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How to Efficiently Produce the Selenium-Enriched Cucumber Fruit with High Yield and Qualities via Hydroponic Cultivation? The Balance between Selenium Supply and CO₂ Fertilization

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Abstract: Hydroponic-producing selenium (Se)-biofortified vegetables in a greenhouse is a convenient and effective way to provide Se-enriched food and overcome hidden hunger. CO₂ fertilization is commonly implemented to increase vegetable yield in greenhouse production. However, this application accompanies decreased mineral concentrations in the edible parts. Here we investigated the effects of [CO₂] and Se supply on the growth, gas exchange, and cucumber fruit quality. A hydroponic experiment with two CO₂ concentrations ([CO₂]) (C1: 410, and C2: 1200 μmol mol⁻¹) and four Se supply levels (Se0: 0, Se1: 0.125, Se2: 0.250, and Se3: 0.500 mg Se L⁻¹) was carried out. A low level of Se supply (Se1: 0.125 mg Se L⁻¹) protected the photosynthetic pigments and stimulated the stomatal opening, especially under [CO₂] fertilization. It leads to a higher net photosynthesis rate (Pn) and transpiration rate (Tr) than other Se treatments. The most significant changes in dry weight, fruit yield, and soluble sugar concentration were also obtained in Se1 under CO₂ fertilization due to the enhanced CO₂ fixation. Meanwhile, the Se concentration in fruit was 0.63 mg kg⁻¹ FW in C2Se1, with the highest Se accumulation and use efficiency. According to the recommended dietary allowance of 55 μg Se day⁻¹ for adults, an intake of 87 g of cucumber grown in C2Se1 is sufficient. Because of the improved Tr and better root structure in Se1, the uptake of mineral nutrients through mass flow and interception was well maintained under CO₂ fertilization. So, the concentrations of N, P, K, Ca, and Mn in cucumber fruits were not significantly decreased by elevated [CO₂] in Se1. However, the concentrations of soluble proteins, S, Mg, Fe, and Zn in cucumber fruits in C2Se1 were lower than those in C1Se1, which was mainly attributed to the dilution effects under CO₂ fertilization. Therefore, a selenite supply of 0.125 mg Se L⁻¹ was found to be the optimal dosage for producing Se-enriched cucumber fruits with high yield and better qualities under CO₂ fertilization (1200 μmol mol⁻¹).

Keywords: crude fiber; photosynthetic pigment; recommended daily intakes; soluble protein; soluble sugar; vitamin C



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1. Introduction

As an essential trace micronutrient, selenium (Se) is a critical component of seleno-amino acids and selenoproteins, which play a crucial role in animals, including humans [1, 2]. Se deficiency is a severe health problem that could cause many human diseases, such as Kashin–Beck disease, Keshan disease, and so on [3,4]. Recent studies also demonstrated that a complement of Se may help reduce the incidence and severity of various viral infections, including severe acute respiratory syndrome coronavirus 2 (SARS-CoV-2), which caused

the global pandemic of coronavirus disease 2019 (COVID-19), due to its anti-inflammatory, antioxidant, and immune-boosting effects [5–7]. The primary source of Se for humans is derived along the food chain from edible plants, and the Se concentration ([Se]) in plants largely depends on the Se content and bioavailability in the soil where the plants grow [8,9]. Therefore, more than one billion people living in the Se deficiency area suffer the long-term deficiency of this essential element [9,10]. Among the methods to meet the dietary demand for Se for these people, biofortification of agricultural products is one of the most economical and safe approaches to providing Se-enriched food and overcoming hidden hunger [11,12].

Vegetables are an essential component of the daily diet [13,14], accounting for nearly one-third of dietary intake and about 10% of Se intake [15]. In food-based dietary guidelines from the World Health Organization (WHO) and 90 countries globally, an intake of 200–250 g of vegetables per person per day is usually recommended for health [16]. For example, in a questionnaire of 143,305 participants from 18 countries on five continents, the average vegetable intake per capita was 2.19 servings per day (95% CI 2.13–2.25) [17]. Similarly, vegetable consumption per capita ranged from 140 to 250 g per day, based on the data from representative surveys in 19 European countries [18]. Hydroponics can efficiently increase the concentration of essential and/or beneficial micronutrients due to biofortification in edible plant organs, promoting human health [19]. Meanwhile, Se-biofortified cucumber (*Cucumis sativus* L.), tomato (*Solanum lycopersicum* L.), cabbage (*Brassica oleracea* L.), lettuce (*Lactuca sativa* L.), chard (*Beta vulgaris* L.), parsley (*Petroselinum crispum* L.), chicory (*Chicorium intybus* L.), and spinach (*Spinacia oleracea* L.) have already been produced by root fertilization or foliar spray with selenite or selenate solutions [20–24]. Accordingly, intake of these Se-biofortified vegetables is a convenient and effective way to match the recommended dietary allowance (RDA) of 55 $\mu\text{g Se day}^{-1}$ for adults, as recommended by the WHO [25,26].

Moreover, the biofortification of vegetables with Se has at least three advantages. Firstly, vegetables are commonly eaten fresh, avoiding the Se lost in the cooking process [27,28]. Secondly, most vegetables can be produced using a hydroponic cultivation system, which brings higher Se utilization efficiency and bioavailability, meanwhile eliminating the adsorption of Se by the Al- and Fe-oxides and organic matters in soils [8,9]. Thirdly, the [Se] in the vegetables could be precisely controlled in a hydroponic system without exceeding the toxic threshold for plants and humans [2,29]. Therefore, producing Se-enriched vegetables is a convenient and reliable way to provide Se-biofortified foods.

On the other hand, CO₂ fertilization is commonly implemented to increase vegetable yields in greenhouse production [30,31]. However, many studies showed a decrease in vegetable mineral concentrations under CO₂ fertilization, resulting from the dilution effects, decreased mass flow and transpiration, and so on [32–34]. To date, studies dealing with the effect of CO₂ fertilization on the [Se] in plants are rare. Moreover, the application rate of Se also had significant impacts on the leaf pigments, the gas exchange parameters, and the plant's mineral nutrient uptake [35,36]. Therefore, it is essential to find the optimum dosage of Se supplement to produce Se-enriched vegetables with high yield and qualities under CO₂ fertilization.

Based on the previous work, the growth of cucumber seedlings was significantly inhibited when treated with a selenite concentration beyond 10 $\mu\text{mol L}^{-1}$ (0.79 mg Se L^{−1}) for 14 days [36]. We also found no cucumber fruits were harvested in the treatments with a selenite concentration beyond 1.00 mg Se L^{−1} with a longer treatment period (70 days) in a pre-experiment. Therefore, a hydroponic experiment on cucumber plants with a whole growth period of 70 days after transplanting (DAT) was carried out, with two CO₂ concentrations ([CO₂]) (410 and 1200 $\mu\text{mol mol}^{-1}$) and four Se supply levels (0, 0.125, 0.250, and 0.500 mg Se L^{−1}). The objectives of the present study were: (1) to investigate the effects of [CO₂] and Se supply levels on the growth and gas exchange parameters of cucumber plants; (2) to characterize the nutritional qualities of cucumber fruits under different [CO₂] and Se supply levels, including the [Se], organic and inorganic nutrients;

and (3) to determine the optimal dosage of Se supply for producing Se-enriched cucumber fruits with high yield and qualities under CO₂ fertilization.

2. Materials and Methods

2.1. Plant Material and Experimental Design

The hydroponic experiment on cucumber plants was carried out in four open-top chambers (OTCs) at the Institute of Soil Science, Chinese Academy of Sciences, P.R. China (32.0596° N, 118.8050° E), which had been used in our previous study [37]. The experiment used a split-plot design, in which [CO₂] was designed as the primary treatment and Se supply levels were the sub-plot treatment. Seeds of cucumber (*Cucumis sativus* L.) cultivar 'Jinmei 3' (Tianjin Kerun Cucumber Research Institute, China) were germinated in a growth chamber at a temperature of 28 °C and a relative humidity of 70% and then sown into trays containing a peat-vermiculite mixture (2:1, *v/v*). After ten days, 64 healthy seedlings of similar size were transplanted into 2.5 L polyvinyl chloride polymer (PVC) pots, with one seedling per pot. Each pot was filled with 2 L Yamazaki nutrient solution for cucumbers [38], and then sixteen of them were moved into each OTC. The nutrient solution was composed of (mg L⁻¹): Ca(NO₃)₂·4H₂O (826), KNO₃ (607), MgSO₄·7H₂O (493), NH₄H₂PO₄ (115), Na₂Fe-EDTA (29.27), H₃BO₃ (2.86), MnSO₄·4H₂O (2.03), ZnSO₄·7H₂O (0.22), CuSO₄·5H₂O (0.08), and (NH₄)₆Mo₇O₂₄·4H₂O (0.02) [37]. After that, the [CO₂] and Se treatments began. Four Se levels (Se0: 0, Se1: 0.125, Se2: 0.250, and Se3: 0.500 mg Se L⁻¹) were set by adding sodium selenite (Sigma-Aldrich) into the nutrient solution with four replicates in each OTC. The [CO₂] in two OTCs was set at 410 µmol mol⁻¹ (C1), and the other two were set at 1200 µmol mol⁻¹ (C2) and were controlled by an infrared gas analyzer (Ultramat 6, Siemens, Munich, Germany) from 8:00 am to 5:00 pm every sunny day. All pots were aerated intermittently by a pump every 30 min per hour, and the nutrient solution was renewed every week. The pots were also rotated within and among OTCs every two weeks to reduce the chamber effects. After 21 days of treatment (DOT), half of the seedlings were harvested, and the other half were transplanted into 5 L PVC pots with 4 L nutrient solution. All the plants were harvested on the 70 DOT. During the 70 DOT, the accumulated [CO₂] treating period was 372 h, and the average [CO₂] was 412.1 ± 4.7 (mean ± sd) and 1196.2 ± 8.3 µmol mol⁻¹ in C1 and C2 treatment during the treating period, respectively.

2.2. Sampling and Measurements

2.2.1. Gas-Exchange Rate Measurements

Cucumber plants' gas exchange properties were measured the day before harvesting (69 DOT). The third leaves from the top of cucumber plants were chosen for the measurement of net photosynthetic rate (P_n), transpiration rate (Tr), and stomatal conductance (G_s) by a portable photosynthesis system (Li-6400, Li-Cor Inc., Lincoln, NE, USA). Four replicates in each treatment were used for measurements. The standard leaf chamber (2 cm × 3 cm) (6400-02B) was used, and the photosynthetic photon flux density was set at 1500 mol m⁻² s⁻¹ by a LED light source. The [CO₂] of the flow-in air was set at the same level as the [CO₂] treatment where the plant was grown. The temperature, relative air humidity, and air flow rate inside the leaf chamber were set at 25 °C, 50%, and 500 µmol s⁻¹, respectively.

2.2.2. Plant Harvest and Weight Determination

All the plants harvested on the 70 DOT were separated into roots, stems, leaves, and fruits and washed with tap water, followed by distilled water. First, the fresh weight (FW) of each part was recorded. Then a portion of fresh leaves and fruits was used immediately to determine the photosynthetic pigment and organic nutrient concentrations, respectively. Next, fresh roots, stems, and leaves were dried at 105 °C for 30 min and then at 75 °C to a constant weight in an electro-thermostatic oven. Finally, the fresh fruits were cut into pieces and frozen immediately, and then lyophilized to obtain the dry samples.

2.2.3. Leaf Pigment Concentration Determination

The photosynthetic pigment concentrations in leaves, including chlorophyll a (Chl a), chlorophyll b (Chl b), and carotenoids in leaves, were determined by the method described by Linchenthaler and Wellburn [39]. Briefly, 0.2 g of fresh leaves was homogenized and extracted with 25 mL of 80% (*v/v*) acetone. Afterwards, the extracts were centrifuged at $10,000 \times g$ for 5 min. The absorbance of the resulting supernatant was then measured at 646, 663, and 470 nm using a microplate spectrophotometer (EPOCH, BioTek Instruments, Inc., Winooski, VT, USA). The Chl a, Chl b, and carotenoid concentrations were calculated according to the formulas given by Lichtenthaler and Wellburn [39].

2.2.4. Fruit Organic Nutrient Concentration Determination

The soluble sugar concentrations in cucumber fruits were determined by the phenol-sulfuric acid method [40]. Briefly, 0.2 g pieces of fresh fruits were homogenized and extracted with 10 mL of hot water (95 °C) twice. Then, the extracts were combined and diluted to 100 mL, and 1 mL of 9% phenol solution and 5 mL of concentrated sulfuric acid were added to 2 mL of the sample solution in turn. The mixture was shaken for 5 min and placed for 25 min, and then the absorbance of the final solution was measured at 485 nm by a microplate spectrophotometer (EPOCH, BioTek Instruments, Inc., Winooski, VT, USA).

The soluble protein concentrations in cucumber fruits were determined by the Coomassie brilliant blue method [41]. Briefly, 0.5 g pieces of fresh fruits were homogenized in 10 mL of distilled water and then centrifuged at $4000 \times g$ for 20 min. Next, 1 mL of the supernatant was mixed with 5 mL of Coomassie brilliant blue G-250. The mixture was shaken for 30 s and placed for 2 min, and then the absorbance of the final solution was measured at 595 nm by a microplate spectrophotometer (EPOCH, BioTek Instruments, Inc., Winooski, VT, USA).

The crude fiber concentrations in cucumber fruits were determined by the acid detergent method [42]. Briefly, 2.0 g pieces of fresh fruits were refluxed in 100 mL acid detergent solution (adding 20 g cetyltrimethylammonium bromide (CTAB) into 1 L 0.5 mol L⁻¹ sulfuric acid) for 1 h. Next, the residue was washed with boiled water three times until the pH of the eluate was neutral, and then the residue was rewashed with acetone three times until the eluate was colorless. Finally, the residue was dried at 100 °C for 5 h, and the DW was recorded.

The vitamin C concentrations in cucumber fruits were determined by the colorimetric method using the 2,6-dichlorophenol method [43]. Briefly, 2.0 g of fresh fruits was ground, extracted with 100 mL of 1% oxalic acid extraction solution, and then filtered. Next, 4 mL of the extract solution was mixed with 2 mL of dye solution (250 mg L⁻¹ 2,6-dichlorophenol indophenol and 205 mg L⁻¹ sodium bicarbonate) and 5 mL of xylene. The mixture was shaken for 30 s and placed for 5 min, and then the absorbance of the xylene phase was measured at 500 nm by a microplate spectrophotometer (EPOCH, BioTek Instruments, Inc., Winooski, VT, USA).

The soluble sugars, protein, and crude fiber concentrations were expressed as milligrams per gram of FW, and the vitamin C concentrations were expressed as milligrams per 100 g of FW.

2.2.5. Fruit Inorganic Mineral Concentration Determination

Lyophilized fruit samples (0.2 g) were digested in H₂SO₄–H₂O₂ at 180 °C for 5 h and measured for nitrogen concentration ([N]) by a discrete auto-analyzer (Smartchem200, Alliance, France) [37]. Another portion of lyophilized fruit samples (0.3 g) was digested in HNO₃–HClO₄ (85:15 *v/v*) at 190 °C for 6 h and used for the concentration determination of phosphorus ([P]), potassium ([K]), calcium ([Ca]), magnesium ([Mg]), sulfur ([S]), iron ([Fe]), manganese ([Mn]), and zinc ([Zn]) by an inductively coupled plasma atomic emission spectrometer (Iris-Advantage, Thermo Elemental, Franklin, MA, USA) [44]. The third portion of the lyophilized fruit sample (0.25 g) was digested in 10 mL of HNO₃–HClO₄ (ultra-pure grade, 9:1 *v/v*) at 160 °C for 6 h, and then 5 mL of 6 mol L⁻¹ hydrochloric acid

was added to maintain the temperature at 120°C until the yellow color disappeared. The digested solution was diluted to 25 mL by ultra-pure water, and [Se] was measured by an atomic fluorescence spectrometer (AFS-8520, Haiguang Co., Ltd., Beijing, China) [45]. A certified reference material (GBW10014/GSB-5; cabbage leaf) was used for method validation.

Because the [Se] in the cucumber fruits grown in Se0 (without Se supply) was below the detection limit, the [Se] and accumulation in Se0 were not given, and Se use efficiency (SeUE) was calculated as follows:

$$\text{SeUE (\%)} = \text{Se accumulation in fruits (mg)} / \text{Se added in the nutrient solutions (mg)} \times 100\%$$

2.3. Statistical Analysis

All data were shown as the mean \pm standard error of four replicates. The means of DWs, gas exchange properties, leaf pigment concentrations, fruit organic and inorganic nutrient concentrations, and the Se concentrations and accumulations in each treatment were compared using the Fisher LSD test ($p < 0.05$) using the application “Paired Comparison Plot” in OriginPro (Version 2021; OriginLab Corp., Northampton, MA, USA). The effects of [CO₂], Se supply levels, and their interactions were quantified using a two-way analysis of variance (ANOVA) in OriginPro. Heat maps displaying fold-changes in the mineral element concentrations in cucumber fruits relative to the mean value of the C1Se0 treatment were generated and clustered according to the Euclidean distance using the application “Heat Map with Dendrogram” in OriginPro. Pearson correlations between [CO₂] level, Se supply level, gas exchange property, and Se concentration, organic and mineral concentrations in cucumber fruits under different [CO₂] and Se supply levels were generated using the application “Correlation Plot” in OriginPro. The variation of gas exchange property, and Se concentration, organic and mineral concentrations in cucumber fruits in response to [CO₂] and Se supply levels were quantified and visualized using the application “principal component analysis” (PCA) in OriginPro.

3. Results

3.1. Dry Weight of Cucumber Plant

Se supply levels had significant effects on the DW of the roots, stems, leaves, fruits, and the whole cucumber plants, as well as the root-to-shoot ratios (R/S), whereas [CO₂] levels only significantly impacted the DW of leaves, fruits, the whole plants, and R/S (Figure 1). Under ambient [CO₂] (C1), the largest DW of root, stem, and leaf was obtained in Se1, Se2, and Se0 treatments, respectively, whereas the largest DW of fruit and whole plants were found in Se0 and Se1 treatments. Under CO₂ fertilization (C2), Se1 treatment always produced the highest DW, and the DW was decreased as the Se supply increased from 0.125 (Se1) to 0.500 (Se3) mg Se L⁻¹. Compared with Se1, the DW of roots, stems, leaves, fruits, and whole cucumber plants significantly decreased by 36.5%, 44.7%, 37.3%, 75.8%, and 53.3% in Se3, respectively. High levels of Se supply (Se2 and Se3) inhabited the DW formation under both ambient and elevated [CO₂], and this inhabitation was more severe in leaf and fruit, resulting in a larger R/S in Se3. On the other hand, CO₂ fertilization increased the DW formation in the above-ground parts of cucumber plants, especially in the Se1 treatment, so a significant decrease in R/S of 31.9% was found in Se1 by CO₂ fertilization.

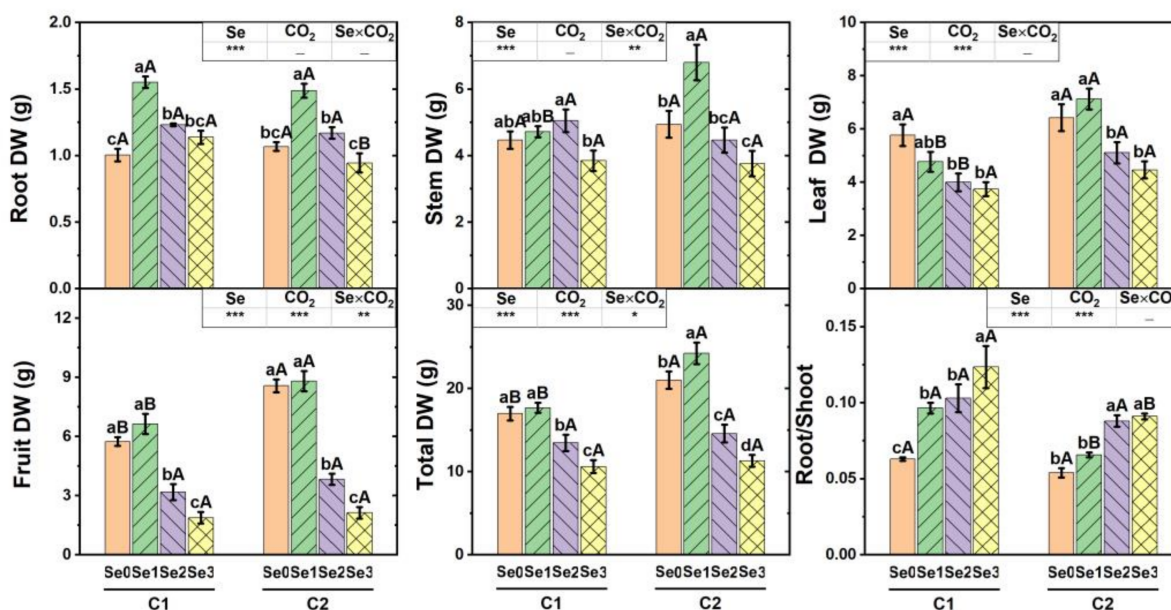


Figure 1. The root, stem, leaf, fruit, and whole plant dry weight (DW), and the root/shoot ratio of cucumber plants grown under different [CO₂] and Se supply levels ($n = 4$). Bars represent standard errors. [CO₂] levels: C1: ambient [CO₂] (412 $\mu\text{mol mol}^{-1}$); C2: elevated [CO₂] (1196 $\mu\text{mol mol}^{-1}$). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and 0.500 mg Se L⁻¹. Means not followed by the same lower-case letters are significantly different among different Se supply levels in the same [CO₂] level, and not followed by the same upper-case letters are significantly different between different [CO₂] levels in the same Se supply level, according to Fisher LSD test at $p < 0.05$. In the internal table, CO₂: [CO₂] level; Se: Se supply level. Asterisks (*) indicate significant differences (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$); – indicates non-significant differences ($p \geq 0.05$).

3.2. Gas Exchange Property and Pigment Concentration in Cucumber Leaf

Gas exchange properties of cucumber plants grown under different [CO₂] and Se supply levels were investigated. As shown in Figure 2, both [CO₂] and Se supply levels had significant effects ($p < 0.01$) on the Pn, Gs, and Tr. CO₂ fertilization significantly increased the Pn of cucumber plants by 49.6% and 61.4% in Se0 and Se1 treatments, respectively. But CO₂ fertilization also significantly decreased the Gs and Tr by 31.4% and 32.7% in Se0, respectively. However, the Gs and Tr were not significantly decreased by CO₂ fertilization in Se1 and Se2 treatments. Under ambient [CO₂], the highest Gs and Tr were found in Se1, but there were few differences in Pn among the four Se supply levels. Under CO₂ fertilization, the highest Pn, Gs, and Tr were obtained in Se1, and the increase was 19.6%, 64.6%, and 57.6% compared with those without Se addition (Se0), respectively. Compared with those in Se1, the Pn, Gs, and Tr were significantly decreased by 37.9%, 64.0%, and 44.7%, respectively, in Se3.

The photosynthetic pigment concentrations in cucumber leaves under the corresponding [CO₂] and Se supply levels were also analyzed (Figure 2). Both [CO₂] and Se supply levels had significantly impacted the Chl a and b concentrations, but neither influenced carotenoid concentration. The Chl b and carotenoids concentrations under ambient [CO₂] and the Chl a and b concentrations under elevated [CO₂] were all lower at a high level of Se supply (Se3) than those in Se0 or Se1 treatments. From Se1 to Se3, the decrease was 21.9% and 22.1% in Chl b and carotenoid concentrations under ambient [CO₂]. Similarly, the decrease was 23.5% and 33.3% in Chl a and b concentrations under elevated [CO₂]. CO₂ fertilization increased the Chl a concentration in Se0 and Se1 treatments and the Chl b concentration in Se1 and Se2 treatments but had no significant effect on all photosynthetic pigment concentrations in Se3 or on carotenoids concentrations in all Se supply treatments.

Specifically, CO₂ fertilization increased the Chl a and b concentrations by 17.6% and 20.0% in Se1, respectively.

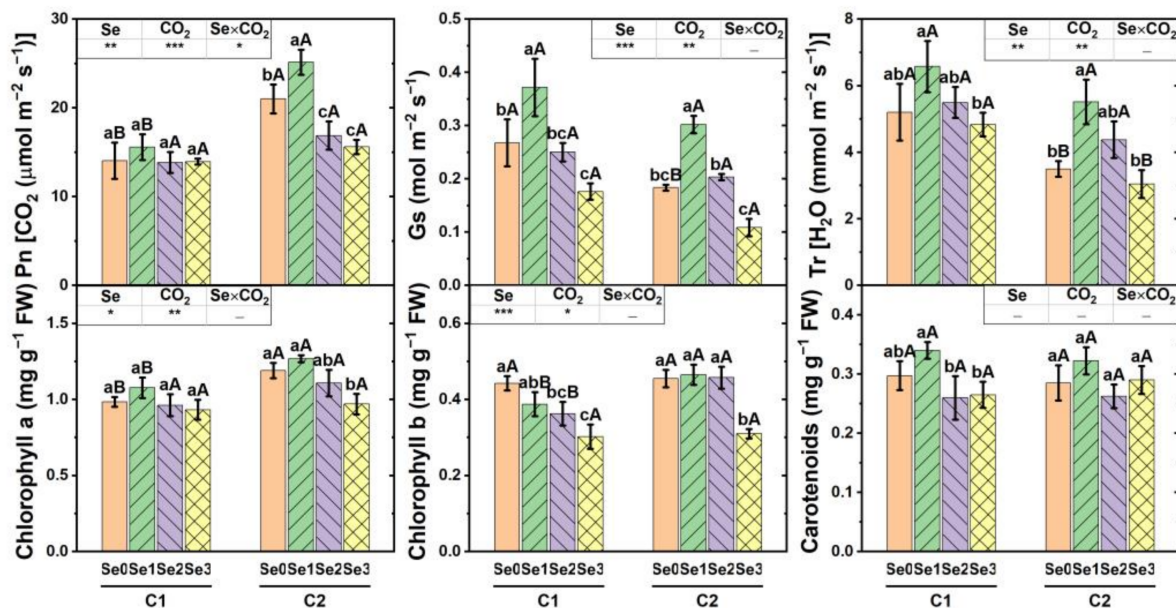


Figure 2. The net photosynthesis rate (Pn), stomatal conductance (Gs), transpiration rate (Tr), and photosynthetic pigment concentrations of cucumber leaves under different [CO₂] and Se supply levels ($n = 12$ for gas exchange properties and $n = 4$ for pigment concentrations). Bars represent standard errors. [CO₂] levels: C1: ambient [CO₂] (412 $\mu\text{mol mol}^{-1}$); C2: elevated [CO₂] (1196 $\mu\text{mol mol}^{-1}$). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and 0.500 mg Se L⁻¹. Means not followed by the same lower-case letters are significantly different among different Se supply levels in the same [CO₂] level, and not followed by the same upper-case letters are significantly different between different [CO₂] levels in the same Se supply level, according to Fisher LSD test at $p < 0.05$. In the internal table, CO₂: [CO₂] level; Se: Se supply level. Asterisks (*) indicate significant differences (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$); – indicates non-significant differences ($p \geq 0.05$).

3.3. Yield and Se Biofortification of Cucumber Fruit

As shown in Figure 3, [CO₂], Se supply levels, and their interaction all significantly affected the cucumber yield per plant. CO₂ fertilization dramatically increased the yield by 32.7% and 37.4% in Se0 and Se1 treatments, respectively, but did not significantly change the yields in Se2 and Se3 treatments. Adding 0.125 mg Se L⁻¹ in the nutrient solution (Se1) brought the highest cucumber yields of 165 and 227 g plant⁻¹ under ambient and elevated [CO₂], respectively. More Se supply resulted in a significant reduction in yield regardless of [CO₂] levels. The cucumber yield was significantly decreased by 52.9% and 73.5% in Se2 and Se3 under ambient [CO₂] compared with that in Se1, respectively, and decreased by 42.2% and 68.8% under elevated [CO₂], respectively.

[Se] in cucumber fruits was remarkably increased as the Se supply increased and was increased from 17.64 mg kg⁻¹ DW in Se1 to 59.18 mg kg⁻¹ DW in Se3 under ambient [CO₂], and was increased from 16.09 mg kg⁻¹ DW in Se1 to 41.57 mg kg⁻¹ DW in Se3 under elevated [CO₂]. CO₂ fertilization significantly decreased the [Se] in fruits by 29.8% only in Se3. The accumulation of Se in fruit was not influenced by either the Se supply levels under ambient [CO₂] or the [CO₂] levels in all three Se treatments. Se accumulation was significantly decreased by 37.2% from Se1 to Se3 under elevated [CO₂]. SeUE was dramatically decreased as the Se supply increased, whereas [CO₂] levels had little effect on it. Compared with Se1, SeUE in Se3 was significantly decreased by 76.0% and 84.3% under ambient and elevated [CO₂], respectively.

Based on the [Se] in cucumber fruits on a DW basis and the water content of fruits, the [Se] in cucumber fruits on an FW basis was calculated (Table 1). Similar to the [Se] on a DW basis, the [Se] on an FW basis was also increased as the Se supply increased, ranging from 0.630 to 2.163 mg kg⁻¹ FW. According to the RDA for adults set by the WHO, an intake of 77~87 g cucumber grown in Se1 or 25~37 g cucumber grown in Se3 every day could meet the RDA of 55 µg Se day⁻¹. Moreover, consuming less than 560~635 g cucumber grown in Se1 or 185~270 g cucumber grown in Se3 per day could ensure the daily intake of Se is below the tolerable upper intake level (UL) of 400 µg Se day⁻¹.

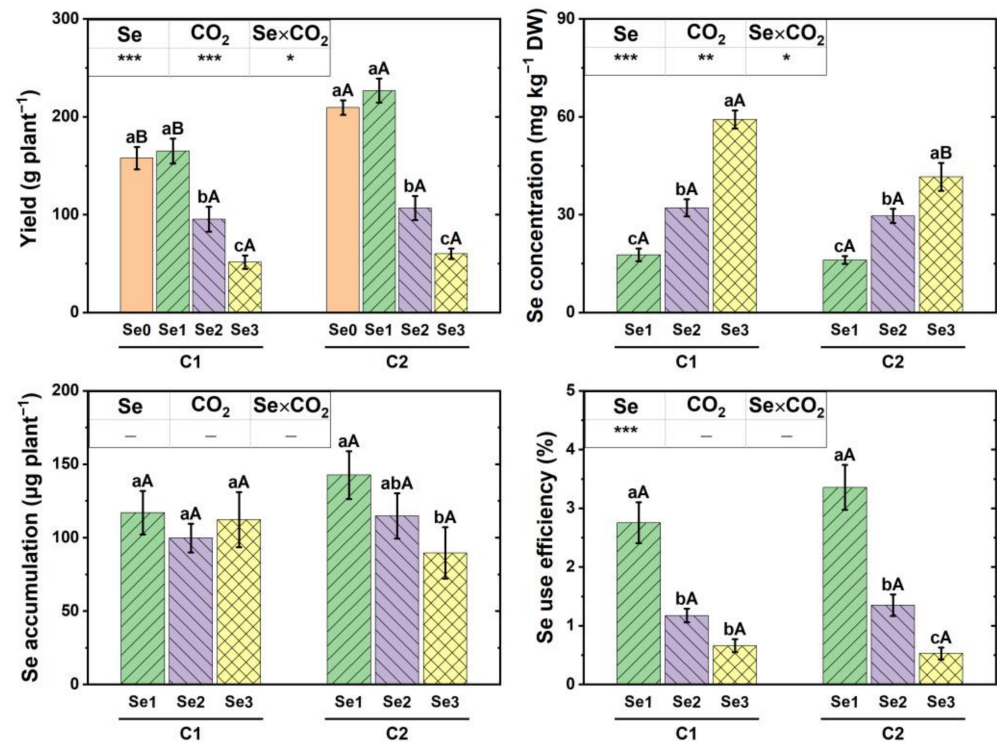


Figure 3. The yield, Se concentration, Se accumulation, and Se use efficiency of cucumber fruits under different [CO₂] and Se supply levels ($n = 4$). Bars represent standard errors. [CO₂] levels: C1: ambient [CO₂] (412 µmol mol⁻¹); C2: elevated [CO₂] (1196 µmol mol⁻¹). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and 0.500 mg Se L⁻¹. Means not followed by the same lower-case letters are significantly different among different Se supply levels in the same [CO₂] level, and not followed by the same upper-case letters are significantly different between different [CO₂] levels in the same Se supply level, according to Fisher LSD test at $p < 0.05$. In the internal table, CO₂: [CO₂] level; Se: Se supply level. Asterisks (*) indicate significant differences (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$); – indicates non-significant differences ($p \geq 0.05$).

Table 1. The Se concentration of cucumber fruits on a fresh weight (FW) basis, and the recommended daily allowances (RDA: 55 µg Se day⁻¹) and the tolerable upper intake levels (UL: 400 µg Se day⁻¹) of fresh cucumber fruits produced under different [CO₂] and Se supply levels.

| Treatments ¹ | [Se] (mg kg ⁻¹ FW) | RDA (g day ⁻¹) | UL (g day ⁻¹) |
|-------------------------|-------------------------------|----------------------------|---------------------------|
| C1Se1 | 0.714 | 77.03 | 560.2 |
| C1Se2 | 1.067 | 51.55 | 374.9 |
| C1Se3 | 2.163 | 25.43 | 184.9 |
| C2Se1 | 0.630 | 87.30 | 634.9 |
| C2Se2 | 1.073 | 51.26 | 372.8 |
| C2Se3 | 1.480 | 37.16 | 270.3 |

¹ [CO₂] levels: C1: ambient [CO₂] (412 µmol mol⁻¹); C2: elevated [CO₂] (1196 µmol mol⁻¹). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, 0.500 mg Se L⁻¹.

3.4. Organic Nutrients in Cucumber Fruit

Se supply levels had significant effects on the soluble sugar, soluble protein, crude fiber, and vitamin C (VC) concentrations in cucumber fruits, whereas $[\text{CO}_2]$ levels only greatly impacted the soluble sugar and soluble protein concentrations (Figure 4). A high level of Se supply (Se3) inhibited the soluble sugar accumulation in cucumber fruits. Compared with the highest soluble sugar concentration in Se2, the decrease was 36.4% and 33.8% in Se3 under ambient and elevated $[\text{CO}_2]$, respectively. CO_2 fertilization significantly increased the soluble sugar concentration by 21.6% and 16.4% in Se0 and Se1 treatments, respectively, but did not have significant changes in the soluble sugar in Se2 and Se3 treatments. There were no significant differences in soluble protein concentrations in cucumber fruits among four Se supply levels under ambient $[\text{CO}_2]$. However, the soluble protein concentration in Se3 was 33.9% lower than that in Se1 under elevated $[\text{CO}_2]$. The crude fiber concentrations in cucumber fruits significantly decreased as the Se supply increased, and the decrease was 42.0% and 55.7% from Se0 to Se3 under ambient and elevated $[\text{CO}_2]$, respectively. CO_2 fertilization only increased the crude fiber concentration by 29.1% in the Se0 treatment. The highest VC concentration in cucumber fruits was obtained in Se2 treatment under both ambient and elevated $[\text{CO}_2]$. VC concentrations in cucumber fruits were not influenced by the $[\text{CO}_2]$ levels, regardless of Se supply levels.

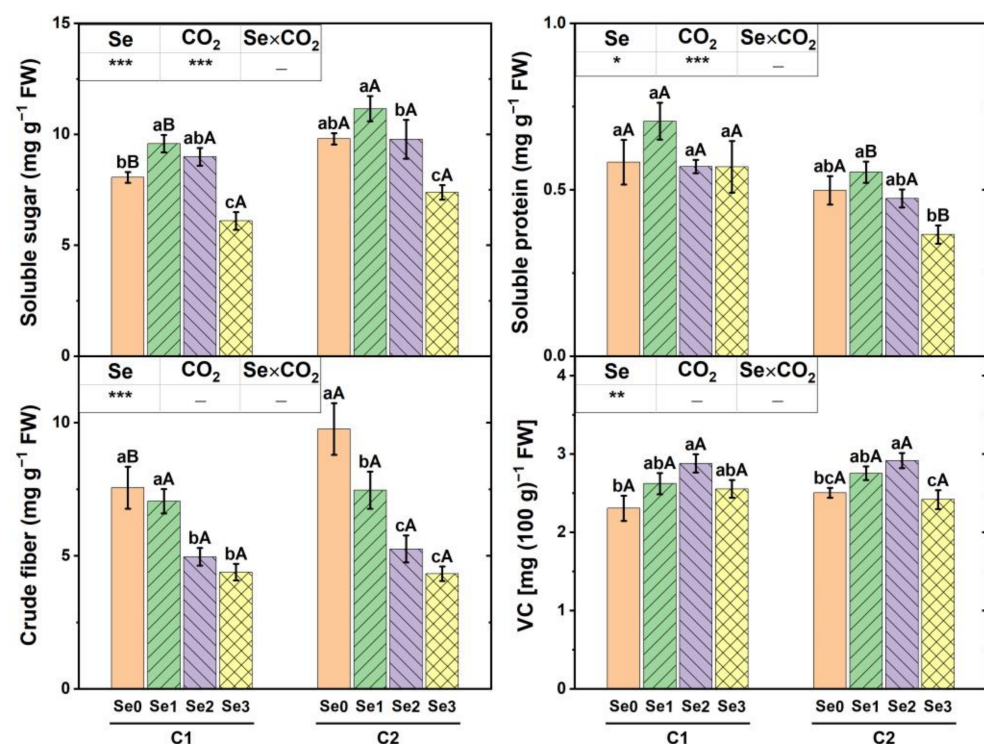


Figure 4. The soluble sugar, soluble protein, crude fiber, and vitamin C (VC) concentrations in cucumber fruits under different $[\text{CO}_2]$ and Se supply levels ($n = 4$). Bars represent standard errors. $[\text{CO}_2]$ levels: C1: ambient $[\text{CO}_2]$ ($412 \mu\text{mol mol}^{-1}$); C2: elevated $[\text{CO}_2]$ ($1196 \mu\text{mol mol}^{-1}$). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and $0.500 \text{ mg Se L}^{-1}$. Means not followed by the same lower-case letters are significantly different among different Se supply levels in the same $[\text{CO}_2]$ level, and not followed by the same uppercase letters are significantly different between different $[\text{CO}_2]$ levels in the same Se supply level, according to Fisher LSD test at $p < 0.05$. In the internal table, CO_2 : $[\text{CO}_2]$ level; Se: Se supply level. Asterisks (*) indicate significant differences (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$); — indicates non-significant differences ($p \geq 0.05$).

3.5. Inorganic Minerals in Cucumber Fruit

As shown in Table 2, the $[\text{CO}_2]$ levels had significant effects on [N], [P], [K], [Mg], [S], [Fe], [Mn], and [Zn] in cucumber fruits, whereas the Se supply levels had significant effects

on [Ca], [Mg], [Mn], and [Zn]. According to the cluster analysis in Figure 5, the above nine mineral elements could be divided into six groups. [N] and [P] in cucumber fruits were not influenced by Se supply levels under ambient and elevated [CO₂] and did not have a decreasing trend by CO₂ fertilization. Se supply levels did not change [K], [Mg], [S], and [Fe] in cucumber fruits under ambient [CO₂] but gradually increased them as the Se supply increased under elevated [CO₂]. The increase of [K], [Mg], [S], and [Fe] in cucumber fruits was 25.2%, 15.4%, 12.0%, and 31.1% from Se0 to Se3 under elevated [CO₂], respectively. CO₂ fertilization dramatically decreased [K] in Se0; [S] in Se0 and Se1; [Mg] and [Fe] in Se0, Se1, and Se2, respectively. High levels of Se supply (Se3) inhibited [Mn] accumulation in cucumber fruits under both ambient and elevated [CO₂], and the decrease was 27.0% and 36.0% from Se0 to Se3, respectively. CO₂ fertilization decreased [Mn] by 22.8% in Se2. [Zn] in cucumber fruits was decreased by adding Se into the nutrient solutions under ambient [CO₂], with a decrease ranging from 14.4% to 18.4% compared with Se0. CO₂ fertilization significantly decreased [Zn] in all four Se levels. The decrease was 28.5%, 23.9%, 17.0%, and 26.8% in Se0, Se1, Se2, and Se3, respectively. In contrast to the other eight mineral elements, [Ca] in cucumber fruits was higher in Se2 and Se3 than that in Se0 and Se1 and had few differences between two [CO₂] levels in the same Se supply level.

Table 2. ANOVA testing the effects of [CO₂] and Se supply levels and their combination on the mineral element concentrations in cucumber fruits.

| Factor ¹ | [N] | [P] | [K] | [Ca] | [Mg] | [S] | [Fe] | [Mn] | [Zn] |
|----------------------|-----------------|-----|-----|------|------|-----|------|------|------|
| CO ₂ | * ² | * | ** | NS | *** | ** | *** | * | *** |
| Se | NS ³ | NS | NS | *** | * | NS | NS | *** | * |
| CO ₂ × Se | NS | NS | NS | NS | NS | NS | NS | NS | NS |

¹ CO₂: [CO₂] level; Se: Se supply level. ² Asterisks (*) indicate significant differences (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$). ³ NS indicates non-significant differences ($p \geq 0.05$).

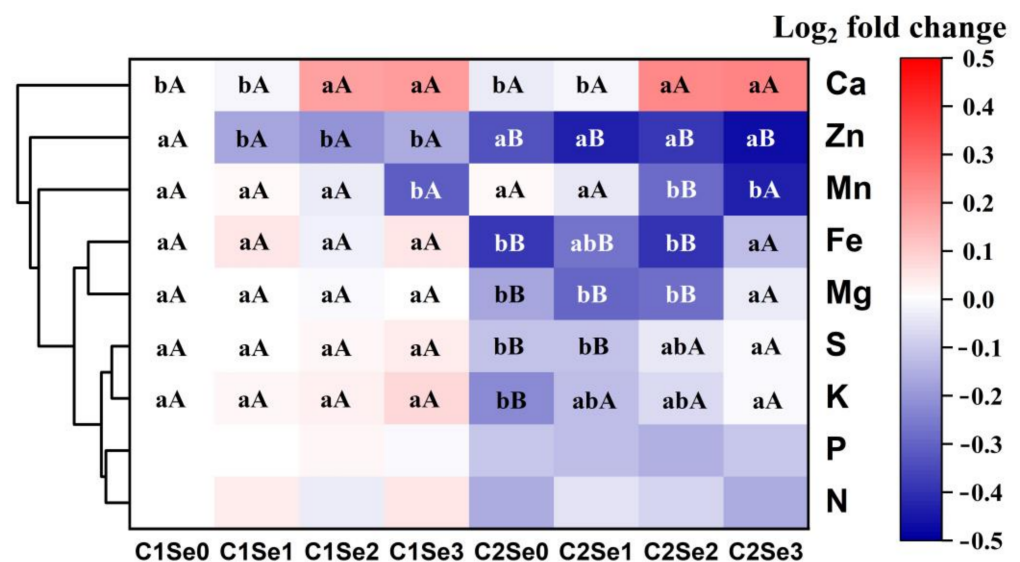


Figure 5. Heat map of changes in the mineral element concentrations in cucumber fruits relative to the mean value in C1Se0 treatment under different [CO₂] and Se supply levels ($n = 4$). [CO₂] levels: C1: ambient [CO₂] ($412 \mu\text{mol mol}^{-1}$); C2: elevated [CO₂] ($1196 \mu\text{mol mol}^{-1}$). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and $0.500 \text{ mg Se L}^{-1}$. Means not followed by the same lower-case letters are significantly different among different Se supply levels in the same [CO₂] level, and not followed by the same uppercase letters are significantly different between different [CO₂] levels in the same Se supply level, according to Fisher LSD test at $p < 0.05$. No letter indicates non-significant differences ($p \geq 0.05$).

3.6. Correlations between [CO₂], Se Supply and Nutritional Qualities of Cucumber Fruits

The Pearson correlation coefficient was calculated to evaluate the relationship between [CO₂], Se supply, gas exchange property, and the nutritional qualities of cucumber fruits (Figure 6). Pn and soluble sugar were positively correlated to [CO₂] with a correlation coefficient of 0.59 and 0.39, respectively. In contrast, Tr and soluble protein were negatively correlated to [CO₂] with a correlation coefficient of −0.48 and −0.54, respectively. All the ten mineral element concentrations in cucumber fruits were negatively correlated to [CO₂] except [Ca]. The negative correlation was significant between [N], [P], [K], [Mg], [S], [Fe], [Zn], and [CO₂], with the corresponding correlation coefficients of −0.43, −0.41, −0.49, −0.66, −0.53, −0.72, and −0.26, respectively.

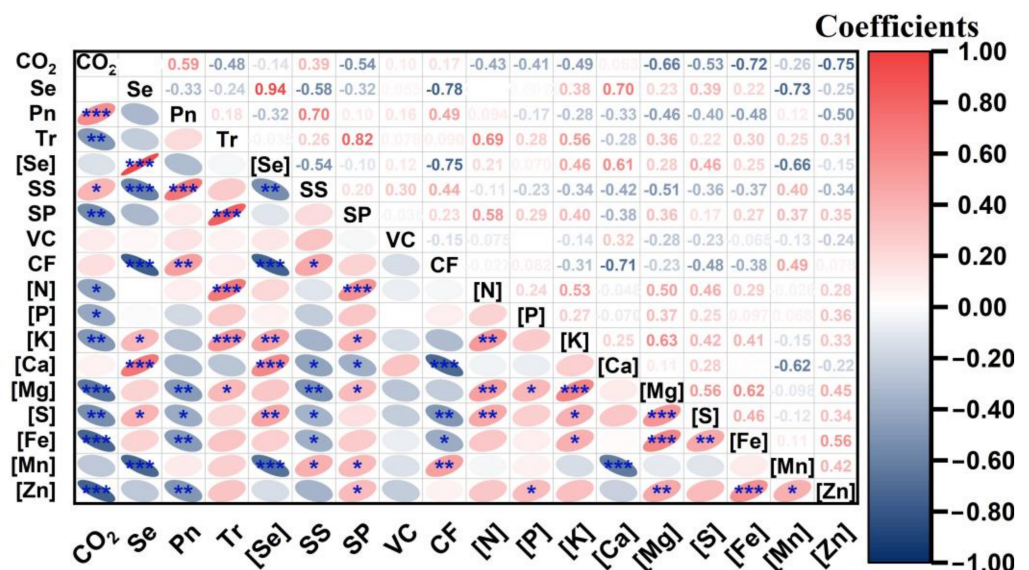


Figure 6. Pearson correlations between [CO₂] level, Se supply level, gas exchange property, Se concentration, organic nutrient concentration, and mineral concentrations in cucumber fruits under different [CO₂] and Se supply levels ($n = 4$). CO₂: [CO₂] level; Se: Se supply level; Pn: net photosynthesis rate; Tr: transpiration rate; SS: soluble sugar concentrations; SP: soluble protein concentrations; VC: vitamin C concentrations; CF: crude fiber concentrations. Asterisks (*) indicate significant levels (*: $p < 0.05$, **: $p < 0.01$, ***: $p < 0.001$).

[Se], [K], [Ca], and [S] in cucumber fruits had significantly positive correlations to the Se supply level, and the correlation coefficients were 0.94, 0.38, 0.70, and 0.39, respectively. Soluble sugar, crude fiber, and [Mn] in cucumber fruits had significantly negative correlations to the Se supply level, and the correlation coefficients were −0.58, −0.78, and −0.73, respectively.

Soluble sugar and crude fiber in cucumber fruits were positively correlated to Pn with a correlation coefficient of 0.70 and 0.49, respectively. All the ten mineral element concentrations in cucumber fruits were negatively correlated to Pn except [N] and [Mn]. The negative correlation was significant between [Mg], [S], [Fe], [Zn], and [CO₂], with the corresponding correlation coefficients of −0.46, −0.40, −0.48, and −0.50, respectively. Soluble sugar, protein, VC, crude fiber, and all the ten mineral element concentrations in cucumber fruits were positively correlated to Tr except [Ca]. The positive correlation was significant between soluble protein, [N], [K], [Mg], and Tr, with the corresponding correlation coefficients of 0.82, 0.69, 0.56, and 0.36, respectively.

The PCA plot was generated to further quantify and visualize the variation of the nutritional qualities of cucumber fruits responding to [CO₂] and Se supply levels. The first two components accounted for 57.9% of the variation, as shown in the PCA plot (Figure 7). The results showed that Pn, soluble sugar, crude fiber, and VC concentrations in cucumber fruits were positively correlated to [CO₂] levels. In contrast, Tr, soluble protein, [N], [P],

[K], [Mg], [S], [Fe], and [Zn] were negatively correlated to $[\text{CO}_2]$ levels. In addition, [Se] and [Ca] in cucumber fruits were positively correlated to Se supply levels. In contrast, Pn, soluble sugar, crude fiber, and [Mn] in cucumber fruits were negatively correlated with Se supply levels.

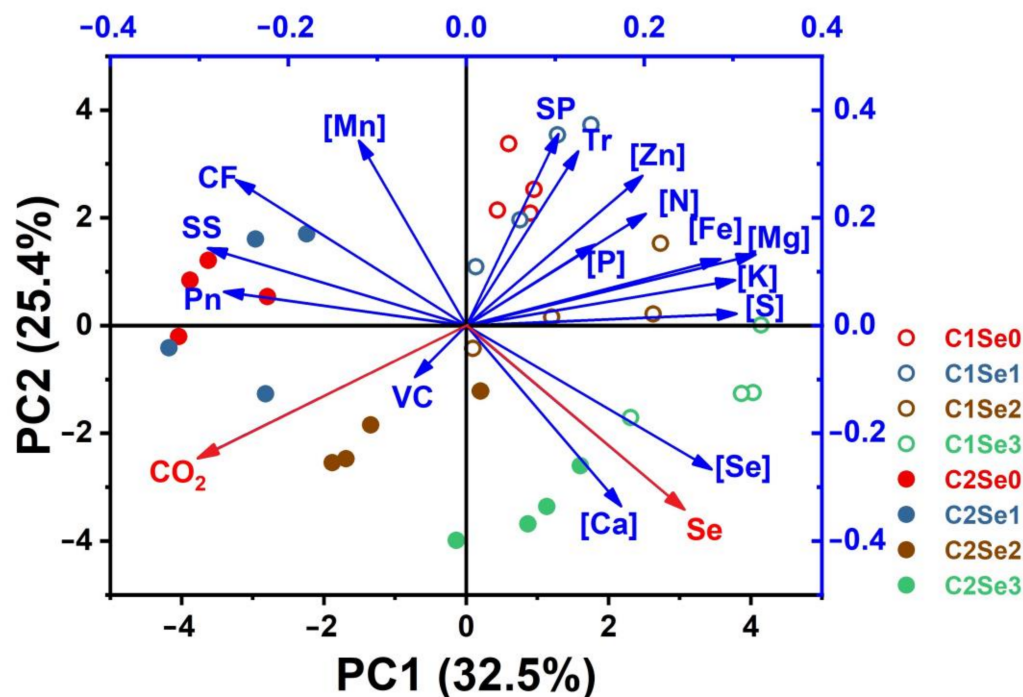


Figure 7. Principal components analysis of $[\text{CO}_2]$ level, Se supply level, gas exchange property, Se concentration, organic nutrient concentration, and mineral concentrations in cucumber fruits under different $[\text{CO}_2]$ and Se supply levels ($n = 4$). CO_2 : $[\text{CO}_2]$ level; Se: Se supply level; Pn: net photosynthesis rate; Tr: transpiration rate; SS: soluble sugar concentrations; SP: soluble protein concentrations; VC: vitamin C concentrations; CF: crude fiber concentrations. $[\text{CO}_2]$ levels: C1: ambient $[\text{CO}_2]$ ($412 \mu\text{mol mol}^{-1}$); C2: elevated $[\text{CO}_2]$ ($1196 \mu\text{mol mol}^{-1}$). Se supply levels: Se0, Se1, Se2, and Se3: 0, 0.125, 0.250, and $0.500 \text{ mg Se L}^{-1}$.

4. Discussion

4.1. Effects of $[\text{CO}_2]$ and Se Supply Levels on the Growth and Gas Exchange Parameters of Cucumber Plants

It is well known that CO_2 fertilization could significantly stimulate growth and increase the biomass accumulation of plants by enhancing the net photosynthetic rate [31,46,47]. In the present work, we observed a beneficial effect of CO_2 fertilization on the DW and Pn, but only in the Se0 and Se1 treatments (Figures 1 and 2). The ineffective CO_2 fertilization on the DW and Pn in Se2 and Se3 treatments indicated it was dose dependent. For example, Hawrylak-Nowak et al. found that a selenite concentration of $10 \mu\text{mol L}^{-1}$ ($0.79 \text{ mg Se L}^{-1}$) significantly decreased the shoot and root FW of cucumber plants compared with that in $6 \mu\text{mol L}^{-1}$ selenite ($0.47 \text{ mg Se L}^{-1}$) treatment [36]. Similarly, Haghighi et al. found that a 4 and 6 mg Se L^{-1} selenite supply significantly inhibited the shoot DW of cucumber plants. The Pn had a declining trend in 6 mg Se L^{-1} selenite treatment compared to 2 mg Se L^{-1} selenite treatment [48]. Those studies showed that a high level of selenite supply could inhibit the Pn and promote the growth of cucumber plants, but the selenite toxicity thresholds were different. In the present work, the selenite toxicity threshold was between 0.125 and $0.250 \text{ mg Se L}^{-1}$ ($1.58\text{--}3.16 \mu\text{mol L}^{-1}$), lower than the previous works [36,48]. These differences may be caused by the more extended selenite treatment period (70 days) and different cucumber varieties used in the present work.

Stimulation on the root, stem, and total DW and Pn of cucumber plants at the low level of Se supply (Se1: $0.125 \text{ mg Se L}^{-1}$) was found in this work (Figures 1 and 2), especially under CO_2 fertilization. Several potential reasons can contribute to this phenomenon. Firstly, an appropriate Se supply could protect the photosynthetic pigments and improve the Pn by enhancing the chloroplast antioxidant defense system (Figure 2), which could increase the tolerance to elevated $[\text{CO}_2]$ [49,50]. Secondly, increased Gs (Figure 2) was probably attributed to the Se-induced K^+ inward currents and stomatal opening [51,52], which were conducive to exploiting the full potential of CO_2 fertilization. Thirdly, the inhibition of Tr by high $[\text{CO}_2]$ was also alleviated due to the stomatal opening (Figure 2), so the uptake and transport of water and mineral nutrients through mass flow were improved [23]. Fourthly, low-dose selenite supply has been reported to shorten the primary root and increase lateral root initiation via hormonal and signaling processes and enhance root activities [35,53]. This facilitates the interception and uptake of mineral nutrients to match the enhanced CO_2 fixation and avoid photosynthesis acclimation under CO_2 fertilization [54].

4.2. Effects of $[\text{CO}_2]$ and Se Supply Levels on the Yield and Se Biofortification of Cucumber Fruits

Overall, the yield of cucumber was increased by CO_2 fertilization, whereas the [Se] in cucumber fruits was decreased with $[\text{CO}_2]$ elevation due to the dilution effects. Therefore, there were few differences in the Se accumulation and use efficiency between the two $[\text{CO}_2]$ levels (Figure 3). Specifically, similar to the DW, the most significant cucumber fruit yield was obtained in low-dose selenite supply (Se1) under CO_2 fertilization, mainly attributed to the enhanced Pn. [Se] in cucumber fruit, both on a DW and FW basis, was positively correlated to the Se supply level but was not significantly decreased by CO_2 fertilization except for Se3 treatment (Figure 3, Figure 6, and Figure 7, Table 1). A possible reason was that the improved Tr and enhanced root activities induced by the low-dose selenite supply (Se1) increased the uptake of Se, which counteracted the dilution effects under CO_2 fertilization. In contrast, high-dose selenite supply (Se3) produced toxic effects on the cucumber roots, which limited the uptake of Se and aggravated the reduction of [Se] under high $[\text{CO}_2]$ levels. This reason could also explain the higher Se accumulation and use efficiency in Se1 than in Se3 under CO_2 fertilization [55].

Most vegetables are non-accumulator plants for Se with high water but low protein content, so [Se] is usually below 100 mg kg^{-1} DW [24]. In previous works, the [Se] in the edible parts could be accumulated to $0.5\text{--}35.8 \text{ mg kg}^{-1}$ DW, $10.0\text{--}170 \text{ mg kg}^{-1}$ DW, $7.86\text{--}150 \text{ mg kg}^{-1}$ DW, and 15.5 mg kg^{-1} DW in biofortified tomato, lettuce, chicory, and spinach, respectively [20–24]. In the present work, [Se] in cucumber fruit could be enriched to $16.09\text{--}59.18 \text{ mg kg}^{-1}$ DW or $0.630\text{--}2.163 \text{ mg kg}^{-1}$ FW (Figure 3, Table 1), which was much higher than that grown in Se-enriched peat ($8.30\text{--}173 \text{ } \mu\text{g kg}^{-1}$ FW) or that foliar-sprayed with Se ($0.03\text{--}0.08 \text{ mg kg}^{-1}$ FW) [23,56]. Therefore, adding Se into the hydroponic system is a convenient and effective way to produce Se-biofortified vegetables due to the direct and active uptake of selenite or selenate through roots [57].

According to the dietary reference intakes from the WHO, the RDA and UL of Se were $55 \text{ } \mu\text{g day}^{-1}$ and $400 \text{ } \mu\text{g day}^{-1}$ for adults, respectively [25,26]. Therefore, an intake of 87.3 g of fresh cucumber fruit grown in C2Se1 can fully cover the RDA and is well consistent with people's dietary habits, assuming an intake of $200\text{--}250 \text{ g}$ of vegetables with three species per day [16]. On the other hand, the cucumber fruit grown in C1Se3 had the highest [Se] of 2.613 mg kg^{-1} FW, so intake of more than 185 g of fresh cucumber will dangerously exceed the UL, leading to a toxic risk for the consumer.

4.3. Effects of $[\text{CO}_2]$ and Se Supply Levels on the Nutritional Qualities of Cucumber Fruits

The organic and inorganic nutrients in the Se-enriched cucumber fruit were also important factors for the Se-biofortified foods. Because of the dilution effects, decreased mass flow and transpiration, and the changes in metabolism, the concentrations of protein, N, Zn, Fe, and Mg in the edible parts of vegetables all had a risk of declining under

CO₂ fertilization, leading to a deteriorated quality [32]. In this work, [K], [S], [Mg], [Fe], and [Zn] in cucumber fruit without Se supply (Se0) were significantly decreased by CO₂ fertilization. All of them had significant negative correlations ($p < 0.01$) to [CO₂] level (Figures 5–7). [N] and [P] in cucumber fruit also had a decreasing trend in C2Se0 compared with those in C1Se0 and were negatively correlated ($p < 0.05$) to [CO₂] level (Figures 5–7). This decline in these seven elements under CO₂ fertilization was mainly caused by the dilution effects [32,37]. Moreover, an increase of [K], [Mg], [S], and [Fe] in cucumber fruits as Se supply increased was observed under elevated [CO₂]. This increase may be attributed to the stronger inhibitory effect on the fruits' DW by high-dose Se supply than that on the uptake of these four elements.

Different from the findings in wheat and rice [58,59], [P] in cucumber fruit remained unaffected by adding selenite into the nutrient solution (Figure 5). Those works showed inhibition of P uptake under selenite supply, derived from the competition between phosphate and selenite due to their similar structures and transporter sharing [36,58]. In contrast, the unchanged [P] in selenite treatment was also reported in alfalfa, cherry tomato, and radish [35,60,61]. The possible explanation is that the phosphate supply (1 mol L⁻¹) is sufficient for cucumber growth, so the uptake of phosphate is hardly affected by selenite [35,62].

Notably, the regulation of [Ca] in cucumber fruit by Se supply was quite different from other mineral elements (Figure 5). [Ca] in cucumber fruit had a significant positive correlation ($p < 0.001$) to the Se supply level (Figures 6 and 7). The increase of [Ca] mediated by Se supply was also observed in *Brassica oleracea*, tomato, and maize [22,63,64]. Because Ca plays an essential role in the integrity of the cell membrane, the increase in [Ca] may be an indicator of the strengthening of the defense of the cell membrane or the rebuilding of the damaged cell membranes caused by Se toxicity [35,65]. Mn and Zn were the only elements whose concentrations in cucumber fruit were decreased when exposed to Se treatment and were negatively correlated to Se supply levels (Figures 5–7). Similar phenomena were found in lettuce and rice grain [66,67], but the underlying mechanism is still unknown.

Concerning the organic nutrients in cucumber fruit, as photosynthates, the concentrations of soluble sugar and crude fiber were increased by CO₂ fertilization in Se0 and positively correlated to [CO₂] (Figures 4, 6 and 7). The concentration of soluble sugar was also Se dose-dependent, higher in low-dose Se supply and lower in high-dose, which agrees with previous studies [23,67,68]. The increase in soluble sugar was mainly attributed to the strengthened Pn by low-dose Se supply [2,50]. However, the concentration of crude fiber was continuously decreased as the Se supply increased, which is consistent with the results in *Festuca arundinacea* Schreb [69]. This decrease in structural carbohydrates associated with the increase of soluble sugar may be caused by the Se-induced regulation of carbohydrates' metabolism [69]. The VC concentration in cucumber fruit was not significantly changed by CO₂ fertilization but was Se dose dependent (Figure 4). Previous studies have revealed that Se aided in improving the activities of the enzymes involved in the biosynthesis and recycling of VC in plants, which was responsible for the increase in VC concentration under low-dose Se supply [10,70,71].

Soluble protein in cucumber fruit only had a decreasing trend in Se0 by CO₂ fertilization but significantly decreased in Se1 and Se3 (Figure 4). A significantly negative correlation between soluble protein and [CO₂] and a non-significantly negative correlation between soluble protein and Se were found (Figures 6 and 7). However, the [N] in cucumber fruit was only negatively correlated to [CO₂], and it was not significantly influenced by the Se supply level (Figures 5–7). Therefore, the reason for the decline in soluble protein was not only the decreased [N] by CO₂ fertilization but also the damaged secondary structure and thermal stability of proteins by Se toxicity [29,72].

Together, in the cucumber fruits grown in C2Se1, soluble sugar was increased; crude fiber, VC, [N], [P], [K], [Ca], and [Mn] were maintained; but soluble protein, [S], [Mg], [Fe], and [Zn] were decreased when compared with C1Se1. Likewise, soluble sugar, soluble protein, VC, [N], [P], [K], [Ca], [Mg], [S], [Fe], [Mn], and [Zn] were all maintained, but only crude fiber was decreased in C2Se1 when compared with C2Se0.

5. Conclusions

According to the present results, although [Se] in cucumber fruits was increased with the Se supply level, the Se supply level in the nutrient solution was not higher or better. Because the regulation of cucumber plant growth by Se is dose-dependent, low-dose Se supply (Se1: 0.125 mg Se L⁻¹) protected the photosynthetic pigments and stimulated the stomatal opening, especially under CO₂ fertilization (C2: 1200 µmol mol⁻¹), and thus promoted Pn, Tr, and DW accumulation, which were all inhibited in high-dose Se supply (Se3: 0.500 mg Se L⁻¹) due to the selenite toxicity. Consequently, the highest fruit yield, Se use efficiency, and soluble sugar concentration were obtained in Se1 under CO₂ fertilization. Intake of 87.3 g of fresh cucumber fruit grown in C2Se1 can fully cover the RDA of 55 µg Se day⁻¹ for adults and is well consistent with people's dietary habits. Most organic and inorganic nutrients in Se-enriched cucumber (C2Se1) were maintained, with only a decrease in soluble protein, [S], [Mg], [Fe], and [Zn]. Therefore, better nutrient solution formulations for this hydroponic production of Se-enriched cucumber need to be developed in future studies to efficiently utilize the benefits of CO₂ fertilization and produce biofortified vegetables with high qualities.

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