

Review



# **Understanding How Reservoir Operations Influence Methane Emissions: A Conceptual Model**

Henriette I. Jager \*<sup>®</sup>, Rachel M. Pilla <sup>®</sup>, Carly H. Hansen, Paul G. Matson <sup>®</sup>, Bilal Iftikhar <sup>†</sup> and Natalie A. Griffiths <sup>®</sup>

Oak Ridge National Laboratory, Oak Ridge, TN 37831-6038, USA; pillarm1@ornl.gov (R.M.P.); hansench@ornl.gov (C.H.H.); matsonpg@ornl.gov (P.G.M.); Iranabilal122@gmail.com (B.I.); griffithsna@ornl.gov (N.A.G.)

\* Correspondence: jagerhi@ornl.gov

<sup>+</sup> Current address: Irrigation Department, Government of Punjab, Lahore 54840, Punjab, Pakistan.

Abstract: Because methane is a potent greenhouse gas (GHG), understanding controls on methane emissions from reservoirs is an important goal. Yet, reservoirs are complex ecosystems, and mechanisms by which reservoir operations influence methane emissions are poorly understood. In part, this is because emissions occur in 'hot spots' and 'hot moments'. In this study, we address three research questions, 'What are the causal pathways through which reservoir operations and resulting water level fluctuations (WLF) influence methane emissions?'; 'How do influences from WLF differ for seasonal drawdown and diurnal hydropeaking operations?'; and 'How does understanding causal pathways inform practical options for mitigation?'. A graphical conceptual model is presented that links WLF in reservoirs to methane emissions via four causal pathways: (1) water-column mixing (2) drying-rewetting cycles, (3) sediment delivery and redistribution, and (4) littoral vegetation. We review what is known about linkages for WLF at seasonal and diurnal resolutions generate research questions, and hypothesize strategies for moderating methane emissions by interrupting each causal pathway. Those related to flow management involve basin-scale management of tributary flows, seasonal timing of hydropeaking (pathway #1), timing and rates of drawdown (pathway #2). In addition, we describe how sediment (pathway #3) and vegetation management (pathway #4) could interrupt linkages between WLF and emissions. We demonstrate the strength of conceptual modeling as a tool for generating plausible hypotheses and suggesting mitigation strategies. Future research is needed to develop simpler models at appropriate timescales that can be validated and used to manage flow releases from reservoirs.

**Keywords:** methane emissions; conceptual model; water-level fluctuations; reservoir operation; greenhouse gas; mitigation

# 1. Introduction

Freshwater networks return a substantial proportion of terrestrial carbon to the atmosphere as methane [1], accounting for half of global emissions [1]. Knowledge about the effects of water-level fluctuations on methane comes from rivers [2,3], lakes [4], and wetlands [5]. In particular, rewetting of river floodplains during warm periods is associated with high methane emissions [3].

Reservoirs are now a part of freshwater networks. A subset of these has structures to control water level fluctuations, which has implications for greenhouse gas emissions. Relationships between reservoir operations and greenhouse-gas (GHG) emissions are complex and poorly understood. As a renewable energy source, hydropower displaces fossil fuels and adds grid stability by filling in for other renewables. However, reservoirs that support hydropower generation also tend to have large surface areas and can emit significant amounts of methane [6–8]. Reconciling the costs and benefits of hydropower



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**Copyright:** © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). generation, particularly in terms of its carbon footprint [9], is crucial to meet sustainable energy production goals.

Reservoirs are complex systems with carbon dynamics influenced by both natural and anthropogenic factors. The initial inundation and subsequent decay of terrestrial organic matter typically occurs over the first decade of a reservoir's life [10,11]. In older reservoirs, methane production is supported by the decomposition of terrestrial organic matter from upstream inputs and the surrounding watershed, as well as by in situ algal production. Methane emissions have been described as sporadic, often occurring as 'hot moments' during which conditions are right for the gas to be produced and emitted [12,13]. The timing of hot moments is influenced by seasonal patterns of flows, carbon and nutrient inputs [14], water-column stratification [15], and water-level fluctuations [16–19].

Fewer than 3% of reservoirs in the US support hydropower [20]. For this subset, GHG emissions are mediated by the timing and amounts of water released, which control reservoir water-level fluctuations (WLF) at seasonal and diurnal time scales. At a seasonal timescale, water levels in storage reservoirs used for flood control are drawn down by releasing water in the fall. This ensures that enough storage capacity is available in the reservoir to store water from spring high flows to fulfill water demand during drier summer conditions. Some hydropower reservoirs are operated to release more water when energy prices are high. The increased release of water during times of peak, unmet electricity demand results in diurnal fluctuations in water level ('hydropeaking') [21]. A review of 33% of US hydropower plants found that roughly 40% of them released flows with significant sub-daily variation during at least part of the year [22]. As variable renewables (solar and wind power) are being added to the grid, twice-daily hydropeaking is becoming more frequent to compensate when variable renewables are not generating ('double peaking') [23]. Dams operated in peaking mode either have more storage than non-peaking projects or, for run-of-river reservoirs with short residence times, they pass stored water from upstream storage reservoirs releasing fluctuating flows [22]. Most run-of-river hydroelectric reservoirs are required to balance inflows and outflows on a daily timescale and can therefore experience significant within-day fluctuations in water level. Both seasonal and diurnal fluctuations in reservoir water level can affect methane emissions through various mechanisms, including changes in hydrostatic pressure [24], mixing dynamics [25], and drying–rewetting cycles [26].

The goal of this study is to improve our understanding of processes linking reservoir operations to carbon dynamics leading to methane emissions. Key questions addressed include:

- What are the causal pathways through which reservoir operations and resulting WLF influence methane emissions?
- How do influences from WLF differ for seasonal drawdown and diurnal hydropeaking operations?
- How does understanding causal pathways inform practical options for mitigation (i.e., reducing methane emissions via scheduled releases)?

We developed a conceptual model to clarify the causal linkages between WLF (specifically those associated with hydropower operations) and methane emissions. Conceptual ecological models (also called 'causal models') are graphic representations (diagrams) with a written description of predicted relationships between anthropogenic stressors and environmental outcomes. They are used as a first step in ecological risk analysis [27]. The diagrams represent hypotheses about causal linkages between the management actions of interest (e.g., reservoir operation) and outcomes (e.g., methane emissions). Although they are qualitative, causal models can be used as a first step in developing quantitative models, such as structural equation models [28] or dynamic systems models [29]. We also review equations associated with the system.

# 2. Conceptual Model

A review of the recent literature on GHG emissions from reservoirs identified key processes and factors that can aid in predicting emissions [9]. Here, we update and augment that review to understand key predictors to consider in future when constructing indicators of methane emissions. Our expanded review includes information from other ecosystems, such as wetlands and floodplains, that are relevant to understanding linkages between WLF and methane emissions. Based on our review and understanding of methane dynamics, we produced a graphical conceptual model of causal linkages between reservoir operations and methane emissions from reservoirs. We organized our review of methane production and transport to the atmosphere around causal pathways (drying–rewetting, sediment delivery and redistribution, plant-mediated emissions, mixing and stratification, turbine withdrawal) with distinctions made between two temporal scales (seasonal dynamics, diurnal hydropeaking). This should be seen as a strawman and used as the basis for further refinement. For each causal pathway, we report relevant equations and describe which methane generation and emission processes are most influenced by each pathway.

Our conceptual model seeks to understand the causal pathways through which methane emissions are linked to WLF at different time scales (Figure 1). Three causal pathways take place mainly in the littoral zone (drying-rewetting, sediment delivery and redistribution, and plant-mediated emissions), whereas mixing mainly takes place in the limnetic zone along with turbine withdrawal. First, WLF can alter the mixing dynamics of reservoirs (causal pathway #1 in Figure 1). WLF can disrupt thermal stratification and vertical oxygen gradients that control oxygen levels below the thermocline and conditions that promote methane production. The second causal pathway identified here is mediated by drying and rewetting of littoral sediments. In the middle box of Figure 1, we distinguish between the dominant effects of WLF during the wetting phase and the drying phase (Figure 2). We also distinguish diurnal (short-term) and seasonal (long-term) effects of drying, where we assume that drawdown and refill represent slower, longer-term changes in water elevation than hydropeaking. Third, WLF can deliver sediment from floodplains and cause resuspension and focusing of sediment in deeper areas of the reservoir [30]. This should lower the potential for methane emissions during rewetting cycles, which can become limited by availability of organic carbon in shallower areas [31,32]. Fourth, alternate drying and rewetting of sediments can result in a loss of littoral vegetation as few plant species are adapted to extended periods of dry and wet soils or sediment erosion. Reduced vegetation reduces the potential for plant-mediated methane transport. Each of these pathways is described in more detail in the sections below.



**Figure 1.** Causal diagram showing linkages between reservoir water level fluctuations and associated changes in methane emissions via different emission pathways (ebullitive, diffusive, and degassing) shown in cyan boxes. Longer- (short-)term responses are shown in green (cyan). Four causal pathways, indicated by black circles with numbers, are (1) water-column mixing, (2) drying rewetting cycles, (3) sediment delivery and redistribution, and (4) vegetation.

In addition to conceptual modeling, we diagramed a simplified systems model with methane pools and fluxes (Figure 2). Although a complete systems model is beyond our scope, we present equations corresponding with the system depicted in Figure 2

for amounts of methane (masses) in three pools, sediment,  $CH4_{(t)}^{sed}$ , deep, hypolimnetic water,  $CH4_{(t)}^{deep}$ , and the epilimnion,  $CH4_{(t)}^{top}$  (Table S1, Equations (S1)–(S3)). We refer to mass flows as 'fluxes' for convenience, and many parameters are rates representing fluxes as a proportion of a donor pool's mass (Table 1). The flux of methane produced in sediment  $P_{(t)}^{sed}$  adds to the pool of methane in sediment,  $CH4_{(t)}^{sed}$ , which can leave the pool in three ways, via (1) sediment diffusion at rate,  $Df_{(t)}^{sed}$ , (2) bubble release from the sediment at rate  $Br_{(t)}$ , or (3) as a flux to the atmosphere through plants, Ap [33] (Figure 2). Rates depend on temperature and other factors not explicitly included. Variables used in the equations are defined in Table 1.



**Figure 2.** Diagram linking pools (rectangles), internal fluxes (dashed arrows), and atmospheric emissions (solid arrows) of methane in a reservoir. Pathways linking methane production in sediment,  $P^{sed}$ , to its release to the atmosphere include direct transport through plant tissue, Ap, and multiple pathways through the water column. The diffusive pathway to the atmosphere involves three steps: (1) diffusion from deep sediment to the hypolimnion,  $D_f^{sed}$ , (2) from the hypolimnion to the epilimnion,  $D_f^{deep}$  the surface flux to the atmosphere, Df. A second pathway is ebullition, Eb, and a third is via degassing during turbine passage, Tb and downstream methane. See text and equations for definitions of fluxes. Ebullition emissions occur if methane produced in sediment,  $P^{sed}$ , forms into bubbles that are released at rate, Br, from the sediment and reach the atmosphere without dissolving. In deeper areas, bubble dissolution, Bd, adds to the pool of dissolved methane in the hypolimnion,  $CH4^{deep}$ . Within and above the oxycline, oxidation, Ox, converts a proportion of dissolved methane to  $CO_2$  and thereby reduces diffusive methane emissions. Some pathways are depicted on only one side of the reservoir cross-section to improve clarity.

| Symbol                       | Description   | Generic Units        |
|------------------------------|---|----------------------|
| Ap                           | Fractional mass discharge (i.e., rate) of methane transfer from sediment to atmosphere via plant aerenchyma from sediment methane pool, $CH4^{sed}_{(t)}$ | Mass/mass/time       |
| Bd                           | Fractional mass discharge (i.e., rate) of methane bubble dissolution from pool $CH4^{bub}_{(t)}$  | Mass/mass/time       |
| Bd <sub>max</sub>            | Maximum <i>Bd</i>   | Same as above        |
| $Br^{sed}_{(t)}$             | Rate of methane bubble release from sediment, fractional mass discharge from sediment pool, $CH4_{(t)}^{sed}$   | Mass/mass/time       |
| $CH4^{bub}_{(t)}$            | Pool of methane contained in bubbles at time <i>t</i>   | Mass                 |
| $CH4_{(t)}^{deep}$           | Pool of methane in the hypolimnion at time <i>t</i>   | Mass                 |
| $CH4^{sed}_{(t)}$            | Pool of methane in the sediment at time <i>t</i>  | Mass                 |
| $CH4_{(t)}^{top}$            | Pool of methane in the epilimnion at time <i>t</i>  | Mass                 |
| $\left[CH4^{*}_{(t)}\right]$ | Threshold sediment methane concentration for ebullition   | Mass/sediment volume |
| Dox                          | Vertical distance (depth) at which 4-6 mm diameter bubbles completely dissolve  | Depth (e.g., m)      |
| $Df^{deep}$                  | Rate of methane diffusion from hypolimnion to the epilimnion  | Mass/mass/time       |
| $Df^{sed}$                   | Rate of methane diffusion from sediment to the hypolimnion  | Mass/mass/time       |
| Df                           | Rate of methane from epilimnion to the atmosphere via diffusion   | Mass/mass/time       |
| $Eb_{(t)}$                   | Ebullition mass discharge from sediment as a proportion of $CH4_{(t)}^{sed}$ at time t  | Mass/mass/time       |
| K <sub>ebu</sub>             | Rate of ebullition when sediment methane concentration exceeds threshold, $\begin{bmatrix} CH4_{(t)}^* \end{bmatrix}$                                     | Mass/mass/time       |
| Ox <sub>(t)</sub>            | Proportion of methane oxidized (in a given pool) per unit time  | Mass/mass/time       |
| Pa                           | Atmospheric pressure near sediment  | Force/area           |
| $P_{(t)}^{sed}$              | Rate of production of methane in sediment per unit mass in sediment at time $t$   | Mass/mass/time       |
| $Tf_{(t)}$                   | Degassing rate from hypolimnetic inflow to the atmosphere at time $t$   | Mass/mass/time       |
| $Tb_{(t)}$                   | Degassing flux of methane to the atmosphere due to turbine passage at time $t$  |                      |

## Table 1. Definitions of symbols.

Dissolved methane concentration in the hypolimnion,  $[CH4^{deep}]$ , increases via diffusion of methane from sediment at rate  $Df^{sed}$ , and dissolution of methane bubbles from pool  $CH4^{bub}$  at rate Bd. The deep-water pool loses methane to oxidation, Ox, diffusion into the surface layer,  $Df^{deep}$ , and degassing due to turbine passage, Tb [Table S1, Equation (S2)]. Degassing depends indirectly on the production of methane,  $P^{sed}$ , in sediment and competing losses from sediment via transport through plant aerenchyma tissue, Ap. In a spatial model representing methane dynamics in littoral areas separately from the limnetic/open-water zone, one could assume that Ap is negligible in the limnetic zone.

Ebullition is an important pathway whereby methane protected inside bubbles avoids oxidation during their ascent through the water column. Ebullition is influenced by the sum of changes in atmospheric and hydrostatic pressure [34]. Decreases in reservoir elevation reduce hydrostatic pressure,  $\Delta Pa$ , experienced by littoral sediments, which can produce a short-term pulse of methane via ebullition [16,17,35,36]. When the rate of methane production in sediment exceeds the rate at which it can diffuse into the water column, bubbles form and ebullition takes place [37]. The rate of bubble release from sediment, *Br*, is controlled by temperature and hydrostatic pressure [see [38], Equation (S5)]. One study reported that methane ebullition followed an exponential increase with decreasing

pressure as water levels dropped from 0.2 to 1.3 m [39]. Conversely, ebullition diminished as water levels rose [39].

The pool of methane contained in bubbles,  $CH4_{(t)}^{bub}$ , is incremented by methane released from sediment and decremented by methane lost to bubble dissolution or ebullition flux from the surface (Figure 2). The flux of methane bubbles released from sediment,  $Br_{(t)}$ , depends on an upper methane concentration threshold [Table S1, Equation (S7)], which, in turn depends on pressure, bubble volume, and temperature [38,40]. Because bubbles form and release only at low ambient hydrostatic pressures, models that consider pressure predict ebullition events better than those relying on sediment methane concentration alone [38,41]. In addition to depth-related hydrostatic pressure, ebullition increases with increasing temperature and decreases with increasing wind speed [42].

Methane in bubbles dissolve at rate Bd, resulting in a lower ebullitive emission by the time the bubbles reach the water's surface and increased concentrations of dissolved methane (Figure 1, pathway #2; Figure 2). Bubble dissolution,  $Bd_{(t)}$  [Table S1, Equation (S8)], depends on bubble depth at release and dissolution rate while traveling through the water column and other factors, such as bubble size [43]. The distribution of initial bubble sizes is an important factor because bubble size is proportional to bubble rise velocity [44]. Bubble size, in turn, is determined by sediment characteristics, decreasing from clay to sand to silt [45]. Thus, WLF may influence bubble size through grain size distributions, an outcome of sediment redistribution (Section 2.3).

When methane bubbles dissolve, aerobic methanotrophs in the water column incorporate carbon into microbial biomass in the oxycline [46]. This can prevent a significant fraction of dissolved methane from reaching the atmosphere [46]. This process has been represented differently by different models. For example, the monthly Reservoir Methane Emissions model (ResME) represents oxidation in oxic portions of sediment layers as a function of bubble dissolution, but it does not represent oxidation of dissolved methane in the water column [47]. ResME describes bubble dissolution from the point of bubble release from sediment as a function of average reservoir depth. A 60-m reservoir depth limit [Table S1, Equation (S8)] is based on the inference that 4–6-mm diameter bubbles will completely dissolve after rising through  $D_{ox} = 60$  m of water [48]. This depth may vary among reservoirs. For example, a model of emissions from a well-studied reservoir in Israel estimated that 60% of simulated methane in bubbles released from sediment reached the atmosphere [49], even though methane was almost-completely oxidized. ResME assumes that the daily fraction of methane in bubbles that dissolves,  $Bd_{max}$ , is less than 20% because some methane produced in littoral sediments will still reach the atmosphere, even in deep reservoirs. The remainder of methane produced is oxidized after bubbles dissolve [47]. Another model, the GHG Reservoir (G-Res) model represents oxidation of dissolved methane, but not bubble dissolution [17].

Model assumptions about depth-dependent oxidation rates in the water column can be supported by isotopic depth profiles of methane [14,46]. Ideally, depth dependence of oxidation should reflect both the increased availability of dissolved oxygen to methanotrophs and the decreased dissolution of methane bubbles in shallower water. Consistent with our two-layer conceptual model (Figure 2), methane in the surface layer is incremented by diffusion from deeper layers and decremented by ebullitive and diffusive surface fluxes to the atmosphere (Figure 2). Diffusive methane emission,  $Df^{top}$ , emanating from the pool of methane in the epilimnion,  $CH4_{(t)}^{top}$ , can be estimated by subtracting the proportion oxidized,  $Ox_{(t)}$  [Table S1, Equation (S6)], and adding methane from bubble dissolution, which occurs at rate  $Bd_{(t)}$ , a function of water depth [Table S1, Equation (S8)]. Oxidation within sediment (under certain conditions), bubble release from sediment to deep water, and subsequent dissolution and oxidation in the water column, would be represented as separate processes.

### 2.1. Water-Column Mixing

WLF induces mixing (pathway #1 in Figure 1), which has different short- and longterm effects on methane production and emission. Typically, turnover events are seen as producing pulses of diffusive emissions in the short term, but oxygenation of deep waters reduces methane concentrations and decreases methane emissions over the long term. For example, in the short-term, diurnal interruptions of stratification in Iron Gate I, a relatively shallow run-of-river reservoir on the Danube River, enabled diffusion of methane in the water column [50]. Advective mixing also influences oxidative methanotrophy by controlling where the supply of oxygen, dissolved methane, and methanotrophs in the water column [46] coincide with thermal conditions that favor methanotrophy.

The proportion of the world's largest reservoirs that stratify is not well known [51], with a recent estimate of 70% based on reservoir morphometry [52]. Both the timing and depth of water releases from reservoirs influence mixing and stratification, especially when the change in volume associated with WLF is a large proportion of reservoir volume at full pool [53]. Selective withdrawal from deep water layers shortens the period of summer stratification by depleting cold water in the hypolimnion and decreasing the stability of the water column [54]. Generally, faster-flowing waters (those with shorter residence times) do not stratify because there is no opportunity for surface heating to produce depth gradients as it does in more stagnant waters. The source of carbon fueling methane production also depends on residence time. In reservoirs with short residence times, methane production is linked to allochthonous carbon, whereas methane production in reservoirs with longer residence times is better predicted by autochthonous carbon (e.g., algae) [47].

Degassing, i.e., emissions from water flowing through turbines, are a poorly quantified, component of reservoirs' carbon footprint of particular relevance to hydropower [55]. Estimates vary widely. One estimate based on the G-Res model suggested that degassing accounts for 31–52% of global methane fluxes from reservoirs [8]. This estimate is likely high because it assumed that all reservoirs stratify and that all flows in reservoirs that support hydropower are drawn from methane-rich hypolimnetic water. Another study of US reservoirs did not find degassing to be the dominant emissions pathway [55]. Degassing is strongly dependent on methane concentration and flow; as water passes through the intakes (or turbines in the case of hydropower), hydrostatic pressure decreases and dissolved methane is released immediately downstream of the discharge [56].

Lima, Ramos, Bambace and Rosa [56] modelled degassing as a fraction,  $\phi$ , of methane passing through turbines. For reservoirs with bottom withdrawals, degassing is proportional to the concentration of methane in the hypolimnion,  $[CH4]_{(t)}^{deep}$ . We refined the equation to represent the fraction of flow,  $Q_{(t)}$ , passing through turbines on day t, by incorporating fractional turbine flow,  $Tf_{(t)}$ , which ranges from zero (no generation, 100% spill) to one (100% generation, no spill) [Table S1, Equation (S10))].

Methane emissions from managed reservoirs are influenced by the timing of water releases (i.e., reservoir operation), the depth of intakes, and the proportion of flow passed through turbines versus non-turbine routes, such as bypass structures and spill. Spillways may only be active for part of the year to cope with seasonal precipitation or snowmelt events [56] or to provide downstream fish passage [24]. To integrate reservoir operations in a way that is consistent with our conceptual model, turbine degassing can be represented by integrating  $Tf_{(t)}$  over time, where  $[CH4]_{(t)}^{deep}$  is a function of antecedent hypoxic conditions [Table S1, Equations (S1) or (S2)]. Note also that methane removal through turbines from the deep-water layer decreases the pool of hypolimnetic methane when withdrawals exceed production (Figure 2). In addition to degassing, elevated methane levels in tailwaters can lead to high downstream emissions that decays with distance downstream [57,58].

## 2.1.1. Seasonal Dynamics (Stratification, Turnover, and Drawdown)

Typical seasonal patterns in stratification and WLF that influence methane emissions are depicted in Figure 3. In temperate regions, deeper reservoirs become thermally stratified

as the gradient between warmer surface waters and colder, denser bottom waters increases in late spring. This leads to a separation of oxygenated water above the thermocline and less-oxygenated water below, where resupply of oxygen to deeper waters via wind-driven mixing or photosynthesis is negligible. Inverse stratification can also occur after ice-over in northern latitudes. During fall turnover, surface temperatures decrease causing shallower waters to reach similar densities as deeper waters, at which point the water is less resistant to advective forcings (e.g., wind, flow) that induce mixing (Box 3 in Figure 3). Similarly, in reservoirs that freeze over, mixing events occur after ice-off in spring (left-most Box 5 in Figure 3).



**Figure 3.** Diagram relating seasonal patterns including stratification, turnover, along with reservoir operations for flood control (winter drawdown and spring refill) as they influence causal pathways for methane emissions. Within seasonal boxes, processes in black decrease whereas those in purple italics increase. The lower Box #5 applies only to dimictic reservoirs (i.e., those that ice over in winter), whereas seasonal refill applies to most managed reservoirs. The circle represents seasonal phenological patterns corresponding to water level drawdown, low winter water levels, and subsequent refill in littoral areas of temperate reservoirs. Credits: Insets drawn by Adam Malin (ORNL) and inspired by Carmignani and Roy [59].

Seasonal dynamics of methane emissions differ significantly in reservoirs that stratify and those that do not, and for those that stratify, reservoir release patterns can have important effects. Typical seasonal patterns with reservoir drawdown lead to lower water releases in summer, which decreases mixing and increases surface heating and water-column stability (i.e., stratification) [60]. While a reservoir is stratified, methane concentrations increase in the hypolimnion, controlled by redox dynamics and temperature. During stratification, methane build-up in the hypolimnion has been observed across a range of reservoir types in the US [16,17,24]. If we assume negligible oxidation occurs in the deep, anoxic portion of the water column, methane accumulation in the hypolimnion can be modeled by adding daily methane diffusion,  $Df^{sed}$  from the anoxic sediment to the deep-water pool and bubble dissolution below the oxycline and subtracting daily amounts removed by turbine degassing, Tb [Equation (1)].

$$CH4_{Tmix}^{deep} = CH4_{Ts}^{deep} + \sum_{t=Ts+1}^{Tmix} \Delta CH4_{(t)}^{deep},$$
(1)

10 of 26

where

$$\Delta CH4_{(t)}^{deep} = CH4_{(t)}^{deep} \left(1 - Tb_{(t)}\right) + Df_{(t)}^{sed} CH4_{(t)}^{sed} + Bd_{(t)} CH4_{(t)}^{bul}$$
$$Tmix = \text{day of fall overturn,}$$

ts = day of stratification, onset

Any significant mixing events during the stratification period would result in oxidation of a proportion, pM, of  $CH4^{deep}$  that could be modeled as an increasing function of the magnitude of WLF. Equation (1) can be generalized by defining R 'runs' of dates with anoxic conditions, r = 1 to R, where the first day of fall turnover occurs at the end of the last, Rth, run, i.e., Tmix = tm(R). Let the start of the stratification (or restratification) period r be ts(r) and the end of the stratified period r be tm(r). Cumulative methane in the deep layer can then be represented by Equation (2).

$$CH4_{Tmix}^{deep} = \sum_{r=1}^{tm(r)} pM_{(r)} \sum_{t=ts(r)}^{tm(r)-1} \Delta CH4_{(t)}^{deep},$$
(2)

where

tm(r) = day of rth mixing event

ts = day of [re] stratification

Fall turnover is considered a potential 'hot moment' for emissions [61]. Methane generated in the sediments and water column of the hypolimnion is trapped in deep water during stratification. Later, the portion of this accumulated methane emitted via diffusion and ebullition pathways can represent a significant fraction of reservoir emissions, especially as a pulse of emissions during turnover [62]. In one Virginia reservoir, ebullition rates peaked just before fall turnover, whereas methane diffusion peaked during turnover [42]. These turnover pulses may be localized hot spots; for example, cold temperatures and replenished dissolved oxygen at the time of fall turnover may depress methane production in shallow areas [14]. Diffusion of some accumulated methane across the thermocline may occur prior to turnover, especially when concentrations below the thermocline are high [14]. Often, a fraction of methane is not oxidized during fall mixing and remains in the dissolved methane pool. The remaining fraction varies (e.g., a reasonable range may be from 50 to 90% [47]) depending on reservoir geomorphology. The opportunity for oxidation is greater in deep hydropower reservoirs, preventing 90% [62] to 99% [49] of dissolved methane in the hypolimnion from reaching the atmosphere via diffusion.

Like emissions during fall turnover, emissions of methane accumulated under ice may peak after spring ice-off. For example, spring pulse emissions were reported from the surface waters of deep boreal reservoirs [63]. In warmer climates, extended anoxic conditions and methane storage during summer produce larger emissions during autumn overturn, whereas in cooler climates, larger pulses are produced during spring turnover [61]. However, methane generation in winter is likely lower due to lower water temperatures.

Superimposed on natural seasonal patterns are the effects of reservoir operations. A large majority of US reservoirs experience at least one annual drawdown > 0.5 m [17]. This typically occurs in autumn in snowmelt-dominated and many rainfall-dominated catchments of the US. Because methane builds up in the hypolimnion during stratification, the timing of artificial reservoir drawdown relative to natural seasonal patterns is important. Several studies have documented increases in ebullitive methane emissions in stratified reservoirs following drawdown prior to fall turnover. Drawdown increases methane ebullition (pathway #2 in Figure 1) by decreasing hydrostatic pressure, which increases bubble formation and release [17]. For example, in Harsha Lake, a eutrophic Midwestern US reservoir, ebullitive methane emissions increased substantially following an experimental fast drawdown in September, especially at shallower sites [16]. Harsha Lake stratifies from

mid-May to mid-October [16]. Because the pulse was short-lived, it constituted only 3% of emissions measured over a 7-month period [16]. Another study of two reservoirs in the Pacific Northwest produced different results. In these reservoirs, that stratified from June to October, ebullitive emissions during an August drawdown [64], accounted for 90% of annual methane emissions [17]. In one reservoir, the annual mass of dissolved methane,  $CH4_{(i)}^{deep}$ , was modeled [Equation (3)] as a function of duration of antecedent hypoxic conditions and a variable indicating whether water was being spilled from the epilimnion to achieve drawdown [24].

$$CH4_{(t+1)}^{deep} = \alpha + \beta \operatorname{Year} + \begin{cases} \gamma & \text{if drawdown} \\ 0 & \text{otherwise} \end{cases}$$
(3)

Drawdown was found to be a significant predictor in the best-performing models. Differences among years were better explained by water-column stability than by algal production [24]. In years with higher water-column stability (i.e., high resistance to mechanical mixing), oxidants in the water column were depleted and methane produced in sediments was able to accumulate in the hypolimnion without being oxidized [24]. However, we note that unlike the reservoirs studied by Deemer and Harrison, most hydropower reservoirs achieve drawdown through hypolimnetic withdrawals. Drawdown is likely to have different effects on upstream emissions in reservoirs with hypolimnetic withdrawals because methane accumulated in the hypolimnion is removed and instead contributes to downstream degassing emissions.

Methane degassing is directly related to flow rates and partial pressure of methane in the reservoir at the intake depth, which are both influenced by seasonal release patterns [65]. Specifically, hypolimnetic releases from hydropower reservoirs promote thermal mixing and disrupt stratification [66,67]. Operations that promote mixing will lower degassing emissions by bringing oxygenated water down to the sediments. Kemenes et al. [68] reported lower methane degassing fluxes from a Brazilian reservoir during the rainy season when stratification was weak and hypolimnetic methane concentrations were low. Low methane degassing fluxes were also reported from several run-of-river reservoirs in the Columbia Basin, US from summer through fall, possibly because they did not stratify [69].

## 2.1.2. Diurnal Hydropeaking

According to our conceptual model, hydropeaking can have positive or negative effects on methane emissions depending on whether the dominant mechanism is reduced hydrostatic pressure (leading to increased ebullition, pathway #2 in Figure 1) in the draw-down zone or water column mixing (leading to lower methane production) (pathway #1 in Figure 1). Positive effects on ebullition rates mediated by reduced hydrostatic pressure as a result of hydropeaking were reported by one study of reservoirs in the Pacific Northwest [17]. The study compared time series of methane concentrations and emissions in two cascade reservoirs, Keno and J.C. Boyle. Keno feeds into J.C. Boyle, which is a peaking reservoir. Diurnal fluctuations in J.C. Boyle produced pulses of methane ebullition that increased emissions by a factor of 3.5 relative to Keno [17].

On the other hand, over a longer time horizon, volatility in water levels might decrease emissions for several reasons. Repeated ebullition events could deplete sediment methane faster than it is produced so that integrated emissions over time might not be higher under peaking operations, especially in reservoirs with low sediment storage capacity for methane [17].

Secondly, hydropeaking shortens the period of stratification in summer and fall [70,71]. For example, in the Three-Gorges reservoir, increased daily WLF (i.e., peaking operation) weakened stratification from June to late September [72]. This was evidenced by increased thermocline thickness (i.e., penetration of temperature and oxygen to deeper depths) accompanied by decreased thickness of the hypolimnion [72]. A shorter period of stratification reduces methane accumulation prior to fall turnover.

Vertical mixing associated with thermopeaking (fast fluctuations in temperature associated with hydropeaking) in summer reduces water-column stability (stratification) and increases dissolved oxygen in the hypolimnion. This mixing also moderates water temperature, decreasing summer highs and increasing winter lows relative to temperatures under non-peaking operations [70]. Both the moderation of summer temperatures and increased oxygenation of the hypolimnion should inhibit methanogenesis in bottom sediments. Internal waves produced by hydropeaking can enhance vertical mixing by an order of magnitude, decreasing the thickness of the hypolimnion in proportion to the square root of outflow volume [67]. Effects of WLF on vertical mixing are also cumulative under daily peaking operations [67]. Even in non-peaking reservoirs, relatively small fluctuations in flow releases can have significant negative effects on stratification and bottom temperatures [73].

Hydropeaking may also influence methane emissions indirectly through its effects on algal production. Although methane produced from algal production is not counted in net carbon emissions [74], the moderation of temperatures and mixing induced by hydropeaking during summer should reduce methane production stimulated by eutrophication and associated hypoxia (pathway #1 in Figure 1). Hydropeaking has been shown to decrease algal production [75]. Likewise, diurnal disruptions in stratification caused by nighttime decreases in temperature also decrease algal production [48]. However, in eutrophic reservoirs, the redistribution of nutrients from deeper water and bottom sediments may stimulate algal blooms [76] and later increase methane generation in sediments by reducing oxygen levels upon decay [77] (pathway #1 in Figure 1).

The potential effects of diurnal hydropeaking on degassing methane emissions are mediated by the diurnal effects of temperature and pressure on gas saturation and fluctuations in turbine flows. One recent study in a 'double-peaking' tropical reservoir found that diurnal flow peaks corresponded with high  $CO_2$  degassing emissions, termed 'carbopeaking' [78]. During the summer stratification period, when the concentrations of  $CO_2$  in the hypolimnetic waters were supersaturated, these peaks occurred twice daily—once in the morning and once in the evening [78]. If it is driven by saturation, methane may follow similar patterns. In addition to pressure changes, changes in flow produce fluctuations in degassing emissions Table 1, Equation (S10). Hydropeaking increases both the baseline values and diurnal fluctuations in total gas saturation compared with free-flowing rivers [79].

In summary, although hydropeaking may cause short-term pulses of ebullitive emissions and resuspend nutrients, when averaged over longer time periods, we hypothesize that methane emissions will decrease in response to hydropeaking by disrupting stratification and depleting sediment methane.

## 2.2. Drying–Rewetting Cycles

WLFs induce drying-rewetting cycles in the littoral zone that influence methane emissions in complex ways (Figure 1, pathway #2). Sediment drying-rewetting cycles can enhance emissions of methane [80], especially during transitional phases. For example, one study reported that alternate drying and rewetting produced higher methane fluxes at frequently flooded sites compared to rarely flooded and non-flooded sites [81].

A key question in understanding the effects of drying–rewetting cycles is the relative residence time of water in the soil profile compared to the rates of relevant biogeochemical reactions, which are largely controlled by temperature [18,82]. The rate of methane production in sediments [83–85] and diffusive sediment fluxes of methane both increase exponentially with increasing temperature [86]. To better understand how the timing of short-term fluctuations influence methane emissions requires understanding the outcome of coupled methane production and oxidation, which is controlled by temperature-dependent rates of methanogenesis and methanotrophy.

#### 2.2.1. Seasonal Dynamics

A global assessment of reservoir recently highlighted the importance of including drawdown areas in carbon footprints [87]. Methane and *CO*<sub>2</sub> show opposite responses to inundation because one predominates under anoxic conditions and the other under well-oxygenated conditions, In contrast to *CO*<sub>2</sub> emissions, methane emissions decrease as drawdown area increases [87]. However, drawdown areas may have a lower impact at hydropower reservoirs than other types, as the median percentage of area in drawdown is 9%, compared to a median of 17% across all reservoirs [87]. Although global extrapolations have estimated methane emissions by extrapolating from the areal extent of drawdown [87], such footprint approaches (i.e., based on area) do not consider dynamics linking methane emissions to seasonal drawdown and rewetting [7,9]. Some of these mechanisms are outlined below.

Prolonged exposure to the atmosphere during drawdown reduces diffusive methane emissions [88] as oxygen infiltrates soil pores. After drawdown, methanogenesis is suppressed because reservoir sediments in littoral areas are exposed to the atmosphere and/or to water rich in favorable electron acceptors that promote aerobic respiration (and produce  $CO_2$ ). This is why, under warm conditions, dry soils become a sink for atmospheric methane [89].

Prolonged inundation during full-pool conditions leads to build-up of methane in sediments and the water column if methane production rates exceed removals [17,38,90]. For example, in the Three Gorges reservoir, where the drawdown area represents one-third of the surface area, methane emissions were higher during the inundation phase than the drawdown phase [91]. When stratification is prolonged, there is sufficient time for methanotrophic microbial populations to develop [92,93].

Methane build-up can be exacerbated by eutrophic conditions [94]. For example, the increase in emissions between pre- and post-drawdown scaled linearly with the log of chlorophyll-a concentration In eutrophic reservoirs with low turnover rates [17]. Following such a build-up of methane, seasonal drawdown can lead to a substantial increase in ebullition in shallow areas especially as hydrostatic pressure is decreased [16,17,95,96]. In addition, dissolution of methane bubbles during drawdown likely contributes to elevated methane concentrations in the water column [24].

Methane oxidation also varies with water level. Anaerobic methane oxidation is concentrated in the narrow band within the oxycline where oxygen and other electron acceptors (e.g.,  $NO_2$ ) and dissolved carbon co-occur [97,98]. WLF disrupt stratification by promoting mixing, as discussed earlier. Methane production decreases exponentially with sediment depth and is less likely to be oxidized while diffusing from shallow sediments and rising for a short distance through the water column.

To understand methane responses to WLF in reservoirs, models should also account for the effects of water saturation on the availability of sediment carbon. Existing models differ in this respect. For example, the monthly ResME reservoir model accounts for methane production from pools of carbon with different ages and decay parameters [47]. However, it does not represent the effects of rewetting or changes with depth in sediment. WETMETH, a model developed for wetlands, does consider wet–dry cycles [99]. In WETMETH, total methane production,  $P_{(t)}^{sed}$ , is estimated by integrating over methane produced in each depth layer of soil (or sediment) column per unit time. Other 1D biogeochemical models represent the rate of methane production as the product of a Q10 function of temperature and the soil carbon available to anaerobic decomposers. Production can be integrated over wetted sediment layers with mean depths, *z* [Table S1, Equation (S5)] or this can be simplified to represent just one soil layer [100,101].

To represent sediments in reservoir margins (i.e., the littoral zone), linkages between reservoir operations, WLF, and sediment saturation depth must be established. For example, WETMETH demonstrated an increase in methane fluxes when the water table in a wetland was high (Figure 4). If sediments along the edges of reservoirs behave similarly to wetlands, we would expect them to produce higher methane fluxes when inundated (e.g., full pool)

compared to conditions under drawdown conditions (Figure 4). At a given reservoir pool elevation, methane emissions reflect a trade-off between oxidation and methane production, both of which decrease with depth as the percent organic matter and methanogens declines (Figure 4). There is evidence that microorganisms involved in methane production can develop strategies to recover 20–40 d following drying–rewetting events, subject to the availability of organic matter [102]. However, we are not aware of studies of short-term recovery of microbial function in response to sub-daily fluctuations.





#### 2.2.2. Diurnal Hydropeaking

Hydropeaking influences methane emissions by decreasing methane production but increasing ebullition (causal pathway #2 in Figure 1). Most field studies of ebullition have been short-term sampling efforts conducted during summer and fall. The question is, 'How does methane production in sediment in the drawdown zones of peaking projects compare to methane production in reservoirs that do not fluctuate flow releases significantly on a sub-daily scale?' Several hypotheses for linking diurnal WLF to annual methane emissions consider whether sediment erosion, methane production and destruction by microbes, or ebullition play a more-dominant role in shaping the net outcome:

- Short-term drying-rewetting cycles will increase bank erosion and focusing of sediment, thereby depleting carbon in sediment and reducing net emissions, or alternatively;
- 2. Short-term drying–rewetting cycles will maximize production of methane (during wetting) and subsequent opportunities to reach the atmosphere without significant depletion (mineralization) of sediment carbon (during drying), and/or;
- 3. Short-term drying-rewetting cycles will disrupt methanogenesis by disturbing microbial communities, and/or;

4. Short-term drying–rewetting cycles will increase ebullitive emissions due to reduced hydrostatic pressure.

#### 2.3. Sediment Delivery and Redistribution

Reservoir fluctuations are also linked to methane emissions via the delivery and redistribution of carbon-laden sediment (pathway #3 in Figure 1). Consistent with the flood-pulse paradigm in river floodplains [103], WLF promote exchange with adjacent terrestrial ecosystems such as floodplains in reservoir tributaries and bays. This adds nutrients and carbon to the main channel. In addition, methane emissions are more strongly related to transport from shallow areas than to reservoir total surface area [104]. Therefore, the aspect ratio, defined as the ratio of shallow to total surface area, is a better morphometric correlate of methane emissions than reservoir area alone [104].

Trade-offs exist between benefits of sediment flushing to downstream ecosystems and the increased risk of higher methane emissions in upstream reservoirs [105]. Sediment flushing can produce 'hot moments' in methane releases [106]. Specific mitigations targeting sediment supply to downstream ecosystems are required for fewer than 3% of privately owned projects in the US [107]. Yet, mobilizing sediments has both short-term and long-term benefits. In the long-term, geomorphologic processes that shape downstream river channels by creating shallow, slow habitat for aquatic biota depend on sediment scouring [108]. Meanwhile, the supply of sediment-bound nutrients, including phosphorus and silica, is needed to support diatoms and other vegetation in downstream ecosystems [109]. Research is needed to understand trade-offs between upstream methane emissions and benefits to downstream biota associated with the frequency of different magnitudes of flushing events [105].

Erosion and focusing of sediments in deeper areas of the reservoir is another mechanism by which sediment dynamics mediate the relationship between WLF and methane emissions. Drawdown zones of tropical water impoundments lost nearly 40% of organic soils over ~80 years due to erosion [110]. Worldwide, reservoirs are accumulating sediment at a rate of about 1% each year [111]. More specifically, storage is depleting at the rate of 1.1% for reservoirs with storage volumes less than 500 m<sup>3</sup> and 0.6% for larger dams [112]. Sediment accumulating in a reservoir reduces the size of the conservation pool (i.e., volume of water required for the purposes of the reservoir, such as hydropower, water supply) and thus sets the useful lifespan of reservoirs [113,114].

Based on our conceptual model, we expect that causal pathway #3 leads to lower the GHG emissions in reservoirs when drawdown and water-level fluctuations are considered. In general, we expect that WLF will increase the net transport (delivery or redistribution) of sediment from shallow to deeper areas. Erosion and sediment transport from land to deposition zones in reservoir bends and bays produces methane emission hotspots and carbon burial zones. However, WLF mainly affects subsequent redistribution of sediment, rather than its initial delivery after precipitation events. Bank erosion typically removes fine material first (i.e., at a lower critical velocity). Water erosion also redistributes littoral organic sediments away from shallow areas, that are prime locations for methanogenesis and ebullition, to deeper water. Spatial zonation of sediments in reservoirs reveals focusing of deposition in deeper areas; beyond around 20 m, resuspension typically does not occur [30]. Movement of sediment to deeper areas and removal from the littoral zone should therefore result in lower methane emissions along the perimeter of the reservoir.

By concentrating sediments, WLF may create hotspots of methane emissions via ebullition [110]. However, this depends on whether the deposited sediments have sufficiently high organic content at a shallow depth and are deposited at a sufficiently slow rate. Bubble formation tends to occur when organic carbon concentrations exceed 2.4% [36] and redistributed (eroded) soils tend to be low in organic matter [115]. Fast deposition can result in both a high bubble release frequency (e.g., at accumulation rates of 2–2.8 cm y<sup>-1</sup> [36]) and high rates of carbon burial, so that carbon is sequestered below the sediment depth at which microbial methanogens are most active. Estimates of the ratio of emissions to burial have included only CO<sub>2</sub> (not methane). For medium to large reservoirs in the conterminous US, the ratio was estimated to be 0.72 [116] and carbon burial exceed emission rates in most temperate reservoirs [117–119]. By simply removing the area constituting drawdown zones from burial and assigning higher CO<sub>2</sub> emissions, a larger estimate showing emissions in excess of burial was produced for all global reservoirs (not just the subset used for hydropower, which typically have smaller drawdown zones) [87]. An improved conceptual framework is needed to understand how WLF in reservoirs influence the balance between emissions and long-term carbon storage.

# 2.3.1. Seasonal Dynamics (Drawdown)

Generally, we assume that drawdown occurs gradually over a period of multiple weeks, rather than days. The effects of drawdown depend on how fast it occurs. Fast drawdown erodes fine sediments, nutrients, and organic matter from exposed sediments [120]. If seasonal drawdown occurs in a short time period (days to weeks), limiting exposure to oxygen, rapid mobilization and focusing of sediment enhance carbon burial rates [36]. Lateral transport of sediment also contributes to carbon burial. For example, in the Three Gorges reservoir, China, WLF resulted in transport of nutrients and carbon-laden sediments from tributaries and bays into deeper areas of reservoirs during drawdown [121]. Decreases in carbon-rich sediment in the littoral zone can decrease methane emissions, but redistribution of sediment within the reservoir also creates hotspots of methane production below the drawdown zone [36]. Reflooding does not increase carbon emissions after prolonged drawdown [88].

## 2.3.2. Diurnal Hydropeaking

Diurnal fluctuations in water level influence erosion via three mechanisms. First, short-term reservoir fluctuations can cause bank slumping [122]. When sediments do not dry out between drawdown events, slumping can result from the added weight of water in pores [122]. Shoreline erosion depends on the flow recession rate, which is usually faster for diurnal WLF than for drawdown. Stability analysis has shown that rapid decreases in water level can produce landslides near the toe slope when sediments are saturated [123]. Second, wave action is an important factor that can undermine bank stability. Hydropeaking generates waves that increase shoreline erosion. At high latitudes, the scouring effects of diurnal fluctuations in winter are especially erosive if surface ice is moving up and down [79]. Beyond this, seiches with amplified energy are sometimes generated by WLF [124–127]. Third, hydropeaking WLF causes changes in sediment biogeochemistry that can result in lower methane production. Scouring removes smaller grain sizes first and exposes new sediments to subsequent erosion, potentially suppressing methanogenesis. WLF can decrease total carbon and nitrogen in sediment, likely due to increased mineralization and nitrification rates, respectively, a pattern that attenuates with distance above a peaking dam [128]. Repeated drying ensures that carbon in sediments are released primarily as  $CO_2$  rather than methane [129]. Although short-term increases in respiration can occur due to a priming effect upon rewetting, repeated wet-dry cycles substantially lower long-term rates of soil C mineralization [130], presumably due to loss of mineralizable organic matter.

## 2.4. Littoral Vegetation

Causal relationships linking aquatic plants to methane production and emission are complex, and research examining the effects of littoral vegetation on methane emissions in reservoirs is slim, especially compared to studies in wetlands. Vegetation stimulates methane production by adding organic carbon (via senescence/litter inputs) and reducing erosion [131] (sediment stabilization and focusing along causal pathway #3 in Figure 1).

The best-known vegetation-mediated causal pathway is shunting methane from sediment interstices to the atmosphere, Ap, (Figure 2) via plant vascular systems [132,133]. Riparian-adapted species have aerenchyma tissue that allows oxygen transport during long-periods of inundation (anoxic conditions) [134]. Venting of methane is highest when plant transpiration is high and when water levels are high [135]. On the other hand, macrophytes and other vegetation with aerenchyma also reduce methane emissions by leaking oxygen into sediment via roots [136], which reduces methanogenesis and increases methane oxidation by methanotrophs [132]. One study reported a single nighttime peak in methane, with lower daytime emissions due to transport of oxygen to the root zone supporting methanotrophy [133].

WLF can reduce littoral vegetation [137] and we hypothesize that this, in turn, would reduce shunting of methane through aerenchyma by plants, but also decrease oxygen and organic-matter inputs to sediments (Figure 1). Transport varies by macrophyte species, with higher transport in those with a higher density of stems and leaves emerging from the water [138]. In addition, decomposition of leaf litter can be a dominant factor leading to higher littoral emissions from vegetated zones. Some evidence suggests that submerged macrophytes can also produce methane in the presence of oxygen [139], although this is not considered to be a primary pathway.

The complex relationship between WLF and shoreline vegetation effects on methane production and emission should not be used to infer that reservoirs should be managed to decrease vegetation because littoral vegetation provides many positive ecosystem services. These range from stabilizing sediment to providing important nursery habitat for fishes. For this reason, some reservoir managers advocate that drawdown patterns be shifted to overlap with the growing season to benefit vegetation [140]. In shallow waterbodies, two alternative stable states are dominance by macrophytes and dominance by algae [141]. These states are separated by a tipping point mediated by fish predation and turbidity [141]. Attempting to reduce methane emissions through eliminating macrophytes could backfire by increasing algal blooms in smaller reservoirs [142,143], which, when they decompose, can increase methane emissions and degrade fish habitat. The balance among these mechanisms is poorly understood and it is unclear whether they also apply in shallow areas of larger reservoirs.

# 2.4.1. Seasonal Drawdown

At the seasonal scale, both prolonged desiccation and prolonged inundation associated with seasonal drawdown and refill have the effect of reducing littoral vegetation, and presumably alter methane emissions (causal pathway #4 in Figure 1). In the 1980s, reservoir drawdown was studied as a means of controlling macrophytes by the US Army Corps of Engineers [137]. Effective 'control' required dewatering of sediments for at least a month [137]. Prolonged inundation is also problematic for many species of plants that are not adapted to extended wet conditions.

#### 2.4.2. Diurnal Hydropeaking

Although duration of dewatering and inundation regulates the impacts of seasonal drawdown on vegetation, the impacts of hydropeaking on vegetation likely increase with the frequency and magnitude of WLF. Erosion of fine sediments leaves sediment with low cohesion that prevents establishment of rooted plants. As discussed in Section 2.3, erosion and scouring are key mechanisms that act against the persistence of aquatic and shoreline vegetation. In general, native riparian trees and grasses (e.g., willow, *Salix* spp.) are better adapted to persist under the altered disturbance regime created by hydropeaking [144]. Effects of hydropeaking on dispersal, germination, establishment and growth, and reproduction of aquatic plants are summarized by Bejarano, Jansson and Nilsson [79].

### 3. Strategies for Reducing Methane Emissions

Harrison et al. [17] distinguish between a 'forcing-controlled regime' in which WLF can be used to regulate emissions and a 'methane production-controlled regime' in which methane production rates overwhelm the sediment's capacity to modulate emissions. This is an important distinction in deciding which strategies to focus on.

#### 3.1. Flow Management

For 'forcing-controlled' regimes, flow management can play an important role in managing emissions. Flow regimes that limit the duration of stratification or promote periodic mixing events that bring oxygen down to the sediment interface and 'reset' the sediments [17], can help to reduce emissions over an annual time horizon. This assumes that increased bottom temperatures associated with mixing will not stimulate methane production in the presence of oxygen or other terminal electron acceptors. Drawdown during times with a well-oxygenated hypolimnion may partially mitigate ebullitive fluxes that would otherwise result from the decrease in hydrostatic pressure [24]. This can be achieved by delaying seasonal drawdown from the end of the summer until after fall turnover (and reservoir mixing) [17]. Alternatively, drawdown before turnover can be achieved by spilling or generating on surface water, which avoids degassing of potentially methane-rich, hypolimnetic water running through turbines [14].

Research is needed to understand whether methane production in shallow areas could be disrupted by managing the timing and duration of ramping events or by preventing intrusion of dissolved methane from shallow areas in side-channels into the epilimnion of the main channel. The risk of triggering ebullition can presumably be reduced by achieving drawdown over a longer period. Restricting constraints on the up ramping phase at times when sediment methane concentrations are high near the forebay may help to minimize ebullition events.

Causal pathway #1 suggests that summer hydropeaking offers another strategy to reduce annual methane emissions if initial sediment methane levels are low. Mixing induced by WLF can disrupt stratification, which reduces methane production and accumulation in the hypolimnion (Note: this can also be achieved by higher summer flows). Fluctuations also promote erosion and mineralization of carbon in organic sediments by exposing organic sediments to oxygen. Adjustments in diurnal reservoir operation could play a significant role in controlling emissions in reservoirs where sediment storage capacity is high relative to annual methane fluxes [17].

Current patterns of hydropower operation and associated WLF are shifting as hydropower 'fills-in' for variable renewables like wind and solar [9,145]. One diurnal pattern that is emerging is 'double peaking' in the early morning before solar comes online and in the evening after sunset. Seasonal patterns in WLF are also likely to shift in response to patterns of wind and solar generation. Adaptive strategies should therefore consider the GHG implications of expected shifts in the renewable portfolio as well as those that are currently optimal.

## 3.2. Selective Withdrawal/Spill/Forebay Oxygenation

During stratification, withdrawal of water from above the thermocline can prevent passing methane-rich water through the turbine and reduce methane emissions through degassing. Some reservoirs are equipped with selective withdrawal structures that allow intake of water from different depths to control for temperature and dissolved oxygen [146,147]. Seasonal and diurnal peaks in degassing can be mitigated at critical times through selective withdrawal, surface spill, or through forebay oxygenation [148]. To minimize degassing emissions, we suggest that monitoring methane concentrations at the depth of intake(s) in the forebay can help guide the timing of surface releases in real time. This is important because most of these options are costly and minimizing the period of deployment is valuable.

#### 3.3. Sediment Management

Our conceptual model (pathway #3) suggests that intercepting sediment loadings could be used to break the link between sediment delivery and methane emissions from reservoirs. Watershed attributes affect the supply of sediment to reservoirs (prior to its erosion and redistribution). Watershed-based practices for sediment management can provide a wide range of benefits to society [149], including minimizing methane emissions

and prolonging the life of hydropower reservoirs. For example, riparian buffers slow flows, increase soil infiltration, and trap sediments from watershed sources, and have been shown to be methane sinks [150,151]. Although the fate of trapped carbon in such buffers might be groundwater or soil respiration (as  $CO_2$ ), less carbon is likely to be transformed to methane, especially if they are rarely inundated.

Given the importance of shallow areas as sources of methane, research is needed to examine how WLF affects sediment inputs and redistribution, which can have cascading effects on methane production, especially in the littoral zone. This is especially important in reservoirs with an abundance of shallow littoral area within the drawdown zone. Sediment management within reservoirs can influence the amount and spatial distribution of organic carbon. Altering sediment redistribution (pathway #3) can involve construction of internal barriers, flushing, dredging, and the use of sediment bypass structures [152]. Understanding seasonal patterns in sediment methane concentration may also be important when planning seasonal flow regimes.

#### 3.4. Basin-Wide Coordination

Reservoirs are not isolated; they often exist within cascades or complexes that are linked by a common river network. To date, few studies of methane emissions have focused on cascades [17,34]. Improved results can be obtained by coordinating releases among reservoirs to achieve water quality objectives across a river basin. For example, in the Cumberland River basin in the southeast US, control of dissolved oxygen throughout the system is achieved by storing colder water in tributary reservoirs later into the fall [153]. Conversely, releasing cold water from upstream reservoirs earlier in summer can strengthen thermocline stability and reduce mixing in downstream reservoirs [153]. In the Tennessee River basin, US high runoff events that occur too early in summer force premature releases from tributary reservoirs and reduce operational flexibility later in the year. Within the physical constraints of the system, basin-wide coordination could seek to reduce methane emissions by intermittently disrupting prolonged hypoxic conditions in downstream reservoirs.

#### 4. Summary

In this paper, we demonstrated the strength of conceptual modeling as a tool for generating plausible hypotheses and suggesting mitigation strategies. By diagramming four causal pathways, we explored how different patterns of reservoir operation might alter methane emissions. In some cases, our causal diagrams showed clear directional controls on emissions, and in others revealed potential for competing directions of influence depending on conditions that should be explored further.

Research questions identified by our analysis include: (1) How can up-ramping be managed to avoid ebullition during times when risk is high?; (2) Do the effects of peaking operations vary among seasons?; (3) Do different diurnal peaking patterns (e.g., single versus double peaking) have different effects on methane emissions?; (4) Does bank erosion play a significant role in decreasing methane emissions from the drawdown zone?; (5) Are there ways that sediment and vegetation management could be used to moderate methane emissions?; (6) Do flow regimes designed to moderate temperatures and or maintain higher levels of dissolved oxygen effectively lower methane emissions from reservoirs?; and (7) Can the information derived from real-time monitoring of forebay methane concentrations be used to prevent degassing emissions?

In addition to these research questions, we produced a series of hypotheses for how methane emissions could be interrupted along each causal pathway (Section 3). These hypotheses require testing and refinement.

# 5. Future Directions

We currently lack models suitable for predicting the effects of reservoir operations on methane emissions. In this study, a conceptual modeling approach was used to examine

causal linkages. However, other types of models can also be useful. It is impossible to construct a model to serve all purposes, and trade-offs exist among the goals of generality, precision, and realism [154]. At the extreme of high realism and precision, CE-QUAL-W2 is a complex lake model that over time has been used to answer many different questions. Of particular relevance is a version that includes sediment diagenesis [155,156] that represents all electron acceptors in the redox cascade, bacteria, and dynamics of multiple algal species. As such, it requires many input parameters and can only be used for well-studied reservoirs. At the other extreme, simpler models (e.g., ResME) focus specifically on representing carbon dynamics and methane, but they currently do not have a fine-enough temporal resolution to represent responses to short-term (subdaily) WLF. Regardless of complexity, validation of models is a challenge because reservoirs with highly resolved temporal and spatial measurements needed to capture hotspots and moments, and long-term data are not available. These models may also not be suitable for exploring pathways involving sediment transport or vegetation.

We see a need for simpler, targeted models to predict hot spots and hot moments. Model-based indicators can serve as proxies along each of the causal pathways (and subpathways) linking reservoir operation to methane emissions. For example, along the 'water-column mixing' pathway, 'habitat' indicators can be developed based simply on where and for how long thermal and oxygen conditions are suitable for methane production and avoiding oxidation under different reservoir operation scenarios. As one nice example of this in a riverine setting, hot spots and hot moments in the Ogeechee River were well-represented simply by mapping the extent of inundated floodplain area during warm periods [3]. In the case of reservoirs, it is more complex because a sub-daily resolution is needed to match that of reservoir releases (the control variable) and to simulate depth stratification and turnover. An important research question is to discern the overall significance of antecedent conditions (hysteresis in responses to WLF and temporal lags). Suitable conditions for methane emissions are not simply a function of reservoir elevation because sediments may have been previously wetted (or not) and microbial communities take time to recover. Indicators for the 'drying-rewetting' pathway would address these non-linear responses. In contrast, questions about the role of sediment and vegetation might best be explored by comparing reservoirs, although there may also be a need for modeling.

**Supplementary Materials:** The following are available online at https://www.mdpi.com/article/ 10.3390/w15234112/s1, Table S1. Equations associated with the systems diagram shown in Figure 2. System boundaries exclude upstream inflows.

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