



Article Effects of LiF-Addition on Sintering Behavior and Dielectric Response of LaPO₄ Ceramics at Microwave and Terahertz Frequency for LTCC Applications

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Abstract: This paper reports on the successful preparation of LaPO₄-*x* wt.% LiF (x = 0-5) ceramics using the traditional solid-state reaction method. The crystal structures, sintering behaviors, and dielectric response at microwave and terahertz frequencies were investigated. XRD results indicate that all the diffraction peaks were attributed to LaPO₄, and no secondary phase was observed. Rietveld refinement was conducted to analyze the variation of the crystal structure of LaPO₄-*x* wt.% LiF. SEM indicates that the addition of LiF significantly decreased the grain size while increasing the apparent density of the ceramics. When x = 3, the optimum microwave dielectric properties $\varepsilon_r = 10.03$, $Q \times f = 81,467$ GHz, and $\tau_f = -43.79$ ppm/°C were achieved in LaPO₄-3 wt.% LiF ceramic at 750 °C. The infrared reflectance spectrum and terahertz time-domain spectroscopy were analyzed and compared with the dielectric properties measured at microwave frequency to investigate the inherent dielectric response. The findings indicate that the dielectric constant attributed to ionic displacement polarization and oxygen vacancy is an essential factor affecting dielectric loss. Moreover, it is worth noting that the LaPO₄-3 wt.% LiF ceramic demonstrates excellent compatibility with silver powders, suggesting its immense potential as a dielectric material in LTCC applications.

Keywords: microwave dielectric properties; rare earth phosphates; Rietveld refinement; infrared reflectance spectrum; LTCC

1. Introduction

With the development of wireless communication and the arrival of the 5G era, the demand for high–speed, high-capacity wireless technology in 5G communications is soaring [1–3]. In the future communication field, the operating frequency will be expanded to micron and millimeter waves. Microwave dielectric ceramics, which are essential components in passive devices, have garnered more attention due to their exceptional performance. The main parameters to measure the properties of microwave dielectric ceramics are the dielectric constant (ε_r), quality factor ($Q \times f$), and temperature coefficient (τ_f). Among these, ε_r is a physical quantity that characterizes polarizability. High-frequency communication is characterized by low signal delay and therefore requires a low dielectric constant in the material. $Q \times f$ described the dielectric loss of materials [4]. A high-quality factor means high signal quality. τ_f determines the stability of signals and requires a value near zero to ensure operating stability at work. In addition, to realize the device application of materials, low-temperature co-firing technology (LTCC) is necessary [5–9].

Rare-earth phosphates exist in crystals of two structures (monazite and xenotime). Therein, high atomic masses (La, Ce, Pr, Nd, Sm, and Eu) tend to form the monazite phase,



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). and other elements (Tb, Dy, Ho, Er, Tm, Yb, and Y) form the xenotime phase. However, the researcher mainly focused on other properties, such as luminescence and catalysis [10–12]. Narasimha et al. reported the microwave dielectric properties of LaPO₄ [13]. Cho et al. published the microwave dielectric properties of a series of rare-earth phosphates [14]. LaPO₄ ceramic exhibits excellent microwave dielectric properties of $\varepsilon_r = 10.4$, $Q \times f = 64,556$ GHz, and $\tau_f = -56.2$ ppm/°C with a sintering temperature of 1400 °C and would be a good candidate for future 5G communications. However, the high sintering temperature impeded the application at the passive device terminal. Therefore, reducing the sintering temperature is meaningful work.

It is well known that the search for new ceramic systems or the addition of sintering aids can be effective in reducing sintering temperatures. Li-based ceramics usually have a low sintering temperature. Kamutzki et al. [15] successfully prepared LiCrSi_2O_6 microwave dielectric ceramics at low temperatures by applying the plasma spark sintering method, obtaining relatively excellent dielectric properties. However, the latter is easier to achieve than the option of adding a sintering aid.

As the universal sintering assistant, LiF has been used in this paper. By adding LiF to the BaFe_{0.5}Nb_{0.5}O₃ ceramic system, Intatha et al. [16] successfully reduced the sintering temperature and improved the sintering characteristics of the ceramics. Song et al. [17] synthesized SrWO₄-2 wt.% LiF ceramic at 850 °C by the traditional solid phase reaction method with excellent microwave dielectric properties of $\varepsilon_r = 9.03$ and $Q \times f = 47,830$ GHz. Hao et al. [18] prepared Li₂TiO₃-2.5 wt.% LiF ceramic with excellent microwave dielectric properties of $\varepsilon_r = 24.01$, $Q \times f = 75,500$ GHz, and $\tau_f = -36.2$ ppm/°C at 950 °C and exhibited good compatibility with Ag. However, adding LiF to LaPO₄ ceramic to reduce the sintering temperature has not been reported.

This article presents the preparation and investigation of LaPO₄-*x* wt.% LiF (x = 0-5) in terms of crystal structure, sintering behaviors, and dielectric properties at microwave and terahertz frequencies. The phase composition, crystal parameters, and sintering behaviors were investigated using XRD, SEM, Rietveld refinement, and apparent density. The infrared reflectance spectrum and terahertz time-domain spectroscopy investigated the intrinsic dielectric mechanism. In addition, the compatibility of samples with Ag has been analyzed.

2. Experimental Procedure

The LaPO₄-*x* wt.% LiF (x = 0, 1, 2, 3, 4, and 5) system was prepared by the traditional solid-state reaction method. High-purity powders of La₂O₃ (99.99%), NH₄H₂PO₄ (99%), and LiF (99%) as the primary raw materials were weighted stoichiometrically. The raw powders were mixed with zirconium balls and alcohol for 24 h. Then the slurry was dried in an oven at 80 °C and passed through a 60-mesh screen. The dried powders were calcined at 1200 °C to form the main crystal phase. Whereafter, LiF was added as a sintering aid and for secondary ball milling. Before being pressed into a pellet with a diameter of 10 mm and a height of 6 mm. To bind the pellets, 12 wt.% paraffin was added. Finally, the pellets were sintered with a heating rate of 5 °C/min.

The phase composition of the sample was identified using an X-ray diffractometer (Model D/MAX-B, Rigaku Co., Tokyo, Japan) at room temperature with Cu K α radiation. The signals were collected in the range of 10°–80°. Archimedes' drainage method was carried out to measure the apparent density of the samples. Scanning electron microscopy (S-4800, Hitachi, Tokyo, Japan) was implemented to observe the microstructure. The microwave dielectric properties (ε_r , $Q \times f$, and τ_f) of the samples were measured by a network analyzer (3656D, Ceyear Co., Qingdao, China) with the Hakki-Coleman method [19]. The ε_r and τ_f were determined by the parallel plate method, and the $Q \times f$ was determined by the resonant cavity method. The temperature coefficient can be calculated using Equation (1).

$$\tau_f = \frac{f_2 - f_1}{(T_2 - T_1) \times f_1} \tag{1}$$

where the resonant frequencies f_1 and f_2 are measured at T_1 (25 °C) and T_2 (85 °C), respectively. The infrared reflectance spectrum was measured using a far infrared spectrometer (FTIR, Bruker IFS66v, Bruker Optics, Ettlingen, Germany) at the National Synchrotron Radiation Laboratory. The terahertz time-domain spectroscopy was measured by a THz-TD spectrometer (Z3, Zomega, Plano, TX, USA) in the State Key Laboratory of New Ceramics and Fine Technology, Tsinghua University [20–22]. The dielectric properties and loss in the terahertz band are calculated as follows: Equations (2)–(7) [23–25].

$$\varepsilon'(\omega) - i\varepsilon''(\omega) = [n(\omega) - ik(\omega)]^2$$
⁽²⁾

$$\mathbf{n}^*(\omega) = \mathbf{n}(\omega) - \mathbf{i}\mathbf{k}(\omega) \tag{3}$$

$$\alpha(\omega) = \frac{2\omega \mathbf{k}(\omega)}{v} \tag{4}$$

$$k(\omega) = \frac{\alpha(\omega) \cdot \lambda}{4\pi} \tag{5}$$

$$\varepsilon = n^2(\omega) - \left[\frac{\alpha(\omega) \cdot \lambda}{4\pi}\right]^2 \tag{6}$$

$$\tan \delta = \frac{\varepsilon''(\omega)}{\varepsilon'(\omega)} = \frac{n(\omega) - \alpha(\omega) \cdot \lambda}{2\pi} \left\{ n^2(\omega) - \left[\frac{\alpha(\omega) \cdot \lambda}{4\pi}\right]^2 \right\}$$
(7)

where $n^*(\omega)$ is the complex refractive index at terahertz frequency, $n(\omega)$ is the index of refraction, $k(\omega)$ is the coefficient of extinction, v is the propagation speed of an electromagnetic wave at terahertz, ω is the frequency of Angle, and $\alpha(\omega)$ is the coefficient of absorption.

3. Results and Discussion

Figure 1 shows the XRD diffraction pattern of LaPO₄-*x* wt.% LiF ceramic. The figure's diffraction peaks are consistent with the standard PDF card (PDF#83–0651), indicating that the crystal structure is a single monoclinic structure with a space group of $P2_1/c$. The absence of the diffraction peak of the second phase demonstrates that the addition of LiF does not affect the formation of the LaPO₄ main stage.

Figure 1b shows that with increasing doping of LiF, the diffraction peaks corresponding to the (200) crystal orientations are shifted towards. According to the Bragg equation, the lattice parameters and cell volume change as the angle shifts. The lattice parameters and cell volume also change as the angle shifts; with increasing *x*, the cell volume increases from 306.115 Å³ to 306.339 Å³ as *x* increases.

Figure 2 exhibits the refined XRD data using Full Prof software (FullProf_Suite Windows (64 bits)). Table 1 summarizes lattice parameters and reliability factors. The smooth red line represents the difference between the measured and theoretical values, while the lower R-factors indicate that the refined results are reliable [26–28]. Figure 3 displays a single-unit cell diagram of LaPO₄ ceramic. The crystal cell comprises [LaO₉] polyhedrons and [PO₄] tetrahedrons. [LaO₉] and [PO₄] connect with oxygen along the *b*-axis. To intuitively understand the microscopic morphology of LaPO₄-*x* wt.% LiF ceramics.



Figure 1. XRD pattern of the LaPO₄-*x* wt.% LiF ceramics at the optimal sintering temperatures. (a) overall X-ray diffraction pattern; (b) local XRD trends.



Figure 2. The refinement patterns of the LaPO₄-*x* wt.% LiF ceramics at the optimal sintering temperatures ((**a**–**f**) corresponding to 1400 °C, 900 °C, 900 °C, 750 °C, 750 °C, 750 °C).



Figure 3. Unit cell diagram of the LaPO₄-3 wt.% LiF ceramic.

	x = 0	<i>x</i> = 1	<i>x</i> = 2	<i>x</i> = 3	x = 4	<i>x</i> = 5
a (Å)	6.8358	6.8359	6.8368	6.8366	6.8367	6.8371
b (Å)	7.0728	7.0743	7.0749	7.0745	7.0746	7.0748
c (Å)	6.5058	6.5043	6.5070	6.5069	6.5068	6.5069
α (°)	90.000	90.000	90.000	90.000	90.000	90.000
β (°)	103.295	103.276	103.268	103.270	103.272	103.273
γ (°)	90.000	90.000	90.000	90.000	90.000	90.000
Ζ	4	4	4	4	4	4
V _m (Å ³)	306.115	306.233	306.337	306.307	306.310	306.339
R_{p}	10.60	10.70	11.50	9.42	9.12	9.08
R_{wp}	12.30	12.30	12.30	10.90	10.40	10.50
Rewp	5.52	5.69	6.26	5.96	6.18	5.92
χ^2	2.23	2.16	1.96	1.82	1.68	1.77

Table 1. The refinement patterns of the LaPO₄-*x* wt.% LiF ceramics at the optimal sintering temperatures.

Figure 4 displays the SEM image, which illustrates the changes in the apparent morphology of LaPO₄ ceramics with the increase in the amount of LiF. As shown in Figure 4a–e, when x = 1,2, the grain size is relatively large and the grain boundary is distinct. However, when $x \ge 3$, the grain size begins to refine, indicating that adding LiF is conducive to grain refinement. It is worth noting that due to the melting point of LiF at 845 °C, a part of LiF melts to form a liquid phase during the ceramic sintering process. When $x \ge 4$, grain boundaries begin to soften, and macroscopic defects such as pores between grains and cracks appear on the surface, which particularly impacts the apparent density and performance [29–31]. When the amount of LiF exists on the grain surface in an amorphous state, and a small number of pores and cracks appear, which has an adverse effect on the densification of ceramics. This is consistent with the trend of relative density change and further indicates that an appropriate amount of LiF can effectively reduce the sintering temperature and porosity. Improve the relative density of ceramics.

Figure 5 shows the apparent density of LaPO₄-*x* wt.% LiF (x = 0-5) ceramics. Between 1250 °C and 1450 °C, the apparent density of the LaPO₄ ceramic matrix increases from 4.13 g/cm⁻¹ to 4.40 g/cm⁻¹ with the increase in temperature, indicating that temperature is an essential factor. Furthermore, the sintering interval of the LaPO₄-*x* wt.% LiF ceramics is 650 °C–1000 °C, meaning that the addition of LiF successfully reduces the sintering temperature of ceramics. The sintering temperature decreases further with the increase in LiF content. When x = 3, the temperature is reduced to 750 °C. At different *x* values, the apparent density increases first and then decrease. Figure 5b shows the relative density of ceramic samples. The shrinkage increases initially with the increase of *x* and reaches its maximum value at x = 3.

Figure 6 is the dielectric constant (ε_r) of LaPO₄-*x* wt.% LiF (*x* = 0–5) ceramics. As the sintering temperature increases, the ε_r first increases and then decreases. It takes *x* = 3 as an example. When the temperature increases from 700 °C to 750 °C, the ε_r increases from 7.08 to 10.03. However, as the temperature rises, the ε_r begins to decrease slowly. The ε_r is affected by several factors, such as dielectric polarization, porosity, and second equality [24,32,33]. However, the X-ray shows that the LaPO₄-*x* wt.% LiF (*x* = 0–5) ceramic is a single pure phase. The change in apparent density is the same as the dielectric constant, so the density is the main factor affecting the dielectric constant of LaPO₄ ceramics. Adding LiF reduces the sintering temperature and promotes the densification of ceramics. Therefore, as the temperature increases, giant permittivity is obtained. Still, when the temperature exceeds the optimal sintering temperature, the increase in grain size destroys the crystal structure, leading to a decrease in density and a consequent reduction in permittivity.



Figure 4. The morphology of the LaPO₄-*x* wt.% LiF ceramics as a function of the *x* values ((**a**–**e**) corresponding to 900 °C, 900 °C, 750 °C, 750 °C, 750 °C).



Figure 5. Density of the LaPO₄-x wt.% LiF ceramics as a function of the x values. (a) Apparent density; (b) relative density.



Figure 6. Dielectric constant (ε_r) of the LaPO₄-*x* wt.% LiF ceramics sintered at various temperatures.

Figure 7 is the quality factor ($Q \times f$) of LaPO₄-x wt.% LiF (x = 0-5) ceramics. The quality factor has the same variation as the dielectric constant. For instance, when x = 3, as an example. With the temperature increasing, the $Q \times f$ increases significantly from 15,946 GHz at 700 °C to 81,467 GHz at 750 °C, and the $Q \times f$ begins to decrease as the temperature continues to rise. Many factors affect the $Q \times f$. These factors fall into two categories: inherent losses and external losses. The lattice vibration mode and crystal structure influence the inherent losses [34,35]. The sintering additive effectively reduces the sintering temperature of LaPO₄ ceramics, but the phase composition is not affected. Thus, the main factor affecting the quality factor is density. Before reaching the optimum temperature point (750 °C), the increase in temperature is beneficial to reduce the number of pores in the ceramic, improve the density, and reduce the material loss inside the ceramic. As the temperature continues to rise, especially to 800 °C, several macroscopic cracks and pores make the interior losse, leading to the $Q \times f$ dropping to 68,791 GHz. The quality factor finally stabilizes within the range of 68,000 GHz to 77,000 GHz.



Figure 7. Quality factor ($Q \times f$) of the LaPO₄-*x* wt.% LiF ceramics sintered at various temperatures.

The temperature coefficient characterizes the thermal stability of materials. Figure 8 summarizes the trend of the temperature coefficient. The temperature coefficients are all negative, and the change of *x* and τ_f does not change significantly. It fluctuates between $-46.40 \text{ ppm}/^{\circ}\text{C}$ and $-34.71 \text{ ppm}/^{\circ}\text{C}$, indicating that adding LiF has little effect on the temperature coefficient. Table 2 shows the microwave dielectric properties of LaPO₄-*x* wt.% LiF ceramics. In conclusion, when *x* = 3, the optimum microwave dielectric



Figure 8. (τ_f) of the LaPO₄-*x* wt.% LiF ceramics at various temperatures.

Composition	S.T (°C)	E _r	Q imes f (GHz)	$ au_f$ (ppm/°C)
x = 0	1400	8.73	44,449	-42.18
x = 1	900	9.81	53,247	-43.14
x = 2	900	9.19	62,844	-46.40
x = 3	750	10.04	81,467	-43.79
x = 4	750	9.61	43,674	-40.41
x = 5	750	8.56	33,873	-34.72

Table 2. List of the microwave dielectric properties of the LaPO₄-*x* wt.% LiF ceramics.

Infrared spectroscopy is an effective technique for characterizing the dielectric loss and response of ceramics. Figure 9a displays the fit of the infrared reflectance spectrum, obtained using the Reffit software and the classical resonator model with three parameters. The fitted values are in good agreement with the measured values. The complex permittivity (ε *) and reflectance (R) can be calculated using Equations (8) and (9) in the harmonic oscillator mode [36,37]:

$$\varepsilon^*(\omega) = \varepsilon' + \varepsilon'' = \varepsilon_{\infty} + \sum_{i=1}^n \frac{\omega_{p_j}^2}{\omega_{o_j}^2 - \omega^2 - i\gamma_j\omega}$$
(8)

$$R(\omega) = \left| \frac{1 - \sqrt{\varepsilon^*(\omega)}}{1 + \sqrt{\varepsilon^*(\omega)}} \right|^2 \tag{9}$$



Figure 9. (a) Measured and fitted infrared reflectance spectrum curves; (b,c) complex permittivity of infrared spectra of the LaPO₄-3 wt.% LiF ceramic.

In the formula, ε_{∞} is the optical permittivity, ω_{pj} and ω_{oj} are the plasma frequency and transverse frequency, respectively, γ_j is the damping factor, and *i* is the imaginary unit. Table 3 shows the vibration modes of 16 different phonons. The optical permittivity obtained by infrared spectrum fitting is 2.341. Figure 9 represents the real (ε') and imaginary (ε'') parts. The scattering of phonons and the overlap of peaks make the real part ($\varepsilon' = 6.83$) less than the measured dielectric constant ($\varepsilon_r = 10.04$) [37–39]. Figure 9c shows the observed dielectric loss (2.16×10^{-4}) in the same order of magnitude as the measured dielectric loss (1.72×10^{-4}), indicating that phonon vibration affects dielectric loss. Moreover, external factors such as porosity, density, and grain distribution also affect dielectric loss. Pores can be reduced or processes optimized to minimize losses.

Mode	LaPO ₄ -3	wt.% LiF	$\varepsilon_{\infty} = 2.341$		
moue	ω_{oj}	ω_{pj}	γ_j	$\Delta arepsilon_j$	
1	126.5378	47.4173	7.1522	0.1404	
2	157.7606	62.9653	8.1873	0.1593	
3	197.1342	152.8907	23.5457	0.6015	
4	214.3998	149.5696	16.9943	0.4867	
5	248.9289	191.8766	30.0668	0.5941	
6	288.8744	175.8852	29.1625	0.3707	
7	456.4929	449.3217	332.2120	0.9688	
8	537.1147	134.5086	22.1472	0.0627	
9	561.8092	52.2352	7.7303	0.0086	
10	577.8740	82.8326	11.3729	0.0205	
11	621.9667	151.3773	26.1880	0.0592	
12	951.8753	93.8768	6.2487	0.0097	
13	994.1020	147.9418	11.1387	0.0221	
14	1035.2952	432.4328	68.0378	0.1745	
15	1091.1393	290.0738	26.4696	0.0707	
16	1145.5237	128.5903	28.4676	0.0126	

Table 3. Phonon parameters obtained from the fitting of the LaPO₄-3 wt% LiF ceramic.

Infrared fitting results are easily limited by infrared pattern recognition. Terahertz timedomain spectroscopy technology investigates the effect of lattice vibration on properties. Figure 10 shows the dielectric response of LaPO₄-3 wt.% LiF sintered at 750 °C in the



frequency band of 0.5 THz~1.1 THz. The dielectric properties of the terahertz band are derived from the refractive index and extinction coefficient.

Figure 10. Optical and dielectric Properties of the LaPO₄-3 wt.% LiF ceramic in THz frequency band: (a) reflectance; (b) absorption coefficient; (c) dielectric constant; and (d) dielectric loss.

Figure 11 shows the change curve of the refractive index (n). The refractive index does not increase significantly with the increase in frequency and is stable between 3.17 and 3.21. The absorption coefficient (α) increases with the increase in frequency, which is associated with the unit volume polarizability. The dielectric constant which extrapolated from terahertz time-domain spectrum was in line with the value at microwave frequency [1,21,35]. At 7 GHz, the dielectric constant of ceramics is 10.03. At 0.5 THz, the dielectric constant of the ceramic is 10.04. Therefore, it can be inferred that the polarization mechanism of dielectric ceramic does not change in the terahertz frequency band, in which the ionic polarization is still dominant.



Figure 11. XRD pattern of the LaPO₄-3 wt.% LiF ceramic co-fired with Ag at 750 °C.

The fitted line in Figure 11 is the linear relationship between dielectric loss and frequency, and the slope of the matched line is 0.010, representing the vibration of the lattice. Considering the defect in the sample, it may be the oxygen vacancy that affects the dielectric loss. When x = 3, S.T. = 750 °C, the excellent optical and dielectric properties are: n = 3.17, $\varepsilon_r = 10.03$, $\alpha = 1.28$ cm⁻¹, tan $\delta = 0.0039$. Therefore, LaPO₄-3 wt.% LiF ceramics with low absorption coefficients and dielectric losses are optional for dielectric filters and lenses applied in the terahertz band.

The low-temperature co-firing ceramic (LTCC) technology has garnered significant attention due to its exceptional high-frequency characteristics, low energy consumption, and compact size. Among the various components of LTCC, the focus of research has been on LTCC materials. LaPO₄ ceramics have a high sintering temperature (1400 °C). Adding 3 wt.% LiF reduces the sintering temperature to 750 °C and has excellent dielectric properties. In this study, 20 wt.% Ag powder (10 μ m) and LaPO₄-3 wt.% LiF ceramic powder at 750 °C are mixed and co-fired at 750 °C for 2 h. Figure 11 shows the XRD results. The results show that no chemical reaction occurs after co-firing LaPO₄-3 wt.% LiF ceramic and Ag, and the co-firing is a success, which provides a new idea for the future application of LTCC [29].

4. Conclusions

In this paper, the LaPO₄-x wt.% LiF ($x = 0 \sim 5$) ceramics are prepared by the traditional solid-state reaction method, and the microwave dielectric properties at different frequency bands are studied. The XRD patterns show that the ceramics crystallized in single phase LaPO₄, no secondary phase observed. Rietveld refinement by using Fullprof software. The results show that the crystal structure of the ceramics is monoclinic and that the phase composition of the mixture has not changed. LiF exists in the liquid phase and fills the void in ceramic. Therefore, with the increase of LiF, the relative density of ceramic is greatly improved compared with the matrix, the sintering temperature reduces to 750 °C, and the quality factor rises to 81,466 GHz. The dielectric constant increases by 10.03, and the temperature coefficient is $-43.78 \text{ ppm}/^{\circ}\text{C}$ (x = 3). In the microwave frequency band, the ion shift polarization determines the dielectric constant, and the dielectric loss is mainly due to the lattice vibration. In the THz frequency band, the main factor affecting the dielectric constant is the ion displacement polarization, while the oxygen vacancy is the main factor affecting the dielectric loss. At 750 °C, LaPO₄-3 wt.% LiF ceramic powder does not react with Ag, and the co-firing is successful, indicating that $LaPO_4$ -x wt.% LiF ceramic is an alternative material for future LTCC technology.

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