Supplementary Materials

Dibutyl Itaconate and Lauryl Methacrylate Copolymers by Emulsion Polymerization for Development of Sustainable Pressure-Sensitive Adhesives

Carlos Rafael Casas-Soto [1], Alain Salvador Conejo-Dávila [1], Velia Osuna [2], David Chávez-Flores [3], José Carlos Espinoza-Hicks [3], Sergio Gabriel Flores-Gallardo [1] and Alejandro Vega-Rios [1,*]

[1] Departament of Engineering and Materials Chemistry, Centro de Investigación en Materiales Avanzados, SC, Miguel de Cervantes No. 120, Chihuahua C.P. 31136, Mexico; carlos.casas@cimav.edu.mx (C.R.C.-S.); alain.conejo@cimav.edu.mx (A.S.C.-D.); sergio.flores@cimav.edu.mx (S.G.F.-G.)

[2] Consejo Nacional de Ciencia y Tecnología (CONACyT)-Centro de Investigación en Materiales Avanzados, SC (CIMAV), Miguel de Cervantes No. 120, Chihuahua C.P. 31136, Mexico; velia.osuna@cimav.edu.mx

[3] Facultad de Ciencias en Química, Universidad Autonóma de Chihuahua, Chihuahua C.P. 31125, Mexico; dchavezf@uach.mx (D.C.-F.); jhicks@uach.mx (J.C.E.-H.)

* Correspondence: alejandro.vega@cimav.edu.mx; Tel.: +52-01-614-439-4831
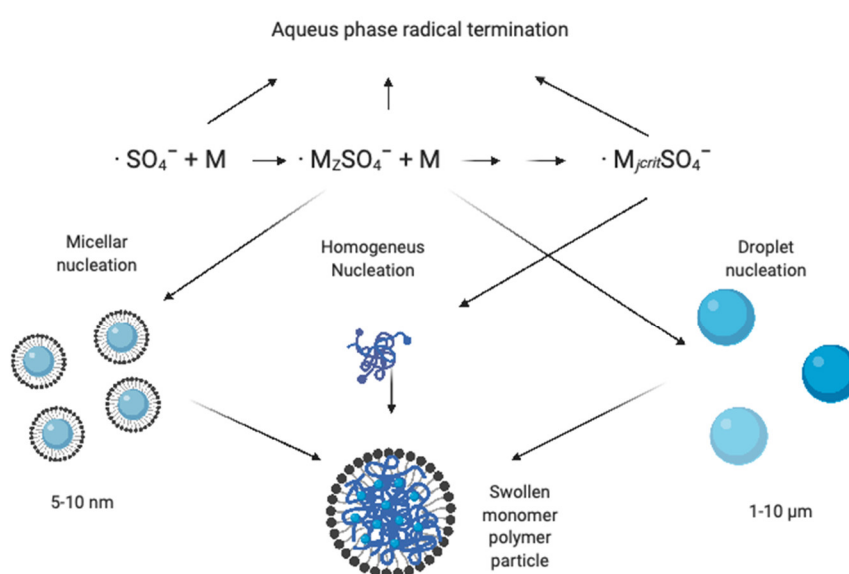
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Figure S1. Mechanisms of particle nucleation formation.
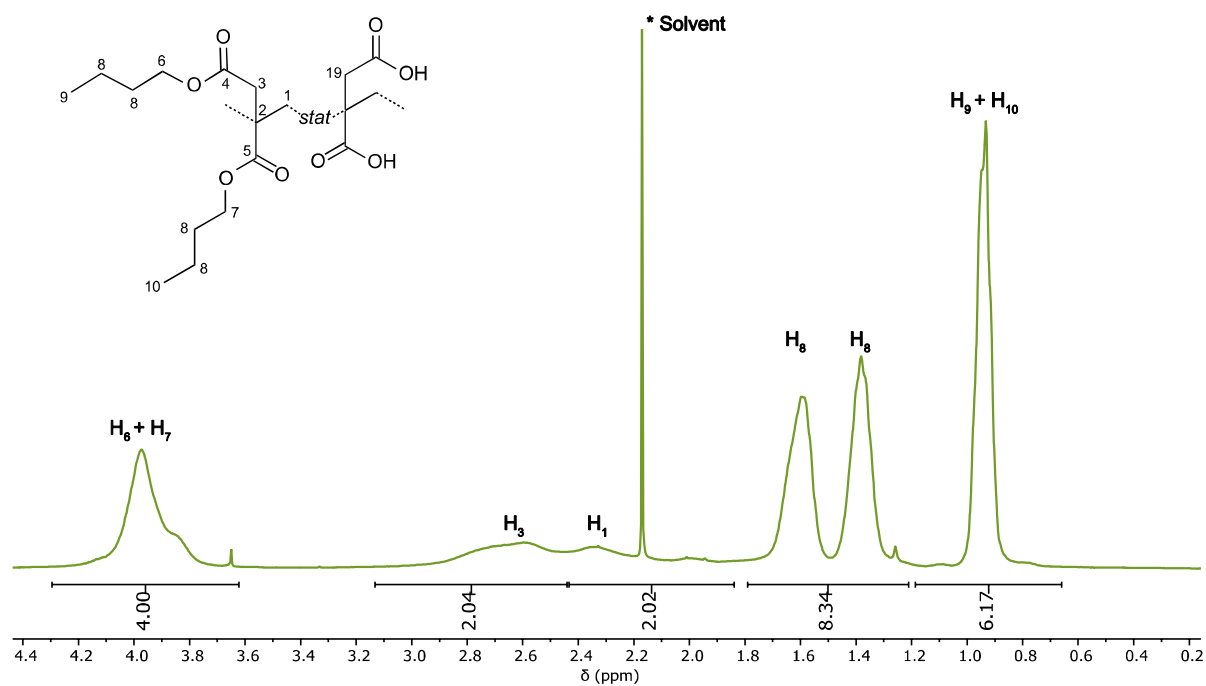
Figure S2. ^{1}H-NMR spectrum of PSA-1000 [poly(DBI-stat-IA) (99 wt. % DBI : 1 wt.% IA)] in CDCl$_3$.
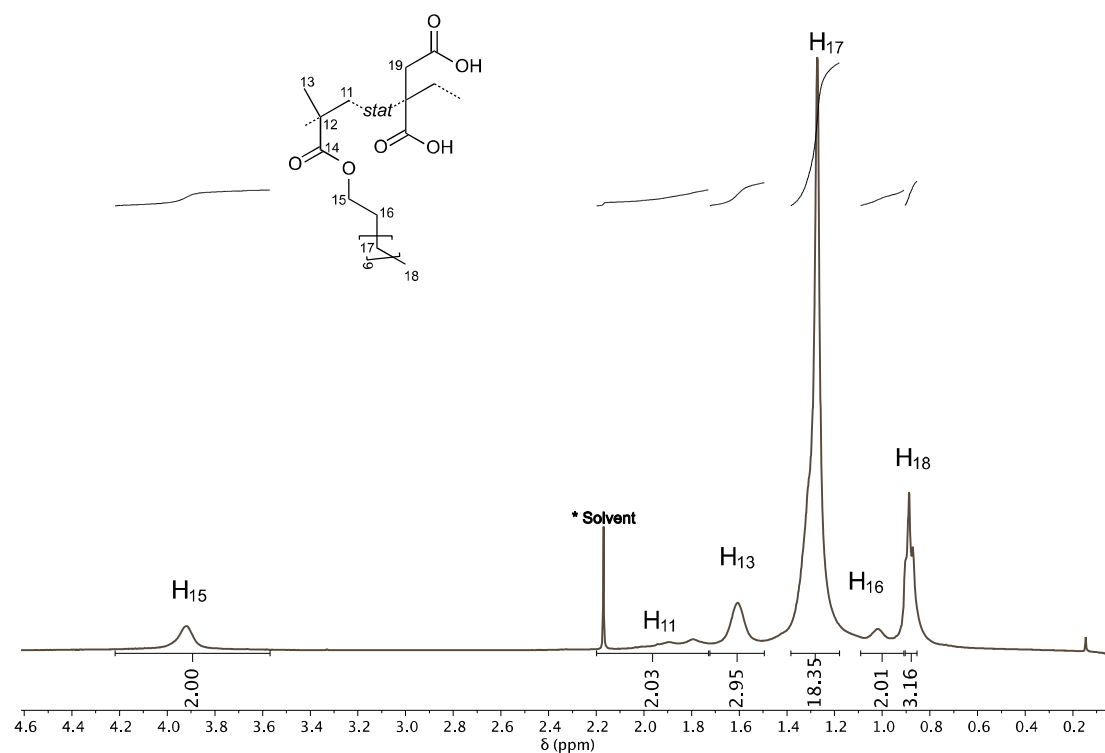
Figure S3. ^{1}H-NMR spectrum of PSA-5000 [poly(LMA-stat-IA) (99 wt. % LMA: 1 wt.% IA)] in CDCl3.
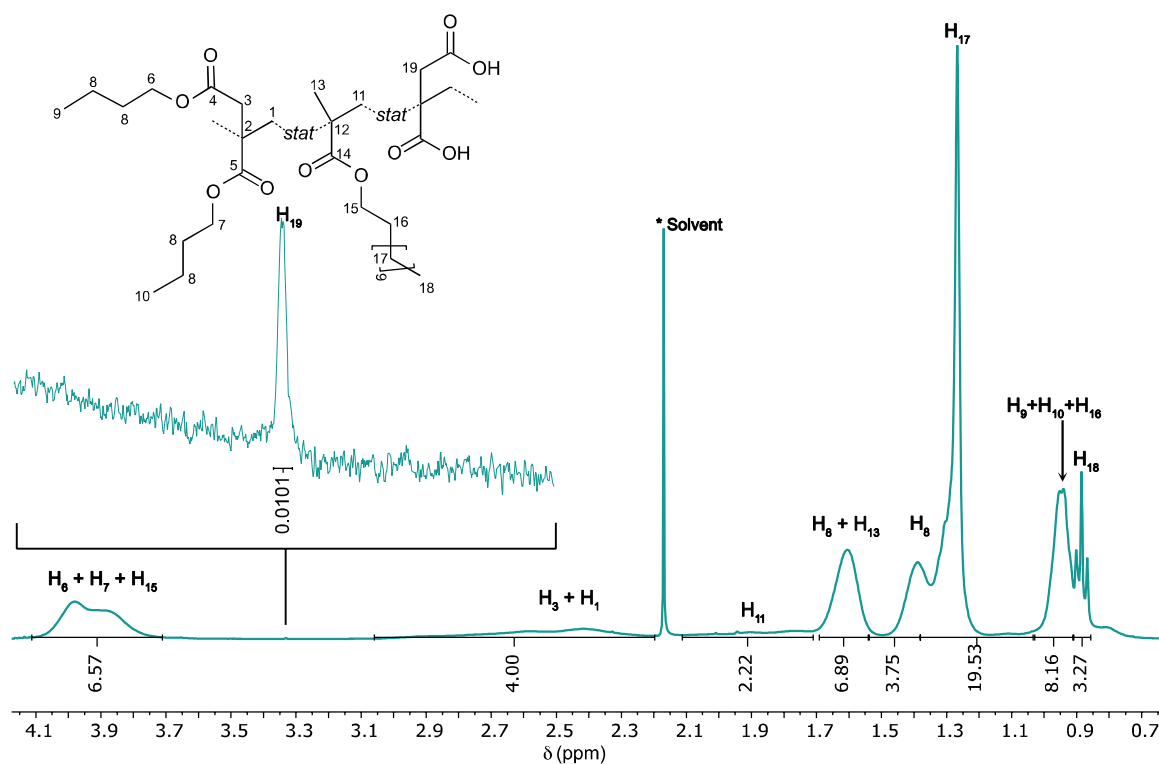
Figure S4. ¹H-NMR spectrum of PSA 3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in CDCl₃.
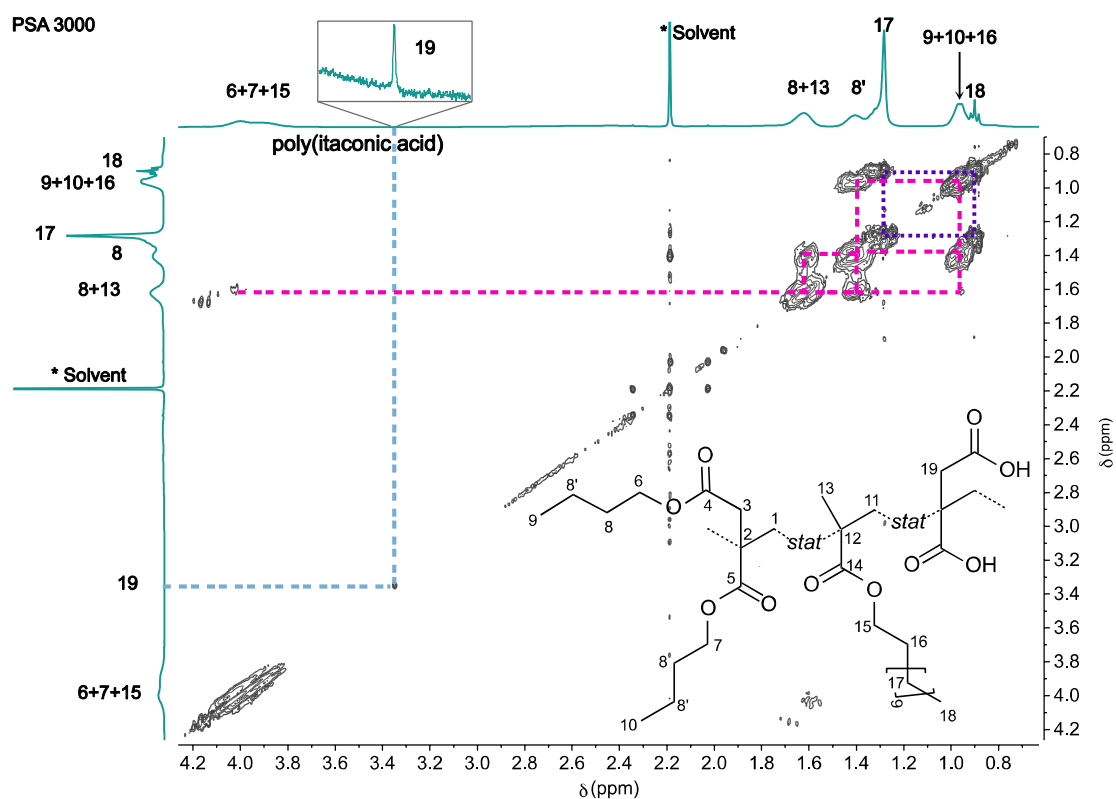
Figure S5. ¹H-¹H COSY of PSA-3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in CDCl₃.
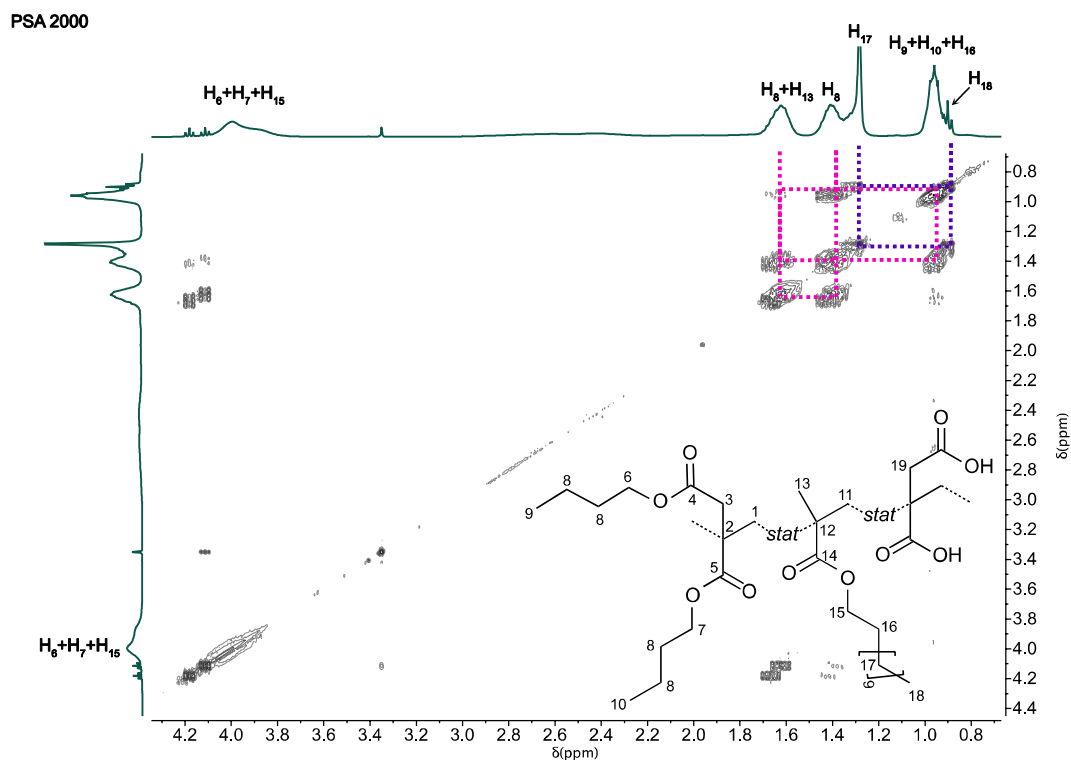
Figure S6. ^{1}H-^{1}H COSY of PSA-2000 [poly(DBI-stat-LMA-stat-IA) (74 wt. % DBI : 25 wt. % LMA: 1 wt.% IA)] in CDCl₃.
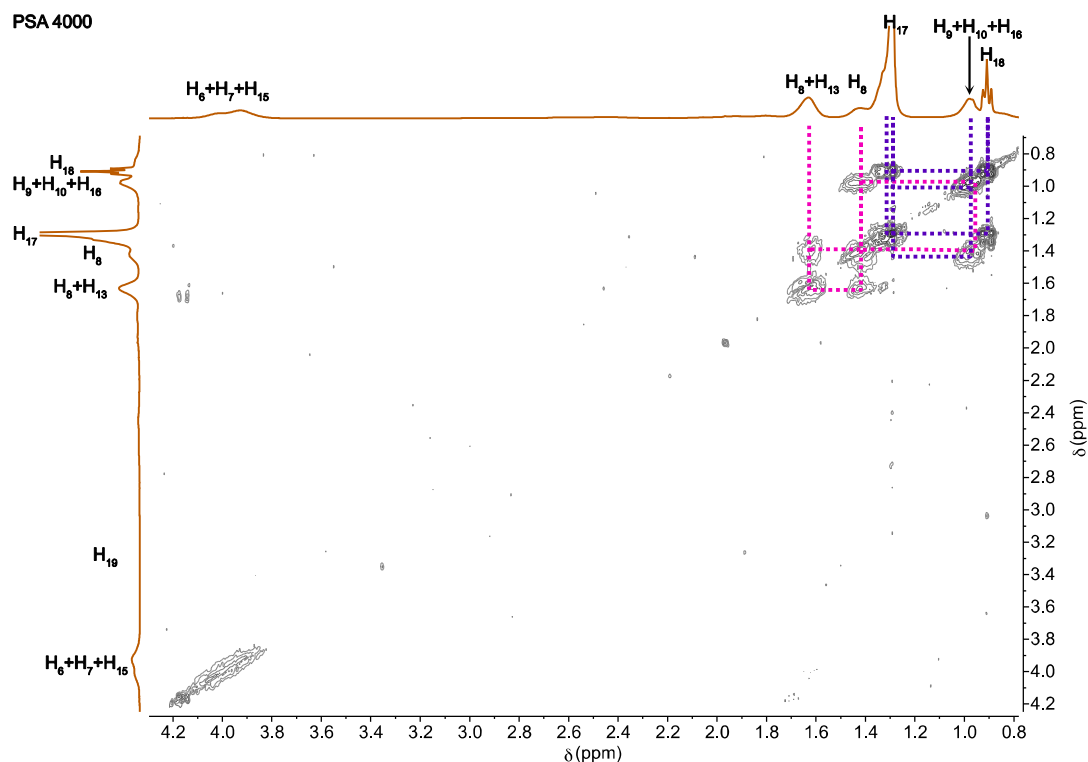
Figure S7. ^{1}H-^{1}H COSY of PSA-4000 [poly(DBI-*stat*-LMA-*stat*-IA) (24 wt. % DBI : 75 wt. % LMA: 1 wt.% IA)] in CDCl₃.
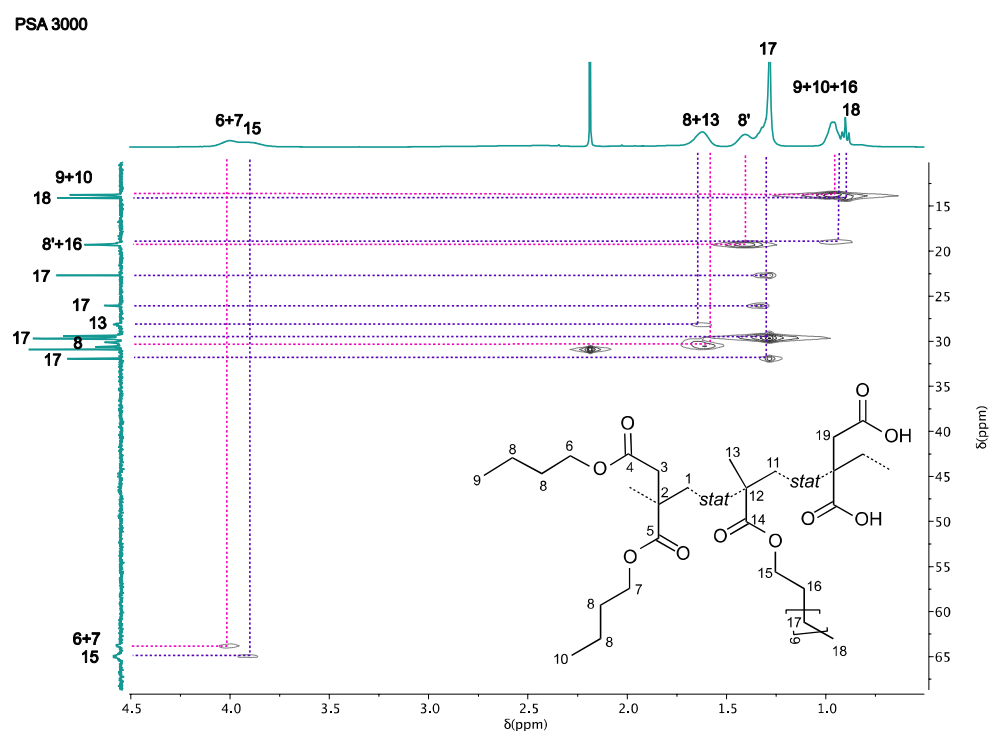
Figure S8. ¹H-¹³C HSQC of PSA-3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in CDCl₃.
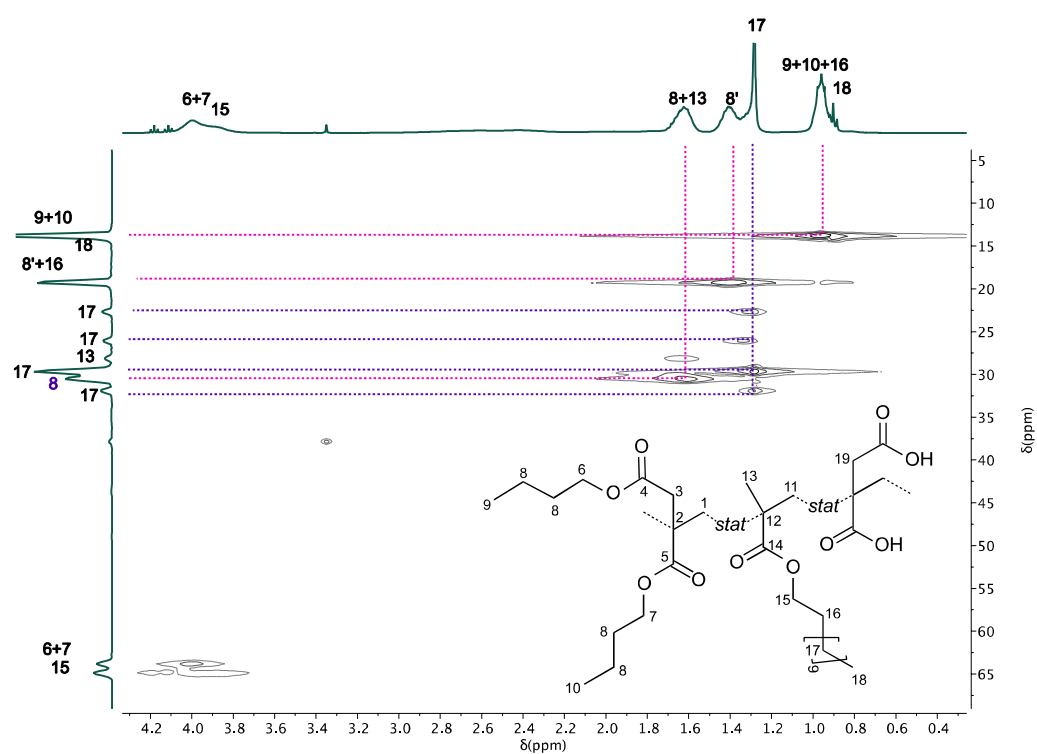
Figure S9. ¹H-¹³C HSQC of PSA-2000 [poly(DBI-stat-LMA-stat-IA) (74 wt. % DBI : 25 wt. % LMA: 1 wt.% IA)] in CDCl₃.
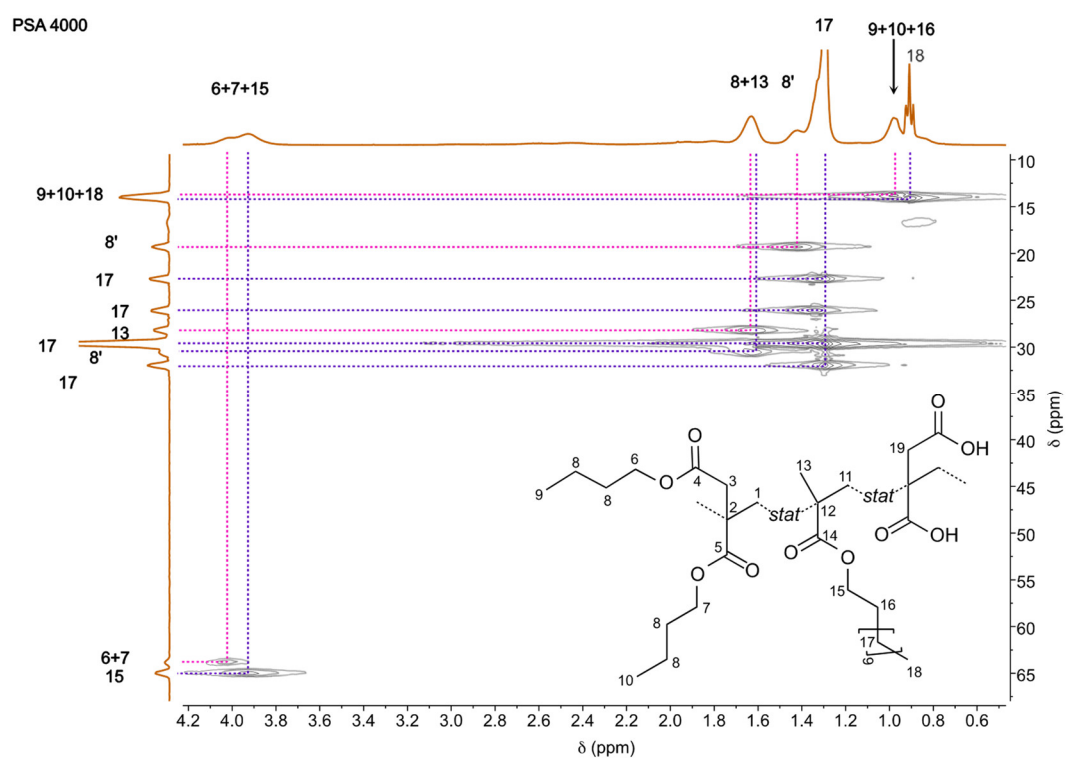
Figure S10. ^{1}H-^{13}C HSQC of PSA-4000 [poly(DBI-*stat*-LMA-*stat*-IA) (24 wt. % DBI : 75 wt. % LMA: 1 wt.% IA)] in CDCl$_3$.
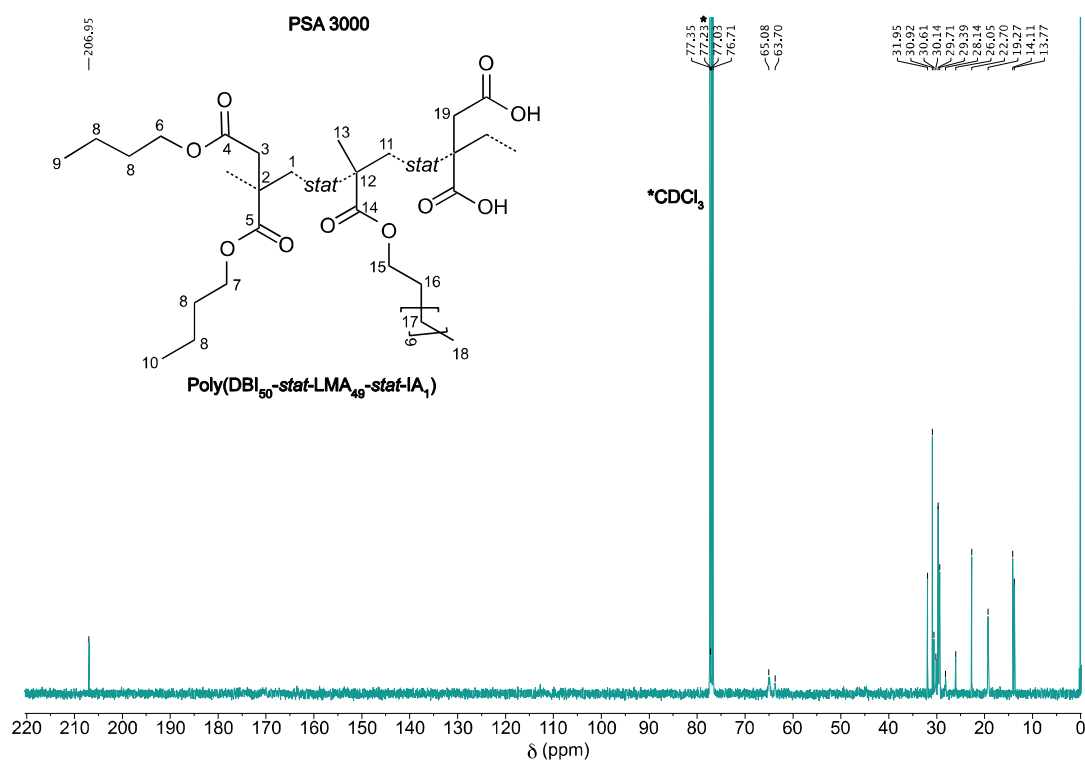
Figure S11. ^{13}C-NMR spectrum of PSA 3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in CDCl$_3$.

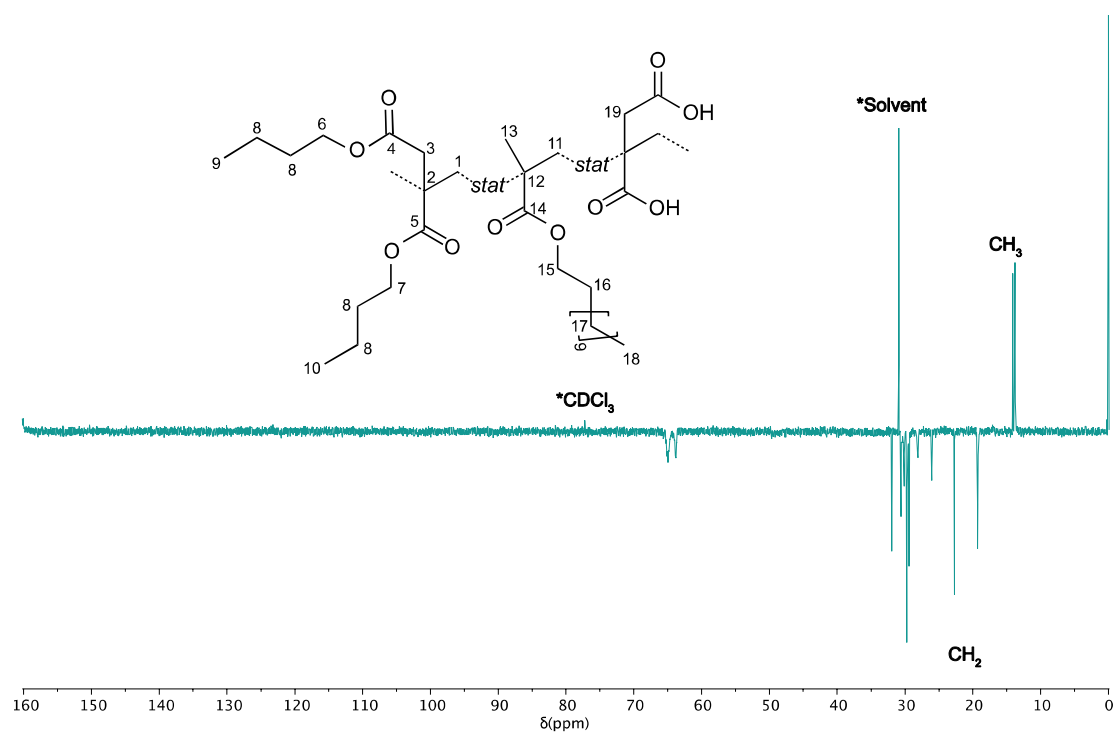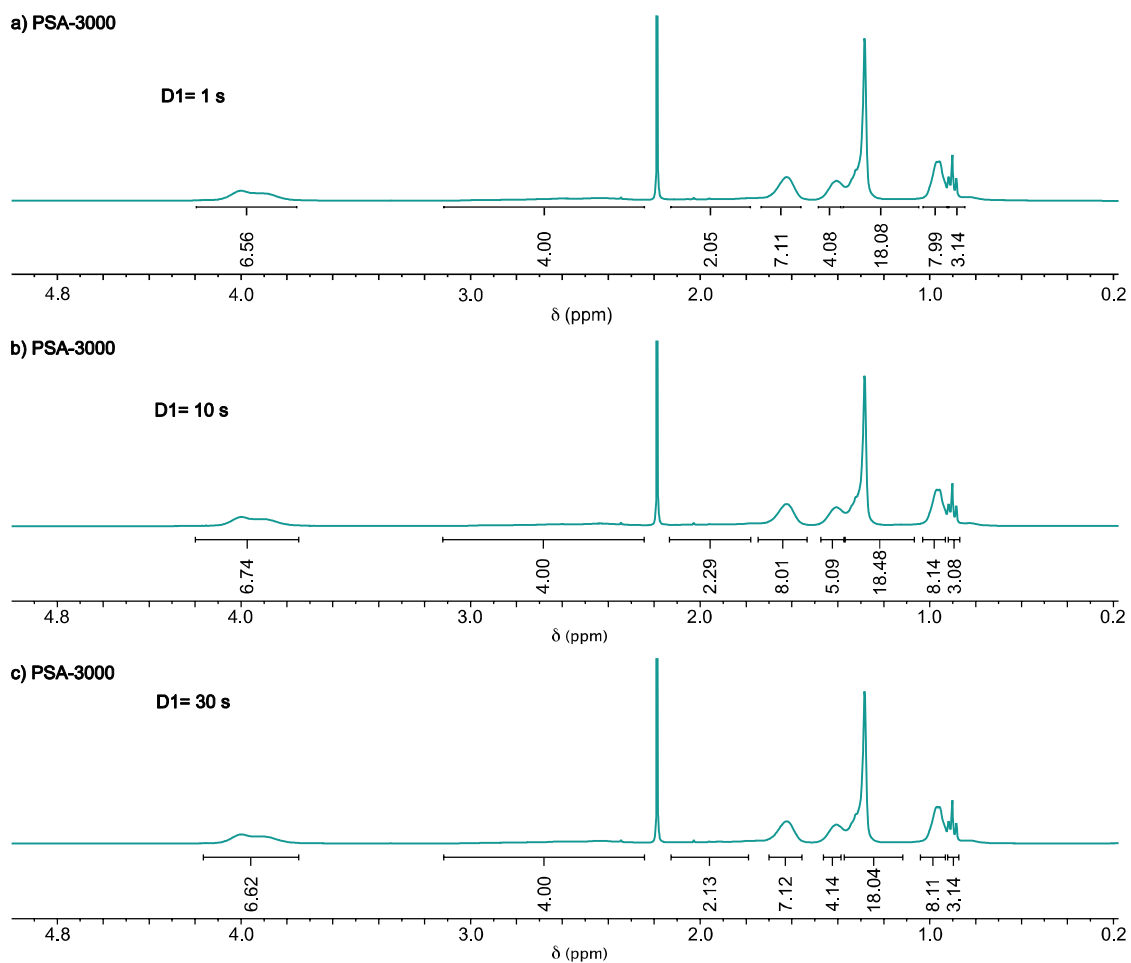**Figure S12.** DEPT-135 of PSA-3000 sample.

Figure S13. Analysis of PSA-3000 at different relaxation times (D1).

Table S1. Area and number of protons used to calculate the composition of copolymers.

Sample	IA		DBI		LMA	
	A_{19}	P 19	A (1+3)	P (1+3)	A_{18}	P 18
DB1	0.0070	2	4.0540	4	0	0
LMMA	0.0145	2	0	0	3.1556	3
PDBI 49	0.0050	2	4.0010	4	3.2662	3
PDBI 74	0.0076	2	2.7045	4	0.9360	3
PDBI 24	0.0064	2	1.0850	4	2.6337	3

Binary copolymers

$$\% \, IA = \frac{\frac{A_{19}}{P19}}{\frac{A_{19}}{P19} + \frac{A_{1+3}}{P(1+3)}} \times 100$$

$$\% \, DBI = \frac{\frac{A_{1+3}}{P(1+3)}}{\frac{A_{19}}{P19} + \frac{A_{1+3}}{P(1+3)}} \times 100$$

$$\% \, LMMA = \frac{\frac{A_{18}}{P(18)}}{\frac{A_{19}}{P19} + \frac{A_{18}}{P(18)}} \times 100$$

Ternary copolymers

$$\% \, IA = \frac{\frac{A_{19}}{P19}}{\frac{A_{19}}{P19} + \frac{A_{1+3}}{P(1+3)} + \frac{A_{18}}{P(18)}} \times 100$$

$$\% \, DBI = \frac{\frac{A_{1+3}}{P(1+3)}}{\frac{A_{19}}{P19} + \frac{A_{1+3}}{P(1+3)} + \frac{A_{18}}{P18}} \times 100$$

$$\% \, LMMA = \frac{\frac{A_{18}}{P(18)}}{\frac{A_{19}}{P19} + \frac{A_{18}}{P(18)} + \frac{A_{1+3}}{P(1+3)}} \times 100$$
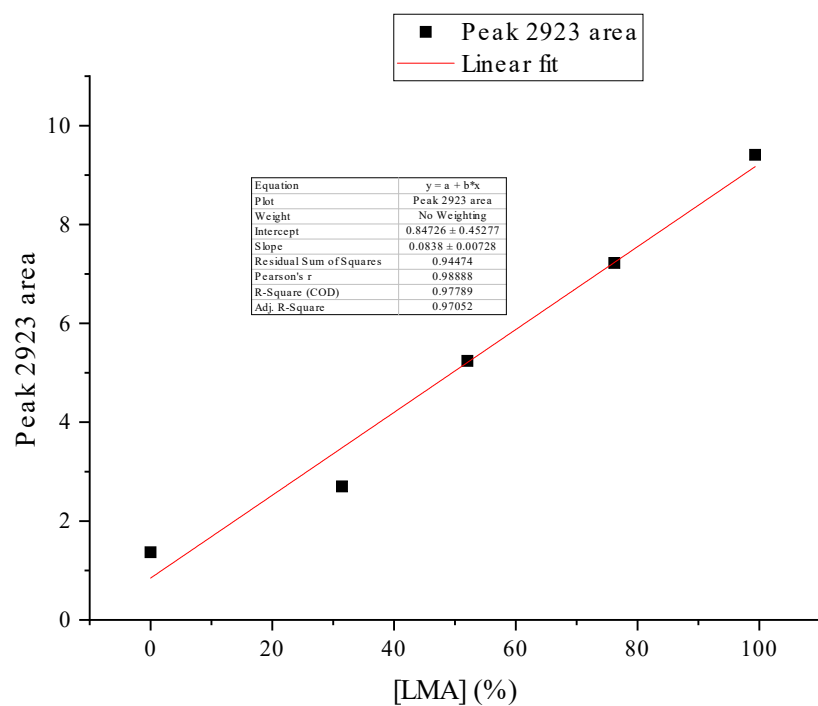
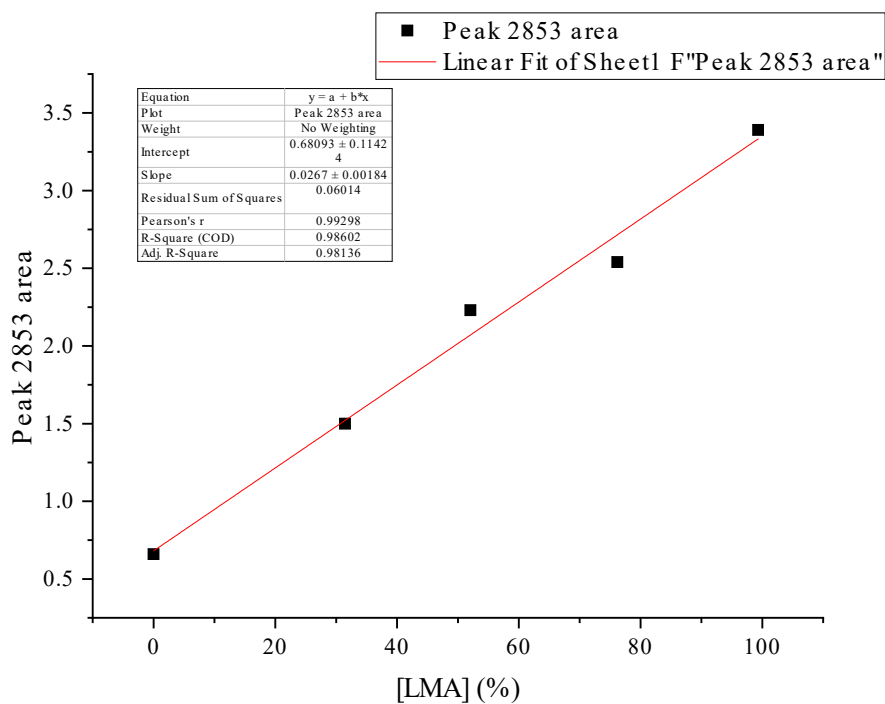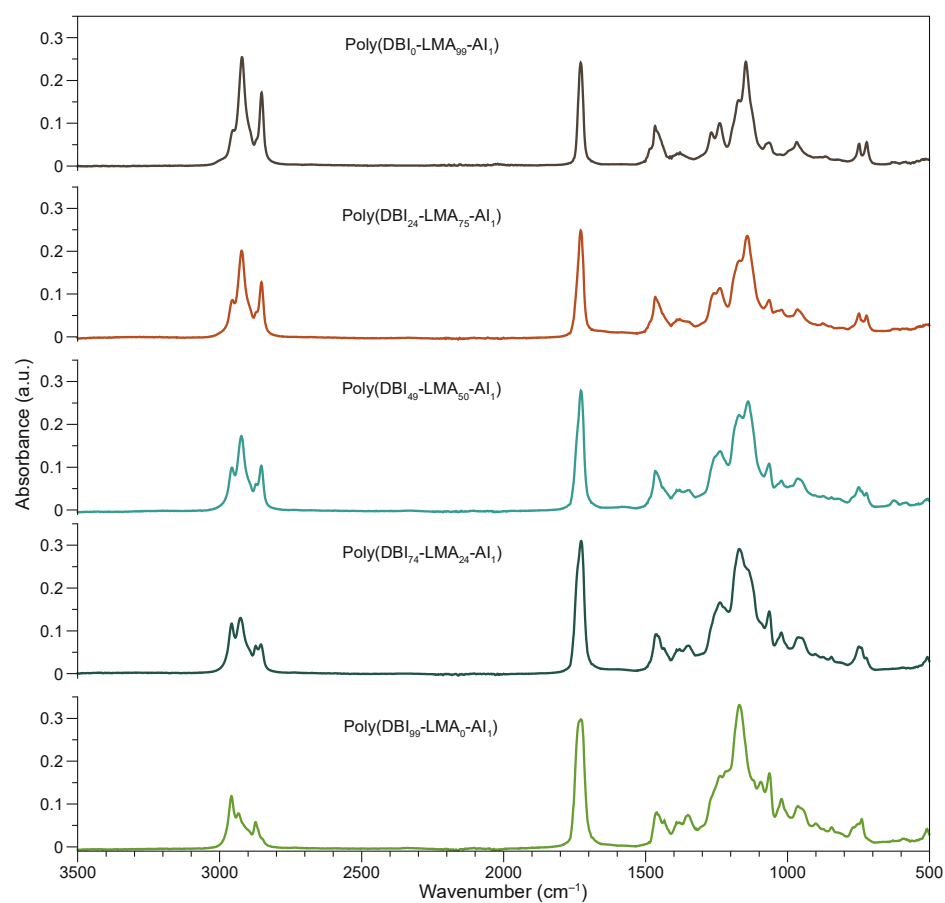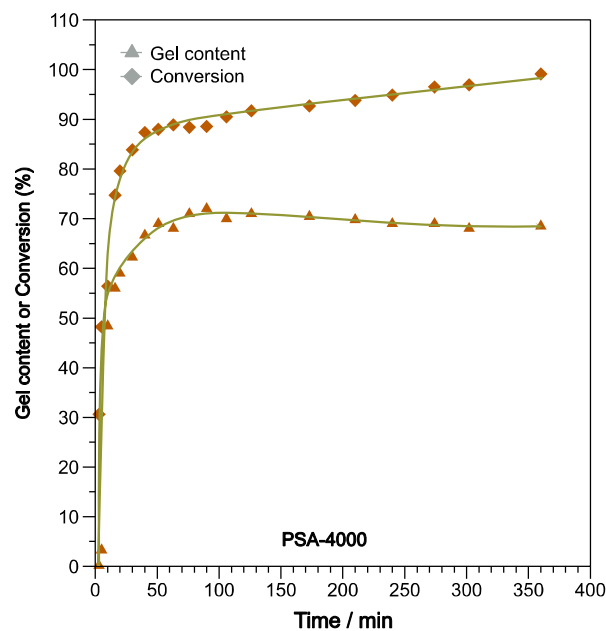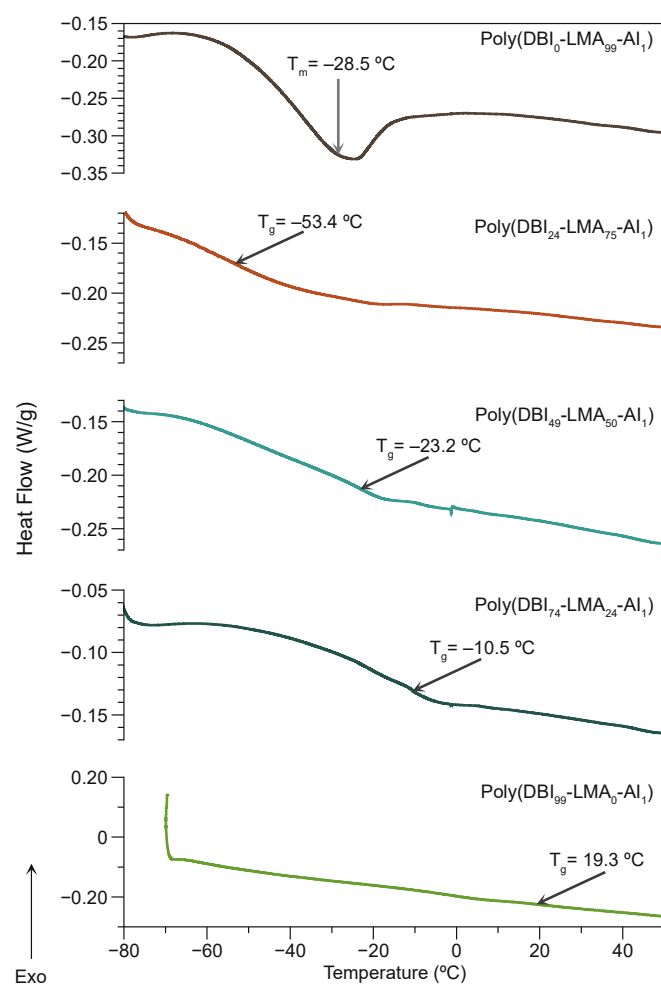**Figure S14.** Area correlation of the band 2923 cm^{-1} as a function of the LMA content.

Figure S15. Area correlation of the band 2853 cm^{-1} as a function of the LMA content.

Figure S16. FTIR spectra of copolymers films. a) PSA-5000; b) PSA-4000; c) PSA-3000; d) PSA-2000; e) PSA-1000.

Figure S17. Kinetic conversion study in relation to the gel content of the PSA-4000 sample.

Figure S18. DSC thermograms of copolymers films. a) PSA-5000; b) PSA-4000; c) PSA-3000; d) PSA-2000; e) PSA-1000.