



Article Environmental and Technological Assessment of Operations for Extraction and Concentration of Metals in Electronic Waste

Josinaldo Dias ^{1,*}^(D), Angelus Giuseppe Pereira Silva ², José Nilson França de Holanda ², Fabíola Martins Delatorre ¹^(D), Amanda Oliveira da Conceição ¹, Gilson Mendonça de Miranda Júnior ¹, Aurora Futuro ³^(D) and Sílvia Cardinal Pinho ³

- ¹ Agricultural Sciences and Engineering Center, Federal University of Espírito Santo, Alegre 29500-000, Brazil
- ² Laboratory of Advanced Materials, State University of the Northern Rio de Janeiro, Campos dos Goytacazes 28013-602, Brazil; angelusdasilva@gmail.com (A.G.P.S.); holanda@uenf.br (J.N.F.d.H.)
- ³ LEPABE, Department of Metallurgical and Materials Engineering, Faculty of Engineering, University of Porto, 4169-007 Porto, Portugal; scpinho@feup.pt (S.C.P.)
- * Correspondence: josinaldo.dias@ufes.br

Abstract: The exponential growth in the consumption of electronics, in combination with the reduction in their useful life, has led to a significant increase in the volume of electronic materials which are discarded. Printed Circuit Boards (PCBs) are modules composed of ceramic, polymer, and metallic materials of high economic value and with a great potential for damage to biotic and anthropic environments when inadequately discarded. The aims of this work were to study the main environmental impacts and the efficiency of mechanical operations in the electronic waste recycling process in optimizing the concentration of metallic copper. For this, the samples obtained were characterized according to their morphology and chemical composition, and subjected to physical and mechanical treatments: dismantling, grinding, separation by granulometry, density, and magnetic property. The environment impacts were estimated by the Life Cycle Assessment of the pre-treatment processes, associated with copper extraction operations in Waste Printed Circuit Boards (WPCBs). According to the results obtained, it is possible to infer that the NM + 1 mm Concentrated product presents in its metallic composition around 78% copper, which contributes to the efficiency of the hydrometallurgical extraction processes. It is noteworthy that the Concentrated class represents about 14% of copper (wt) taking into account the amount of 3.789 kg of PCB waste input material. The potential for reinsertion of the dust in the material recovery process is highlighted, given that the chemical composition of this particulate has a significant presence of metals. In general, it was found that due to the higher concentration (wt%) of the sample being allocated in the range of -0.15 mm + 0.05 mm, the elements Fe (18.30%), Si (10.73%), Ca (9.21%), and Cu (8.89%) stand out regarding the participation of the elemental composition of this fraction and also regarding the general composition of the sample. Furthermore, the generation of dust during the fragmentation process can be associated with occupational respiratory diseases when not managed. About the Life Cycle Assessment, in seven of the nine categories evaluated, crushing and screening activities accounted for more than 90% of the recorded impact values. In general, it is estimated that in the pre-treatment phase it generated 15.4 kg CO₂ eq.

Keywords: electronic waste; waste management; copper recovery

1. Introduction

The recycling of electronic waste has become a goal for extractive industries due to the increase in the quantity and economic value associated with the recovery of materials. Urban mining, as the electronics recycling activity is called, allows for obtaining more precious metals than the conventional mining process [1].



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). Printed Circuit Boards (PCBs) are widely found as components of electronic devices, being composed of several metallic, polymeric, and ceramic materials [2].

According to [3] the elementary composition of PCBs varies according to the type and its applications. In general, PCBs contain approximately 28% metals, 23% polymers, and the remaining percentage corresponds to ceramic materials. About 10–20% of the PCB is made of copper, which forms the conductive layer for electrical connection between different components. Precious metals, especially Au and Pd, are used as contact materials in joints. Typical Pb/Sn solders, which are used to connect surface components, represent 4 to 6% of the total board mass. The components affixed to the PCBs also have different metals in their compositions, including: Ga, In, Ti, Si, Ge, As, Sb, Se, Te, Ta, etc. [4,5].

Due to the diverse and complex nature of PCB waste, the characterization in terms of type, structure, components, and composition becomes important to establish the process for recycling [6]. The recovery of metals from electronic waste requires the execution of pre-treatment initially, aiming at releasing the metals existing inside the PCBs, which will later be extracted by metallurgical processes such as hydrometallurgy, biohydrometallurgy, or pyrometallurgy.

The dismantling of different electronic components fixed superficially on the PCBs is a very important step in the recycling process [7]. After that, the steps for processing PCB waste are basically composed of four distinct phases, namely [8,9]: comminuting, particle size classification, magnetic separation, and electrostatic separation.

Ref. [10] also states that during the PCB metal recovery process, there is a heterogeneous composition that it presents, which dictates the careful need to select recovery methods and parameters, such as the hydrometallurgical and biohydrometallurgical processes that demand unique chemical/biochemical leaching characteristics (oxi-reduction potential, pH, ionic strength, bacterial strain, mode of agitation, etc.).

According to [11,12], toxic components, especially Pb, Cd, Hg, As, and Cr, are significantly present in PCBs. Ref. [13] reported that electronic waste recycling workers had chromosomal alterations in a proportion twenty times greater than those who were not exposed to electronic waste. It confirms the high degree of toxicity of these elements to human health.

Thus, the recycling of materials from PCBs represents the mitigation of environmental and social impacts, ensuring that it is tangible to obtain recoverable metal concentrates.

Considering that any system, product, process, or activity generates environmental impacts from the extraction of natural resources, until the end of their useful life. They are returned to the environment in the form of aqueous (effluents) and atmospheric emissions, waste, or energy. Life Cycle Assessment (LCA) is used and characterized by its excellence in evaluating the environmental performance of products and services [14].

This study aims to assess the efficiency of mechanical operations in the electronic waste recycling process in optimizing the concentration of metallic copper, considering the use of density separation by shaker table, and the environment impacts associated.

Thus, the evaluated hypothesis consists of using a pre-treatment route to enrich the metallic fraction of copper, reducing the environmental impacts associated with the recovery of metals in WPCBs.

This work presents its relevance in studying a physical and mechanical treatment route which contributes to the improvement of recovery parameters, such as material grain size and selection of the hydrometallurgical route most suitable for the metal of interest. Furthermore, the work contributes to the characterization of the particulate material, defined as dust, obtained during the electronic waste fragmentation stage. The composition of this by-product is relevant for the identification of the main component metals, in addition to stimulating the insertion of this material in the metal recovery process. In addition, the work is important in studying the environmental impacts associated with the metal recovery route considered in this study, creating opportunities for environmental improvements in copper extraction processes in electronic waste.

2. Materials and Methods

2.1. Materials

In order to ensure the uniformity of the sample, PCBs from computers, also known as motherboards, were used in this study. However, the boards may vary in terms of their basic composition, even if they are obtained from the same type of electronic equipment. Thus, the Stemi 2000—C Stereo Microscope was used to identify and characterize the boards by image, according to their internal structure.

The materials used in the study consist of PCB samples submitted to the dismantling process for the removal of plastic components. For the internal characterization of the studied sample, the Stemi 2000—C Stereo Microscope was used to analyze the material after a transverse section.

2.2. Physical and Mechanical Processing

The steps adopted by this methodology consisted of: (i) manual collection and dismantling; (ii) comminuting; (iii) particle size classification; (iv) separation by density; (v) magnetic separation.

The PCBs' waste was cut into 5 cm² pieces using a guillotine. Then, they were ground using a hammer shredder (EWZ M400) with an 8 mm diameter grid (primary milling), and a shredder of RETSCH SM 200 knives with a 4 mm diameter grid for smaller sizes (secondary crushing). After this process, it was possible to obtain a particulate sample, which was subjected to selection according to the size of its particles, comprising the intervals +1 mm, -1 mm + 0.45 mm, -0.45 mm. The classification was carried out with the aid of a vibrating stirrer (Retch AS 200).

Following the sample's separation by granulometry, the material was subjected to dynamics of separation by density, allowing the obtainment of four distinct product classes in terms of granulometry and composition of materials. Each of the classes was identified as follows: Concentrated, Mixed, Sterile, and Organic Matter. The Wilfley shaker table is characterized by an inclined surface, which has linear extensions called rifles. There is an asymmetric movement in the longitudinal direction and a film of water that crosses the table surface transversely (Figure 1).



Table movement

Figure 1. Density separation scheme by the Wilfley shaker table.

After separation by the shaker table, a manual magnetic device was used to classify the materials in terms of their magnetic characteristics. Products that presented magnetic characteristics were allocated to the Magnetic materials group (M), whereas products that did not present magnetic properties, or that presented low magnetic capacity, were allocated to the group of non-magnetic materials (NM).

Physical and Chemical Characterization of Waste Printed Circuit Boards Dust

Due to the generation of dust during the fragmentation process, this material was characterized, captured through paper filters with diameter of 37 mm, and moistened with water. The filters were then submitted to a lab dryer for drying for a period of 24 h, at a temperature of 50 °C. Then, the scraping procedure was performed on each filter used, accumulating all the material in a single sample.

The purpose of this procedure was to characterize physically and chemically the particulate material, with low granulometry, which was in suspension inside the fragmentation equipment. For the characterization of the general dust sample, the Scanning Electron Microscopy—SEM examination was performed, using the environmental Scanning Electron Microscope, high resolution (Schottky), and also the Energy Dispersive Spectroscopy—EDS with Microanalysis by X-ray and Electron Diffraction Pattern Analysis: Quanta 400 FEG ESEM/EDAX Genesis X4M. The sample was coated with a thin film of Au/Pd, by sputtering, using the SPI Module Sputter Coater equipment.

2.3. Physical and Chemical Characterization of Waste Printed Circuit Boards

After the physical and mechanical processing phase, the samples were segmented according to magnetic property, particle size, and product class (defined after the density separation process). With the aid of the Stemi 2000—C Stereo Microscope, the samples were analyzed visually, allowing the previous selection of the material of interest in this study. In order to collaborate with the visual analysis, elementary characterization of the samples was also carried out based on X-Ray fluorescence using the Oxford X-MET 7000 equipment. The Concentrated and Mixed classes were prioritized, due to the greater potential for concentration in metals, especially copper.

For the quantification of metals, the technique of submitting the samples to a chemical solution composed of nitric acid (HNO₃) and hydrochloric acid (HCl) in the ratio of 1:3—by [15] was used. Chemical digestions were performed in triplicates with the aid of *Kjeldahl* tubes, following the solid/liquid ratio of 1 g/15 mL, runtime 3 h, at 90 °C. After cooling to room temperature, the solutions were filtered and diluted with deionized water to a volume of 100 mL.

After digestion in aqua regia and filtering, the supernatant residue retained on paper filters was taken to an oven for 24 h at 50 °C. Then, the residual material was weighed, and the initial weight of the sample was subtracted from the weight of the residual material. It should be noted that the class of insoluble materials comprises polymeric and ceramic materials, whereas the materials solubilized during the process correspond to metallic materials.

The quantification of copper was determined by Atomic Absorption Spectrometry using the UNICAM 969 AA spectrometer equipment. For analysis of the organic fraction of the sample, Total Carbon (TC) was determined with a TC analyzer (TOC-V, SHIMADZU), by the [16].

2.4. Life Cycle Assessment—LCA

The objective of this LCA study was to estimate the Life Cycle Impacts of the pretreatment processes, associated with Cu extraction operations in WPCBs using the standards and general principles of the [17].

Regarding the definition of the system boundary for the evaluation, the pre-treatment was considered. The boundary of the system initially studied comprises two sets of processes, as shown in Figure 2: Process I (crushing and sieving operations) and Process II (density separation and magnetic separation).



Figure 2. System boundary considered for LCA of the pre-treatment process.

Taking into account that the pre-treatment phase is common to all downstream processes, the LCA of this set of processes was carried out in isolation in order to assess their specific environmental impact. Thus, the Functional Unit was established in terms of obtaining 1 kg of product CNM + 1 mm (with a concentration of 78% Cu), since this was the input product for the leaching processes that took place posteriori.

In the Life Cycle Inventory related to each process, as shown in Tables 1 and 2, data were collected, measured, calculated or estimated, allowing the quantification of inputs and outputs, which contributed to the LCIA. The software used to support this work was SimaPRO[®] 8.0, which allowed obtaining additional information by consulting the Ecoinvent 3.0 database.

Process I		
Description	Value	Unit
Inputs (materials and energy)		
WPCB	7.16	kg
Energy	29.5	kWh
Output		
Shredded PCB + 1 mm	4.33	kg
Shredded PCB – 1 mm	2.51	kg
Waste for treatments—Dust/losses	0.32	kg

Table 1. Process I—Life Cycle Inventory considering functional unit = 1 kg CNM + 1 mm (78% Cu).

Process II		
Description	Value	Unit
Inputs (materials and energy)		
Shredded PCB + 1 mm	4.33	kg
Energy	0.36	kWh
Water	0.33	m ³
Output		
CNM + 1 mm (78% Cu)	1	kg
Shredded PCB + 1 mm M	0.08	kg
Mixed shredded PCB + 1 mm	2.99	kg
Waste for treatment	0.25	kg
Effluents	0.32	m ³

Table 2. Process II—Life Cycle Inventory considering functional unit = 1 kg CNM + 1 mm (78% Cu).

3. Results and Discussion

The analysis based on the cross-section performed on the printed circuit board (Figure 3) allowed the visualization of four copper layers inside, which are interspersed with layers of fiberglass. In the outermost regions, it is possible to identify the presence of an overlay layer composed of epoxy resin. In view of the internal structure shown and the materials that compose it, it is possible to infer that the analyzed plate corresponds to the FR-4 model, which is indicated by the literature as the type of plate used mainly in the manufacture of computers and electronics.



Figure 3. Cross section of the FR-4 type PCB.

3.1. Physical and Mechanical Processing: Copper Concentration

The comminuting process started with the entry of approximately 4.5 kg of dismantled and guillotined material. In sequence, the material already crushed by the hammer shredder was submitted to the knife shredder which allowed the reduction of the sample grain size. The process occurred in duplicate.

During the crushing process, it was possible to record an estimated loss of about 0.192 kg of material. Thus, in general terms, it was found that the overall losses during the comminuting process correspond to about 4.3% of the total sample entering the process. These losses are associated with materials retained close to the internal structure of the equipment and materials that are easily crushed and end up dissipating in the form of dust during the process.

After the comminuting process, the material was classified according to seven particle sizes, respectively, +1 mm; -1 mm + 0.71 mm; -0.71 mm + 0.45 mm; -0.45 mm + 0.25 mm; -0.25 mm + 0.15 mm; -0.15 mm + 0.05 mm; and -0.05 mm. The granulometric separation took place with the aid of the vibrating separator (RETSCH AS 200) and the corresponding sieves.

According to Table 3, the distribution, in percentage of mass of the general sample, had its highest concentration in the particle size fraction +1 mm, representing about 64% of the total sample mass. The remaining 36% were distributed in the other fractions, not exceeding the representativeness of 10% of the total mass of the general sample in each fraction.

Particle Size Fraction	Yield (wt%)
+1 mm	63.27
-1 mm + 0.71 mm	7.38
-0.71 mm + 0.45 mm	9.24
-0.45 mm + 0.25 mm	5.42
-0.25 mm + 0.15 mm	3.67
-0.15 mm + 0.05 mm	5.93
-0.05 mm	5.09

Table 3. Samples particle size fraction obtained.

The dense liquid technique was used to separate solid particles through the difference in density. The particles that needed to be separated were immersed in a liquid that had a density value between the density values of the components that were to be separated. The heavier particles go to the bottom, while the lighter particles float on the surface. The process proves to be quite efficient, as it makes it possible to separate metals from polymers and ceramics.

However, to ensure greater efficiency in the density separation process, it was noticed through a pilot test that the samples of low granulometry, which had little representation in relation to the total mass of the overall sample, did not respond efficiently to the process. Thus, it was decided to group the general material into only three particle size fractions.

Thus, after separation by secondary particle size fraction, where the +1 mm, -1 mm + 0.45 mm, and -0.45 mm class intervals were adopted, the sample was subjected to density separation using a shaker table with a flow rate of $9.508 \times 10^{-6} \text{ m}^3/\text{s}$. Water was used as a liquid medium to carry out the density separation activity. Four product classes were obtained from this step: Concentrated I, Mixed (M), Sterile (S), and Organic Matter (OM).

The Concentrated class stands out for presenting greater potential for enriching metals through the process of separation by densities, since higher density materials, when compared to the density of the medium, follow the oscillatory movement of the table along the rifles, leading to mostly the same compartment.

The Mixed class has some relevance regarding the presence of metals. However, it is also characterized by having polymeric and ceramic materials, which makes this class less favorable to the enrichment of metals, when compared to the Concentrated class.

The Sterile and Organic Matter classes are characterized by the intense accumulation of ceramic and polymeric materials, i.e., with low metal enrichment potential. In this case, the materials had a density lower than the density of water, suffering immediate fluctuations during the process.

Together, the Concentrated and Mixed material classes corresponded to 60% of the total mass of the sample submitted to the density separation process. Table 4 shows the mass distribution (%) of the classes obtained in the density separation.

Table 4. Mass distribution of classes obtained in density separation.

Classes	Yield (wt%)
Concentrated	24
Mixed	36
Sterile	36
Organic matter	4

After separation by density, the general sample was segmented according to particle size fraction and class. Figure 4 shows the mass distribution of the overall sample, in percentage. The participation of +1 mm particle size in the composition of three of the four observed classes is highlighted. In addition, taking into account the Concentrated and



Mixed classes with +1 mm particle size, which have greater potential for metallic enrichment, it is clear that together they correspond to about 44% by mass of the overall sample.

Figure 4. Yield distribution (wt%) according to the class and particle size fraction of the sample.

Subsequently, magnetic separation was performed, through which two groups of materials were identified: Non-Magnetic Materials (NM) and Magnetic Materials (M). About 7% of the sample, considering the four classes, corresponded to materials with magnetic characteristics (M), whereas NM materials presented values around 93% of the mass of the general sample.

Based on the information shown in Figure 5, it was observed that the greatest presence of magnetic materials (M) was recorded in the particle size fraction—0.45 mm, reaching a value of 23% of the total (wt) of this particle size range. The size fraction +1 mm was mainly composed of Non-Magnetic Materials (NM), about 97% of the total weight of this sample.



Figure 5. Yield distribution (wt%) according to the size fraction and magnetic composition of the sample.

Taking into account the classes obtained after separation by density, it was found that in the Concentrated, Mixed, and Sterile fractions, the NM materials of particle size +1 mm predominated their composition, representing about 16%, 27%, and 21%, respectively, in relation to the mass of the overall sample (Figure 6).



Figure 6. Yield distribution (wt%) according to the size fraction, classes, and magnetic composition of the sample.

However, the Sterile and Organic Matter classes were characterized by low metal accumulation potential, according to the dynamics of the shaker table. Thus, the Concentrated and Mixed classes stood out due to their potential for the accumulation of metals, including copper (Cu), and these two classes are the central targets of this study. Figure 7 and Table 5 systematize the entire operational flow of the pre-treatment phase, showing the inventory of each operation.



Figure 7. Inventory of inputs and outputs in the pre-treatment phase.

Size Fraction	Concentrated Size Fraction (g)		Mi (Mixed (g)		Sterile (g)		Organic Matter (g)	
	Μ	NM	Μ	NM	Μ	NM	Μ	NM	
+1 mm	49.84	604.47	15.34	1009.81	2.13	785.06	-	44.04	
-1 mm + 0.45 mm	37	95.83	14.49	151.28	1.87	294.99	-	45.9	
-0.45 mm	55.16	19.75	59.73	122.25	16.65	314.17	15.8	33.56	
Total	142	720.05	89.56	1283.34	20.65	1394.22	15.8	123.5	

Table 5. Yield distribution (wt) according to granulometry, class, and magnetic characteristic.

The Concentrated and Mixed classes became the target of the study due to the significant presence of metals in their compositions. By selecting the Concentrated and Mixed classes, twelve different products were obtained, varying according to the particle size and the presence of magnetic properties or not, as shown in Table 6, where the total mass of the sample corresponded to 2234.95 g (the sum of the masses of all products in the Concentrated and Mixed classes).

Table 6. Yield distribution.

Class	Size Fraction (mm)	Product <i>i</i>	Magnetic Properties	wt (g)
	. 1	1	NM	604.47
	+1 mm -	2	М	49.84
	1	3	NM	95.83
Concentrated	-1 mm + 0.43 mm	4	М	37.00
	0.45	5	NM	19.75
	-0.45 mm	6	М	55.16
	. 1	7	NM	1009.81
	+1 mm	8	М	15.34
	1	9	NM	151.28
Mixed	-1 mm + 0.45 mm	10	М	14.49
		11	NM	122.25
	–0.45 mm	12	М	59.73

The samples selected for visual analysis and chemical composition were defined during the quartering process. As it is a material with great heterogeneity and different particle sizes, the need to ensure the greatest possible uniformity in the samples to be submitted for analysis is highlighted. Manual quartering consisted of reducing the mass of the samples by dividing the global sample into parts with lower mass in order to obtain the final sample according to the purpose of the assay.

The samples were sent for three complementary analyses, namely: quantification of the Total Carbon (TC); visual analysis of the physical structure of the constituent particles of each sample; chemical analysis using aqua regia digestion and Atomic Absorption Spectrometry. The result of the analysis estimated a low presence of carbon in the samples of the Concentrated class, reaching its highest value of 1.9% of Total Carbon, in the products CNM + 1 mm and CNM – 0.45 mm. It is noted, as expected after the density separation process, that the Mixed class had a greater presence of carbon in its samples, with its highest value, 24.1% of Total Carbon, in the MNM + 1 mm product. This is due to the greater presence of polymeric material in the Mixed class when compared to the Concentrated class. Figure 8 shows the values recorded for each product analyzed in relation to the concentration of Total Carbon (%) in the analyzed samples.



Figure 8. Total Carbon Concentration (TC) in the samples.

Thus, a significant increase in the existence of polymer and ceramic materials in the Mixed class can be seen when compared to the Concentrated Class, which was significantly recorded in the products MNM + 1 mm, MM + 1 mm, MNM - 0.45 mm, and MM - 0.45 mm.

Thus, the samples considered in the study were sent to chemical studies, carried out by applying the sample to the aqua regia solution and observing their reactions and behavior (Figure 9). Then, according to results obtained from the Atomic Absorption Spectrometry, the product Concentrated Non-magnetic (CNM) + 1 mm stood out, which presented 78% of the Cu concentration in the studied sample, according to the solubilized metallic fraction. The Mixed Non-magnetic (MNM) -1 mm + 0.45 mm product also presented relevant values for the copper concentration in the composition, about 67%. As expected, based on the characteristics exhibited by metallic copper, it was confirmed that the NM fraction exhibited an enrichment of this metal.



Figure 9. Quantification of insolubilized material during digestion by aqua regia.

It was also observed that the reduction in the presence of copper in the CNM fraction was associated with the sample's particle size, i.e., by reducing the particle size, the presence of copper in the CNM sample also reduced. The same effect does not occur with the MNM product, since the MNM - 1 mm + 0.45 mm presents the highest copper concentration value in this class. Analyzing Figure 10, it is possible to obtain the concentrations of the samples considered in the study.



Figure 10. Copper concentration in each analyzed product.

Given the representativeness of the Concentrated class for this study, the Atomic Absorption Spectrometry analysis was performed to quantify other metals in this class. Thus, Table 7 shows the metallic composition of each product in the Concentrated class, according to the presence of the following metals: copper, iron, zinc, nickel, silver, gold, manganese, and others.

Table 7. Composition of the metallic fraction of the Concentrated class.

Class and Size Fraction	Cu (%)	Zn (%)	Mn (%)	Au (%)	Fe (%)	Ag (%)	Ni (%)	Others (%)
CNM + 1 mm	78.48	15.00	0.008	0.01	0.04	0.02	0.21	6.23
CM + 1 mm	37.46	3.1	0.13	0.02	44.9	0.02	3.7	10.67
CNM – 1 mm + 0.45 mm	42.34	8.07	0.01	0.02	0.04	0.02	0.12	49.38
CM – 1 mm + 0.45 mm	37.30	0.35	0.09	0.05	31.6	0.01	2.83	27.77
CNM - 0.45 mm	37.36	1.46	0.03	0.07	0.06	0.02	0.14	60.86
CM - 0.45 mm	39.02	0.3241	0.11	0.07	55.3	0.04	1.16	3.97

Considering the quantification of existing metallic materials in each product (Figure 7) associated with the quantification of the Cu concentration in the samples (Figure 8), it was possible to estimate the Cu weight value, based on each product studied (Table 4). Thus, Equation (1) was obtained to calculate the total fraction of existing copper, considering all the samples studied:

$$TFCu = \sum_{i=1}^{12} (MF_i * CuF_i * wt_i)$$
⁽¹⁾

where,

$$TFCu = Total \ Fraction \ Cu \ (g) \tag{2}$$

$$MF_i = Metallic \ Fraction \ of \ sample \ i \ (\%)$$
(3)

$$CuF_i = Cu \ fraction \ of \ sample \ i \ (\%)$$
(4)

$$Wt_i = Weight of fraction i (g)$$
 (5)

Table 8 stratifies the Cu values by weight for each product according to the concentration obtained by Atomic Absorption Spectrometry. In view of these data, it became possible to infer that the Concentrated and Mixed classes, together, represent about 772 g of copper, which, in general terms, taking into account the amount of 3789.12 g of input material (eliminating losses and the pilot test), represents about 20% of copper in relation to the total weight of PCB waste used in the study. It is noteworthy that the Concentrated class alone presents 14% of the copper (wt) of the input sample.

Cu Fraction Size Fraction Magnetic Product Metallic **Cu Fraction** Fraction Metallic Class (mm) Fraction Fraction (%) (%) Weight (g) (i) Fraction (g) (g) 92 604.47 433.76 NM 1 78 556.11 +1 mm 2 Μ 93 37 49.84 46.35 17.15 NM 3 95 43 95.83 91.04 39.15 Concentrated -1 mm + 0.45 mm97 37 35.89 4 37.00 13.28 Μ NM 5 94 37 19.75 18.56 6.87 –0.45 mm 39 55.16 52.95 20.65 Μ 6 96 NM 7 55 26 1009.81 555.39 144.40 +1 mm Μ 8 59 18 15.34 9.05 1.63 9 NM 73 67 151.28 110.43 73.99 Mixed -1 mm + 0.45 mmΜ 10 83 31 14.49 12.03 3.73 122.25 50.12 NM 11 41 23 11.52 -0.45 mm 12 51 18 59.73 30.46 5.48 Μ **Total Fraction** 2234.95 1568.38 771.61

Table 8. Quantification of the accumulated copper fraction.

Thus, the CNM + 1 mm product has become of great interest for this system due to the potential concentration of metals, especially copper. Figure 11 shows the images related to the +1 mm particle size fraction, CNM. It is possible to identify the existence of particles that have a green hue on their surface with orange edges (highlighted in red). These particles possibly originate from the PCB substrate, which is summarily constituted by epoxy resin (covered by a solder mask), copper layers, deposited on the substrate, and fiberglass.



Figure 11. Stereo Microscope image of the CNM + 1 mm product: (**a**) image size 5 mm and (**b**) image size 2 mm.

By analyzing the orange-colored particles, possibly copper particles, it was possible to perceive different geometries. In addition to the Cu particles originating from the PCB substrate, it is possible to notice (Figure 11b) the presence of copper from PCB inductors, in a rod shape.

Considering all losses during the physical treatment process, the input sample for the chemical analysis corresponded to 3.789 kg. In view of these data, it was possible to infer that the Concentrated and Mixed classes, together, represent about 772 g of copper, which in general terms represents about 20% of copper in relation to the total of PCB waste (wt%) used in the study. It is noteworthy that the Concentrated class alone presents 14% of the copper (by mass) of the input sample.

Analysis of Dust Obtained in the Fragmentation Process

During the pre-treatment phase, specifically in the fragmentation phase, the production of particulate material with low granulometry was noted, which was defined as dust from the comminution process.

According to the generation of dust during the fragmentation process, which can present potential for reinsertion in the material recovery process as well as being associated with occupational respiratory diseases when not well managed, the characterization of this material, captured during the comminuting process by means of 37 mm diameter paper filters, was moistened with water. The purpose of this procedure was to characterize the particulate material physically and chemically, which was in suspension inside the fragmentation equipment.

As shown in Figure 12a,b, through the SEM analysis, the particles that compound the sample have diversified geometries. The elongated, rod-shaped particles are believed to be fiberglass particles, found mainly in the internal coating of the copper layers of PCBs.



Figure 12. Scanning Electron Microscopy. (a) Global image—point Z1; (b) Approximate image (500×)—points Z2, Z3, and Z4.

Thus, with the aid of the EDS analysis, it was possible to study the particles Z1, Z2, Z3, and Z4. Figure 13a shows the Z1 particle, which presents Ba and Ti in its composition. According to [18], this particle probably originates from a ferroelectric material that is frequently used by the electronics industry. This material is in the form of barium titanate (BaTiO₃), which is characterized by being a ceramic material that exhibits photorefractive effects and dielectric properties.



Figure 13. Electron Dispersion Spectroscopy (EDS): (a) point Z1; (b) point Z2; (c) point Z3; (d) point Z4.

According to [19], the substrate of the PCBs is mainly composed of C and O, in addition to metals such as Si and Al, that are released during the grinding process. According to Figure 13b obtained by the EDS test at point Z2, it is possible to infer that the analyzed particle possibly represents a part of the substrate of the PCB, since it presents the typical elementary characterization.

In view of the images obtained by the SEM (Figure 12b), it is possible to perceive the geometric peculiarity of the Z3 particle, where it is elongated. When observing the information generated from the EDS analysis (Figure 13c) in the Z3 particle, the presence of the elements Si, O, C, Al, and Ca is highlighted. In view of these characteristics, the Z3 particle is possibly a fiber of glass, where the elements mentioned above exist in the form of oxides such as SiO₂ and Al₂O₃.

In the analysis of point Z4 (Figure 13d), it is noted the relevant presence of Sn and Pb in the sample, because these elements are used frequently in the production of PCBs to weld other components to the surface of the board.

After characterization by images, the general dust sample was subjected to the separation process by granulometry fractions, where four class intervals were defined. Since the largest particle size presented by the sample was below 0.45 mm, the following classes were assigned: -0.45 mm + 0.25 mm, -0.25 mm + 0.15 mm, -0.15 mm + 0.05 mm, and -0.05 mm.

Figure 14 shows the distribution of the mass, in percentage, presented by each granulometry fraction in relation to the general sample. Approximately 70% of the entire sample obtained showed an interval granulometry of -0.15 mm + 0.05 mm.



Figure 14. Dust sample distribution (wt%).

With the support of the X-ray fluorescence equipment Oxford X-MET 7000, the elemental chemical analysis of each particle size fraction was obtained, which is shown in Figure 15.



Figure 15. Chemical composition (wt%) dust sample according to the size fraction (mm).

In general, it was found that due to the higher concentration (wt%) of the sample being allocated in the range of -0.15 mm + 0.05 mm, the elements Fe (18.30%), Si (10.73%), Ca (9.21%), and Cu (8.89%) stood out regarding the participation of the elemental composition of this fraction and also regarding the general composition of the sample. However, in a punctual analysis, it is possible to notice the inversely proportional tendency of the presence of Fe in relation to the particle size of the classes, that is, the smaller the particle size of the material, the greater the relative presence of Fe in the fraction, registering 2.34%

(wt) in particles with the largest grain size and reaching 29.70% (wt) in the class with the smallest particle grain size.

The opposite occurs with Cu, which registers its largest participation (28.38%) in the composition of the fraction with the largest particle size (-0.45 mm + 0.25 mm), decreasing its presence as the particle size also decreases.

3.2. Life Cycle Assessment

Analyzing the total values of the impacts generated in the pre-treatment phase for copper recovery in PCB waste, shown in Figure 16, it becomes possible to infer that the crushing and sieving operations, which make up process I, had a majority of participation in all impact categories evaluated. In seven of the nine categories evaluated, crushing and screening activities accounted for more than 90% of the recorded impact values. Table 9 presents the LCIA results for the pre-treatment phase. Thus, the importance of developing new technologies and methods in the pre-treatment phase is highlighted, which prioritize energy efficiency and value the recovery of the fraction of dust generated in this stage.



Figure 16. Life Cycle Impact Assessment for the pre-treatment phase.

Impact Category	Unit	Total	Crushing and Sieving	Density and Magnetic Separation
Climate change	kg CO ₂ eq	15.4	15.1	0.236
Fossil depletion	kg oil eq	4.1	4.04	0.0613
Occupation of agricultural land	m ² a	3.77	3.04	0.2
Human toxicity	kg 1.4-DB eq	3.24	3.04	0.2
Ionizing radiation	kBq U235 eq	3.24	3.04	0.2
Metal depletion	kg Fe eq	0.246	0.216	0.03
Ecotoxicity in fresh water	kg 1.4-DB eq	0.0979	0.0781	0.0198
Terrestrial ecotoxicity	kg 1.4-DB eq	0.000695	0.000676	$1.89 imes10^{-5}$
Ozone depletion	kg CFC-11 eq	$1.2 imes 10^{-6}$	$1.19 imes 10^{-6}$	$1.71 imes 10^{-8}$

Table 9. Life Cycle Impact Assessment of the pre-treatment phase.

In a particular analysis of Process I, it was found that the impacts associated with it are greatly influenced by the way of obtaining the energy used in this stage, especially by the energy matrix presented by each geographic area. This is due to the use of equipment that requires more energy consumption when compared to the activities that make up Process II.

On the other hand, the density separation operation, performed on a shaker table, required the use of water which generated effluents as an output of the process. In addition, waste disposed of in underground deposits contributed to Process II, reaching around 20% of the total share in the Freshwater Ecotoxicity Impact category.

The crushing process is necessary because it has the characteristic of releasing metals in the PCB, mainly copper, which is also present in the inner layers of the PCB. However, most of the materials processed in the grinding phase are of ceramic or polymeric origin, which makes subsequent processes for the recovery of metals in quantity and quality difficult.

It was possible to obtain a class of the sample with non-magnetic characteristics, granulometry greater than 1 mm, and a concentration of 78% Cu due to the separation operations carried out in Process II, which contributed to the accumulation of metals, in particular copper, in a specific fraction, facilitating hydrometallurgical or pyrometallurgical processes.

4. Conclusions

Therefore, it is concluded that density separation significantly contributes to increasing the efficiency of the pre-treatment phase and physical separation operations in the process of recovering materials from electronic waste.

Based on the results obtained, it is possible to infer that the NM + 1 mm concentrated product presents in its metallic composition around 78% copper, which contributes to the efficiency of the hydrometallurgical extraction processes.

According to the generation of dust during the fragmentation process, this can be associated with occupational respiratory diseases when not managed. The potential for the reinsertion of this by-product in the material recovery process is highlighted, given that the chemical composition of this particulate has a significant presence of metals.

The study highlighted the recovery of metals in electronic waste as a potential practice, due to its collaborative power to reduce product volumes in the end of life phase, in addition to providing extractive reduction of natural resources and mitigating environmental impacts associated with their final disposal of waste.

It should be noted that the processes carried out for the recovery of materials must be economically, socially, and environmentally viable, and their associated impacts must be continuously measured and monitored.

Considering the physical–mechanical processes of the pre-treatment phase for copper concentration in PCB waste, crushing and sieving operations are responsible for 90% of the environmental impacts associated with this system. This is strongly linked to the way energy is obtained and the energy efficiency of the equipment used.

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