

## Article

# Effect of Thermal Exposure on Mechanical Properties of Al-Si-Cu-Ni-Mg Aluminum Alloy

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**Abstract:** The microstructure morphology and evolution of mechanical properties are investigated in this study. The results show that the phases displayed no clear change after thermal exposure at 250 °C for 200 h. The tensile strength of the as-cast alloy showed a downward trend in different degrees with the increase in the tensile temperature, while the influence of elongation was opposite to the tensile strength. In addition, the tensile strength tended to be stable after thermal exposure at 250 °C for 100 h. The main creep mechanism of the as-cast alloy at a low temperature and low stress ( $T \leq 250$  °C;  $\sigma \leq 40$  MPa) is grain-boundary creep. The Monkman–Grant empirical formula was used to fit the relationship between the creep life and the minimum creep rate, and the fitting results are:  $t_r \cdot \dot{\epsilon}_{min}^{0.95} = 0.207$ .

**Keywords:** thermal exposure; casting alloys; mechanical properties; creep mechanism

## 1. Introduction

Because of its good specific strength and excellent mechanical properties, heat-resistant aluminum alloys have been widely used in aerospace engineering, vehicle engineering, and marine engineering. Al-Si-Cu-Ni-Mg alloy is a typical cast heat-resistant aluminum alloy, which is used to produce engine pistons. Therefore, the alloy usually works at a high temperature and alternating load. The service life of the piston is determined by the high-temperature endurance performance of the alloy [1–5].

With the increase in engine power, new challenges are presented to the high-temperature stability and endurance strength of piston materials. The development of a heat-resistant aluminum alloy with better high-temperature performance is imminent [6,7]. Thermal exposure experiments have been widely used by domestic and foreign scholars to verify the high-temperature stability of heat-resistant aluminum alloys. For instance, the eutectic silicon and Al<sub>3</sub>Ni phases of Al<sub>3</sub>Zr/Al-8Si-2Ni composites were spheroidized after being exposed at 450 °C for 150 h. The elongation of the Al<sub>3</sub>Zr/Al-8Si-2Ni composites was positively correlated with the heat exposure time, and the tensile strength tended to stabilize after decreasing [8]. A study reported that the average number of fatigue cycles before failure of the alloy was  $2.24 \times 10^7$  after thermal exposure at 425 °C/100 h, which was 3 times higher than the  $8.21 \times 10^6$  fatigue life of T6 heat treatment. The  $\delta$ -Al<sub>3</sub>CuNi phase and  $\gamma$ -Al<sub>7</sub>Cu<sub>4</sub>Ni were thermally stable intermetallic compounds, which significantly improved the high-temperature fatigue life of Al-Si piston alloys [9]. Zhang et al. [10] observed that the steady-state creep rate of the studied alloy decreased threefold when the loading changed from 70 MPa to 90 MPa with a constant temperature of 90 °C. Under a constant stress of 90 MPa, the steady creep rate of the studied alloy increased from  $1.94 \times 10^{-7}$  to  $1.48 \times 10^{-6}$  with the temperature rising from 90 °C to 150 °C. The grain size was basically unchanged under different compressive creep stresses. Zhao et al. [11] reported that the increase in YS was attributed to the precipitation of  $\beta''$ ,  $Q'$ , and  $\theta'$  nanophases in an  $\alpha$ -Al matrix after aging at 175 °C for 4 h. After thermal exposure at 350 °C for 100 h, the nanophases were mainly dissolved in the  $\alpha$ -Al matrix, and the ultra-fine eutectic silicon



**Citation:** Chen, F.; Liu, C.; Zuo, L.; Wu, Z.; He, Y.; Dong, K.; Li, G.; He, W. Effect of Thermal Exposure on Mechanical Properties of Al-Si-Cu-Ni-Mg Aluminum Alloy. *Crystals* **2023**, *13*, 236. <https://doi.org/10.3390/cryst13020236>

Academic Editors: Khitouni Mohamed and Joan-Josep Suñol

Received: 3 January 2023

Revised: 19 January 2023

Accepted: 20 January 2023

Published: 30 January 2023



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network was fractured and coarsened, resulting in the hardness of the alloy decreasing from 110 HV to 61 HV and the YS dropping to 117 MPa. The current literature reports that the high-temperature mechanical properties after thermal exposure are positively correlated with Ti and Cu content, which is attributed to the increase in the number of precipitated particles. The high strength value of the alloy after thermal exposure at 350 °C for 0.5 h is attributed to the number of nano sizes. The high heat resistance of T(Al<sub>20</sub>Cu<sub>2</sub>Mn<sub>3</sub>) and Fe(Al<sub>15</sub>(FeMn)<sub>3</sub>Si<sub>2</sub>) makes the strength stable after 100 h thermal exposure [12].

At present, the research on the high-temperature endurance strength of heat-resistant aluminum alloys mainly focuses on fatigue failure, and high-temperature creep performance is another key factor in the high-temperature service life of alloys [13–17]. Therefore, the aim of this study is to find out the thermal stability of the high-temperature endurance strength of Al-Si-Cu-Ni-Mg piston alloys, including the change law of the creep properties of the alloys when exposed to high temperature, and then guide subsequent study of heat-resistant aluminum alloy.

In this work, the microstructures of the cast alloy before and after thermal exposure were characterized using optical microscopy and scanning electron microscopy. The different temperature tensile properties of the cast alloy were investigated. The evolution law of the tensile strength during thermal exposure, as well as the creep mechanism of the alloy before and after thermal exposure, were investigated in this study. This study provides a basis for subsequent study on the influence of the high-temperature properties of Al-Si-Cu-Ni-Mg alloy.

## 2. Experimental Procedures

The chemical composition of the designed alloys is shown in Table 1. The alloys were prepared by melting pure Mg, Al, and Al-20Si (wt. %), Al-50Cu (wt. %), and Al-10Ni (wt. %) master alloys in an electric smelting furnace. Before melting the alloy, the crucible and mold were kept at 200 °C for 2 h. The Al-50Cu and Al-10Ni were kept at 150 °C for 2 h to remove moisture in the raw materials. First, pure Al and master alloys were added to the crucible and melted at 740 °C. Then pure Mg wrapped in aluminum foil was added to the melt at approximately 720 °C. The melt was refined using the refining agent C<sub>2</sub>Cl<sub>6</sub> at 740 °C. During the process of refining, the melt was stirred to ensure the refining effect and homogeneous chemical composition.

**Table 1.** Composition of cast Al-Si-Cu-Ni-Mg alloy.

Elements	Si	Cu	Ni	Mg	Mn	Zn	Ti	Fe	Ce	Al
wt. %	12.34	3.81	1.97	0.79	0.53	0.26	0.23	0.18	0.23	Bal

The metallographic samples were ground with different levels of sandpapers, polished with a PG-2B metallographic polishing machine for final polishing, and then etched with Keller's reagent (a mixed solution of 2.5% HNO<sub>3</sub>, 1.5% HCl, 1% HF, and 95% H<sub>2</sub>O) for 10 s. The fracture surface of the tensile specimen was observed with NOVA NanoSEM 230 scanning electron microscopy (Hong Kong, China). The tensile test was carried out on a WDW-10S (Jinan, China) universal tensile testing machine with a tensile rate of 0.4 mm/min at room temperature or 250 °C. The thermal exposure experiment was carried out in a tubular furnace with a temperature error of ± 2 °C. The thermal exposure temperature was set at 250 °C, and the time was 0–200 h. Each sample was measured after holding for 15 min at the set temperature, and the results are the average value of three parallel samples. The creep test was carried out on a CSS-3902 creep testing machine (Nanjing, China). The test temperature was 200–300 °C, and the stress was 20 MPa and 28 MPa. The sample was kept at the preset temperature for 1 h before loading. Two creep samples were tested for each condition, and no repeatability test was carried out if there was no abnormality. The geometric sizes of the tensile creep sample and the creep testing device are shown in Figures 1 and 2, respectively.

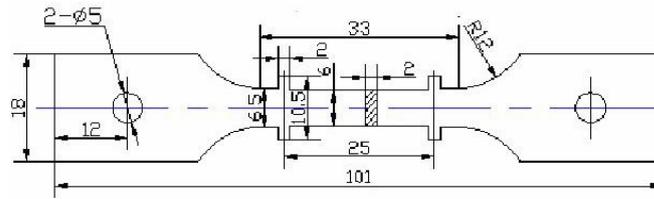


Figure 1. The geometric dimensions of tensile creep samples [18].

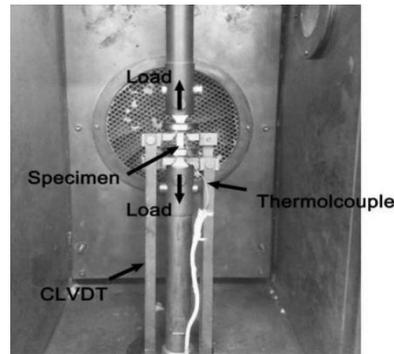


Figure 2. The tension creep testing setup.

### 3. Results and Discussion

#### 3.1. Microstructure and Morphology of Alloys in Different States

Figure 3 depicts the microstructure morphology of Al-Si-Cu-Ni-Mg alloy under different characterization methods. Figure 3a shows that the  $\alpha$ -Al was in the form of coarse dendrites, and most of the eutectic silicon was in the shape of short rods. Uniformly distributed black needle-like strengthening phases and some fishbone-like AlSiFeNiCu phases were detected. The characteristics of the fishbone-like phases were consistent with those reported in the related literature [19]. Figure 3b presents that primary silicon and eutectic silicon were randomly distributed on the Al matrix. The fishbone-like phase is observable in Figure 3a, and the network-like precipitation phase can be found in Figure 3b.

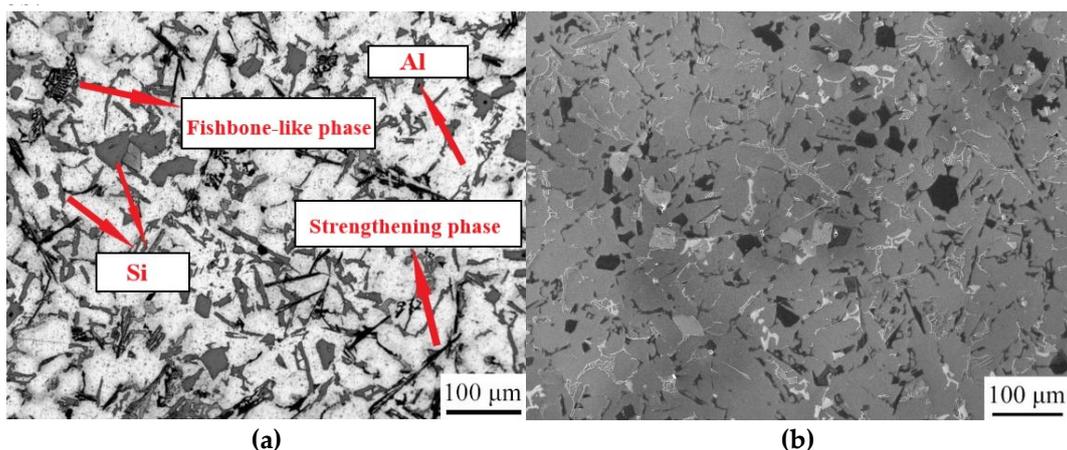
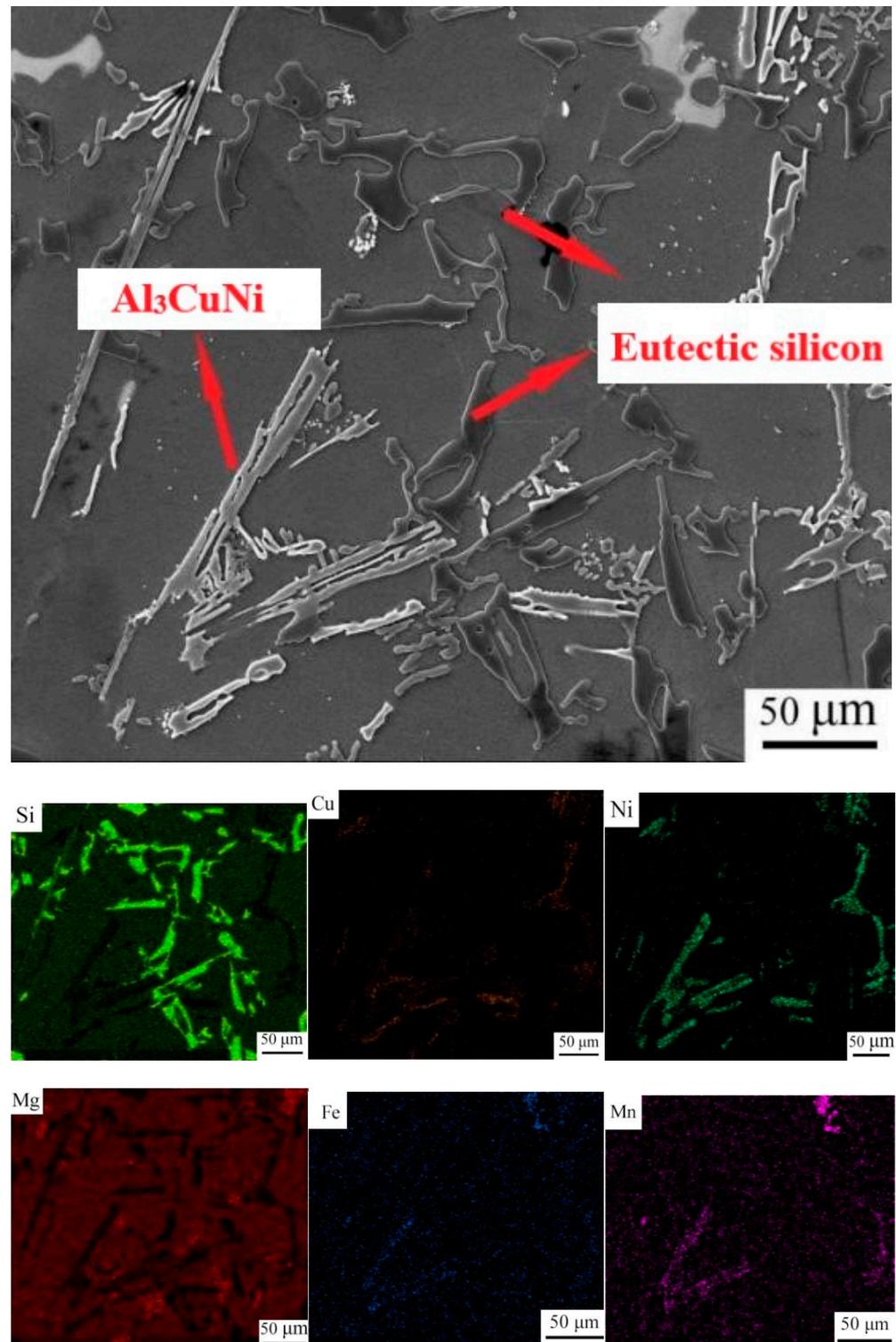


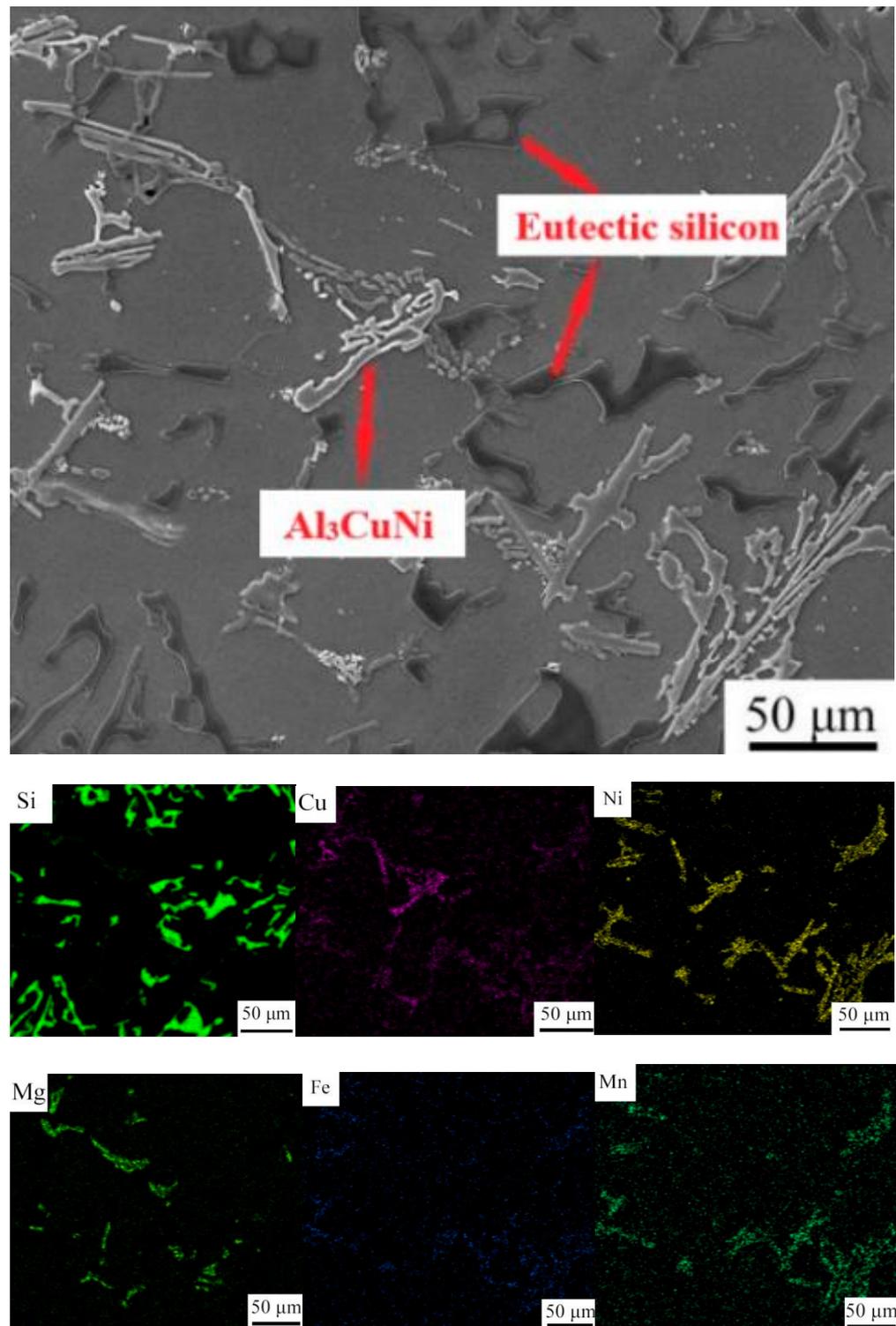
Figure 3. Microstructure of Al-Si-Cu-Ni-Mg alloy under (a) metallography and (b) scanning electron microscopy.

Compared with the as-cast alloy in Figure 4, the elements in the thermal exposed alloy occurred with different degrees of segregation and aggregation as shown in Figure 5. With the temperature increase, the solute atoms of the precipitation phase diffused, which promoted the nucleation and growth of  $Q$ -Al<sub>5</sub>Cu<sub>2</sub>Mg<sub>8</sub>Si<sub>6</sub>, displayed with the Mg element obviously aggregated. The distribution of Cu and Ni elements was more uniform, whereas

there was an increase in the distribution of slightly roughened rod-shaped eutectic Si. In conclusion, after 200 h thermal exposure at 250 °C, the microstructures at the micro scale demonstrated no significant changes.



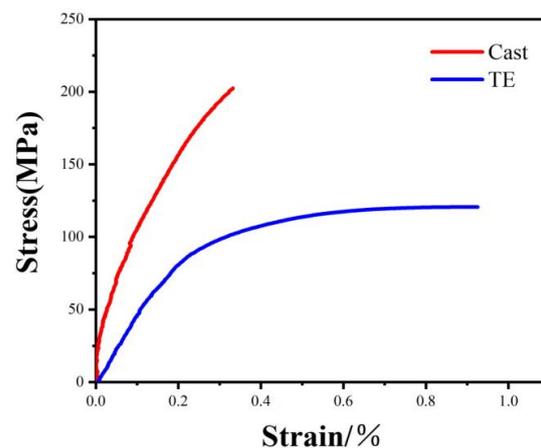
**Figure 4.** SEM morphology of as-cast Al-Si-Cu-Ni-Mg alloy and distribution of alloy constituent elements.



**Figure 5.** SEM morphology of thermally exposed Al-Si-Cu-Ni-Mg alloy and the distribution of alloy constituent element.

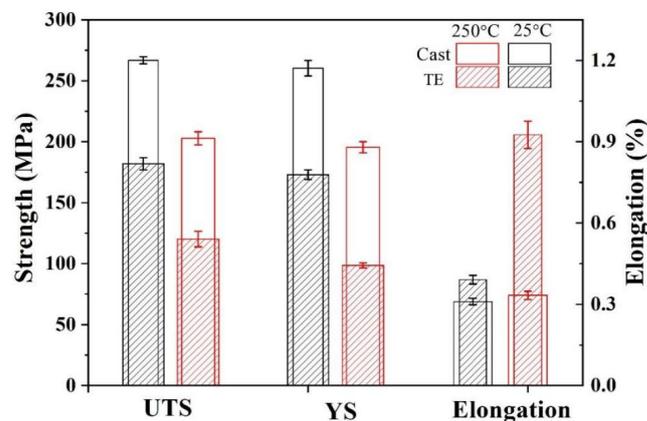
### 3.2. Evolution of Tensile Properties during Thermal Exposure

Figure 6 shows the typical stress–strain curves of the Al-Si-Cu-Ni-Mg alloy under different conditions tested at 250 °C. The ultimate tensile strength (UTS) of the alloy measured at 250 °C decreased from 202.8 MPa to 120.2 MPa after thermal exposure at 250 °C for 200 h, which is a 40.7% decrease. The elongation of the alloy measured at 250 °C increased from 0.33% to 0.95% after thermal exposure.



**Figure 6.** The typical stress–strain curves of Al-Si-Cu-Ni-Mg alloy under different conditions measured at 250 °C (TE represents thermal exposure at 250 °C for 200 h).

The tensile properties of the alloy with different states measured at different temperatures are displayed in Figure 7. The UTS of the as-cast alloy decreased from 266.8 MPa to 202.8 MPa when the temperature was increased from 25 °C to 250 °C, which is a drop of about 24%. The UTS of the TE alloy decreased from 266.8 MPa to 170.2 MPa at room temperature, with a decrease of about 36.2%. The elongation of the as-cast alloy increased slightly when the temperature was increased from 25 °C to 250 °C.



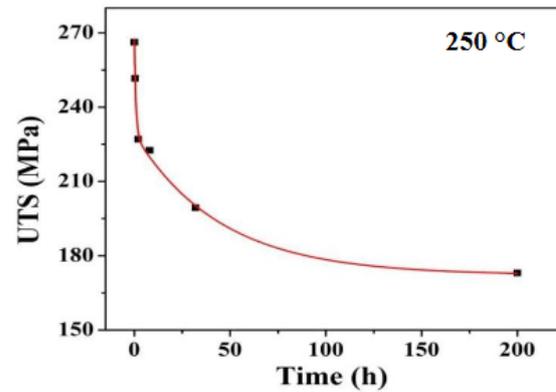
**Figure 7.** Tensile properties of the Al-Si-Cu-Ni-Mg alloy measured at different states (TE represents thermal exposure at 250 °C for 200 h).

The UTS of the TE alloy dropped from 170.2 MPa to 120.2 MPa when the temperature was increased from 25 °C to 250 °C, which is a significant drop of 29.4%. The elongation of the TE alloy increased greatly from 0.39% to 0.92% when the temperature was increased from 25 °C to 250 °C. With a test temperature of 250 °C, dislocations can climb to a certain extent under the action of stress, which provides more possibilities for alloy cross-slip and reduces the resistance of the cross-slip process. The climbing of a dislocation leads to a decrease in mechanical properties, as can be seen in the tensile strength decreasing accordingly.

As illustrated in Figure 8, the UTS of the alloy was tested at room temperature after thermal exposure at 250 °C for different times. The UTS of the studied alloys decreased greatly in the initial stage and then basically tended to be stable when the exposure temperature reached 32 h [12]. The UTS of the alloy was 173 MPa after thermal exposure at 250 °C for 200 h. Compared with the as-cast alloy, the UTS decreased by about 35%. The

relation between the UTS ( $\sigma$ ) at room temperature and the thermal exposure time ( $t$ ) at 250 °C fitted using ExpAssoc is shown as follows:

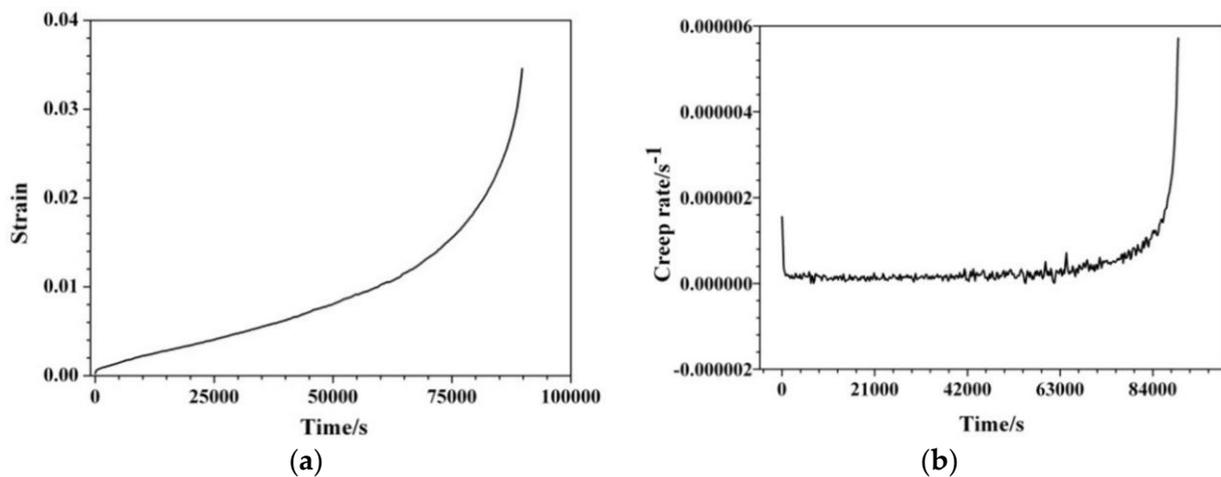
$$\sigma_t^{250} = \sigma_0^{250} - 38(1 - \text{Exp}\left(-\frac{t}{0.84}\right)) - 56.8(1 - \text{Exp}\left(-\frac{t}{45.2}\right)), R^2 = 0.98 \quad (1)$$



**Figure 8.** The ultimate tensile strength of the Al-Si-Cu-Ni-Mg alloy at room temperature fitted with exposure time at 250 °C for different times.

### 3.3. Creep Behavior before and after Thermal Exposure

The typical creep curve and creep-rate curve of the alloy are displayed in Figure 9. The creep rate decreased sharply in the first stage as a result of the work-hardening phenomenon. The creep stage where work hardening and dynamic recovery were in equilibrium was called the second stage of creep. The main feature of this stage was that the creep rate was constant and minimum. The creep rate of the material increased rapidly until fracturing, called the accelerated creep stage.



**Figure 9.** Typical creep curve (a) and creep-rate curve (b) of Al-Si-Cu-Ni-Mg alloy.

The minimum creep rates of the alloys at different states are presented in Table 2. As shown in Table 2, the minimum creep rates of the as-cast and thermally exposed alloys were similar at 250 °C. However, the minimum creep rate increased by almost an order of magnitude when the creep stress increased from 20 MPa to 28 MPa at the same temperature.

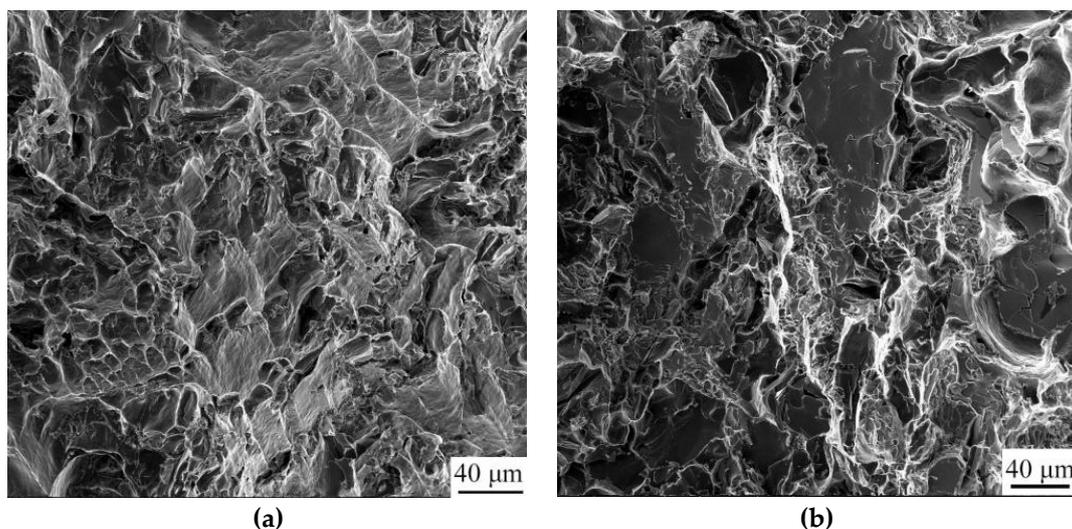
**Table 2.** Part of the minimum creep rate of Al-Si-Cu-Ni-Mg alloy before and after heat exposure under different creep conditions.

Alloy State	Temperature	Creep Rate ( $s^{-1}$ )	
		20 MPa	28 MPa
As-cast alloy	250 °C	$6.69 \times 10^{-9}$	$1.02 \times 10^{-8}$
Thermal exposed alloy	250 °C	$3.78 \times 10^{-9}$	$1.10 \times 10^{-8}$

Table 2 also displays the minimum creep rate of the alloy after thermal exposure at 250 °C for 200 h. At 20 MPa/250 °C, the minimum creep rate of the alloy was around  $10^{-9} \cdot s^{-1}$ . However, the minimum creep rate of the alloy under the same conditions was lower than the minimum creep rate of the as-cast state, indicating that the creep resistance of the alloy at a low temperature was slightly improved after the thermal exposure. Compared with the minimum creep rate of the as-cast alloy, the minimum creep rate of the thermal exposed alloy was increased by an order of magnitude, in which the creep stress increased from 20 MPa to 28 MPa. Therefore, the creep resistance of the alloy significantly deteriorated.

The creep fracture morphology of the alloy at 20 MPa and 28 MPa is displayed in Figure 10. In contrast with Figure 10a, the quantity of dimples in Figure 10b significantly decreases. Apparently, Figure 10b has almost no dimples and a large area of cleavage planes.

Owing to the interactions of dislocation creating long-range stress fields in the alloy, the dislocations relied on shear stress to overcome these stress fields and move. Hence, the alloy with the higher applied stress load had a faster creep rate and worse creep resistance. The applied load condition of 28 MPa offered a larger shear stress, which provided more possibilities for the dislocation of the alloy. The larger applied load made the high-strength, hard and brittle phase more prone to brittle fracture, and the  $\alpha$ -Al was less stressed. The large-scale precipitates had already initiated cracks and led to cracking when plastic deformation occurred, so a mass of dimples formed after plastic deformation, which is observable in Figure 10a but not in Figure 10b. Under the interaction of the stress field and high temperature, the strengthening phases shown in Figure 10b became globular more rapidly than those shown in Figure 10a, and the ability to deform was also decreased, which is explained by the phenomenon of a large-area cleavage surface occurring and no dimples forming after fracturing.



**Figure 10.** The creep fracture morphology of Al-Si-Cu-Ni-Mg alloy at (a) 20 MPa/250 °C and (b) 28 MPa/250 °C.

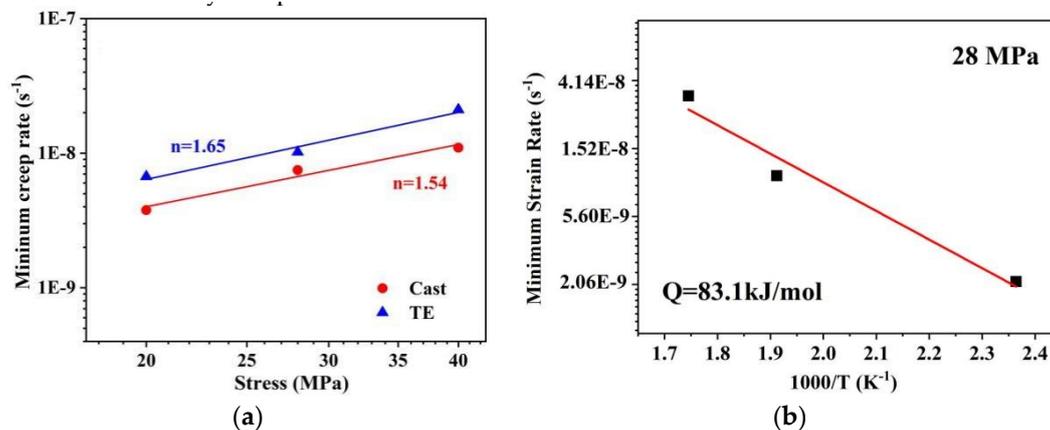
With 20 MPa of applied stress loaded, dislocation was hindered as a result of less shear stress distribution, and the tendency of stress concentration was less. The forces distributed in the  $\alpha$ -Al and strengthening phase were relatively balanced. Compared with 28 MPa, the coarsening rate of the strengthening phase was lower with the combined action of the small stress field and temperature. The deformation performance declined slowly, and the strengthening phase reached the stress limit only after the plastic deformation of the  $\alpha$ -Al produced cracks, manifesting as plenty of dimples and the cleavage surface being small. This could explain why the creep resistance deteriorated signally with the creep stress increasing from 20 MPa to 28 MPa at the same creep temperature.

The minimum creep rate and creep stress of materials are closely related to the creep temperature, and their corresponding relationship is expressed by the classical power law Equation (2) [16]:

$$\dot{\epsilon} = A \cdot \sigma^n \cdot \exp\left(-\frac{Q}{RT}\right) \quad (2)$$

where  $\dot{\epsilon}$  is the minimum creep rate,  $A$  is the materials constant,  $\sigma$  is the creep stress,  $R$  is the gas constant ( $R = 8.31 \text{ J/mol}$ ),  $T$  is the absolute temperature,  $n$  is the creep stress exponent, and  $Q$  is the creep activation energy. Stress exponent  $n$  and activation energy  $Q$  are often used to infer the creep mechanism of alloys. The creep mechanism corresponding to stress exponent  $n$  and activation energy  $Q$  are summarized simply as follows: When the grain-boundary creep  $n \leq 2$ , which is divided into grain-boundary diffusion and grain-boundary slip, the creep activation energy  $Q$  corresponding to this creep mechanism is 82 kJ/mol. When  $n \approx 3$ , this indicates the dislocation slip mechanism. When  $n$  is 4–6, this indicates a dislocation climbing mechanism, and the minimum creep rate is related to the diffusion of vacancy.

Using Equation (2), the creep stress exponent of the alloy at different temperatures was obtained. Figure 11 displays that the creep stress exponent of the as-cast alloy was about 1–2, and it is speculated that the creep mechanism of the material under this creep condition was grain-boundary creep. After thermal exposure at 250 °C for 200 h, the creep stress exponent of the alloy at 250 °C was 1.54, so the creep mechanism was grain-boundary creep.



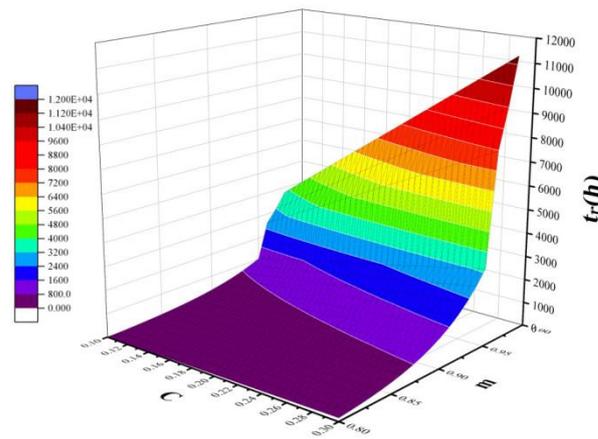
**Figure 11.** Al-Si-Cu-Ni-Mg alloy (a) creep stress exponent and (b) creep activation energy (TE represents thermal exposure at 250 °C for 200 h).

With the condition of invariable stress, the creep activation energy of the material at a temperature of 250 °C was obtained by changing the creep temperature; the result is shown in Figure 11b. Figure 11b displays that the creep activation energy of the alloy at 250 °C is 83.1 kJ/mol, which is very close to the theoretical grain-boundary diffusion activation energy of 82 kJ/mol. It also illustrates that the creep mechanism of the Al-Si-Cu-Ni-Mg alloy at 250 °C is grain-boundary creep.

After the steady-state creep, the accelerated creep stage occurred until the final failure of the material. The Monkman–Grant model is the most widely used among many material creep failure prediction models, which are summarized as the following Equation (3):

$$t_r \cdot \dot{\epsilon}_{min}^m = C \tag{3}$$

where  $t_r$  is the creep life,  $\dot{\epsilon}_{min}$  is the minimum creep rate, and  $m$  and  $C$  are both material constants. In order to further understand the influence of material constants  $m$  and  $C$  on the creep life of the material, let the minimum creep rate =  $1.10 \times 10^{-8} \text{ s}^{-1}$ , and the results of the influence of the material constants  $m$  and  $C$  on the creep life are presented in Figure 12.

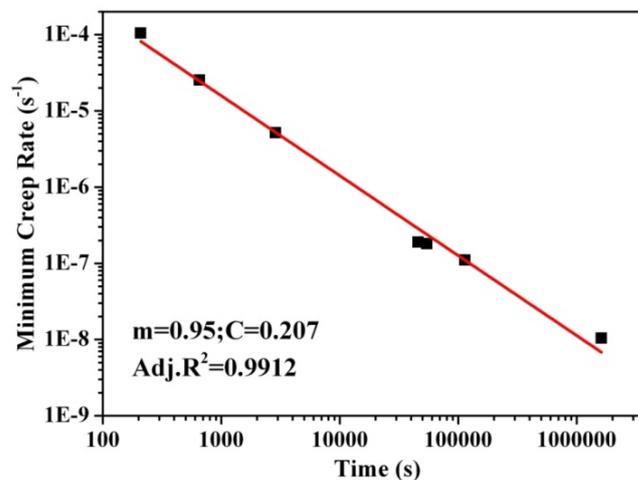


**Figure 12.** The influence of  $C$  and  $m$  constants in the Monkman–Grant model on the fracture life under the condition of constant minimum creep rate.

As shown in Figure 12, under the condition of constant  $m$ , creep life and material constant  $C$  increase linearly, and the increase is inconspicuous, while under the condition of constant  $C$ , creep life and material constant  $m$  increase exponentially, and the closer the  $m$  value is to 1, the more obvious its effect on creep life is.

The fitting results are displayed in Figure 13, showing that the data fitting degree is as high as 99.12%, and the material constant  $m$  is also close to 1, indicating the good creep resistance of the material to a certain extent. The Monkman–Grant Equation (4) is written as follows:

$$t_r \cdot \dot{\epsilon}_{min}^{0.95} = 0.207 \tag{4}$$



**Figure 13.** Monkman–Grant fitting of creep life of Al-Si-Cu-Ni-Mg alloy.

From Equation (4), it was calculated that for the creep life to reach 1000 h, the minimum creep rate should be lower than  $2.4 \times 10^{-8} \text{ s}^{-1}$ . The relationship between the specific creep life and the corresponding minimum creep rate limit value are shown in Table 3.

**Table 3.** Correspondence table of creep life and minimum creep rate.

$t_r$ (h)	10	100	1000
$\dot{\epsilon}$ ( $\text{s}^{-1}$ )	$3.0 \times 10^{-6}$	$2.7 \times 10^{-7}$	$2.4 \times 10^{-8}$

#### 4. Conclusions

The effect of thermal exposure on the mechanical properties of Al-Si-Cu-Ni-Mg aluminum alloy has been investigated in the present work. The main conclusions of this work are summarized as follows:

1. The main phases at the micro scale of the Al-Si-Cu-Ni-Mg alloy were stable during thermal exposure at 250 °C.
2. The UTS of the studied alloys decreased greatly in the initial stage and then basically tended to be stable after thermal exposure at 250 °C for about 100 h.
3. The creep resistance deteriorated signally when the creep stress was increased from 20 MPa to 28 MPa at the same creep temperature.
4. The main creep mechanism of the as-cast Al-Si-Cu-Ni-Mg alloy at a low temperature and low stress ( $T \leq 250 \text{ °C}$ ;  $\sigma \leq 40 \text{ MPa}$ ) was grain-boundary creep. The Monkman–Grant empirical formula was used to fit the relationship between the creep life and the minimum creep rate of the cast Al-Si-Cu-Ni-Mg alloy, and the fitting results were:  $t_r \cdot \dot{\epsilon}_{min}^{0.95} = 0.207$ .

**Author Contributions:** Conceptualization, F.C.; methodology, C.L.; software, L.Z.; validation, Z.W.; formal analysis, Y.H.; investigation, K.D.; resources, Z.W.; data curation, L.Z.; writing—original draft preparation, G.L.; writing—review and editing, F.C.; visualization, W.H.; supervision, L.Z.; project administration, L.Z.; funding acquisition, Y.H. All authors have read and agreed to the published version of the manuscript.

**Funding:** This research was funded by the Natural Science Foundation of the Jiangsu Higher Education Institutions of China (Grant No. 20KJB430015) and the Natural Science Foundation of Jiangsu Province (Grant No. BK20201467).

**Data Availability Statement:** Not applicable.

**Acknowledgments:** This work was supported by the Natural Science Foundation of the Jiangsu Higher Education Institutions of China (Grant No. 20KJB430015). This work was supported by the Natural Science Foundation of Jiangsu Province (Grant No. BK20201467).

**Conflicts of Interest:** The authors declare no conflict of interest.

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