

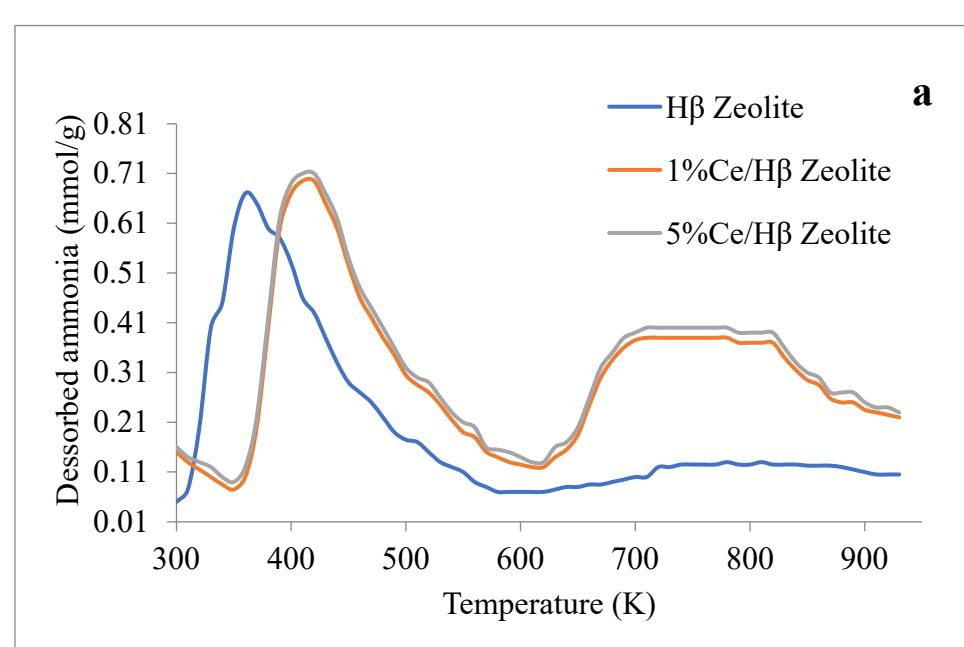
Supplementary Material

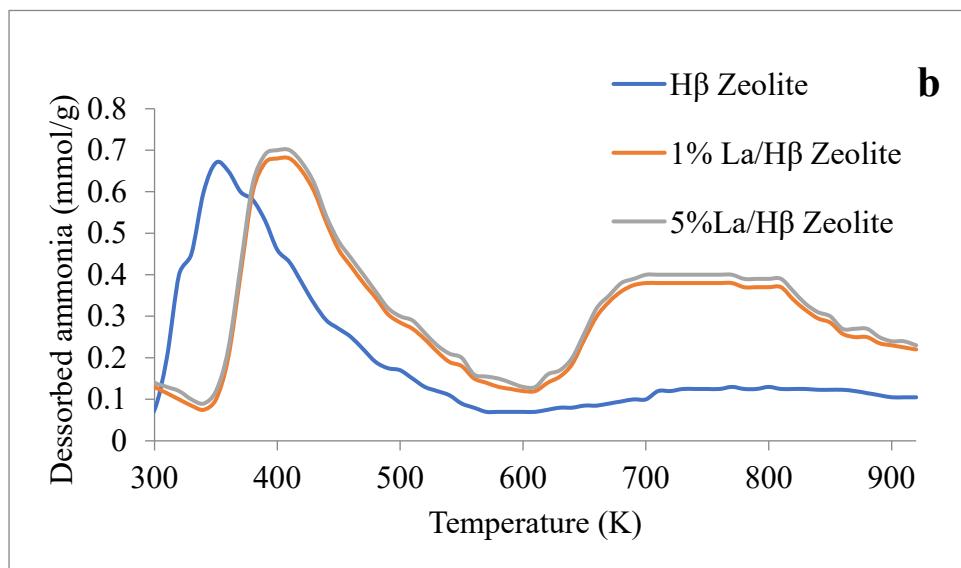
GAS-PHASE DEOXYGENATION OF BIOMASS PYROLYSIS TAR CATALYZED BY RARE EARTH METAL LOADED H β ZEOLITE

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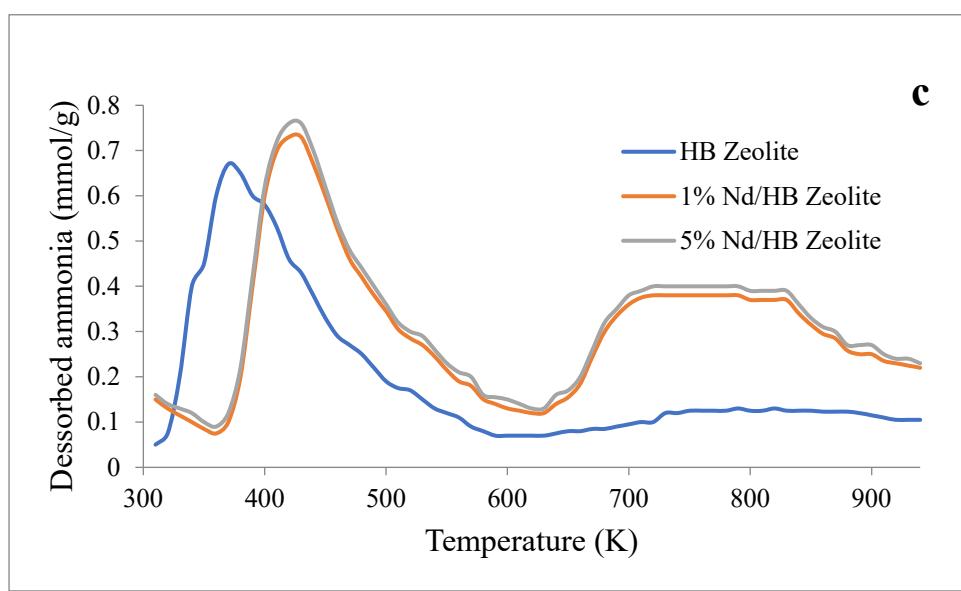
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Figure S1. Ammonia-TPD profiles for pure H β zeolite, 1wt% loaded H β zeolite, and 5wt% loaded H β zeolite before and after their regeneration at 800°C for 2h (a) Ce loaded H β zeolite, (b) La loaded H β zeolite, and (c) Nd loaded H β zeolite (All catalysts were calcined for 6h at 550°C).

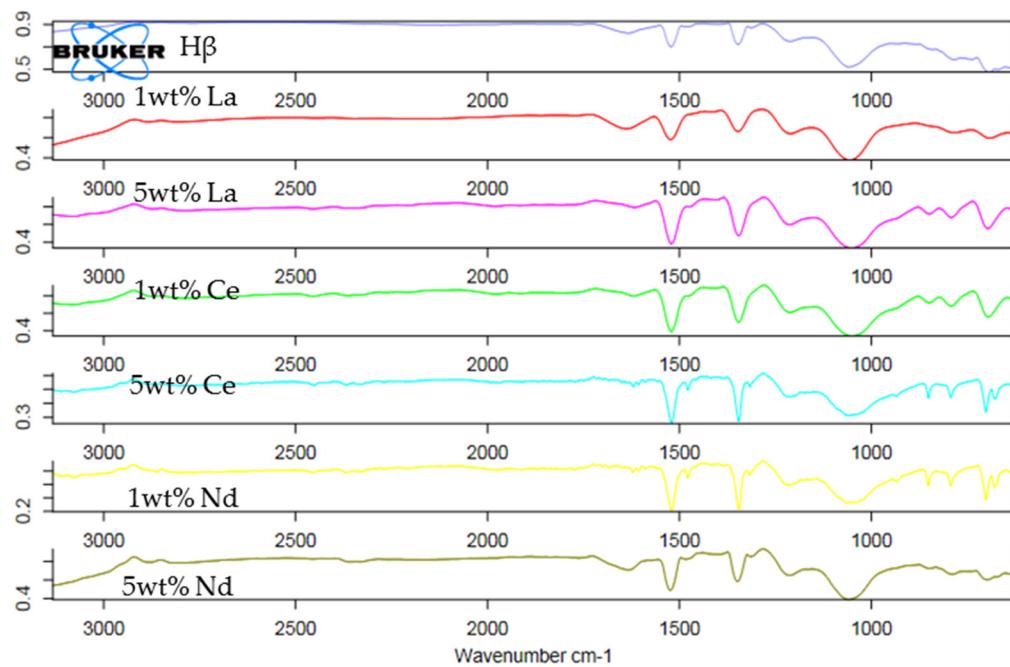


Figure S2. Pyridine-Fourier-Transform Infra-Red Spectroscopy for pure H β zeolite, 1wt% Ce/ H β zeolite, 5wt% Ce/ H β zeolite, 1wt% La/ H β zeolite, 5wt% La/ H β zeolite, 1wt% Nd/ H β zeolite, and 5wt% Nd/ H β zeolite (All catalysts were calcined for 6h at 550°C).

Table S1. GC-FID analysis of liquid phase products of the GDO reaction of acetone.

Retention time, min	Compound Class	Compound	Formula
1.53		Benzene	C ₆ H ₆
1.696		Toluene	C ₇ H ₈
2.08		O-xylene	C ₈ H ₁₀
2.11		P-xylene	C ₈ H ₁₀
2.30	Aromatics	Ethylbenzene	C ₈ H ₁₀
2.46		Mesitylene	C ₉ H ₁₂
2.49		1-Ethyl-4-methylbenzene	C ₉ H ₁₂
2.51		1-Ethyl-3-methylbenzene	C ₉ H ₁₂
2.53		1-Ethyl-2-methylbenzene	C ₉ H ₁₂

2.71		2,3-Dihydro-methyl-1H-indene	C ₁₀ H ₁₂
2.83		P-Cymene	C ₁₀ H ₁₄
2.97		2,2-Dimethylindene, dihydro-	2,3- C ₁₁ H ₁₄
1.313	Aldehydes	Acetaldehyde	C ₂ H ₄ O
1.466	Ketones	Acetone	C ₂ H ₆ O
2.19		Methyl isobutyl ketone	C ₄ H ₈ O
1.637		Methanol	CH ₃ OH
1.843	Alcohols	Ethanol	C ₂ H ₆ O
1.76		Isopropanol	C ₂ H ₈ O
1.52	Alkanes	Butane	C ₄ H ₁₀
1.58		N-Pentane	C ₅ H ₁₂
1.62	Alkenes	4-Methyl-1-pentene	C ₆ H ₁₂
2.12	Esters	5-Hexenyl acetate	C ₈ H ₁₄ O ₂
1.193	Furans	2-Methylfuran	C ₅ H ₆ O

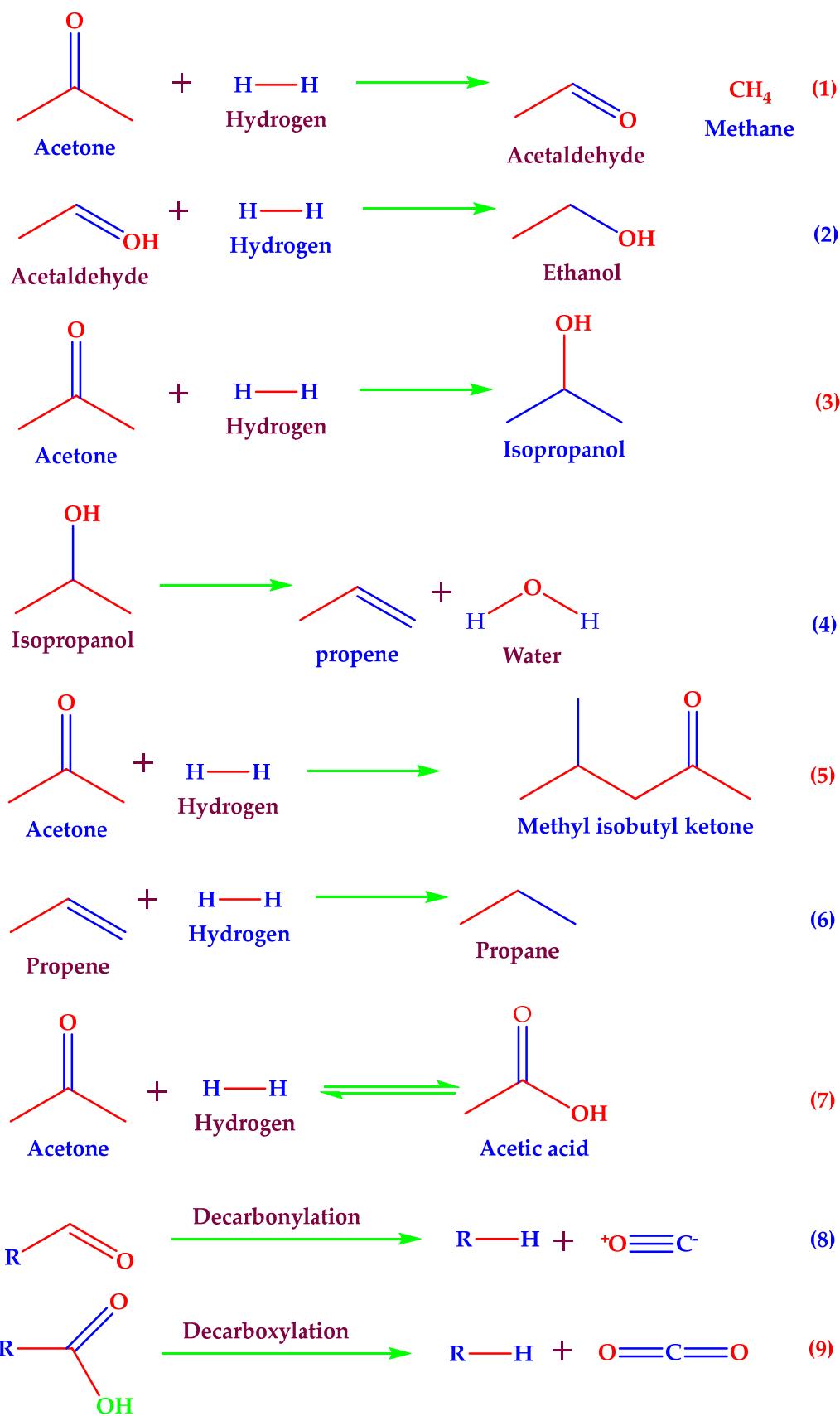


Figure S3. Reaction network for GDO of acetone model compound at 400°C using 1wt% Ce/ H β zeolite as the catalyst. 38
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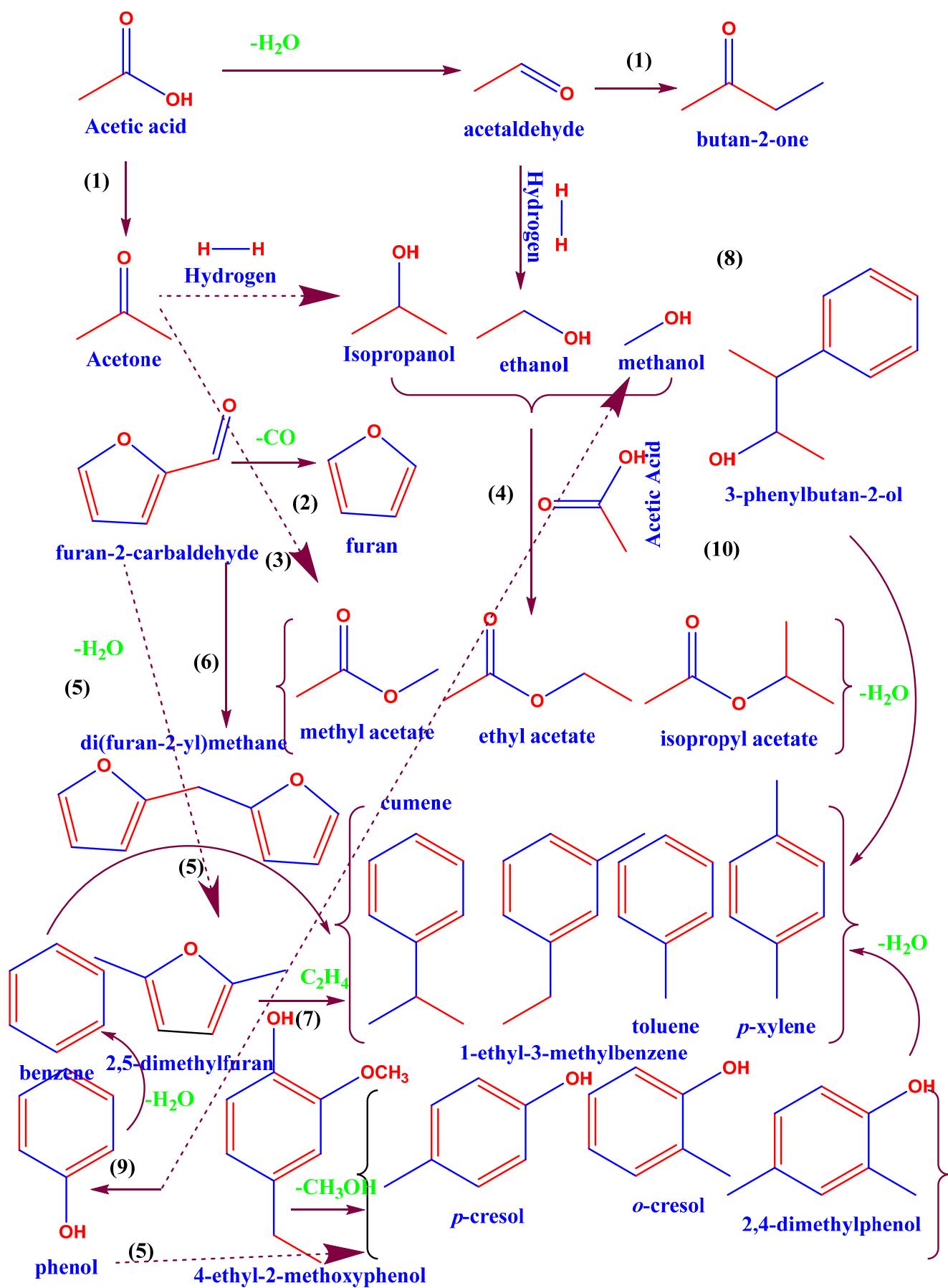


Figure S4. Main interaction pathways for biomass pyrolysis of tar GDO using 1wt% Ce/ H β zeolite as the catalyst at 400°C and process time of 3h (Decarbonylation and decarboxylation steps are not shown).

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