

# Iron (II) metallo-supramolecular polymers based on thieno[3,2-*b*]thiophene for electrochromic applications

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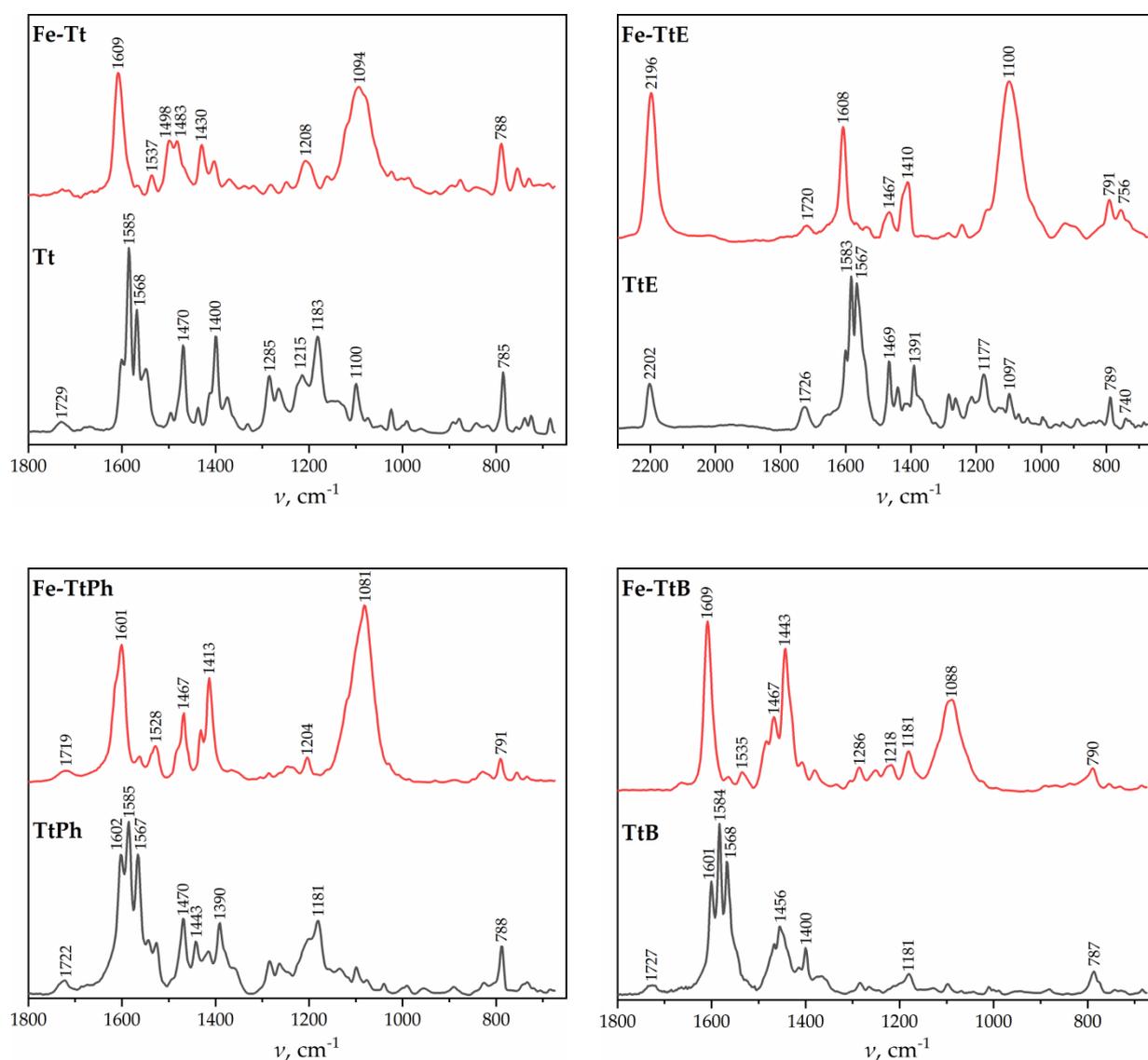


Figure S1. IR spectra of unimers and corresponding Fe-MSPs.

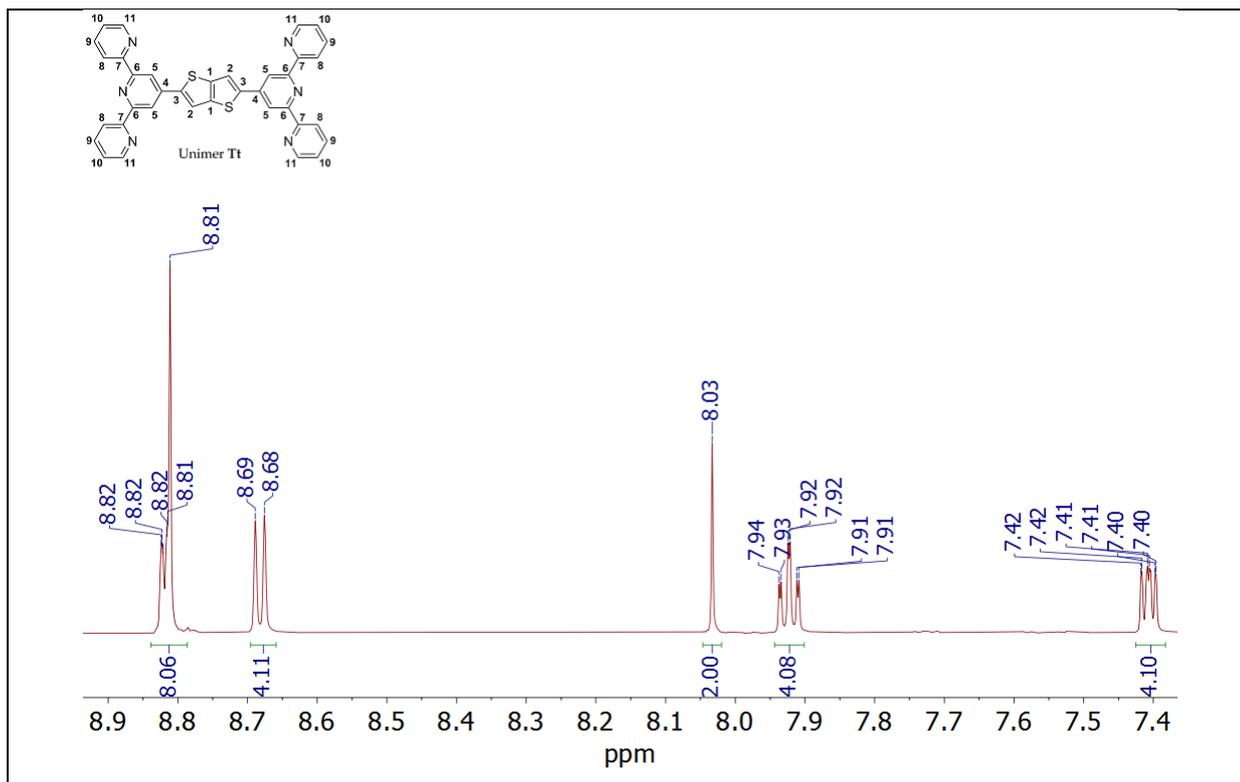


Figure S2.  $^1\text{H}$  NMR spectra of unimer Tt.

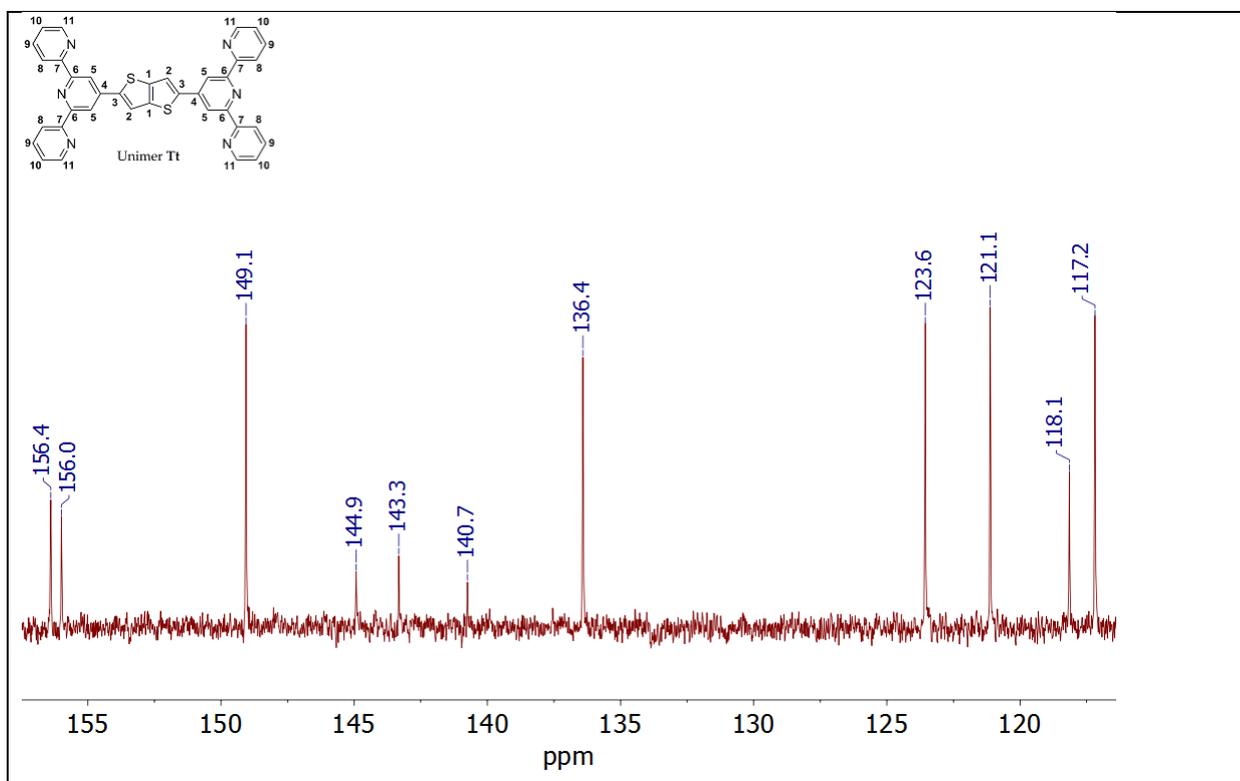


Figure S3.  $^{13}\text{C}$  NMR spectra of unimer Tt.

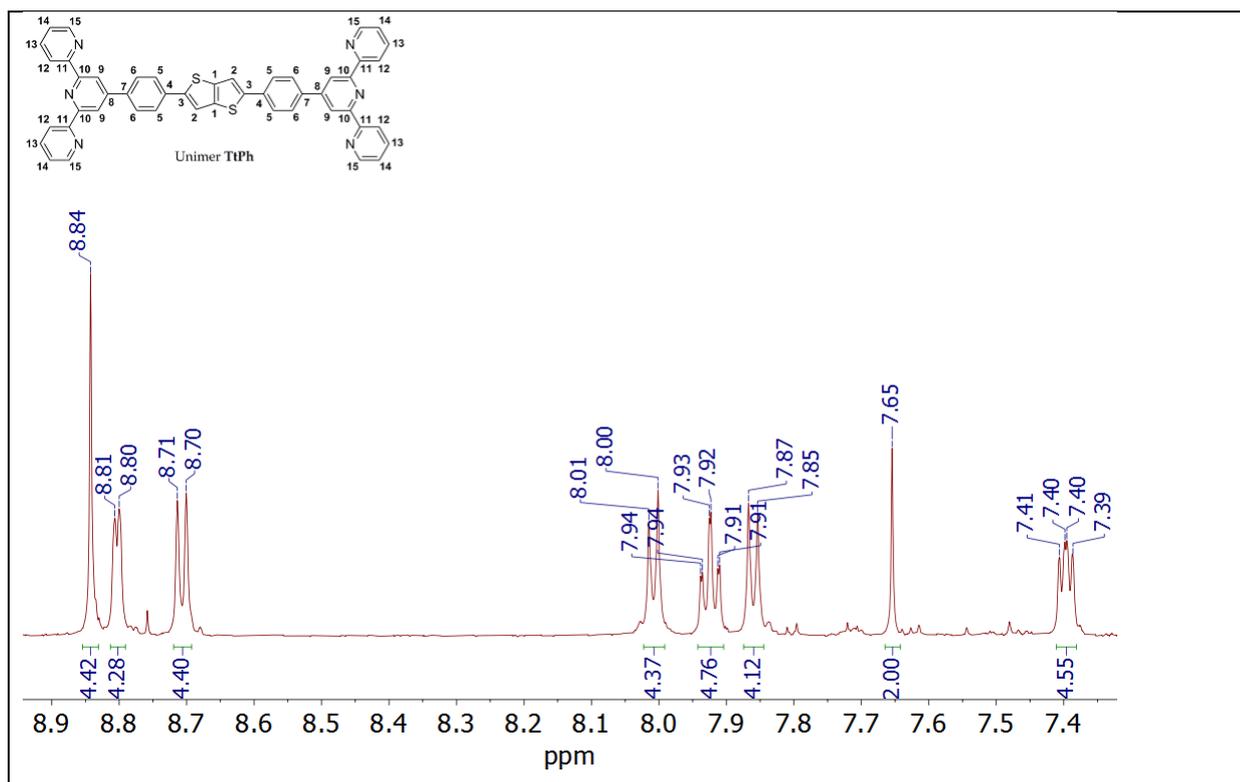


Figure S4. <sup>1</sup>H NMR spectra of unimer TtPh.

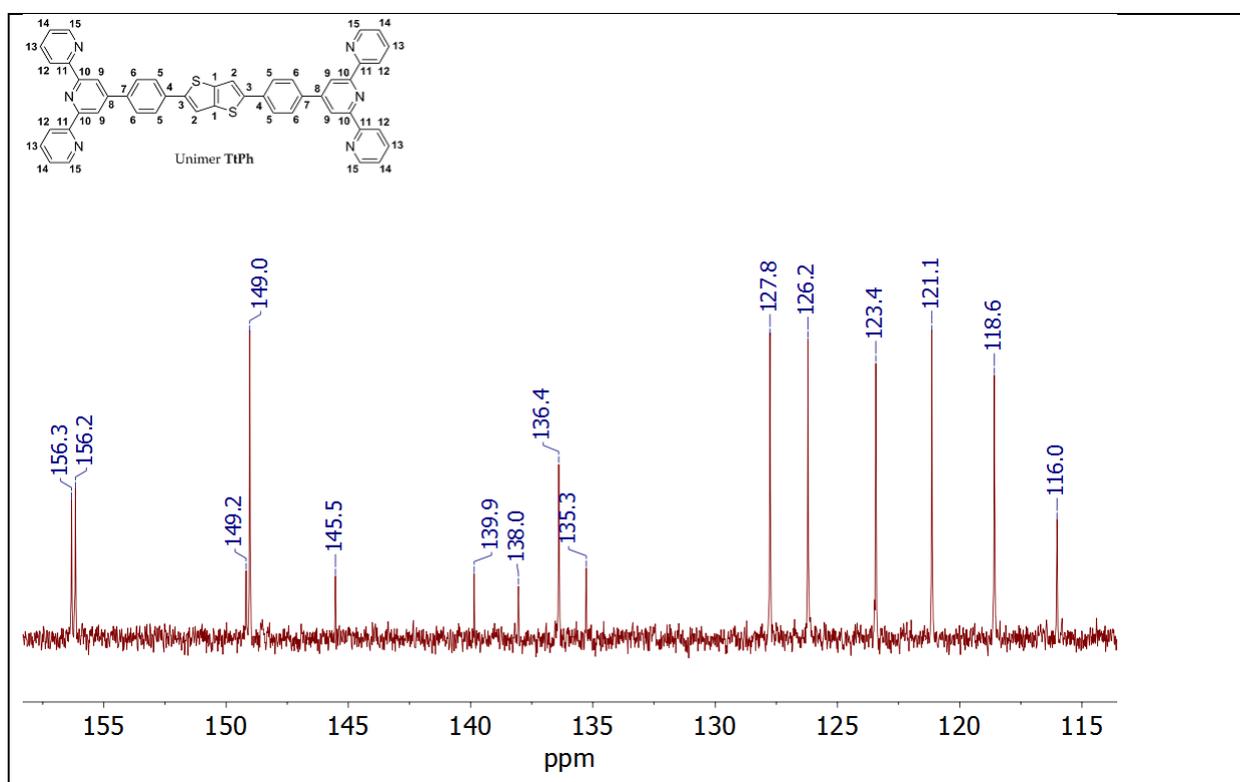


Figure S5. <sup>13</sup>C NMR spectra of unimer TtPh.

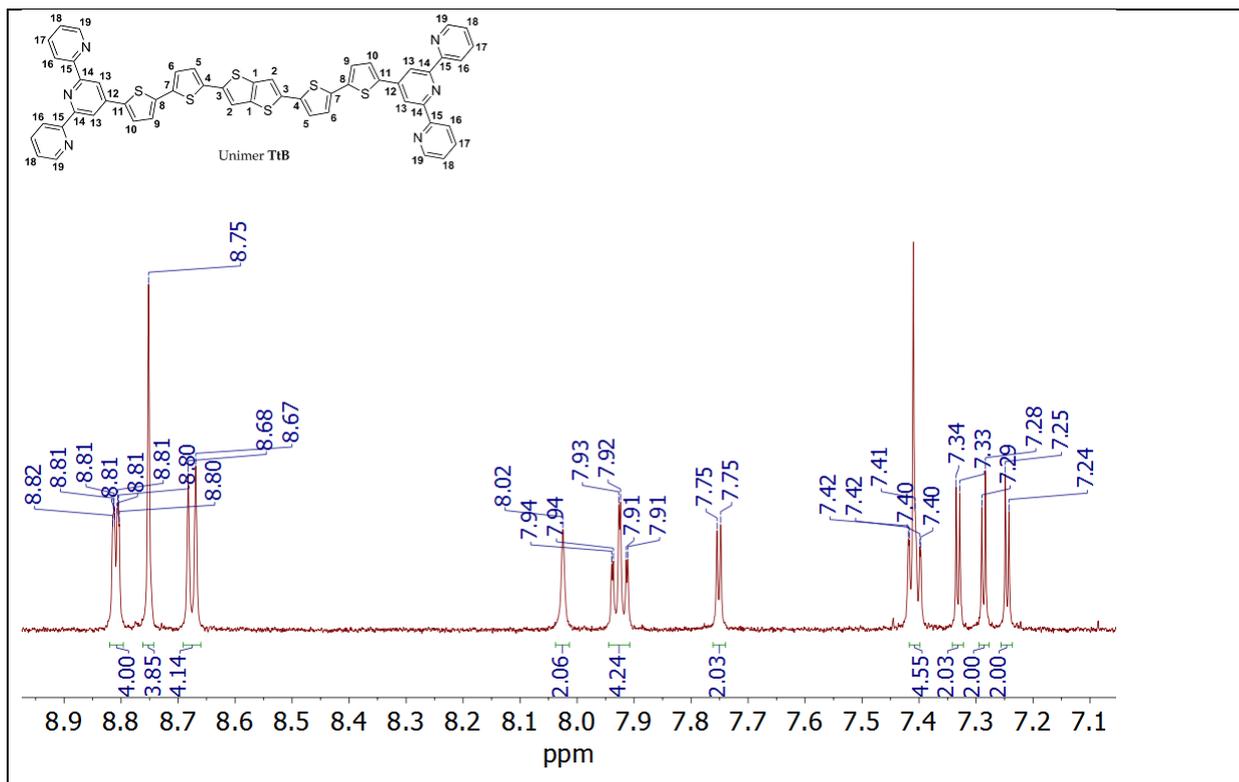


Figure S6. <sup>1</sup>H NMR spectra of unimer TtB.

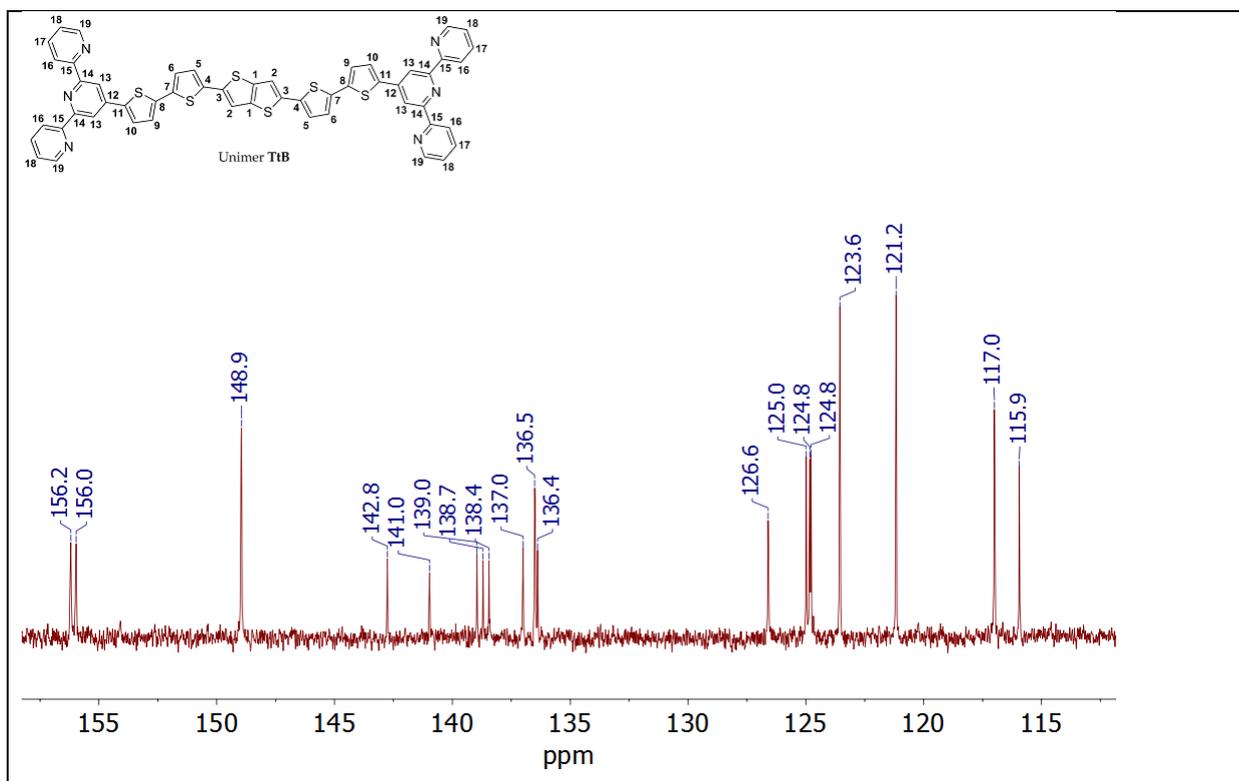


Figure S7. <sup>13</sup>C NMR spectra of unimer TtB.

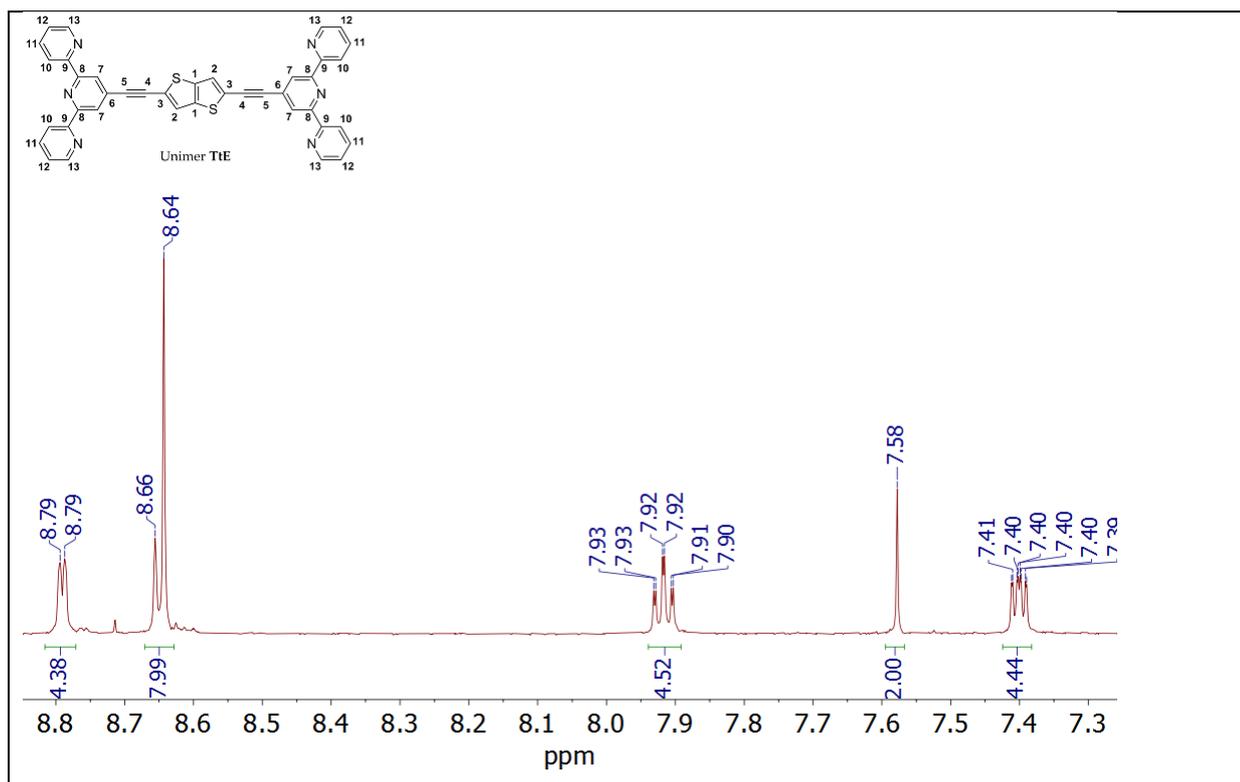


Figure S8.  $^1\text{H}$  NMR spectra of unimer TtE.

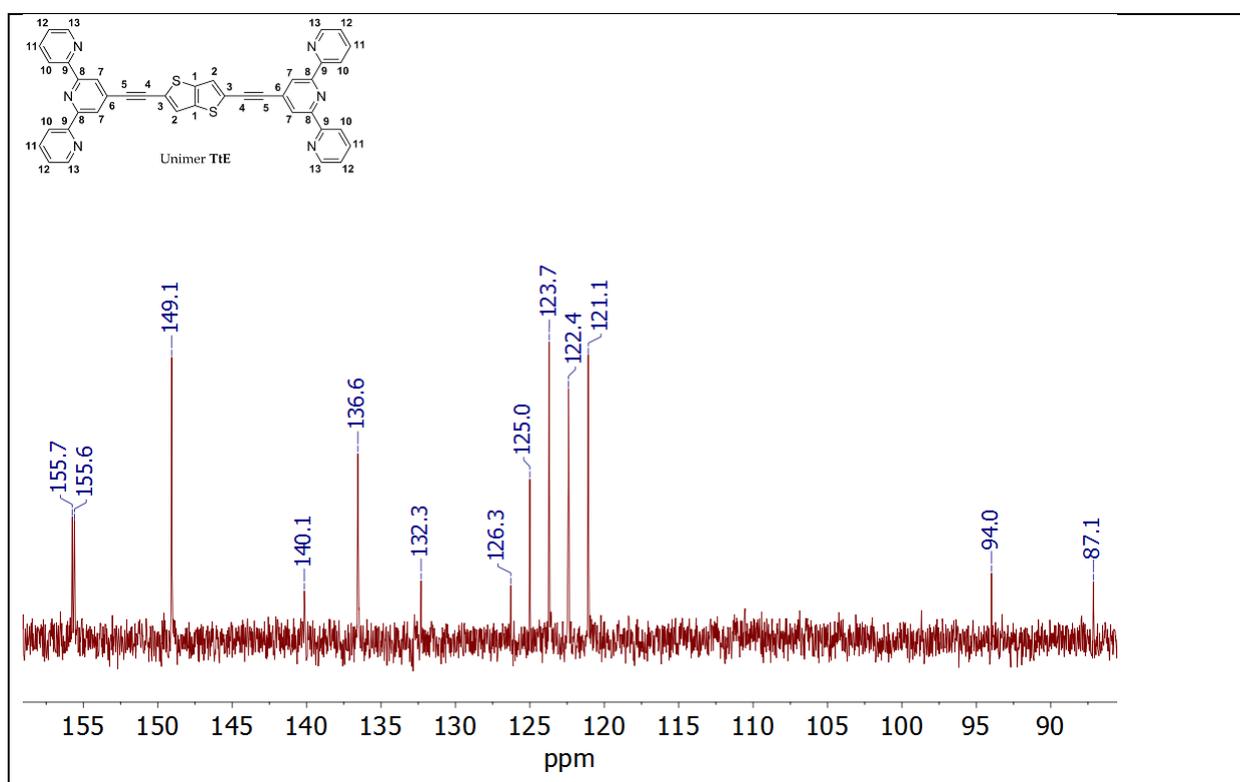
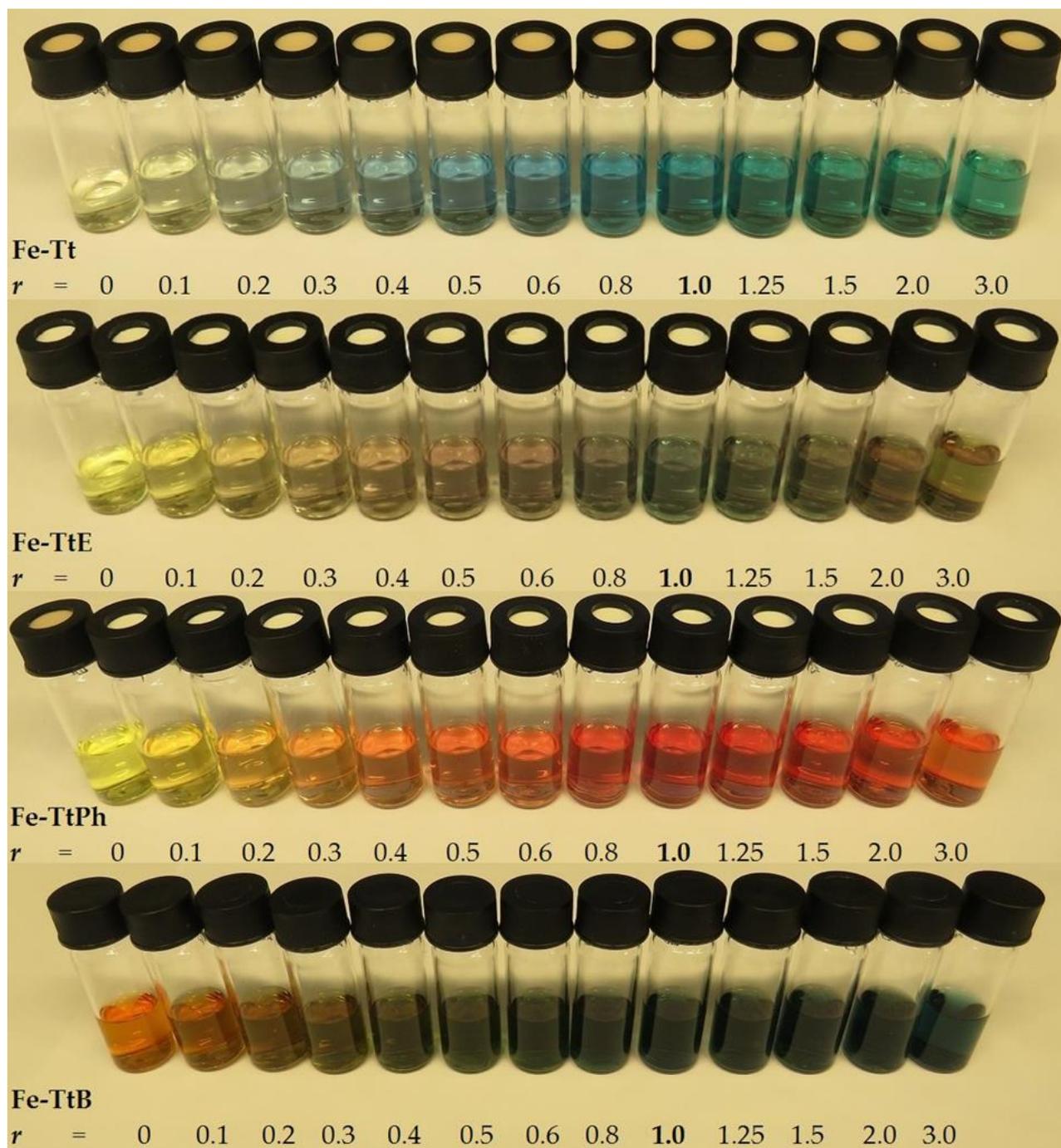
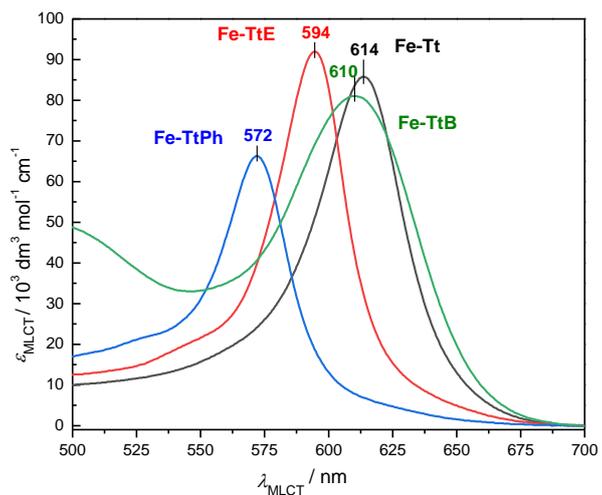


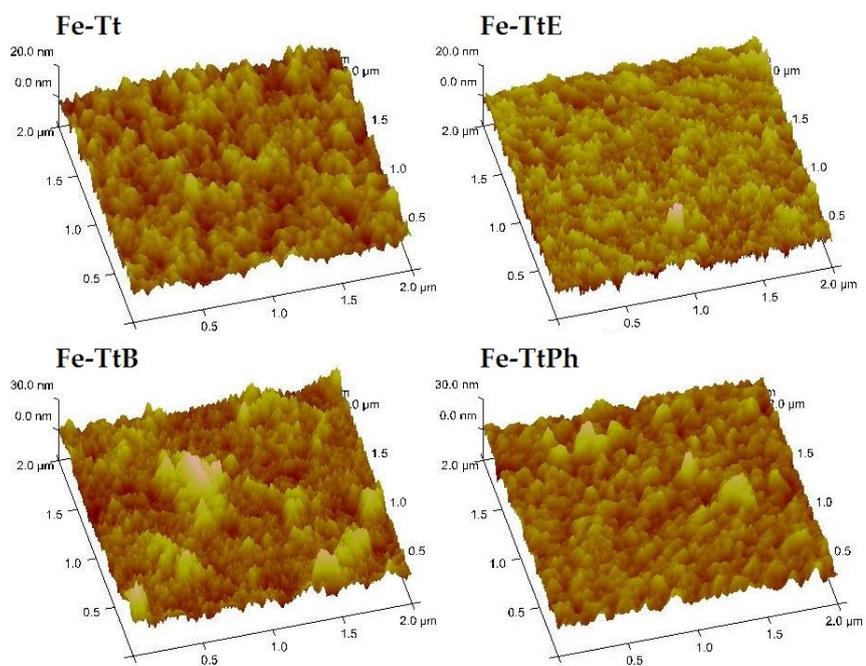
Figure S9.  $^{13}\text{C}$  NMR spectra of unimer TtE.



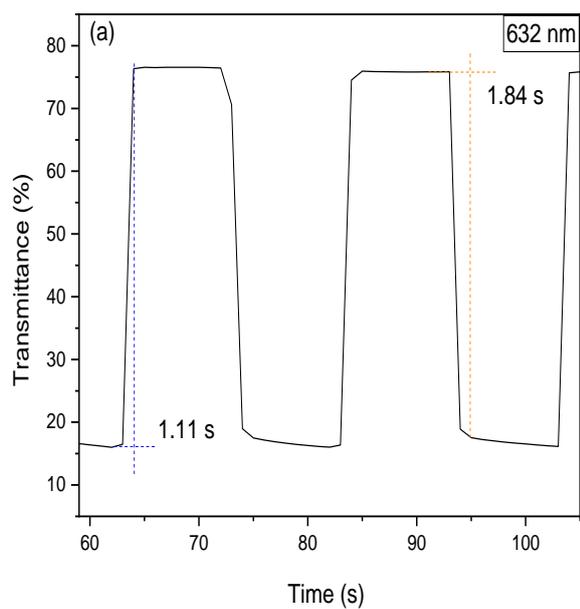
**Figure S10.** Solutions with gradually increasing ratio  $r = [\text{Fe}^{2+}]/[\text{U}]$  clearly show the effect of linker on the color of metallo-supramolecular oligomers and polymers.



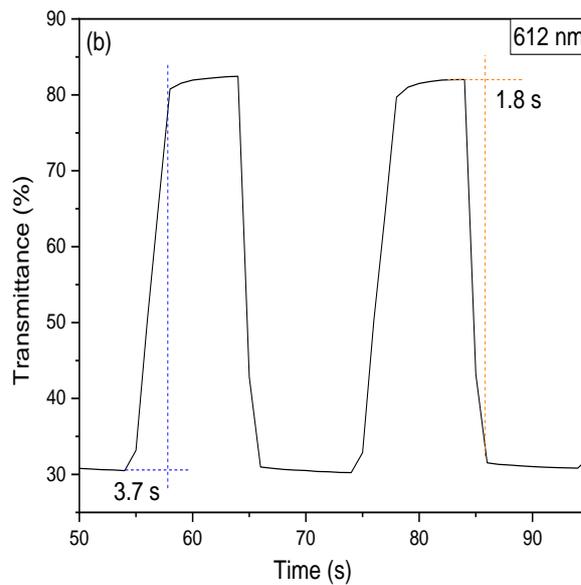
**Figure S11.** Comparison of the position and intensity of MLCT bands of Fe-MSPs ( $r = 1$ ) solutions.



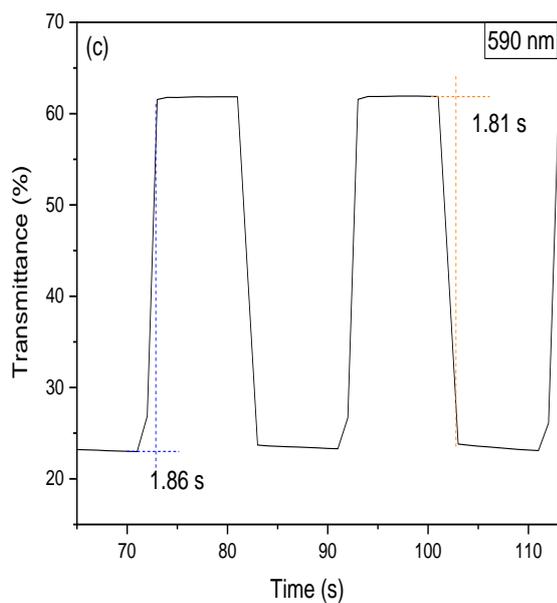
**Figure S12.** AFM images of Fe<sup>2+</sup>-MSP films (tapping mode, clockwise).



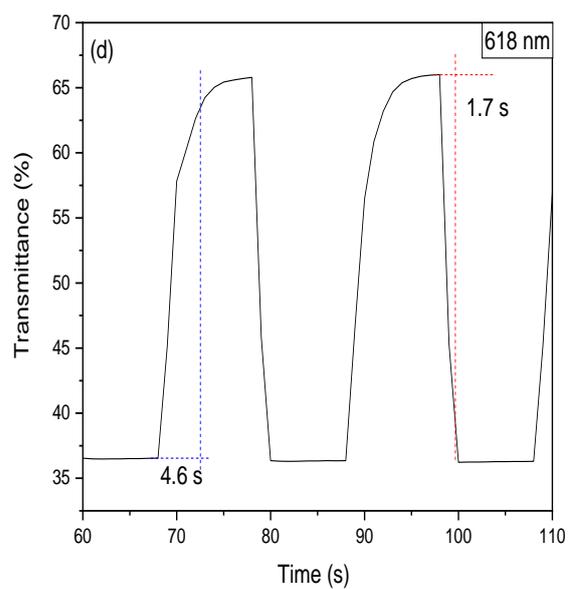
**Fe-Tt**



**Fe-TtE**

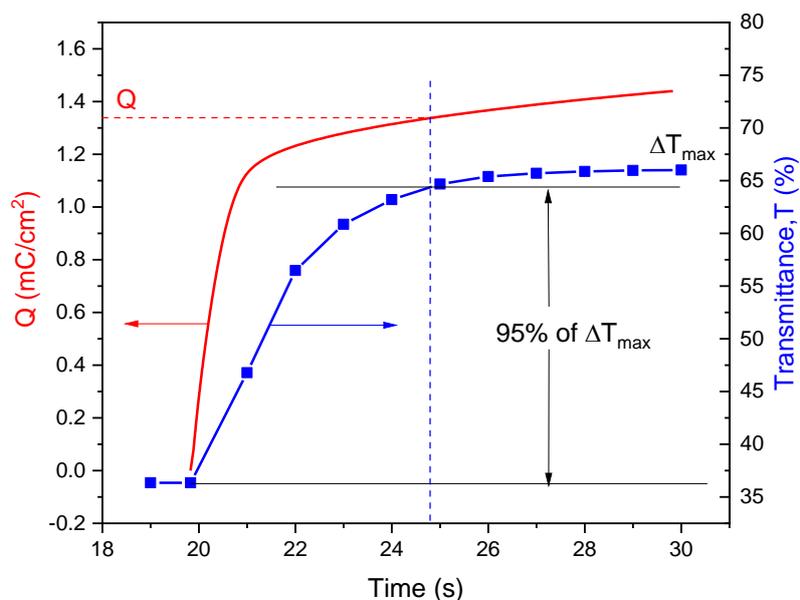


**Fe-TtPh**

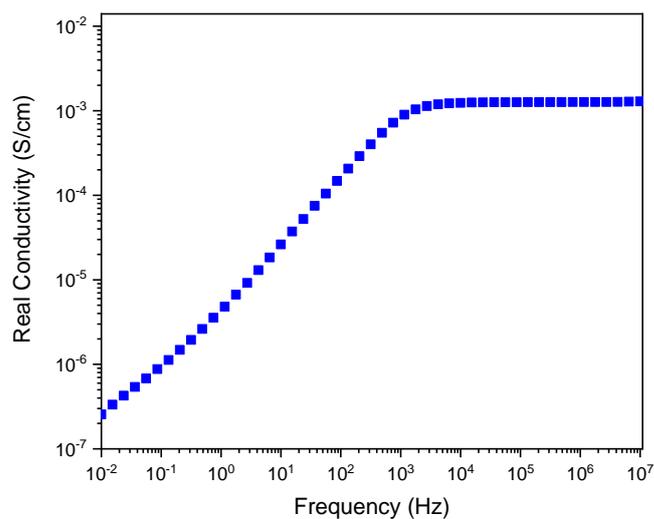


**Fe-TtB**

**Figure S13.** Determination of the bleaching and coloring time of electrochromic films from 95 % of saturated transmission. (a) **Fe-Tt**; (b) **Fe-TtE**; (c) **Fe-TtPh**; and (d) **Fe-TtB**.



**Figure S14.** Time courses of the transmittance,  $T$ , at  $\lambda = 618$  nm of the **Fe-TtB** electrochromic layer and the charge passed through the layer,  $Q$ , during its bleaching. The charge needed to achieve 95% of the final saturated transmittance was used for the calculation of the coloration efficiency



**Figure S15.** Frequency dependence of the real part of the conductivity of gel electrolyte.

**Randles-Ševčík equation** [35,36] for temperature of 25 °C is:

$$i_p = 269 \cdot c \cdot D^{1/2} \cdot v^{1/2}$$

where:  $i_p$  is the current density of CV peak,  $c$  the molar concentration (in mol/L) and  $D$  diffusion coefficient (in  $\text{cm}^2\text{s}^{-1}$ ) of electroactive species and  $v$  the scan rate (in  $\text{V}\cdot\text{s}^{-1}$ ).