## Supplementary Materials: Regenerable Subnanometer Pd Clusters on Zirconia for Highly Selective Hydrogenation of Biomass-Derived Succinic Acid in Water

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**Figure S1.** The aberration-corrected high-angle annual dark-filed scanning transmission electron microscopy images of 0.2 Pd/ZrO<sub>2</sub> catalyst.



Figure S2. Nitrogen adsorption and desorption isotherms of  $0.2 \text{ Pd/ZrO}_2$ .



Figure S3. The X-ray diffraction patterns of  $Pd/ZrO_2$  before and after reaction.



Figure S4. Catalytic activities of succinic acid hydrogenation over  $0.2 \text{ Pd/ZrO}_2$  catalysts (a) and different Pd loadings (b,c).



Figure S5. The thermogravimetric analysis patterns of  $0.2 \text{ Pd/ZrO}_2$  and used  $0.2 \text{ Pd/ZrO}_2$  catalysts.

Catalysts and reaction solution	Pd Loading by ICP (wt. %)	BET (m <sup>2</sup> ·g <sup>-1</sup> )
1.0 Pd/ZrO <sub>2</sub>	0.88	208
$0.2 \ Pd/ZrO_2$ a	0.22	216
$0.2 \ Pd/ZrO_2 \ ^b$	0.21	205
0.05 Pd/ZrO <sub>2</sub>	0.042	220
Reaction solution	0	-

Table S1. Physicochemical properties of Pd/ZrO<sub>2</sub> catalysts.

<sup>a</sup> fresh catalyst; <sup>b</sup> used catalyst.