

# Dibutyl Itaconate and Lauryl Methacrylate Copolymers by Emulsion Polymerization for Development of Sustainable Pressure-Sensitive Adhesives

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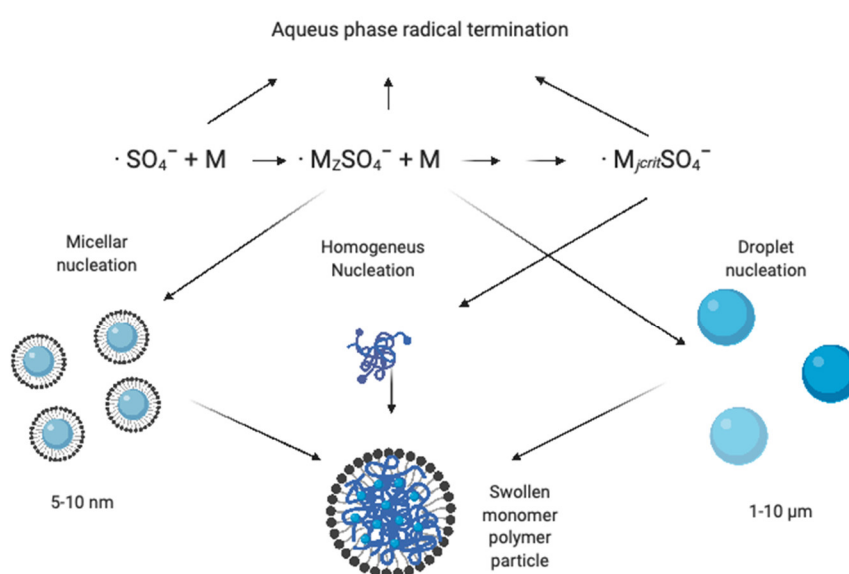
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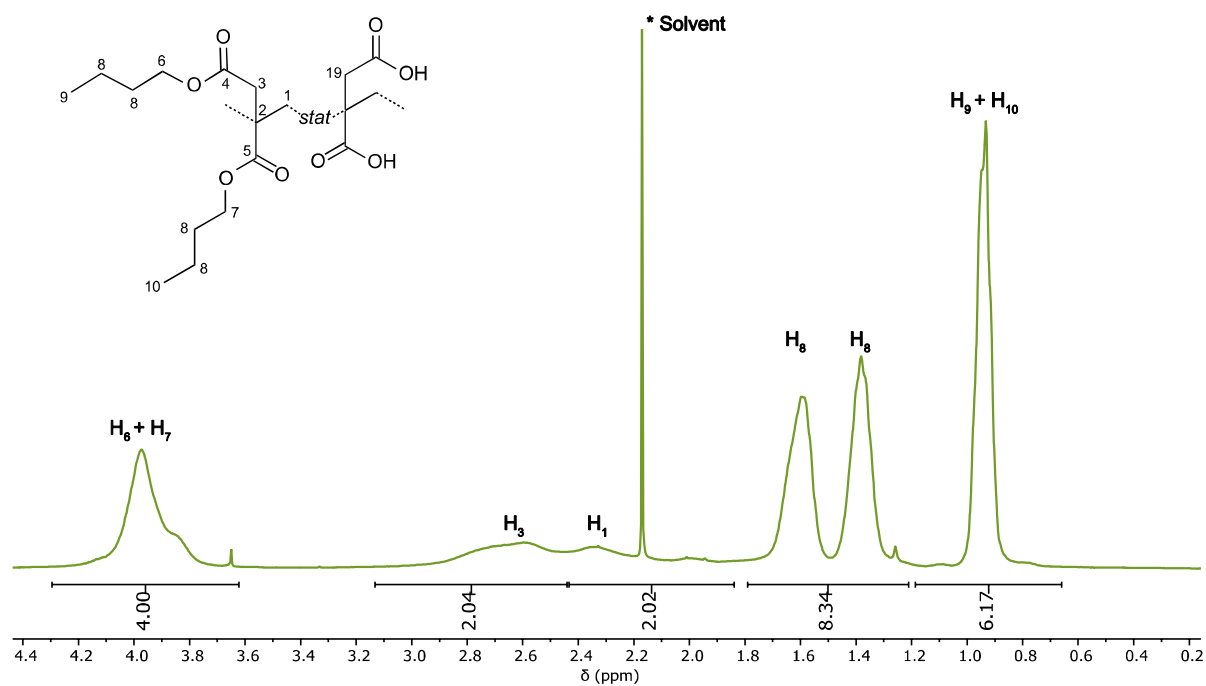
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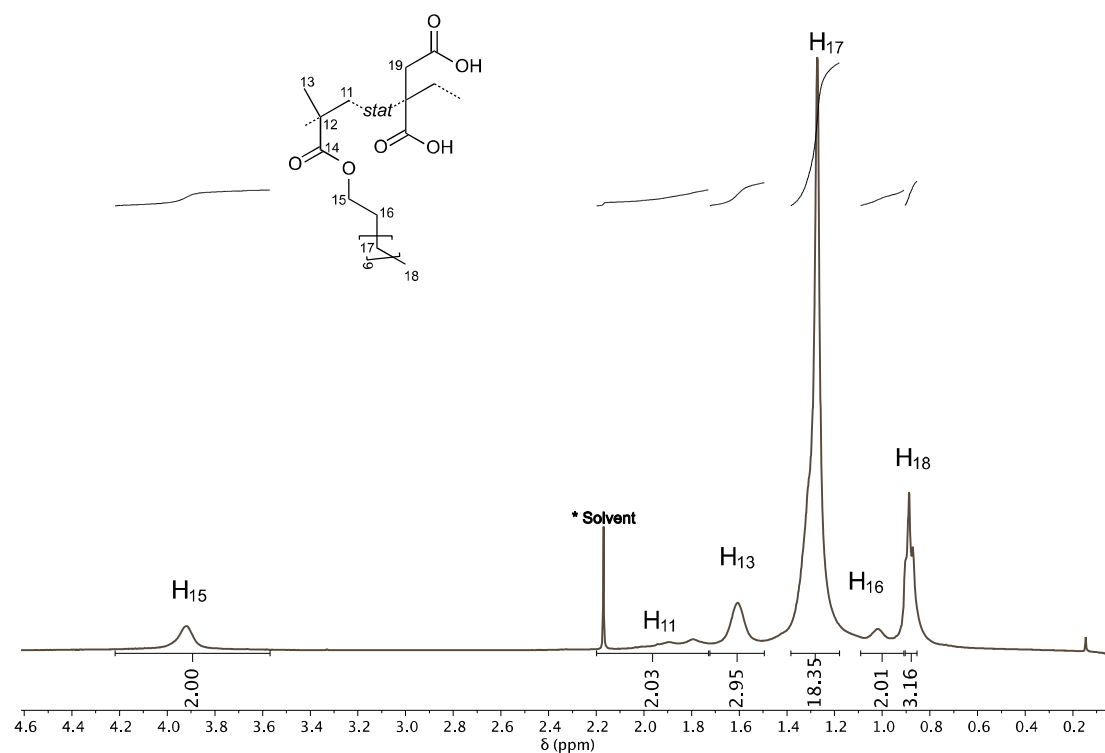
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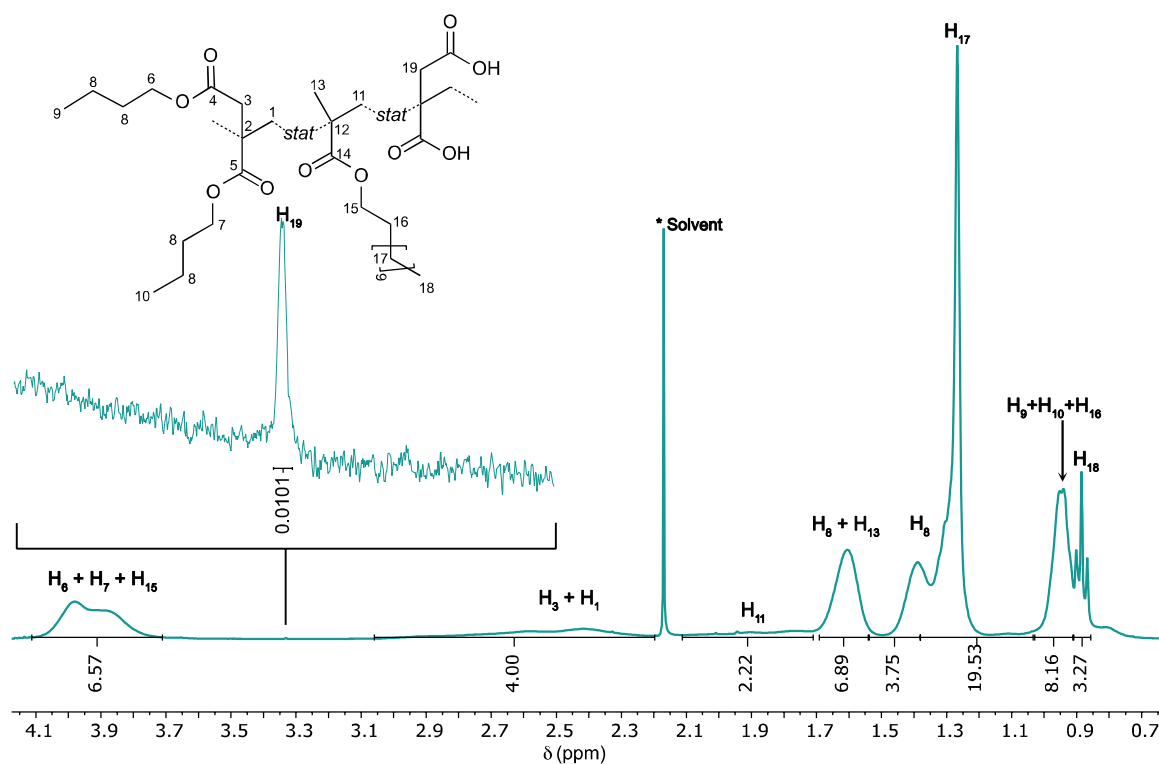
**Figure S1.** Mechanisms of particle nucleation formation.



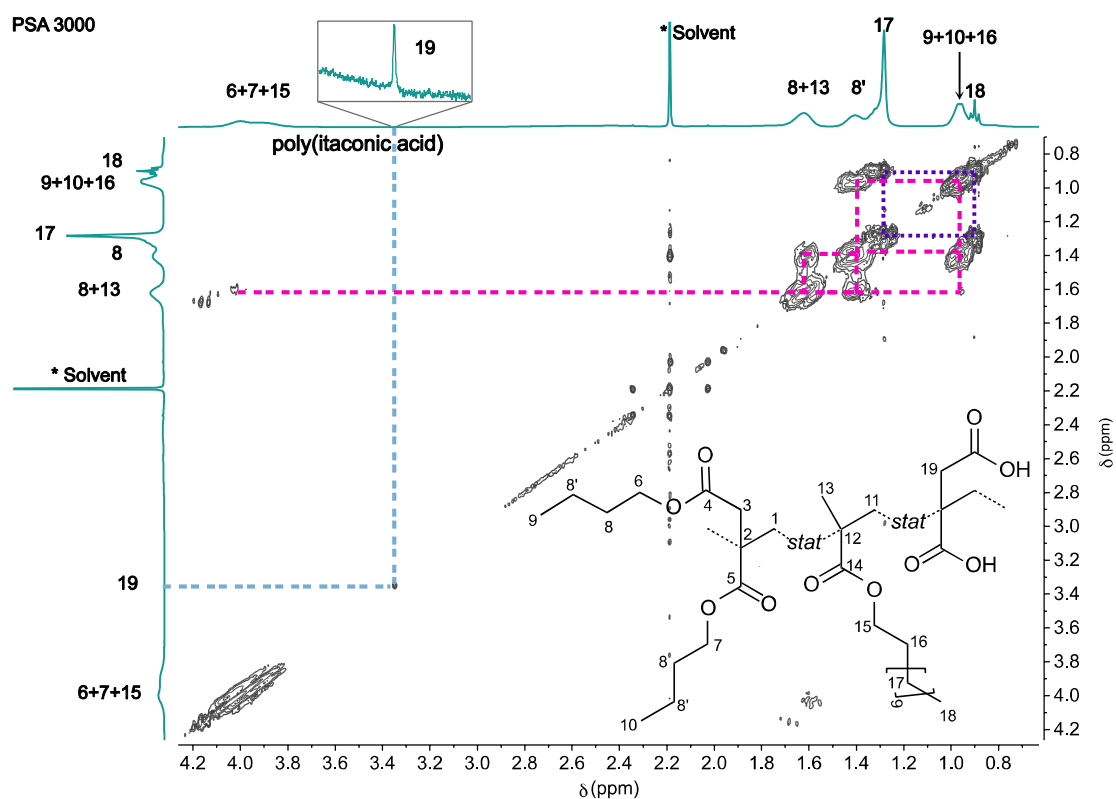
**Figure S2.**  $^1\text{H}$ -NMR spectrum of PSA-1000 [poly(DBI-stat-IA) (99 wt. % DBI : 1 wt.% IA)] in  $\text{CDCl}_3$ .



**Figure S3.**  $^1\text{H}$ -NMR spectrum of PSA-5000 [poly(LMA-stat-IA) (99 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .



**Figure S4.**  $^1\text{H}$ -NMR spectrum of PSA 3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .



**Figure S5.**  $^1\text{H}$ - $^1\text{H}$  COSY of PSA-3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .

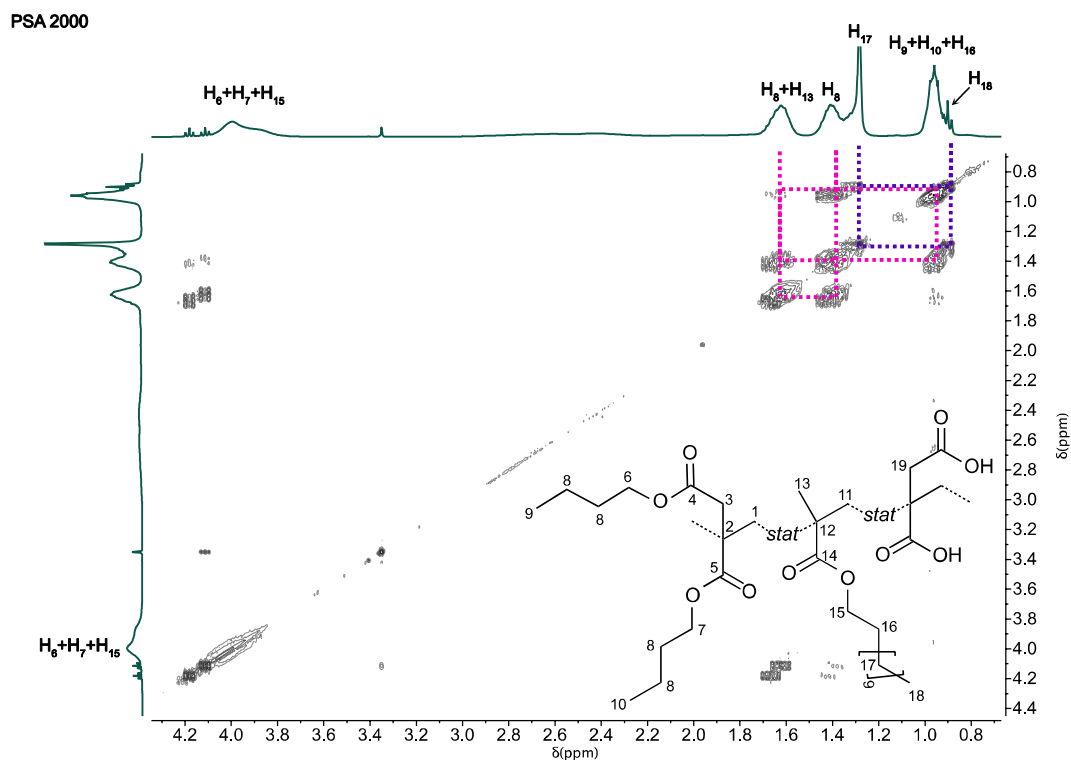


Figure S6.  $^1\text{H}$ - $^1\text{H}$  COSY of PSA-2000 [poly(DBI-*stat*-LMA-*stat*-IA) (74 wt. % DBI : 25 wt. % LMA : 1 wt.% IA)] in  $\text{CDCl}_3$ .

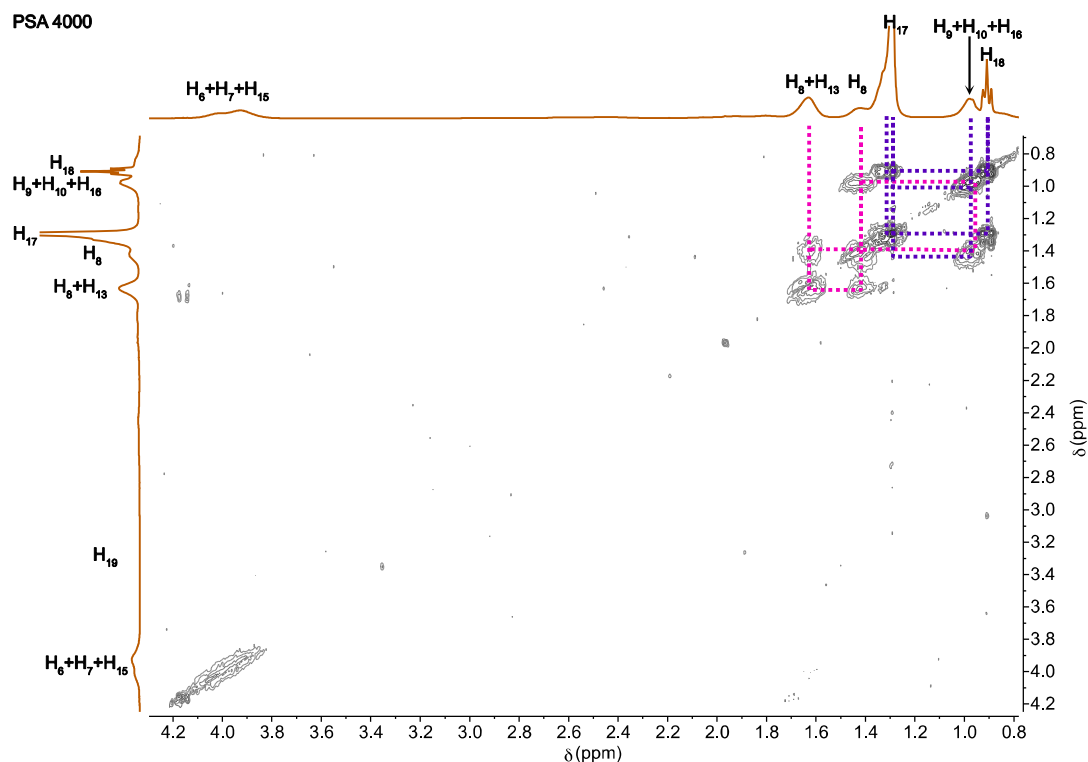


Figure S7.  $^1\text{H}$ - $^1\text{H}$  COSY of PSA-4000 [poly(DBI-*stat*-LMA-*stat*-IA) (24 wt. % DBI : 75 wt. % LMA : 1 wt.% IA)] in  $\text{CDCl}_3$ .

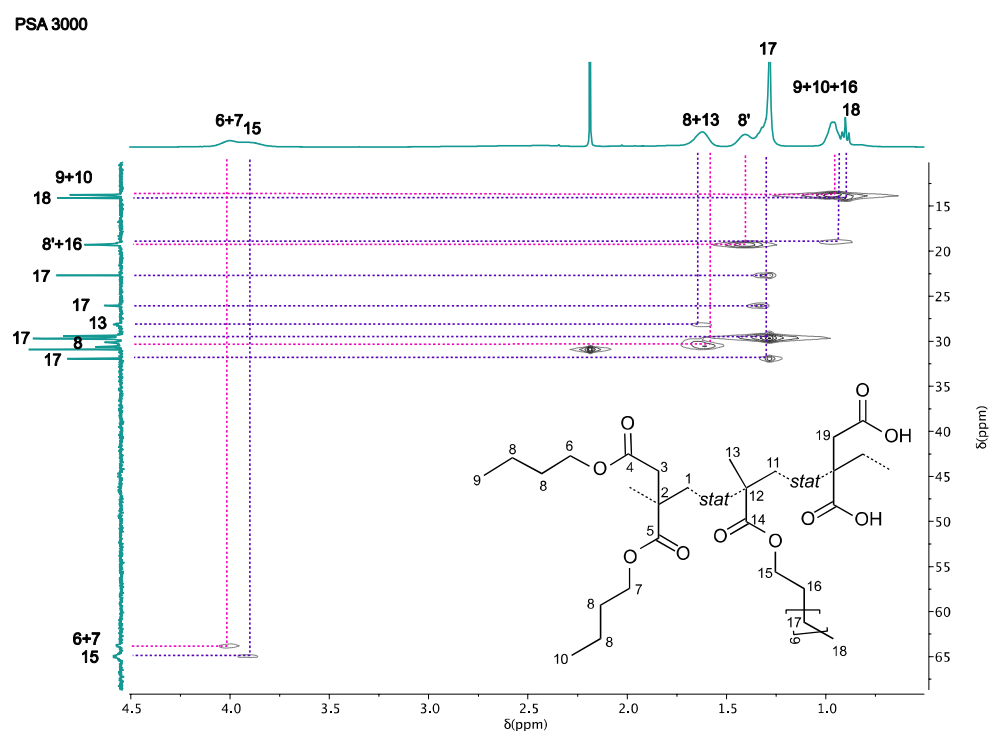


Figure S8.  $^1\text{H}$ - $^{13}\text{C}$  HSQC of PSA-3000 [poly(DBI-stat-LMA-stat-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .

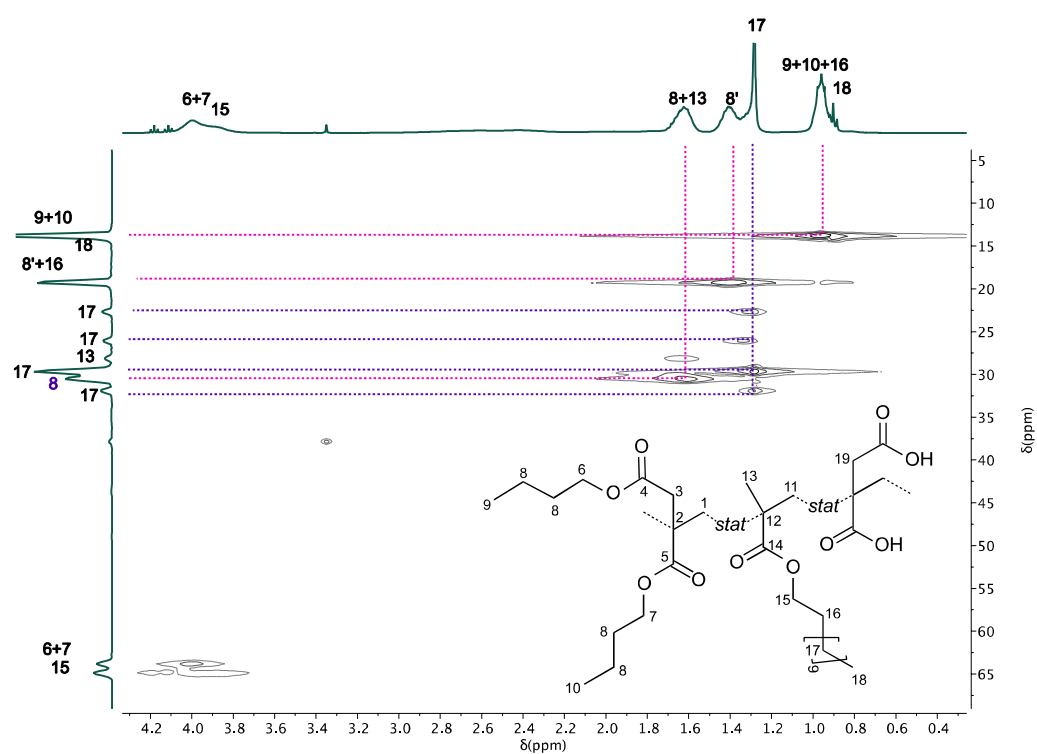
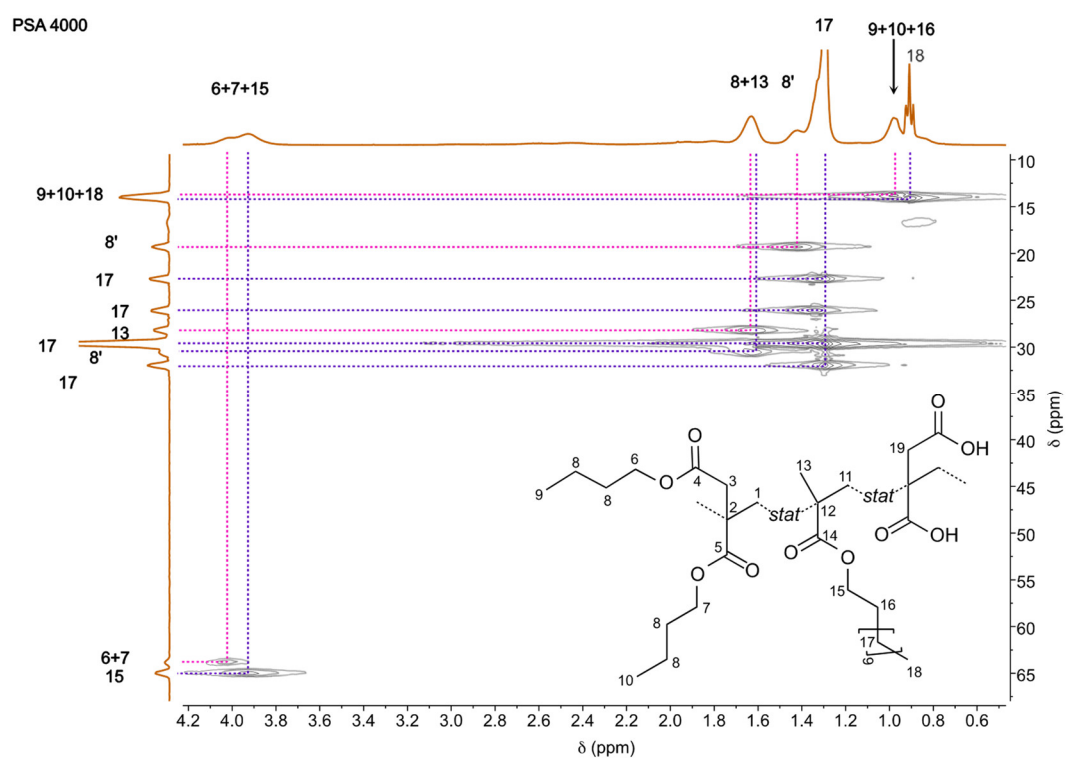
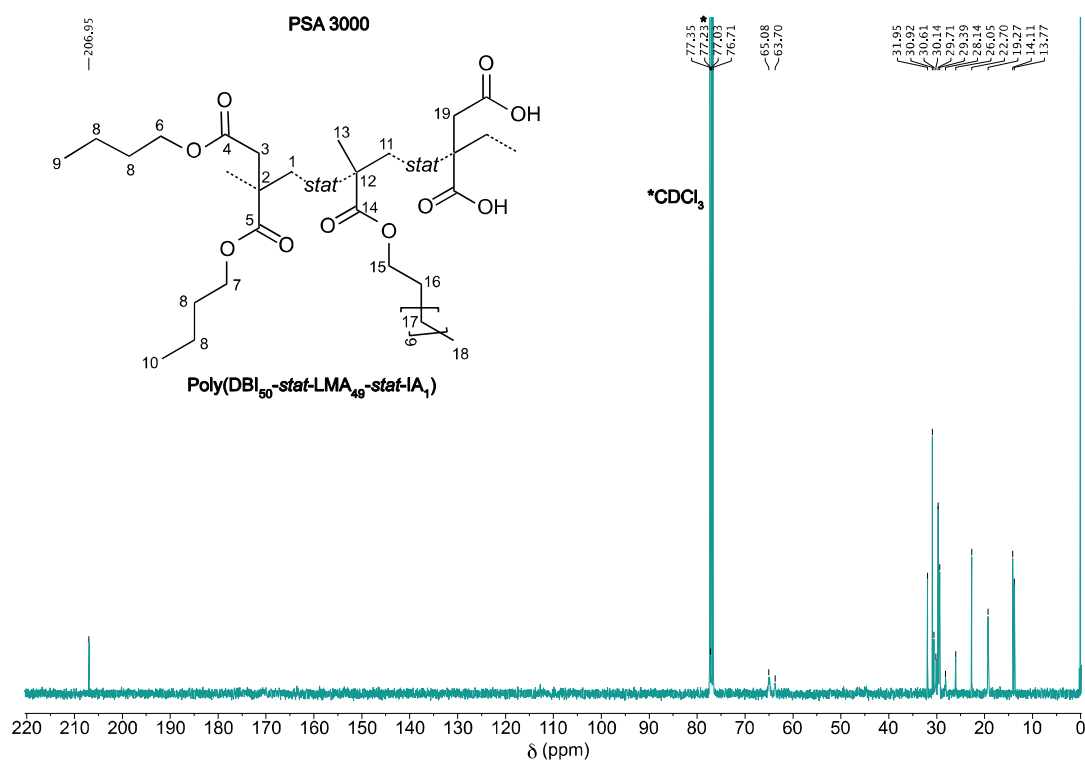


Figure S9.  $^1\text{H}$ - $^{13}\text{C}$  HSQC of PSA-2000 [poly(DBI-stat-LMA-stat-IA) (74 wt. % DBI : 25 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .



**Figure S10.**  $^1\text{H}$ - $^{13}\text{C}$  HSQC of PSA-4000 [poly(DBI-*stat*-LMA-*stat*-IA) (24 wt. % DBI : 75 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .



**Figure S11.**  $^{13}\text{C}$ -NMR spectrum of PSA 3000 [poly(DBI-*stat*-LMA-*stat*-IA) (49 wt. % DBI : 50 wt. % LMA: 1 wt.% IA)] in  $\text{CDCl}_3$ .

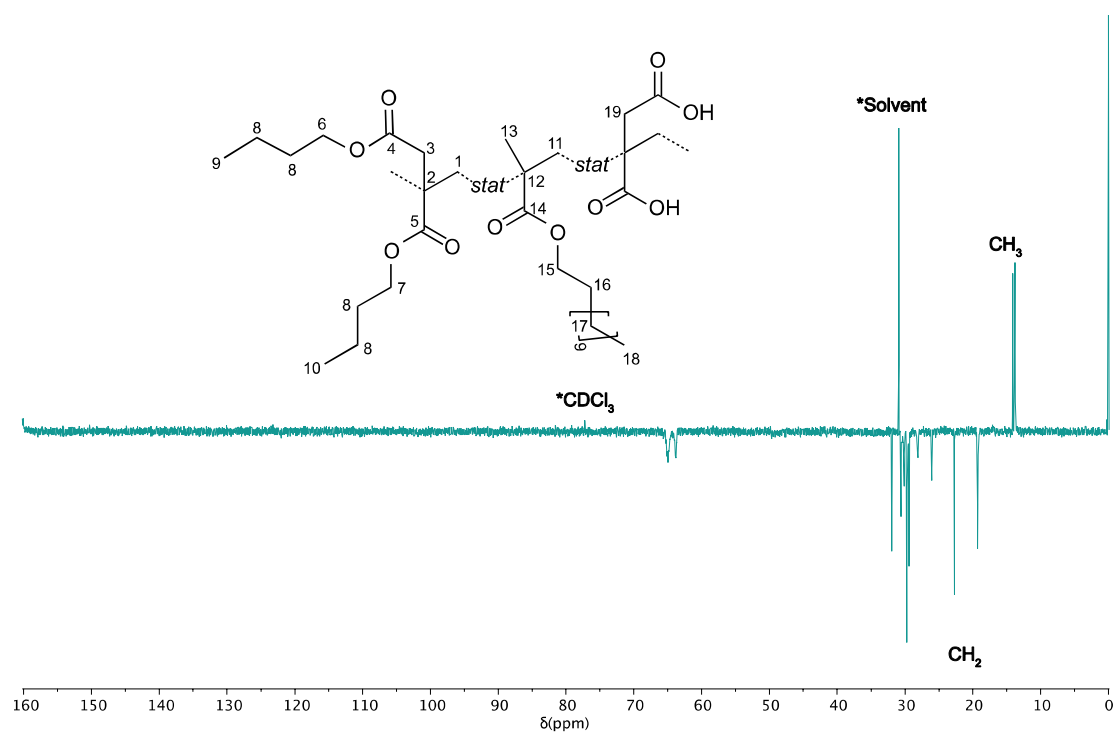
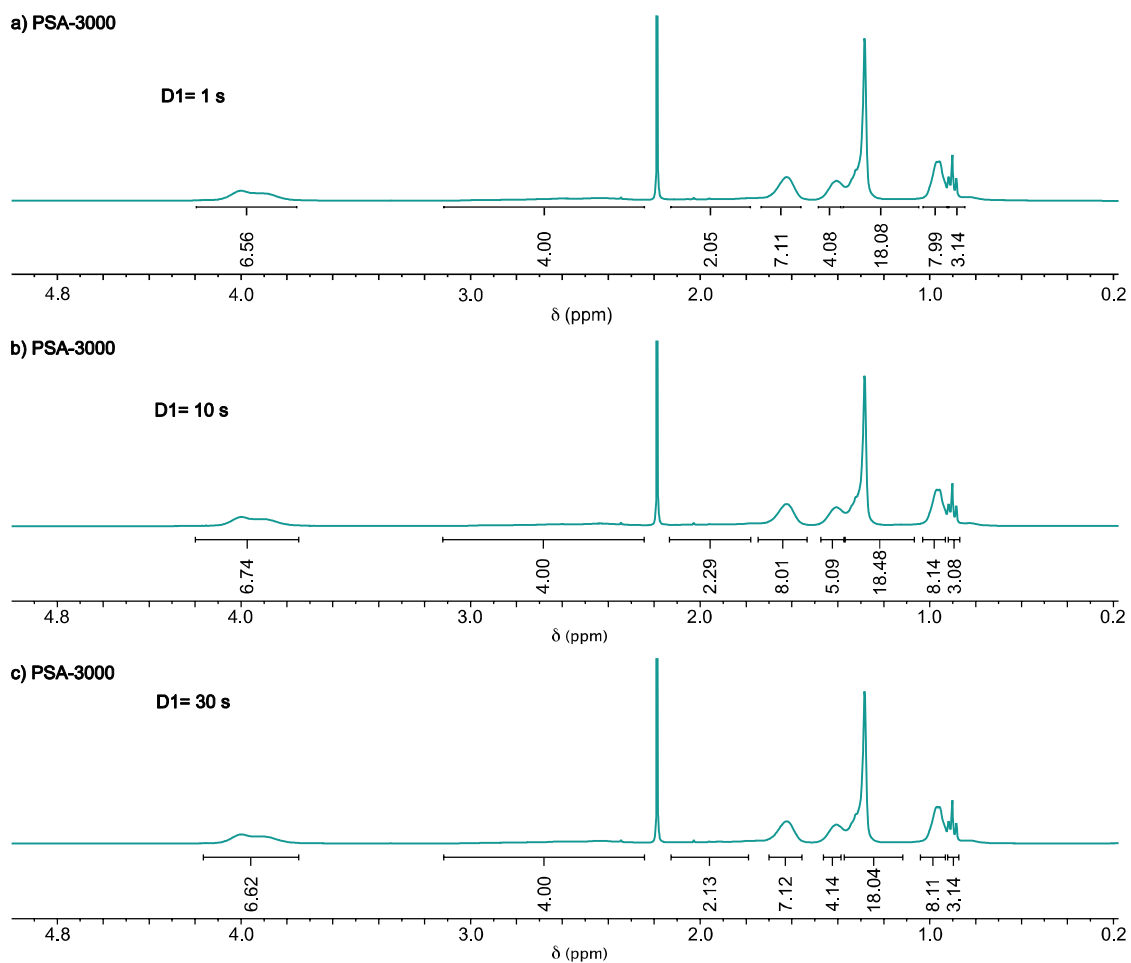


Figure S12. DEPT-135 of PSA-3000 sample.

Figure S13. Analysis of PSA-3000 at different relaxation times ( $D1$ ).



**Table S1.** Area and number of protons used to calculate the composition of copolymers.

Sample	IA		DBI		LMA	
	A <sub>19</sub>	P 19	A (1+3)	P (1+3)	A <sub>18</sub>	P 18
DBI	0.0070	2	4.0540	4	0	0
LMMA	0.0145	2	0	0	3.1556	3
PDBI 49	0.0050	2	4.0010	4	3.2662	3
PDBI 74	0.0076	2	2.7045	4	0.9360	3
PDBI 24	0.0064	2	1.0850	4	2.6337	3

Binary copolymers

$$\% IA = \frac{\frac{A_{19}}{P_{19}}}{\frac{A_{19}}{P_{19}} + \frac{A_{1+3}}{P(1+3)}} \times 100$$

$$\% DBI = \frac{\frac{A_{1+3}}{P(1+3)}}{\frac{A_{19}}{P_{19}} + \frac{A_{1+3}}{P(1+3)}} \times 100$$

$$\% LMMA = \frac{\frac{A_{18}}{P(18)}}{\frac{A_{19}}{P_{19}} + \frac{A_{18}}{P(18)}} \times 100$$

Ternary copolymers

$$\% IA = \frac{\frac{A_{19}}{P_{19}}}{\frac{A_{19}}{P_{19}} + \frac{A_{1+3}}{P(1+3)} + \frac{A_{18}}{P(18)}} \times 100$$

$$\% DBI = \frac{\frac{A_{1+3}}{P(1+3)}}{\frac{A_{19}}{P_{19}} + \frac{A_{1+3}}{P(1+3)} + \frac{A_{18}}{P(18)}} \times 100$$

$$\% LMMA = \frac{\frac{A_{18}}{P(18)}}{\frac{A_{19}}{P_{19}} + \frac{A_{18}}{P(18)} + \frac{A_{1+3}}{P(1+3)}} \times 100$$

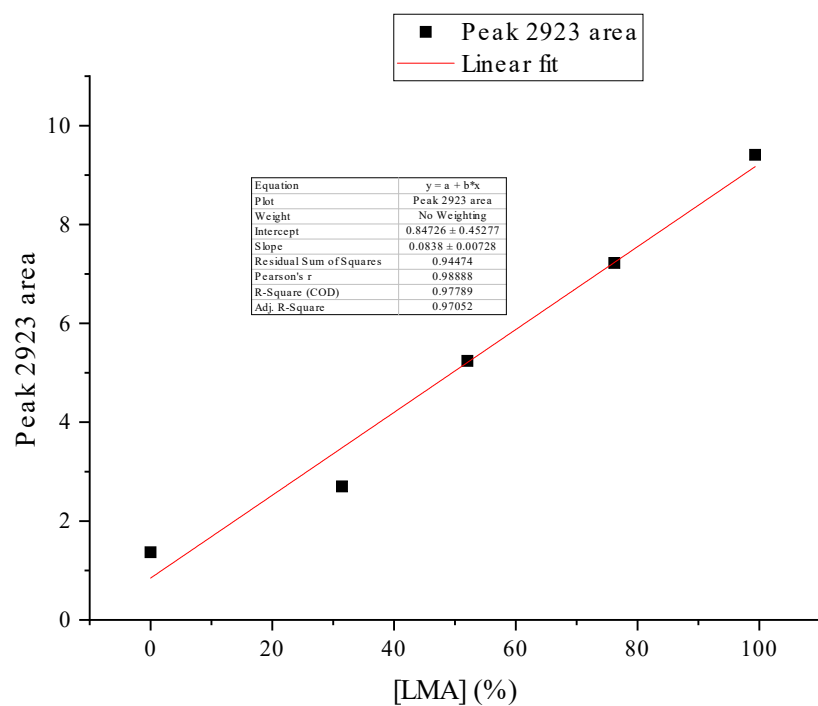


Figure S14. Area correlation of the band 2923  $\text{cm}^{-1}$  as a function of the LMA content.

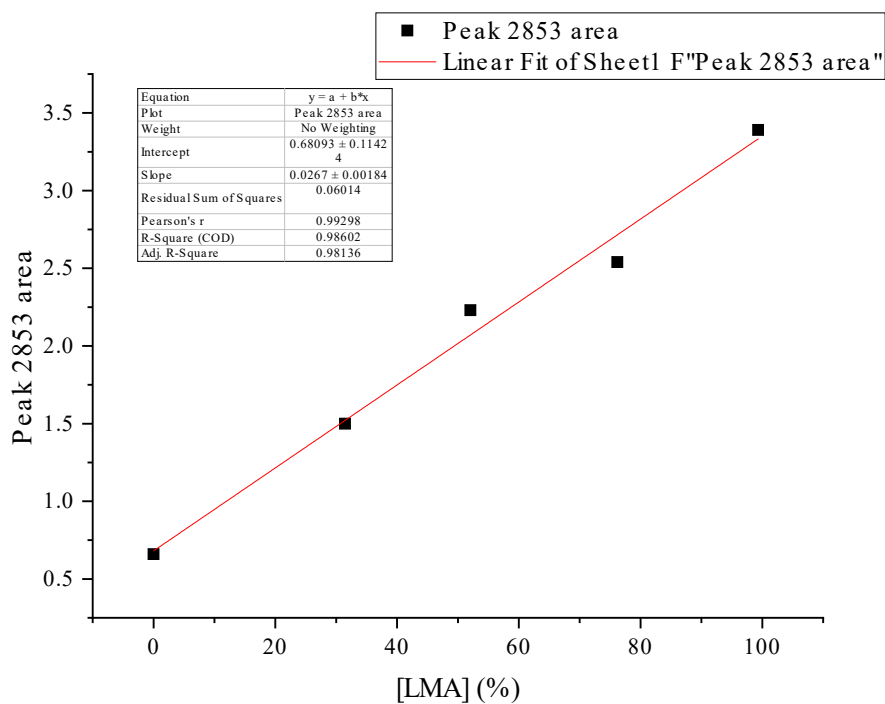
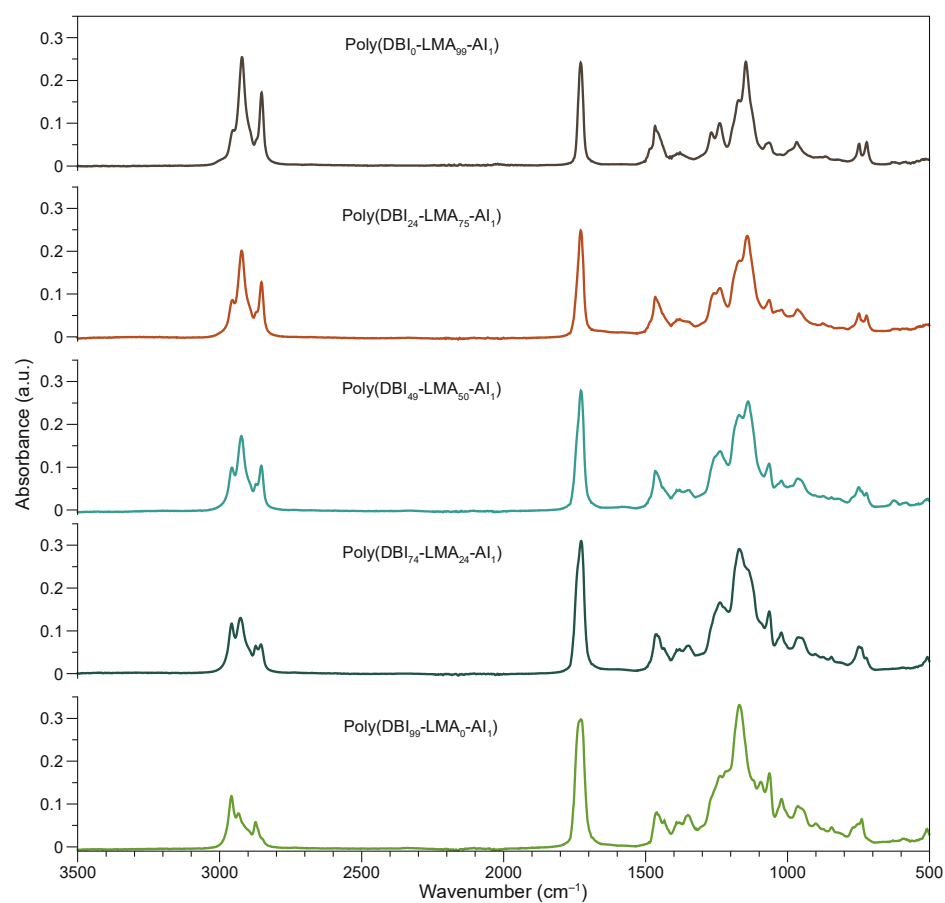
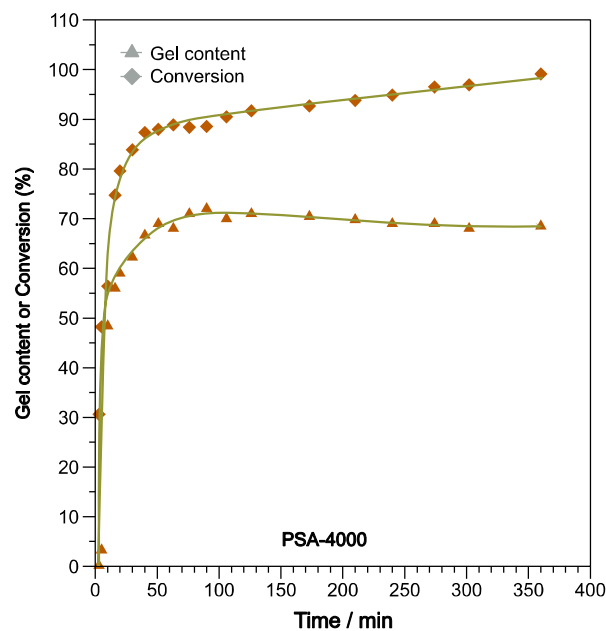


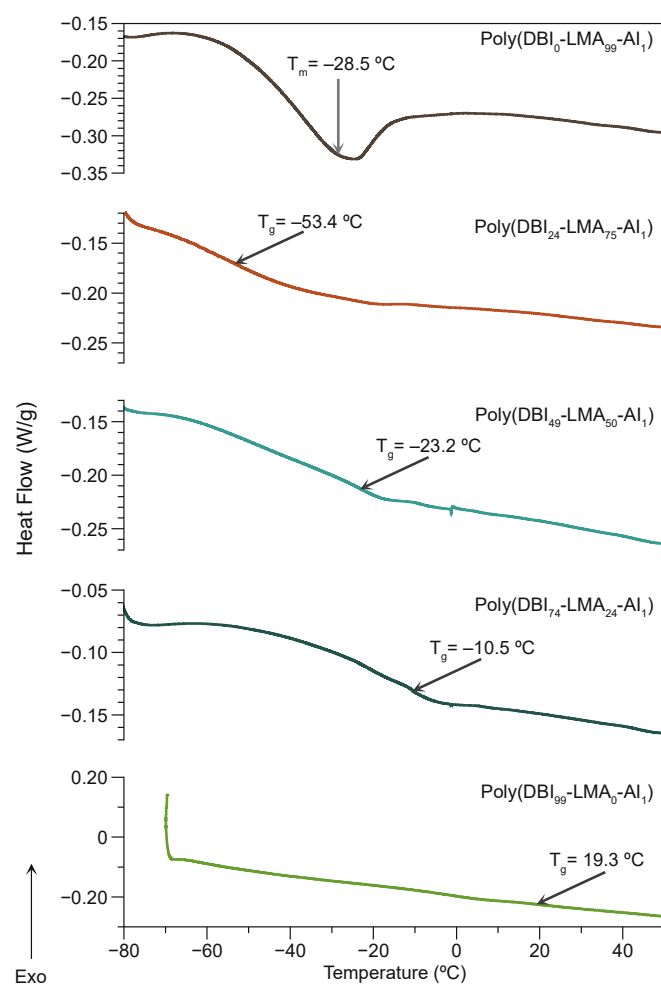
Figure S15. Area correlation of the band 2853  $\text{cm}^{-1}$  as a function of the LMA content.



**Figure S16.** FTIR spectra of copolymers films. a) PSA-5000; b) PSA-4000; c) PSA-3000; d) PSA-2000; e) PSA-1000.



**Figure S17.** Kinetic conversion study in relation to the gel content of the PSA-4000 sample.



**Figure S18.** DSC thermograms of copolymers films. a) PSA-5000; b) PSA-4000; c) PSA-3000; d) PSA-2000; e) PSA-1000.