Title:Guerbet Reactions for Biofuel Production from ABE fermentation Using Bifunctional Ni-MgO-Al₂O₃ Catalysts

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Supporting Information



Fig. S1. (a) and (b): SEM images of Ni-MgO-Al₂O₃ catalyst (Mg/Al=3).



Fig. S2. XRD patterns of Ni–MgO–Al₂O₃ (Mg/Al=3, Ni content = 6 wt.%) catalysts before and after reactions.



Fig. S3. (a) Pore size distributions and (b) N2 adsorption/desorption isotherms of fresh and spent

catalysts.

reactions.

Sample	Surface area (m ² /g)	Pore volume (cm ³ /g)	Mean pore size (nm)
Fresh	237.5	0.68	5.7
Cycle ₁	234.4	0.66	5.6
Cycle ₂	229.4	0.65	5.6
Cycle ₃	222.5	0.64	5.5

Table S1. BET results of Ni-MgO-Al₂O₃ catalysts (Ni content = 6 wt.%) before and after



Fig. S4. (a) TEM image of spent catalyst Ni–MgO–Al₂O₃ (Mg/Al=3) catalysts. (b) Histogram of particle size distribution of Ni nanoparticles.



Figure S5. CO₂-TPD profiles of the fresh, used Ni–MgO–Al₂O₃ catalysts.



Figure S6. Results of the catalytic coupling of ABE mixture. Reaction conditions: 1.5g cat: Ni–MgO–Al₂O₃, 240°C for 20 hours. Values determined by FID, using internal standard.