

Supplementary Materials

Sustainable Hydrogenation of Vinyl Derivatives Using Pd/C Catalysts

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Figure S1. Multi-reactor setup used for liquid-phase hydrogenation of vinyl derivatives under hydrogen pressure.

TEM images of Pd/C catalysts.

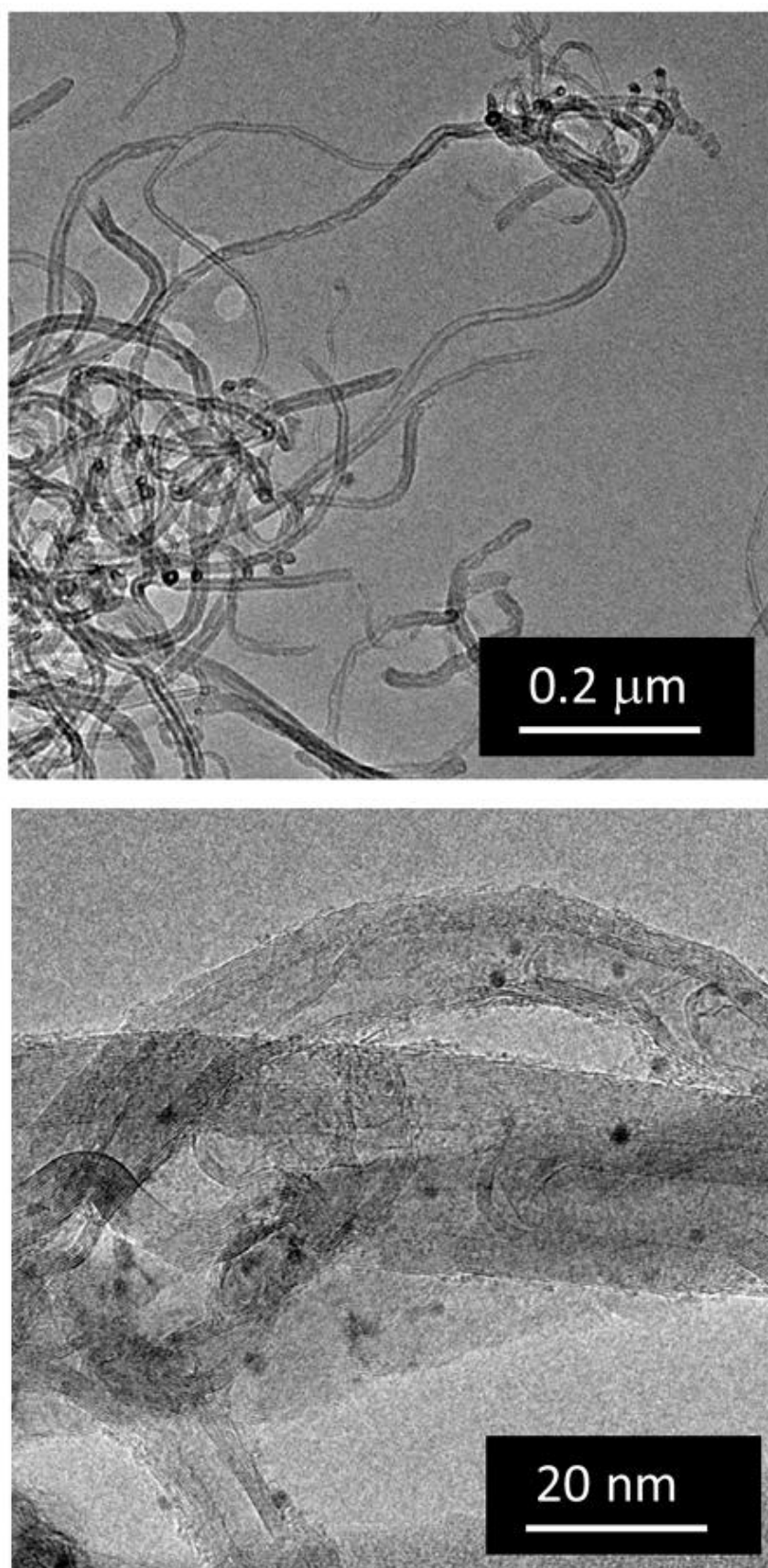


Figure S2. TEM images of 2% Pd-CCs/MWCNTs catalyst. See also Fig. 1a.

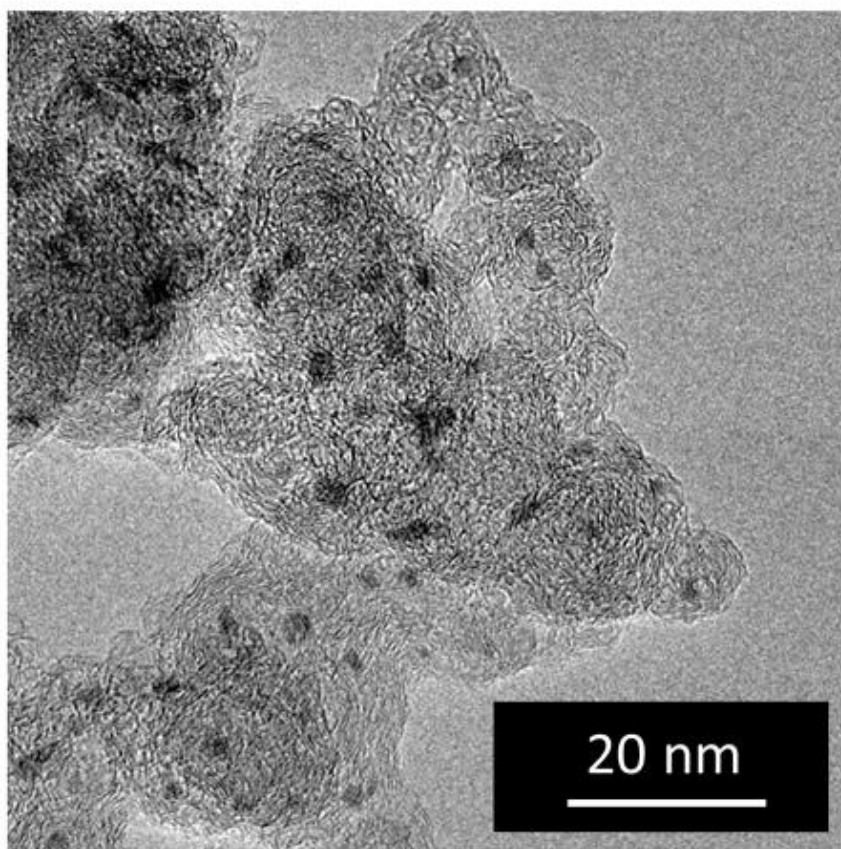
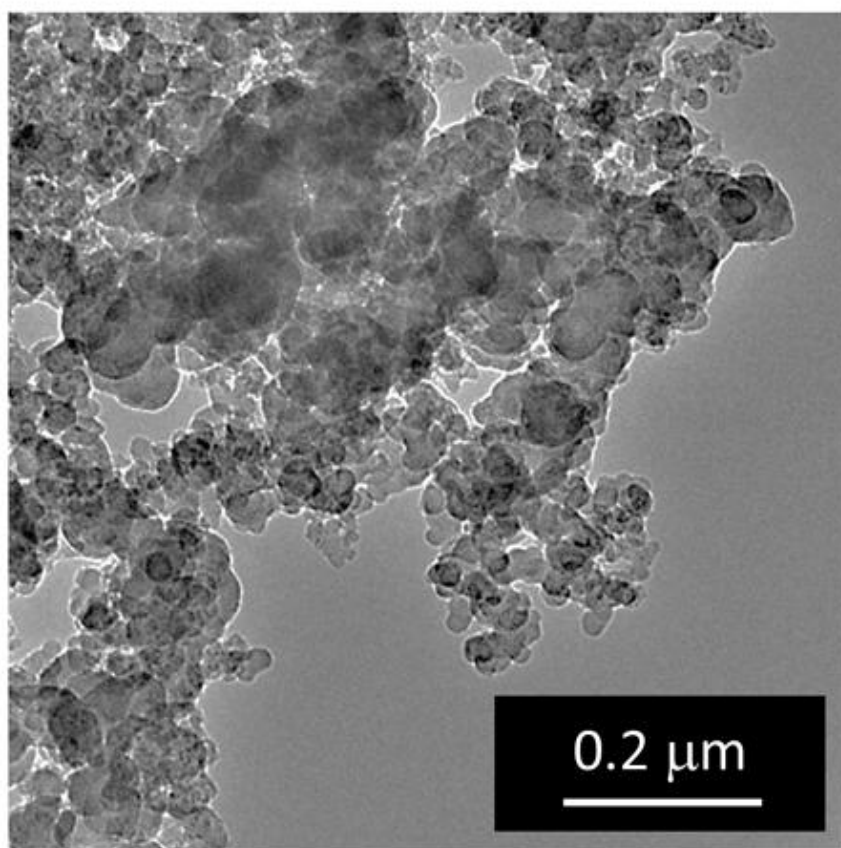


Figure S3. TEM images of 2% Pd-CCs/P278-E catalyst. See also Fig. 1b.

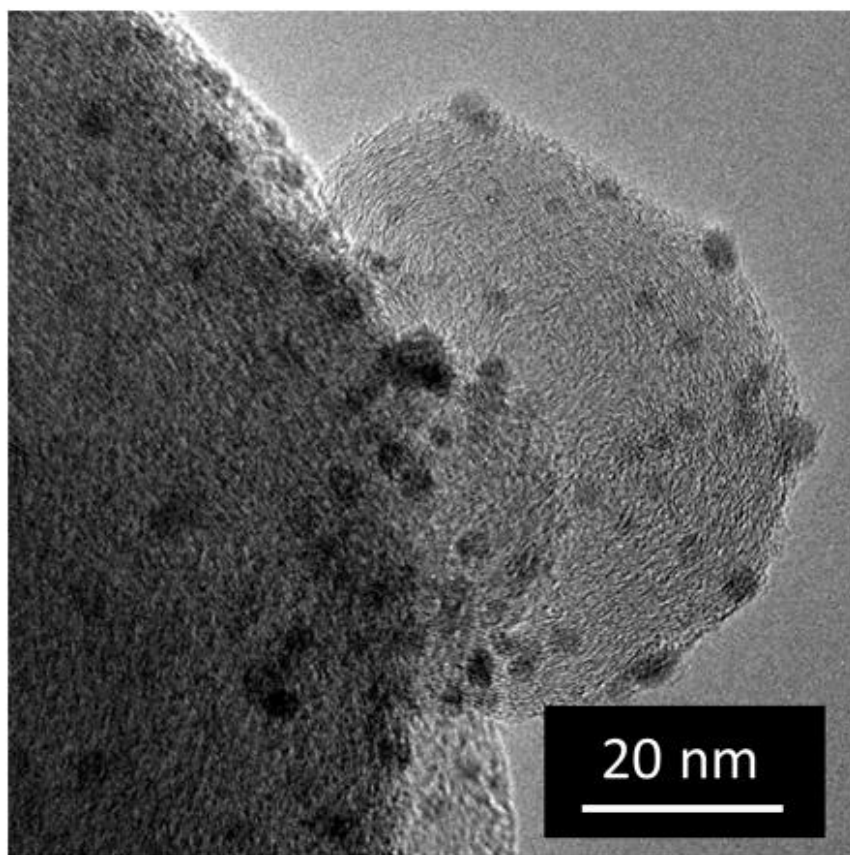
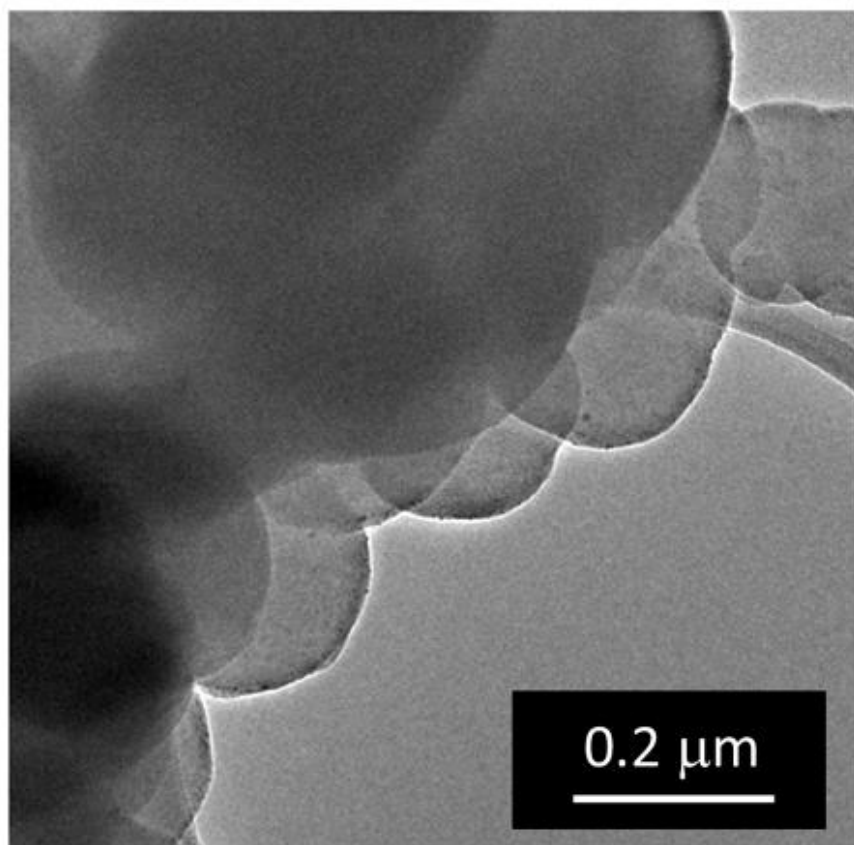


Figure S4. TEM images of 2% Pd-CCs/T900 catalyst. See also Fig. 1c.

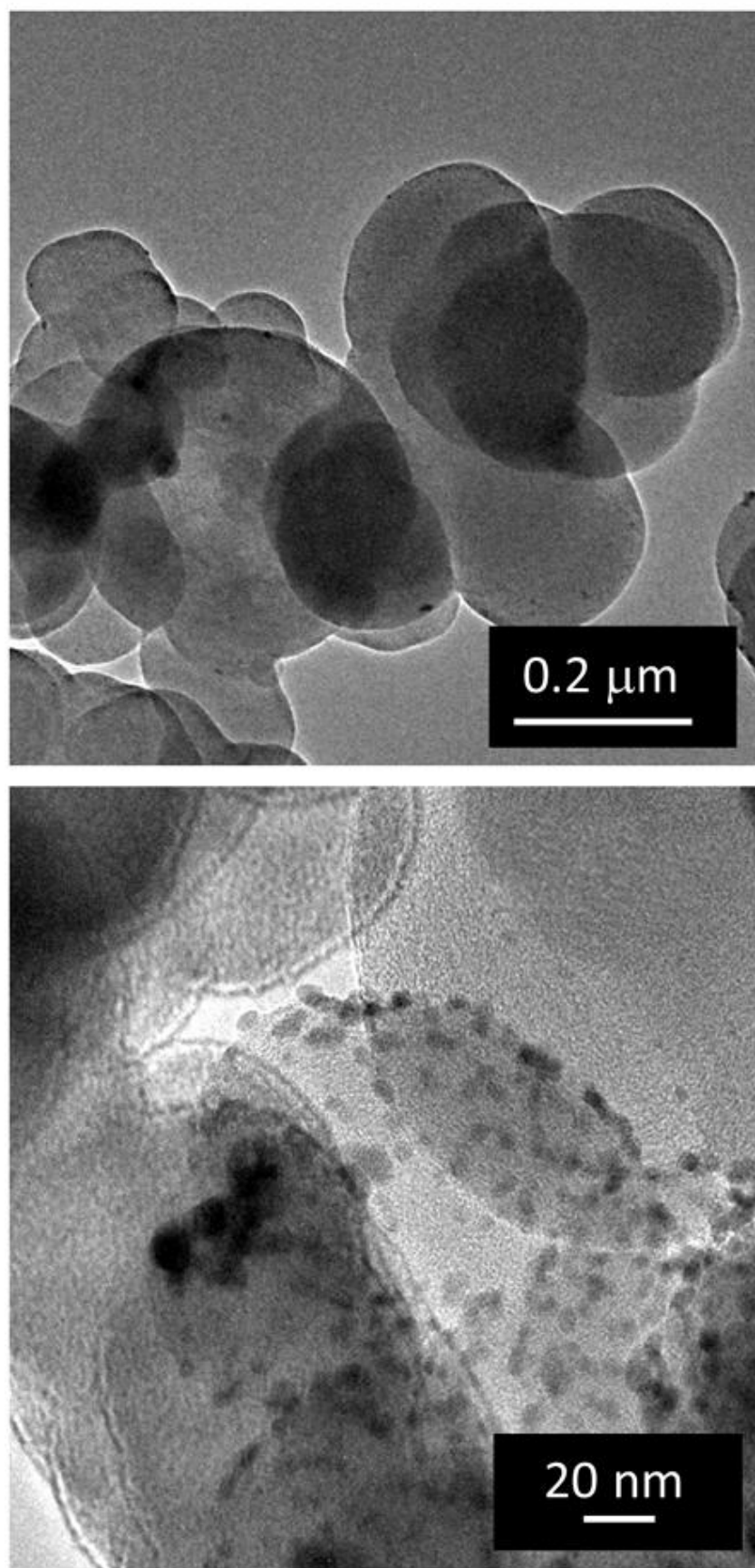


Figure S5. TEM images of 2% Pd-PHCs/T900 catalyst. See also Fig. 1d.

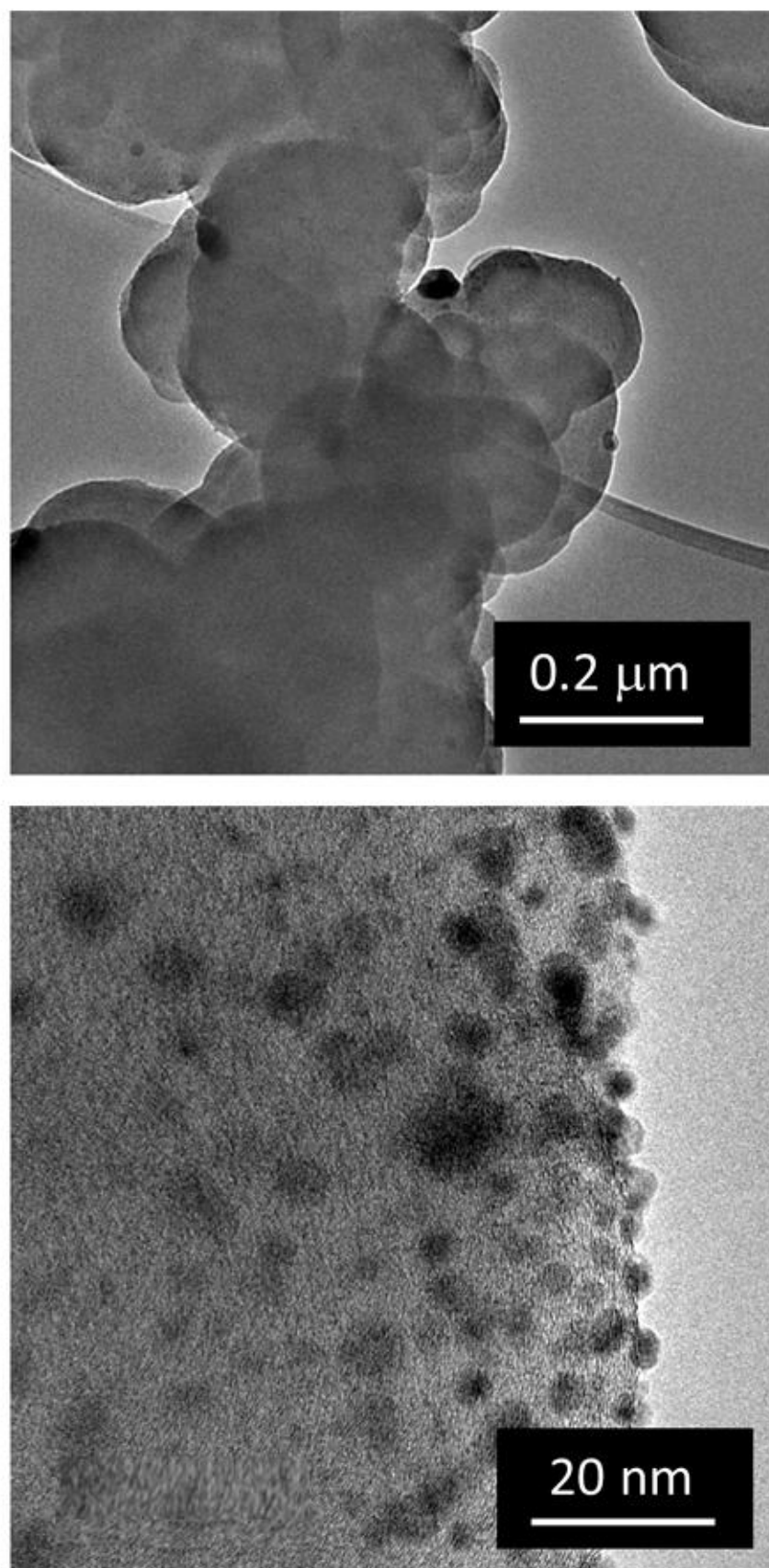


Figure S6. TEM images of 2% Pd-Ac/T900 catalyst. See also Fig. 1e.

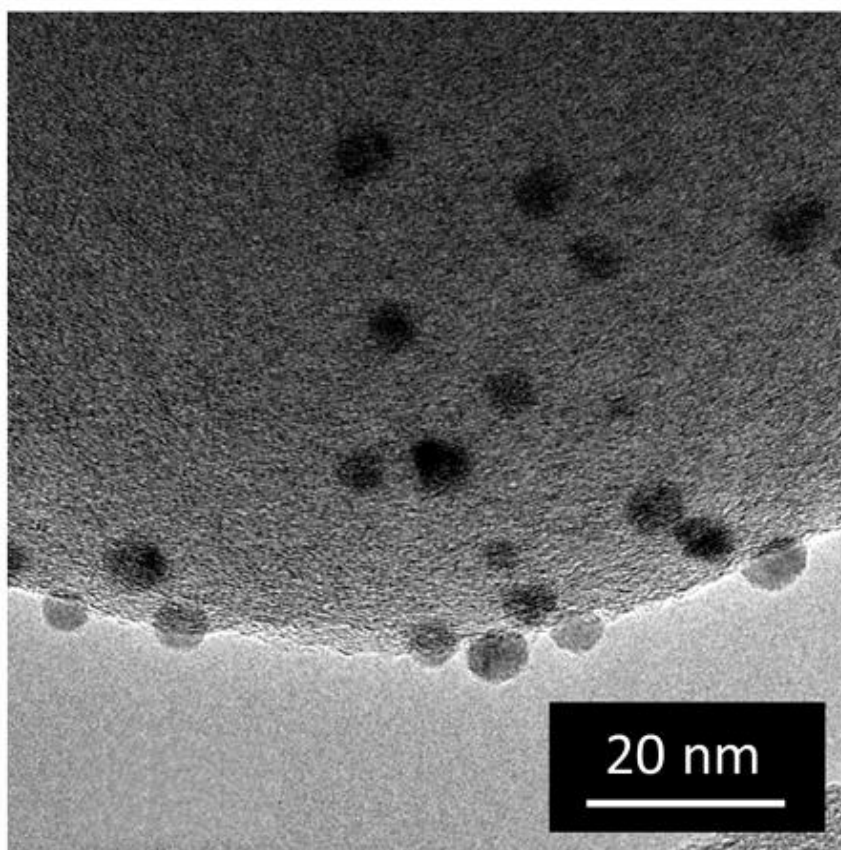
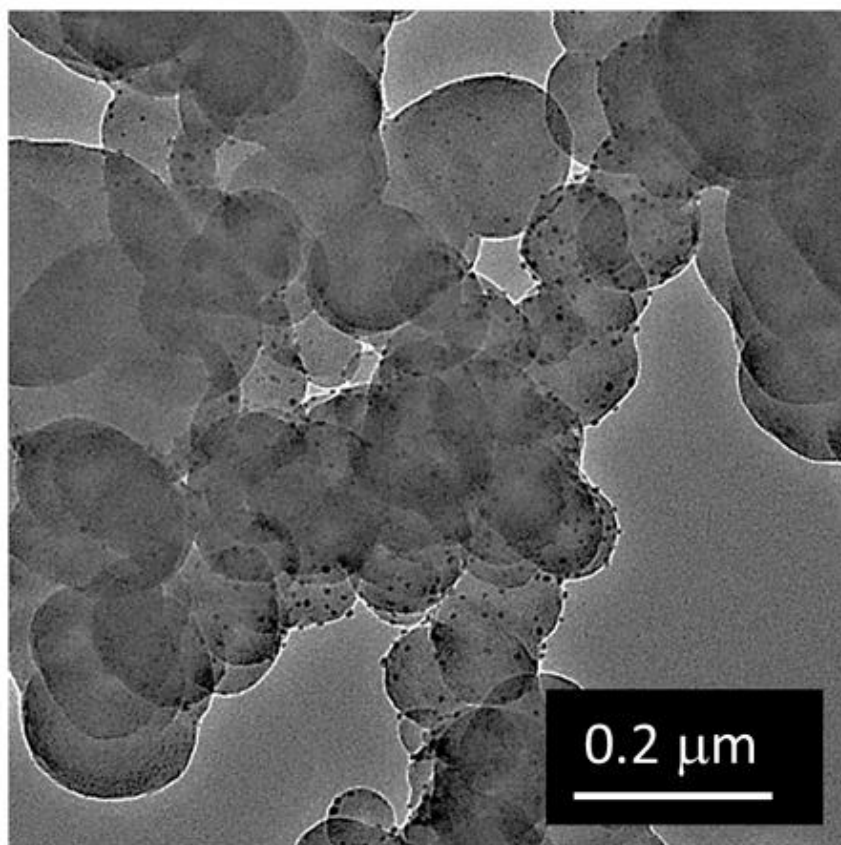


Figure S7. TEM images of 2% Pd-dba/T900 catalyst. See also Fig. 1f.

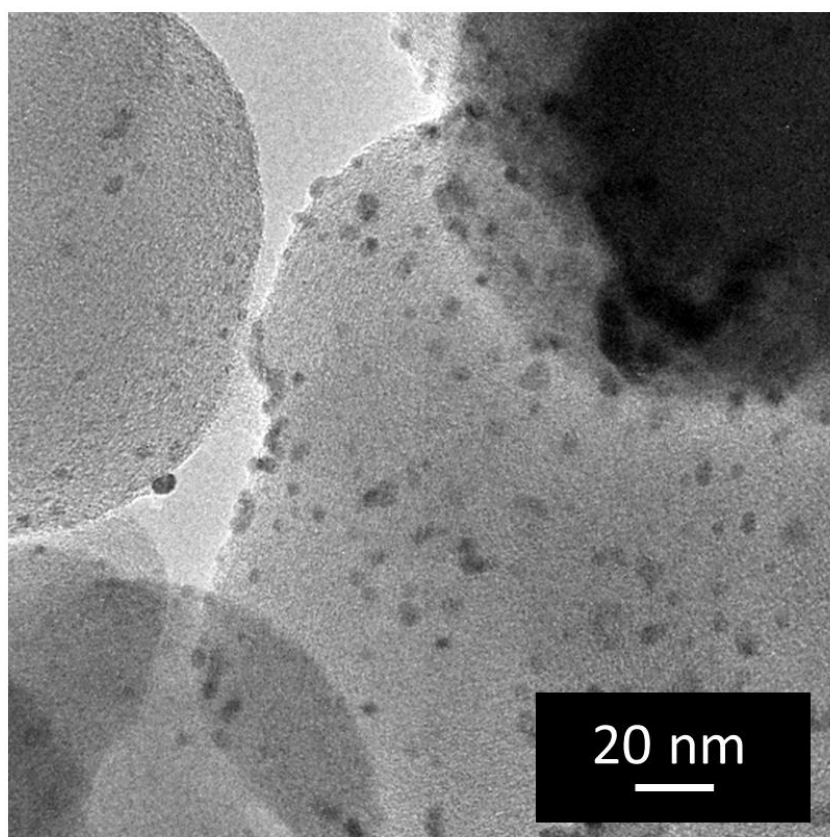
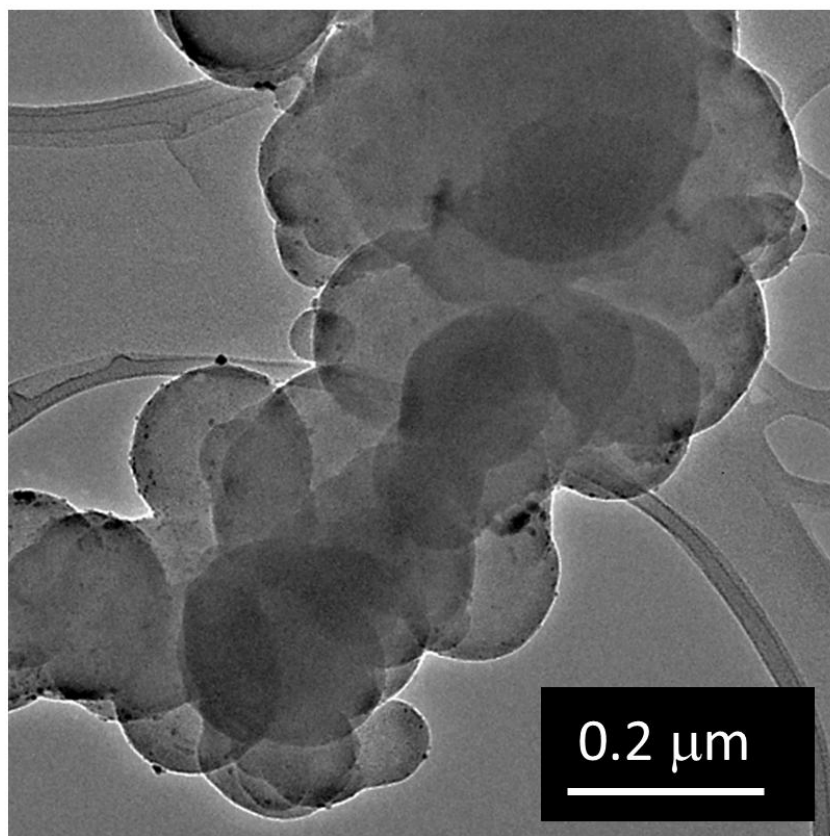
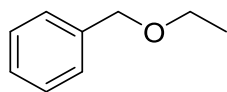


Figure S8. TEM images of the spent 2% Pd-PHCs/T900 catalyst after five cycles of the NVE hydrogenation.

Identification of the hydrogenation products.

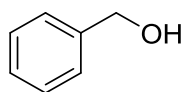
Benzyl ethyl ether (BEE).



^1H NMR (400 MHz, MeOH), δ_{H} /ppm: 6.95-6.85 (m, 5H), 4.09 (s, 2H), 3.15 (q, $J = 7.0$ Hz, 2H), 0.82 (t, $J = 7.0$ Hz, 3H). ^{13}C NMR (101 MHz, MeOH), δ_{C} /ppm: 138.6, 128.4, 127.9, 127.7, 72.7, 65.8, 14.4. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 136 (7) M^+ , 135 (11), 107 (17), 105 (5), 92 (84), 91 (100), 89 (5), 79 (34), 77 (19), 65 (16), 51 (9).

The spectral data obtained are in agreement with those previously reported [1-3].

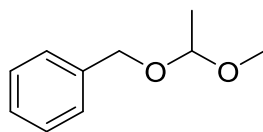
Benzyl alcohol (BAL).



^1H NMR (400 MHz, MeOH), δ_{H} /ppm: 6.82-6.96 (m, 5H), 4.20 (s, 2H). ^{13}C NMR (101 MHz, MeOH), δ_{C} /ppm: 141.7, 128.3, 127.3, 127.0, 64.4. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 108 (84) M^+ , 107 (59), 106 (7), 105 (10), 92 (23), 91 (42), 90 (9), 89 (10), 80 (10), 79 (100), 78 (14), 77 (70), 65 (11), 63 (8), 53 (7), 52 (8), 51 (24), 50 (11).

The spectral data obtained are in agreement with those previously reported [3, 4].

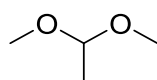
1-Benzyl-1-methoxyethane.



^{13}C NMR (101 MHz, MeOH), δ_{C} /ppm: 128.4, 127.6, 100.7, 67.8, 52.0, 18.7. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 134 (29), 108 (12), 107 (14), 106 (6), 105 (12), 92 (21), 91 (100), 89 (6), 79 (17), 78 (5), 77 (19), 65 (15), 63 (5), 59 (36), 51 (12), 44 (20), 43 (7), 40 (65).

The ^{13}C NMR data are in agreement with those previously reported [5].

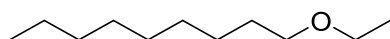
1,1-Dimethoxyethane (DME).



^1H NMR (400 MHz, MeOH), δ_{H} /ppm: 4.14 (q, $J = 5.3$ Hz, 1H), 2.90 (s, 6H), 0.84 (d, $J = 6.3$ Hz, 3H). ^{13}C NMR (101 MHz, MeOH), δ_{C} /ppm: 101.8, 52.1, 18.2. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 90 (0.1) M^+ , 75 (47), 59 (100), 57 (7), 47 (10), 43 (23), 41 (6).

The spectral data obtained are in agreement with those previously reported [3, 4].

Nonyl ethyl ether (NEE).

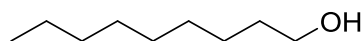


^1H NMR (400 MHz, MeOH), δ_{H} /ppm: 3.22 (q, $J = 7.0$ Hz, 2H), 3.16 (t, $J = 6.7$ Hz, 2H), 1.29 (quin, $J = 7.0$ Hz, 2H), 1.03 (m, 12H); 0.91 (t, $J = 7.0$ Hz, 3H), 0.63 (t, $J = 6.7$ Hz, 3H). ^{13}C NMR (101 MHz, MeOH), δ_{C} /ppm: 70.8, 66.1, 32.0, 29.8, 29.7, 29.6, 29.4, 26.2, 22.7,

14.5, 13.5. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 126 (18), 98 (27), 97 (23), 85 (9), 84 (18), 83 (28), 82 (13), 71 (15), 70 (40), 69 (33), 68 (17), 59 (100), 57 (23), 56 (42), 55 (33), 47 (19), 43 (32), 42 (13), 41 (42), 39 (8).

The MS data are in agreement with those previously reported [6].

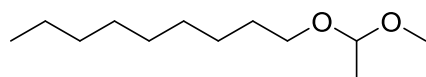
Nonyl alcohol (NAL).



^1H NMR (400 MHz, MeOH), $\delta_{\text{H}}/\text{ppm}$: 1.19 (quin, $J = 6.9$ Hz, 2H), 0.97 (m, 12H), 0.57 (t, $J = 7.0$ Hz, 3H). ^{13}C NMR (101 MHz, MeOH), $\delta_{\text{C}}/\text{ppm}$: 62.2, 32.7, 32.1, 29.7, 29.6, 29.4, 25.9, 22.7, 13.4. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 98 (30), 97 (33), 84 (26), 83 (41), 82 (16), 71 (14), 70 (79), 69 (68), 68 (26), 67 (12), 57 (37), 56 (100), 55 (93), 54 (9), 53 (6), 44 (8), 43 (69), 42 (36), 41 (80), 40 (12), 39 (19).

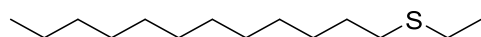
The spectral data obtained are in agreement with those previously reported [3, 4].

1-Nonyloxy-1-methoxyethane.



^{13}C NMR (101 MHz, MeOH), $\delta_{\text{C}}/\text{ppm}$: 101.1, 66.0, 52.0, 29.9, 29.6, 26.3, 18.7. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 187 (5), 171 (4), 85 (14), 71 (24), 70 (7), 69 (8), 61 (8), 59 (100), 57 (23), 56 (11), 55 (15), 43 (33), 42 (7), 41 (20).

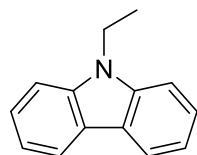
Dodecyl ethyl sulfide (DES).



^1H NMR (400 MHz, MeOH), $\delta_{\text{H}}/\text{ppm}$: 2.37 (t, $J = 6.9$ Hz, 2H), 2.18 (q, $J = 7.0$ Hz, 2H), 1.24 (quin, $J = 7.2$ Hz, 2H), 1.12-0.85 (m, 18H), 0.90 (t, $J = 7.4$ Hz, 3H), 0.58 (t, $J = 6.9$ Hz, 3H). ^{13}C NMR (101 MHz, MeOH), $\delta_{\text{C}}/\text{ppm}$: 32.1, 31.5, 29.7, 29.7, 29.7, 29.5, 29.4, 28.9, 28.8, 25.6, 22.7, 14.2, 13.4. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 230 (26) M^+ , 201 (100), 97 (11), 83 (14), 75 (25), 69 (18), 62 (14), 55 (22), 43 (16), 41 (19).

The MS data are in agreement with those previously reported [3].

9-Ethylcarbazole (9-EC).



^1H NMR (400 MHz, C_6H_6), $\delta_{\text{H}}/\text{ppm}$: 8.05 (d, $J = 7.6$ Hz, 2H), 3.71 (q, $J = 7.1$ Hz, 2H), 0.91 (t, $J = 7.2$ Hz, 3H). ^{13}C NMR (101 MHz, C_6H_6), $\delta_{\text{C}}/\text{ppm}$: 139.8, 125.3, 123.0, 120.3, 118.6, 108.1, 36.7, 12.9. MS (EI, 70 eV), m/z ($I_{\text{rel}}/\%$): 195 (49) M^+ , 181 (15), 180 (100), 167 (8), 166 (10), 152 (17), 140 (8), 139 (6).

The spectral data obtained are in agreement with those previously reported [3, 4].

References

1. Joshi, G.; Adimurthy, S. New Method for the Synthesis of Benzyl Alkyl Ethers Mediated by FeSO₄. *Synth. Commun.* **2011**, *41*, 720-728.
2. Quintard, J.-P.; Elissondo, B.; Mouko-Mpegna, D. Quelques aspects de la reactivite de l' α -chloro α -ethoxymethyltributyletain: etherification reductrice des aldehydes aromatiques et mise en evidence de formyltributyletain lors de la reaction d'hydrolyse. *J. Organomet. Chem.* **1983**, *251*, 175-187.
3. NIST Mass Spectrometry Data Center. URL <https://webbook.nist.gov>.
4. AIST: Spectral Database for Organic Compounds. URL <http://sdbb.db.aist.go.jp>.
5. Senaweera, S.; Cartwright, K.C.; Tunge, J.A. Decarboxylative Acetoxylation of Aliphatic Carboxylic Acids. *J. Org. Chem.* **2019**, *84*, 12553-12561.
6. Luska, K.L.; Julis, J.; Stavitski, E.; Zakharov, D.N.; Adams, A.; Leitner, W. Bifunctional nanoparticle-SILP catalysts (NPs@SILP) for the selective deoxygenation of biomass substrates. *Chem. Sci.* **2014**, *5*, 4895-4905.