



# Article Heavy Metals Can Affect Plant Morphology and Limit Plant Growth and Photosynthesis Processes

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Abstract: Soil heavy metal pollution caused by human activities has become one of the most critical environmental issues with a global concern. Phytoremediation is widely used due to its low cost and environmental friendliness. However, the impact of heavy metals on plant growth remains unclear. This study investigated the effects on the growth and photosynthetic activity of Picris divaricata Vant. under different cadmium concentrations using a hydroponics cultivation system. The results showed that the growth and photosynthetic processes of *P. divaricata* exhibited a phenomenon of promotion in low Cd concentrations and inhibition in high Cd concentrations. Under a low to medium Cd concentration ( $\leq$ 25  $\mu$ M), there was no Cd toxicity in terms of plant growth, but high concentrations of Cd inhibited plant growth. The Fe content of leaves gradually increased as the Cd concentration increased; it reached 201.8 mg kg<sup>-1</sup> in 75  $\mu$ M Cd. However, there was no significant difference in Mn between the 75  $\mu$ M Cd treatment and the control (p > 0.05). The contents of carotenoid ranged between 3.06 and 3.26 mg/g across the different Cd treatments, showing no significant differences. The treatment with 5–75 µM Cd did not directly affect the photosynthesis of *P. divaricata*. Higher Cd concentrations reduced the stomatal density on the of P. divaricata leaves, resulting in stomatal and mesophyll conductance limitations, indirectly affecting P. divaricata photosynthesis. These research results provide a reference for evaluating and selecting heavy metal tolerant plants and provide environmentally friendly approaches to remediate heavy metal pollution.

Keywords: heavy metal pollution; Picris divaricata Vant.; cadmium; photosynthesis; biomass

#### 1. Introduction

With industrial development, human activities have led to an enormous increase in the content of heavy metals in soil [1]. In the early 1990s, the average amount of cadmium (Cd) entering the soil through various sources reached 22,000 tons per year [2]. However, Cd is a non-essential and toxic trace element in plants, and it is easily absorbed by plant roots and transported to the leaves and the fruits [3]. After Cd enters the plant, it may impact plant growth and photosynthesis. The photosynthetic reaction sites of plants are most susceptible to heavy metal ions, and Cd is considered the most effective inhibitor of photosynthesis [4]. Cd has been shown to affect the net photosynthetic rate of many plants, such as tomatoes, rice, corn, and Indian mustard. Although only a small proportion of Cd<sup>2+</sup> can enter the chloroplast, it can strongly inhibit pigment functions, chloroplast structure, and photosynthesis's light and dark reactions. In addition, Cd disrupts the integrity of the chloroplast membrane system [5], leading to a significant increase in osmolytes. It was reported that low-concentration Cd treatment resulted in a sparse structure of the thylakoids in maize chloroplasts [6]. As Cd concentration increased, the thylakoids and grana disappeared, and the chloroplasts became spherical and shrunk. However, studies have also shown that increasing the concentration of Mn ions in the culture medium under Cd stress can partially restore the damaged structure of chloroplasts [7]. The inhibition



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**Copyright:** © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). of chlorophyll synthesis by Cd is considered the initial mechanism by which Cd affects photosynthesis [8]. The total chlorophyll content of *P. australis* decreased by 30% and 60% compared to the control, under 50 and 100  $\mu$ M Cd, respectively [9]. In addition to reducing the total chlorophyll content, Cd directly affects the photon capture by chlorophyll molecules, leading to the reduction and eventual shutdown of photosynthesis. Certain studies have demonstrated that Cd can replace the carotenoid binding in PSII [10], thus affecting carotenoids that play an important role in plant absorption of light energy and chlorophyll protection.

Presently, research on plants that have metal hyperaccumulator properties mainly focuses on heavy metal absorption, transportation, storage, and detoxification mechanisms by different super-enriched plants at the physiological and molecular levels [11,12]. Moreover, the inhibitory effect of heavy metal stress on photosynthesis has been extensively demonstrated [13]. However, Cd can also promote growth and increase photosynthetic rates in certain species. *Picris divaricata Vant.* is a Zn/Cd polymetallic-tolerant and hyperaccumulator plant that is able to grow in Lanping County, Yunnan Province, where China's largest lead and zinc mine is located. It has high heavy metal tolerance, enrichment ability, and relatively large biomass [14]. In this study, we compared and analyzed the growth and photosynthetic response of *P. divaricata* under Cd stress using a hydroponics-based cultivation system and investigated the mechanisms underlying photosynthesis maintenance under these conditions. The research results provide a basis for heavy metal phytoextraction approaches using hyperaccumulator plants and insights into the physiological adaptation mechanisms of plants in extreme environments.

#### 2. Materials and Methods

#### 2.1. Experimental Design

Seedlings of *P. divaricata* were collected from the Jinding lead–zinc mining area in Lanping County, Nujiang Prefecture, Yunnan Province. Seedlings with similar growth were transferred to plastic pots containing 500 mL of 20% Hoagland nutrient solution for pre-cultivation. After 14 days of plant pre-cultivation, five Cd concentrations were applied in the above nutrient solution, namely 0 (CK), 5, 10, 25, 50, and 75  $\mu$ M (added as CdCl<sub>2</sub>·2.5 H<sub>2</sub>O). Each treatment has 3 replicates, with three plants per replicate (Figure 1). The nutrient solution was changed every 3 days, and continuous aeration was maintained. The greenhouse growth conditions were day/night temperature of 25/18 °C, relative humidity of 70–75%, and light intensity of 100  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup>, with 14 h of light and 10 h of darkness per day.

#### 2.2. Biomass Measurements and Root Growth Analyses of P. divaricata under Cd Stress

After 14 days of plant pre-cultivation, the entire *P. divaricata* seedlings were removed from the medium, and the plant roots were soaked in 5 mM CaCl<sub>2</sub> for 5 min to remove heavy metal ions adsorbed on the surface of the roots. The samples were collected from 3 plants in each pot and pooled for the analyses. Then, the samples were washed with deionized water, and, after separating the above-ground parts and the roots, their fresh weight was measured separately. The plants were then dried at 105 °C for 30 min and at 75 °C for 48 h to a constant weight, and then the dry weight was measured.

The *P. divaricata* root morphological characteristics were measured using the Win-RHIZO root analysis system manufactured by Regent Instruments in Canada. The root sample was placed in a 30 cm  $\times$  40 cm transparent glass tray, and water was added, forming a 3–4 mm layer to disperse the root system fully. The root system was scanned with a double-sided light source and analyzed by dedicated digital software (WinRHIZO 2008a). Morphological indicators such as root length, root surface area, root volume, and average root diameter were obtained.



(d) 25 µM Cd

(e) 50 µM Cd

(f) 75 µM Cd

**Figure 1.** Growth of *P. divaricata* in different levels of Cd. The (**a**, **b**, **c**, **d**, **e**, and **f**) correspond to the 0, 1, 5, 10, 25, 50, and 75 μM Cd treatments, respectively.

# 2.3. Determination of Photosynthetic Parameters

A total of 1~2 fully expanded leaves were selected from the plants in each pot for the determination of photosynthetic parameters with a photosynthesis meter (LI-6400XT, LI-COR, Lincoln, NE, USA). Using a combined red and blue light source leaf chamber, the net photosynthetic rate (AN), stomatal conductance (gs), intercellular CO<sub>2</sub> concentration (Ci), and other gas exchange parameters were measured under different light density flux (PPFD) settings (2000, 1800, 1500, 1200, 1000, 800, 700, 600, 500, 400, 200, 100, 50, 20, and 0  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup>). The measurement conditions were leaf temperature of 25 °C, CO<sub>2</sub> concentration 380  $\pm$  5  $\mu$ mol/mol, and airflow rate of 500  $\mu$ mol·s<sup>-1</sup>.

#### 2.4. Determination of Chlorophyll Fluorescence Parameters

The chlorophyll of leaves was determined by 80% acetone extraction spectrophotometry. Remove the weighed leaves from the refrigerator at -80 °C and place them in a mortar. Then add a small amount of calcium carbonate and 2–3 mL of 80% acetone to grind into a homogenate. Filter the extract into a 25 mL colorimetric tube wrapped in tin foil, rinse the mortar, grinding rod, and residue several times with a small amount of 80% acetone, pour into a funnel, and filter together. Dilute to 25 ml with 80% acetone and shake well. Measure the extinction of the extraction solution at wavelengths 663, 646, and 470 nm using 80% acetone as the blank.

The chlorophyll fluorescence parameters of the plants were measured with the LI-6400XT photosynthesis meter. The measured leaf position was the same as those selected for photosynthetic gas exchange parameter measurements. The specific steps followed were:

- (1) After dark adaptation of the selected leaf for 30 min, a light with a wavelength of 630 nm was flashed on the leaf to measure the initial fluorescence ( $F_0$ ). Then, light with a wavelength of 630 nm and an intensity of approximately 6000 µmol·m<sup>-2</sup>·s<sup>-1</sup> was used to saturate the reaction centers of photosystem II (Duration 0.8 s, Intensity 7, Blue 10%, Modulation 20 KHz, Filter 50 KHz) to determine the maximum fluorescence ( $F_m$ ).
- (2) The leaves sampled for the measurements were exposed to continuous photochemical active light (constant light in a greenhouse with a light intensity of 80  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup> for 30 min at steady-state fluorescence ( $F_s$ )). Then, the light intensity was increased to 6000  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup> to measure the maximum fluorescence ( $F_m'$ ) after adaptation under saturated light irradiation. Then, the photochemically active light was stopped, and far-red light with a wavelength of 740 nm was applied (Duration 6 s, Intensity 8, Modulation 0.25 KHz, Filter 1 KHz) to measure the minimal fluorescence,  $F_o'$ .
- (3) Other chlorophyll fluorescence parameters of PSII, such as the maximum photochemical efficiency ( $F_v/F_m$ ), actual photochemical efficiency of PSII ( $\Phi_{PSII}$ ), the photochemical quenching ( $q_P$ ), non-photochemical quenching (NPQ), and electron transfer rate (ETR) of PSII were calculated according to the following formulas:

$$F_{\rm v}/F_{\rm m} = (F_{\rm m} - F_{\rm o})/F_{\rm m} \tag{1}$$

$$F_{\rm v}'/F_{\rm m}' = (F_{\rm m}' - F_{\rm o}')/F_{\rm m}'$$
 (2)

$$\Phi_{\rm PSII} = (F_{\rm m}' - F_{\rm s}') / F_{\rm m}'$$
(3)

$$q_{\rm P} = (F_{\rm m} - F_{\rm s}) / (F_{\rm m}' - F_{\rm o}') \tag{4}$$

$$NPQ = (F_m - F_m')/F_m' = F_m/F_m' - 1$$
(5)

$$ETR = \Phi_{PSII} \times PPFD \times 0.5 \times 0.84$$
(6)

#### 2.5. Stomatal Pore Observation by Scanning Electron Microscopy

Samples with a width of 2–3 mm were excised from the middle transverse section of a plant leaf. The samples were washed with deionized water and immersed in a 1% Na<sub>2</sub>S aqueous solution. They were then subjected to vacuum infiltration for 5 min to appropriately soak with the solution. After soaking at room temperature for another 25 min, they were rinsed twice with 0.1 mmol L<sup>-1</sup> phosphate buffer (pH 7.2). A single-sided blade pre-cooled with liquid nitrogen was used to cut the sample and obtain an undisturbed transverse section, which was then quickly transferred to a freeze-drying machine (JFD-320, JEOL, Tokyo, Japan) for drying. The dried samples were mounted on the plastic platform, sprayed with carbon, and observed under the scanning electron

microscope (SEM, Model Quanta 400, FEI, Eindhoven, The Netherlands) under thermal field emission.

#### 2.6. Fluorescence CO<sub>2</sub> Response Curve

The chlorophyll fluorescence  $CO_2$  response curves of plants were measured by the LI-6400XT photosynthesis analyzer and a fluorescence leaf chamber (LI-COR, Lincoln, NE, USA). Please refer to annex 1 for specific measurement methods (File S1).

#### 2.7. Statistical Analysis

Statistical analyses were performed using SPSS 16.0 (IBM, New York, NY, USA). Descriptive statistics were used to calculate the means and standard deviations for each set of replicates. One-way ANOVA with LSD-t test was performed to test the differences in the biomass, root growth, and metals of different Cd treatments. A one-way ANOVA was also used to analyze the effect of Cd treatments on PSII, chlorophyll a, chlorophyll b, and carotenoid contents. All data were tested for normal distribution and homogeneity of variance.

### 3. Results

### 3.1. Effects of Cd Stress on the Biomass, Root Growth, and Metals Content of P. divaricata

Low to medium Cd concentrations had a beneficial effect on the total biomass and root growth of *P. divaricata*, which were gradually reduced as the Cd concentration increased to high levels (Figure 2). The FW of *P. divaricata* was the highest in the 5  $\mu$ M Cd treatment, 1.27-fold higher than that of the CK, but gradually decreased with the increase in Cd concentration. There was no significant difference between the 10 and 25  $\mu$ M Cd treatments (p > 0.05). The FWs of *P. divaricata* under 5, 10, and 25  $\mu$ M Cd were 2-, 1.33-, and 1.24-fold higher than the CK, respectively. The FW of *P. divaricata* under the 50 and 75  $\mu$ M Cd treatments was significantly lower compared to 25  $\mu$ M Cd. The DW of *P. divaricate* followed the same patterns as the FW under the different Cd treatments. However, the DW of *P. divaricata* exhibited a more pronounced Cd concentration response. In the 5–25  $\mu$ M Cd treatment, the DW was 1.85–2.19-fold higher than that of the CK. The length, average diameter, total surface area, and total volume of the roots increased with the increase in Cd concentration (Table 1). In the 5–25  $\mu$ M Cd treatments, they were significantly higher than those of CK (p < 0.05). However, they decreased significantly in the 50 and 75  $\mu$ M Cd treatments (p < 0.05).



Figure 2. Biomass of P. divaricata under different Cd concentrations.

Treatment (Cd μM)	Root Length (cm)	Average Root Diameter (mm)	Total Surface Area of Root System (cm <sup>2</sup> )	Total Volume of Root System (cm <sup>3</sup> )
0	$1681.8 \pm 154.3 \ ^{\rm b}$	$0.30\pm0.03$ $^{\rm a}$	$156.46 \pm 5.80$ <sup>b</sup>	$1.17\pm0.17$ <sup>c</sup>
5	$2555.4 \pm 356.8 \ ^{\rm a}$	$0.31\pm0.02~^{\mathrm{a}}$	$252.92\pm52.91~^{\rm a}$	$2.00\pm0.56~^{ m ab}$
10	$2429.2\pm139.2~^{\rm a}$	$0.35\pm0.02~^{\mathrm{a}}$	$262.90 \pm 23.20~^{a}$	$2.27\pm0.33$ <sup>a</sup>
25	$2741.0\pm196.8~^{\rm a}$	$0.34\pm0.01$ a	$288.28\pm10.03~^{\rm a}$	$2.43\pm0.07$ $^{\mathrm{a}}$
50	$1501.3 \pm 385.8$ <sup>b</sup>	$0.33\pm0.00~^{\mathrm{a}}$	$155.79 \pm 38.02$ <sup>b</sup>	$1.27\pm0.30~\mathrm{^{bc}}$
75	$1678.0 \pm 473.5 \ ^{\rm b}$	$0.31\pm0.03$ a	$165.18 \pm 57.93 \ ^{\rm b}$	$1.30\pm0.55~^{\mathrm{bc}}$

**Table 1.** Root system growth of *P. divaricata* in different Cd treatments. Values represent the mean and the variance of three independent replicates. Different letters in the same column indicate values which significantly differ according to one-way ANOVA at p < 0.05.

There were no significant changes in the Mg, Ca, and Cu in leaves of *P. divaricata* under Cd treatment (Table 2). The Fe content of leaves were 144.1 and 153.9 mg kg<sup>-1</sup> in the 5–10  $\mu$ M Cd treatment, showing a significant difference from the control (p < 0.05). However, the Fe content of leaves gradually increased as the Cd concentration increased, and it reached 201.8 mg kg<sup>-1</sup> in 75  $\mu$ M Cd. Similarly, the content of Mn in the 5–25  $\mu$ M Cd treatment was lower than the control; they were 60.4%, 78.2%, and 52.3% of the control, respectively. There was no significant difference in Mn between 75  $\mu$ M Cd treatment and control (p > 0.05).

**Table 2.** Metals contents in *P. divaricata*. Values represent the mean and the variance of three independent replicates. Different letters in the same column indicate values which significantly differ according to one-way ANOVA at p < 0.05.

Treatment (Cd μM)	Fe (mg kg <sup>-1</sup> )	Mg (g kg $^{-1}$ )	Mn (mg kg $^{-1}$ )	Ca (g kg <sup>-1</sup> )	Cu (mg kg <sup>-1</sup> )	Zn (mg kg $^{-1}$ )
0	$227.4\pm52.8~^{\rm a}$	$6.58\pm0.47$ $^{\rm a}$	$234.5\pm123.1$ $^{\rm a}$	$22.9\pm2.59~^{a}$	$3.11\pm0.50~^{a}$	$37.2\pm10.2~^{\rm a}$
5	$144.1\pm13.0~^{\rm b}$	$6.57\pm0.62$ $^{\rm a}$	$307.7\pm164.9$ $^{\rm a}$	$22.0\pm1.83$ <sup>a</sup>	$2.09\pm1.62~^{\rm a}$	$30.8\pm2.57$ $^{\mathrm{a}}$
10	$153.9 \pm 45.8 \ ^{\mathrm{b}}$	$5.77\pm0.44$ $^{\rm a}$	$222.1\pm45.6~^{\rm a}$	$20.1\pm0.33$ <sup>a</sup>	$3.27\pm0.15$ $^{\rm a}$	$35.0\pm7.04$ <sup>a</sup>
25	$164.7\pm14.6~^{\mathrm{ab}}$	$5.28\pm1.12$ $^{\mathrm{a}}$	$259.1\pm76.1~^{\rm a}$	17.8 $\pm$ 0.73 $^{\mathrm{a}}$	$2.95\pm1.14~^{\rm a}$	$22.9\pm8.50~^{a}$
50	$173.8\pm40.6$ <sup>ab</sup>	$5.06\pm0.26$ a	$449.3\pm174.1$ a	$20.2\pm4.62$ a	$4.62\pm0.85$ a	$22.1\pm14.3$ a
75	$201.8\pm21.8~^{ab}$	$5.81\pm1.47$ $^{\rm a}$	$281.7\pm132.1~^{a}$	$20.4\pm5.66~^{a}$	$4.40\pm2.57$ $^{a}$	$26.7\pm11.4~^{a}$

#### 3.2. Effects of Cadmium on Leaf Pigment Content of P. divaricata

The chlorophyll a, chlorophyll b, and carotenoid contents in the leaves of *P. divaricata* were increased as the Cd treatment concentration increased, and then they decreased under higher Cd concentrations (Figure 3). They were significantly higher in the 25  $\mu$ M Cd concentration compared to the CK (p < 0.05). Whereas, under 50 and 75  $\mu$ M of Cd, they were significantly lower compared to CK (p > 0.05). There was no significant difference in chlorophyll a/b between the 5 and 10  $\mu$ M Cd treatments (p > 0.05). Similarly, there were also no differences in chlorophyll a/b between the 50 and 75  $\mu$ M Cd treatments (p > 0.05). The contents of carotenoid ranged between 3.06 and 3.26 mg/g across the different Cd treatments, showing no significant differences (p > 0.05).

#### 3.3. Distribution Pattern of Blade Cross-Section

Compared to the control, the 5 and 10  $\mu$ M Cd treatments resulted in a thickening of the *P. divaricata* leaf cross-section, with thicknesses of 147.3 and 142.0  $\mu$ M, respectively (Figures 4 and 5). As the concentration of Cd applied increased, the thickness of cross-section of the leaves became compact, and the intercellular spaces became smaller. The lowest value was observed at 75  $\mu$ M Cd at 103.3  $\mu$ m, which was 22.3% lower than that of CK, at 133.0  $\mu$ M.







(a) CK

(**b**) 5 µM Cd

(**c**) 10 µM Cd



(**d**) 25 µM Cd

(e) 50  $\mu$ M Cd

(**f**) 75 µM Cd

**Figure 4.** Scanning electron microscopy micrographs of the upper and lower epidermis and stomata of control and 75  $\mu$ M Cd treated *P. divaricata* (magnification ×250 and 3000). (a) The upper epidermis of control (magnification ×250); (b) the lower epidermis of control (magnification ×250); (c) the stomata in the lower epidermis of control (magnification ×250); (d) the upper epidermis of 75  $\mu$ M Cd (magnification ×250); (e) the lower epidermis of 75  $\mu$ M Cd (magnification ×250); (f) stomata in lower epidermis of 75  $\mu$ M Cd (magnification ×3000); (f) stomata in lower epidermis of 75  $\mu$ M Cd (magnification ×3000). White arrows, stoma. Bars (a,b,d,e) = 100  $\mu$ m.



**Figure 5.** The thickness of mesophyll cross-sections of leaves in control and Cd-treated *P. divaricata plants.* The n is the number of samples.

# 3.4. Effects of Cadmium on the Photochemical Reactions of P. divaricata

3.4.1. Effects of Cadmium on the Chlorophyll Fluorescence Parameters of P. divaricata

The  $F_v/F_m$  increased with the increase in Cd treatment concentrations (Figure 6). The maximum photochemical quantum yield ( $F_v/F_m$ ) remained around 0.8 across all treatments. The  $F_m'$ , and  $F_v'$  increased with increasing Cd concentration up to 10  $\mu$ M Cd, while they decreased under 25–75  $\mu$ M Cd, showing an inverted "U"-shaped trend under Cd treatment. Compared with the control, there were no significant differences in  $F_v'/F_m'$  between each Cd treatment and the CK.



**Figure 6.** Chlorophyll fluorescence parameters ( $F_v/F_m$ ,  $F_v'/F_m'$ ,  $\Phi_{PSII}$ ) in *P. divaricata* under different Cd concentrations.

The  $\Phi_{PSII}$  of the plants in the Cd treatments ranged between 0.5 and 0.6, and there were no significant differences between each Cd treatment and the CK (Figure 7). No significant differences could be identified in the ETR between the Cd treatments and the CK. Similarly, no significant differences were observed in  $q_P$  between the Cd treatments and CK, except for the 10  $\mu$ M Cd concentration. The NPQ was also unaffected by Cd, showing no significant differences among the Cd treatments.



**Figure 7.** Effects of cadmium on fluorescence parameters ( $\Phi_{PSII}$ , ETR,  $q_P$ , NPQ) of *P. divaricata* under different Cd concentrations.

3.4.2. Effects of Cadmium on the Excited-State Energy Distribution and the Photoreaction Quantum Yield in *P. divaricata* 

The P used for the photochemical reactions did not exhibit significant changes under different Cd treatments, and the *p* value remained between 0.53 and 0.59 (Figure 8). The heat dissipation D of the antenna pigment–protein complex showed no significant difference between Cd treatments and CK. However, the D values in the leaves under 75 and 5  $\mu$ M Cd were 0.34 and 0.27, showing significant differences (*p* < 0.05). The excess excitation energy Ex absorbed by PSII was the lowest at 10 and 50  $\mu$ M Cd, significantly different compared to CK. On the other hand, there was no difference in Ex between the 5 and 75  $\mu$ M Cd treatments.



**Figure 8.** Energy distribution of PSII in *P. divaricata* under different Cd concentrations. Significant differences (p < 0.05) among different Cd treatments are indicated with different lower-case letters.

# 3.5. Effects of Cadmium on the Photosynthetic Parameters of P. divaricata

# 3.5.1. Effects of Cadmium on Photosynthetic Parameters and the Light Response Curve of *P. divaricata*

The net photosynthetic rate (Pn) of *P. divaricata* under different Cd treatments showed a consistent trend. It initially increased with the increase in photon flux density (PPFD) and gradually decreased after being suppressed by light (Figure 9). The maximum Pn value decreased with increased Cd concentration. The maximum Pn value was reached under  $5 \mu$ M Cd at 7.00  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup> and was significantly higher compared to that of CK at 6.07  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup>. The maximum Pn values under the 10, 25, 50, and 75  $\mu$ M Cd treatments decreased significantly with the increase in Cd concentration, reaching 5.99, 5.13, 4.93, and 3.86  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup>, respectively. This indicates that the photosynthesis of *P. divaricata* was inhibited as the Cd concentration increased.



**Figure 9.** Gas exchange parameters of *P. divaricata* leaves under different Cd treatments and different photosynthetic photon flux densities (PPFD). (**a**, **b**, **c**, and **d**) correspond to the net photosynthetic rate ( $P_n$ ), stomatal conductance ( $g_s$ ), intercellular CO<sub>2</sub> concentration ( $C_i$ ), and transpiration rate ( $T_r$ ), respectively.

Similar to Pn, the stomatal conductance ( $g_s$ ) and transpiration rate ( $T_r$ ) showed a trend of initially increasing and then decreasing with the increase in PPFD. There was an inhibition of gs and  $T_r$  at 800  $\mu$ mol·m<sup>-2</sup>·s<sup>-1</sup>. The trend in the intercellular CO<sub>2</sub> concentration (Ci) with increasing PPFD was opposite to that of AN,  $g_s$ , and  $T_r$ , initially decreasing and then gradually increasing.

# 3.5.2. Analysis of the Factors Affected by Cadmium That Influence the Photosynthesis of *P. divaricata*

Using the fluorescence  $CO_2$  response curve, the corresponding limiting values of the photosynthetic limiting factors of *P. divaricata* were calculated under different Cd concentrations (Table 3). The 5  $\mu$ M Cd concentration was set as the reference, and, as a result, its various limit values were all 0. Because diffusion limitation ( $D_L = S_L + MC_L$ ) accounted for the majority of the total limitation value ( $T_L$ ) (73–96%), both  $D_L$  and  $T_L$ increased with the Cd treatment concentration increase. Their values and trends were similar to that of Pn. There were no significant differences between 10  $\mu$ M Cd and the CK. The biochemical limitation ( $B_L$ ) accounted for a small percentage of the total photosynthesis limitation, at only 4–27%. The mesophyll conductance limitation ( $MC_L$ ) accounted for most of the non-stomatal limitation factors ( $N_L$ ;  $N_L = MC_L + B_L$ ).

Table 3. Photosynthesis limitation parameters (%) of *P. divaricata* under different Cd concentrations.

Treatment (Cd μM)	SL	MCL	BL	NL	$D_L$	TL
0	$9.0\pm1.3$ c	$4.2\pm3.3$ c	$5.1\pm1.2$ a	$9.7\pm2.7$ c	$13.6\pm2.3$ c	$18.7\pm2.4$ <sup>c</sup>
10	$12.1\pm0.3$ <sup>c</sup>	$2.6\pm0.4$ <sup>c</sup>	$0.6\pm0.0~^{ m c}$	$3.1\pm0.4$ <sup>c</sup>	$14.7\pm0.1~^{ m c}$	$15.3\pm0.2~^{\rm c}$
25	$17.0\pm1.9$ <sup>b</sup>	$11.7\pm2.1~^{ m b}$	$0.5\pm0.2~^{ m c}$	$12.2\pm2.0~^{\rm c}$	$28.7\pm1.8$ <sup>b</sup>	$39.2\pm2.0$ <sup>b</sup>
50	$23.0\pm1.5~^{\rm b}$	$18.5\pm3.0$ <sup>b</sup>	$2.9\pm0.8$ <sup>b</sup>	$21.4\pm4.0$ <sup>b</sup>	$41.5\pm3.3$ <sup>b</sup>	$44.4\pm3.7~^{ m b}$
75	$28.8\pm2.0~^{a}$	$36.8\pm7.2\ ^{a}$	$3.1\pm0.4$ <sup>b</sup>	$40.0\pm7.2~^{\rm a}$	$65.7\pm6.2$ $^{a}$	$68.8\pm6.3$ $^{a}$

The 5  $\mu$ M Cd treatment was used as a reference, for which all limitations were set to 0. SL, stomatal limitations; MCL, mesophyll conductance limitations; B<sub>L</sub>, biochemical limitations; N<sub>L</sub>, non-stomatal limitations; D<sub>L</sub>, diffusional limitations; T<sub>L</sub>, total limitations. Data are the mean  $\pm$  SE of three replicates. Different letters in the same column indicate values that are significantly different according to one-way ANOVA at *p* < 0.05.

#### 4. Discussion

# 4.1. Cadmium Resulted in Growth Promotion in Low Concentrations and Growth Inhibition in High Concentrations in P. divaricata

The root system is an important plant organ that absorbs mineral nutrients and water. Under abiotic stress conditions, the plant root system is the first organ to be under stress. Plant roots have certain developmental plasticity [15], and plants can alter the morphology and distribution of their roots to adapt to environmental stress [16]. Cadmium exerts a strong toxicity to plants, and its toxicity symptoms are first expressed in plant roots [17]. Root length is an indicator of the potential for plants to absorb nutrients or water [18], while root area is an indicator of root absorption efficiency [19]. The heavy metal absorption efficiency by plant roots directly affects their metal enrichment ability. The metal hyperaccumulator model plant *T. caerulescens*, when compared to non-metal hyperaccumulator plants such as *T. arvense*, has more lateral roots, denser and longer root hairs (>2.1 mm), and longer effective root length [20], which enhance the ability of T. caerulescens to absorb and utilize metals faster and more efficiently for normal growth. Similarly, compared with the CK, the root length and the root surface area of the metal hyperaccumulator *Sedum alfredii* significantly increased under Cd treatment ( $\geq 50 \ \mu$ M). However, Cd did not affect root diameter and volume [21]. The study showed that low to medium concentrations of Cd (5, 10, 25  $\mu$ M) promoted root growth of *P. divaricata*. However, at higher concentrations (50 and 75  $\mu$ M), the root system growth was inhibited to a certain extent, indicating that *P. divaricata*, as a metal hyperaccumulator plant, has strong adaptability to Cd.

The apparent growth status and biomass of *P. divaricata* also showed that Cd had "growth promotion in low concentrations and growth inhibition in high concentrations" effects in *P. divaricata*, consistent with previous experimental results [22]. In toxicology, the phenomenon where chemicals exhibit beneficial effects on organisms at low doses (such as stimulating growth and development) but exhibit negative effects at high doses is known as hormesis [23]. However, in general, the stimulus–response amplitude and the stimulus–dose range for a hormesis response are very small [24]. The maximum stimulus–response amplitude generally does not exceed twice that of the control and is usually only 30% to 60% higher

than the control; the dose range of the stimulus is generally around one percent to one-fifth of the no-observed-adverse-effect level [25]. In this study, P. divaricata exhibited toxicity symptoms under the 50  $\mu$ M Cd treatment; however, 5–25  $\mu$ M Cd stimulated growth, which is higher than the normal stimulus dose range to induce hormesis. Moreover, the fresh and dry weights of the below-ground parts of *P. divaricata* treated with Cd were 2.2- and 2.6-fold higher than the control, respectively, and the dry weight of the above-ground part was 1.5-fold higher than the control. Moreover, the root length and root surface area under Cd treatment were up to 63% and 84% higher than the control, respectively. The maximum stimulus response amplitude in P. divaricata was also greater than what is commonly observed in hormesis responses. The effect of Cd resulting in growth promotion in low concentrations and growth inhibition in high concentration in *P. divaricata* is likely not a hormesis effect as described in toxicology, and an overcompensation mechanism cannot explain the growth-promoting effects of Cd in *P. divaricata*. This phenomenon observed in metal hyperaccumulator plants may be related to their long-term biological evolution in environments with extremely heavy metal concentrations. A certain concentration of heavy metals may have become a necessary condition for the normal growth of metal hyperaccumulator plants, and this hypothesis also needs to be further investigated in the future.

#### 4.2. Effects of Cadmium on P. divaricata Photosynthesis

The photosynthetic organs of plants are very sensitive to heavy metal stress [26]. Cd can affect various processes of photosynthesis. Zhu et al. concluded that Cd affects photosynthesis in two different ways: (1) directly, by affecting the chloroplast structure, photosynthetic pigments, fluorescence parameters, electron transfer, and the Calvin cycle enzyme activities [27,28]; and (2) indirectly, through factors such as water status, stomatal opening and closing, and other factors that can affect intercellular and intracellular CO<sub>2</sub> contents [29]. In this study, Pn was inhibited at high Cd levels, indicating that Cd did not directly affect the photosynthesis of *P. divaricata*. At the same time, deficiencies in different nutrient elements can affect chlorophyll and fluorescence parameters. In this study, the element content was similar to the leaf pigment content; therefore, Cd did not significantly reduce it. This also indicates that Cd did not affect photosynthesis indirectly through nutrient element deficiencies.

Stomatal density was positively correlated with stomatal conductance g<sub>s</sub>, net photosynthetic rate Pn, and water use efficiency [30]. Stomatal closure is an important factor controlling biological carbon fixation [31]. In this study, the plant water content was altered under 50–75  $\mu$ M Cd. This significant decrease is directly related to stomatal conductance, mesophyll conductance, stomatal density, stomatal opening and closure, and leaf density. Tang et al. suggested that when SL increases, Ci decreases [32]. We can conclude that stomatal closure is the main reason for the photosynthetic rate decrease. In this study, we can conclude that stomatal limitations were one of the reasons for the reduction of the photosynthetic rate of *P. divaricata*. Our research results are consistent with those of many studies under different stress conditions [33]. However, certain studies have demonstrated that Cd could reduce AN, GS, and E in rice [34] and peanut [35] but could increase stomatal density and Ci. These differences may be related to the different plants studied. The mesophyll conductance (gm) is considered to play a more important role than stomatal conductance in leaf photosynthesis [36]. Many studies have shown that mesophyll conductance  $g_m$  is relatively very low in high Cd concentration, which results in the chloroplast matrix  $CO_2$  concentration (Cc) being relatively lower compared to the intercellular  $CO_2$  (Ci) concentration, further confirming its limiting effect on photosynthesis [37]. In this study, g<sub>m</sub> decreased due to the reduction of mesophyll and chloroplast surfaces and changes in leaf morphology caused by Cd, consistent with the research results under water, salt, and Zn stress [38,39].

### 5. Conclusions

This study demonstrated that *P. divaricata* has strong Cd tolerance and Cd accumulation capacity. *P. divaricata* showed a phenomenon of "growth promotion in low concentrations and growth inhibition in high concentrations" under Cd treatment. At Cd concentrations below 25  $\mu$ M, there was no Cd toxicity in plant growth and photosynthesis, but high concentrations of Cd inhibited plant growth and photosynthetic physiological processes. The above-ground and below-ground parts of *P. divaricata* differed in their response to Cd stress. In the above-ground tissues, membrane lipid peroxidation were mainly eliminated through the action of various antioxidants. At the same time, in the roots, mainly Cd stress tolerance was exerted by reduced enzyme production. Diffusion limitations (stomatal limitation and mesophyll conductance limitation) mainly contributed to the total photosynthetic limitations under Cd treatment. To conclude, 5–75  $\mu$ M Cd did not directly affect the photosynthesis of *P. divaricata*. The inhibition of the photosynthetic efficiency of *P. divaricata* under high Cd concentrations was indirectly caused by the impact of Cd on leaf morphology.

**Supplementary Materials:** The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/agronomy13102601/s1. Refs. [40–43] are listed in File S1.

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