

## Supplementary Documents

# Glycerol Hydrogenolysis to Produce 1,2-propanediol in Absence of Molecular Hydrogen Using a Pd Promoted Cu/MgO/Al<sub>2</sub>O<sub>3</sub> Catalyst

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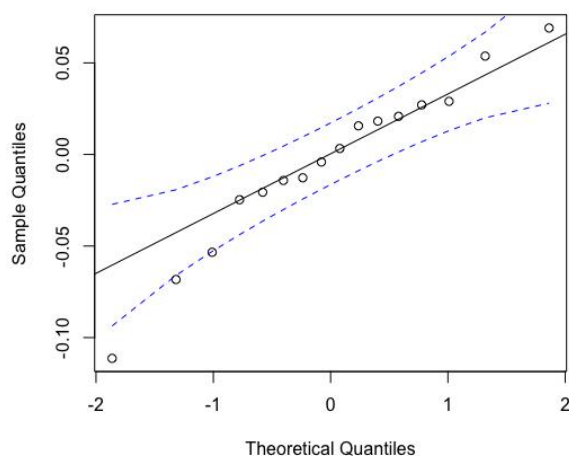
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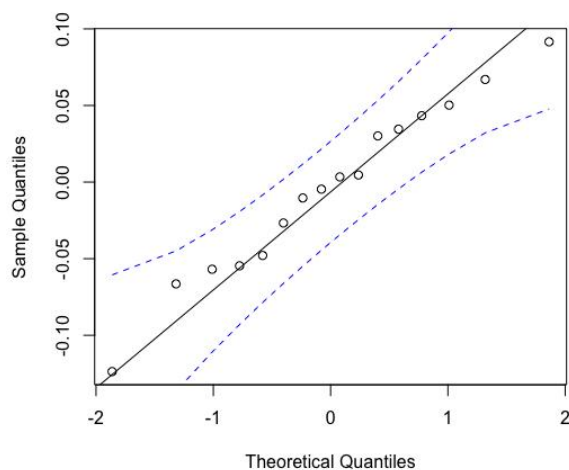
† Deceased November 2, 2018, this paper is dedicated to the memory of Professor Garry L. Rempel.

**Table S1** Criteria for a Fractional Factorial Design [1]

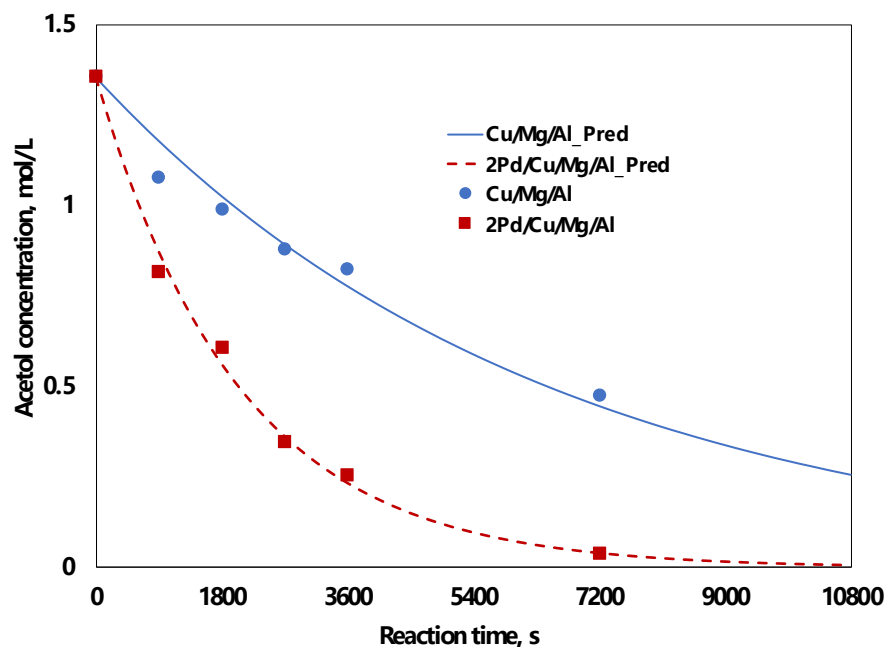
Number of Factors <i>k</i>	Fraction	Number of Runs	Design Generators	Number of Factors <i>k</i>	Fraction	Number of Runs	Design Generators
3	$2_{III}^{1-1}$	4	$C = \pm AB$	10			$H = \pm ABCG$
4	$2_{IV}^{1-1}$	8	$D = \pm ABC$				$J = \pm ACDE$
5	$2_{IV}^{1-1}$	16	$E = \pm ABCD$		$2_V^{-3}$	128	$K = \pm ACDF$
	$2_{III}^{-2}$	8	$D = \pm AB$				$G = \pm BCDF$
			$E = \pm AC$				$H = \pm ACDF$
6	$2_{VI}^{-1}$	32	$F = \pm ABCDE$				$J = \pm ABDE$
	$2_{IV}^{-2}$	16	$E = \pm ABC$		$2_{IV}^{-4}$	64	$K = \pm ABCE$
			$F = \pm BCD$				$F = \pm ABCD$
	$2_{III}^{-3}$	8	$D = \pm AB$				$G = \pm ABCE$
			$E = \pm AC$				$H = \pm ABDE$
			$F = \pm BC$				$J = \pm ACDE$
7	$2_{VII}^{-1}$	64	$G = \pm ABCDEF$		$2_{IV}^{-5}$	32	$K = \pm BCDE$
	$2_{IV}^{-2}$	32	$E = \pm ABC$				$E = \pm ABC$
			$G = \pm ABDE$				$F = \pm BCD$
	$2_{IV}^{-3}$	16	$E = \pm ABC$				$G = \pm ACD$
			$F = \pm BCD$				$H = \pm ABD$
			$G = \pm ACD$				$J = \pm ABCD$
	$2_{III}^{-4}$	8	$D = \pm AB$	11	$2_{III}^{-6}$	16	$K = \pm AB$
			$E = \pm AC$				$G = \pm CDE$
			$F = \pm BC$				$H = \pm ABCD$
			$G = \pm ABC$				$J = \pm ABF$
8	$2_V^{-2}$	64	$G = \pm ABCD$				$K = \pm BDEF$
			$H = \pm ABEF$		$2_{IV}^{-5}$	64	$L = \pm ADEF$
	$2_{IV}^{-3}$	32	$F = \pm ABC$				$F = \pm ABC$
			$G = \pm ABD$				$G = \pm BCD$
			$H = \pm BCDE$				$H = \pm CDE$
	$2_{IV}^{-4}$	16	$E = \pm BCD$				$J = \pm ACD$
			$F = \pm ACD$				$K = \pm ADE$
			$G = \pm ABC$		$2_{IV}^{-6}$	32	$L = \pm BDE$
			$H = \pm ABD$				$E = \pm ABC$
9	$2_{VI}^{-2}$	128	$H = \pm ACDFG$				$F = \pm BCD$
			$J = \pm BCEFG$				$G = \pm ACD$
	$2_{IV}^{-3}$	64	$G = \pm ABCD$				$H = \pm ABD$
			$H = \pm ACEF$				$J = \pm ABCD$
			$J = \pm CDEF$				$K = \pm AB$
	$2_{IV}^{-4}$	32	$F = \pm BCDE$		$2_{III}^{-7}$	16	$L = \pm AC$
			$G = \pm ACDE$				
			$H = \pm ABDE$				
			$J = \pm ABCE$				
	$2_{III}^{-5}$	16	$E = \pm ABC$				
			$F = \pm BCD$				
			$G = \pm ACD$				
			$H = \pm ABD$				
			$J = \pm ABCD$				



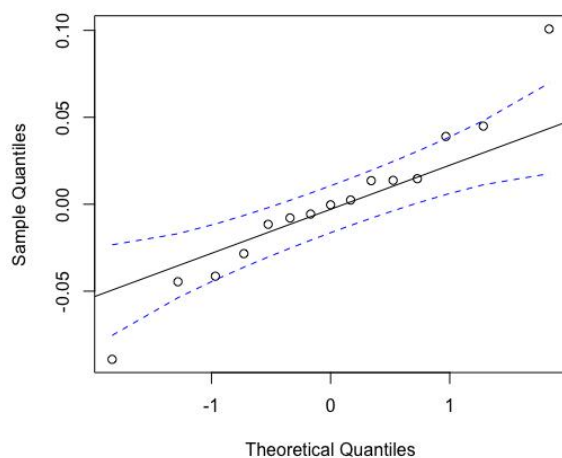
**Figure S1** The normal probability plot of residual for the glycerol hydrogenolysis process with in situ hydrogen from methanol steam reforming. Catalyst: 1Pd/Cu/MgO/Al<sub>2</sub>O<sub>3</sub>, Cu/Mg/Al (molar) = 22.5/67.5/10. Conditions: 220 °C, 15bar N<sub>2</sub>, 20 wt% Glycerol, Water/Methanol(molar)=1.2, 3 wt% catalyst, 500RPM.



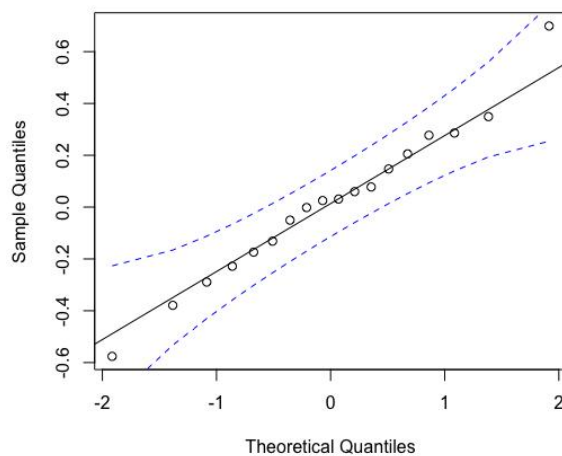
**Figure S2** The normal probability plot of residual for the glycerol hydrogenolysis process with in situ hydrogen from methanol steam reforming. Catalyst: Cu/MgO/Al<sub>2</sub>O<sub>3</sub>, Cu/Mg/Al (molar) = 22.5/67.5/10. Conditions: 220 °C, 15bar N<sub>2</sub>, 20 wt% Glycerol, Water/Methanol(molar)=1.2, 3 wt% catalyst, 500RPM.



**Figure S3** The concentration of different component over the reaction time for acetol hydrogenation.



**Figure S4** The normal probability plot of residual for the acetol hydrogenation. Catalyst: Cu/MgO/Al<sub>2</sub>O<sub>3</sub>.



**Figure S5** The normal probability plot of residual for the acetol hydrogenation. Catalyst:  $2\text{Pd/Cu/MgO/Al}_2\text{O}_3$ .

## References

- [1]. Montgomery, D.C.; Runger, G.C. In *Design of experiments with several factors*; Applied statistics and probability for engineers; John Wiley & Sons, Inc.: 2002; pp 560.