

Supplementary Materials

Synthesis of Cobalt-Doped Iron Phosphate Crystal on Stainless Steel Mesh for Corrosion Resisted Oxygen Evolution Catalyst

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Figure S1. Picture of SSM that corroded away after the hydrothermal reaction in 4.14 wt % H_3PO_4 solution, except for the part inserted into a homemade Teflon holder.

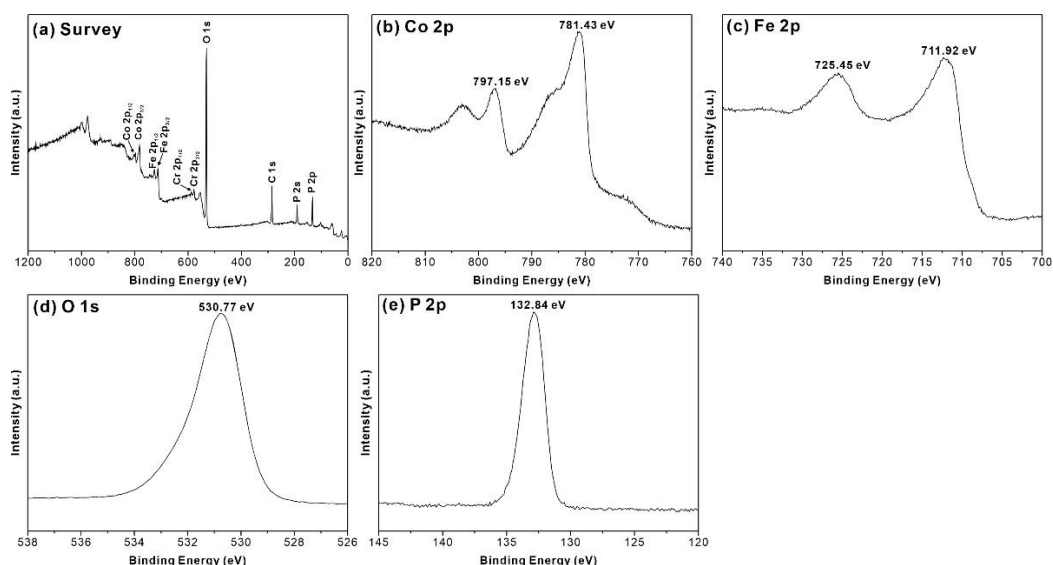


Figure S2. (a) XPS survey spectrum for iron phosphate oxide hydroxide crystal. (b-f) XPS spectra of iron phosphate oxide hydroxide crystal in the (b) Co 2p, (c) Fe 2p, (d) O 1s, and (e) P 2p.

Table S1. Comparison of OER activity of 0.84-CoFePi with recently reported Co and Fe based catalysts.

Catalysts	substrate	Electrolyte	Overpotential (mV) @ 10 mAcm ⁻²	Reference
0.84-CoFePi	Stainless steel mesh	1 M KOH	300.3	This work
Amorphous α -phase NiCo hydroxide	Stainless steel foil (planar)	1 M KOH	257	[1]
Amorphous α -phase Ni(OH) ₂			273	
Amorphous α -phase Co(OH) ₂			305	
Co _x O _y /NC	Nitrogen-doped carbon	0.1 M KOH	430	[2]
PE-Co ₃ O ₄ NS/Ti	Ti foil	0.1 M KOH	300	[3]
Fe adsorbed CoO _x	Glassy-carbon-disk	1 M KOH	309	[4]
Mn-Co LDH/Graphene	Graphene	1 M KOH	330	[5]
CoFeP@C	N, P dual-doped carbon matrix	1 M KOH	336	[6]
CoNi-CoO@NC/CP	Carbon paper	0.1 M KOH	392	[7]

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