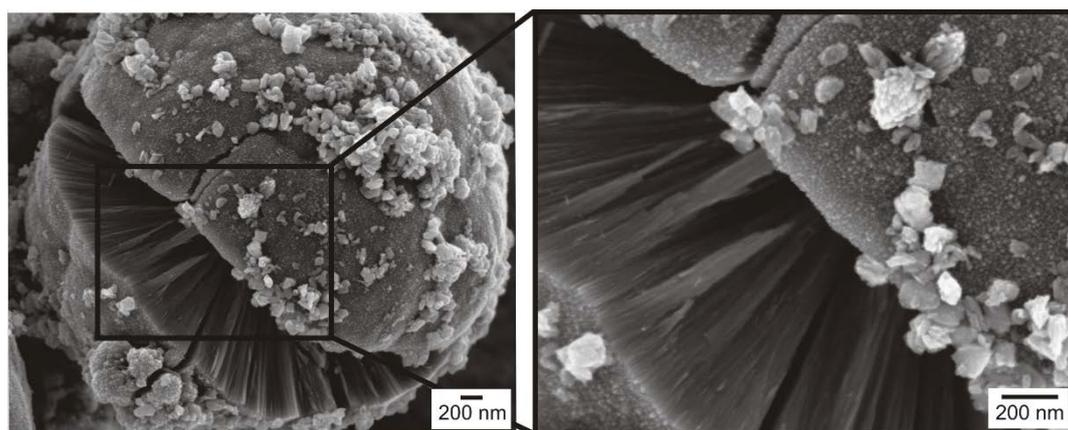


# Supplementary Materials: Photocatalytic TiO<sub>2</sub> Nanorod Spheres and Arrays Compatible with Flexible Applications

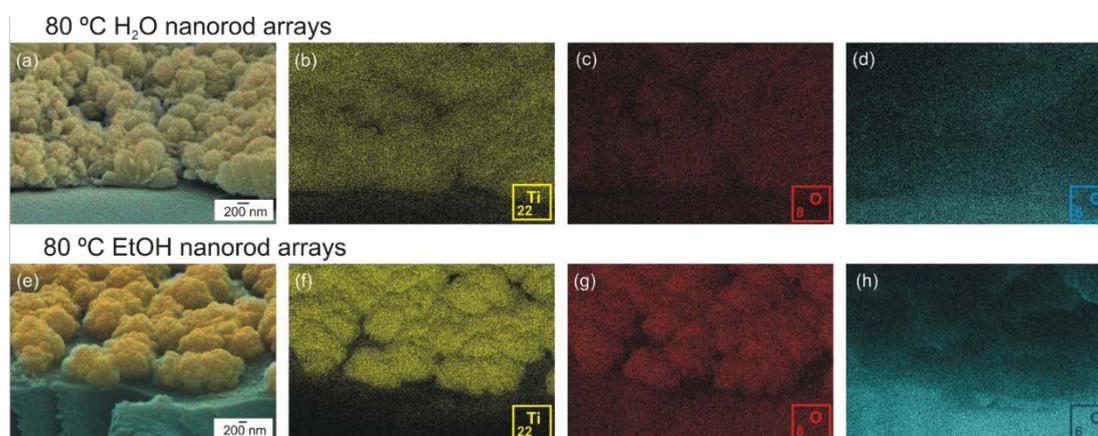
Daniela Nunes, Ana Pimentel, Lidia Santos, Pedro Barquinha, Elvira Fortunato and Rodrigo Martins

The scanning electron microscopy (SEM) images in Figure S1 shows a cracked micro-sized sphere that evidences the nanorods radially arranged.



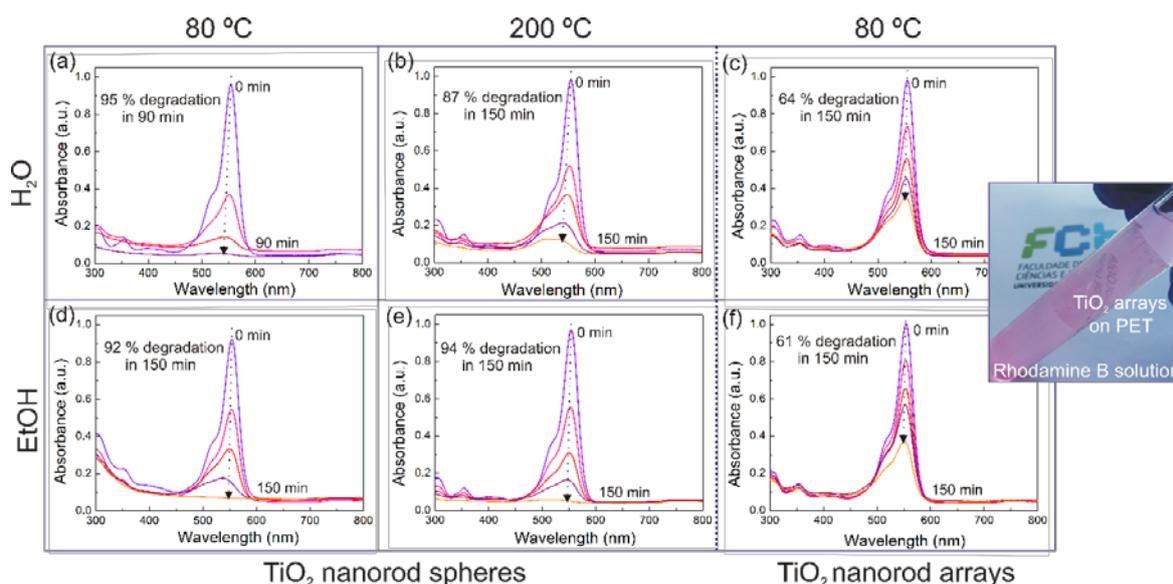
**Figure S1.** SEM images of the 80 °C H<sub>2</sub>O nanorod spheres showing the nanorods radially arranged. The inset evidences the nanorod structures inside the sphere.

The energy dispersive X-Ray spectroscopy (EDS) analyses showed homogeneous distributions of Ti and O in both materials produced. The C map is presented as the substrate is from polymer origin (Figure S2).



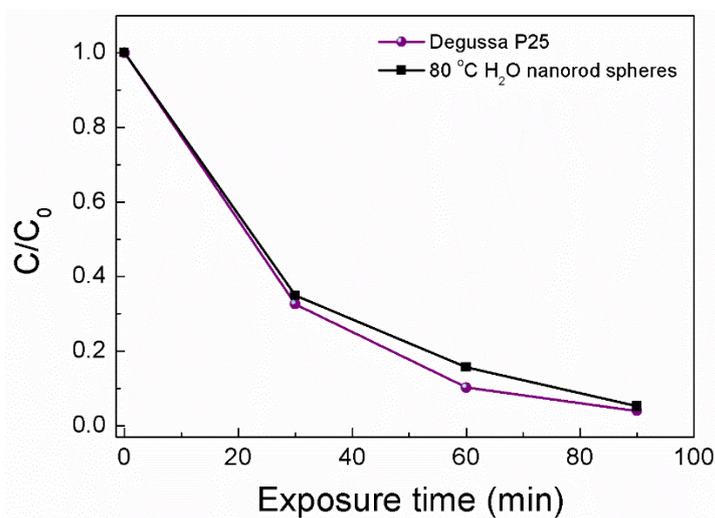
**Figure S2.** Cross-section SEM images (artificial colored) of the: (a) 80 °C H<sub>2</sub>O; and (e) 80 °C EtOH nanorod arrays grown on ZnO seeded PET substrates, together with the corresponding X-ray maps of Ti: (b,f); O (c,g); and C (d,h).

Figure S3 shows the rhodamine B absorbance spectra at different ultraviolet (UV) exposure times with the different TiO<sub>2</sub> materials acting as photocatalytic agents.



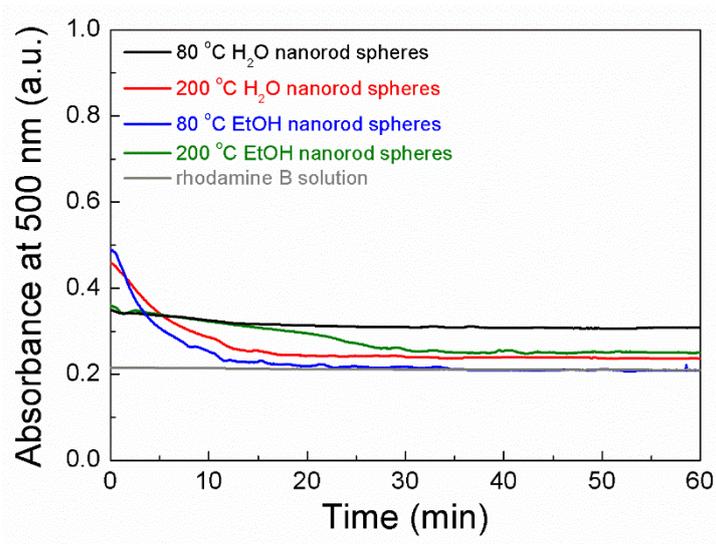
**Figure S3.** Rhodamine B absorbance spectra at different UV exposure times for the TiO<sub>2</sub> nanorod spheres produced with water at: (a) 80 °C; and (b) 200 °C; and with ethanol at: (d) 80 °C; and (e) 200 °C. The TiO<sub>2</sub> nanorod arrays grown on PET were also tested as photocatalysts at 80 °C for: (c) water; and (f) ethanol. The photograph illustrates the PET substrate with the TiO<sub>2</sub> nanorod arrays covering a tube containing the pollutant solution. The absorbance spectra have not been normalized.

Rhodamine B degradation ratio ( $C/C_0$ ) vs. UV exposure time for both the 80 °C H<sub>2</sub>O nanorod spheres and Degussa P25 powders (Figure S4). The spheres show comparable degradation behavior under UV radiation of the most used commercial TiO<sub>2</sub> powder.



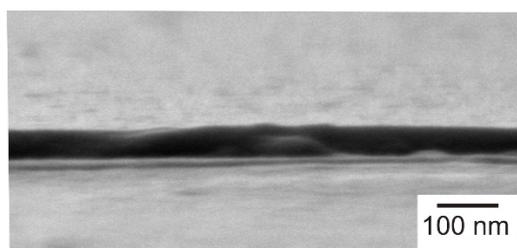
**Figure S4.** Rhodamine B degradation ratio ( $C/C_0$ ) vs. UV exposure time for the 80 °C H<sub>2</sub>O nanorod spheres and Degussa P25.

The blank rhodamine B solution and the rhodamine B solution with the powder photocatalysts were tested over time with a fixed wavelength of 500 nm using a PerkinElmer lambda 950 UV/VIS/NIR spectrophotometer. The absorbance over time confirmed that the 80 °C H<sub>2</sub>O nanorod spheres show the higher absorption at the 500 nm wavelength suggesting that it can be more effective for applications in the visible/solar radiation (Figure S5).



**Figure S5.** Absorbance measurements over time at 500 nm for all the TiO<sub>2</sub> nanorod spheres.

Figure S6 shows the SEM image of the ZnO seed layer prior to microwave synthesis. The ZnO seed thickness average is  $50.7 \pm 3.6$  nm.



**Figure S6.** Cross-section SEM image of the ZnO seed layer prior to microwave synthesis.