

# Synthesis, Crystal Structure, and Hirshfeld Surface Analysis of Hexachloroplatinate and Tetrachlorouranylate of 3-Carboxypyridinium. Halogen Bonds and $\pi$ -Interactions vs Hydrogen Bonds

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**Table S1.** Bond Lengths for I at 100 K.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Pt <sup>1</sup>	Cl1	2.3180(8)	N1	C2	1.340(4)
Pt <sup>1</sup>	Cl1 <sup>1</sup>	2.3180(8)	N1	C6	1.337(4)
Pt <sup>1</sup>	Cl2 <sup>1</sup>	2.3090(8)	C2	C3	1.391(4)
Pt <sup>1</sup>	Cl2	2.3090(8)	C3	C4	1.388(5)
Pt <sup>1</sup>	Cl3	2.3107(8)	C3	C7	1.495(5)
Pt <sup>1</sup>	Cl3 <sup>1</sup>	2.3107(8)	C4	C5	1.384(5)
O <sup>1</sup>	C7	1.236(4)	C5	C6	1.371(5)
O <sup>2</sup>	C7	1.323(5)			

<sup>1</sup>1-X, 1-Y, 1-Z

**Table S2.** Bond Angles for I at 100 K.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Pt1	Cl1 <sup>1</sup>	180.0	Cl3 <sup>1</sup>	Pt1	Cl1	89.57(3)
Cl2	Pt1	Cl1	89.06(3)	Cl3 <sup>1</sup>	Pt1	Cl3	180.00(3)
Cl2	Pt1	Cl1 <sup>1</sup>	90.94(3)	C6	N1	C2	122.6(3)
Cl2 <sup>1</sup>	Pt1	Cl1 <sup>1</sup>	89.06(3)	N1	C2	C3	119.4(3)
Cl2 <sup>1</sup>	Pt1	Cl1	90.94(3)	C2	C3	C7	117.2(3)
Cl2	Pt1	Cl2 <sup>1</sup>	180.0	C4	C3	C2	118.7(3)
Cl2	Pt1	Cl3	90.89(3)	C4	C3	C7	124.1(3)
Cl2	Pt1	Cl3 <sup>1</sup>	89.11(3)	C5	C4	C3	120.1(3)
Cl2 <sup>1</sup>	Pt1	Cl3	89.11(3)	C6	C5	C4	119.0(3)
Cl2 <sup>1</sup>	Pt1	Cl3 <sup>1</sup>	90.89(3)	N1	C6	C5	120.2(3)
Cl3 <sup>1</sup>	Pt1	Cl1 <sup>1</sup>	90.43(3)	O1	C7	O2	124.0(3)
Cl3	Pt1	Cl1 <sup>1</sup>	89.57(3)	O1	C7	C3	119.8(3)
Cl3	Pt1	Cl1	90.43(3)	O2	C7	C3	116.2(3)

<sup>1</sup>H-X, <sup>1</sup>H-Y, <sup>1</sup>H-Z

**Table S3.** Torsion Angles for **I** at 100 K.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C2	C3	C4	0.0(5)	C3	C4	C5	C6	-0.9(5)
N1	C2	C3	C7	-179.3(3)	C4	C3	C7	O1	151.1(3)
C2	N1	C6	C5	0.1(5)	C4	C3	C7	O2	-28.8(5)
C2	C3	C4	C5	0.6(5)	C4	C5	C6	N1	0.5(5)
C2	C3	C7	O1	-29.8(5)	C6	N1	C2	C3	-0.4(5)
C2	C3	C7	O2	150.4(3)	C7	C3	C4	C5	179.8(3)

**Table S4.** Bond Lengths for **I** at 296 K.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Pt1	Cl1 <sup>1</sup>	2.3220(6)	N1	C2	1.340(3)
Pt1	Cl1	2.3220(6)	N1	C6	1.335(3)
Pt1	Cl2	2.3144(5)	C2	C3	1.375(3)
Pt1	Cl2 <sup>1</sup>	2.3145(5)	C3	C4	1.392(3)
Pt1	Cl3	2.3139(5)	C3	C7	1.496(3)
Pt1	Cl3 <sup>1</sup>	2.3139(5)	C4	C5	1.375(4)
O1	C7	1.231(3)	C5	C6	1.365(4)
O2	C7	1.317(4)			

<sup>1</sup>H-X, <sup>1</sup>H-Y, <sup>1</sup>H-Z

**Table S5.** Bond Angles for **I** at 296 K.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1 <sup>1</sup>	Pt1	Cl1	180.0	Cl3	Pt1	Cl2 <sup>1</sup>	89.18(2)
Cl2 <sup>1</sup>	Pt1	Cl1	90.87(2)	Cl3	Pt1	Cl3 <sup>1</sup>	180.0
Cl2	Pt1	Cl1 <sup>1</sup>	90.86(2)	C6	N1	C2	122.5(2)
Cl2	Pt1	Cl1	89.14(2)	N1	C2	C3	120.3(2)
Cl2 <sup>1</sup>	Pt1	Cl1 <sup>1</sup>	89.14(2)	C2	C3	C4	117.7(2)
Cl2	Pt1	Cl2 <sup>1</sup>	180.0	C2	C3	C7	118.2(2)
Cl3 <sup>1</sup>	Pt1	Cl1 <sup>1</sup>	90.49(2)	C4	C3	C7	124.1(2)
Cl3	Pt1	Cl1	90.49(2)	C5	C4	C3	120.5(2)
Cl3	Pt1	Cl1 <sup>1</sup>	89.51(2)	C6	C5	C4	119.4(3)
Cl3 <sup>1</sup>	Pt1	Cl1	89.51(2)	N1	C6	C5	119.6(3)
Cl3 <sup>1</sup>	Pt1	Cl2 <sup>1</sup>	90.82(2)	O1	C7	O2	124.0(2)
Cl3 <sup>1</sup>	Pt1	Cl2	89.18(2)	O1	C7	C3	119.5(2)
Cl3	Pt1	Cl2	90.81(2)	O2	C7	C3	116.5(2)

<sup>1</sup>H-X,<sup>1</sup>H-Y,<sup>1</sup>H-Z

**Table S6.** Hydrogen Bonds for **I** at 296 K.

D	H	A	d(D-H)/Å	d(H-A)/Å	d(D-A)/Å	D-H-A/°
O2	H2B	Cl1 <sup>1</sup>	0.86(2)	2.71(3)	3.531(3)	161(8)
O2	H2B	Cl3 <sup>1</sup>	0.86(2)	2.86(7)	3.434(3)	126(7)
O2	H2C	Cl3 <sup>2</sup>	0.87(2)	2.66(4)	3.451(3)	152(6)
N1	H1A	O1 <sup>3</sup>	0.86	1.93	2.762(3)	162.3

$$^1\text{-X}, ^1\text{-Y}, ^2\text{-Z}; ^2\text{-1/2+X}, ^1\text{/2-Y}, ^1\text{/2+Z}; ^3\text{-1/2+X}, ^3\text{/2-Y}, ^{-1/2+Z}$$

**Table S7.** Torsion Angles for **I** at 296 K.

Table 7 Torsion Angles for NicPt_RT.									
A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C2	C3	C4	0.3(4)	C3	C4	C5	C6	-1.1(4)
N1	C2	C3	C7	-178.9(2)	C4	C3	C7	O1	151.5(2)
C2	N1	C6	C5	0.2(4)	C4	C3	C7	O2	-28.2(4)
C2	C3	C4	C5	0.6(4)	C4	C5	C6	N1	0.7(4)
C2	C3	C7	O1	-29.4(4)	C6	N1	C2	C3	-0.7(4)
C2	C3	C7	O2	151.0(3)	C7	C3	C4	C5	179.7(3)

**Table S8.** Bond Lengths for **II** at 100 K.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
U1	Cl1 <sup>1</sup>	2.6461(8)	N1	C2	1.341(4)
U1	Cl1	2.6461(8)	N1	C6	1.342(4)
U1	Cl2	2.6694(7)	C2	C3	1.383(4)
U1	Cl2 <sup>1</sup>	2.6694(7)	C3	C4	1.394(4)
U1	O1 <sup>1</sup>	1.780(2)	C3	C7	1.512(4)
U1	O1	1.780(2)	C4	C5	1.389(4)
O2	C7	1.235(4)	C5	C6	1.378(4)
O3	C7	1.334(4)			

$$^1\text{-X}, ^1\text{-Y}, ^{-}\text{Z}$$

**Table S9.** Bond Angles for **II** at 100 K.

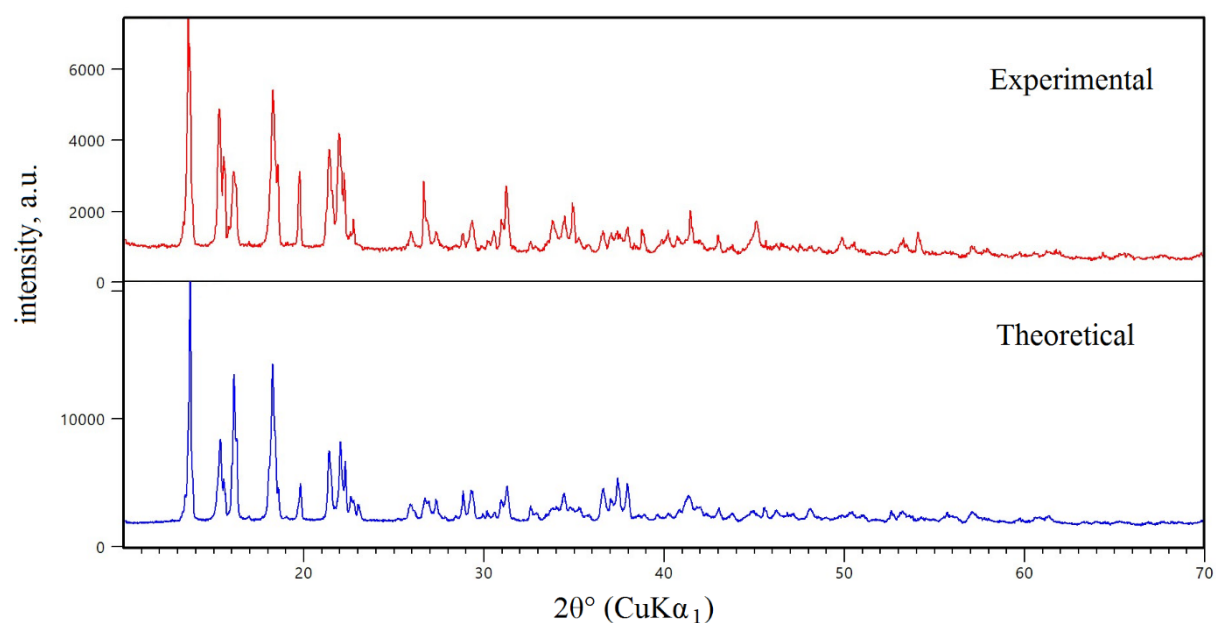
Table 5 Bond Angles for U9.							
Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1 <sup>1</sup>	U1	Cl1	180.0	O1 <sup>1</sup>	U1	Cl2	90.14(8)
Cl1	U1	Cl2	89.47(2)	O1	U1	O1 <sup>1</sup>	180.0
Cl1	U1	Cl2 <sup>1</sup>	90.53(2)	C2	N1	C6	123.3(3)
Cl1 <sup>1</sup>	U1	Cl2	90.53(2)	N1	C2	C3	119.2(3)
Cl1 <sup>1</sup>	U1	Cl2 <sup>1</sup>	89.47(2)	C2	C3	C4	118.9(3)
Cl2 <sup>1</sup>	U1	Cl2	180.000(17)	C2	C3	C7	122.9(3)
O1	U1	Cl1	89.81(8)	C4	C3	C7	118.2(3)
O1	U1	Cl1 <sup>1</sup>	90.19(8)	C5	C4	C3	120.1(3)
O1 <sup>1</sup>	U1	Cl1 <sup>1</sup>	89.81(8)	C6	C5	C4	118.8(3)
O1 <sup>1</sup>	U1	Cl1	90.19(8)	N1	C6	C5	119.6(3)
O1 <sup>1</sup>	U1	Cl2 <sup>1</sup>	89.86(8)	O2	C7	O3	124.4(3)
O1	U1	Cl2	89.86(8)	O2	C7	C3	118.4(3)

O1	U1	Cl2 <sup>1</sup>	90.14(8)	O3	C7	C3	117.2(3)
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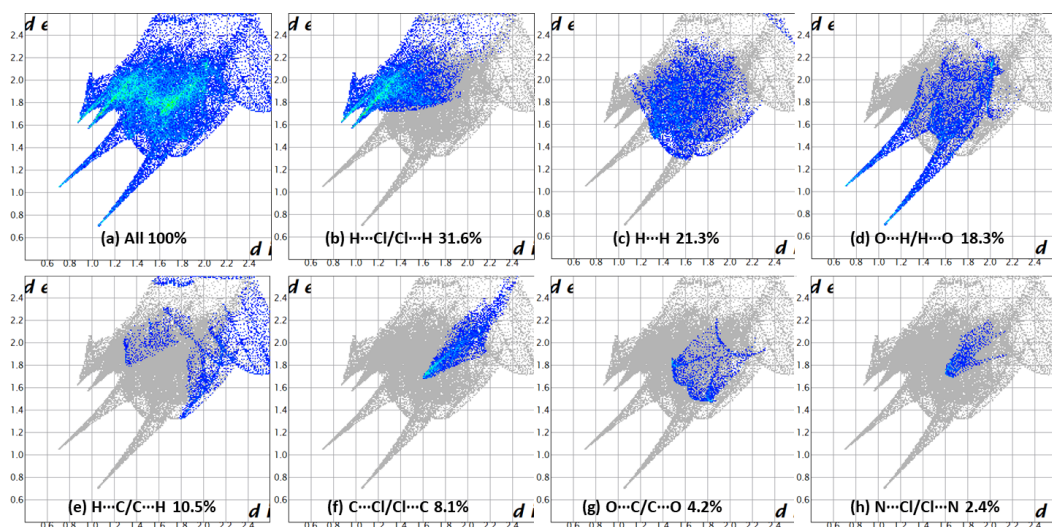
<sup>1</sup>1-X,1-Y,-Z

**Table S10.** Torsion Angles for **II** at 100 K.

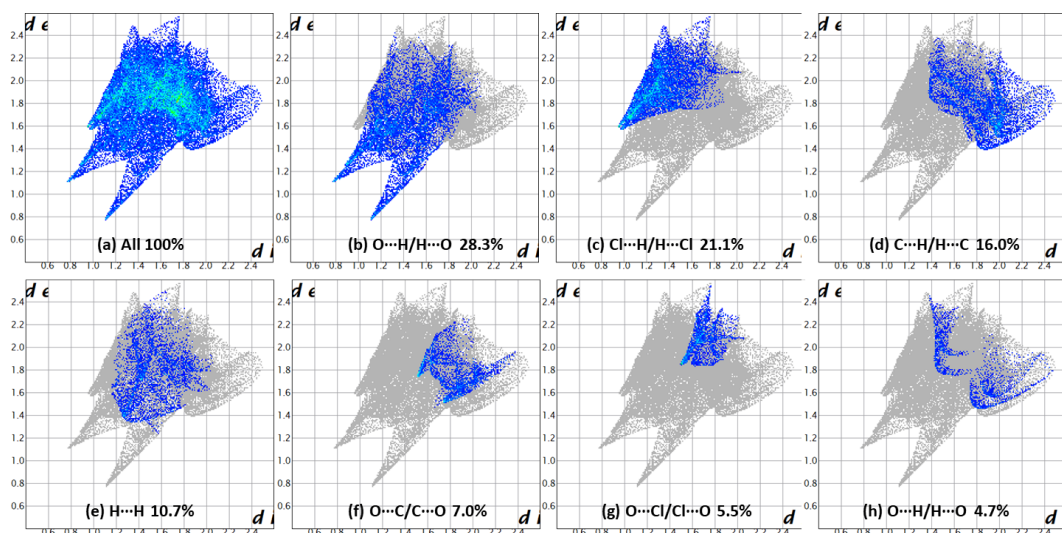
A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C2	C3	C4	1.8(4)	C3	C4	C5	C6	0.4(5)
N1	C2	C3	C7	-178.1(3)	C4	C3	C7	O2	15.5(4)
C2	N1	C6	C5	-1.6(5)	C4	C3	C7	O3	-162.5(3)
C2	C3	C4	C5	-2.0(5)	C4	C5	C6	N1	1.4(5)
C2	C3	C7	O2	-164.6(3)	C6	N1	C2	C3	0.0(5)
C2	C3	C7	O3	17.4(4)	C7	C3	C4	C5	177.9(3)



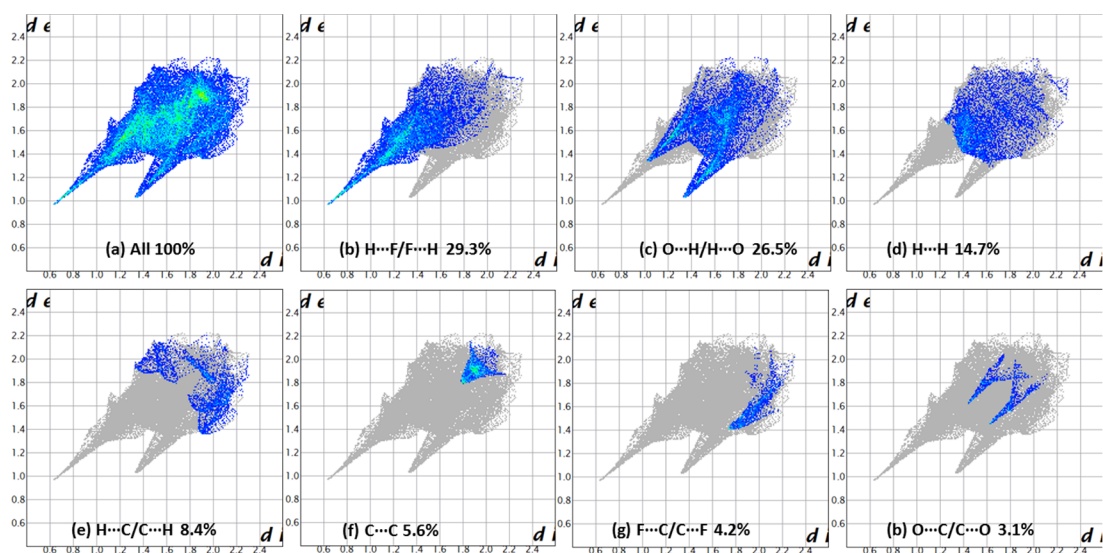
**Figure S1.** Experimental and theoretical powder XRD spectra of **I** at 296 K.



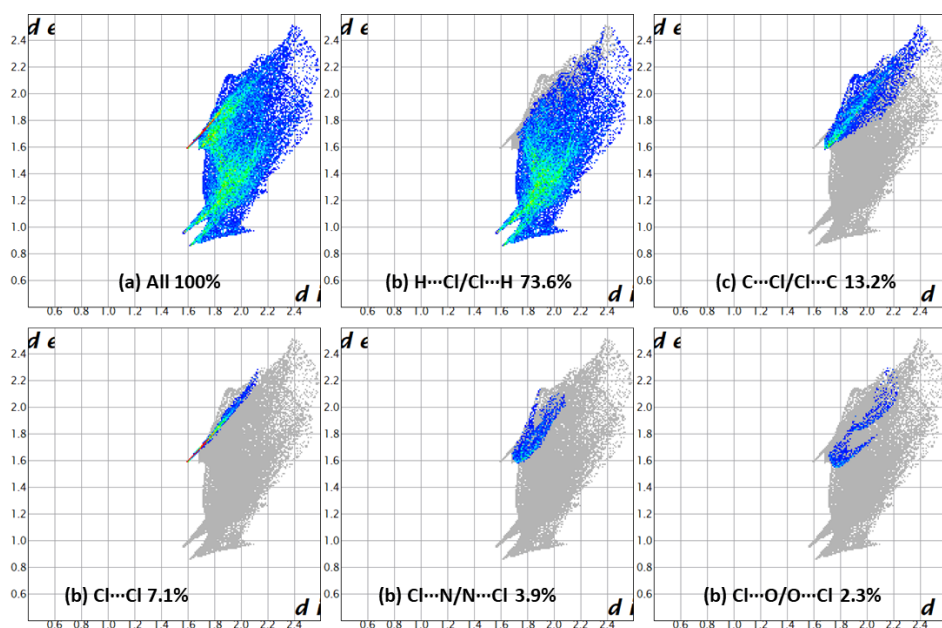
**Figure S2.** The 2D fingerprint plots of interatomic interactions for cation in **I** at 100 K show the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.



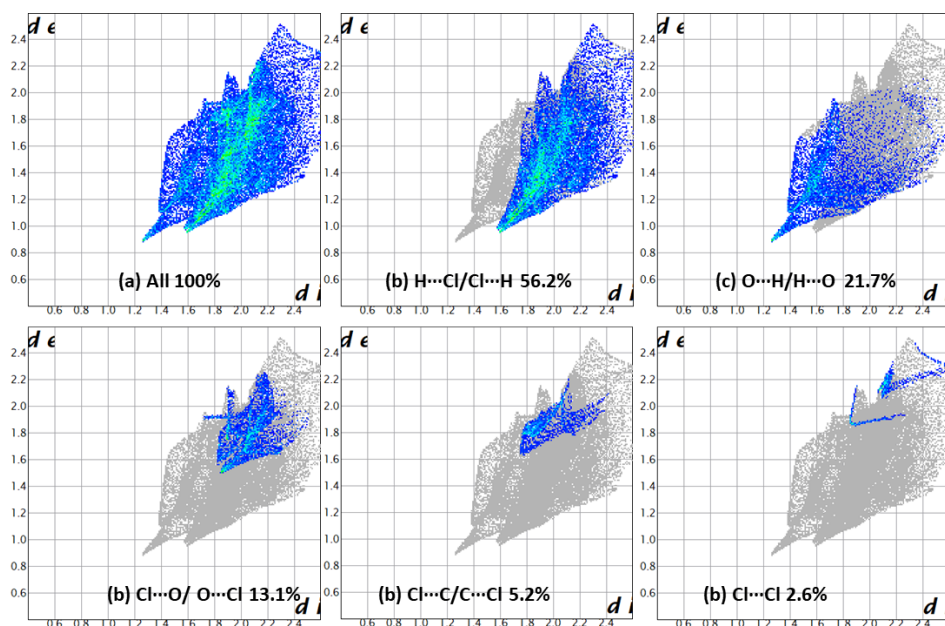
**Figure S3.** The 2D fingerprint plots of interatomic interactions for cation in **II** at 100 K show the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.



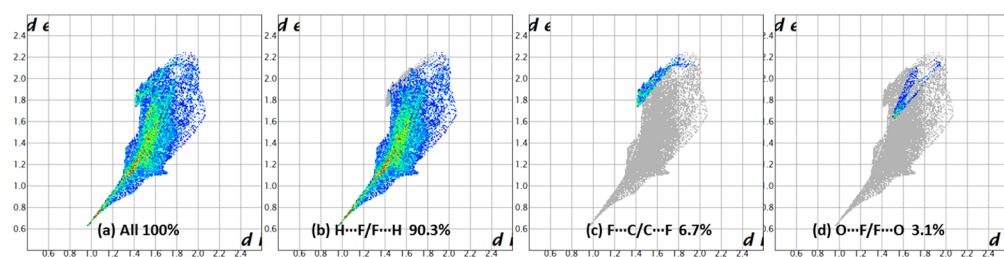
**Figure S4.** The 2D fingerprint plots of interatomic interactions for cation in **III** at 100 K showing the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.



**Figure S5.** The 2D fingerprint plots of interatomic interactions for anion in **I** at 100 K show the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.



**Figure S6.** The 2D fingerprint plots of interatomic interactions for anion in **II** at 100 K show the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.



**Figure S7.** The 2D fingerprint plots of interatomic interactions for anion in **III** at 100 K showing the percentages of contacts contributed to the total Hirshfeld surface area of the molecules.