace gas (50 µL–1.0 mL) was analyzed by the gas chromatography. For the quantification of N₂O gas, a standard curve was made using an absolute calibration method [27]. A series of concentrations (0, 0.498, 4.98, 49.8, and 498 ng/mL nitrogen gas) of standard N₂O gas were injected as a 1.0 mL volume for generating the standard curve to cross the origin. In this quantification of N₂O, a high precision is needed to monitor atmospheric levels of N₂O with a detection limit of 0.05 ng/L (100 ppb).

2.7. Effects of Hill Reaction-Inhibiting Herbicides on N₂O Quenching Chitinophaga

The effect of paraquat (1), diquat (4), and an iron-chelator tropolone (6), were selectively investigated on the N₂O quenching effect of a *Chitinophaga* that was isolated from the soil of Andisol corn farmland in Shinhidaka, Hokkaido, Japan, using an N₂O quenching assay system [30]. A 110 µL aliquot of a 10 mM sample solution dissolved in dimethylsulfoxide (DMSO) was added to Winogradsky's mineral solution-based 0.5% gellan gum soft gel culture (10 mL) supplemented with 0.5 mg/mL sucrose and 0.7 mg/mL aspartic acid as carbon and nitrogen sources, respectively. The culture was adjusted to pH 6.0 and sealed in gas-chromatography vials.

The gellan gum gel bed contained 100 µM of the test compound. The sample in DMSO as 10- and 100-fold diluted with Milli-Q to prepare 10 and 1 µM test compound-containing media. To all of the assay media, 890 µL of N₂O quenching *Chitinophaga* cell suspension was inoculated. In the blank medium, 890 µL Milli-Q was added instead. Finally, N₂O standard gas (standard N₂O gas, GL Sciences, Tokyo, Japan) was injected with a gas-tight syringe into the headspace (21.6 mL) of the gas-chromatography vials to make a final concentration of 12,000 ppmv. The culture in the assay vial was incubated at 25 °C for 6 days in the dark. After 3-day- and 6-day-incubation, the concentration of the remaining N₂O gas in the headspace was quantified by ECD-gas-chromatography in comparison with the culture vials to which no chemical was added, or the blank.

2.8. Statistical Analysis

The cumulative N₂O emissions were expressed as an arithmetic mean and standard deviation (±SD). Statistical analyses were done by Student *t*-tests.

3. Results

3.1. Acceleration of N₂O Emission in Soil Bed Culture Supplemented with Nitrogen Substrate and Sucrose

As the Winogradsky's mineral solution-based gellan gum bed, supplementation of nitrogen substrates, particularly with 2 mM NO₃⁻, and carbon source such as 0.2% sucrose resulted in the active acceleration of N₂O emission from the soil bed in the gas-chromatography vial (Figure 2) [28]. Although 0.01–1.0% sucrose content was also tested in parallel, N₂O emission from the soil bed culture was unstable, including control (0% sucrose), showing that the soil is thus far from a homogeneous material.

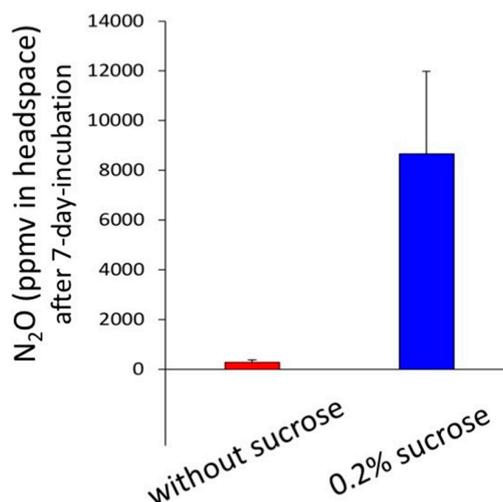


Figure 2. Acceleration of N₂O emission from Andisol soil bed by sucrose supplementation. Effect of 0.2% sucrose supplementation on N₂O emission from raw soil bed (5.0 g) from the Andisol corn farm placed in a gas-chromatography vial. Bars are standard deviation (\pm SD, $n = 5$). Soils were also supplemented with 0.2 mM NH₄NO₃.

3.2. Suppression of N₂O Emission from Soft Gel Medium Adding Hill Reaction-Inhibiting Herbicides

Among the herbicides (2–5) and tropolone (6), only diquat (4) showed a clear N₂O-emission-suppressing effect at 10 μ M (Figure 3). After the 1-week-incubation, diquat (4) reduced N₂O emission into less than a half, 106 ppmv (equivalent to 68 μ g/mL medium/d) (cf. 276 ppmv in control, equivalent to 117 μ g/mL medium/d). At 2-week-incubation, emitted N₂O level in the cultured medium treated with 10 μ M diquat was 161 ppmv (103 μ g/mL medium/d) (cf. 320 ppmv in control, 205 μ g/mL medium/d). Conversely, other chemical compounds tested at 10 μ M did not show any inhibitory effect on N₂O emission by *Pseudomonas* sp. 05CFM15-6D cultured in the soft gel medium.

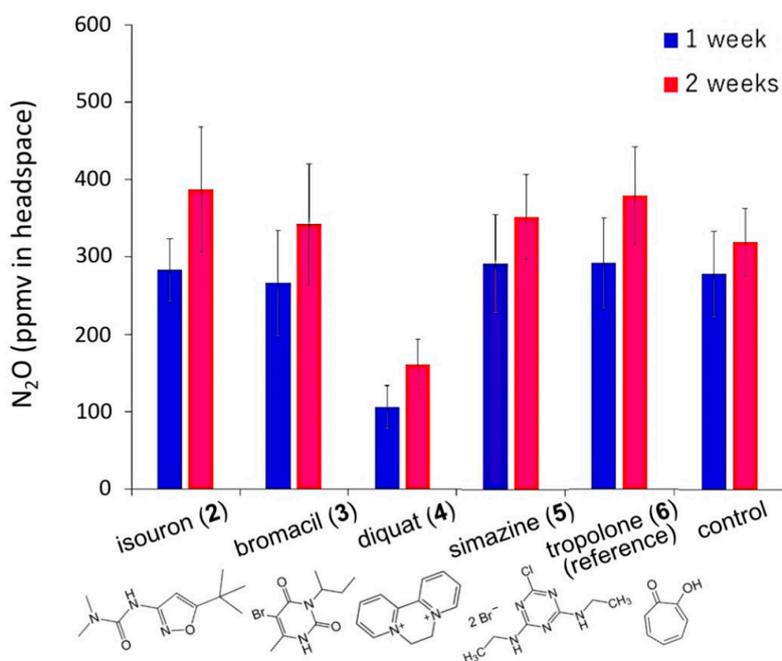


Figure 3. Suppression of N₂O emission from Winogradsky's medium-based gellan gum bed inoculated with an incomplete denitrifier *Pseudomonas* sp. by the addition of Hill reaction-inhibiting herbicides. Ingredients of the medium are shown in Materials and Methods. Bars are standard deviation (\pm SD, $n = 3$). Control contained 0.1% DMSO in the soft gel medium.

3.3. Suppression of N₂O Emission from an Autoclaved Soil of N₂O Emission Hotspot Andisol Followed by Inoculation with *Pseudomonas* sp. 05CFM15-6D

Approximately 195 ppmv level N₂O (251 ng/g soil bed/d) was emitted from the NH₄NO₃-supplemented soil bed. In addition, approximately 340 ppmv N₂O (equivalent to 437 ng/d/g soil bed) was emitted from the soil bed supplemented with 0.2 mM (NH₄)₂SO₄. Paraquat (1) at a final concentration of 50 μM showed a remarkable suppressing effect on N₂O emission from the soil bed supplemented with 0.2 mM (NH₄)₂SO₄, in which the average concentration of N₂O in the headspace was 58 ppmv (75 ng/d/g soil bed). Conversely, at the same concentration of paraquat, there was less suppression of N₂O emission (97 ppmv, equivalent to 124 ng/d/g soil bed) from the 0.2 mM NH₄NO₃⁻ supplemented soil bed. Isouron (2) also showed a similar inhibitory activity against N₂O emission from the soil bed [98 ppmv (126 ng/d/g soil bed) and 127 ppmv (163 ng/d/g soil bed) for soil beds supplemented with 0.2 mM (NH₄)₂SO₄ and 0.2 mM NH₄NO₃, respectively). Diquat (4) did not show any significant N₂O emission suppression. Similarly, bromacil (3) and simazine (5) did not inhibit N₂O emission. Although tropolone (6) is not a Hill reaction inhibitor but a potent iron-chelator [31], this iron-chelating compound was also tested as a reference compound. However, it did not show any suppressing effect on N₂O emission (Figure 4).

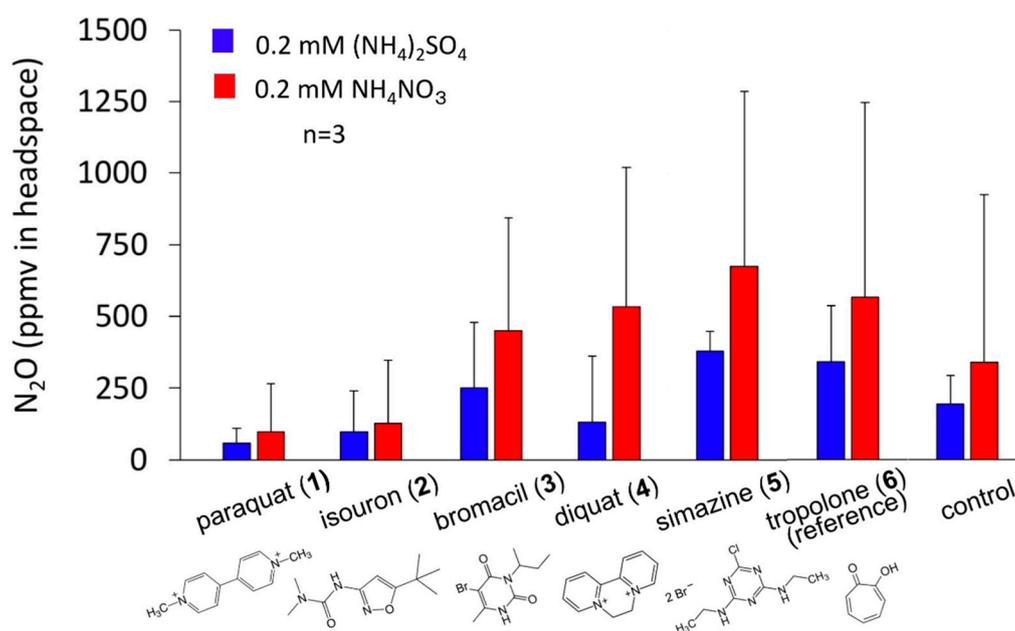


Figure 4. Suppression of N₂O emission from Andisol soil bed by the addition of the Hill reaction-inhibiting herbicides. Incubation was performed for seven days. Bars are standard deviation (\pm SD, $n = 3$).

3.4. Suppressing Action of the Herbicides on Actively N₂O Quenching *Chitinophaga* Isolated from an Andisol Corn Farm Soil

Unlike N₂O emission, the N₂O quenching effect of the *Chitinophaga* bacterium was not inhibited by 1, 10, and 100 μM paraquat (1). Diquat (4), a bipyridylium-type Hill reaction-inhibiting herbicide, at a final concentration of 1 or 10 μM did not suppress N₂O quenching, but 100 μM of 4 reduced the effectiveness of N₂O quenching by *Chitinophaga* after the 3-day-incubation, with 60% of the N₂O remaining in the headspace. Whereas, in all the vials at the three different concentrations, N₂O was quenched to almost zero level at day 6. At a final concentration of 100 μM, treatment with tropolone (6) resulted in nearly 80% inhibition of the N₂O quenching, although no effect was observed at 1 or 10 μM. This suppression of N₂O quenching continued for 6 days, and at day 6, 40% N₂O remained in the headspace (Figure 5).

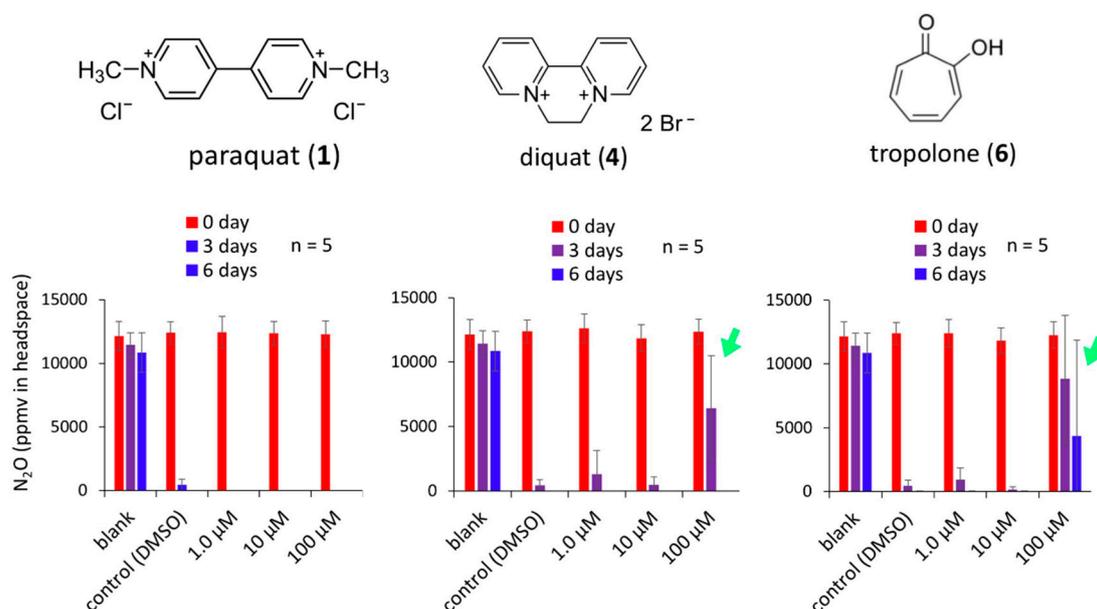


Figure 5. Effect of bipyridylium cation-type Hill reaction-inhibiting herbicides, paraquat, and diquat, on an N_2O quencher, *Chitinophaga* strain isolated from Andisol corn farming soil (green arrows). Bars are standard deviation ($\pm SD$, $n = 5$).

4. Discussion

The results of alternative sucrose supplementation to the raw bulk soil, showing no direct proportion to N_2O emission (Figure 2), suggested that soil microbial ecosystems in the bulk soil consisted of diverse community members. Among them, some may show a faster response to sucrose rather than the incomplete denitrifiers. Hence, the soil, that is far from a homogeneous material, probably showed such an alternative response of N_2O emission to the carbon source added.

On the other hand, N_2O emission-inhibiting effect in the Winogradsky's mineral solution-based gellan gum soft gel medium was observed only in diquat (4) among the herbicides (2–5) and tropolone (6) tested at 10 μM (Figure 3). Using gellan gum gel bed for the culture-based N_2O emission assay, paraquat (1) and diquat (4), both of which are bipyridylium cation-type Hill reaction inhibitors, showed clear suppression of N_2O emission by the incomplete denitrifiers of *Pseudomonas* species [21]. Other chemicals (2, 3, 5, and 6) did not show any inhibitory effect on N_2O emission at 10 μM , indicating that pyridilium cation moiety is necessary for denitrification inhibition.

In an autoclaved soil bed, however, two electron-transport inhibiting herbicides, paraquat (1) and isouron (2) at a final concentration of 50 μM suppressed N_2O generation from the 2 mM $NH_4NO_3^-$ supplemented model soil bed of fertilized Andisol farmland bulk soil (Figure 4). However, three of the electron-transport inhibiting herbicides (3–5), including diquat (4), did not show any potent suppressing effect on N_2O emission. Despite a similar bipyridylium cation structure with 1, diquat (4) did not show any inhibiting activity against N_2O emission in the soil beds. Thus, N_2O -producible incomplete denitrification with NO_3^- as the substrate was inhibited by 1 and 2 only. These clear differences in the suppressing effects of different Hill reaction inhibitors on the emission of N_2O from the soil bed culture suggest a specific inhibition of certain oxidoreductase, highly associated with the denitrification process and/or the nitrification process [20]. Conversely, the results shown in Figure 4 probably indicate that not only incomplete denitrifiers but also thermo-tolerant nitrifier is another player in N_2O emission in the farm soil. In the autoclaved soil, *amoA*-harboring thermo-tolerant archaea may be surviving and emerging as major nitrifiers in the soil bed [32].

In contrast, the inhibitory activity of 100 μM diquat (4) on N_2O quenching by *Chitinophaga* may indicate that N_2O production and N_2O quenching in the soil are separable responses [33]. Oxidoreductases associated with inorganic nitrogen metabolism are often specific to a Hill reaction

inhibiting herbicide. Tropolone (6), which inactivates iron-containing heme-dependent cytochrome c or iron-sulfur-cluster containing oxidoreductases showed a relatively high suppression of N₂O quenching by *Chitinophaga*.

Thus, the current data demonstrates the roles of some Hill reaction-inhibiting or other herbicides in the suppression of N₂O flux from agricultural farming soil [22,23]. This implication is important because bifunctional herbicides that can effectively prevent the loss of nitrogen by denitrification as well as control the growth of weed can possibly be developed with a molecular design and bio-rational screening systems [9,10,34]. Thus far to our knowledge, the mechanism and mode of action of biological denitrification inhibition are rarely known, but some reliable reports have been published [35,36].

5. Conclusions

In this study, we demonstrated that a 50 µM level of two electron transport-inhibiting (Hill reaction-inhibiting) herbicides, paraquat (1), and isouron (2), which have been approved as regal herbicides, showed potent suppression of N₂O emission from the farm soil bed modified for fertilized conditions. This result implied that herbicides 1 and 2 positively contributed to environmental sustainability via suppression of nitrogen loss and N₂O emission from fertilized soils. However, three of the herbicides (3–5) did not show any suppressing effect on N₂O emission. This suggests that some electron-transport inhibiting chemicals are associated with selective steps of inorganic nitrogen metabolism. This specific inhibition targeting certain oxidoreductases, particularly the ones associated with the denitrification process, may provide new approaches that can be used to suppress N₂O flux from agricultural soil and more importantly to prevent the loss of nitrogen from fertilized soil by denitrification.

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Conflicts of Interest: The authors declare no conflict of interest.

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