

Supplementary Materials

Direct Etherification Reaction of Glycerol Using Alkali Metal Cation (Li^+ , Na^+ and K^+) Containing X-Type Zeolites as Heterogeneous Catalysts: Optimization of the Reaction Conditions

Seo Kyung Park, Dae Won Kim, Su Yeon Lee and Je Seung Lee *

Department of Chemistry, Kyung Hee University, 26 Kyungheedaero, Dongdaemun-gu, Seoul 02447, Korea; tjrud1015@naver.com (S.K.P.); david.pinkim@khu.ac.kr (D.W.K.); tndus4631@naver.com (S.Y.L.)

* Correspondence: leejs70@khu.ac.kr; Tel.: +82-2-961-0458

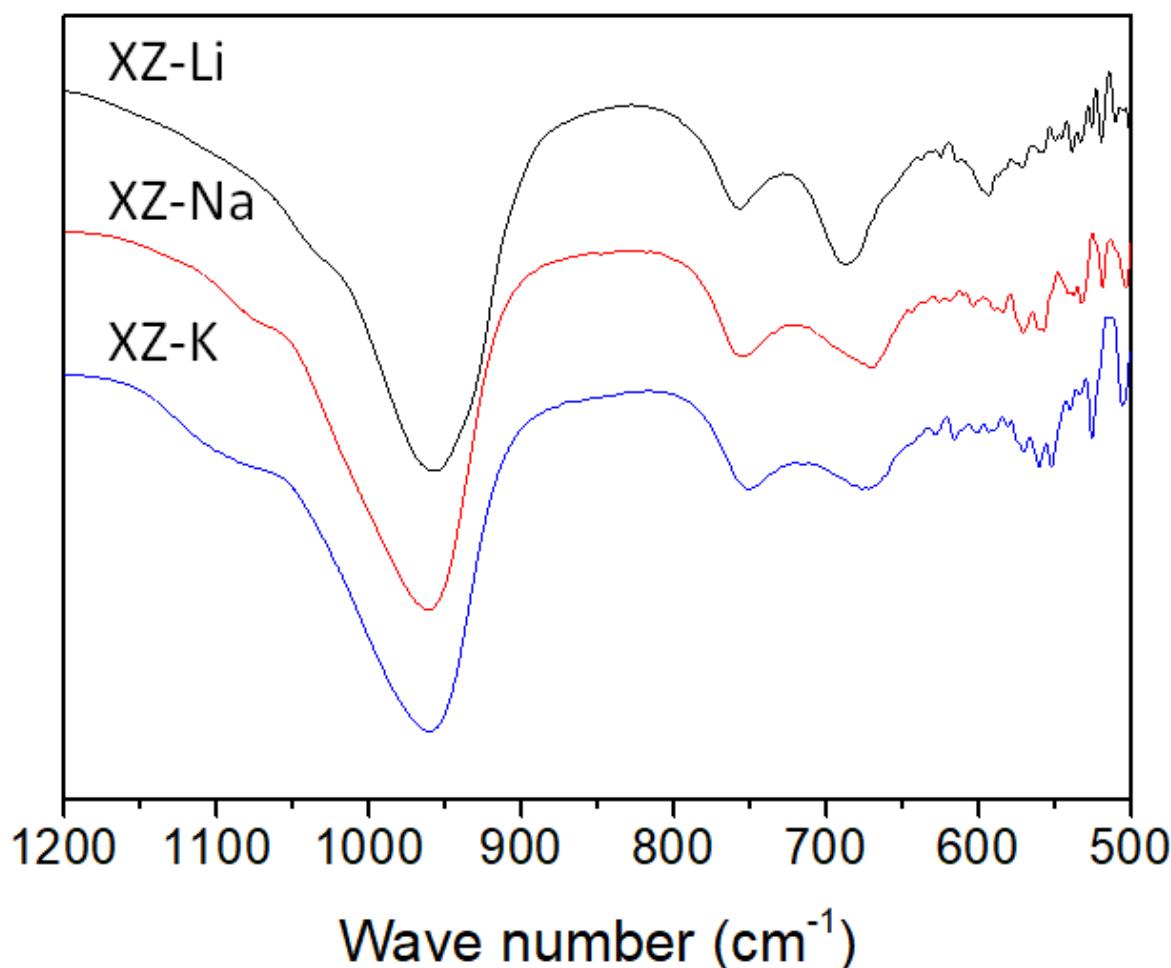


Figure S1. FT-IR spectra of zeolite XZ-M (M = Li, Na, and K).

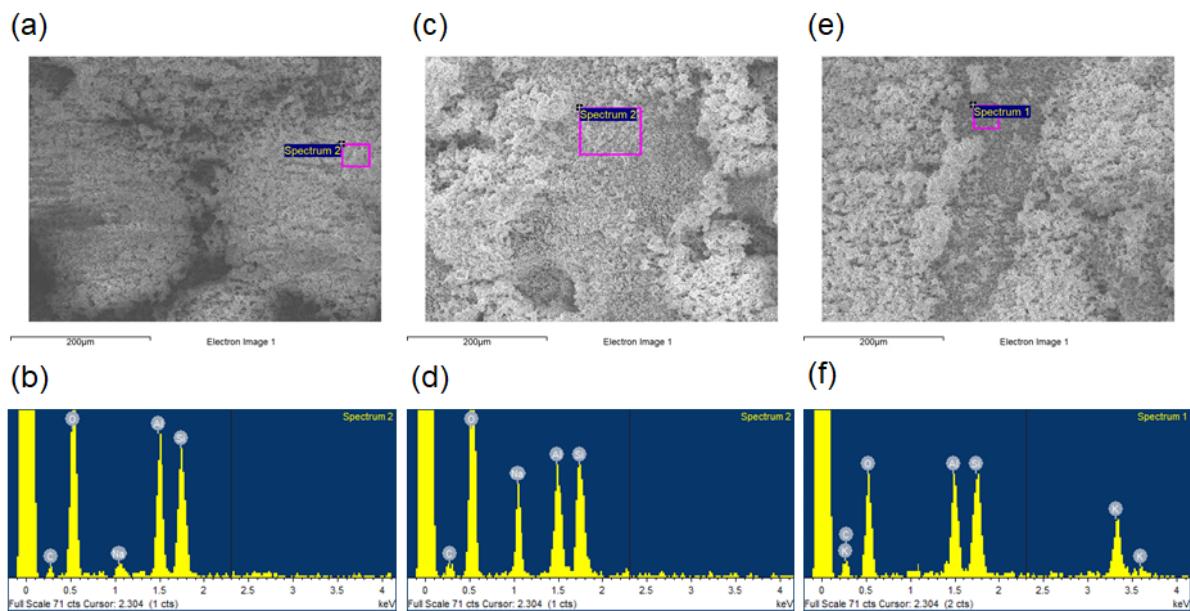


Figure S2. SEM images and EDX data of zeolite (a) and (b) XZ-Li, (c) and (d) XZ-Na, (e) and (f) XZ-K, respectively.

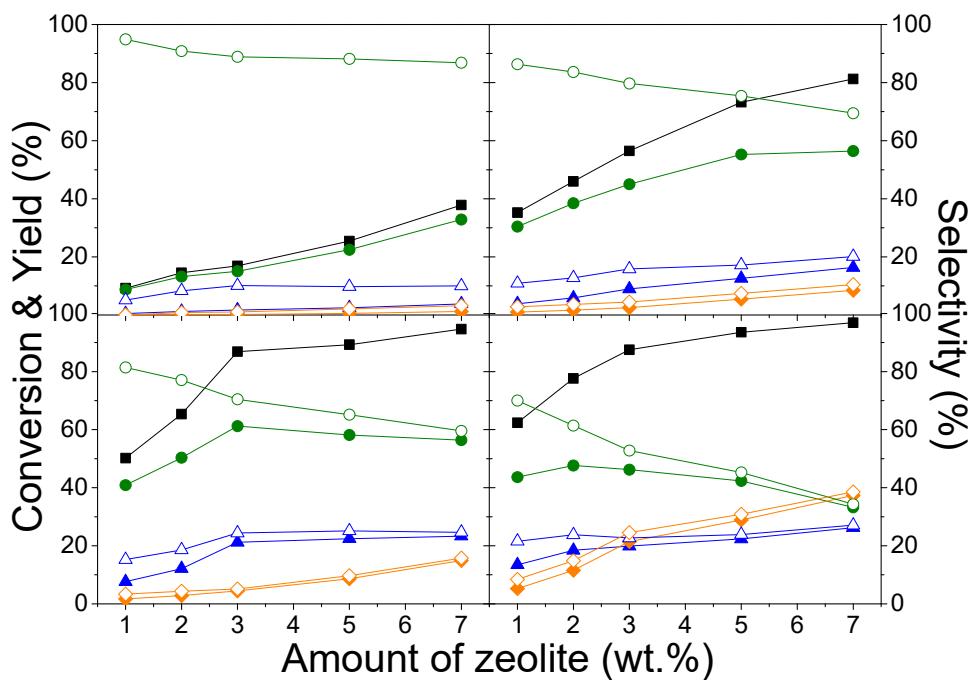


Figure S3. The influence of the amount of XZ-Li on the etherification reaction of glycerol at (a) 260, (b) 270, (c) 280, and (d) 290 °C (-■-: Conversion of glycerol, -●-: Yield of DG, -▲-: Yield of TG, -◆-: Yield of others -○-: Selectivity of DG, -△-: Selectivity of TG, -◇-: Selectivity of others, reaction time: 2 h).

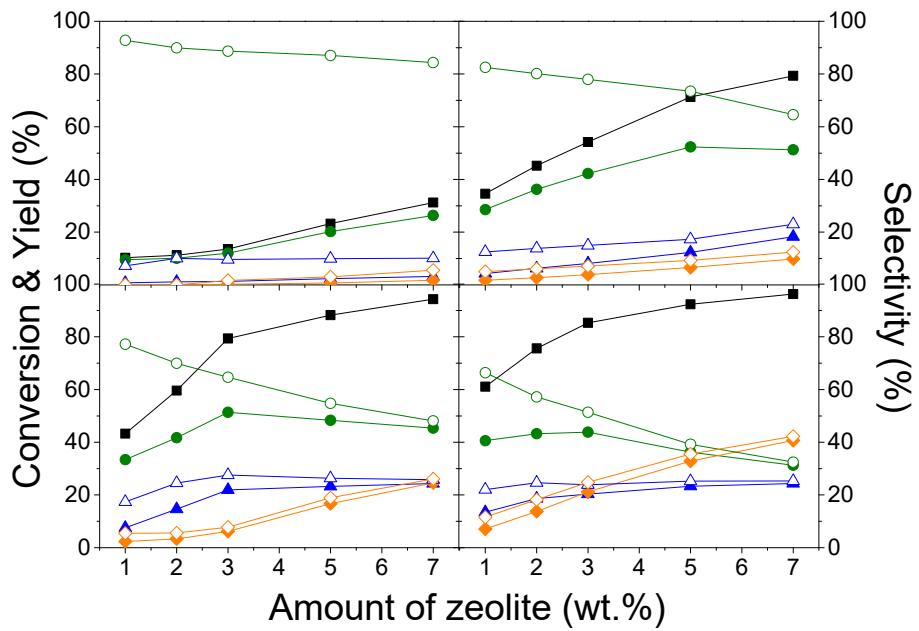


Figure S4. The influence of the amount of XZ-Na on the etherification reaction of glycerol at (a) 260, (b) 270, (c) 280, and (d) 290 °C (-■-: Conversion of glycerol, -●-: Yield of DG, -▲-: Yield of TG, -◆-: Yield of others -○-: Selectivity of DG, -Δ-: Selectivity of TG, -◇-: Selectivity of others, reaction time: 2 h).

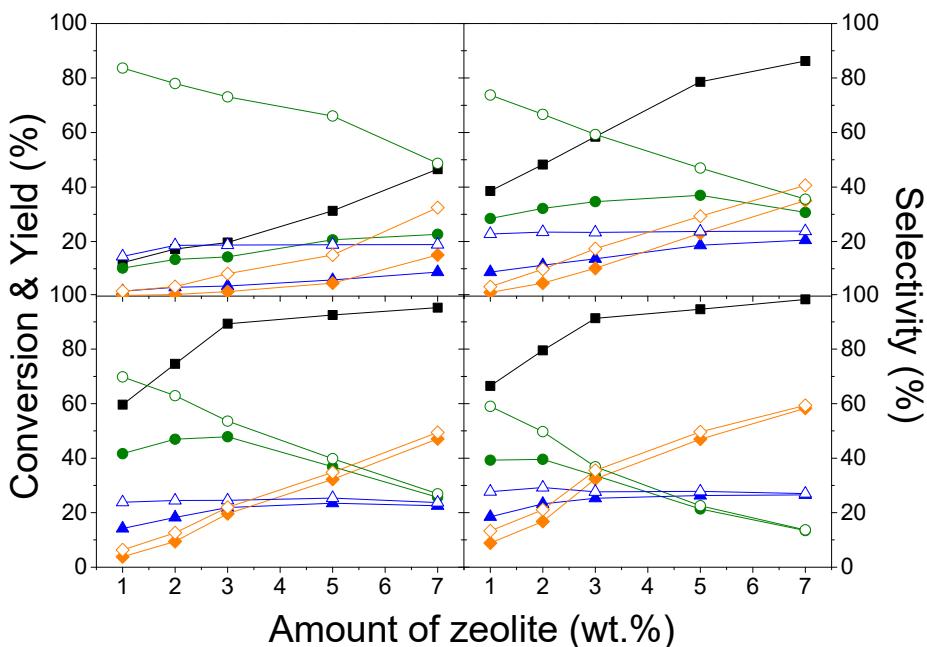


Figure S5. The influence of the amount of XZ-K on the etherification reaction of glycerol at (a) 260, (b) 270, (c) 280, and (d) 290 °C (-■-: Conversion of glycerol, -●-: Yield of DG, -▲-: Yield of TG, -◆-: Yield of others -○-: Selectivity of DG, -Δ-: Selectivity of TG, -◇-: Selectivity of others, reaction time: 2 h).

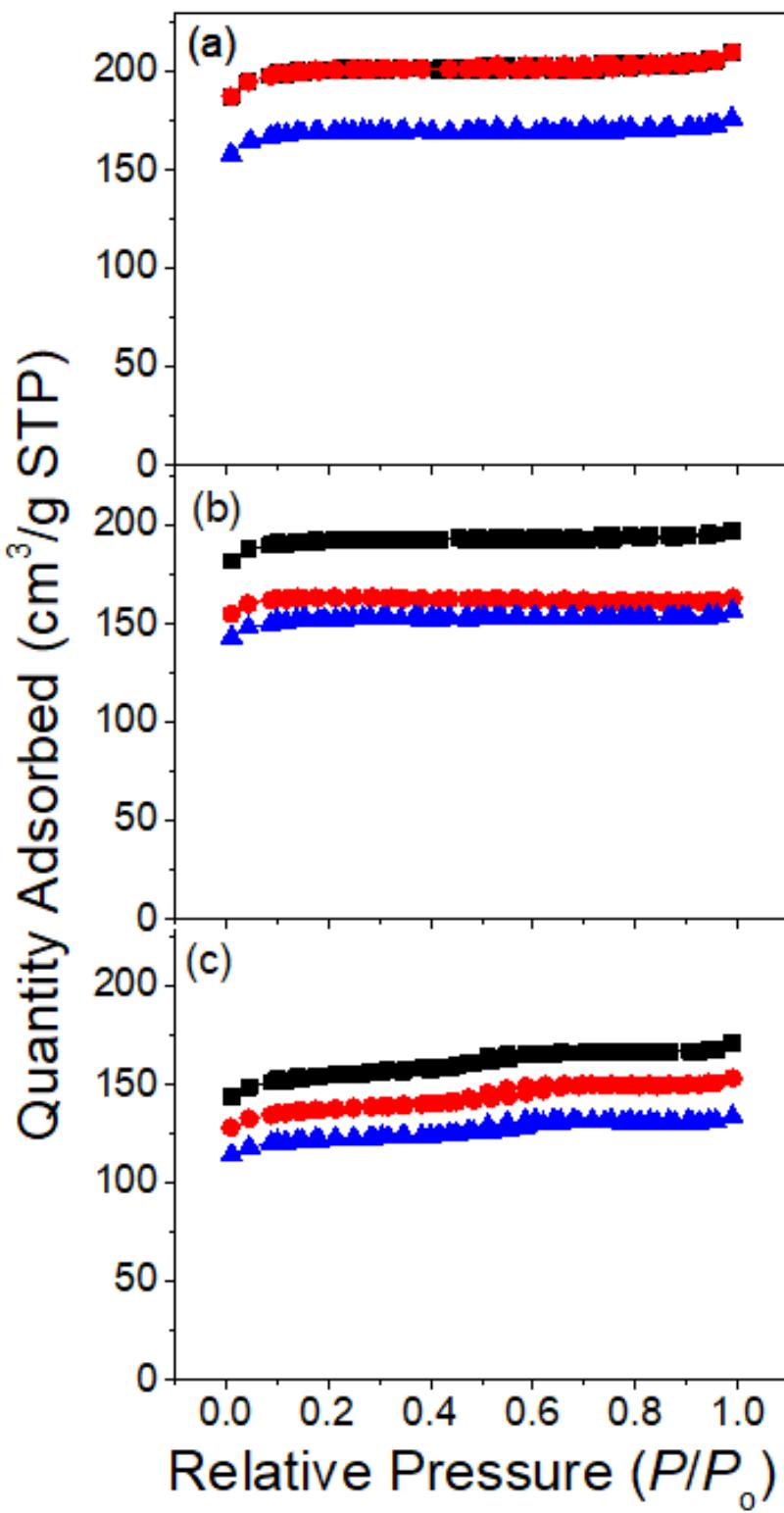


Figure S6. N₂ sorption isotherms of (a) XZ-Li, (b) XZ-Na, and (c) XZ-K (black: Pristine zeolite, red: After the first reaction, blue: After the fifth reaction).

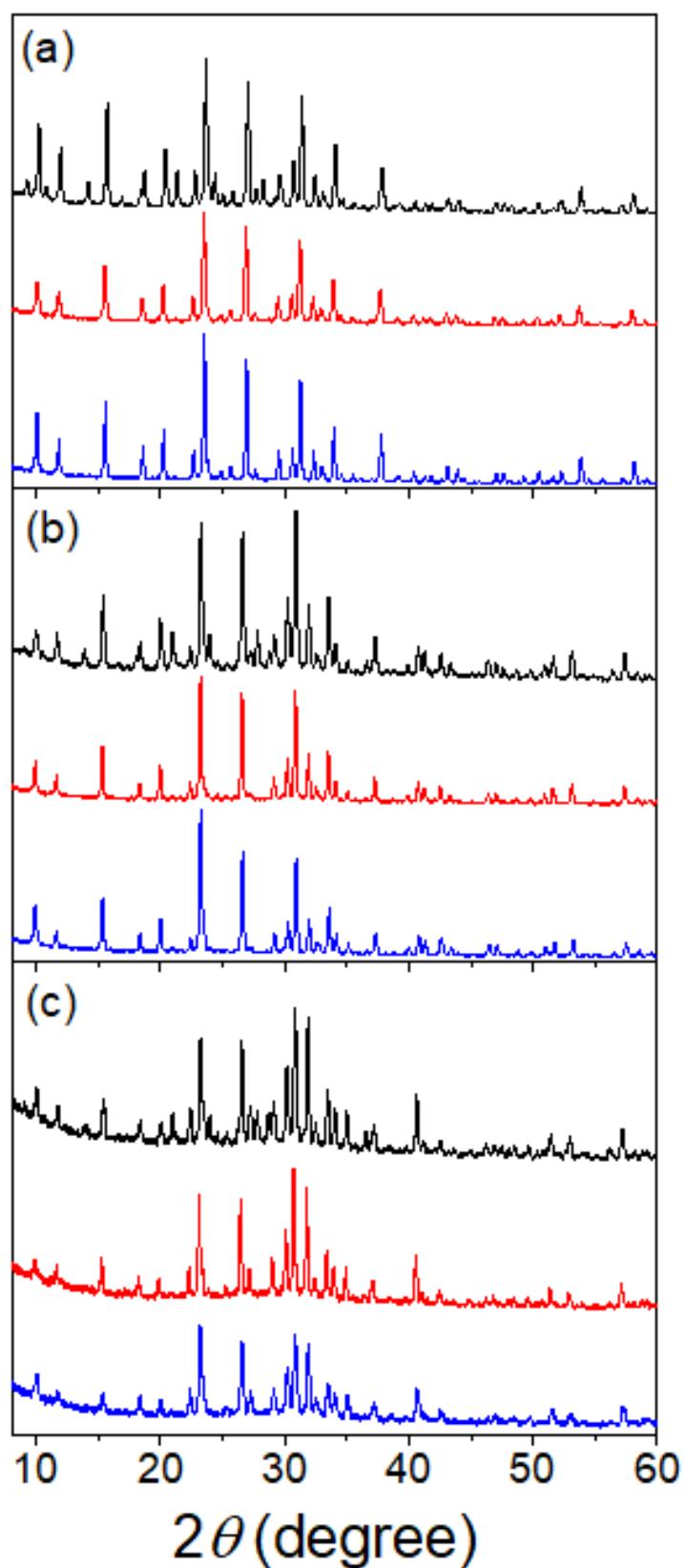


Figure S7. XRD patterns of (a) XZ-Li, (b) XZ-Na, and (c) XZ-K (black: Pristine zeolite, red: After the first reaction, blue: After the fifth reaction).

Table S1. Structural characterization of XZ-M.

Catalyst	Atomic ratio (%)					Surface area (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Crystallinity (%)
	O	Na	Al	Si	K			
XZ-Li	60.94	1.81	11.53	16.12	-	596	0.32	84.2
XZ-Na	64.65	10.26	10.47	14.62	-	541	0.27	87.7
XZ-K	61.11	-	11.56	16.23	11.10	465	0.26	78.9

Table S2. Recycling of XZ-M for the direct etherification of glycerol.

	Cycle No.	1	2	3	4	5
XZ-Li	Conversion of glycerol	86.9	58.8	87.2	83.1	81.2
	Yield of DG	61.2	60.1	61.0	58.2	56.8
	Yield of TG	21.2	20.6	20.9	19.9	19.5
XZ-Na	Conversion of glycerol	79.4	73.2	74.2	73.4	71.2
	Yield of DG	51.4	47.4	48.0	47.5	46.1
	Yield of TG	21.9	20.2	20.5	20.3	19.7
XZ-K	Conversion of glycerol	89.3	86.4	84.1	81.2	78.3
	Yield of DG	47.9	45.8	44.6	43.0	41.5
	Yield of TG	21.9	20.7	20.2	19.5	18.8